

Boosting alkaline hydrogen evolution performance of Co₄N porous nanowires by interface engineering of CeO₂ tuning

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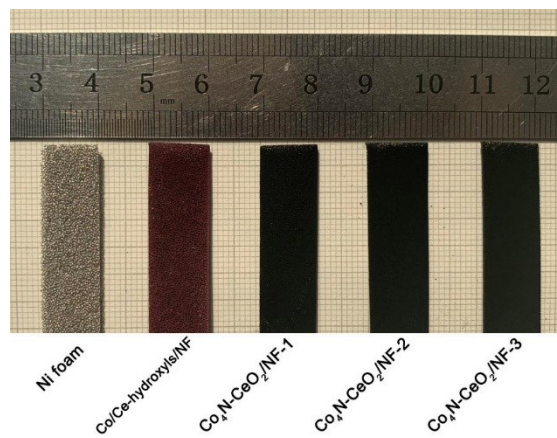


Figure S1. Optical pictures of as-prepared samples.

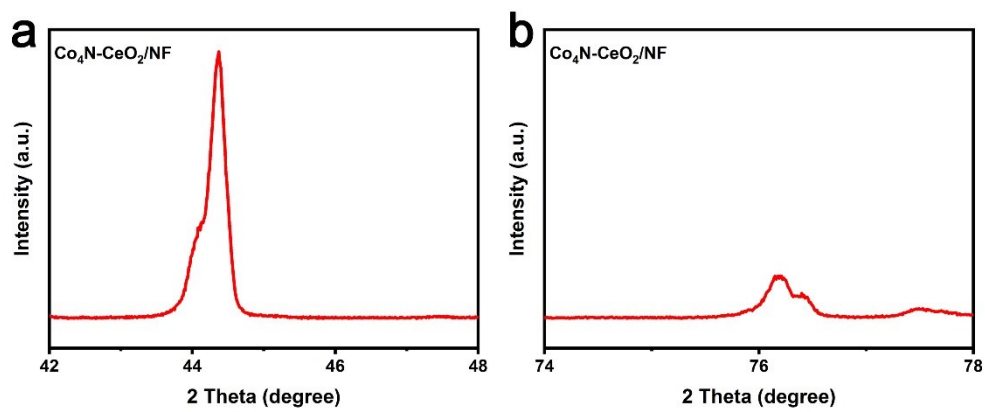


Figure S2. The enlarged XRD patterns of the Co₄N-CeO₂/NF.

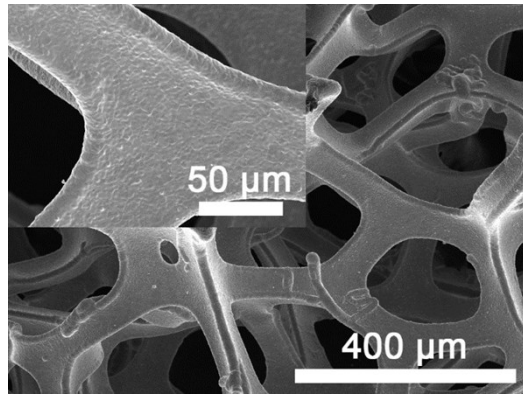


Figure S3. Typical SEM images of the starting Ni foam.

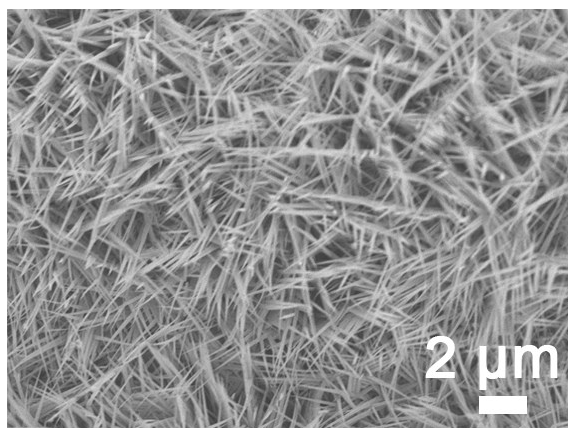


Figure S4. SEM image of $\text{Co}_3\text{O}_4\text{-CeO}_2/\text{NF}$ nanowires on Ni foam.

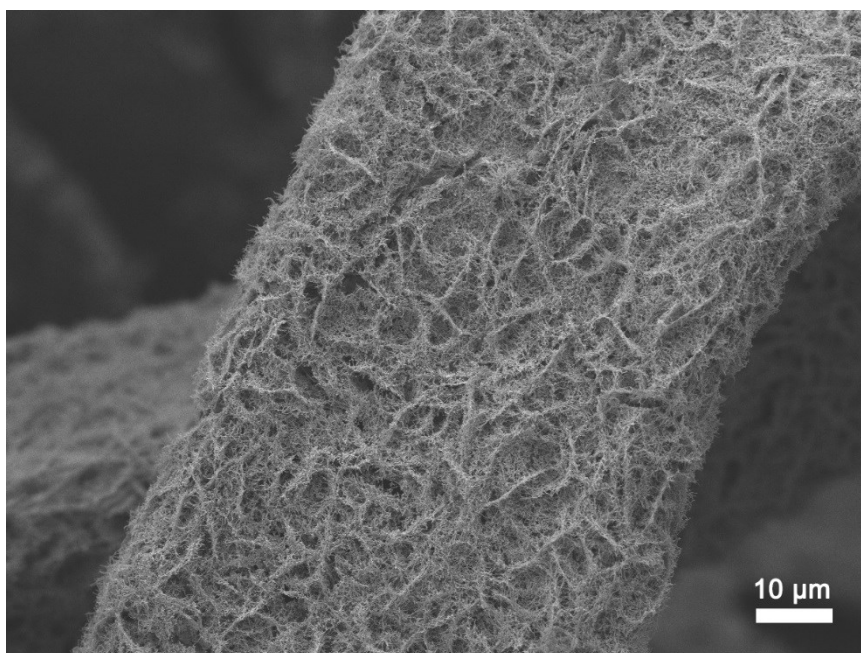


Figure S5. The low-resolution SEM image of $\text{Co}_4\text{N-CeO}_2/\text{NF-2}$ nanoarrays on Ni foam

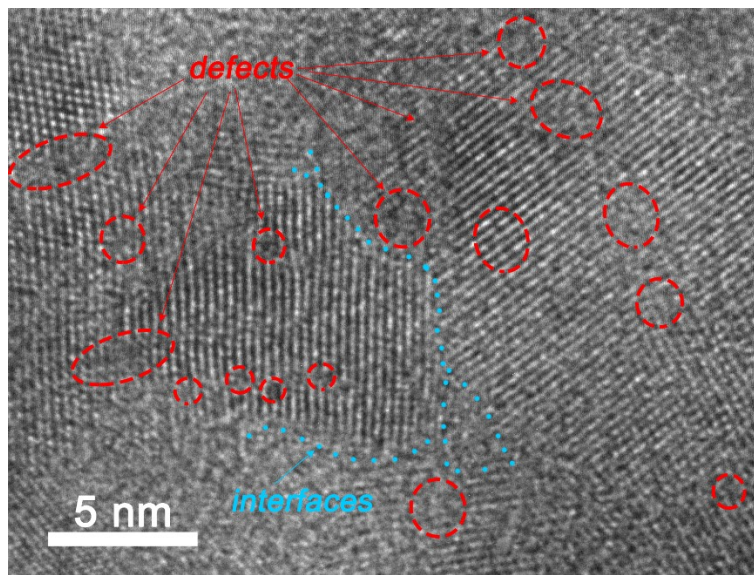


Figure S6. HRTEM images of Co₄N-CeO₂/NF.

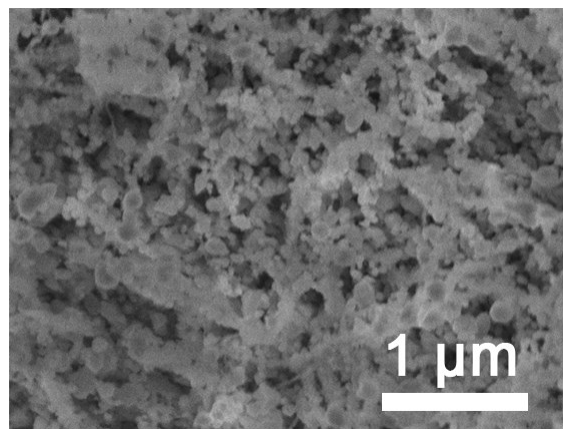


Figure S7. SEM image of pure Co₄N on Ni foam.

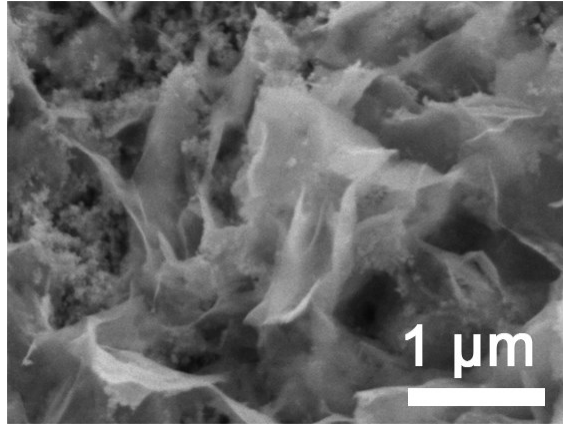


Figure S8. SEM images of pure CeO₂ on Ni foam.

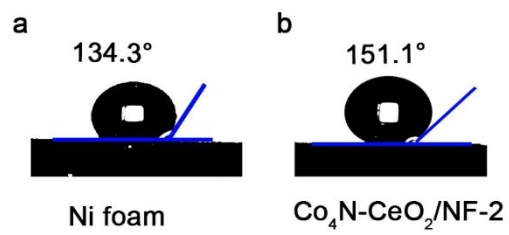


Figure S9. Digital images of air bubbles on (a) blank Ni foam and (b) Co₄N-CeO₂/NF-2.

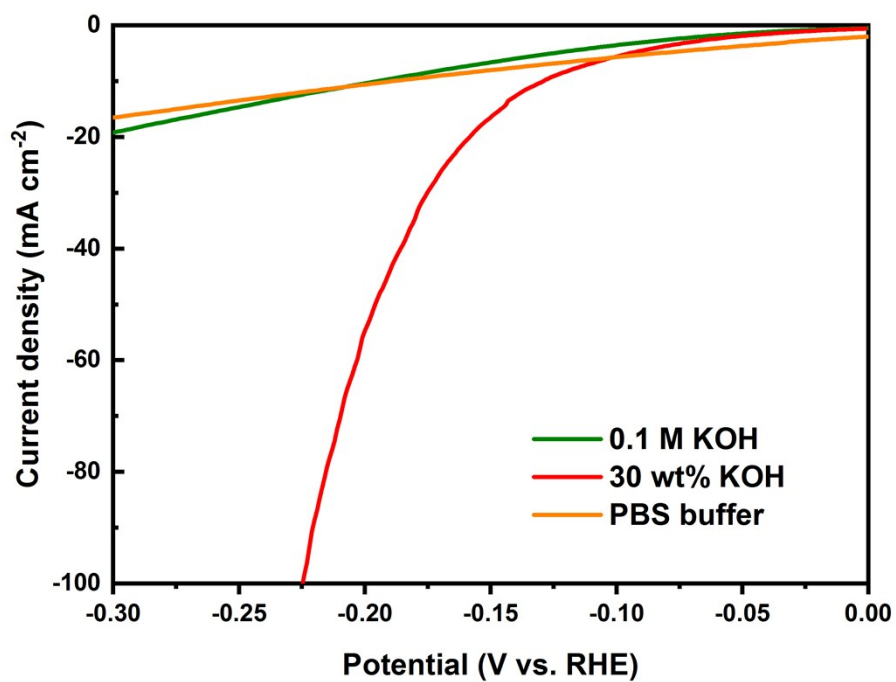


Figure S10. HER polarization curves of Co₄N-CeO₂/NF-2 electrodes in different electrolyte of 0.1 M KOH, 30 wt% KOH and PBS buffer solution.

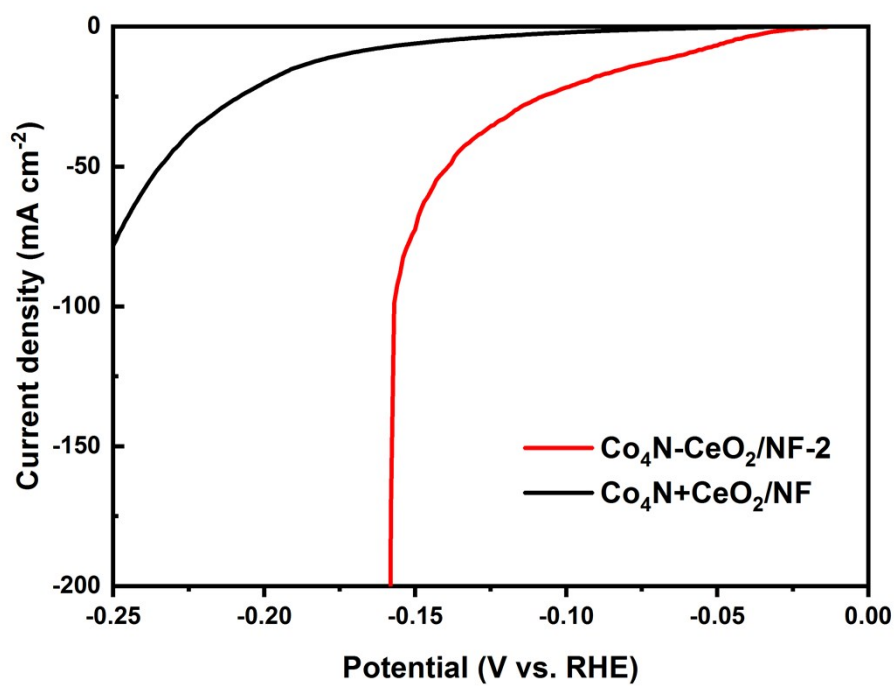


Figure S11. HER polarization curves of the Co₄N-CeO₂/NF-2 electrode compared to the physical mixture of Co₄N and CeO₂ casted on the Ni foam.

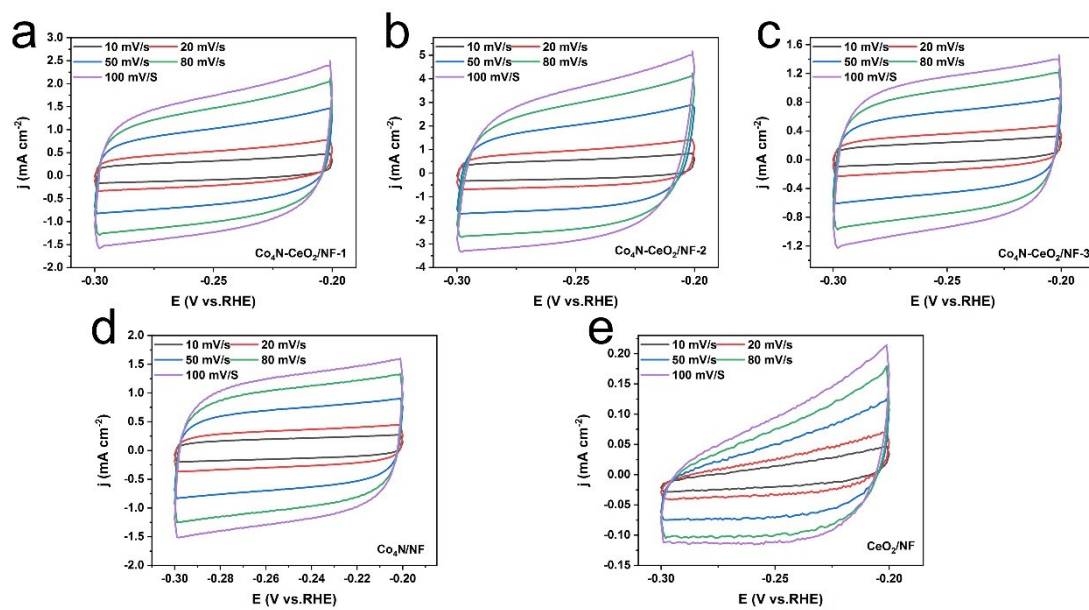


Figure S12. Typical cyclic voltammograms at different scan rates. (a) $\text{Co}_4\text{N-CeO}_2/\text{NF-1}$, (b) $\text{Co}_4\text{N-CeO}_2/\text{NF-2}$, (c) $\text{Co}_4\text{N-CeO}_2/\text{NF-3}$, (d) $\text{Co}_4\text{N}/\text{NF}$ and (e) CeO_2/NF with scan rates ranging from 10 mV/s to 100 mV/s.

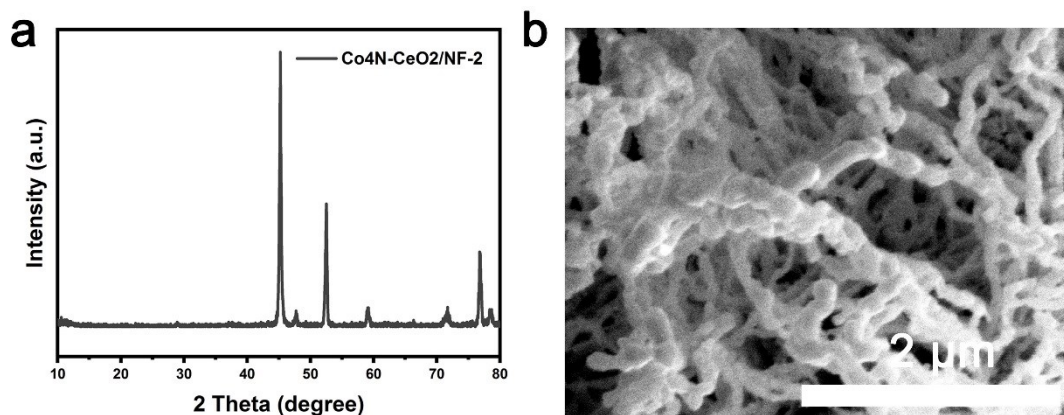


Figure S13. Characterizations after the HER stability test. (a) The XRD pattern and (b) SEM image of $\text{Co}_4\text{N-CeO}_2/\text{NF-2}$ after HER stability test.

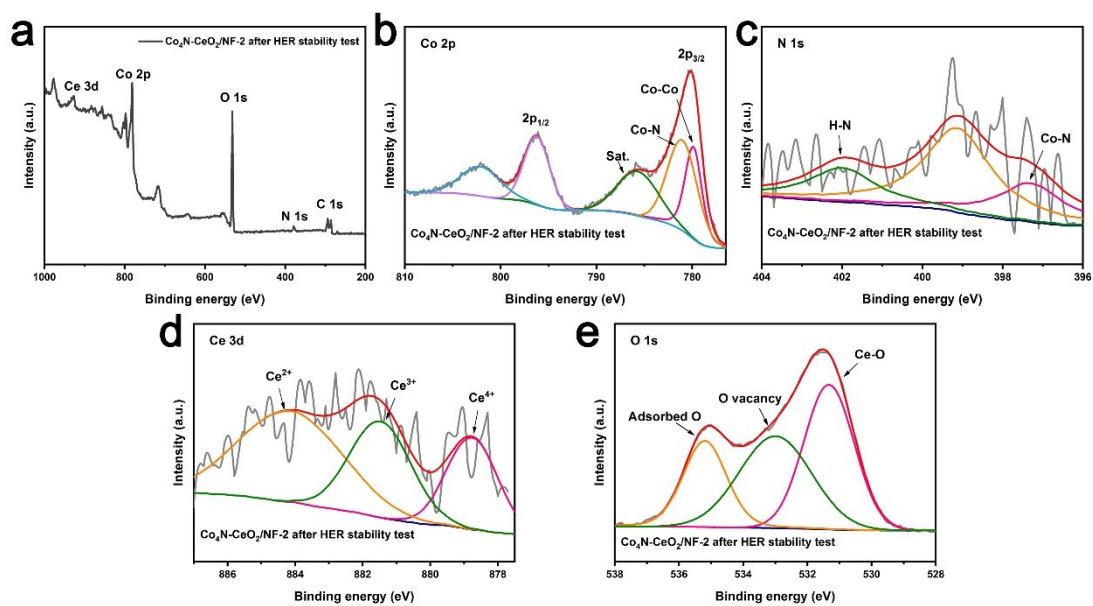


Figure S14. XPS spectra of (a) survey, (b) Co 2p, (c) N 1s, (d) Ce 3d, and (e) O 1s for the $\text{Co}_4\text{N-CeO}_2/\text{NF-2}$ electrode after HER stability test.

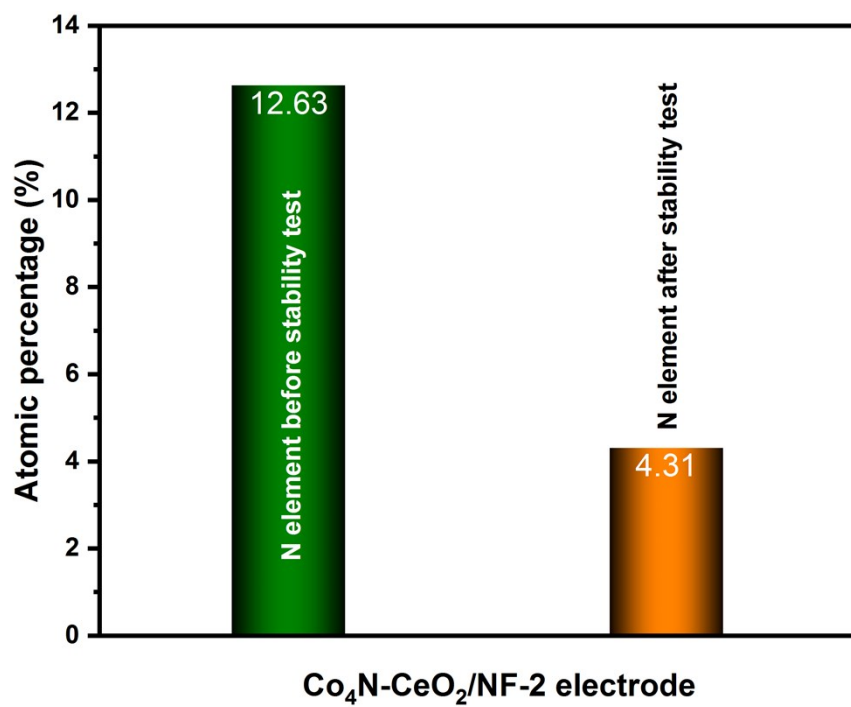


Figure S15. The atomic percentage of N element before and after HER stability test.

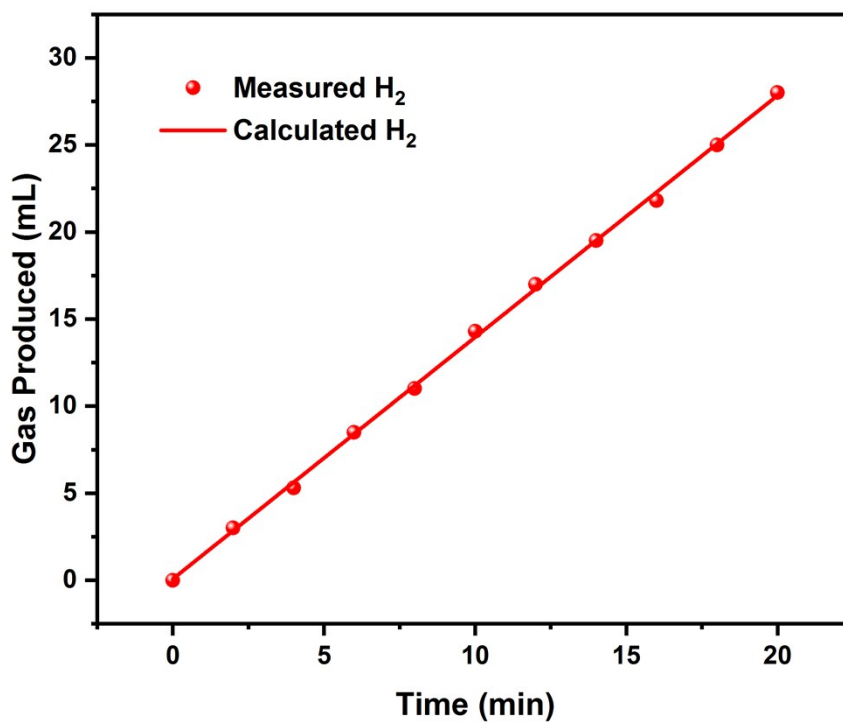


Figure S16. Experimental and theoretical calculated amount of the evolved H₂ over the Co₄N-CeO₂/NF-2 electrode at a constant current density of 200 mA cm⁻² in 1.0 M KOH electrolyte.

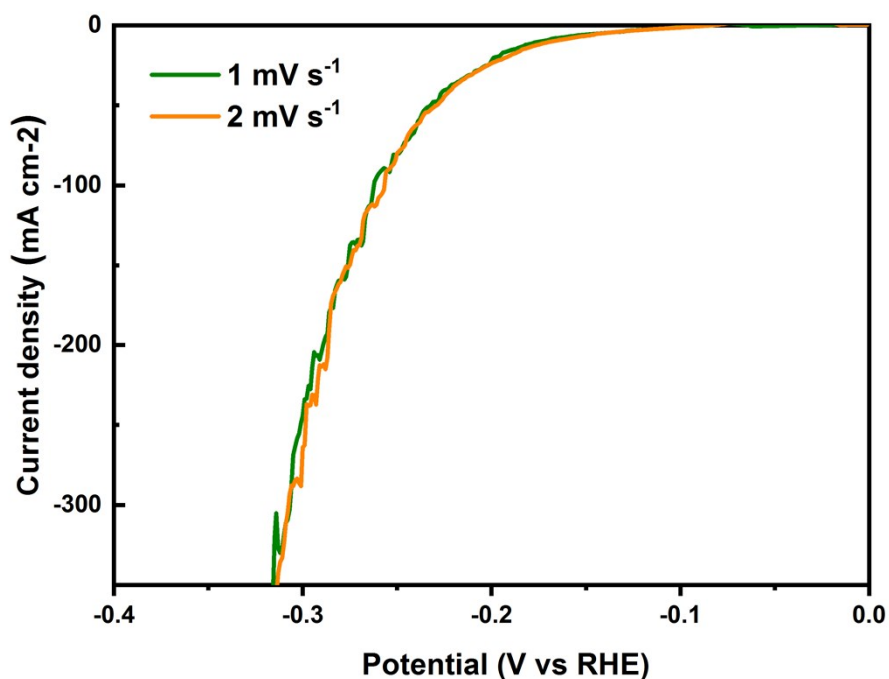


Figure S17. The HER polarization curves without iR -correction of the $\text{Co}_4\text{N-CeO}_2/\text{NF-2}$ different sweep speeds.

Table S1. Comparison of the electrocatalytic HER activity for $\text{Co}_4\text{N-CeO}_2/\text{NF}$ and reported catalysts in 1.0 M KOH electrolyte.

Catalysts	Substrate	$\eta_{j=10}$ (mV)	b (mV dec^{-1})	Refs
Ni/CeO ₂	CNT	91	---	S1
CoN/Ni ₃ N	Carbon cloth	68	69	S2
Cu ₄ N/Ni ₃ N	Carbon cloth	71.4	106.5	S3
Mo ₂ N/Mo ₂ C	GC	152	68	S4
CoP/NiCoP	Ti foil	133	88	S5
Co ₃ O ₄ -CeO ₂	Carbon cloth	90	57.8	S6
NF@NiFe LDH/CeO _x	Ni foam	154	101	S7
Ni ₃ N-CeO ₂	Ti mesh	80	122	S8
Co ₄ N-CeO ₂	Ni foam	52	56.8	This work

Table S2. Comparison of the electrocatalytic HER activity for Co₄N-CeO₂/NF and reported catalysts at high current density of 50 or 100 mA cm⁻² in 1.0 M KOH electrolyte.

Catalysts	Substrate	$\eta_{j=50/100}$ (mV)	Refs
Ni ₃ S ₂ /MnO ₂	Ni foam	$\eta_{100}=197$	S9
NiFe LDH@Ni ₃ N	Ni foam	$\eta_{100}=142$	S10
Ni _x Co _{3-x} S ₄ /Ni ₃ S ₂	Ni foam	$\eta_{100}=258$	S11
Co-doped CeO ₂	Cu foam	$\eta_{100}=132$	S12
CeO ₂ -Cu ₃ P	Ni foam	$\eta_{50}=228$	S13
NiFe LDH/CeO _x	Ni foam	$\eta_{100}=267$	S14
Sn-Ni ₃ S ₂	Ni foam	$\eta_{100}=171$	S15
CoNi/CoFe ₂ O ₄	Ni foam	$\eta_{100}=189$	S16
Co ₄ N-CeO ₂	Ni foam	$\eta_{100}=149$	This work

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