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## **Supporting information**

## Oxygen-induced defect-healing and photo-brightening of halide perovskite semiconductors: science and application

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**Fig. S1** (A) Oxygen passivation of silicon NCs: (a) Possible surface passivation process during controlled oxidation and (b) Dependence of field effect carrier mobility on Si NC oxidation time. Reprinted with permission from Ref.[1]. Copyright 2014 Elsevier Ltd. (B) Oxygen passivation of Sb<sub>2</sub>Se<sub>3</sub> solar cell: (a) *J-V* curves of the CdS/Sb<sub>2</sub>Se<sub>3</sub> solar cells grown with oxygen partial pressures varying from 0 to  $5.2 \times 10^{-3}$  Pa and (b) Sketchy band diagram between Sb<sub>2</sub>Se<sub>3</sub> and CdS without and with the addition of oxygen in the whole Sb<sub>2</sub>Se<sub>3</sub> films deposition. Reprinted with permission from Ref.[2]. Copyright 2015 Wiley-VCH. (C) PL enhancement of monolayer MoS<sub>2</sub> before and after treatment with H<sub>2</sub>O<sub>2</sub>. Reprinted with permission from Ref.[3]. Copyright 2017 The Royal Society of Chemistry. (D) Oxygen bonding induced strong PL enhancement of MoS<sub>2</sub>: charge density difference of O<sub>2</sub> molecule physisorbed on perfect monolayer MoS<sub>2</sub> (a) and chemisorbed on defective monolayer MoS<sub>2</sub> after oxygen plasma irradiation with different durations (c). Reprinted with permission from Ref.[4]. Copyright 2014 American Chemical Society.



**Fig. S2** Schematic picture of the dynamic (MA orientation) and static (Cl doping) healing mechanisms. The crossover from spatially localized to delocalized excited states (ES) in defective perovskites is reflected by the extension of charge densities in

the vicinity of the vacancies (shown in magenta color). The electron (hole) density is shown by red (yellow) circles on lead (iodine) ions. Reprinted with permission from Ref.[5]. Copyright 2018 Wiley-VCH.



**Fig. S3** (A) UPS spectra of a MAPbI<sub>3-x</sub>Cl<sub>x</sub> perovskite film before (black) and after (red) oxygen exposure to 50 mbar. (a) SECO and same valence band spectra displayed on (b) a linear and (c) a logarithmic intensity scale for valence band onset determination. Reprinted with permission from Ref.[6]. Copyright 2018 Wiley-VCH. (B) Electrochemical process of MAPbI<sub>3-x</sub>Cl<sub>x</sub> under control environment (a) and the evolution of resistance (b, c) and capacitance values (d, e) of MAPbI<sub>3-x</sub>Cl<sub>x</sub> film under different atmosphere. Reprinted with permission from Ref.[7]. Copyright 2016 American Chemical Society.



**Fig. S4** Microscale PL properties in dry and humid air. (a) Confocal PL map of a MAPbI<sub>3</sub> film in dry air normalized to the maximum intensity. (b, c) Monitoring the emission (PL count rate) over time under illumination from: (b) a bright grain (blue

circle in (a)) and (c) dark grain (pink circle in (a)) under dry air and under humidified ( $\approx$ 45% relative humidity) air. The PL intensity for each trace over time is given relative to the starting value for the bright grain in air, which is normalized to 1. (d, e) PL decays from the same (d) bright and (e) dark grains under dry air before and after the light soaking. (f, g) PL decays from the same bright (f) and dark (g) grains under dry air, humidified air, and after the light soaking in humidified air. (B) Surface atomic and electronic structures. Schematic representation of the local atomic-scale configurations of the surface termination layer (top row) and calculated band structures (bottom row) for the (110) surface of MAPbI<sub>3</sub>. (a) Pristine uncharged surface, (b) negatively charged iodine vacancy into which the following molecules are adsorbed; (c) N<sub>2</sub>, (d) H<sub>2</sub>O, and (e) O<sub>2</sub>. Reprinted with permission from Ref.[8]. Copyright 2018 Wiley-VCH.



**Fig. S5** (A) Simulated optimized geometry at 0 K (top panel) and a representative snapshot at 300 K of (a) pristine CsPbBr<sub>3</sub> (001) surface, (b) Pb-dimer, and (c) O<sub>2</sub> passivated Pb-dimer, respectively. (B) Charge densities obtained from a representative snapshot at 300 K for the key orbitals participating in charge trapping and recombination of (a) pristine CsPbBr<sub>3</sub> (001), (b) Pb-dimer, and (c) O<sub>2</sub> passivated systems. Reprinted with permission from Ref.[9]. Copyright 2019 American Chemical Society. (C) Projected density of states PDOS of (a) the pristine MAPbI<sub>3</sub> (001) surface, (b) Iv, (c) O<sub>2</sub>-doped I<sub>v</sub>, and (d) O-doped I<sub>v</sub> systems. The Fermi level is set to zero. The superoxide level contributes to the system HOMO. (D) PDOS of the MAPbI<sub>3</sub> (001) system containing the I<sub>v</sub> vacancy doped with charged oxygen species: (a) O<sup>1-</sup> anion, (b) O<sup>2-</sup> anion, (c) O<sub>2</sub><sup>1-</sup> anion, and (d) O<sub>2</sub><sup>2-</sup> anion. The Fermi level is set to zero. The corresponding nonradiative electron-hole recombination dynamics. Reprinted with permission from Ref.[10]. Copyright 2019 American Society.



**Fig. S6** TRPL for MAI-treated MAPbI<sub>3</sub> films without quencher layer (spiro-MeOTAD) (A) and with spiro-MeOTAD (B). Schematic of the VB alignment between spiro-MeOTAD and MAPbI<sub>3</sub> with and without MAI treatment (C). Reprinted with permission from Ref.[11]. Copyright 2019 American Chemical Society.



**Fig. S7** A MAPbBr<sub>3</sub> SC with gold electrodes (a) and its photocurrents (b) measured under cyclical vacuum and atmosphere. Reprinted with permission from Ref.[12]. Copyright 2017 The Author(s). (B) Side-view (a) and structure (b) of a MAPbBr<sub>3</sub> SC device. *I-V* curves of the MAPbBr<sub>3</sub> SC device under dark (c) and under laser illumination (d) in air and vacuum. Variation in PL intensity of MAPbBr<sub>3</sub> SCs from air-vacuum-air environments (e). PL intensity variation in different gases (f).

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