

Electronic Supplementary information

Low-cost and eco-friendly Br-doped $\text{Cu}_7\text{Sn}_3\text{S}_{10}$ thermoelectric compound with zT around unit

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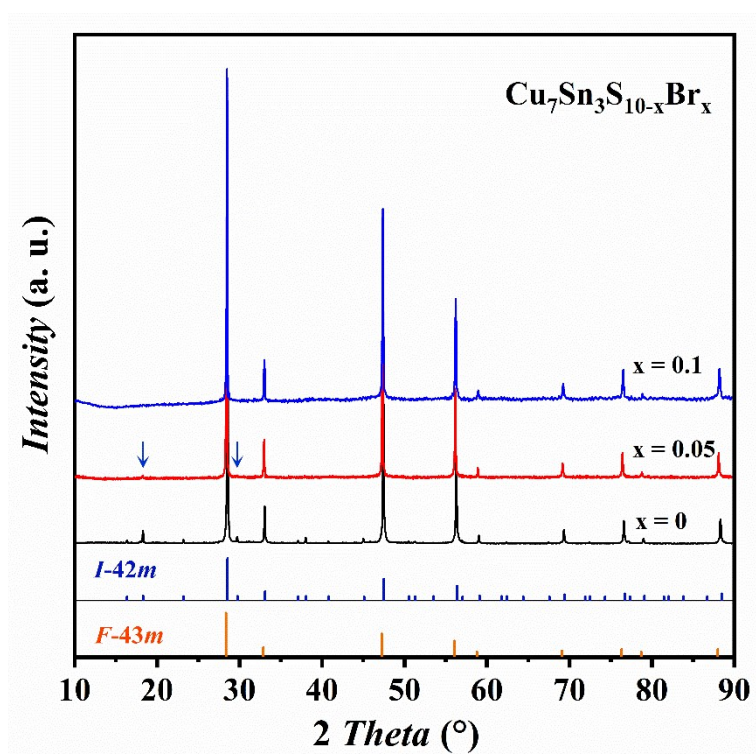


Fig. S1 Powder X-ray diffraction pattern (PXRD) of $\text{Cu}_7\text{Sn}_3\text{S}_{10-x}\text{Br}_x$ ($x=0, 0.05$, and 0.1) samples. The blue arrows represent the positions of the diffraction peaks for tetragonal phase.

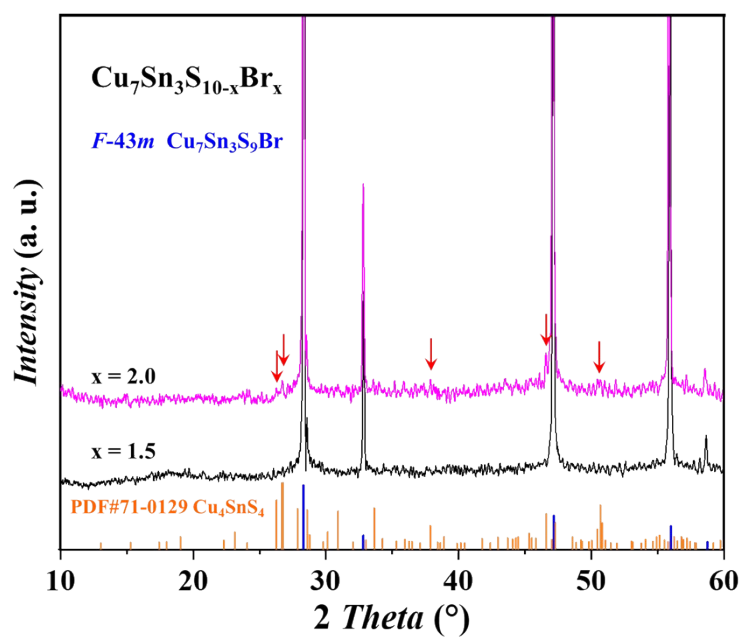


Fig. S2 Magnified powder X-ray diffraction pattern (PXR) of $\text{Cu}_7\text{Sn}_3\text{S}_{10-x}\text{Br}_x$ ($x=1.5$ and 2.0) samples in the low intensity range. The red arrows represent the positions of the diffraction peaks for Cu_4SnS_4 .

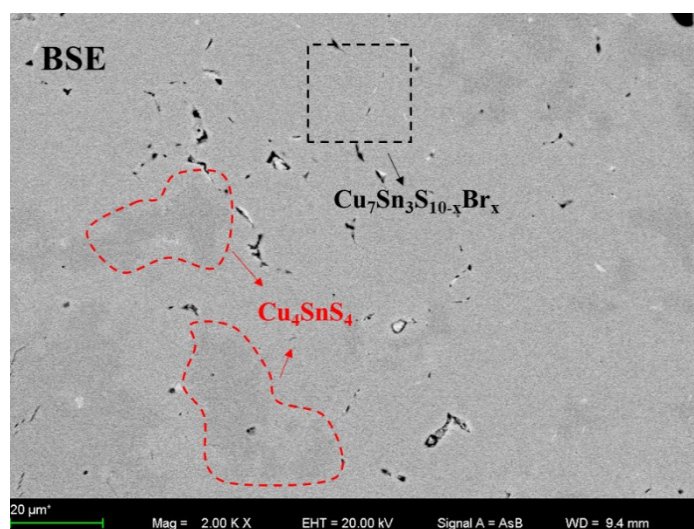


Fig. S3 Backscattered electron (BSE) image performed on $\text{Cu}_7\text{Sn}_3\text{S}_8\text{Br}_2$.

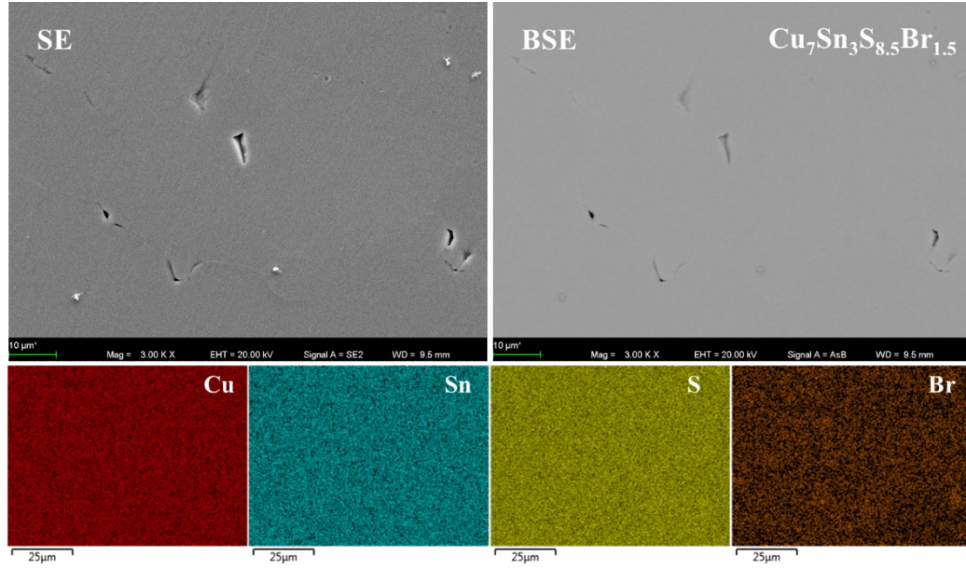


Fig. S4 Secondary electron (SE) image, Backscattered electron (BSE) image and Energy Dispersive X-ray Spectroscopy (EDS) mapping performed on $\text{Cu}_7\text{Sn}_3\text{S}_8\text{Br}_{1.5}$.

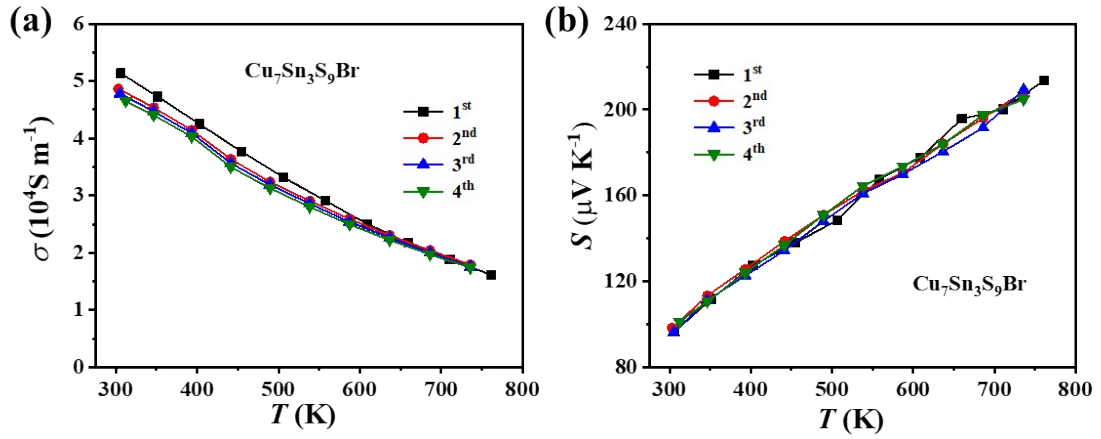


Fig. S5 Cycling test on electrical transport properties of $\text{Cu}_7\text{Sn}_3\text{S}_9\text{Br}$ sample.

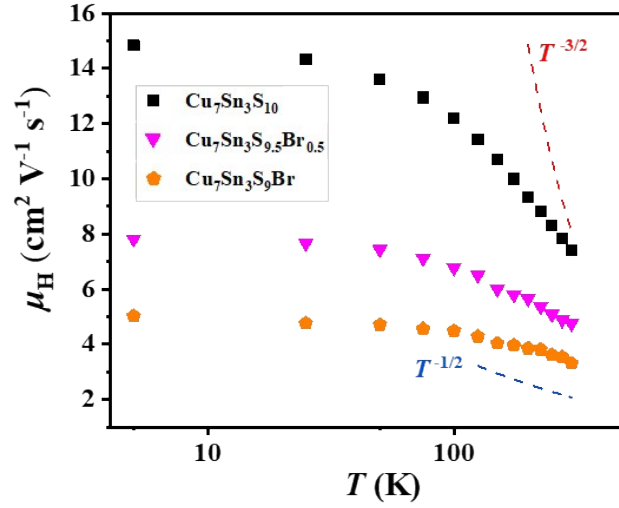


Fig. S6 Temperature dependence of Hall carrier mobility (μ_H) for $\text{Cu}_7\text{Sn}_3\text{S}_{10-x}\text{Br}_x$ ($x=0, 0.5$, and 1.0) samples from 5 K to 300 K. The dashed lines represent the acoustic phonon scattering ($\mu_H \propto T^{-1.5}$) and alloy scattering ($\mu_H \propto T^{-0.5}$), respectively.

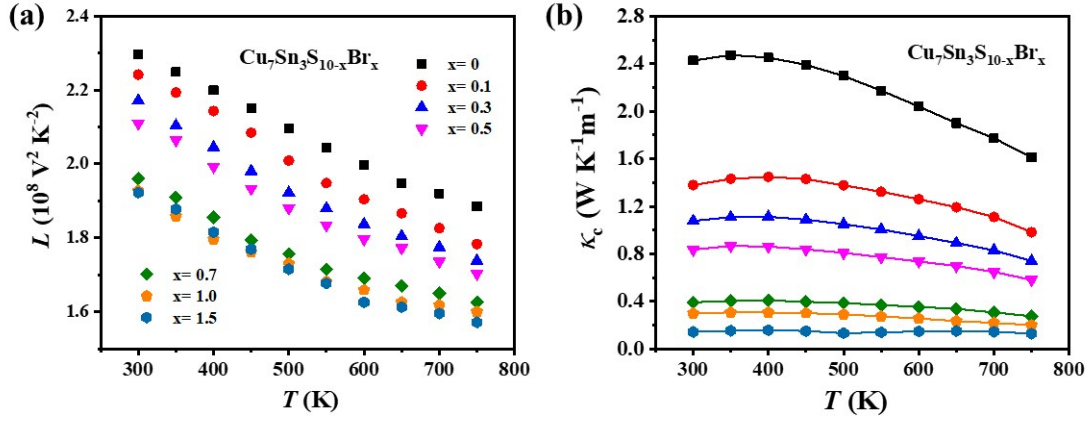


Fig. S7 Temperature dependences of (a) Lorenz number (L) and (b) carrier thermal conductivity (κ_c) for $\text{Cu}_7\text{Sn}_3\text{S}_{10-x}\text{Br}_x$ ($x=0, 0.1, 0.3, 0.5, 0.7, 1.0$, and 1.5).

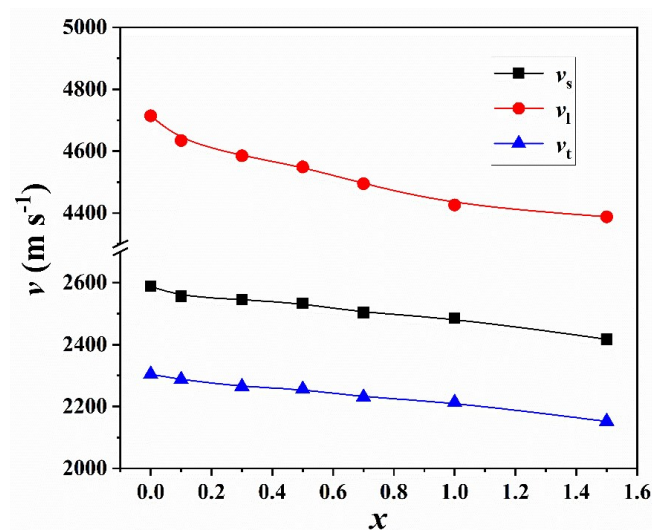


Fig. S8 Transverse sound velocity (v_t), longitudinal wave sound velocity (v_l), and average sound velocity (v_s) for Br-doped $\text{Cu}_7\text{Sn}_3\text{S}_{10}$ samples.

Table S1 Actual chemical compositions of the $\text{Cu}_7\text{Sn}_3\text{S}_{10-x}\text{Br}_x$ samples determined by EDS.

Nominal chemical compositions	Actual chemical compositions
$\text{Cu}_7\text{Sn}_3\text{S}_{10}$	$\text{Cu}_{6.51}\text{Sn}_3\text{S}_{9.89}$
$\text{Cu}_7\text{Sn}_3\text{S}_{9.9}\text{Br}_{0.1}$	$\text{Cu}_{6.62}\text{Sn}_3\text{S}_{9.42}\text{Br}_{0.11}$
$\text{Cu}_7\text{Sn}_3\text{S}_{9.7}\text{Br}_{0.3}$	$\text{Cu}_{6.68}\text{Sn}_3\text{S}_{9.42}\text{Br}_{0.28}$
$\text{Cu}_7\text{Sn}_3\text{S}_{9.5}\text{Br}_{0.5}$	$\text{Cu}_{6.72}\text{Sn}_3\text{S}_{9.42}\text{Br}_{0.38}$
$\text{Cu}_7\text{Sn}_3\text{S}_{9.3}\text{Br}_{0.7}$	$\text{Cu}_{6.59}\text{Sn}_3\text{S}_{9.39}\text{Br}_{0.44}$
$\text{Cu}_7\text{Sn}_3\text{S}_9\text{Br}$	$\text{Cu}_{6.93}\text{Sn}_3\text{S}_{9.22}\text{Br}_{0.46}$
$\text{Cu}_7\text{Sn}_3\text{S}_{8.5}\text{Br}_{1.5}$	$\text{Cu}_{6.49}\text{Sn}_3\text{S}_{9.30}\text{Br}_{0.83}$
$\text{Cu}_7\text{Sn}_3\text{S}_8\text{Br}_2$	$\text{Cu}_{6.75}\text{Sn}_3\text{S}_{8.96}\text{Br}_{0.63}$

Table S2 Atomic coordinates and isotropic displacement parameters for $\text{Cu}_7\text{Sn}_3\text{S}_9\text{Br}$.

Atom	Site	x/a	y/b	z/c	$B_{\text{eq}} [\text{\AA}^2]$	Occupancy
Cu1	4a	0	0	0	1.24(10)	0.6999 Cu+0.3 Sn
S1	4c	1/4	1/4	1/4	0.7(2)	0.8999 S+0.1 Br

Table S3. Crystallographic information for Cu₇Sn₃S₉Br.

Composition	Cu _{0.69} Sn _{0.3} S _{0.9} Br _{0.1}
Formula weight	116.30
Space group	<i>F3m</i> (No. 216)
<i>a</i> /Å	5.4409(2)
Unit cell volume /Å ³	161.06(2)
<i>F</i> (000) /e	212.8
<i>Z</i>	4.0
μ/cm^{-1}	611.73
Calculated density (g cm ⁻³)	4.8217(5)
Radiation, wavelength(Å)	CuK α , 1.54056
<i>T</i> (K)	295
Data range 2 θ (°)	5 – 100
No. of reflections	11
No. of refined structure parameters	7
Profile function	Pseudo-Voigt
Refinement mode	Full profile
<i>R</i> _i	0.0539
<i>R</i> _p	0.2932
<i>R</i> _{wp}	0.0857
Goodness of fit	13.140

Table S4 Selected Bond Lengths (Å) and Angles (°) for Cu₇Sn₃S₉Br.

Cu1-S1	2.4014(1)		
S1-Cu1-aS1	109.47(1)	Cu1-S1-aCu1	109.47(1)
S1-Cu1-bS1 [0000]	109.47(1)	Cu1-S1-bCu1 [0000]	109.47(1)
S1-Cu1-cS1 [0000]	109.47(1)	Cu1-S1-cCu1 [0000]	109.47(1)
aS1-Cu1-bS1 [0000]	109.47(1)	aCu1-S1-bCu1 [0000]	109.47(1)
aS1-Cu1-cS1	109.47(1)	aCu1-S1-cCu1	109.47(1)
bS1-Cu1-cS1	109.47(1)	bCu1-S1-cCu1	109.47(1)

Symmetry codes: +x, +y, +z; +z, +x, +y; +y, +z, +x; -y, +x, -z; -x, +z, -y; -z, +y, -x; +x, -y, -z;
+z, -x, -y; +y, -z, -x; -y, -x, +z; -x, -z, +y; -z, -y, +x; -x, +y, -z; -z, +x, -y; -y, +z, -x; +y, +x, +z;
+x, +z, +y; +z, +y, +x; -x, -y, +z; -z, -x, +y; -y -z, +x; +y, -x, -z; +x, -z, -y; +z, -y, -x;

Table S5 Density-of-state effective mass m_d^* for Cu₇Sn₃S_{10-x}Br_x and Cu₇Sn₃S_{10-x}Cl_x at

300 K and 750 K.

Compositions	m_d^*/m_e @300 K	m_d^*/m_e @750 K
Cu ₇ Sn ₃ S ₁₀	3.5	4.2
Cu ₇ Sn ₃ S _{9.9} Br _{0.1}	4.0	4.9
Cu ₇ Sn ₃ S _{9.7} Br _{0.3}	4.0	4.8
Cu ₇ Sn ₃ S _{9.5} Br _{0.5}	4.1	4.8
Cu ₇ Sn ₃ S _{9.7} Br _{0.3}	4.6	5.3
Cu ₇ Sn ₃ S _{9.0} Br _{1.0}	4.7	5.9
Cu ₇ Sn ₃ S _{8.5} Br _{1.5}	4.6	6.8
Cu ₇ Sn ₃ S _{9.7} Cl _{0.3}	4.3	5.4
Cu ₇ Sn ₃ S _{9.1} Cl _{0.9}	5.4	7.1

Table S6 Detailed parameters used for the Callaway model. Here y (= x -0.1) is the normalized Br-doping content in Cu₇Sn₃Br_{0.1}S_{9.9-y}Br_y.

y	0	0.2	0.4	0.6	0.9	1.4
z(=y/9.9)	0	0.0202	0.0404	0.06061	0.09091	0.14141
1-z	1	0.9798	0.9596	0.93939	0.90909	0.85859
M_S	32.065	32.065	32.065	32.065	32.065	32.065
M_{Br}	79.904	79.904	79.904	79.904	79.904	79.904
r_S	1	1	1	1	1	1
r_{Br}	1.15	1.15	1.15	1.15	1.15	1.15
Γ_{mass}	0	0.00695	0.01339	0.01934	0.02738	0.03863
ε	--	635	635	635	635	635
Γ_{strain}	--	0.04706	0.09545	0.14493	0.22062	0.34871
Γ	--	0.05401	0.10884	0.16426	0.24801	0.38734
v_s (m s ⁻¹)	2556	2545	2533	2503	2485	2417
Θ_D (K)	280.9	279.7	278.5	275.1	273.2	265.7
u (300 K)	--	1.17472	1.67188	2.06579	2.54788	3.22849
u (500 K)	--	0.80962	1.15227	1.42375	1.75601	2.22508
u (750 K)	--	0.61603	0.87674	1.08331	1.33612	1.69303

Table S7 Fitting parameters used in theoretical zT prediction based on SPB model

T (K)	m_d^*/m_e	μ_0 (cm ² V ⁻¹ s ⁻¹)	κ_L (W m ⁻¹ K ⁻¹)
300	4.0	17.0	1.4
	5.0	7.0	0.80
750	5.0	3.3	0.60
	6.0	1.5	0.40