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Electronic Supplementary Material for:

## FeS<sub>2</sub>-anchored transition metal single atoms for high-efficient overall water splitting: a DFT computational screening study

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Content includes: one Table and six Figures.

**Table S1** The binding energy  $(E_b)$ , cohesive energy  $(E_c)$ , Mulliken charge (Q), magnetic moment (M), average bond length  $(d_{M-S})$  and d-band center  $(\varepsilon_d)$  of TM atom adsorption on FeS<sub>2</sub>(100) surface. **Fig. S1** Schematic illustration of the structure model of FeS<sub>2</sub> surface. (a) Top view. (b) Side view. a, b, c, and d denote hollow, S<sub>4</sub>, S<sub>3</sub>, and Fe sites, respectively. S<sub>3</sub> and S<sub>4</sub> denote 3-fold and 4-fold coordinated sulfur atoms, respectively.

Fig. S2 The most stable structure of TM single atoms adsorption on the  $FeS_2(100)$  surface.

Fig. S3 Band-edge positions and bandgaps of M@FeS<sub>2</sub> catalyst.

Fig. S4 Partial density of states (PDOS) of M@FeS2 catalysts: (a) Sc@FeS2, (b) Ti@FeS2, (c)

V@FeS<sub>2</sub>, (d) Cr@FeS<sub>2</sub>, (e) Mn@FeS<sub>2</sub>, (f) Co@FeS<sub>2</sub>, (g) Ni@FeS<sub>2</sub>, (h) Cu@FeS<sub>2</sub>, and (i) Zn@FeS<sub>2</sub>. The dashed lines denote the Fermi level.

**Fig. S5** Potential energy evolution of M@FeS<sub>2</sub> catalysts at 500 K in the water-phase environment. (a) Sc@FeS<sub>2</sub>, (b) Ti@FeS<sub>2</sub>, (c) V@FeS<sub>2</sub>, (d) Cr@FeS<sub>2</sub>, (e) Mn@FeS<sub>2</sub>, (f) Fe@FeS<sub>2</sub>, (g) Co@FeS<sub>2</sub>, (h) Ni@FeS<sub>2</sub>, (i) Cu@FeS<sub>2</sub>, and (j) Zn@FeS<sub>2</sub>.

Fig. S6 Gibbs free energy diagram of the HER over S atom of M@FeS<sub>2</sub> catalysts.

TM atoms	$E_{\rm b}({\rm eV})$	$E_{\rm c}({\rm eV})$	Q (e)	$M\left(\mu_{\mathrm{B}} ight)$	$d_{\text{M-S}}$ (Å)	$\varepsilon_{\rm d}({\rm eV})$
Sc	-4.93	-4.16	0.71	-0.016	2.504	-2.13
Ti	-5.54	-5.91	0.35	0.000	2.275	-2.13
V	-5.79	-7.11	0.09	0.000	2.165	-1.77
Cr	-5.57	-7.44	-0.11	2.976	2.107	-1.47
Mn	-5.15	-6.77	-0.22	0.000	2.085	-1.64
Fe	-4.61	-6.30	0.32	1.701	2.072	-1.59
Co	-4.22	-5.68	-0.32	0.000	2.084	-1.88
Ni	-3.34	-4.68	-0.03	-0.026	2.141	-1.72
Cu	-1.96	-3.53	0.10	0.002	2.327	-2.36
Zn	-0.78	-0.46	0.31	0.001	2.505	-6.21

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**Fig. S4** Partial density of states (PDOS) of M@FeS<sub>2</sub> catalysts: (a) Sc@FeS<sub>2</sub>, (b) Ti@FeS<sub>2</sub>, (c) V@FeS<sub>2</sub>, (d) Cr@FeS<sub>2</sub>, (e) Mn@FeS<sub>2</sub>, (f) Co@FeS<sub>2</sub>, (g) Ni@FeS<sub>2</sub>, (h) Cu@FeS<sub>2</sub>, and (i) Zn@FeS<sub>2</sub>. The dashed lines denote the Fermi level.



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