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## SmBaCo<sub>2</sub>O<sub>5+ $\delta$ </sub> double perovskite with epitaxially grown Sm<sub>0.2</sub>Ce<sub>0.8</sub>O<sub>2- $\delta$ </sub>

## nanoparticles as the promising cathode for solid oxide fuel cells

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## **Supporting information**



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Figure S1. Rietveld refinement of room temperature XRD patterns of SBCO and SBCO-SDC



Figure S2. Polarization resistance  $(R_d)$  versus reciprocal temperature for SBCO-SDC and SBCO electrodes in air.

Cathode	anode	Electrolyte	Power		
		thickness /	Temperatur	density	nafananaa
			e / °C	/ mW	reference
		μm		cm <sup>-2</sup>	
$NdBa_{0.5}Sr_{0.5}Co_2O_{5+\delta}$	Ni–GDC	300	850	904	[1]
$Pr_{1.1}Ba_{0.9}Co_2O_{5+\delta}$	Ni-SDC	300	800	732	[2]
$PrBa_{0.5}Sr_{0.5}Co_2O_{5+\delta}$	Ni-GDC	300	800	1021	[3]
PrBaC <sub>2</sub> O <sub>5+ð</sub> -SDC				758	
$NdBaC_{2}O_{5+\delta}\text{-}SDC$	Ni-SDC	300	800	707	[4]
$SmBaC_2O_{5+\delta}$ -SDC				685	

**Tab. S1**. Comparison of peak power density for LSGM electrolyte-supported cells between  $SmBaCo_2O_{5+\delta}$ -SDC NPs and selected double perovskite cathodes.

$GdBaC_2O_{5+\delta}$ -SDC				608	
$YBaCo_{1.4}Cu_{0.6}O_{5+\delta}$	Ni-GDC	300	850	815	[5]
$YBa_{0.5}Sr_{0.5}Co_{1.4}Cu_{0.6}O_{5+\delta}$	Ni–GDC	300	850	398	[6]
$SmBa_{0.5}Sr_{0.5}CoCuO_{5+\delta}$	NiCu–GDC	300	850	857	[7]
$NdBaCoFeO_{5+\delta}30SDC$	Ni-SDC	300	800	892	[8]
$PrBa_{0.8}Ca_{0.2}Co_2O_{5+\delta}$	$PrBaMn_2O_{5+\delta}$	250	700	460	[9]
$GdBa_{0.4}Sr_{0.6}Co_2O_{5+\delta}$	Ni-GDC	500	800	490	[10]
$NdBaCo_{2/3}Fe_{2/3}Cu_{2/3}O_{5+\delta}$	Ni-GDC	300	800	719	[11]
$PrBa_{0.5}Sr_{0.5}Co_{1.5}Fe_{0.5}O_{5+\delta}$	Ni-SDC	300	850	697	[12]
SmBaCo <sub>2</sub> O <sub>5+ð</sub> -SDC NPs	Ni-GDC		850	977	
			800	806	This
		300	750	592	work
			700	408	
			650	230	

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