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Supplementary information

Recent advances in layered Ln₂NiO_{4+δ} nickelates: fundamentals and prospects for their applications in protonic ceramic fuel and electrolysis cells

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Figure S1. Relationships between compositions of nickelate materials with their oxygen-ionic transport features: oxygen-ionic conductivity of the A-site doped nickelates measured at 900 °C (a), 800 °C (b) and 700 °C (c); oxygen-ionic conductivity of the B-site doped lanthanum nickelates (d) and praseodymium nickelates (e); ionic transference number (f) and activation energy values (y) of ionic-conductivity for the basic and doped nickeltaes. These data were extracted from Table 2.

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Figure S2. Oxygen surface exchange and diffusion coefficient values as well as the corresponding apparent activation energies for $Ln_2NiO_{4+\delta}$ -based compounds. These data were presented on the basis of Table 3.



Figure S3. Concentration dependences of average thermal expansion coefficients for some undoped and A-site doped $La_2NiO_{4+\delta}$ (a), $Nd_2NiO_{4+\delta}$ (b), $Pr_2NiO_{4+\delta}$ (c) as well as for B-site doped $La_2NiO_{4+\delta}$ (d) and $Nd_2NiO_{4+\delta}$ (e). These data were extracted from Table 5.



Figure S4. Electrochemical behavior of the nickelate electrodes measured for symmetrical cells: polarization resistances of the A-doped nickelates at 500 °C (a), 600 °C (b) and 700 °C (c); polarization resistances of the undoped La₂NiO_{4+ δ} (d) Cudoped La₂NiO_{4+ δ} (e); activation energy values of the overall polarization conductivity. These data were presented on the basis of Table 4.



Figure S5. Performance of PCFCs with cathodes based on $Pr_2NiO_{4+\delta}$ (a,c) or $La_2NiO_{4+\delta}$ (b,d): polarisation resistances (a,b) and maximal power densities (c,d). These data were presented on the basis of Tables 9 and 10. MD is the multi-doped oxides. Composites are two-phase systems containing the undoped $La_2NiO_{4+\delta}$ component (weight ratios can be found in the corresponding tables).