

Electronic supplementary information

Ferroelectric polarization and thin-layered structure synergistically promoting CO₂ photoreduction of Bi₂MoO₆

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Table S1. The crystal lattice parameters of the series of Bi₂MoO₆ photocatalysts.

Sample	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	Vol (Å ³)
BMO	5.48338	16.20732	5.49011	487.91
BMO-P	5.49058	16.19847	5.48245	487.60
BMO-U	5.49556	16.20143	5.48236	488.13
BMO-U-P	5.49223	16.21196	5.48442	488.33

Table S2. Comparison with related photocatalysts for CO₂ reduction.

Photocatalyst	Co-catalyst	Product	R _{max} ^[a]	Ref.
Bi ₂ MoO ₆ -U-P	-	CO	14.38	This work
Bi ₂₄ O ₃₁ Cl ₁₀ -OV	-	CO	0.90	1
BiOI	-	CO	5.18	2
Mg-In LDH	-	CO	4.00	3
In ₂ Ge ₂ O ₇ (En)	Pt	CO	0.90	4
β -Ga ₂ O ₃	-	CO	1.46	5
CuGaO ₂	-	CO	9.00	6
TiO ₂	Rh	CO	5.10	7

[a] Maximum formation rate reported for the product ($\mu\text{mol g}^{-1} \text{h}^{-1}$).

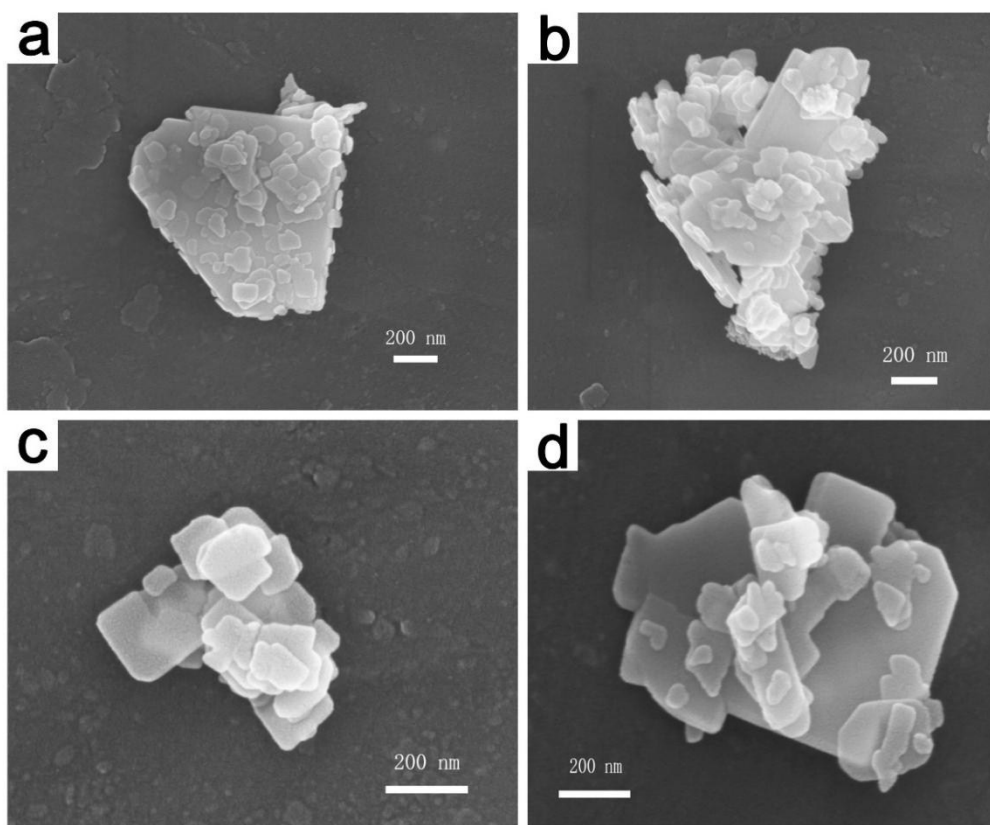


Figure S1. The SEM images of (a) BMO, (b) BMO-P, (c) BMO-U and (d) BMO-U-P samples.

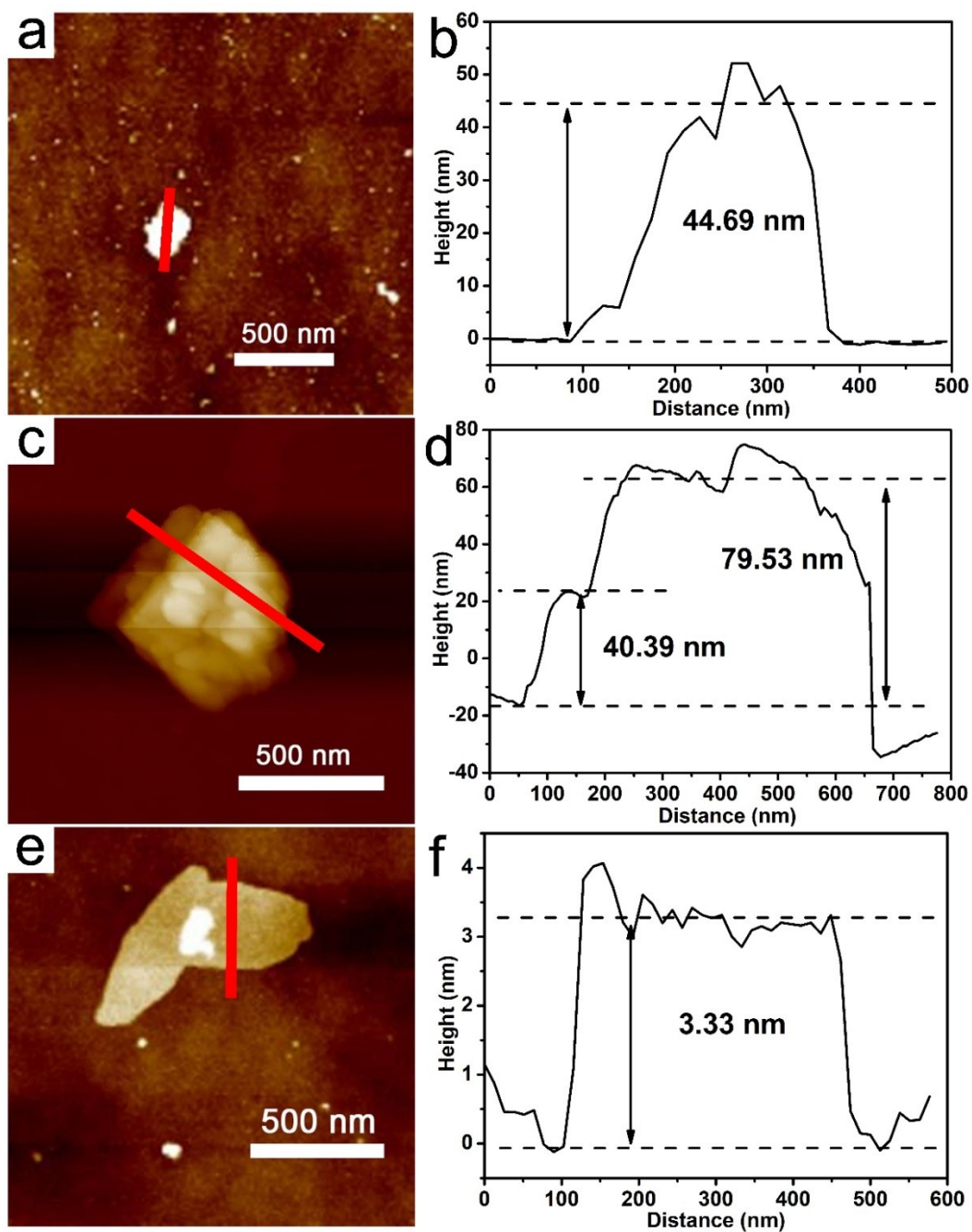


Figure S2. (a) The AFM image and (b) the corresponding height graph of BMO. (c) The AFM image and (d) the corresponding height graph of BMO-P. (e) The AFM image and (f) the corresponding height graph of BMO-U.

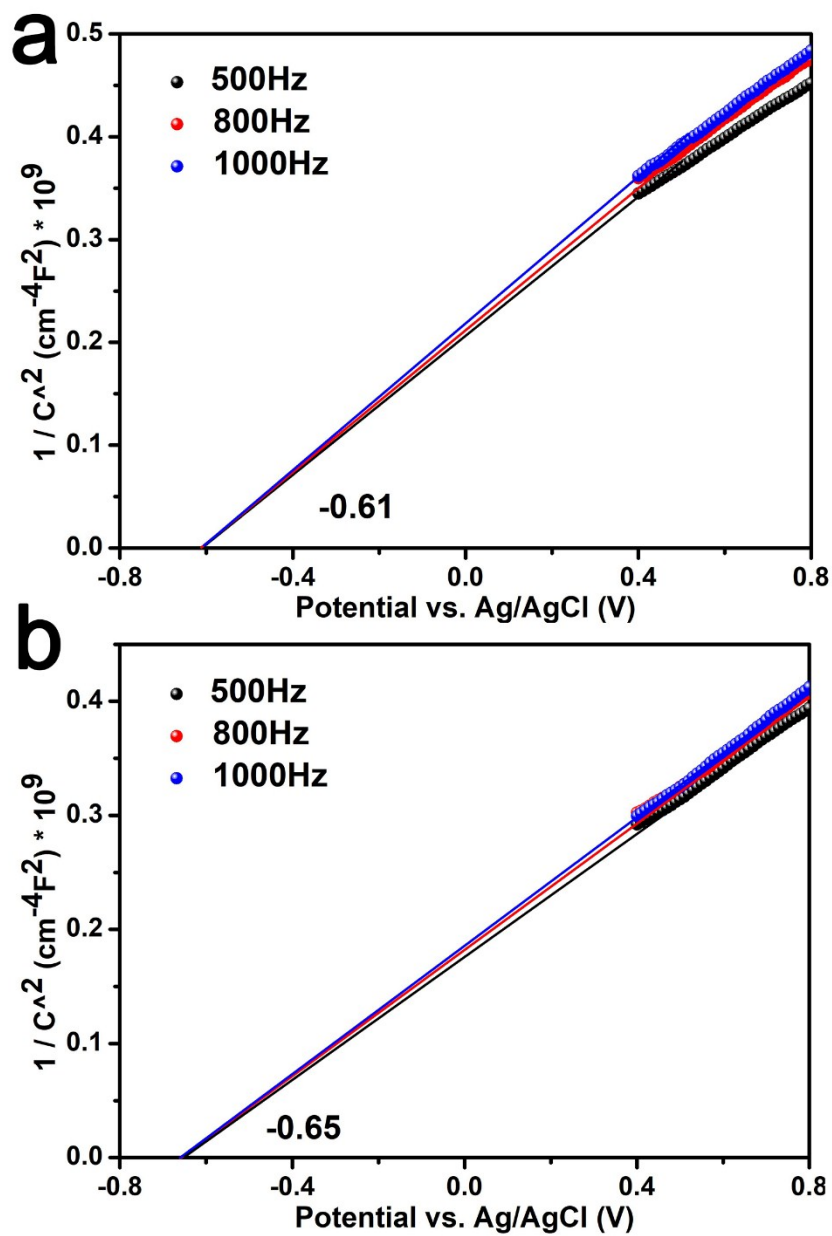


Figure S3. The Mott-Schottky plots of (a) BMO-P, and (b) BMO-U under three different frequencies.

References

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