

Supplementary information

## High-performance Gas Sensor Array for Indoor Air Quality Monitoring: The Role of Au Nanoparticles in WO<sub>3</sub>, SnO<sub>2</sub>, and NiO-based Gas Sensors

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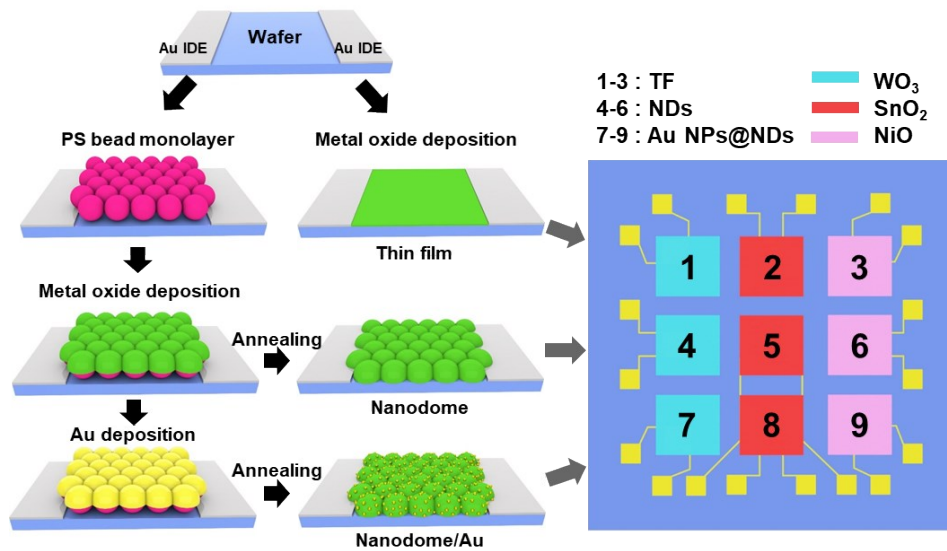


Fig. S1 Fabrication process of 3x3 gas sensor array

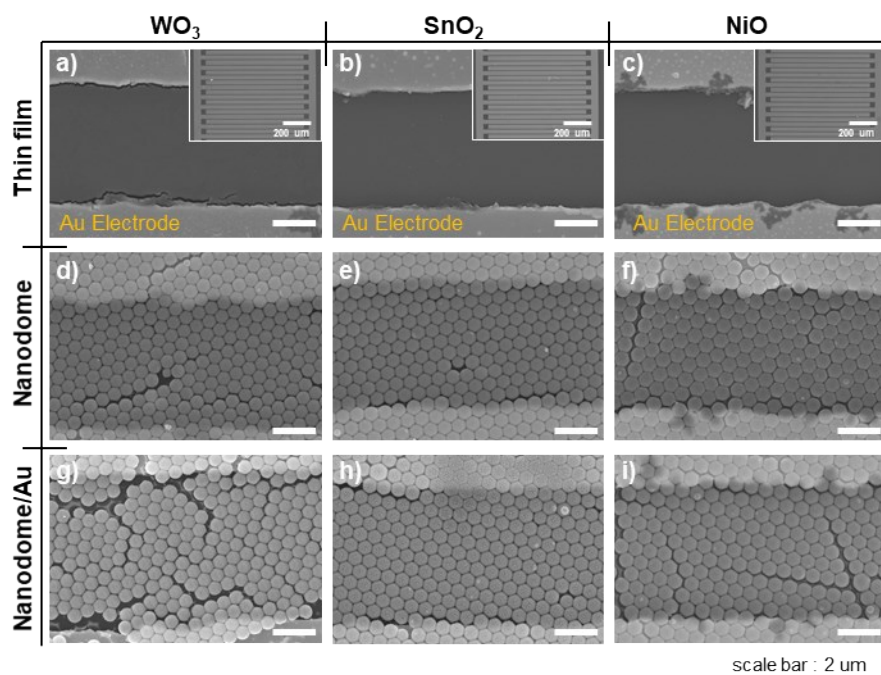
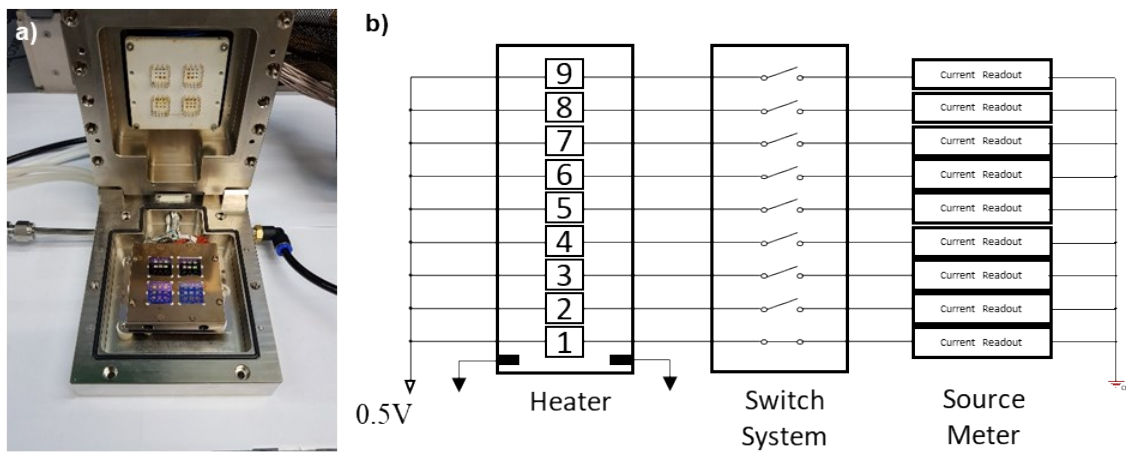
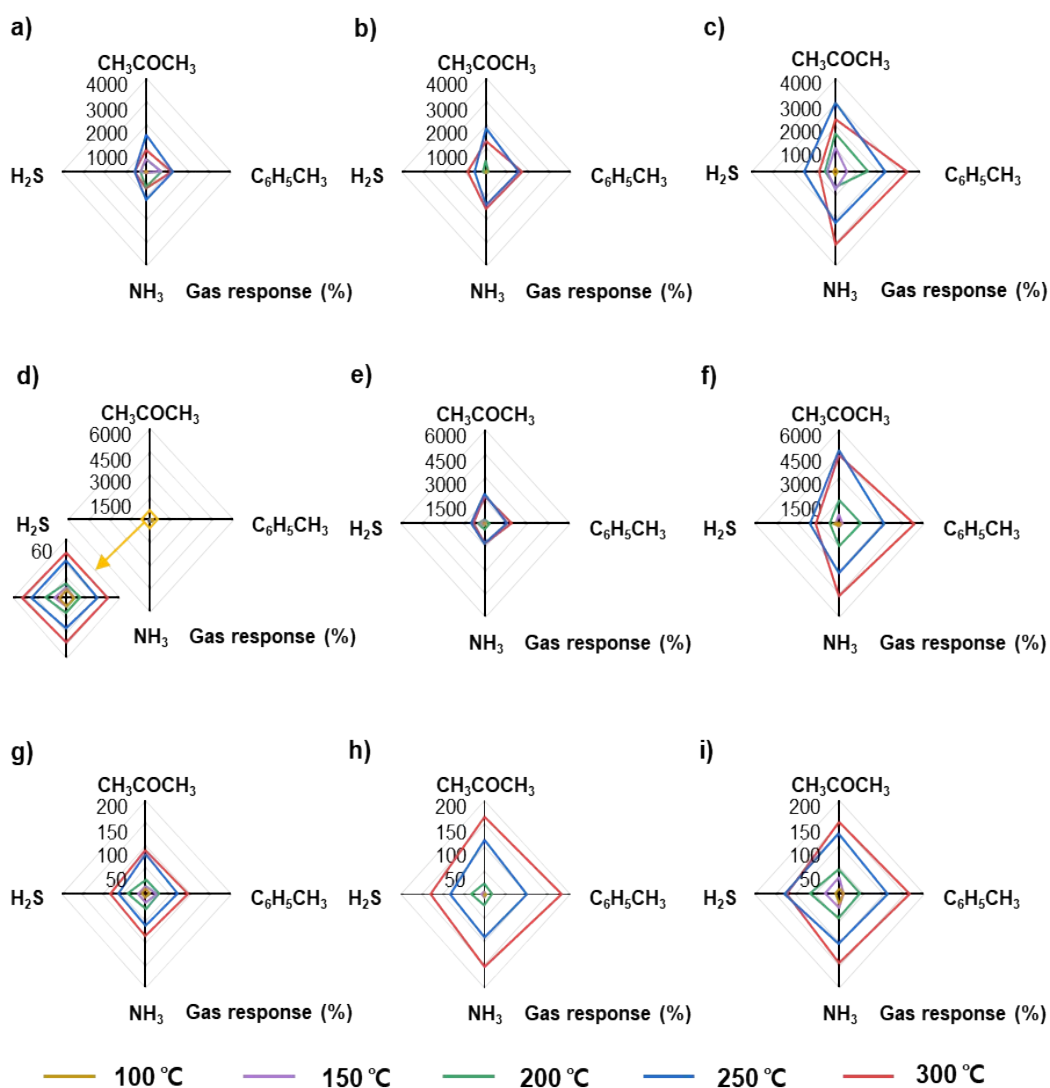


Fig. S2 SEM micrographs of the a-c) TF, d-f) NDs, g-i) Au NPs@ $\text{WO}_3$ ,  $\text{SnO}_2$  and  $\text{NiO}$  NDs. The inset image show the whole IDE substrate.



**Fig. S3** a) Digital image of the gas response measurement setup and the b) interface circuitry for the gas sensing measurements of the 3x3 sensor array.



**Fig. S4** Polar plot of gas responses for a-c)  $\text{WO}_3$ , d-f)  $\text{SnO}_2$ , g-i)  $\text{NiO}$  TFs, NDs, and Au NPs@NDs as a function of temperature.



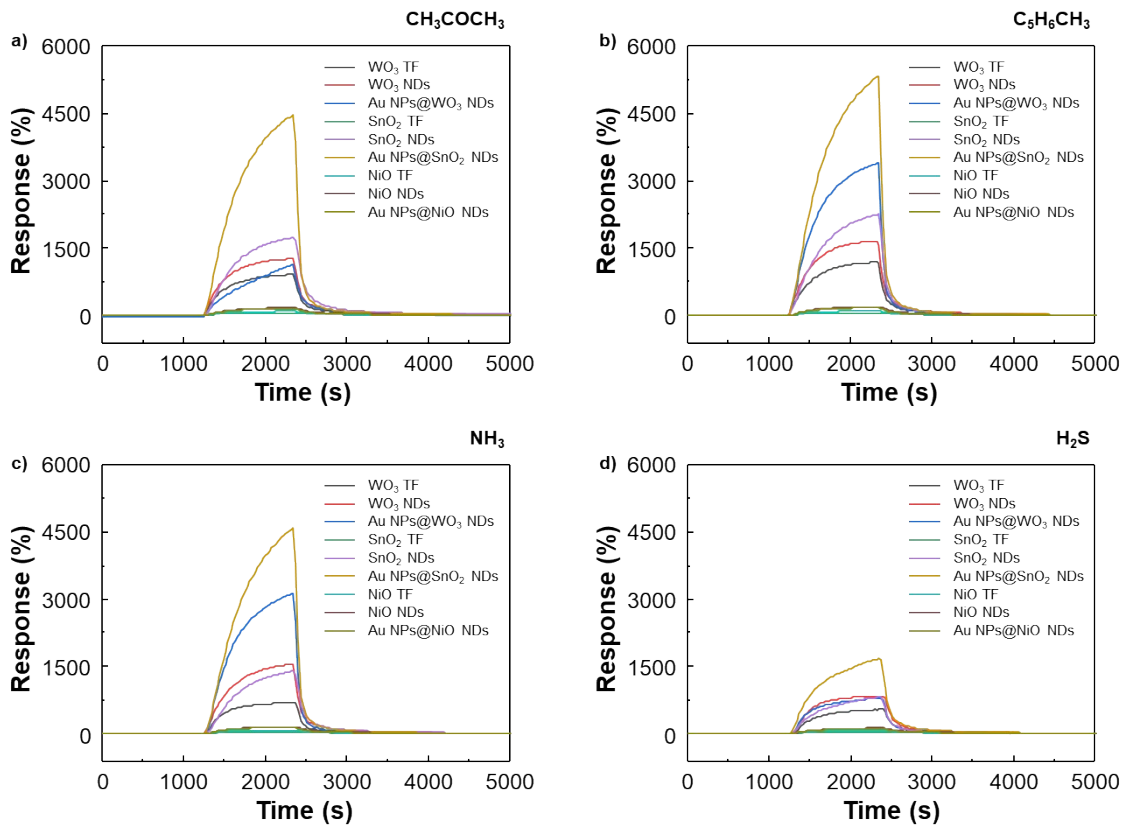


Fig. S5 Real-time response of the sensor array to 10 ppm a)  $\text{CH}_3\text{COCH}_3$ , b)  $\text{C}_6\text{H}_5\text{CH}_3$ , c)  $\text{NH}_3$  and d)  $\text{H}_2\text{S}$  at 300 °C.

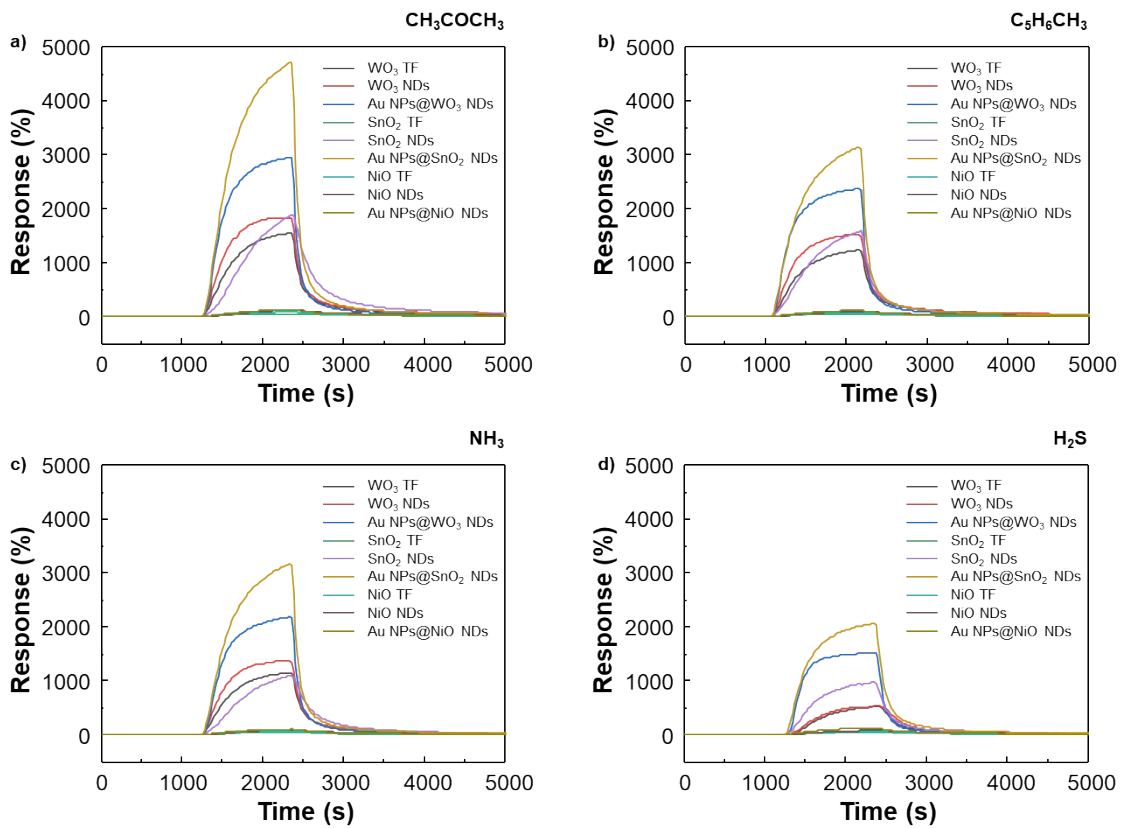


Fig. S6 Real-time response of the sensor array to 10 ppm a)  $\text{CH}_3\text{COCH}_3$ , b)  $\text{C}_6\text{H}_5\text{CH}_3$ , c)  $\text{NH}_3$  and d)  $\text{H}_2\text{S}$  at 250 °C.

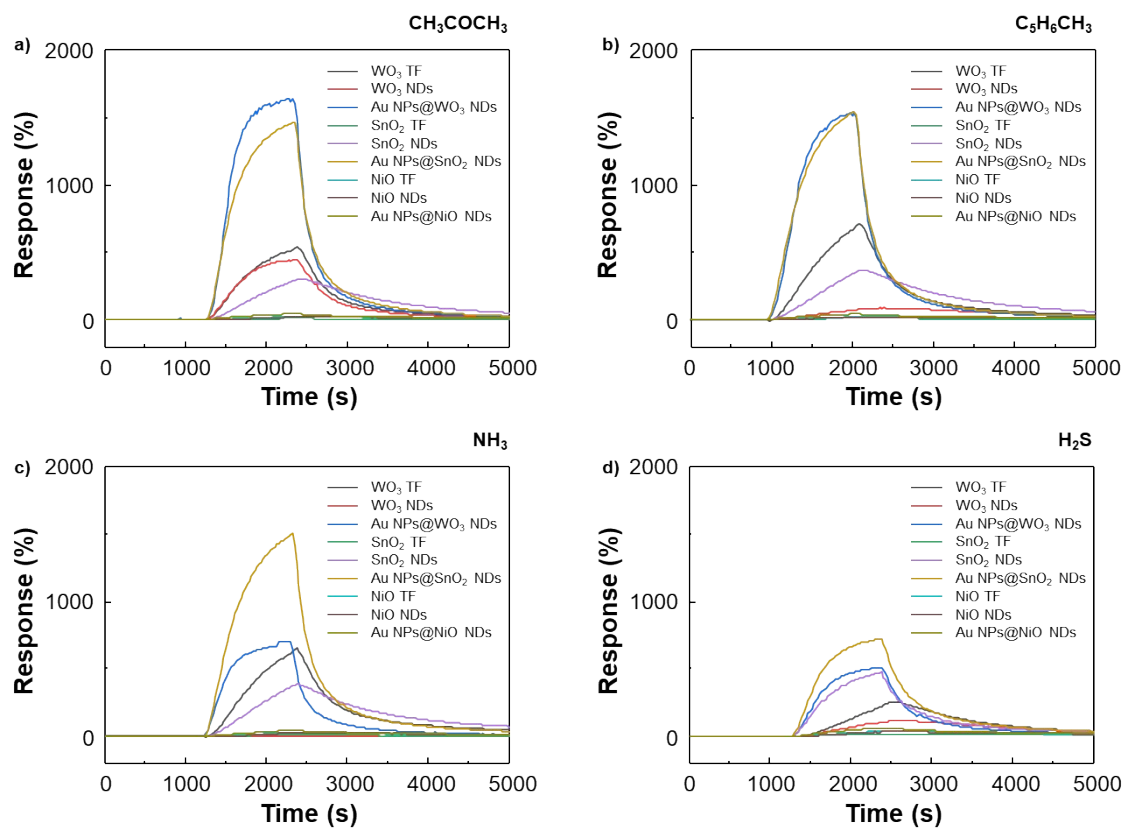


Fig. S7 Real-time response of the sensor array to 10 ppm a)  $\text{CH}_3\text{COCH}_3$ , b)  $\text{C}_6\text{H}_5\text{CH}_3$ , c)  $\text{NH}_3$  and d)  $\text{H}_2\text{S}$  at 200 °C.

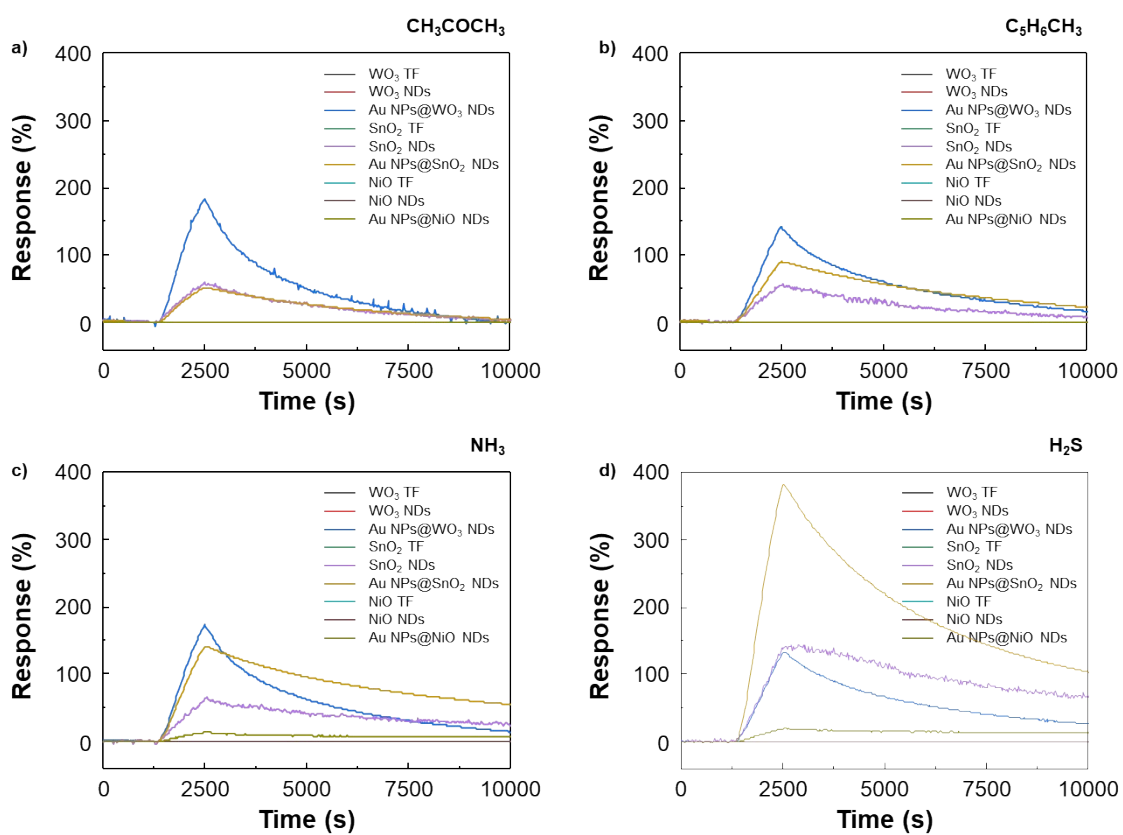


Fig. S8 Real-time response of the sensor array to 10 ppm a)  $\text{CH}_3\text{COCH}_3$ , b)  $\text{C}_6\text{H}_5\text{CH}_3$ , c)  $\text{NH}_3$  and d)  $\text{H}_2\text{S}$  at 150 °C.

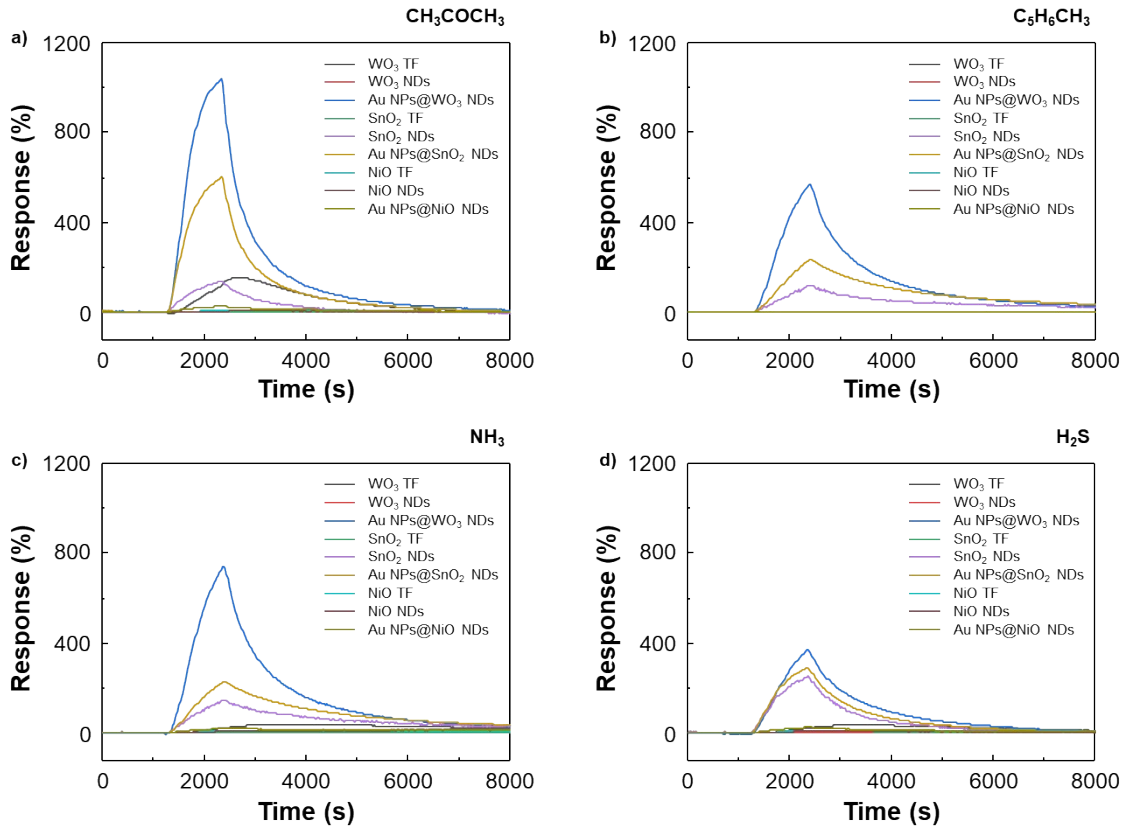


Fig. S9 Real-time response of the sensor array to 10 ppm a)  $\text{CH}_3\text{COCH}_3$ , b)  $\text{C}_6\text{H}_5\text{CH}_3$ , c)  $\text{NH}_3$  and d)  $\text{H}_2\text{S}$  at 100 °C

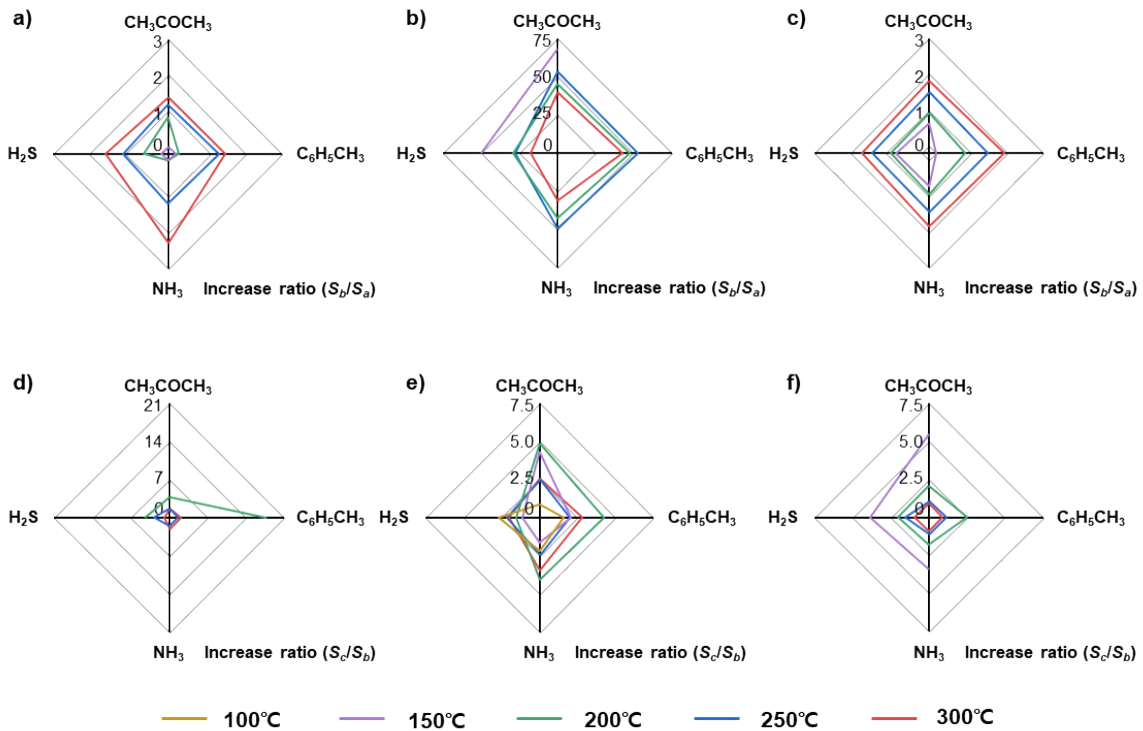
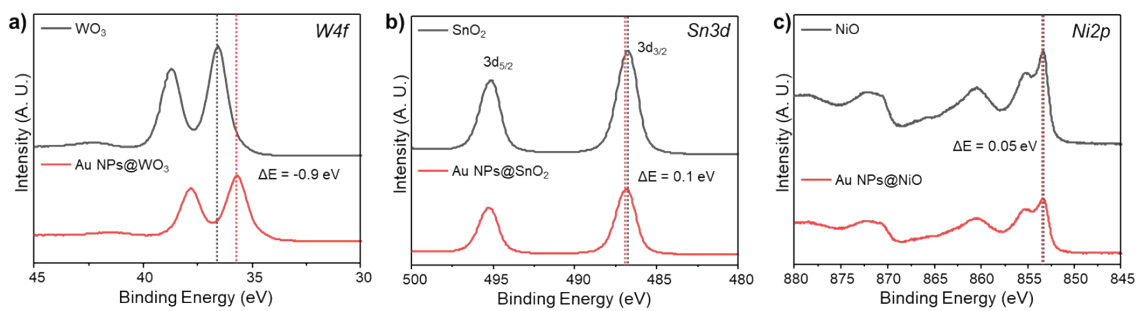


Fig. S10 Polar plots of the increase ratio ( $S_b/S_a$ ) between the response of TFs and the NDs of a)  $\text{WO}_3$ , b)  $\text{SnO}_2$ , and c)  $\text{NiO}$ . Polar plots of the increase ratio ( $S_c/S_b$ ) between the response of NDs and the Au NPs@NDs of d)  $\text{WO}_3$ , and e)  $\text{SnO}_2$ , f)  $\text{NiO}$ .  $S_a$ ,  $S_b$  and  $S_c$  represent the responses of the TFs and NDs and Au NPs@NDs.



**Fig. S11** XPS spectra for a) W4f of  $\text{WO}_3$  and  $\text{Au NPs@WO}_3$ , b) Sn3d of  $\text{SnO}_2$  and  $\text{Au NPs@SnO}_2$  c) Ni2p of  $\text{NiO}$  and  $\text{Au NPs@NiO}$ .



Gas	Metal oxide	Type	Resistance ( $\Omega$ )				
			Temperature ( $^{\circ}\text{C}$ )				
			100	150	200	250	300
$\text{CH}_3\text{COCH}_3$	$\text{WO}_3$	TF	-	$2.62 \times 10^7$	$2.05 \times 10^7$	$1.64 \times 10^7$	$4.60 \times 10^6$
		NDs	-	-	$9.07 \times 10^7$	$7.92 \times 10^7$	$2.48 \times 10^7$
		Au NPs@NDs	$5.86 \times 10^7$	$6.23 \times 10^7$	$5.29 \times 10^7$	$4.40 \times 10^7$	$1.04 \times 10^7$
	$\text{SnO}_2$	TF	-	$3.49 \times 10^2$	$3.71 \times 10^2$	$4.76 \times 10^2$	$4.70 \times 10^2$
		NDs	$2.45 \times 10^6$	$3.18 \times 10^6$	$1.69 \times 10^4$	$6.90 \times 10^4$	$5.10 \times 10^4$
		Au NPs@NDs	$1.62 \times 10^3$	$2.33 \times 10^4$	$7.42 \times 10^4$	$2.19 \times 10^5$	$1.96 \times 10^5$
	$\text{NiO}$	TF	-	$6.70 \times 10^5$	$9.43 \times 10^4$	$2.00 \times 10^4$	$7.39 \times 10^3$
		NDs	-	$2.94 \times 10^7$	$4.57 \times 10^6$	$6.33 \times 10^5$	$1.63 \times 10^5$
		Au NPs@NDs	-	$1.20 \times 10^7$	$1.36 \times 10^6$	$2.15 \times 10^5$	$6.11 \times 10^4$
$\text{C}_6\text{H}_5\text{CH}_3$	$\text{WO}_3$	TF	-	-	$4.25 \times 10^7$	$1.28 \times 10^7$	$5.26 \times 10^6$
		NDs	-	-	$4.09 \times 10^6$	$6.48 \times 10^7$	$2.73 \times 10^7$
		Au NPs@NDs	$8.05 \times 10^7$	$6.16 \times 10^7$	$5.38 \times 10^7$	$3.20 \times 10^7$	$2.47 \times 10^7$
	$\text{SnO}_2$	TF	-	-	$3.88 \times 10^2$	$4.51 \times 10^2$	$4.66 \times 10^2$
		NDs	$5.49 \times 10^6$	$8.20 \times 10^6$	$3.68 \times 10^4$	$2.33 \times 10^5$	$2.98 \times 10^5$
		Au NPs@NDs	$3.06 \times 10^3$	$5.05 \times 10^3$	$8.42 \times 10^4$	$1.41 \times 10^5$	$2.26 \times 10^5$
	$\text{NiO}$	TF	-	$6.42 \times 10^6$	$8.26 \times 10^4$	$2.34 \times 10^4$	$7.73 \times 10^3$
		NDs	-	-	$3.78 \times 10^6$	$7.69 \times 10^5$	$1.69 \times 10^5$
		Au NPs@NDs	$8.29 \times 10^7$	-	$1.21 \times 10^6$	$2.46 \times 10^5$	$6.42 \times 10^4$
$\text{NH}_3$	$\text{WO}_3$	TF	-	$7.50 \times 10^6$	$8.82 \times 10^6$	$5.00 \times 10^6$	$2.15 \times 10^6$
		NDs	-	-	$3.39 \times 10^6$	$2.56 \times 10^6$	$1.42 \times 10^6$
		Au NPs@NDs	$2.15 \times 10^7$	$2.15 \times 10^7$	$8.88 \times 10^6$	$4.56 \times 10^6$	$2.03 \times 10^6$
	$\text{SnO}_2$	TF	-	-	$3.86 \times 10^2$	$4.36 \times 10^2$	$4.72 \times 10^2$
		NDs	$3.87 \times 10^6$	$5.79 \times 10^6$	$2.41 \times 10^4$	$5.02 \times 10^4$	$6.15 \times 10^4$
		Au NPs@NDs	$3.06 \times 10^3$	$4.44 \times 10^3$	$7.30 \times 10^4$	$1.75 \times 10^5$	$2.59 \times 10^5$
	$\text{NiO}$	TF	-	$3.88 \times 10^5$	$1.06 \times 10^5$	$2.32 \times 10^4$	$7.69 \times 10^3$
		NDs	-	$1.62 \times 10^7$	$8.93 \times 10^6$	$7.89 \times 10^5$	$1.65 \times 10^5$
		Au NPs@NDs	$8.08 \times 10^7$	$5.39 \times 10^6$	$1.41 \times 10^6$	$2.55 \times 10^5$	$6.48 \times 10^4$
$\text{H}_2\text{S}$	$\text{WO}_3$	TF	-	$4.76 \times 10^7$	$3.57 \times 10^7$	$1.96 \times 10^7$	$8.54 \times 10^6$
		NDs	-	-	$1.20 \times 10^7$	$9.53 \times 10^6$	$6.23 \times 10^6$
		Au NPs@NDs	$4.29 \times 10^7$	$3.43 \times 10^7$	$1.35 \times 10^7$	$6.67 \times 10^6$	$3.05 \times 10^6$
	$\text{SnO}_2$	TF	-	$4.14 \times 10^2$	$4.48 \times 10^2$	$5.27 \times 10^2$	$5.20 \times 10^2$
		NDs	$1.40 \times 10^7$	$8.91 \times 10^6$	$1.03 \times 10^7$	$1.53 \times 10^7$	$9.26 \times 10^6$
		Au NPs@NDs	$9.60 \times 10^3$	$5.74 \times 10^3$	$9.38 \times 10^3$	$1.88 \times 10^4$	$1.51 \times 10^4$
	$\text{NiO}$	TF	-	$5.18 \times 10^5$	$8.51 \times 10^4$	$2.02 \times 10^4$	$8.32 \times 10^3$
		NDs	-	$1.97 \times 10^7$	$3.00 \times 10^6$	$5.48 \times 10^5$	$1.67 \times 10^5$
		Au NPs@NDs	$3.13 \times 10^7$	$6.61 \times 10^6$	$9.60 \times 10^5$	$1.94 \times 10^5$	$6.57 \times 10^4$

Table S1 Resistance of each sensors with operating temperatruue.

$\Delta (E_{TF} - E_{NDs})$ (kJ/mol)	WO <sub>3</sub>	SnO <sub>2</sub>	NiO
CH <sub>3</sub> COCH <sub>3</sub>	-15.9	4.67	-14.8
C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>	-46.3	0.902	-28.9
NH <sub>3</sub>	-14.2	-1.92	-13.2
H <sub>2</sub> S	10.0	6.22	-12.1

**Table S2** Difference in the activation energy between TF and NDs ( $E_{TF} - E_{NDs}$ ) of each metal oxide and gas combination.