

Supplementary information

Pt decorated hierarchical Sb₂WO₆ microsphere as a surface functionalized photocatalyst for the visible-light-driven reduction of nitrobenzene to aniline

Yingzhang Shi,^a Huan Wang,^a Zhiwen Wang,^a Tai kang Wu,^a Yujie Song,^a Binbin Guo,^{a,b} Ling Wu^{a,b*}

^a State Key Laboratory of Photocatalysis on Energy and Environment, Fuzhou University, Fuzhou, Fujian 350116, P. R. China.

^b State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Science, Fuzhou, Fujian 350002, P. R. China.

Corresponding author: Ling Wu

E-mail: wuling@fzu.edu.cn

Tel: +86-591-22865835

Table S1 XPS fitting results for the prepared SWO-NS and Pt/SWO-NS.

Samples	Peak	Assignment	BE (eV)	Fraction (%)
SWO-NS	W 4f _{5/2} / 4f _{7/2}	W ⁶⁺	37.70/ 35.50	57.61
		W ⁵⁺	36.40/ 34.40	42.39
Pt/SWO-NS		W ⁶⁺	37.70/ 35.50	63.51
		W ⁵⁺	36.60/ 34.60	36.49
Pt/SWO-NS	Pt 4f _{5/2} / 4f _{7/2}	Pt ⁰	71.10/74.45	5.46
		Pt ²⁺	72.20/75.55	94.54

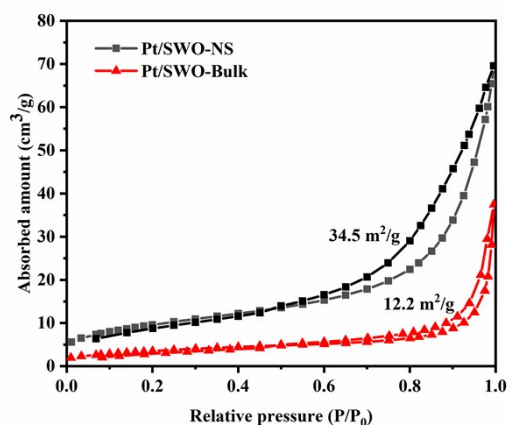


Fig. S1 Nitrogen (N₂) adsorption-desorption isotherms of SWO-NS and SWO-Bulk.

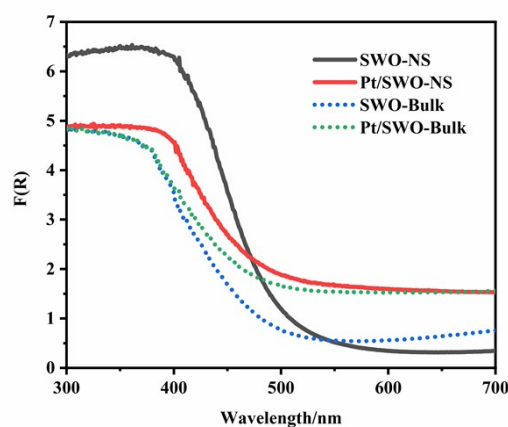


Fig. S2 UV-vis DRS of SWO-NS, Pt/SWO-NS, SWO-Bulk, Pt/SWO-Bulk.

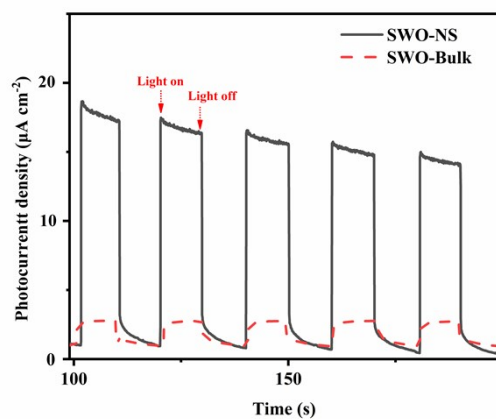


Fig. S3 Photocurrent responses of SWO-NS and SWO-Bulk under visible light irradiation.

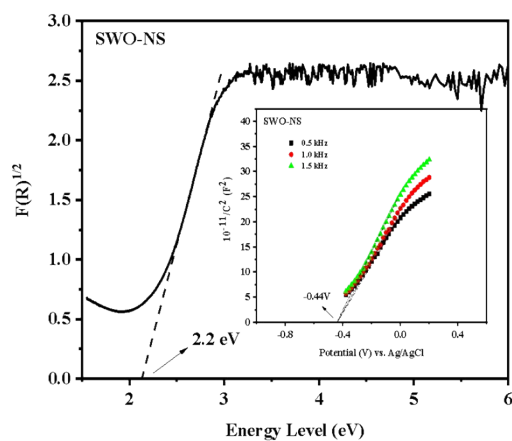


Fig. S4 $(F(R))^{0.5}$ as a function of photon energy ($h\nu$), where F is the Kubelka-Munk function. and the inset is the Mott-Schottky plots of SWO-NS.

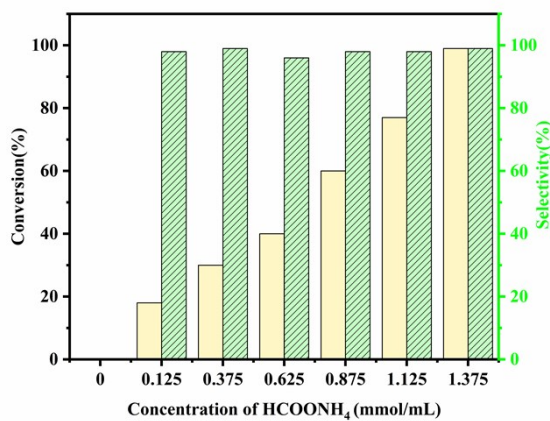


Fig. S5 Conversion of nitrobenzene and selectivity of aniline with different concentration of

HCOONH₄. Reaction condition: Catalyst 10 mg, HCOONH₄/methanol solution 1.5 mL, nitrobenzene 0.1 mmol, Ar, 1 atm, $\lambda \geq 420$, time 2 h, room temperature.

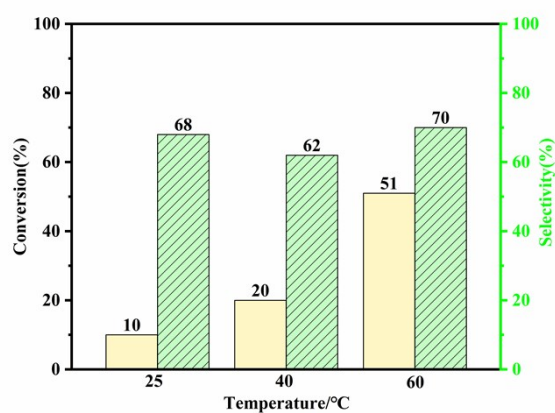


Fig. S6 Conversion of nitrobenzene and selectivity of aniline with different reaction temperature in the dark. Reaction condition: Catalyst 10 mg, 1.375 mmol/mL HCOONH₄/methanol solution 1.5 mL, nitrobenzene 0.1 mmol, Ar, 1 atm, time 2 h, room temperature.

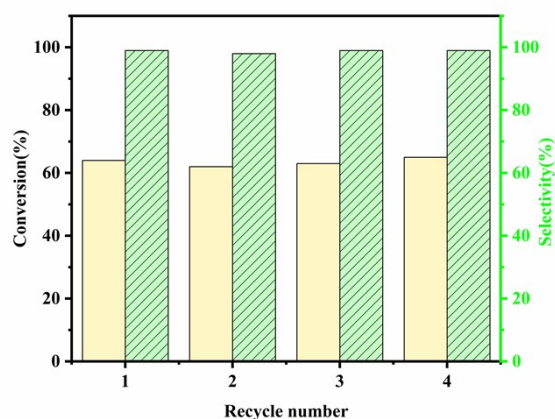


Fig. S7 Recycle experiment in constant temperature. Reaction condition: Catalyst 10 mg, 1.375 mmol/mL HCOONH₄/methanol solution 1.5 mL, nitrobenzene 0.1 mmol, Ar, 1 atm, $\lambda \geq 420$, time 2 h, 298K. Reaction temperature was controlled via a jacket connected to a circulating water source to avoid the thermal effect under the light irradiation.

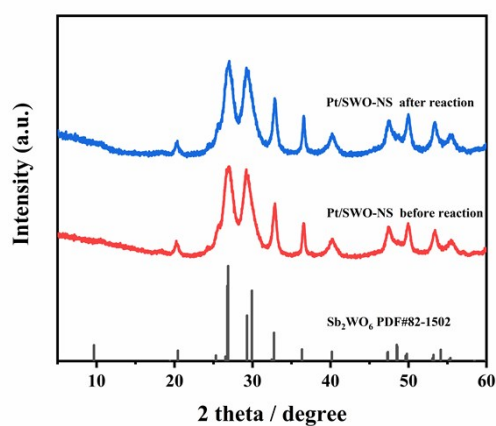


Fig. S8 XRD patterns of the Pt/SWO-NS sample (before and after reaction).

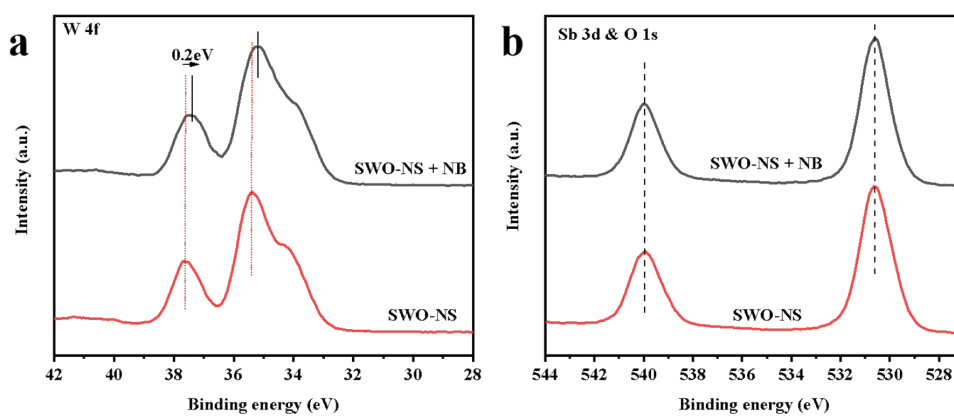


Fig. S9 XPS spectrum of SWO-NS and SWO-NS + NB. (a) W 4f; (b) Sb 3d & O 1s.

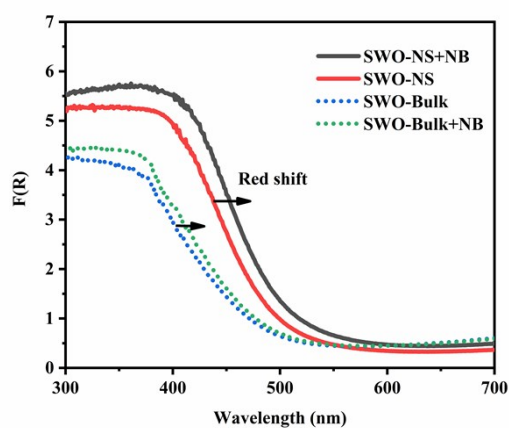


Fig. S10 UV-vis diffuse reflectance spectra of SWO-NS, SWO-Bulk, SWO-NS/NB and SWO-Bulk/NB.

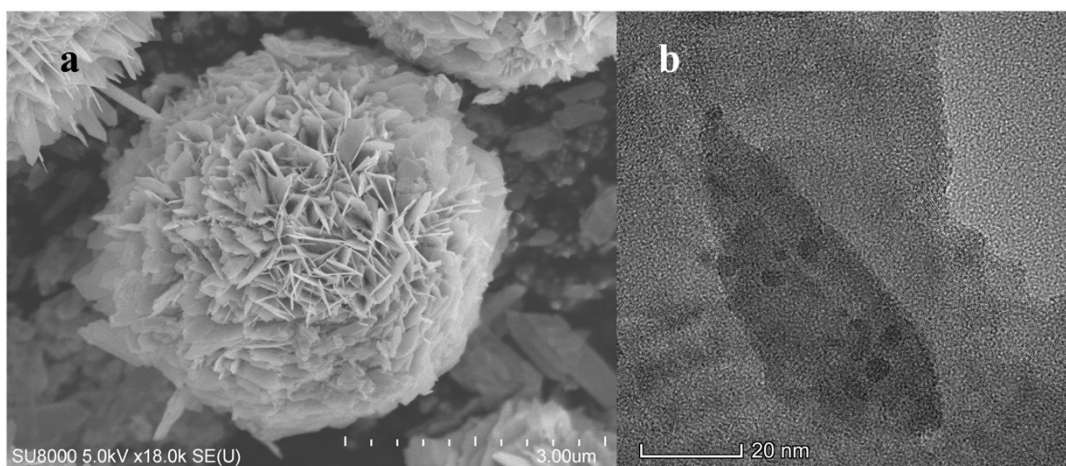


Fig. S11 SEM (a) and TEM (b) images of Pt/SWO-NS after the catalysis.

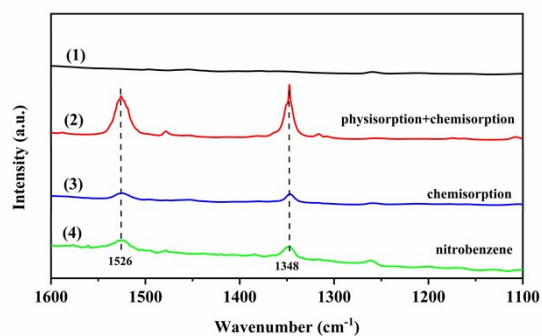


Fig. S12 In-situ FTIR spectra of Pt/SWO-NS adsorbed NB molecules. 1) After degassing in vacuum at 180 °C for 3 h; 2) adsorption of nitrobenzene for 0.5 h at room temperature (physorption +chemisorption); 3) further degassing in vacuum at 180 °C for 5 min (chemisorption); 4) adsorption of nitrobenzene on KBr.