

Supplementary Materials

rGO Wrapped Trimetallic Sulfide Nanowires as Efficient Bifunctional Catalyst for Electrocatalytic Oxygen Evolution and Photocatalytic Organic Degradation

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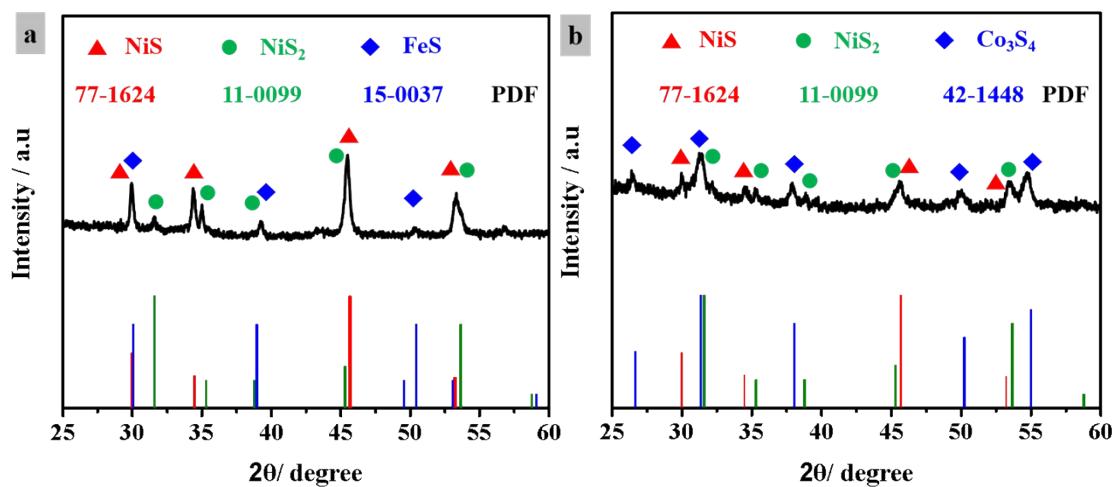


Fig. S1. XRD pattern of the Ni-Co-S-160/rGO and Ni-Fe-S-160/rGO.

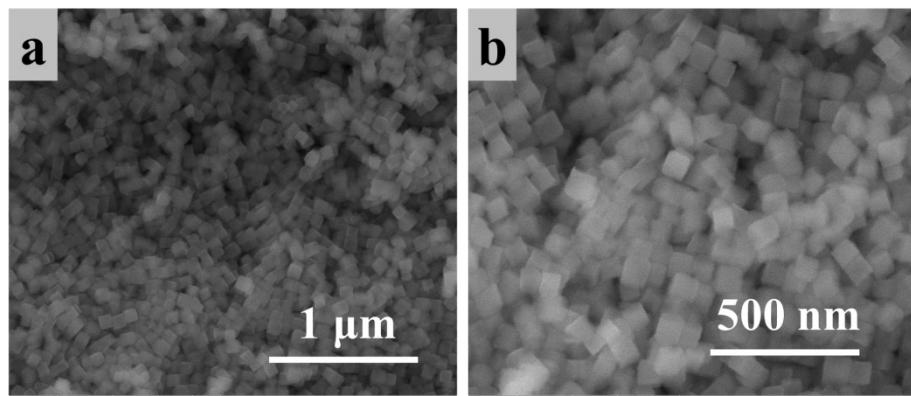


Fig. S2. The SEM images of precursor Ni-Co-Fe-PBA/GO.

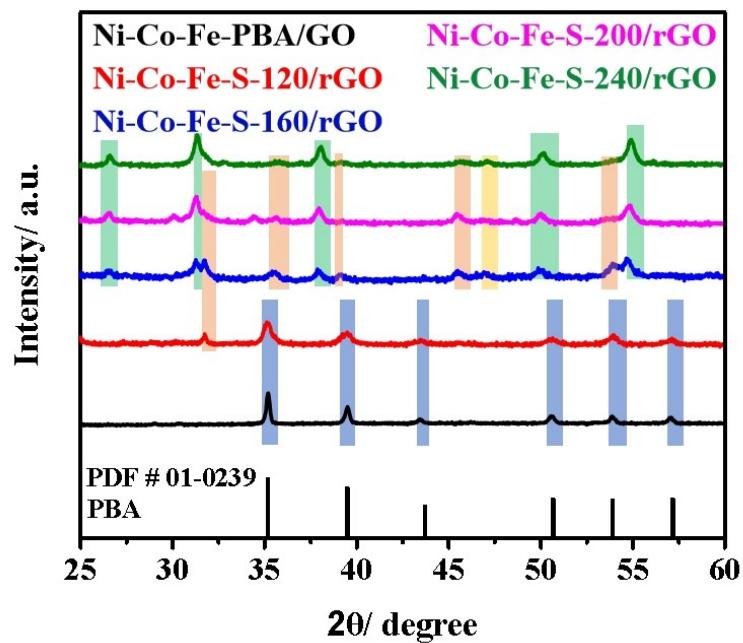


Fig. S3. The XRD patterns of Ni-Co-Fe-PBA/GO, Ni-Co-Fe-S-120/rGO, Ni-Co-Fe-S-160/rGO, Ni-Co-Fe-S-200/rGO, Ni-Co-Fe-S-240/rGO.

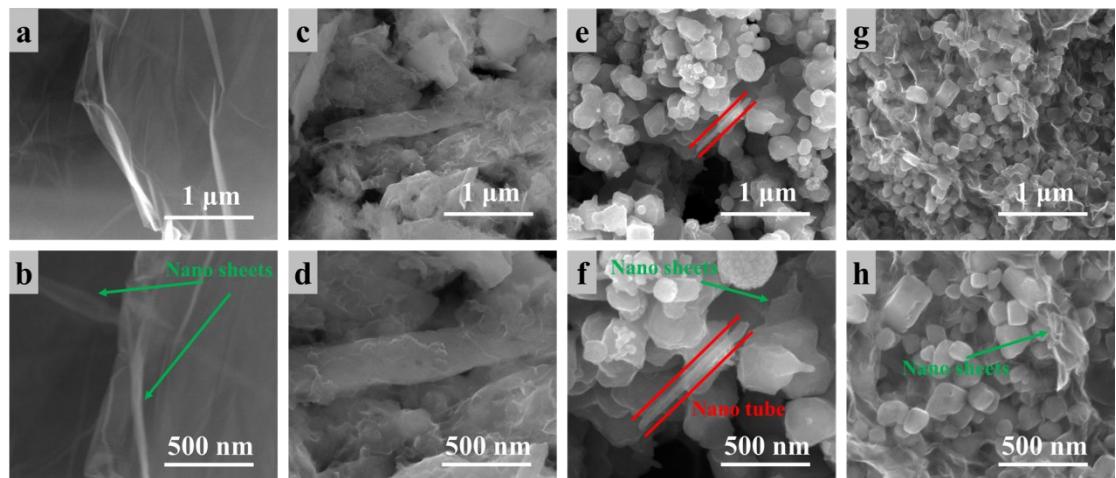


Fig. S4. The SEM images of a, b) GO, c, d) Ni-Co-Fe-S-160, e, f) Ni-Co-S-160/rGO and g, h) Ni-Fe-S-160/rGO

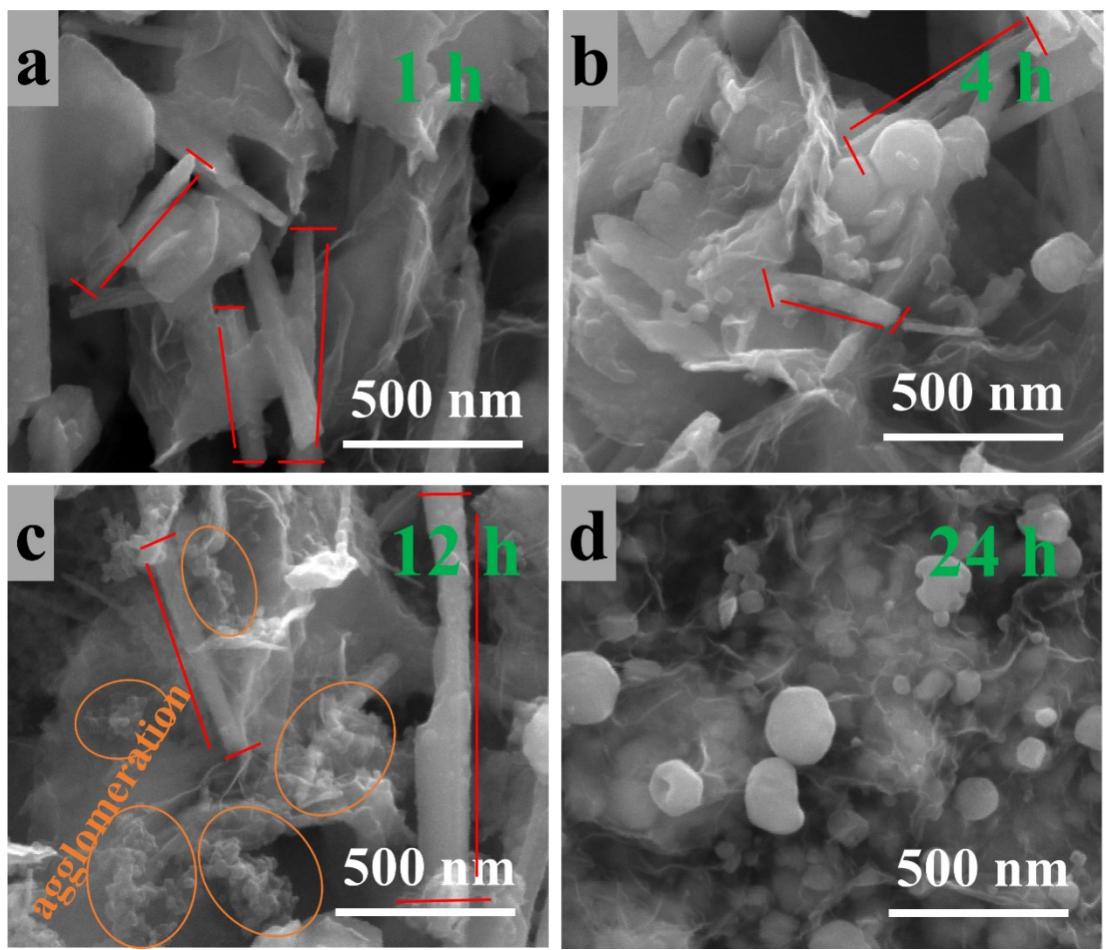


Fig. S5. The SEM images of Ni-Co-Fe-S-160/rGO with different reaction times a) 1 h, b) 4 h, c) 12 h and d) 24 h.

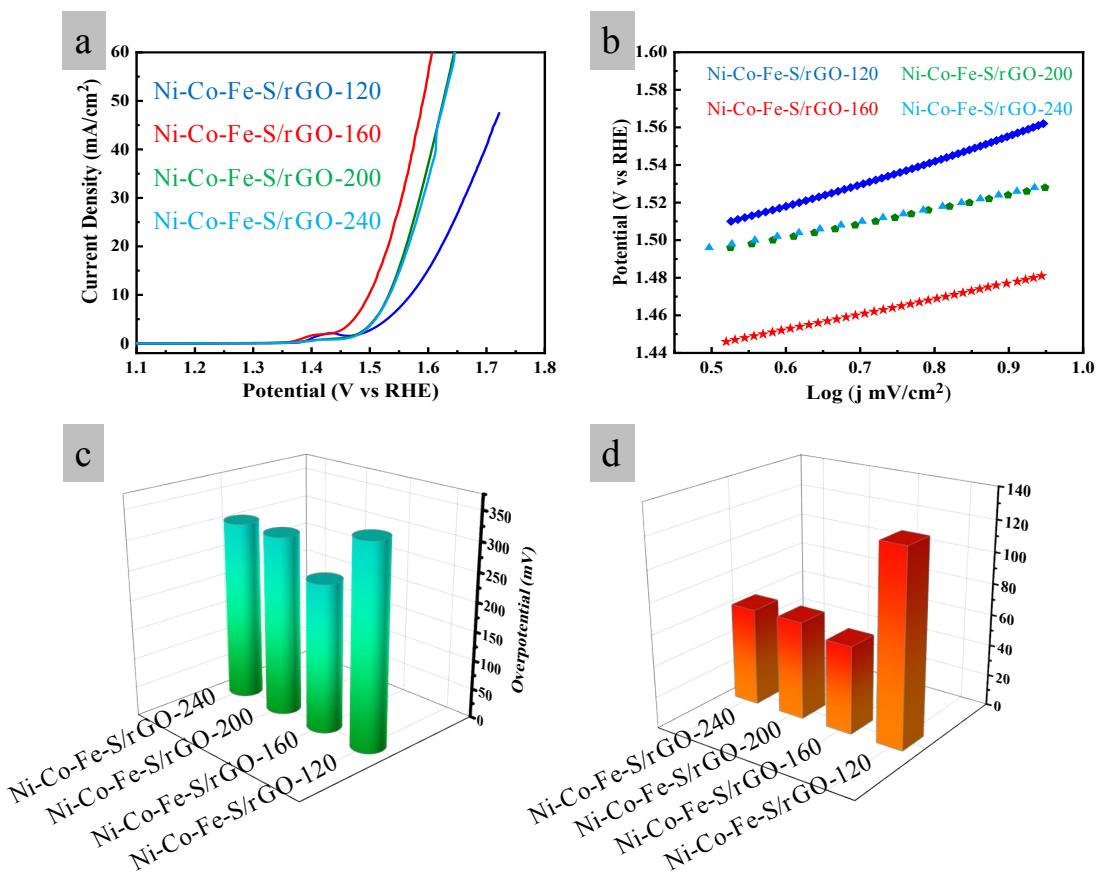


Fig. S6. a) Polarization curves, b) Tafel plots and c, d) performance comparison of the Ni-Co-Fe-S/rGO-**120**, Ni-Co-Fe-S/rGO-**160**, Ni-Co-Fe-S/rGO-**200** and Ni-Co-Fe-S/rGO-**240**.

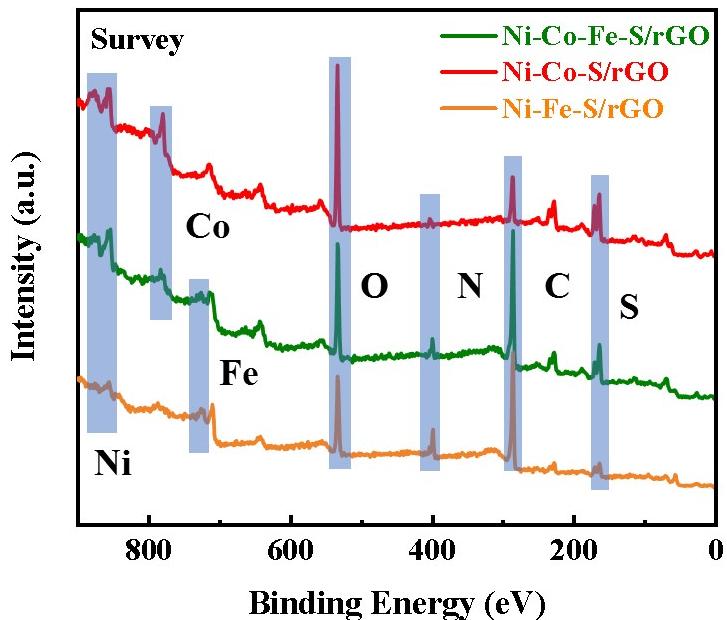


Fig. S7. XPS survey curve of the Ni-Co-Fe-S/rGO, Ni-Co-S/rGO and Ni-Fe-S/rGO.

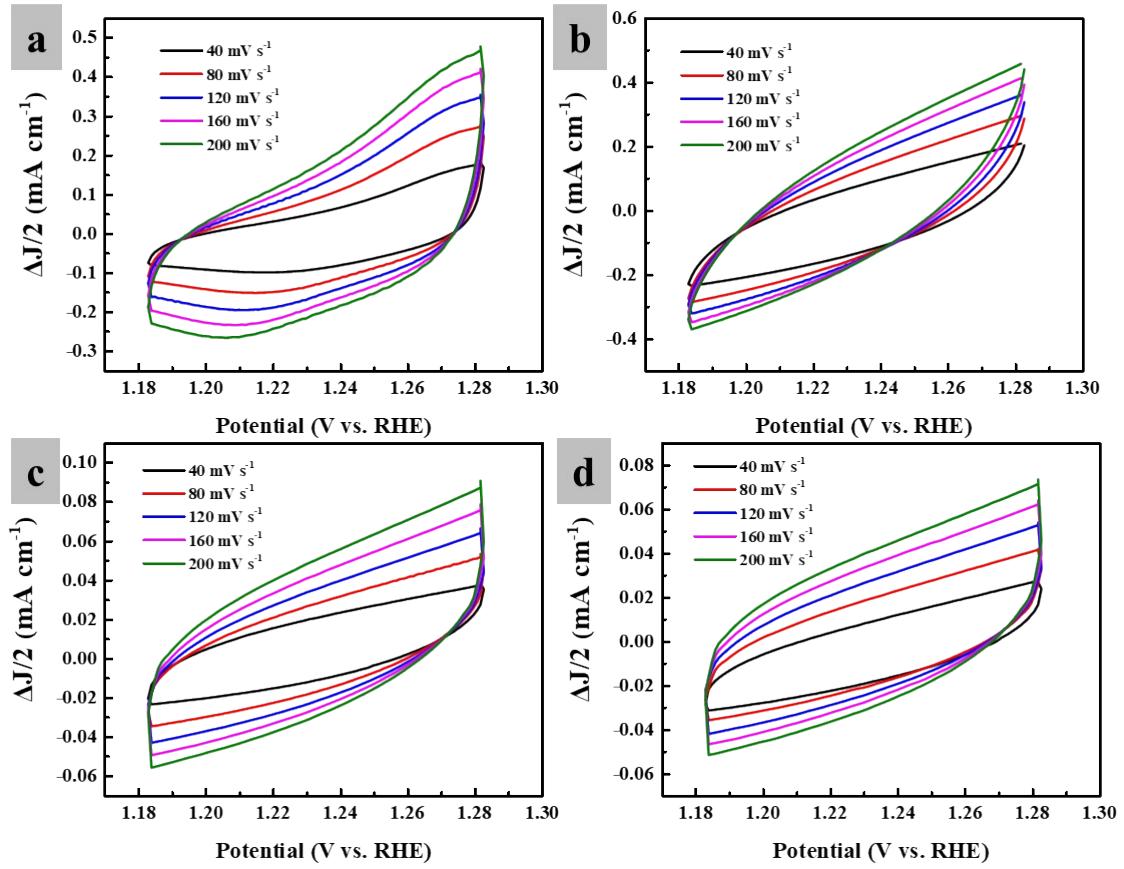


Fig. S8. Voltammograms of the a) Ni-CoFe-S/rGO, b) Ni-CoFe-S, c) Ni-Co-S/rGO and Ni-Fe-S/rGO for OER.

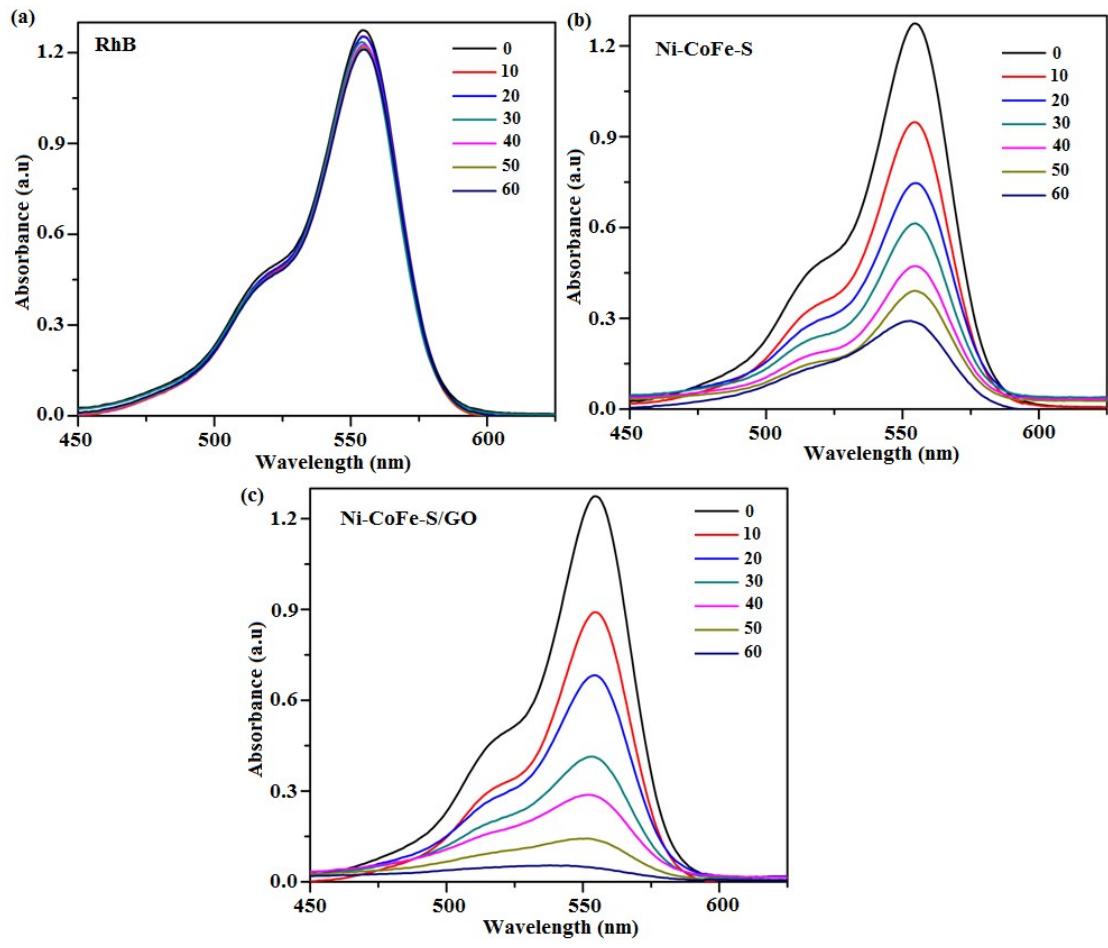


Fig. S9. UV-visible spectra of a) RhB, Ni-Co-Fe-S and Ni-Co-Fe-S/rGO vs. illumination time.

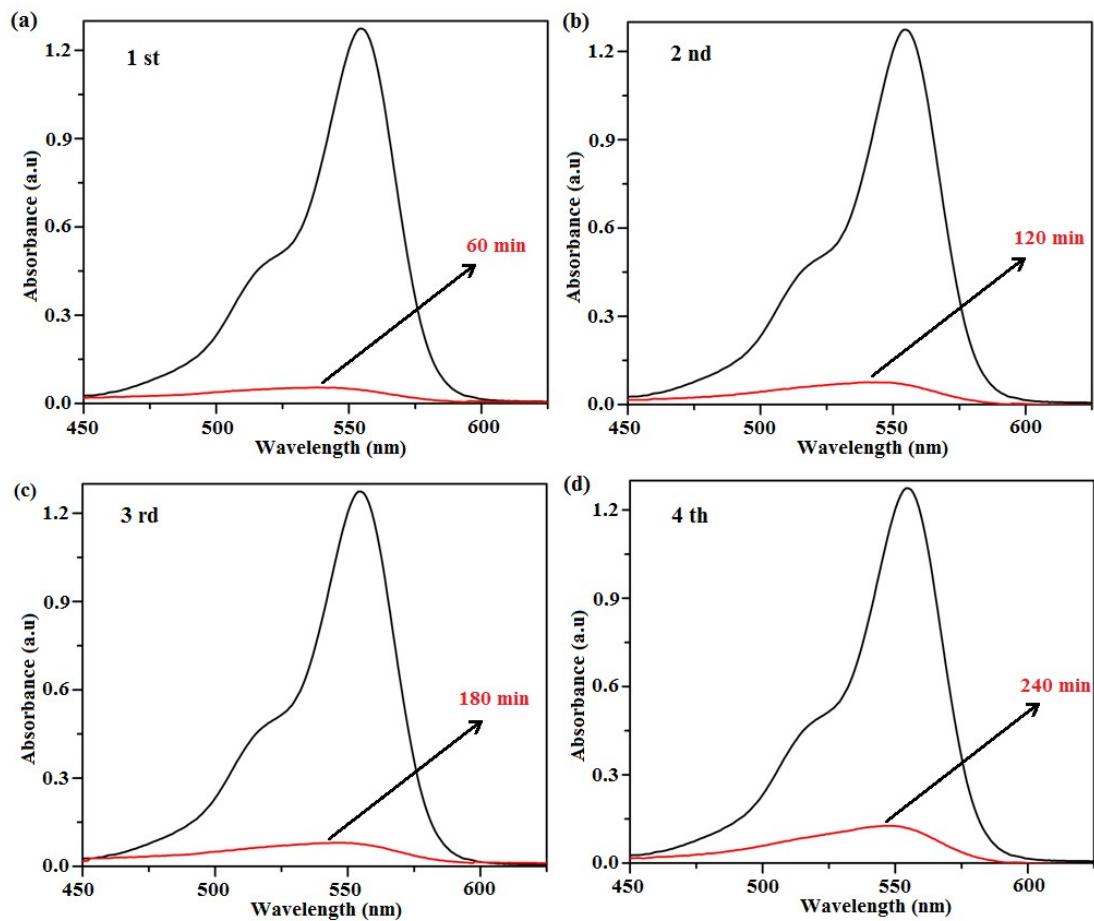


Fig. S10. UV-visible spectra of Ni-Co-Fe-S/GO with different illumination time.

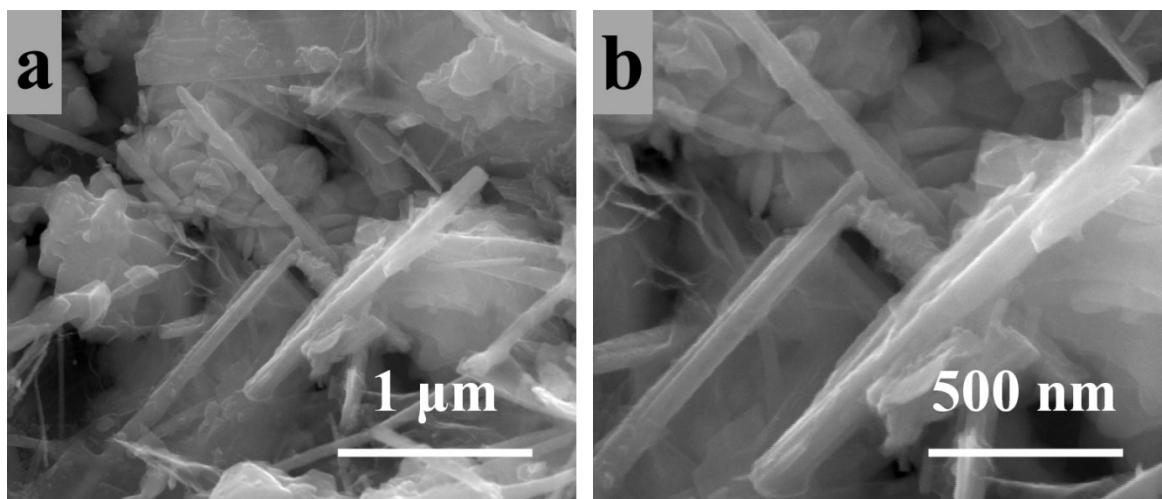


Fig. S11. SEM images after photo-catalytic degradation process.

Tab. S1. The atomic percent of S, Ni, Co and Fe from EDS and ICP-OES analysis.

Method	Characterization				Atomic Ratio (%)			
	S	Ni	Co	Fe				
EDS	58.2	25.2	6.4	10.2				
ICP-OES	60.3	23.6	6.2	9.9				

Tab. S2. The XPS peak position of Ni-Co-Fe-S/rGO, Ni-Co-S/rGO and Ni-Fe-S/GO.

Catalysts	Peak position (eV)					
	Ni-S	Ni²⁺/Ni³⁺	Co-S	Fe²⁺/Fe³⁺	S 2p_{1/2}	S 2p_{3/2}
Ni-Co-Fe-S/rGO	852.62	855.93	778.06	710.88	161.85	160.70
Ni-Co-S/rGO	853.09	856.76	778.32	/	161.64	160.52
Ni-Fe-S/rGO	852.46	855.93	/	709.70	162.01	160.89