Supporting information

Modelling of the heterogenous ion transport in conducting polymer supercapacitors

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Fig. S1: Attempts to reproduce experimental (Exp.) CV results at 10 mV/s. (a) For the homogeneous model the effective cation and anion diffusivities within the electrode are varied. (b) Using the heterogeneous model; by reducing effective cation and anion diffusivity in ~27 % of electrode domain (right portion, see inset) by factor , while keeping 1.3 E-09 and 0.9 E-09 as cation and anion diffusivity respectively in the remaining (~63 %) of the electrode domain.



Fig. S2: Evolution of cation and anion concentration (in molar) during self-discharge tests simulated by the heterogeneous model including ORR. 0 s: Charging starts at 100 mA/g. 194 s: Charging cut-off potential is reached, switch to open circuit potential measurement. 500 s: halfway through ion redistribution. 1000 s: majority of ion redistribution has occurred. 5000 s end of test. Positive electrode on top, negative below.



Fig. S3: Transition of spatial hole concentration along both electrode lateral centre lines (as shown inset - in molar), at critical times during self-discharge tests produced by the heterogeneous model including ORR. 0 s: Charging starts at 100 mA/g. 194 s: Charging cut-off potential is reached, switch to open circuit potential measurement. 500 s: mid-way of ion redistribution. 1000 s: after majority of ion redistribution. 5000 s end of test. Positive electrode on top, negative below.



Fig. S4: Comparison of real model results with or without (w. or w/o.) Equation 8 (Eq. 8): (a) Steady-state cyclic voltammograms at lower scan rates. (b) Nyquist plot from EIS simulation. (b) Bode plots from EIS simulation. EIS is simulated at 0 V with 10 mV perturbation.



Fig. S5: Nyquist plot of experimental CPSC data and the fitted model. The model fits the data well within the 100 Hz to 50 mHz frequency range. The experimental data is from a different batch of CPSCs than the data used in the rest of the manuscript. A contact resistance was added to the model to account for the contact resistance between the current collector and the CP electrodes.



Fig. S6: Experimental charge-discharge at 700 mA/g with varying cut-off potential from 0.6 to 2 V in 0.2 V steps.

Description	Electrode Domains	Separator Domain
D _h + (cm²s⁻¹) ¹	$0.205 \left(0.93 - \frac{1.1}{1 + exp^{\text{ind}} \left(\frac{(h^+ - 71)}{23} \right)} \right)$	-
$\frac{D_{c^+}}{c^-}$ (cm ² s ⁻	1.3E-9 / 0.9E-9 high 6.5E-14 / 4.5E-14 low	1.0E-9 / 0.7E-9++
$v_{f,i}$	0.23 for ions 0.67 for holes	0.85*
Е	28 for ions ² 40 for holes ¹	28
^C _v (mol/(m ³ V))	291 ³	-
^C fix (M)	0.18	-
^k ₀ (s ⁻¹)	1.6E-5	

Table S1: Nominal operating parameters and variables for the model.

*From manufacturer manual. – Not applicable. ++ Bruggeman porosity correction of values from Mao. et. al. 4.

REFERENCES

- 1 K. Tybrandt, I. V. Zozoulenko and M. Berggren, *Sci. Adv.*, 2017, **3**, eaao3659.
- 2 H. Tokuda, S. Tsuzuki, Md. A. B. H. Susan, K. Hayamizu and M. Watanabe, *J. Phys. Chem. B*, 2006, **110**, 19593–19600.
- 3 J. Edberg, O. Inganäs, I. Engquist and M. Berggren, J. Mater. Chem. A, 2017, 6, 145–152.
- 4 Y. Mao and K. Damodaran, Chem. Phys., 2014, 440, 87–93.