

## Supporting Information

# Durable Photoelectrochemical CO<sub>2</sub> Reduction with Water Oxidation using a Visible-Light Driven Molecular Photocathode

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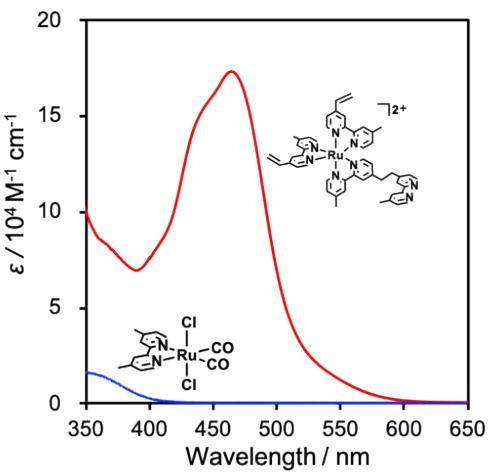
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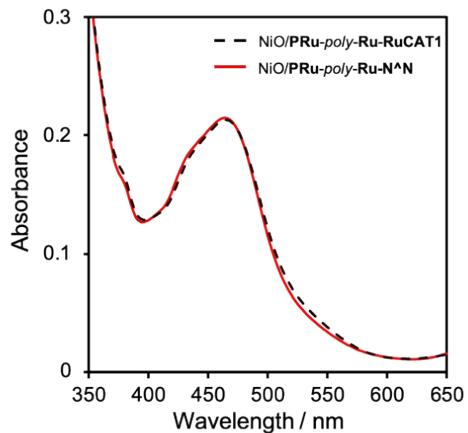
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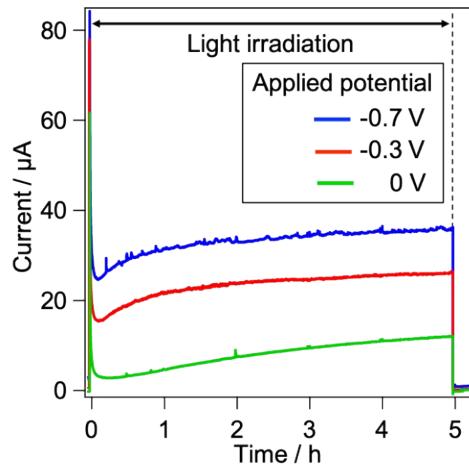
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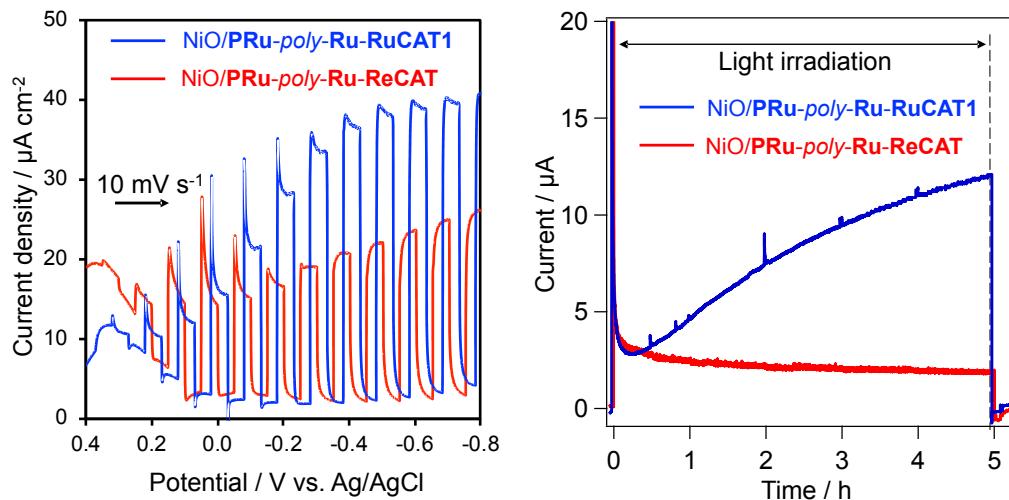
**Figure S1.** Absorption spectra of **VRu-N<sup>N</sup>** (red line) and *cis*-(CO)-*trans*-(Cl)-Ru(dmb)(CO)<sub>2</sub>Cl<sub>2</sub> (blue line) in MeCN solutions.



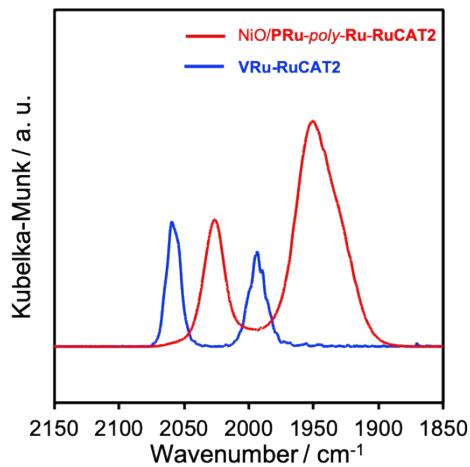
**Figure S2.** UV-vis absorption spectra of **NiO/PRu-poly-Ru-N<sup>N</sup>** (red line), and **NiO/PRu-poly-Ru-RuCAT1** (black dashed line). FTO electrode was employed as the background.



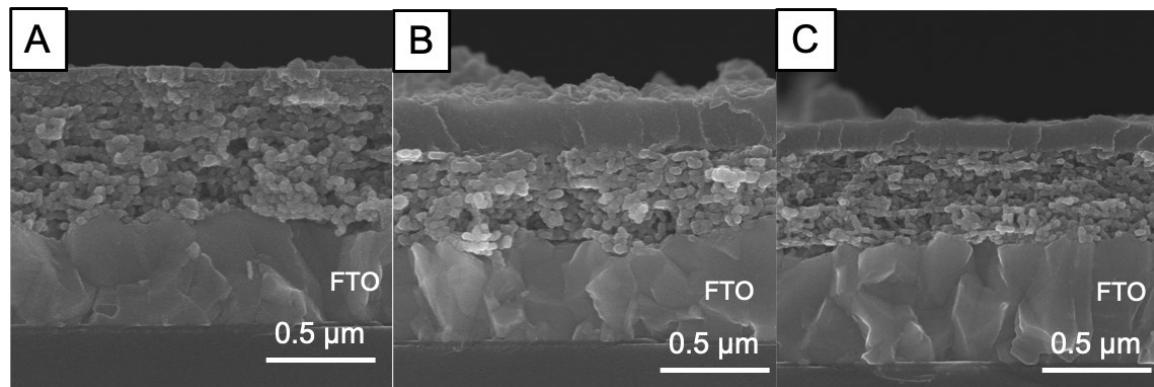
**Figure S3.** Time courses of photocurrent using NiO/PRu-poly-Ru-RuCAT1 (electrode area: 2.5 cm<sup>-2</sup>) at  $E = -0.7$  (blue line),  $-0.3$  (red line), and  $0$  (green line) V vs. Ag/AgCl under light irradiation (460 nm <  $\lambda_{ex}$  < 650 nm, 27 mW cm<sup>-2</sup>) in a CO<sub>2</sub>-purged NaHCO<sub>3</sub> (50 mM) aqueous solution (pH = 6.6).



**Figure S4.** Current-potential curves and time courses of photocurrent at  $E = 0$  V vs. Ag/AgCl using NiO/PRu-poly-Ru-RuCAT1 and the polymer photocathode with Re catalyst (NiO/PRu-poly-Ru-Re)<sup>1</sup> under light irradiation (460 nm <  $\lambda_{ex}$  < 650 nm, 27 mW cm<sup>-2</sup>) in a CO<sub>2</sub>-purged NaHCO<sub>3</sub> (50 mM) aqueous solution (pH = 6.6).



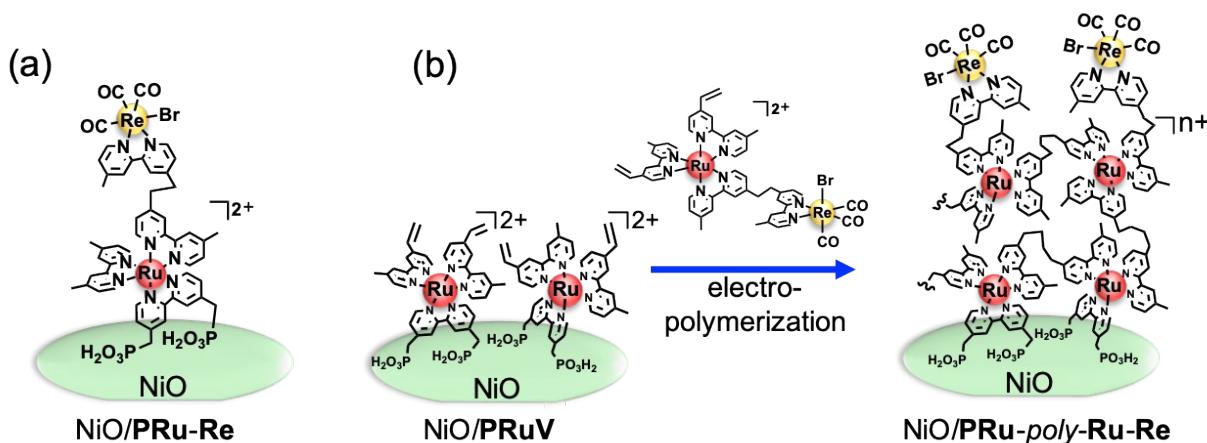
**Figure S5.** FT-IR spectra of NiO/**PRu-poly-Ru-RuCAT2** (red line) and **VRu-RuCAT** on NiO electrode (blue line). A diffuse reflection unit was used for the measurements and a bare NiO electrode was employed as the background.



**Figure S6.** Cross-sectional scanning electron microscopy (SEM) images of A) NiO/**PRu-poly-Ru-RuCAT1**, B) NiO/**PRu-poly-Ru-RuCAT2** and C) NiO/**PRu-poly-Ru-Re**.

**Table S1.** Value of  $n_{\text{cat}}$ ,  $n_{\text{total}}$ , and  $n_{\text{PRuV}}$ .

Entry	$n_{\text{PRuV}} / \text{nmol}$	$n_{\text{total}} / \text{nmol}$	$n_{\text{cat}} / \text{nmol}$
1	11.6	82	35.2
2	12.9	78	32.6
3	12.3	87	37.4
4	13.3	94	40.8



**Scheme S1.** Preparation scheme for (a) NiO/PRu-ReCAT and (b) NiO/PRu-poly-Ru-Re.

## Reference

1. R. Kamata, H. Kumagai, Y. Yamazaki, G. Sahara and O. Ishitani, *ACS Appl. Mater. Interfaces*, 2019, **11**, 5632-5641.