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Electronic supplementary information

Lattice-Tailored Low-Temperature Processed Electron Transporting Materials Boost the

Open-Circuit Voltage of Planar CsPbBr3 Solar Cells up to 1.654 V

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Device structure	$J_{ m sc}$	V _{oc}	FF	PCE	Def
Device structure	(mA cm ⁻²)	(V)	(%)	(%)	Kel.
FTO/c-TiO ₂ /CsPbBr ₃ /spiro-MeOTAD/Au	5.65	1.536	62.4	5.42	S 1
ITO/NiO/CH ₃ NH ₃ PbBr ₃ /ICBA/PrC60MA/Ag	5.13	1.50	69.5	5.35	S2
ITO/PEDOT:PSS/CH ₃ NH ₃ PbBr ₃ /ICBA/Ca/Al	6.04	1.61	77	7.50	S3
FTO/c-TiO ₂ /m-TiO ₂ /GQD/CsPbBr ₃ /Carbon	8.12	1.458	82.1	9.72	S4
FTO/c-TiO ₂ /m-TiO ₂ /CsPb _{0.97} Sm _{0.03} Br ₃ /Carbon	7.48	1.594	85.1	10.14	S5
FTO/c-TiO ₂ /m- TiO ₂ /CsPb _{0.97} Sm _{0.03} Br ₃ /Cu(Cr,Ba)O ₂ /Carbon	7.81	1.615	85.5	10.79	S6
FTO/c-TiO ₂ /m-TiO ₂ /CsPbBr ₃ /MnS/Carbon	8.28	1.52	83	10.45	S7
FTO/c-TiO ₂ /CsPbBr ₃ /spiro-MeOTAD/Ag	9.78	1.498	74.47	10.91	S 8
FTO/c-TiO ₂ /CsPbBr ₃ /Carbon	7.37	1.545	82.2	9.35	S9
FTO/c-TiO ₂ /m-TiO ₂ /CsPbBr ₃ /CuInS ₂ /ZnS QDs/LPP-Carbon	7.73	1.626	86.3	10.85	S10
FTO/SnO ₂ /TiO _x Cl _{4-2x} /Cs _{0.91} Rb _{0.09} PbBr ₃ /Carbon	7.96	1.629	80.5	10.44	S11

Table S1 Summary of the photovoltaic parameters of single-junction solar cells with high V_{oc} .



Fig. S1 XRD patterns of Sb-doped TiO₂ nanocrystals with different Sb dosages in precursor solution.



Fig. S2 Top-view image of the Sb-TiO₂ film deposited on FTO substrate and the corresponding elemental mapping images.



Fig. S3 (a) XPS spectra of TiO_2 films with and without Sb dopant. High-resolution XPS spectra of (b) Sb 3d and (c) Cl 2p core level.



Fig. S4 (a) Defects in the pristine TiO_2 , (b) Sb^{3+} substitution at the Ti^{3+} sites passivates oxygen vacancy defects.



Fig. S5 UV-vis absorption spectra and corresponding bandgaps of Sb-doped TiO_2 films with increased Sb dosages in precursor solution.

ETL	σ (mS cm ⁻¹)	$\mu (\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1})$	$n_{\rm t} ({\rm cm}^{-3})$
TiO ₂	1.19 × 10 ⁻³	5.44 × 10 ⁻⁴	3.15×10^{17}
Sb-TiO ₂	1.99 × 10 ⁻³	6.58 × 10 ⁻⁴	2.67×10^{17}

Table S2 Electrical parameters of the $\rm TiO_2$ and Sb-TiO_2 films



Fig. S6 Top-view SEM image of CsPbBr3 film deposited on traditional mesoporous TiO2 substrate.



Fig. S7 *J-V* curves of CsPbBr₃ solar cells based on TiO_2 and Sb-doped TiO_2 with different doping amounts.



Fig. S8 J-V curves of CsPbBr₃ solar cells based on Sb-TiO₂ with different Sb-TiO₂ dispersion concentrations.

device	$V_{\rm oc}$ (V)	$J_{\rm sc}$ (mA cm ⁻²)	FF (%)	PCE (%)
TiO ₂	1.586	6.64	78.4	8.26
0.25% Sb	1.584	6.52	83.4	8.62
0.5% Sb	1.654	6.70	80.4	8.91
1% Sb	1.560	6.56	81.5	8.35
2% Sb	1.515	6.23	82.3	7.77

Table S3 Photovoltaic parameters of inorganic CsPbBr₃ PSCs based on TiO_2 and Sb-doped TiO_2 ETLs.

Concentration	$V_{\rm oc}$ (V)	$J_{\rm sc}$ (mA cm ⁻²)	FF (%)	PCE (%)
10 mg mL ⁻¹	1.526	6.18	79.1	7.48
15 mg mL ⁻¹	1.554	6.41	82.4	8.22
20 mg mL ⁻¹	1.654	6.70	80.4	8.91
25 mg mL ⁻¹	1.522	6.42	82.1	8.03

Table S4 Photovoltaic parameters of inorganic CsPbBr₃ PSCs based on Sb-TiO₂ ETLs at different Sb- TiO_2 dispersion concentrations.



Fig. S9 J-V curve of inorganic CsPbBr₃ PSC based on the high-temperature processed TiO₂ ETL.

device	$V_{\rm oc}$ (V)	$J_{\rm sc}$ (mA cm ⁻²)	FF (%)	PCE (%)
TiO ₂ -1	1.606	6.37	76.5	7.83
TiO ₂ -2	1.607	6.49	77.0	8.02
TiO ₂ -3	1.608	5.90	83.5	7.92
0.25% Sb-1	1.619	6.52	80.3	8.47
0.25% Sb-2	1.609	6.22	80.1	8.02
0.25% Sb-3	1.630	6.22	75.8	7.69
0.5% Sb-1	1.652	6.44	78.4	8.34
0.5% Sb-2	1.629	6.47	75.3	7.94
0.5% Sb-3	1.654	6.70	80.4	8.91
0.5% Sb-4	1.616	6.46	79.5	8.29
0.5% Sb-5	1.604	6.25	81.4	8.17
0.5% Sb-6	1.606	6.20	85.4	8.50
0.5% Sb-7	1.608	5.92	80.7	7.69
0.5% Sb-8	1.617	6.55	83.6	8.85
0.5% Sb-9	1.624	6.56	80.0	8.52
0.5% Sb-10	1.603	6.34	79.5	8.07
0.5% Sb-11	1.611	5.64	84.4	7.67
0.5% Sb-12	1.603	5.66	85.0	7.71

Table S5 Photovoltaic parameters of inorganic CsPbBr₃ PSCs based on TiO₂ and Sb doped TiO₂ with V_{oc} over 1.6 V.

0.5% Sb-13	1.604	6.34	82.5	8.39	
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Fig. S10 J-V curves of the optimized inorganic CsPbBr₃ PSCs based on TiO₂ and Sb-TiO₂ under forward and reverse scan directions.

Table S6 Photovoltaic parameters of optimized inorganic CsPbBr₃ PSCs based on TiO₂ and Sb-TiO₂ under forward and reverse scan directions.

device	$V_{\rm oc}$ (V)	$J_{\rm sc}$ (mA cm ⁻²)	FF (%)	PCE (%)
TiO ₂ Reverse	1.586	6.64	78.4	8.26
TiO ₂ Forward	1.553	6.79	51.8	5.46
Sb-TiO ₂ Reverse	1.654	6.70	80.4	8.91
Sb-TiO ₂ Forward	1.607	6.79	70.4	7.68

Table S7 TRPL decay lifetimes of CsPbBr₃ films on TiO₂ and Sb-TiO₂ coated FTO substrates. τ_1 and τ_2 correspond to the slow and fast decay components, respectively.

Samples	τ_1 (ns)	A ₁ (%)	τ_2 (ns)	$A_{2}(\%)$	$\tau_{ave} \left(ns \right)$
TiO ₂ /CsPbBr ₃	0.88	58.5	10.53	41.5	1.42
Sb-TiO ₂ /CsPbBr ₃	0.49	67.6	8.66	32.4	0.71

References

- S1. Q. A. Akkerman, M. Gandini, F. Di Stasio, P. Rastogi, F. Palazon, G. Bertoni, J. M. Ball, M. Prato, A. Petrozza and L. Manna, *Nat. Energy*, 2016, 2, 16194.
- S2. S. Chen, Y. Hou, H. Chen, M. Richter, F. Guo, S. Kahmann, X. Tang, T. Stubhan, H. Zhang, N. Li, N. Gasparini, C. O. R. Quiroz, L. S. Khanzada, G. J. Matt, A. Osvet and C. J. Brabec, *Adv. Energy Mater.*, 2016, 6, 1600132.
- S3. C. G. Wu, C. H. Chiang and S. H. Chang, *Nanoscale*, 2016, **8**, 4077.
- S4. J. Duan, Y. Zhao, B. He and Q. Tang, Angew. Chem. Int. Ed., 2018, 57, 3787.
- S5. J. Duan, Y. Zhao, X. Yang, Y. Wang, B. He and Q. Tang, Adv. Energy Mater., 2018, 8, 1802346.
- S6. J. Duan, Y. Zhao, Y. Wang, X. Yang and Q. Tang, Angew. Chem. Int. Ed., 2019, 58, 16147.
- S7. X. Li, Y. Tan, H. Lai, S. Li, Y. Chen, S. Li, P. Xu and J. Yang, ACS Appl. Mater. Interfaces, 2019, 11, 29746.
- S8. G. Tong, T. Chen, H. Li, L. Qiu, Z. Liu, Y. Dang, W. Song, L. K. Ono, Y. Jiang and Y. Qi, *Nano Energy*, 2019, 65, 104015.
- S9. T. Xiang, Y. Zhang, H. Wu, J. Li, L. Yang, K. Wang, J. Xia, Z. Deng, J. Xiao, W. Li, Z. Ku, F. Huang, J. Zhong, Y. Peng and Y.-B. Cheng, *Sol. Energy Mater. Sol. Cells*, 2019, 206, 110317.
- S10. J. Duan, Y. Wang, X. Yang and Q. Tang, Angew. Chem. Int. Ed., 2020, 59, 4391.
- S11. Q. Zhou, J. Duan, Y. Wang, X. Yang and Q. Tang, J. Energy Chemistry, 2020, 50, 1.