

Electronic Supplementary Information for

**Computational Screening of Homo and Hetero
Transition Metal Dimer Catalysts for Reduction
of CO₂ to C₂ Products with High Activity and
Low Limiting Potential**

**Dachang Chen^{a,b}, Zhiwen Chen^b, Zhuole Lu^b, , Ju Tang^a, Xiaoxing Zhang^{a,d*}, Chandra
Veer Singh^{b,c*}**

^a School of Electrical Engineering and Automation, Wuhan University, Wuhan 430072, China

^b Department of Materials Science and Engineering, University of Toronto, Toronto, Ontario,
M5S 3E4, Canada

^c Department of Mechanical and Industrial Engineering, University of Toronto, Toronto,
Ontario, M5S 3G8, Canada

^d Hubei Key Laboratory for High-efficiency Utilization of Solar Energy and Operation
Control of Energy Storage System, Hubei University of Technology, Wuhan 430068, China

*Corresponding author, E-mail: xiaoxing.zhang@outlook.com;
chandraveer.singh@utoronto.ca

Table of Content

Figures	S4
Figure S1. The adsorption free energy of *H, *OH and *CO against various C ₂ related intermediates.	S4
Figure S2. The pathway for CRR to C ₂ products on Cr-Fe based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.....	S5
Figure S3. The pathway for CRR to C ₂ products on Cr-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S6
Figure S4. The pathway for CRR to C ₂ products on Mn-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S7
Figure S5. The pathway for CRR to C ₂ products on Fe-Fe based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S8
Figure S6. The pathway for CRR to C ₂ products on Fe-Co based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S9
Figure S7. The pathway for CRR to C ₂ products on Co-Co based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S10
Figure S8. The pathway for CRR to C ₂ products on Co-Ni based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S11
Figure S9. The pathway for CRR to C ₂ products on Co-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S12
Figure S10. The pathway for CRR to C ₂ products on Ni-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE.	S13
Figure S11. The reaction diagrams of CRR on other DACs. The value larger than +0.8 eV means that the corresponding reaction is difficult to happen.....	S14
Figure S12. Comparison of energy barriers between the coupling process and protonation process on DACs, and the relationship with the descriptors.....	S15
Figure S13. Configurations of protonation process of *CO-CHO or *CO-CO on DACs (LH mechanism).....	S16
Figure S14. Protonation process of *CO-CHO or *CO-CO on DACs (ER mechanism) at applied potential.....	S17
Tables	S18
Table S1. Average binding energies (eV) of metal atoms for TM ₂ N ₆ -graphene system and average cohesive energy in bulk metals for doped two metal atoms.....	S18
Table S2. The values of thermal correction containing the zero-point energy, the enthalpic correction, and the entropy correction of the relevant molecules, and the adsorbed species. The temperature is set as 298 K.....	S19
Table S3. Values of G(*H), G(*OH) on various of DACs (the yellow background means that the DAC is excluded for CRR) and magnetic moments.....	S20
Table S4. Values of G(*COOH), G(*CO) on various of DACs and magnetic moments.....	S21
Table S5. Values of G(*CHO), G(*COH) on various of DACs and magnetic moments.....	S22
Table S6. Values of G(*CO-CO), G(*CO-CHO) and G(*CO-COH) on various of DACs and magnetic moments.....	S23

Table S7. Values of G(*COCO), G(*COCHO) and G(*COCOH) on various of DACs and magnetic moments.....	S24
Table S8. Values of G(*COCH ₂ O), G(*COHCHO), G(*COHCH ₂ O), G(*CHOHCH ₂ O), G(*CHCH ₂ O), G(*CH ₂ CH ₂ O), G(*CH ₂ CH ₂ OH), G(*CH ₂ CH ₂ OH) and G(*O) on various of DACs and magnetic moments	S25
Table S9. Values of adsorbed species on Co-Co and magnetic moments	S26

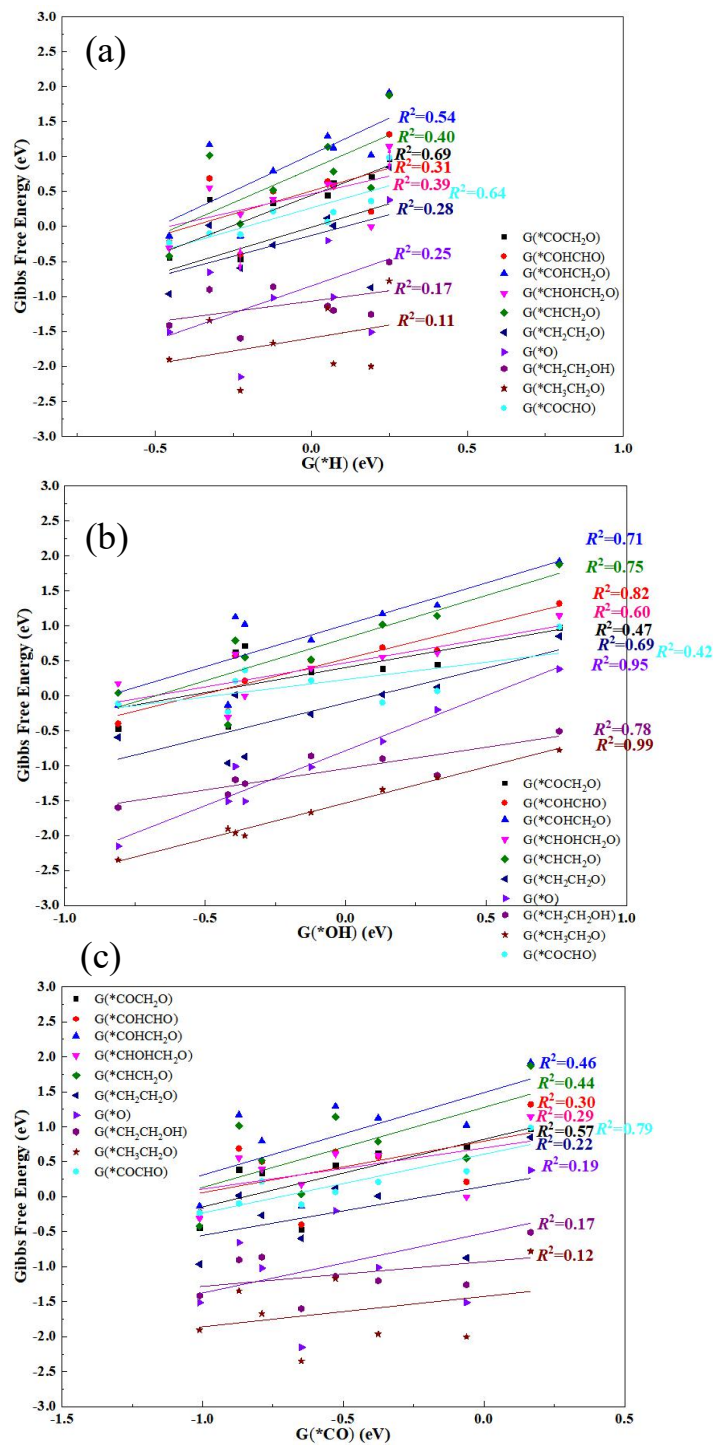


Figure S1 The adsorption free energy of *H, *OH and *CO against various C₂ related intermediates; (a) $G(*H)$ versus every C₂ related intermediate; (b) $G(*OH)$ versus every C₂ related intermediate. The linearity is marked with different color for different intermediates; (c) $G(*CO)$ versus every C₂ related intermediate. The linearity is marked with different color for different intermediates

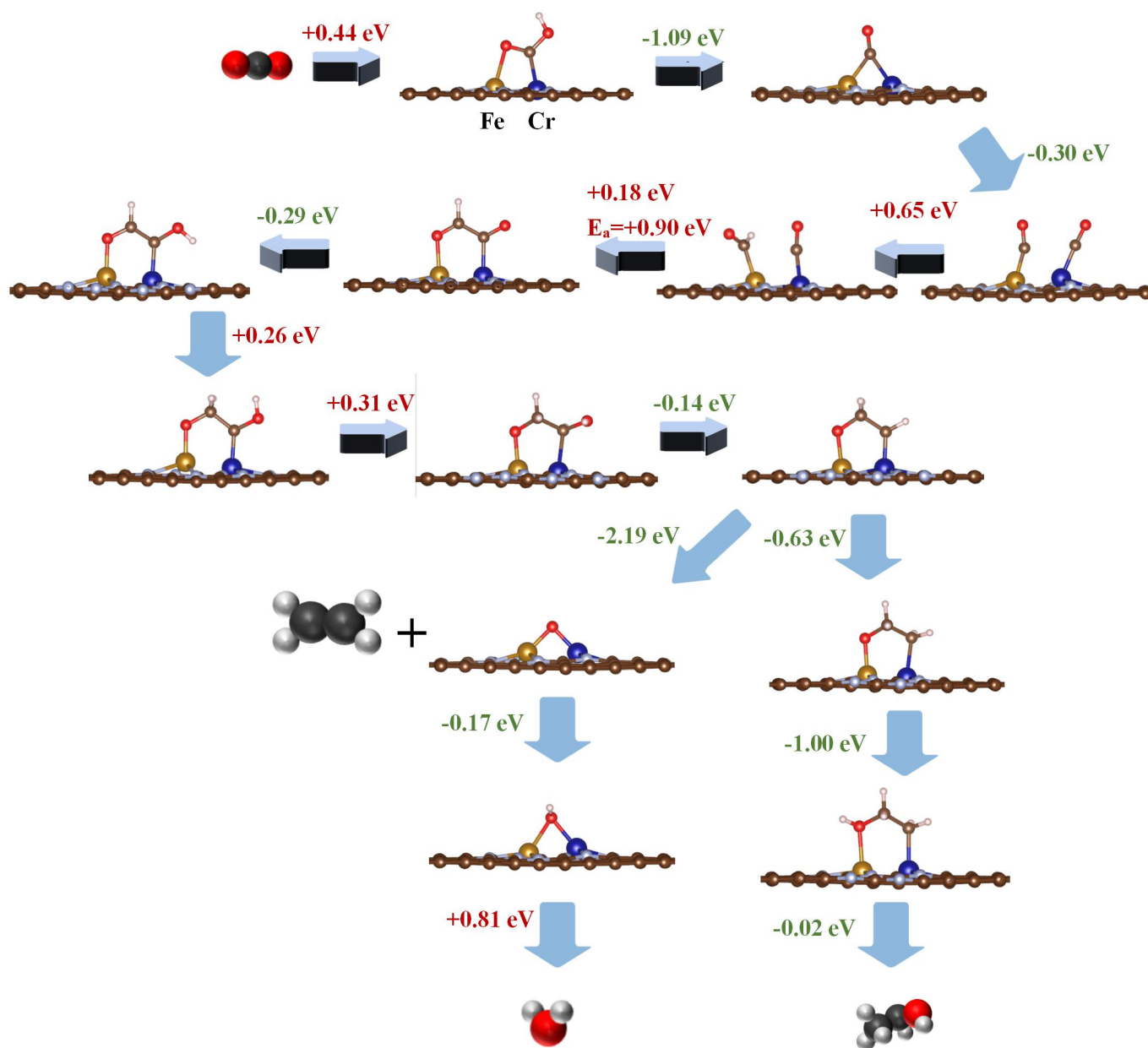


Figure S2 The pathway for CRR to C₂ products on Cr-Fe based DAC with the minimum limiting potential. The calculated free energies are all at $U = 0 \text{ V}$ versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

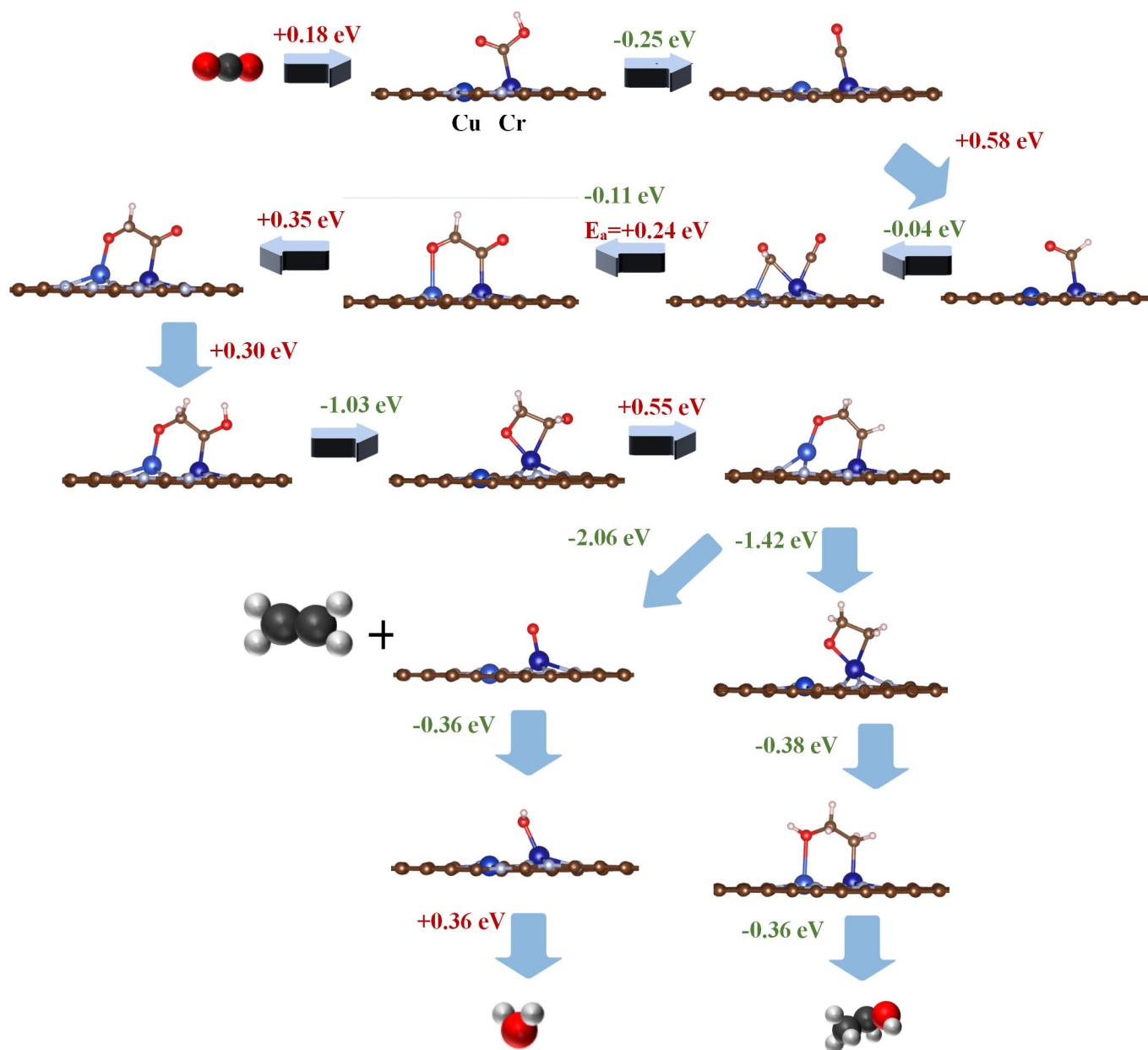


Figure S3 The pathway for CRR to C₂ products on Cr-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

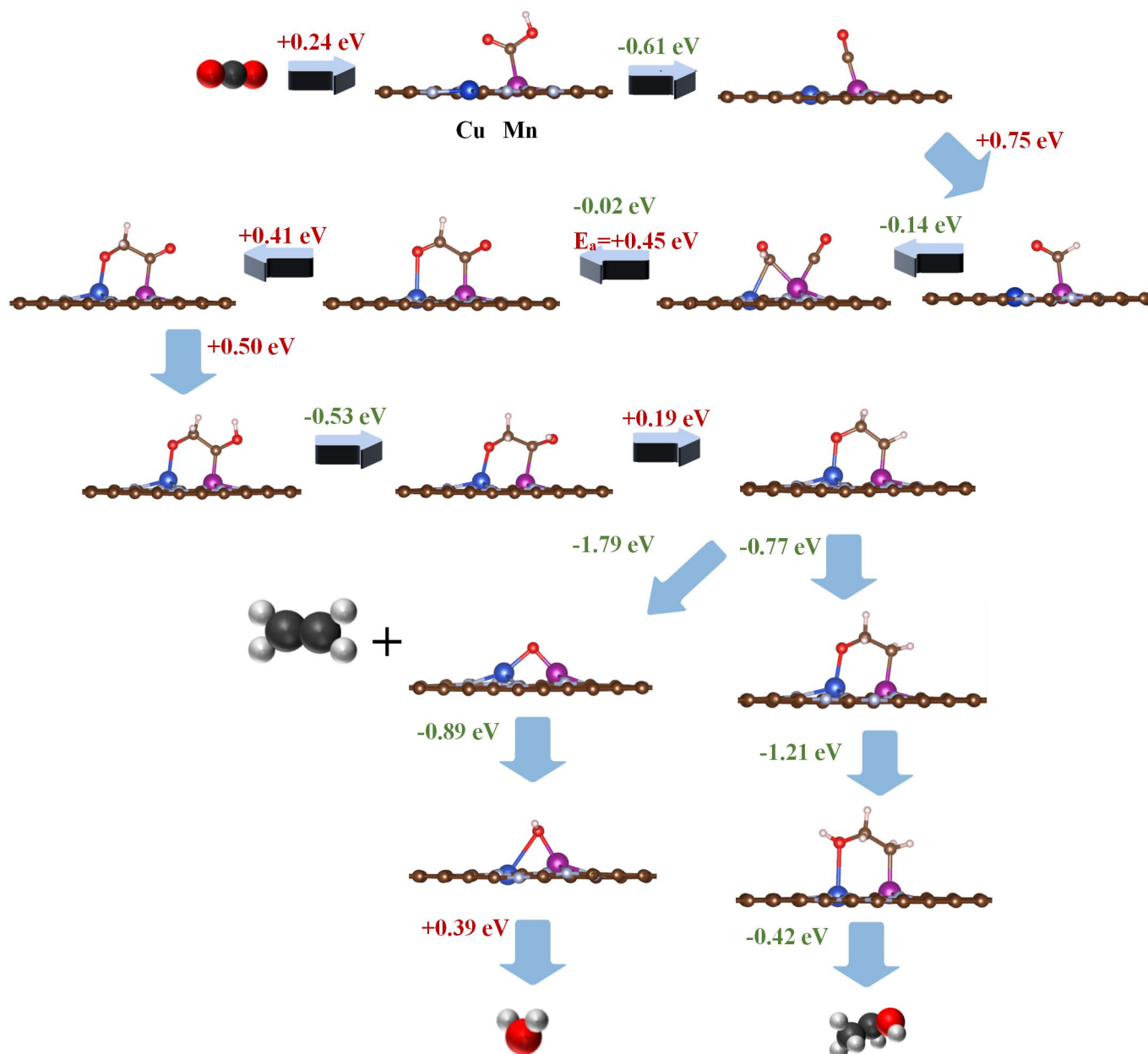


Figure S4 The pathway for CRR to C₂ products on Mn-Cu based DAC with the minimum limiting potential. The calculated free energies are all at $U = 0 \text{ V}$ versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

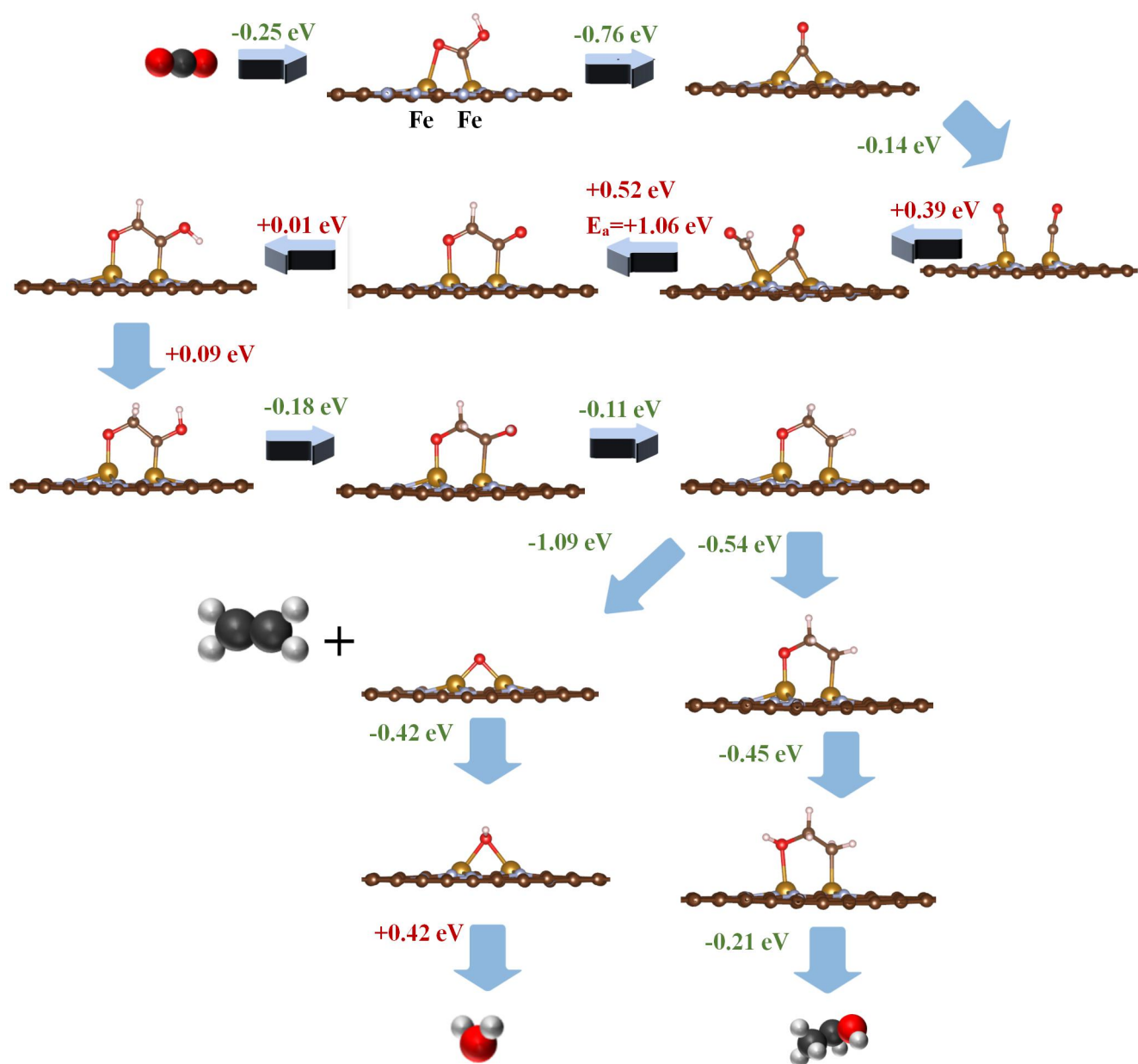


Figure S5 The pathway for CRR to C₂ products on Fe-Fe based DAC with the minimum limiting potential. The calculated free energies are all at U = 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

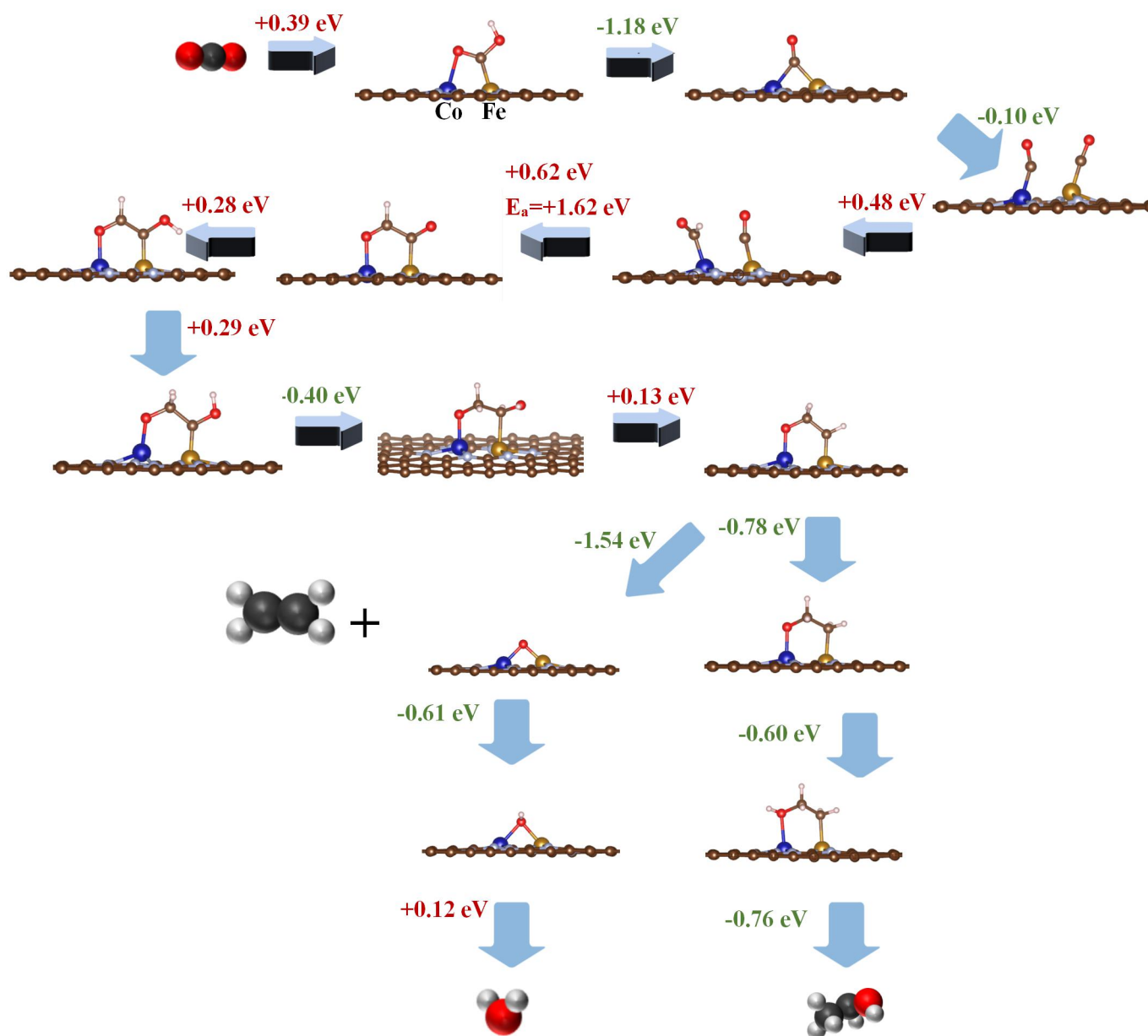


Figure S6 The pathway for CRR to C₂ products on Fe-Co based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

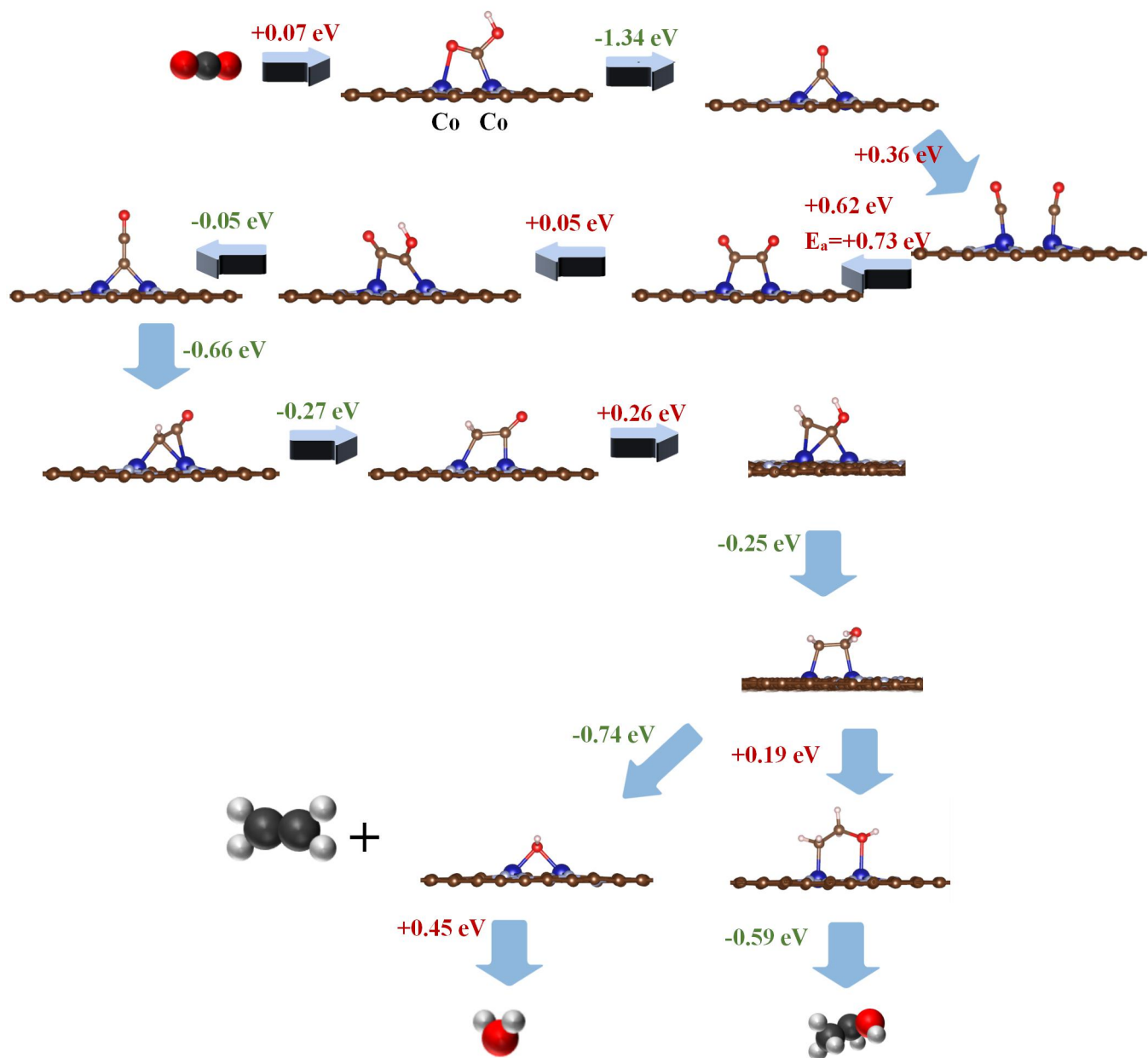


Figure S7 The pathway for CRR to C₂ products on Co-Co based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

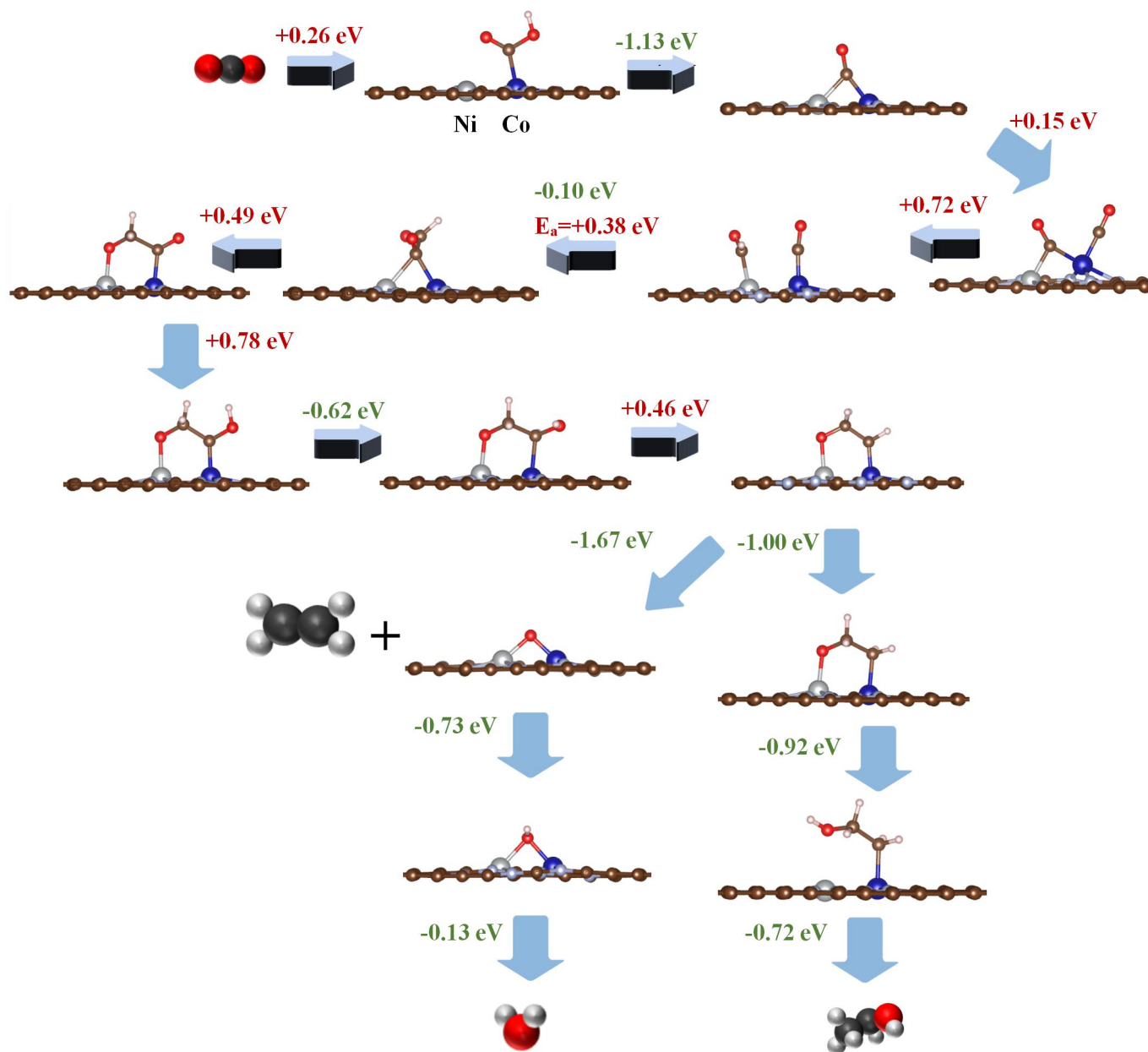


Figure S8 The pathway for CRR to C₂ products on Co-Ni based DAC with the minimum limiting potential. The calculated free energies are all at U = 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

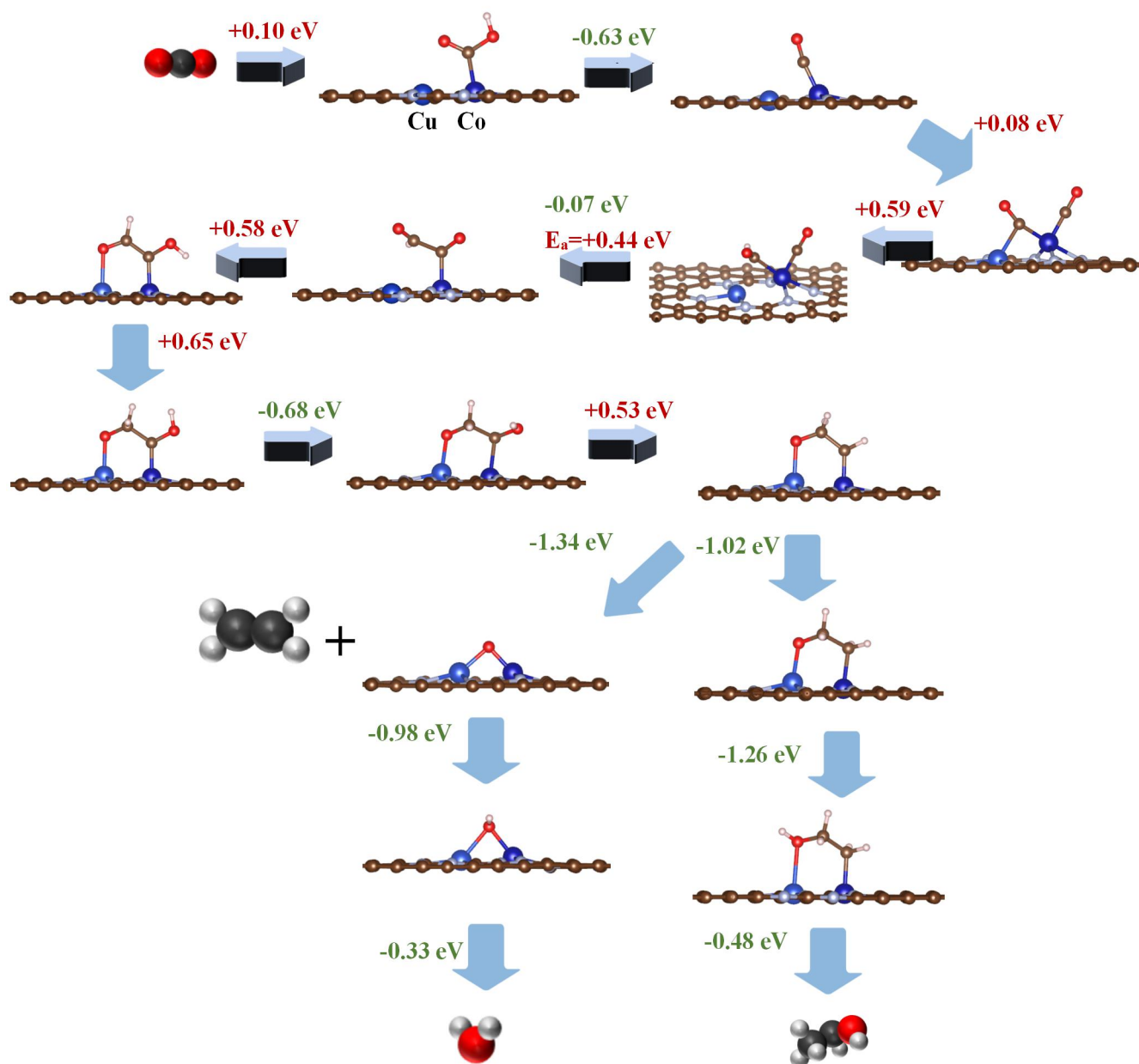


Figure S9 The pathway for CRR to C_2 products on Co-Cu based DAC with the minimum limiting potential. The calculated free energies are all at $U = 0$ V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

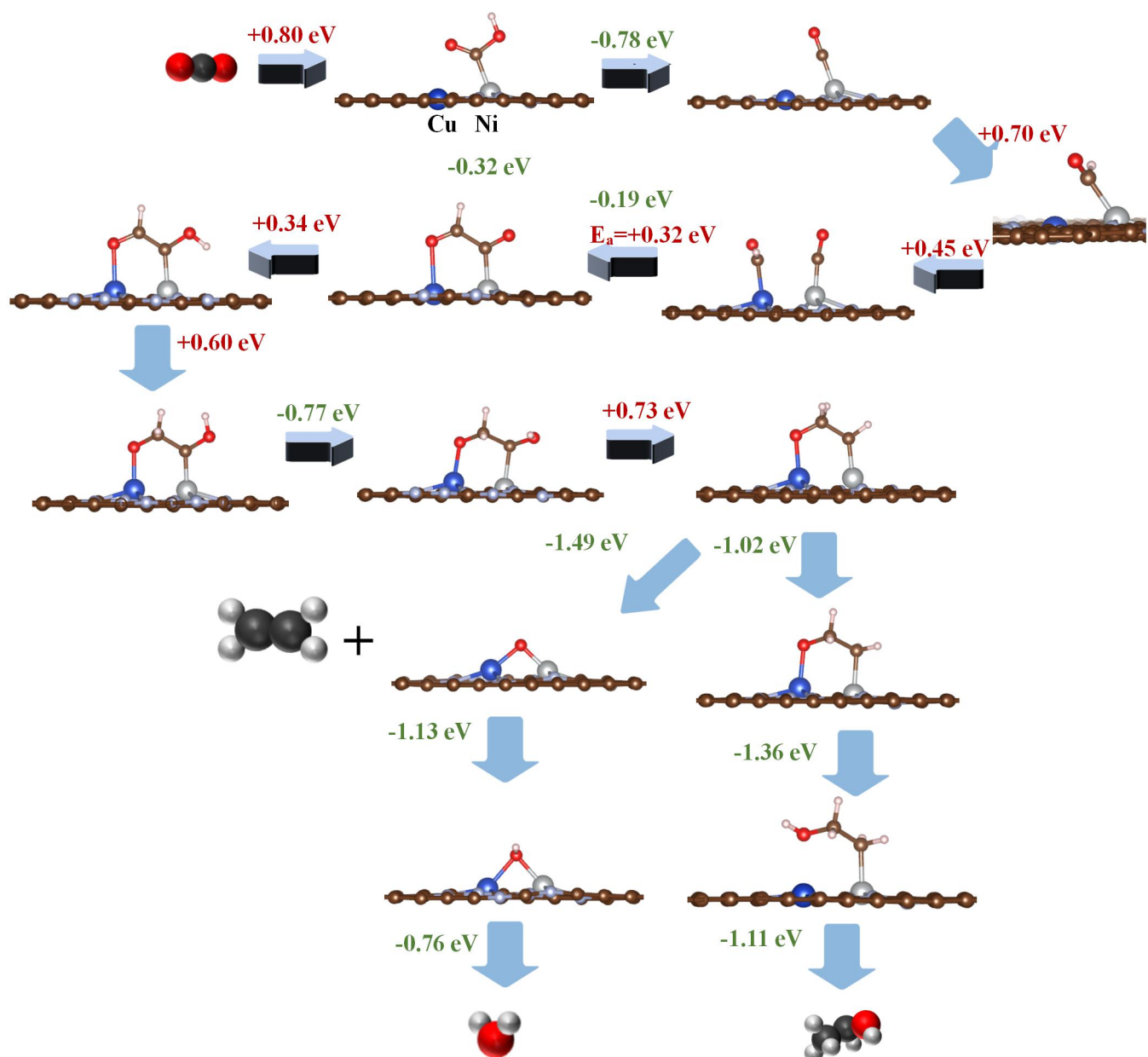


Figure S10 The pathway for CRR to C₂ products on Ni-Cu based DAC with the minimum limiting potential. The calculated free energies are all at U= 0 V versus RHE. (The red atoms are O, the brown atoms are C, the white atoms are H and the silver atoms in DAC are N)

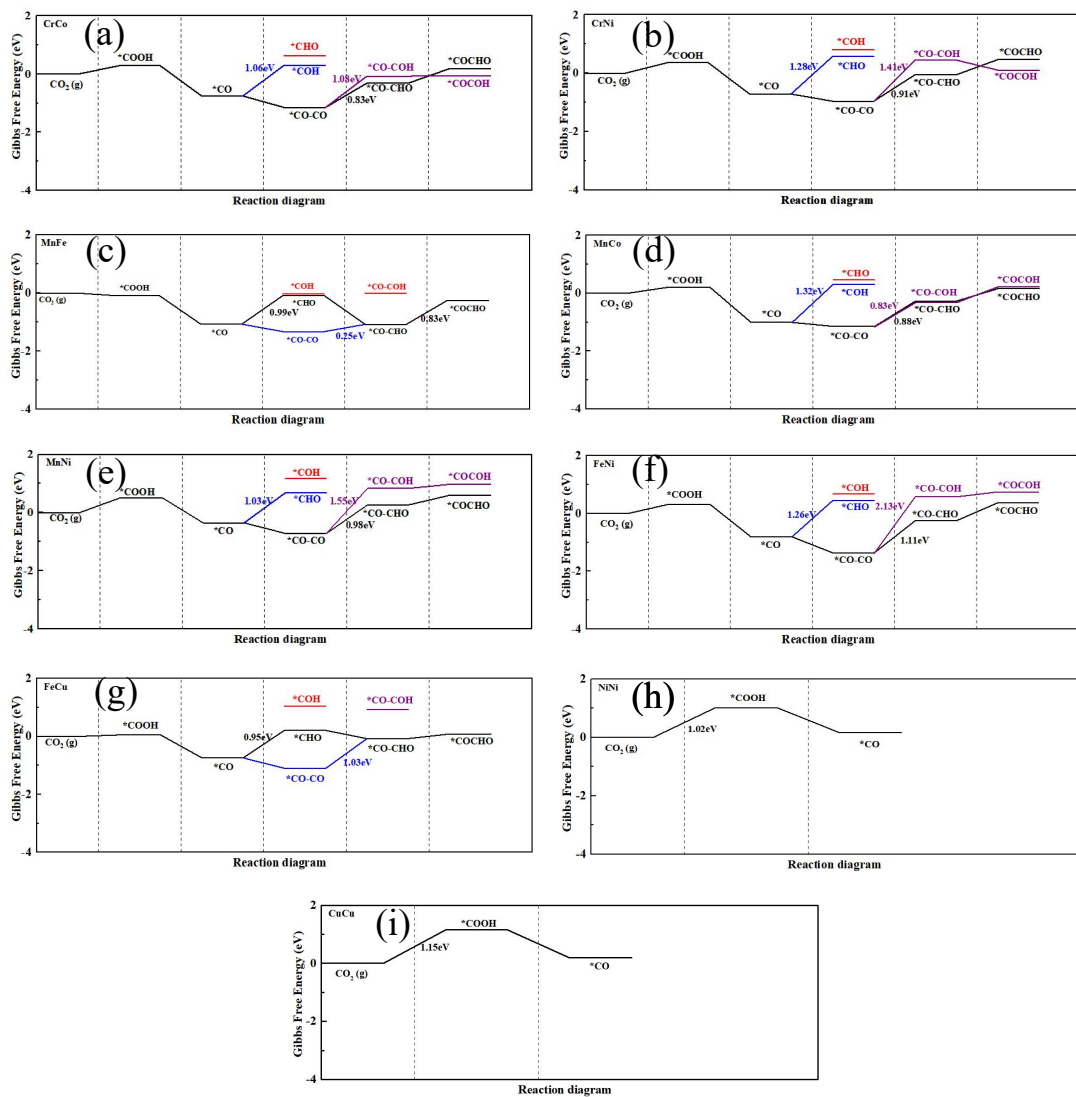


Figure S11 The reaction diagrams of CRR on other DACs. The value larger than +0.8 eV means that the corresponding reaction is difficult to happen.

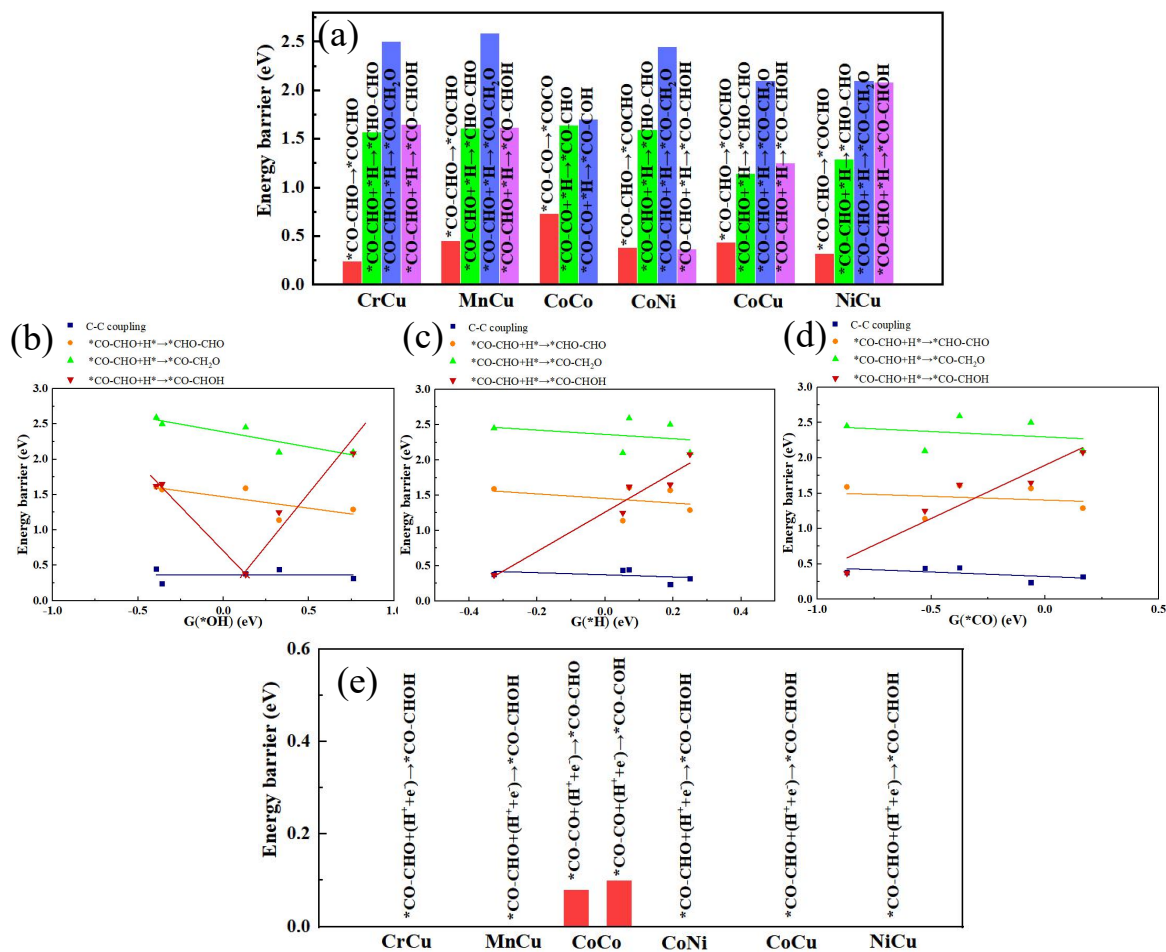


Figure S12 Comparison of energy barriers between the coupling process and protonation process on DACs, and the relationship with the descriptors: (a) comparison between the energy barriers of different reactions (LH mechanism); (b) relationship between the energy barriers and $G(*OH)$; (c) relationship between the energy barriers and $G(*H)$; (d) relationship between the energy barriers and $G(*CO)$; (e) energy barriers of protonation process via ER mechanism at applied potential.

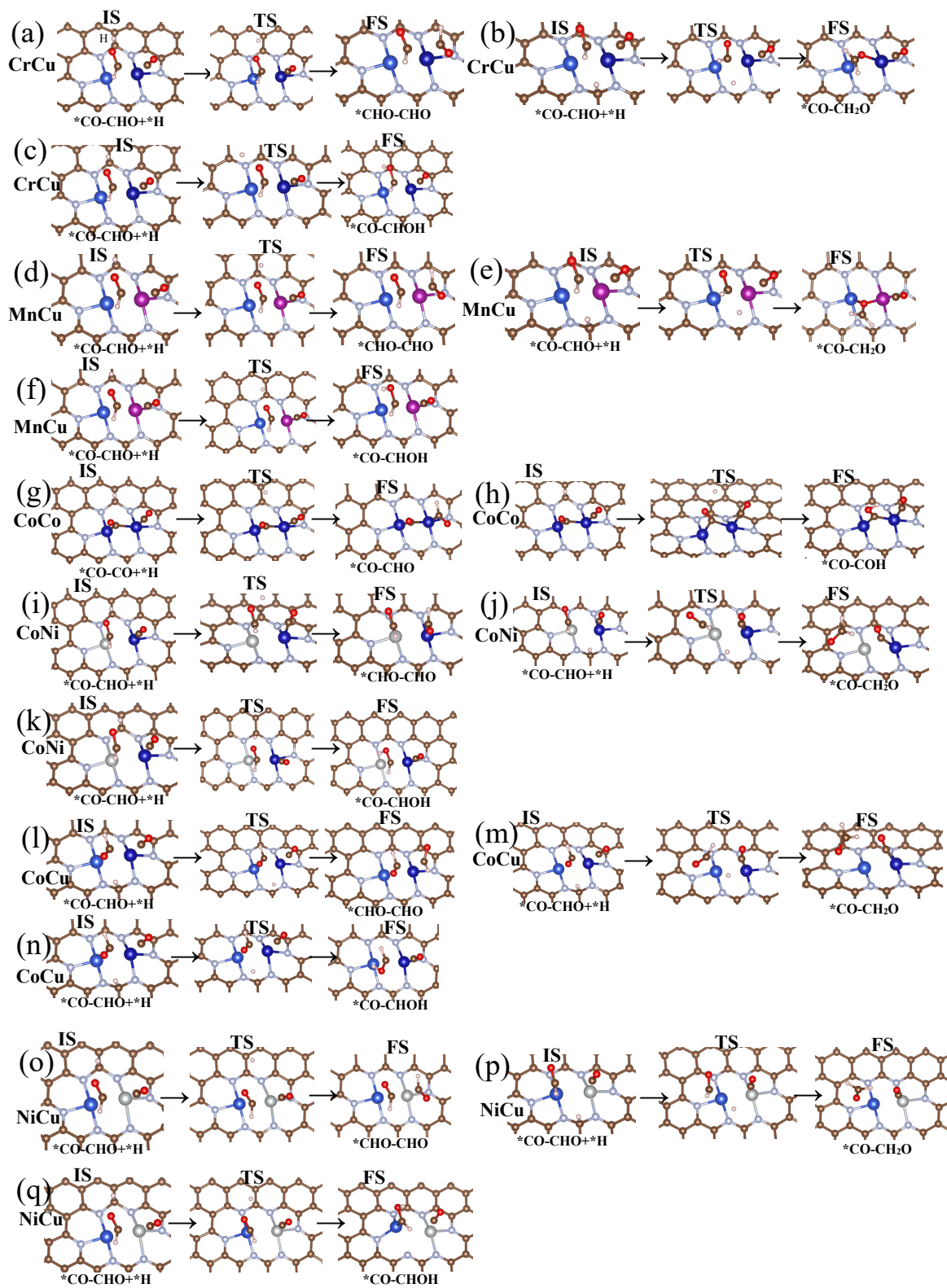


Figure S13 Configurations of protonation process of $*CO-CHO$ or $*CO-CO$ on DACs (LH mechanism).

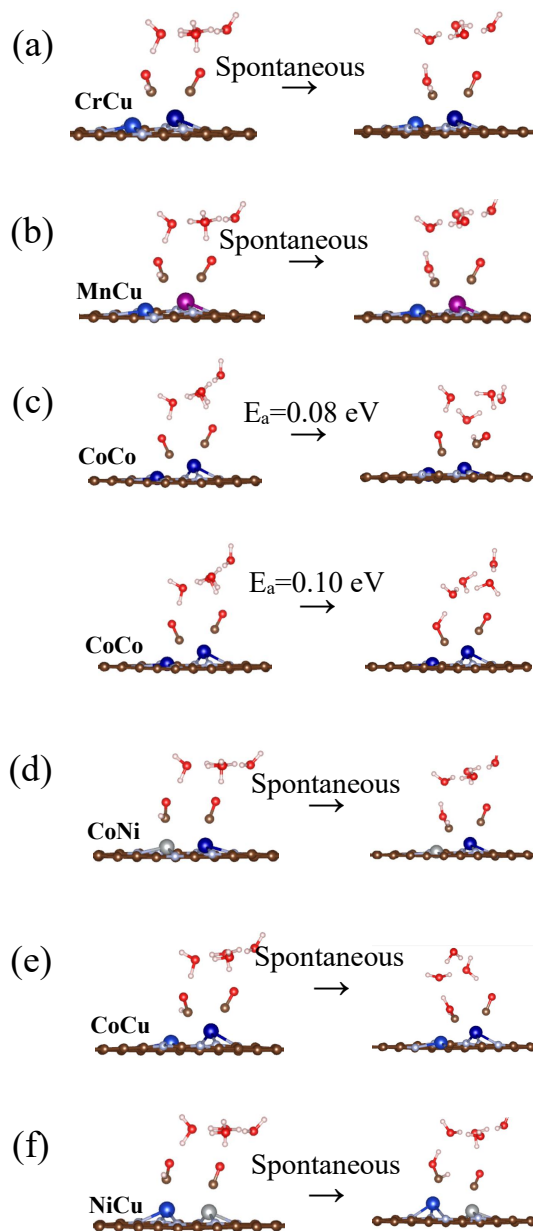


Figure S14 Protonation process of *CO-CHO or *CO-CO on DACs (ER mechanism) at applied potential: (a) *CO-CHO + (H⁺ + e⁻) → *CO-CHOH on Cr-Cu; (b) *CO-CHO + (H⁺ + e⁻) → *CO-CHOH on Mn-Cu; (c) *CO-CO + (H⁺ + e⁻) → *CO-CHO on Co-Co; *CO-CO + (H⁺ + e⁻) → *CO-COH on Co-Co; (d) *CO-CHO + (H⁺ + e⁻) → *CO-CHOH on Co-Ni; (e) *CO-CHO + (H⁺ + e⁻) → *CO-CHOH on Co-Cu; (f) *CO-CHO + (H⁺ + e⁻) → *CO-CHOH on Ni-Cu;

Table S1. Average binding energies (eV) of metal atoms for TM₂N₆-graphene system and average cohesive energy in bulk metals for doped two metal atoms

TM atoms	Average binding energy (eV)	Average cohesive energy (bulk metal) (eV)	Difference value between binding energy and cohesive energy (bulk metal) (eV)	Magnetic moment (μ_B)
CrCr	-4.66	-4.10	-0.56	7.99
CrMn	-4.59	-3.51	-1.08	8.15
CrFe	-5.14	-4.19	-0.95	6.23
CrCo	-5.67	-4.25	-1.42	1.50
CrNi	-5.45	-4.27	-1.18	2.70
CrCu	-4.72	-3.80	-0.92	3.53
MnMn	-4.58	-2.92	-1.66	7.06
MnFe	-5.25	-3.60	-1.65	5.57
MnCo	-5.62	-3.66	-1.96	2.51
MnNi	-5.58	-3.68	-1.90	3.40
MnCu	-4.58	-3.21	-1.37	3.51
FeFe	-5.93	-4.28	-1.65	4.16
FeCo	-6.26	-4.34	-1.92	3.20
FeNi	-5.96	-4.36	-1.6	2.92
FeCu	-5.06	-3.89	-1.17	2.16
CoCo	-6.32	-4.39	-1.93	2.42
CoNi	-6.15	-4.42	-1.73	1.03
CoCu	-5.31	-3.94	-1.37	0.66
NiNi	-6.06	-4.44	-1.62	0.00
NiCu	-5.11	-3.97	-1.14	0.21
CuCu	-4.22	-3.49	-0.73	0.80

Table S2. The values of thermal correction containing the zero-point energy, the enthalpic correction, and the entropy correction of the relevant molecules, and the adsorbed species. The temperature is set as 298 K.

Species	Zero-point energy (eV)	Enthalpic correction at 298K (eV)	Entropy at 298K (eV)	Correction for liquid phase (eV)	Solvent correction (eV)
CO ₂ (g)	0.31	0.13	0.66	/	/
H ₂ O (l)	0.57	0.1	0.58	-0.09	/
H ₂ (g)	0.27	0.09	0.40	/	/
C ₂ H ₅ OH (l)	2.11	0.20	0.83	-0.07	/
C ₂ H ₄ (g)	1.36	0.13	0.71	/	/
*H	0.17	0.03	0.02	/	/
*OH	0.37	0.05	0.08	/	-0.20
*COOH	0.6	0.08	0.22	/	-0.10
*CO	0.19	0.05	0.18	/	/
*CHO	0.51	0.08	0.08	/	/
*COH	0.57	0.06	0.09	/	-0.10
*CO-CO	0.41	0.13	0.29	/	/
*COCO	0.42	0.11	0.24	/	/
*CO-CHO	0.70	0.11	0.30	/	/
*COCHO	0.72	0.10	0.29	/	/
*CO-COH	0.74	0.13	0.24	/	-0.10
*COCOH	0.72	0.11	0.28	/	-0.10
*COCH ₂ O	1.05	0.12	0.23	/	/
*COHCHO	1.03	0.13	0.26	/	-0.10
*COHCH ₂ O	1.33	0.13	0.26	/	-0.10
*CHOHCH ₂ O	1.66	0.15	0.26	/	-0.10
*CHCH ₂ O	1.15	0.11	0.22	/	/
*CH ₂ CH ₂ O	1.55	0.12	0.18	/	/
*CH ₂ CH ₂ OH	1.86	0.12	0.26	/	-0.10
*CH ₃ CH ₂ O	1.89	0.12	0.24	/	/
*COHCOH	1.00	0.13	0.31	/	-0.10
*CCO	0.34	0.08	0.18	/	/
*CHCO	0.63	0.09	0.19	/	/
*CCOH	0.63	0.10	0.21	/	-0.10
*CH ₂ CO	0.92	0.10	0.13	/	/
*CHCOH	0.89	0.12	0.20	/	-0.10
*CH ₂ COH	1.24	0.13	0.17	/	-0.10
*CH ₂ CHOH	1.55	0.13	0.22	/	-0.10

Table S3. Values of $G(^*H)$, $G(^*OH)$ on various of DACs and magnetic moments, the DACs of which $G(^*H)$ or $G(^*OH)$ lower than -0.9 eV are excluded

TM atoms	$G(^*H)$ (eV)	Magnetic moment (μ_B)	$G(^*OH)$ (eV)	Magnetic moment (μ_B)
CrCr	-0.83	7.10	-1.79 (<-0.9 eV)	6.90
CrMn	-1.05 (<-0.9 eV)	0.34	-1.43 (<-0.9 eV)	6.23
CrFe	-0.23	1.48	-0.81	0.29
CrCo	-0.34	2.84	-0.86	2.72
CrNi	-0.32	3.57	-0.55	1.96
CrCu	0.19	2.89	-0.36	2.96
MnMn	-0.71	5.88	-1.13 (<-0.9 eV)	6.11
MnFe	-0.52	4.63	-0.80	4.48
MnCo	-0.50	3.41	-0.54	3.18
MnNi	0.07	2.97	0.00	2.45
MnCu	0.07	3.31	-0.39	3.50
FeFe	-0.46	3.24	-0.42	3.29
FeCo	-0.12	2.61	-0.12	1.25
FeNi	-0.19	1.68	-0.10	1.13
FeCu	0.09	1.56	-0.18	2.24
CoCo	-0.47	1.05	-0.45	0.00
CoNi	-0.33	0.00	0.13	0.59
CoCu	0.05	0.00	0.33	1.23
NiNi	-0.01	0.00	0.86	0.21
NiCu	0.25	0.50	0.76	0.69
CuCu	0.59	0.00	0.79	1.18

Table S4. Values of $G(*\text{COOH})$, $G(*\text{CO})$ on various of DACs and magnetic moments

TM atoms	$G(*\text{COOH})$ (eV)	Magnetic moment (μ_B)	$G(*\text{CO})$ (eV)	Magnetic moment (μ_B)
CrFe	0.44	0.98	-0.65	0.54
CrCo	0.29	0.55	-0.75	2.00
CrNi	0.36	3.48	-0.71	2.64
CrCu	0.19	2.60	-0.06	2.17
MnFe	-0.09	4.57	-1.07	4.75
MnCo	0.21	3.32	-1.01	1.66
MnNi	0.52	2.32	-0.35	2.50
MnCu	0.24	2.36	-0.38	2.08
FeFe	-0.25	3.21	-1.01	1.37
FeCo	0.39	2.73	-0.79	2.32
FeNi	0.32	1.55	-0.81	1.40
FeCu	0.05	0.24	-0.74	0.47
CoCo	0.07	1.52	-1.27	0.00
CoNi	0.26	0.00	-0.87	0.00
CoCu	0.10	0.00	-0.53	0.00
NiNi	1.02	0.00	0.17	0.00
NiCu	0.80	0.50	0.03	1.08
CuCu	1.15	0.66	0.19	0.04

Table S5. Values of G(*CHO), G(*COH) on various of DACs and magnetic moments

TM atoms	G(*CHO) (eV)	Magnetic moment (μ_B)	G(*COH) (eV)	Magnetic moment (μ_B)
CrFe	0.74	0.72	0.56	1.54
CrCo	0.64	0.02	0.31	2.67
CrNi	0.57	1.59	0.81	2.03
CrCu	0.52	2.47	1.31	1.13
MnFe	-0.08	4.64	-0.03	2.56
MnCo	0.46	3.11	0.31	2.12
MnNi	0.68	2.11	1.17	1.35
MnCu	0.37	2.12	1.22	1.43
FeFe	-0.07	3.20	0.28	2.08
FeCo	0.59	2.63	0.50	0.80
FeNi	0.46	1.51	0.68	0.00
FeCu	0.21	0.49	1.05	0.00
CoCo	0.18	1.47	0.03	0.00
CoNi	0.29	0.00	0.99	0.20
CoCu	0.15	0.00	1.49	0.00
NiNi	0.50	0.00	2.11	1.63
NiCu	0.72	0.49	2.14	0.00
CuCu	1.26	0.00	2.42	0.00

Table S6. Values of $G(*\text{CO-CO})$, $G(*\text{CO-CHO})$ and $G(*\text{CO-COH})$ on various of DACs and magnetic moments

TM atoms	$G(*\text{CO-CO})$ (eV)	Magnetic moment (μ_B)	$G(*\text{CO-CHO})$ (eV)	Magnetic moment (μ_B)	$G(*\text{CO-COH})$ (eV)	Magnetic moment (μ_B)
CrFe	-0.94	1.56	-0.30	0.31	0.11	0.11
CrCo	-1.15	1.81	-0.31	1.50	-0.07	1.08
CrNi	-0.96	1.49	-0.05	1.53	0.45	1.25
CrCu	-0.57	1.09	0.48	1.70	0.80	0.25
MnFe	-1.33	1.82	-1.08	2.67	-0.01	1.27
MnCo	-1.15	1.14	-0.27	1.45	-0.32	0.00
MnNi	-0.72	0.00	0.26	1.31	0.83	0.00
MnCu	-0.85	0.00	0.23	0.88	0.73	0.00
FeFe	-1.15	0.72	-0.75	1.58	0.32	1.16
FeCo	-0.89	0.00	-0.41	0.42	0.64	0.48
FeNi	-1.36	0.00	-0.25	0.00	0.58	0.00
FeCu	-1.10	0.24	-0.07	0.00	0.93	0.73
CoCo	-0.92	0.06	-0.93	0.00	-0.06	0.00
CoNi	-0.72	0.95	0.00	0.00	0.66	0.35
CoCu	-0.45	0.47	0.14	0.95	1.75	0.23
NiNi	0.70	1.70	0.91	0.64	1.39 (*COCOH)	0.00
NiCu	0.26	0.99	1.17	1.05	1.81 (*COCOH)	1.70
CuCu	0.14	0.00	1.21	0.00	1.49 (*COCOH)	0.69

Table S7. Values of $G(*\text{COCO})$, $G(*\text{COCHO})$ and $G(*\text{COCOH})$ on various of DACs and magnetic moments

TM atoms	$G(*\text{COCO})$ (eV)	Magnetic moment (μ_B)	$G(*\text{COCHO})$ (eV)	Magnetic moment (μ_B)	$G(*\text{COCOH})$ (eV)	Magnetic moment (μ_B)
CrFe	-0.12	0.00	-0.11	0.76	0.19	0.00
CrCo	1.02	0.34	0.19	0.35	-0.05	0.98
CrNi	0.94	0.96	0.47	1.14	0.11	1.82
CrCu	0.98	1.83	0.37	2.29	0.66	2.54
MnFe	-0.37	0.73	-0.25	4.44	0.07	0.68
MnCo	-0.11	2.51	0.16	2.82	0.22	1.84
MnNi	0.59	2.10	0.60	2.03	0.97	2.62
MnCu	1.03	2.11	0.21	2.01	0.60	3.12
FeFe	-0.26	2.28	-0.23	2.83	-0.11	1.88
FeCo	0.05	1.75	0.22	2.44	0.27	1.11
FeNi	-1.36 (*CO-CO)	0.00	0.36	1.46	0.75	0.00
FeCu	0.03	0.00	0.08	0.37	0.58	1.95
CoCo	-0.29	0.00	0.01	1.26	-0.24	0.00
CoNi	0.12	0.00	-0.10	0.00	0.97	0.00
CoCu	0.50	0.27	0.07	0.00	1.15	0.00
NiNi	1.17	0.44	0.47	0.00	1.39	0.00
NiCu	1.46	0.00	0.98	0.00	1.81	1.70
CuCu	0.66	0.00	1.20	0.39	1.86	0.69

Table S8. Values of $G(^*COCH_2O)$, $G(^*COHCHO)$, $G(^*COHCH_2O)$, $G(^*CHOHCH_2O)$, $G(^*CHCH_2O)$, $G(^*CH_2CH_2O)$, $G(^*CH_2CH_2OH)$, $G(^*CH_2CH_2OH)$ and $G(^*O)$ on various of DACs and magnetic moments

TM atoms	$G(^*C$ OCH ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ OHCH O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ OHCH ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ HOHC H ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ HCH ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ H ₂ CH ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ H ₂ CH ₂ OH) (eV)	Magne tic mome nt (μ_B)	$G(^*C$ H ₃ CH ₂ O) (eV)	Magne tic mome nt (μ_B)	$G(^*O)$ (eV)	Magne tic mome nt (μ_B)
CrFe	-0.47	0.84	-0.40	0.79	-0.14	1.58	0.17	0.00	0.04	0.00	-0.60	0.00	-1.60	0.54	-2.35	0.32	-2.15	0.69
CrCu	0.72	2.99	0.21	1.77	1.02	1.89	0.00	1.93	0.55	2.45	-0.87	2.01	-1.26	2.65	-2.00	2.69	-1.51	1.68
MnCu	0.62	2.21	0.57	1.67	1.12	1.62	0.59	2.52	0.79	1.49	0.01	2.49	-1.20	2.93	-1.96	3.47	-1.01	2.74
FeFe	-0.44	3.63	-0.22	2.44	-0.14	3.45	-0.31	4.07	-0.42	3.48	-0.96	4.06	-1.41	3.32	-1.90	5.01	-1.51	3.30
FeCo	0.34	3.16	0.50	2.46	0.80	2.45	0.39	3.13	0.52	2.63	-0.27	3.20	-0.86	3.26	-1.67	1.36	-1.02	0.60
CoNi	0.39	0.81	0.69	1.51	1.17	0.64	0.55	0.86	1.01	0.64	0.02	1.00	-0.90	0.00	-1.35	0.83	-0.65	0.94
CoCu	0.45	0.00	0.65	1.05	1.29	0.00	0.61	0.00	1.14	0.00	0.12	0.00	-1.14	0.00	-1.17	1.02	-0.20	2.00
NiCu	0.97	0.00	1.32	2.50	1.92	1.72	1.14	1.08	1.87	1.70	0.85	0.00	-0.51	0.64	-0.78	0.59	0.38	1.09

Table S9. Values of adsorbed species on Co-Co and magnetic moments

	Adsorbed species (eV)	Magnetic moment (μ_B)
G(*COHCOH)	0.78	0.00
G(*CCO)	-0.30	0.73
G(*CHCO)	-0.96	0.00
G(*CCOH)	0.46	1.08
G(*CH ₂ CO)	-1.23	0.00
G(*CHCOH)	0.02	0.00
G(*CH ₂ COH)	-0.97	0.00
G(*CH ₂ CHOH)	-1.22	0.00
G(*CH ₂ CH ₂ OH)	-1.03	1.88
G(*O)	-1.31	1.07