

Supporting Information

Melamine-assisted Synthesis of Fe₃N Featuring Highly Reversible Crystalline-Phase Transformation for Ultrastable Sodium Ion Storage

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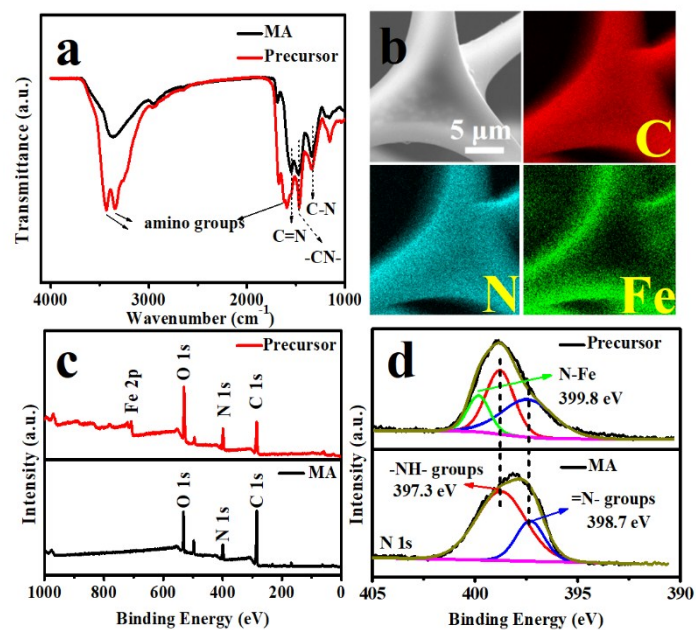


Fig. S1 (a) FTIR spectrum, (c) XPS survey spectrum and (d) high-resolution N 1s XPS spectrum of MA sponge and the precursor; (b) EDS mapping of the precursor.

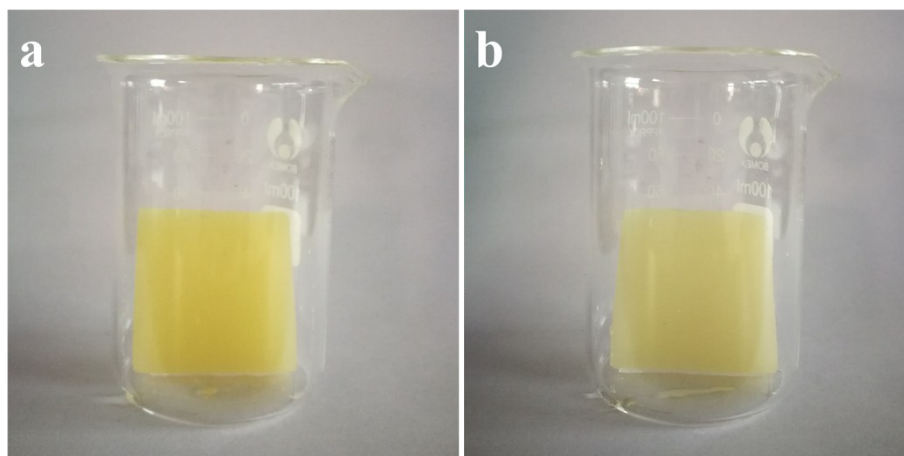


Fig. S2 Digital photos of the precursors before (a) and after (b) extensive sonication in deionized water.

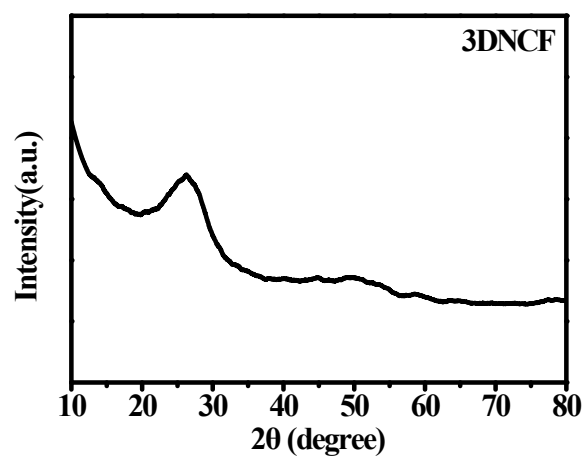


Fig. S3 XRD patterns of the 3DNCF.

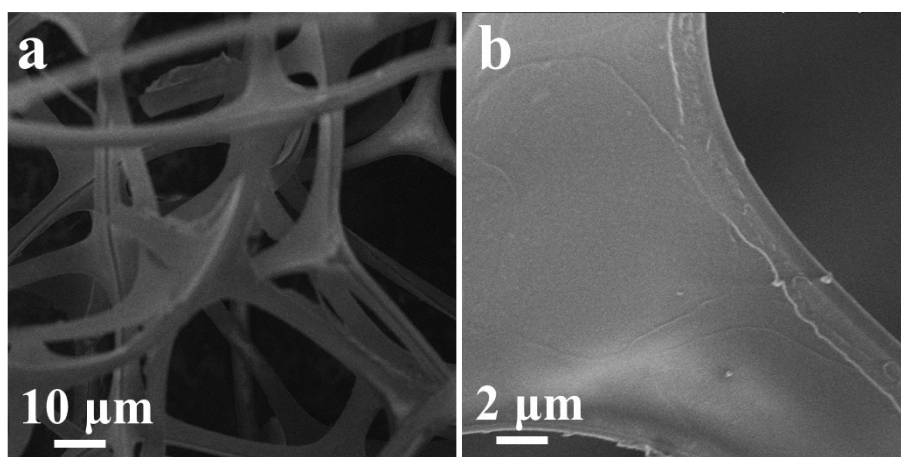


Fig. S4 SEM images of the 3DNCF.

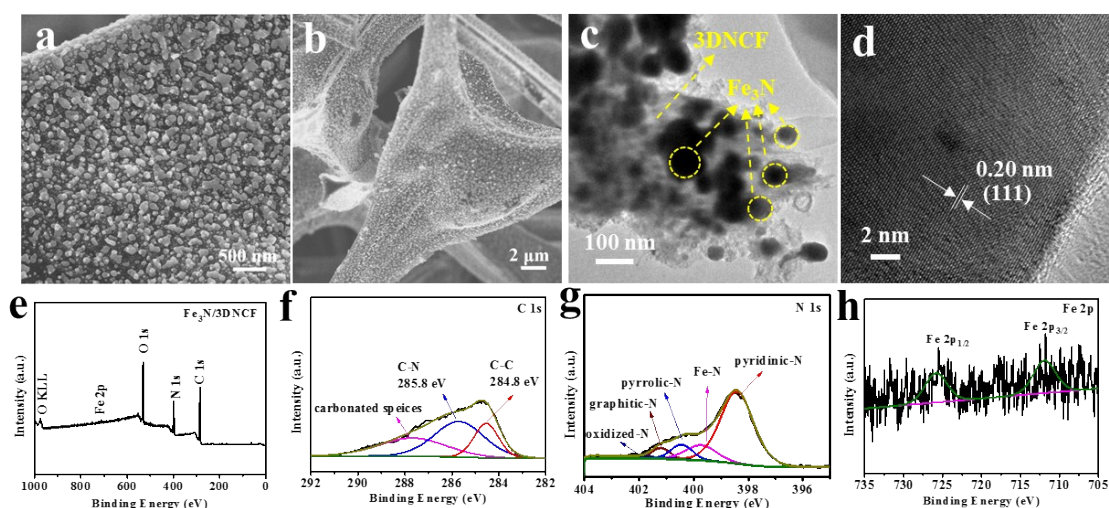


Fig. S5 SEM (a-b), TEM (c-d) images and (e) XPS survey spectrum and high-resolution elemental (f) C 1s, (g) N 1s and (h) Fe 2p spectra of $\text{Fe}_3\text{N}/3\text{DNCF}$ composite without adding glucose.

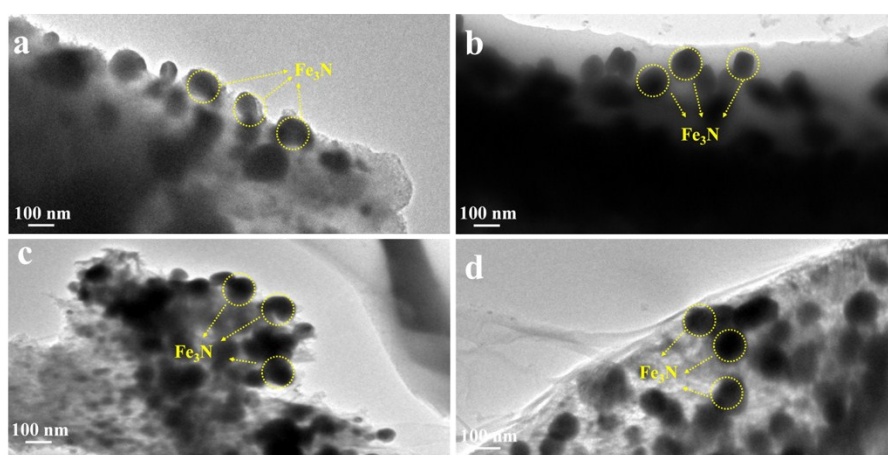


Fig. S6 *Ex situ* TEM images of $\text{Fe}_3\text{N}@C/3\text{DNCF}$ anode for the first cycle at a discharged state of a) 0.55 V and b) 0.01 V, and at a charged state of c) 1.0 V and d) 3.0 V.

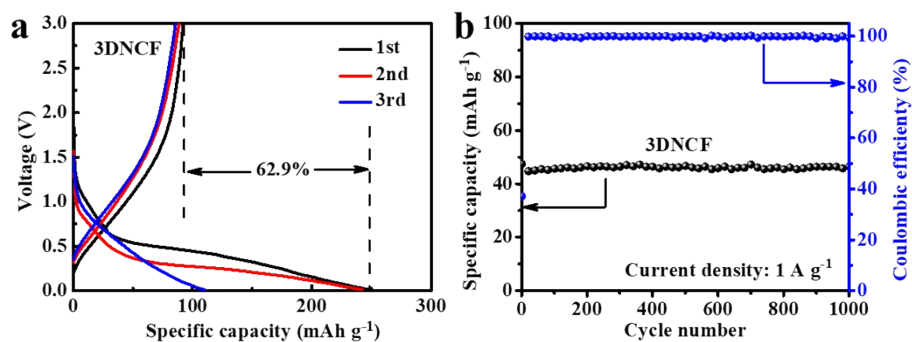


Fig. S7 (a) The galvanostatic charge and discharge voltage profile at 0.1 A g^{-1} , (b) cycle performances of 3DNCF at 1 A g^{-1} .

Table S1. ICP result of the $\text{Fe}_3\text{N}@C/3\text{DNCF}$ and $\text{Fe}_3\text{N}/3\text{DNCF}$ composites

Material	Analyte	Conc.Units	Fe (wt. %)	Fe_3N (wt. %)
$\text{Fe}_3\text{N}@C/3\text{DNCF}$	Fe	1.4 mg/L	25.38	27.5
$\text{Fe}_3\text{N}/3\text{DNCF}$	Fe	1.35 mg/L	24.37	26.4

Table S2 Comparison of synthesis approach and cycling performance of Fe₃N@C/3DNCF with previously reported metal nitrides anodes for SIBs.

	Materials	Synthesis approach	Capacity Retention	Reference
Ref. S2	G-VNQD-500	Hydrothermal and annealing at NH ₃ atmosphere	280 mA h g ⁻¹ after 800 cycles at 240 mA g ⁻¹	Adv. Energy Mater. 2016, 6(6): 1502067
Ref. S4	Fe ₃ N@C	Electrospinning and two step annealing	280 mA h g ⁻¹ after 300 cycles at 400 mA g ⁻¹	Adv. Mater. 2018, 1800525
Ref. S6	Mo ₂ N	Hydrothermal and annealing at NH ₃ atmosphere	180 mA h g ⁻¹ after 300 cycles at 500 mA g ⁻¹	Energy Storage Materials, 2019, doi: 10.1016/j.ensm.2019.04.007
Ref. S1	VNQD@NC HSs	Hydrothermal and annealing at NH ₃ atmosphere	306 mAh g ⁻¹ after 1400 cycles at 1000 mA g ⁻¹	J. Mater. Chem. A, 2019, 7(15): 9289-9296
Ref. S3	Sn ₃ N ₄	two step ammonolysis process and annealing at NH ₃ atmosphere	152 mA h g ⁻¹ after 50 cycles at 200 mA g ⁻¹	J. Mater. Chem. A, 2016, 4, 5081
Ref. S5	Ni ₃ N	Solution-based method and annealing at NH ₃ atmosphere	80 mA h g ⁻¹ after 20 cycles at 423 mA g ⁻¹	J. Mater. Chem. A, 2013, 1, 6441–6445
	Fe ₃ N@C/3DNCF	Solution-based method and one step annealing	375 mA h g ⁻¹ after 2000 cycles at 1000 mA g ⁻¹	Our work

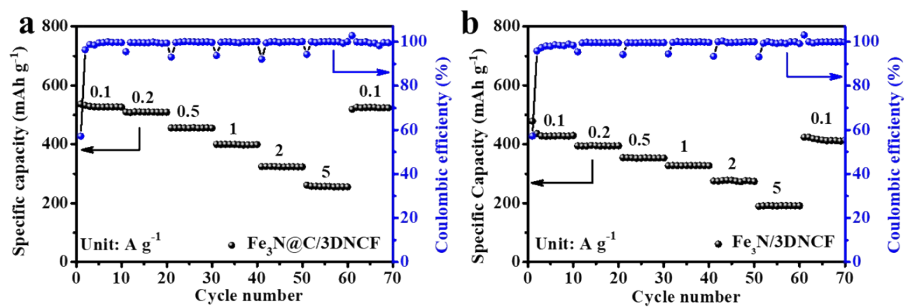


Fig. S8 The rate performance of (a) Fe₃N@C/3DNCF and (b) Fe₃N/3DNCF.

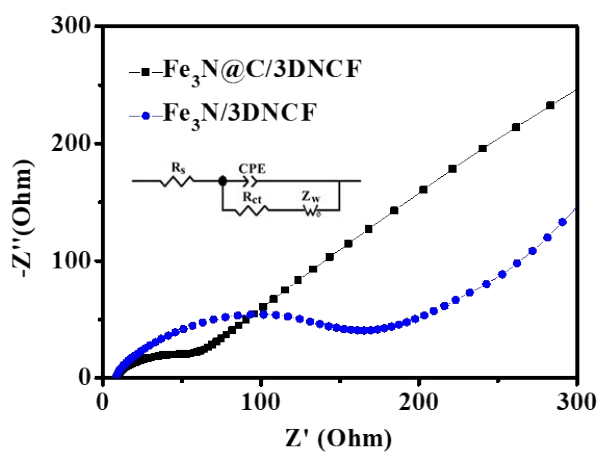


Fig. S9 Recorded electrochemical impedance spectra of Fe₃N@C/3DNCF and Fe₃N/3DNCF. The inset is the corresponding equivalent circuit.

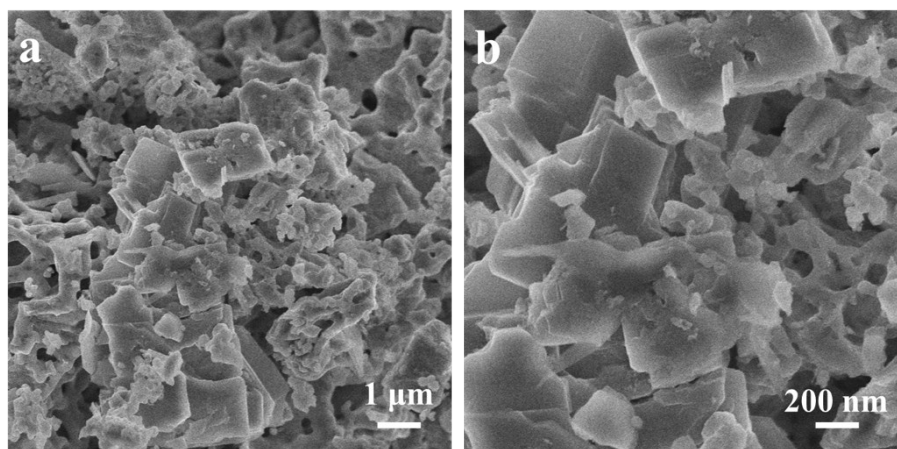


Fig. S10 SEM images of Na₃V₂(PO₄)₃@C composites.

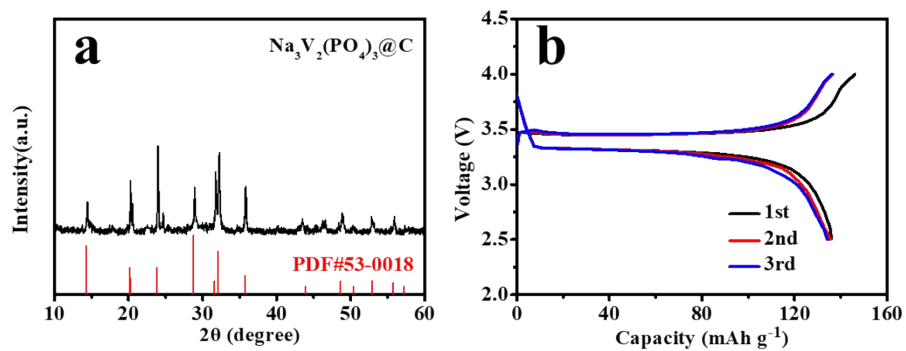


Fig. S11 (a) Typical XRD and (b) charge-discharge profiles of the $\text{Na}_3\text{V}_2(\text{PO}_4)_3@\text{C}$ at 0.1 A g^{-1} .

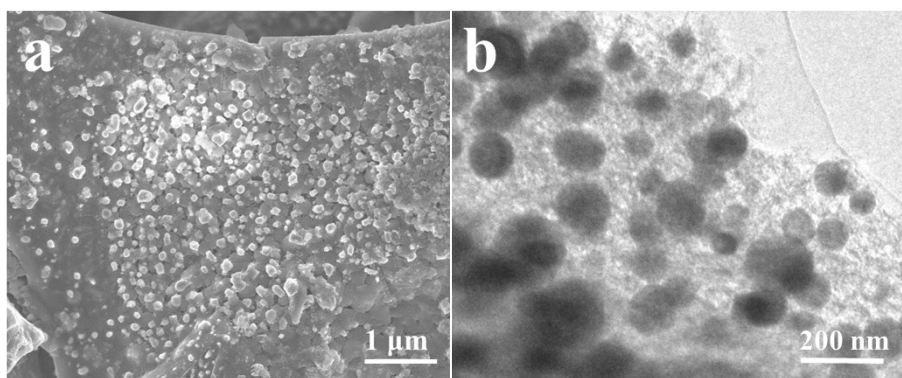


Fig. S12 The SEM and TEM images of $\text{Fe}_3\text{N}@\text{C}/3\text{DNCF}$ after 2000 cycles.

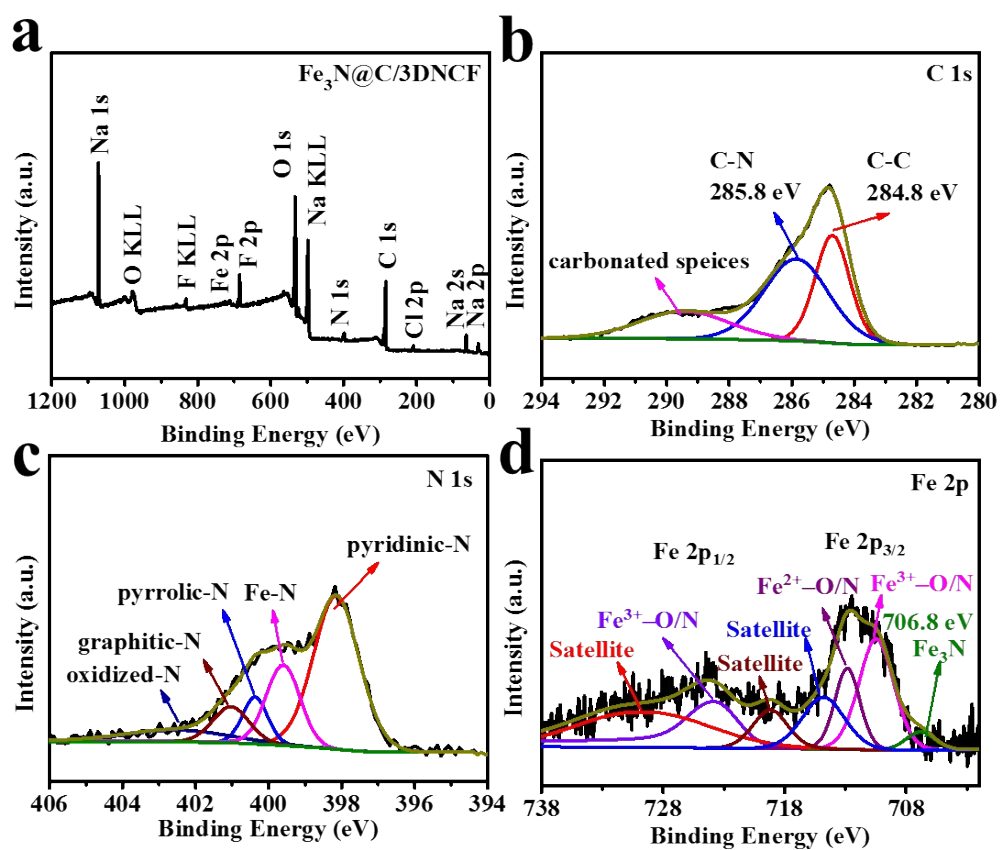


Fig. S13 High-resolution XPS spectra of $\text{Fe}_3\text{N@C/3DNCF}$ after 2000 cycles. (a) Survey XPS spectra. High-resolution XPS spectra: (b) C 1s; (c) N 1s and (d) Fe 2p spectrum of $\text{Fe}_3\text{N@C/3DNCF}$.