Supporting Information for "<u>The influence of oxygen</u> vacancy and Ce³⁺ ion positions on the properties of small gold clusters supported on CeO_{2-x}(111)"

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1. Structures of Au clusters on CeO_{2-x}

Table S 1: Structures of Au_n/CeO_{2-X}

Entry	# Au atoms	top	side	V _o position
1	1			under
2	1			next
3	2			next
4	3			under/corner
5	3			next (a)
6	3			next (b)







Figure S 1: Vacancy positions next to the Au cluster of Au₃/CeO_{2-x}.

2. DOS and COHP plots



Figure S 2: DOS (top) and COHP (bottom) of Au₁/CeO_{2-x}; localisation of Ce³⁺ in 2₁ position; average COHP of all respective pairs within 4.0 Å.



Figure S 3: DOS of Au₁/CeO_{2-x}; localisation of Ce³⁺ in 1_1 position



Figure S 4: DOS of Au_1/CeO_{2-x} ; localisation of Ce^{3+} in 3_1 position.



Figure S 5: DOS (top) and COHP (bottom) of Au₂/CeO_{2-x}; average COHP of all respective pairs within 4.0 Å.



Figure S 6: DOS (top) and COHP (bottom) plot of Au₄/CeO_{2-x}; vacancy under Au₄; average COHP of all respective pairs within 4.0 Å.



Figure S 7: DOS (top) and COHP (bottom) plot of Au₈/CeO_{2-x}; average COHP of all respective pairs with a maximum distance of 4.0 Å; vacancy under central Au atom.



Figure S 8: DOS (top) and COHP (bottom) plot of Au_{θ}/CeO_{2-x} ; average COHP of all respective pairs with a maximum distance of 4.0 Å; vacancy under corner Au atom.



Figure S 9: DOS (top) and COHP (bottom) plot of Au_{10}/CeO_{2-x} ; vacancy under edge Au atom; average COHP of all respective pairs with a maximum distance of 4.0 Å.



Figure S 10: DOS (top) and COHP (bottom) plot of Au₁₁/CeO_{2-x}; average COHP of all respective pairs with a maximum distance of 4.0 Å.

2. Iso-surfaces



Figure S 11: Iso-surface of the electron density of the highest occupied band of Au₁/CeO_{2-x}; iso-value 0.003 e $Å^{-3}$.



Figure S 12: Iso-surface of the electron density of the second highest occupied band of Au₁/CeO_{2-x}; iso-value 0.003 e Å⁻³.



Figure S 13: Iso-surface of spin density (α - β) of Au₂/CeO_{2-x}; iso-value 0.003 e Å⁻³.



Figure S 14: Iso-surface of the lowest unoccupied band of Au₂/CeO_{2-x}; iso-value 0.003 e $Å^{-3}$.



Figure S 15 Iso-surface of the electron density of the highest occupied band of Au₃/CeO_{2-x}; vacancy under Au₃; iso-value 0.003 e $Å^{-3}$.



Figure S 16: Iso-surface of the electron density of the second highest occupied band of Au₃/CeO_{2-x}; vacancy under Au₃; isovalue 0.003 e $Å^{-3}$.



Figure S 17: Iso-surface of spin density (α - β) of Au₃/CeO_{2-x}; vacancy next to cluster; iso-value 0.003 e Å⁻³.



Figure S 18: Spin density (α - β) of Au₄/CeO_{2-x}; iso-value 0.003 e Å⁻³.



Figure S 19: Iso-surface of the electron density of the third highest occupied band of Au₄/CeO_{2-x}; vacancy under Au₄; isovalue 0.003 e $Å^{-3}$.



Figure S 20: Spin density (α - β) of Au₄/CeO_{2-x}; vacancy next to Au₄; iso-value 0.003 e Å⁻³.



Figure S 21: Spin density (α - β) of Au₇/CeO_{2-x}; vacancy under to Au₇; iso-value 0.003 e Å⁻³.

Au8 a-b density



Figure S 22: Iso-surface of the electron density of the fourth highest occupied band of Au₈/CeO_{2-x}; vacancy under corner atom; isovalue 0.003 e $Å^{-3}$.



Figure S 23: Spin density (α - β) of Au₁₁/CeO_{2-x}; vacancy under to Au₁₁; iso-value 0.003 e Å⁻³.

2. Electrostatic potential plots

Table S 2: Iso-surfaces of the electron density (iso-value: 0.001 a.u) mapped with the (electrostatic) local potential; colour scale between $\geq -2.2 \text{ eV}$ (blue) and $\leq -4.0 \text{ eV}$ (red); isolated Au₈ clusters; structures taken from Au₈/CeO_{2-x}.





Table S 3: Iso-surfaces of the electron density (iso-value: 0.001 a.u) mapped with the (electrostatic) local potential; colour scale between $\geq -2.0 \text{ eV}$ (blue) and $\leq -4.5 \text{ eV}$ (red); isolated Au₁₁ clusters; structures taken from Au₈/CeO₂ and Au₈/CeO_{2-x}.

Table S 4: Iso-surfaces of the electron density (iso-value: 0.001 a.u) mapped with the (electrostatic) local potential; colour scale between $\geq -2.0 \text{ eV}$ (blue) and $\leq -3.8 \text{ eV}$ (red); isolated Au₁₁ clusters; structures taken from Au₈/CeO₂ and Au₈/CeO_{2-x}.

