

Supporting Information

Promoting sulphur conversion chemistry with Tri-modal porous N, O-codoped carbon for stable Li-S batteries

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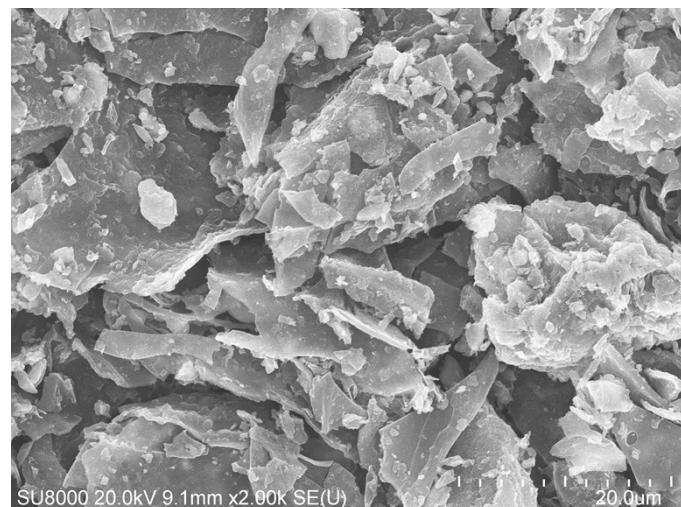


Figure S1. SEM image of the collected bagasse.

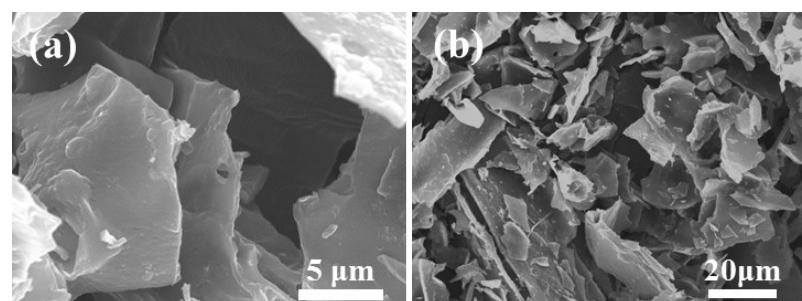


Figure S2. SEM images of pristine bagasse after pre-carbonization at 600 °C.

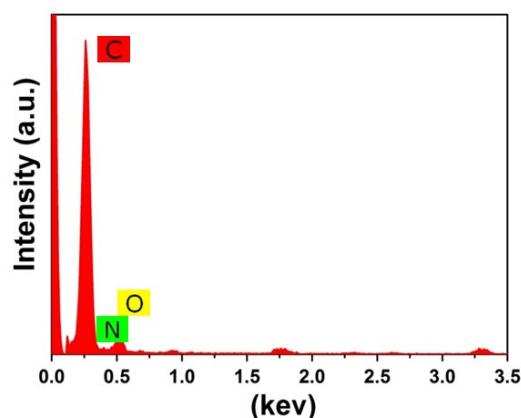


Figure S3. EDS spectrum of the TD-HDC sample.

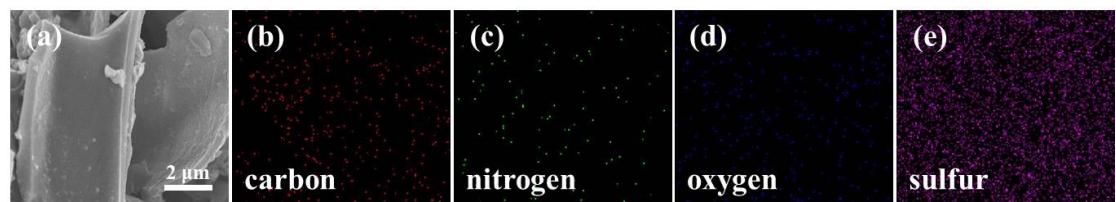


Figure S4. SEM and EDS mapping images of the S/TD-HDC sample.

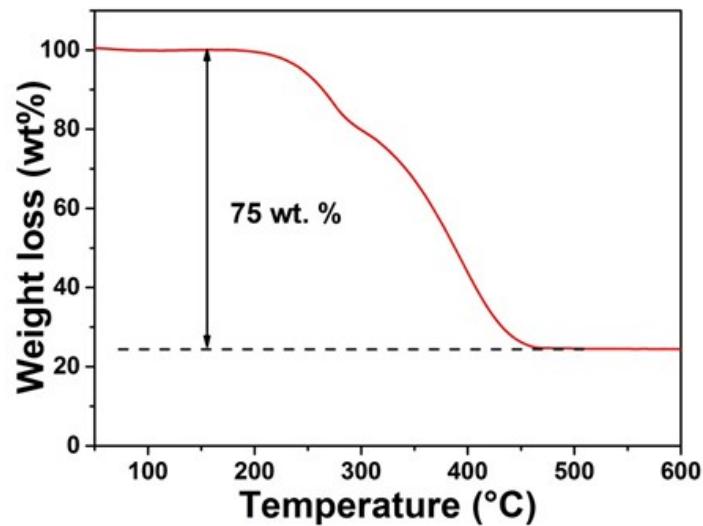


Figure S5. The TGA curve of the obtained S/TD-HDC under the N_2 atmosphere.

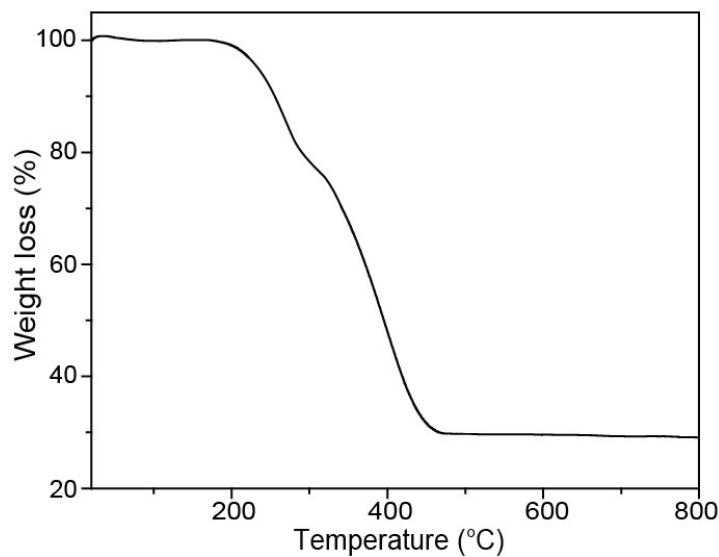


Figure S6. The TGA curve of the obtained S/HDC under the N_2 atmosphere.

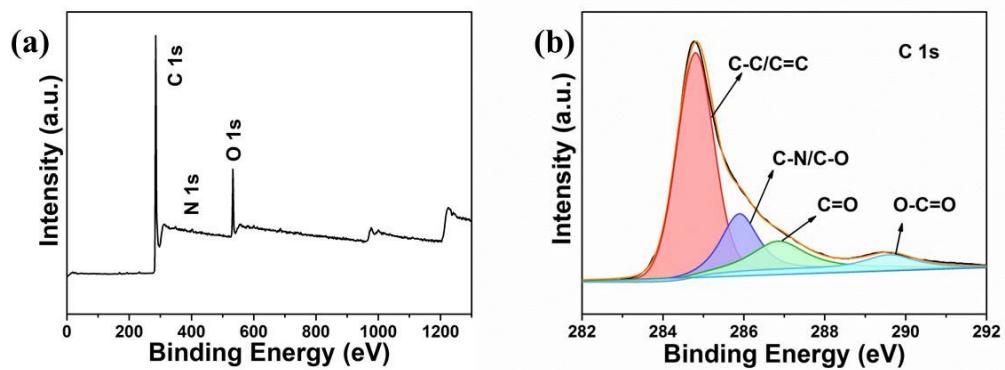


Figure S7. (a) Survey and (b) high-resolution C 1s XPS spectra of TD-HDC.

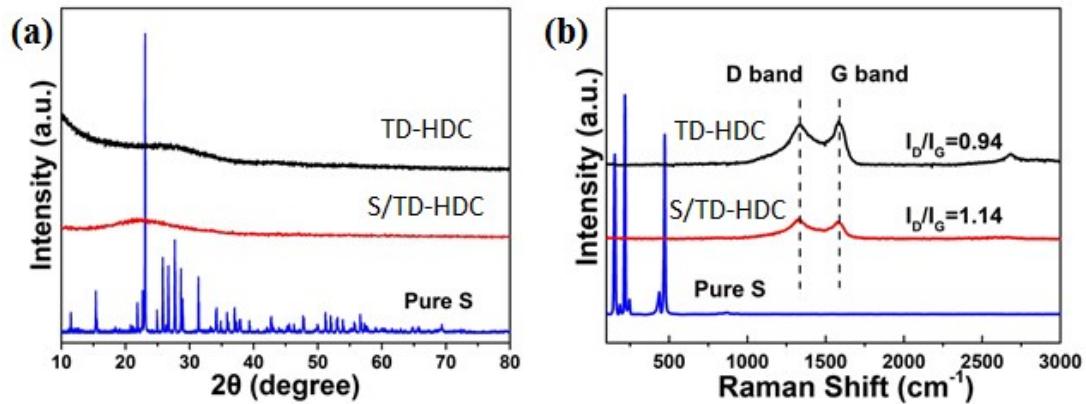


Figure S8. XRD and Raman patterns of the S/TD-HDC.

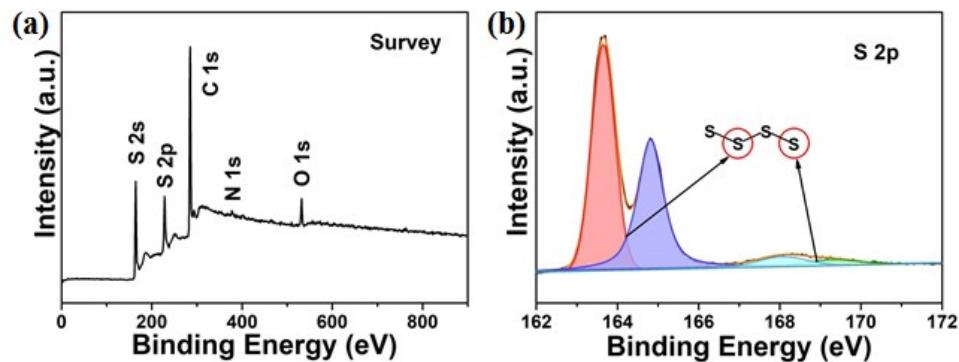


Figure S9. (a) Survey and (b) high-resolution S 2p spectra of S/TD-HDC.

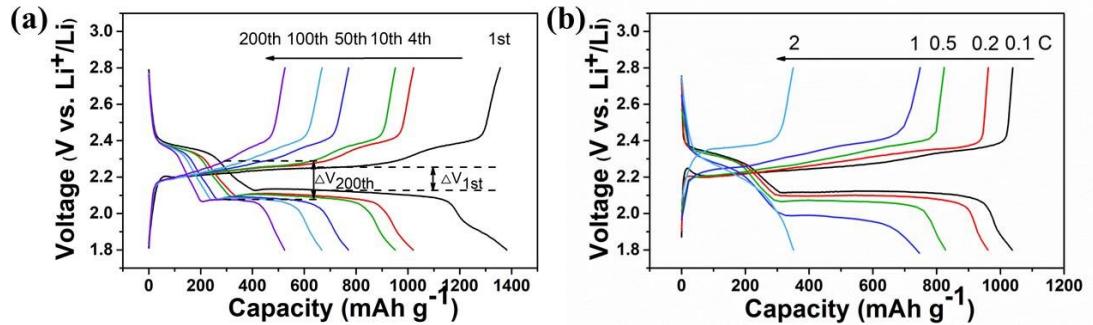


Figure S10. Discharge and charge profiles of the S/HDC cathode at (a) 0.2 C and (b) different rates, respectively.

Table S1 Specific surface areas and pore textural parameters of the as-obtained porous carbon materials.

Samples	$S_{total} (\text{m}^2 \text{g}^{-1})$	$V_{total} (\text{cm}^3 \text{g}^{-1})$	$V_{micro} (\text{cm}^3 \text{g}^{-1})$	$V_{meso} (\text{cm}^3 \text{g}^{-1})$
HDC	2043.8	0.9758	0.7267	0.2491
TD-HDC	1758.1	1.2237	0.5392	0.6845

Table S2 Elemental analysis of HDC and TD-HDC from XPS results.

Samples	C (at%)	N (at%)	O (at%)
HDC	91.4	0.86	7.74
TD-HDC	88.01	1.1	10.89

Table S3. Comparisons of TD-HDC and other carbon materials reported in the literatures serving as sulfur hosts for LSBs.

Materials	Sulfur Content (wt%)	Electrochemical performance	Refs.
N, O codoped hollow carbon spheres	66%	905, 706, 587 and 422 mAh g ⁻¹ @ 0.2, 0.5, 1, and 2 C	1
Hierarchically porous carbon materials	48.4%	980, 800, 650 and 500 mAh g ⁻¹ @ 0.1, 0.3, 0.5 and 1 C	2
N, O codoped nonporous carbonaceous material	40%	726, 540, 546, and 558 mAh g ⁻¹ @ 0.2, 0.5, 1, and 2 C	3
N doped hollow carbon nanospheres	90.4%	1139, 1003, 884, 711, and 476 mAh g ⁻¹ @ 0.2, 0.3, 0.5, 1, and 2 C	4
N-doped porous carbon cages	69.58%	1047.3, 840.8, 663.4, 515.6 mAh g ⁻¹ @ 0.2, 0.5, 1, and 2 C	5
Activated porous carbon	63%	750, 680, 615, and 520 mAh g ⁻¹ @ 0.2, 0.5, 1, and 2 C	6
Fish-scale porous carbon	—	1071, 864, 539, and 413 mAh g ⁻¹ @ 0.1, 0.2, 0.5, and 1 C	7
N-doped hierarchical porous carbon	53.8%	1188.6, 1011.3, 781.5 and 668.1 mAh g ⁻¹ @ 0.2, 0.5, 1, and 2 C	8
Defect-rich hierarchically porous carbon	72.21	1288, 1005, 884, 771 and 694 mAh g ⁻¹ @ 0.1, 0.2, 0.5, 1, and 2 C	9
TD-HDC	75%	1160.8, 1053.7, 960.3, 855.6 and 648.8 mAh g ⁻¹ at 0.1, 0.2, 0.5, 1 and 2 C	This work

Table S4 The impedance parameters of the two Li–S cells based on equivalent circuit fitting of the experimental data.

Sample	Cycle number	R _e (Ω)	R _s (Ω)	R _{ct} (Ω)	W _o (Ω)
HDC	Before cycling	2.9	—	74.4	19.6
	After 200 cycles	12.7	36.5	21.5	26.1
TD-HDC	Before cycling	5.9	—	58.7	12.7
	After 200 cycles	7.9	22.7	12.5	18.1

Reference

1. Y. Peng, Y. Zhang, J. Huang, Y. Wang, H. Li, B. J. Hwang and J. Zhao, *Carbon*, 2017, **124**, 23.
2. B. Zhang, M. Xiao, S. Wang, D. Han, S. Song, G. Chen and Y. Meng, *ACS Appl. Mater. Interfaces*, 2014, **6**, 13174.
3. K. Mi, S. Chen, B. Xi, S. Kai, Y. Jiang, J. Feng, Y. Qian and S. Xiong, *Adv. Funct. Mater.*, 2017, **27**, 1604265.
4. J. Hou, X. Tu, X. Wu, M. Shen, X. Wang, C. Wang, C. Cao, H. Pang and G. Wang, *Chem. Eng. J.*, 2020, **401**, 126141.
5. S. Zeng, G. M. Arumugam, X. Liu, Y. Yang, X. Li, H. Zhong, F. Guo and Y. Mai, *Small*, 2020, **16**, 2001027.
6. A. A. Arie, H. Kristianto, E. C. Cengiz and R. Demir-Cakan, *Ionics*, 2019, **25**, 2121.
7. M. Gao, C. Su, M. He, T. Glossmann, A. Hintennach, Z. Feng, Y. Huang and Z. Zhang, *J. Mater. Chem. A*, 2017, **5**, 6725.
8. Q. H. Q. Xiao, G. R. Li, M. J. Li, R. P. Liu, H. B. Li, P. F. Ren, Y. Dong, M. Feng and Z. W. Chen, *J. Energy Chem.*, 2020, **44**, 61.
9. L. Guan, H. Hu, L. Li, Y. Pan, Y. Zhu, Q. Li, H. Guo, K. Wang, Y. Huang, M. Zhang, Y. Yan, Z. Li, X. Teng, J. Yang, J. Xiao, Y. Zhang, X. Wang and M. Wu, *ACS Nano*, 2020, **14**, 6222.