

Supplementary Information

Synthesis, band structure and photocatalytic property of Sillén–Aurivillius oxychlorides $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$, $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ and $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ with triple-, quadruple- and quintuple-perovskite layers

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Supporting Figures S1–S20

Supporting Tables S1–S7

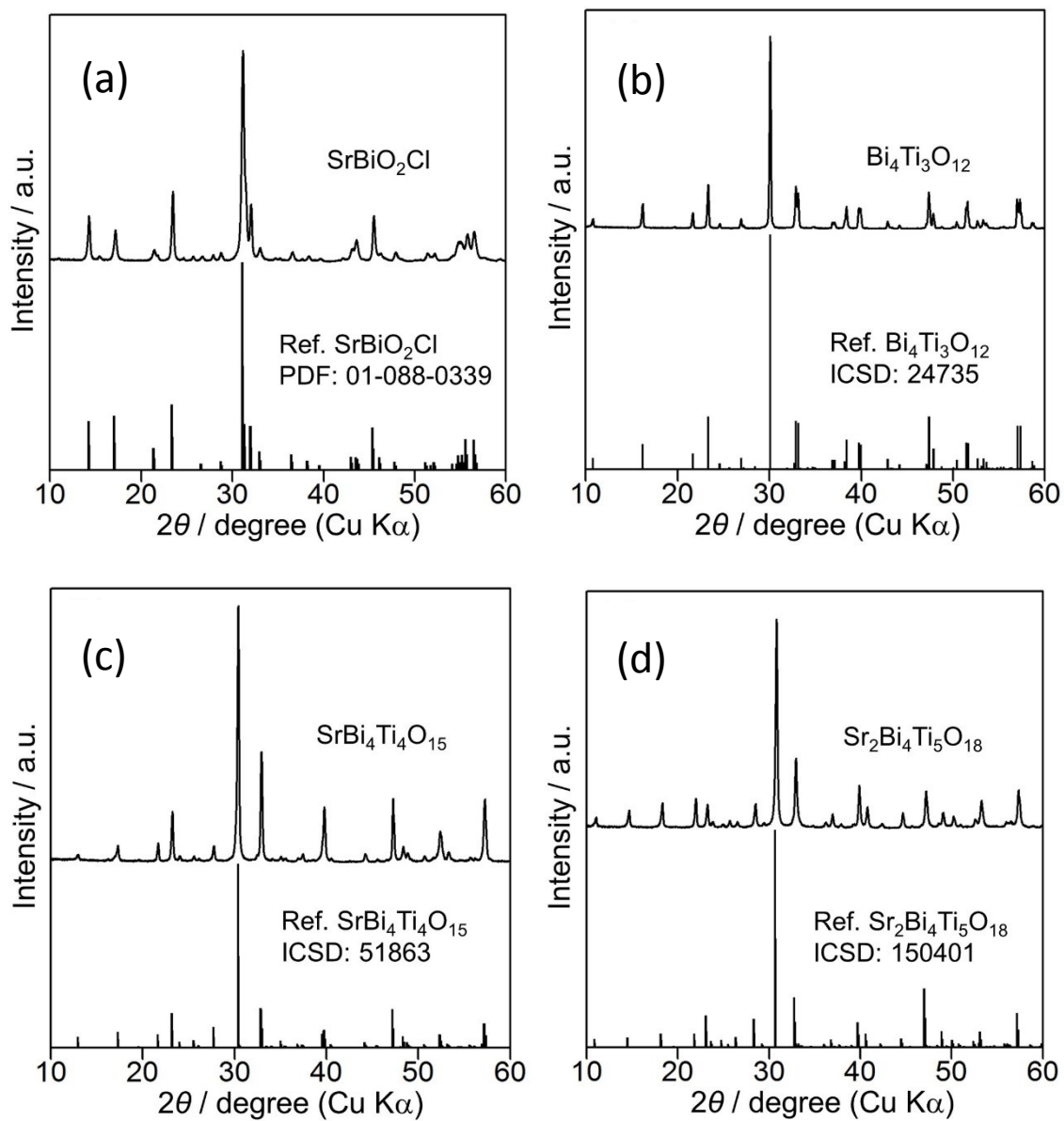


Fig. S1 XRD patterns of (a) SrBiO_2Cl , (b) $\text{Bi}_4\text{Ti}_3\text{O}_{12}$, (c) $\text{SrBi}_4\text{Ti}_4\text{O}_{15}$ and (d) $\text{Sr}_2\text{Bi}_4\text{Ti}_5\text{O}_{18}$.

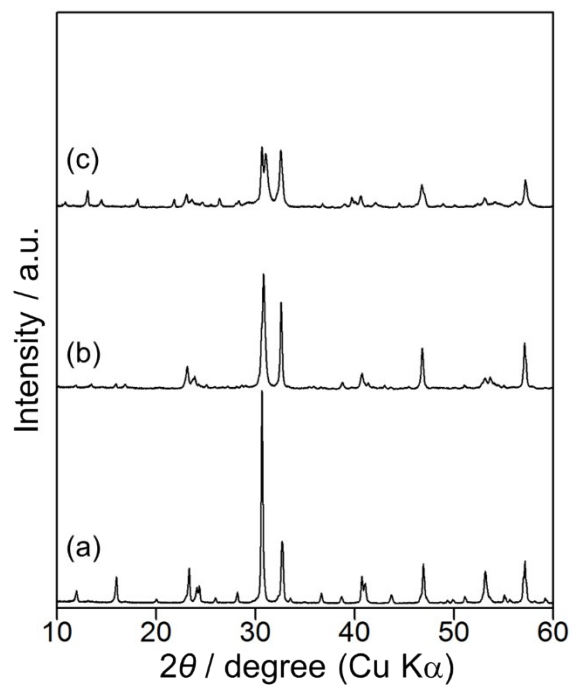


Fig. S2 Laboratory XRD patterns of Sr-containing Sillén–Aurivillius compounds. (a) $\text{Bi}_5\text{SrTi}_3\text{O}_{14}\text{Cl}$ ($n = 3$), (b) $\text{Bi}_5\text{Sr}_2\text{Ti}_4\text{O}_{17}\text{Cl}$ ($n = 4$) and (c) $\text{Bi}_5\text{Sr}_3\text{Ti}_5\text{O}_{20}\text{Cl}$ ($n = 5$). Note that (b) and (c) are not fully crystallized.

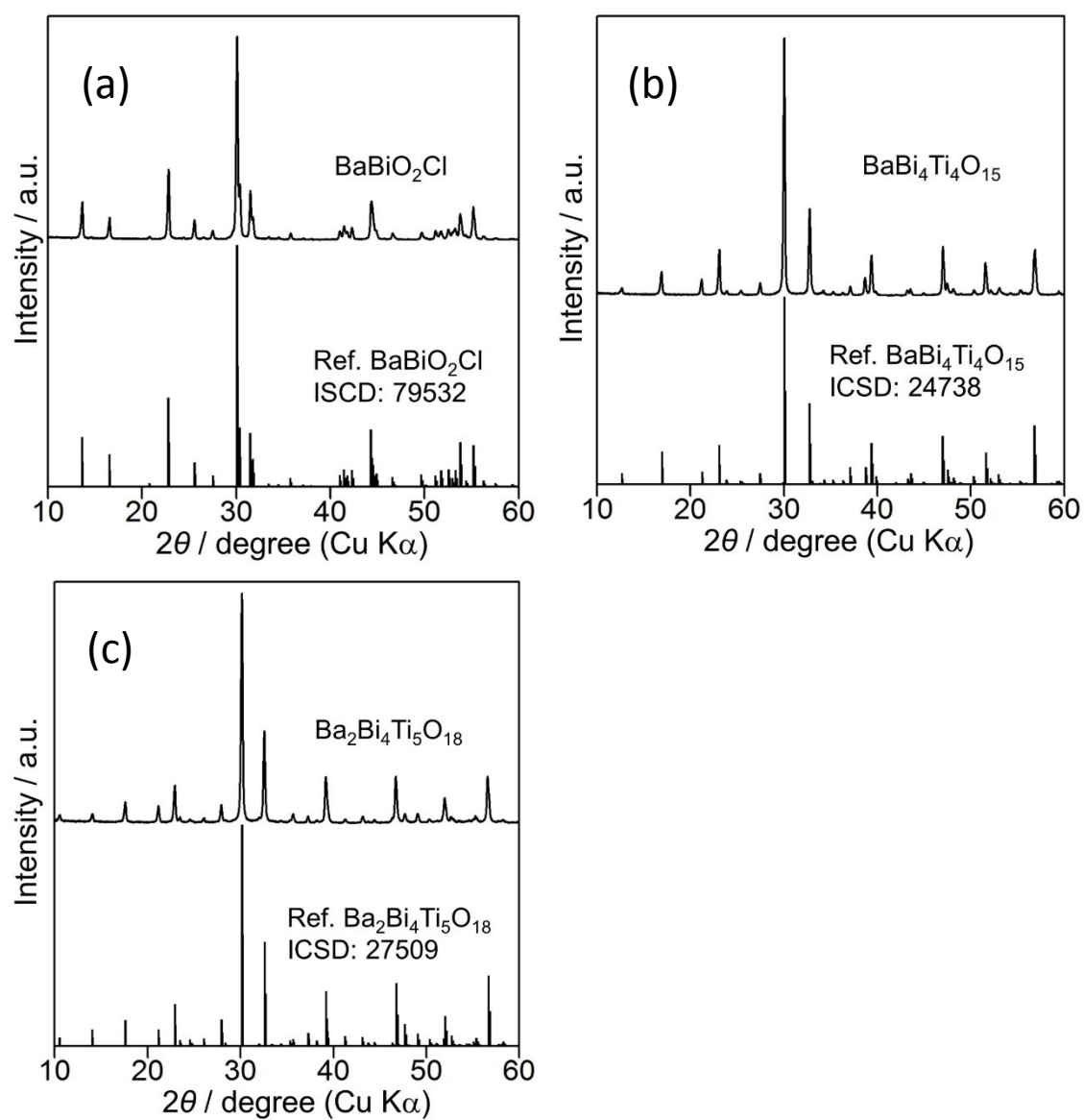


Fig. S3 XRD patterns of (a) BaBiO₂Cl, (b) BaBi₄Ti₄O₁₅ and (c) Ba₂Bi₄Ti₅O₁₈.

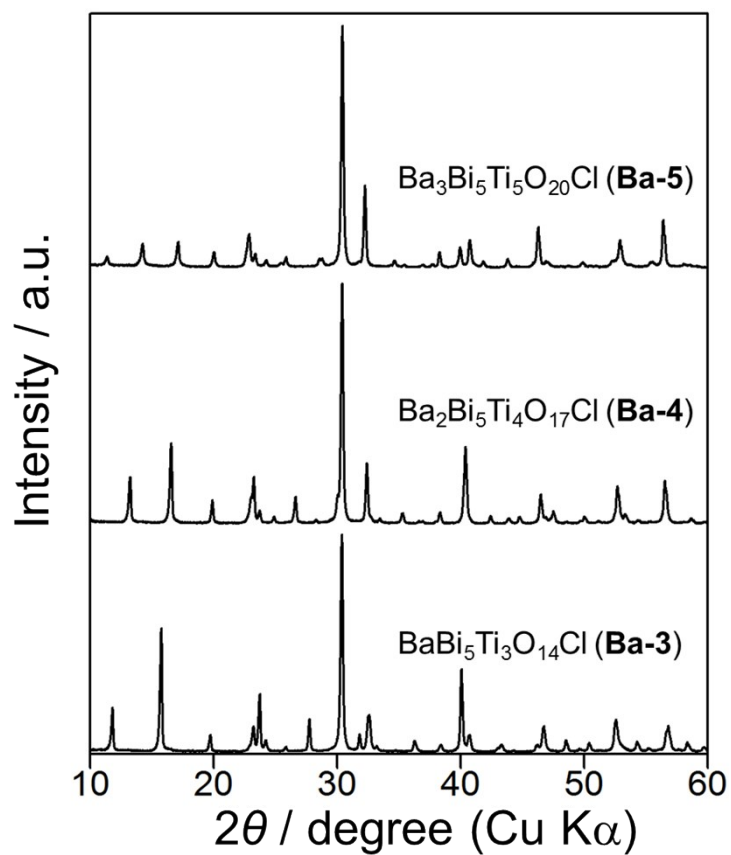


Fig. S4 Laboratory XRD patterns of BaBi₅Ti₃O₁₄Cl (**Ba-3**), Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**) and Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**).

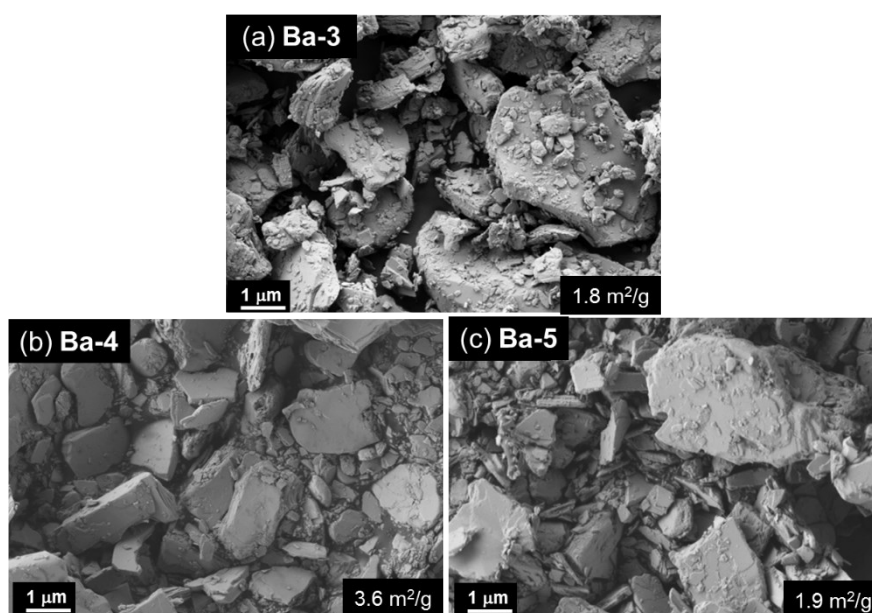


Fig. S5 SEM images of BaBi₅Ti₃O₁₄Cl (**Ba-3**), Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**) and Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**) and their specific surface areas.

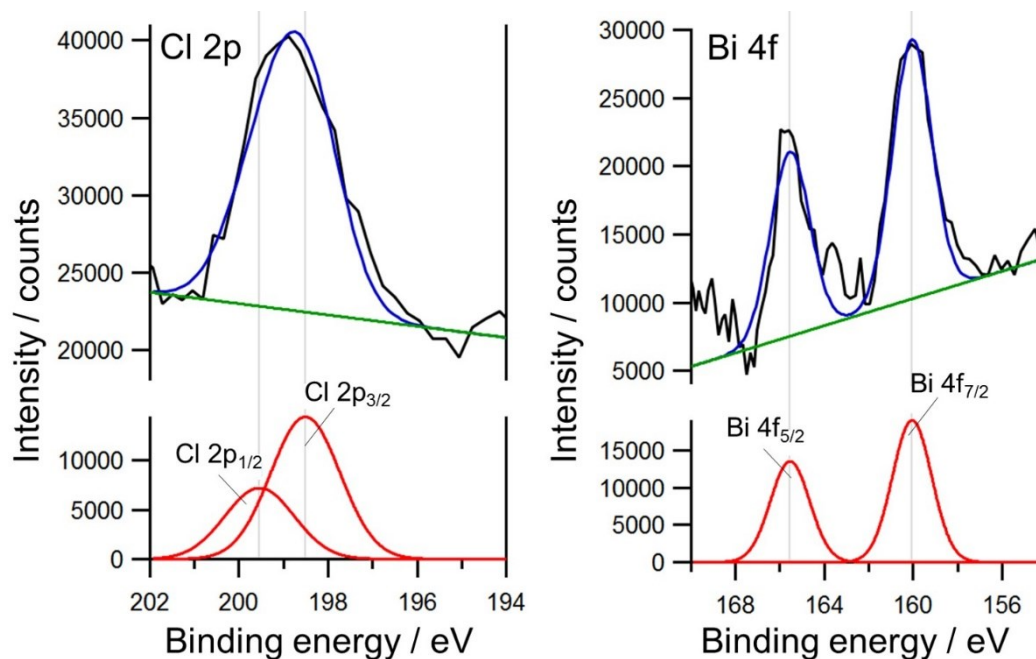


Fig. S6 EUPS spectra of BaBi₅Ti₃O₁₄Cl (**Ba-3**). (left) Cl 2p (Cl 2p_{1/2} area: 14038; Cl 2p_{3/2} area: 28076; total area: 42114) and (right) Bi 4f (Bi 4f_{5/2} area: 29932; Bi 4f_{7/2} area: 42096; total area: 72028).

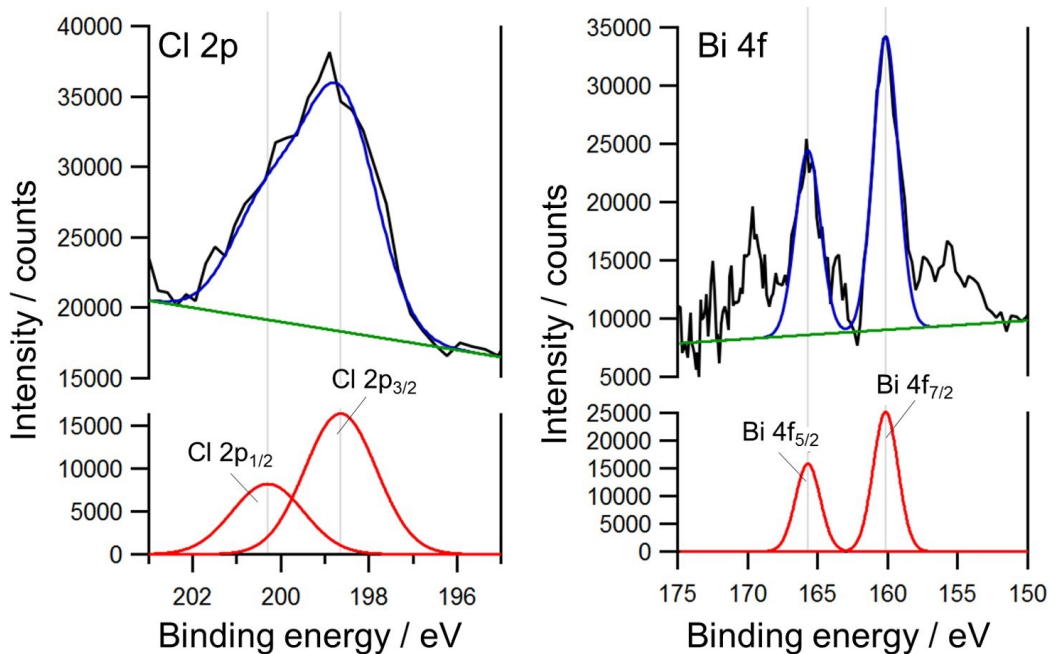


Fig. S7 EUPS spectra of $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ (**Ba-4**). (left) Cl 2p (Cl $2p_{1/2}$ area: 16714; Cl $2p_{3/2}$ area: 33428; total area: 50142) and (right) Bi 4f (Bi $4f_{5/2}$ area: 35083; Bi $4f_{7/2}$ area: 55879; total area: 90962).

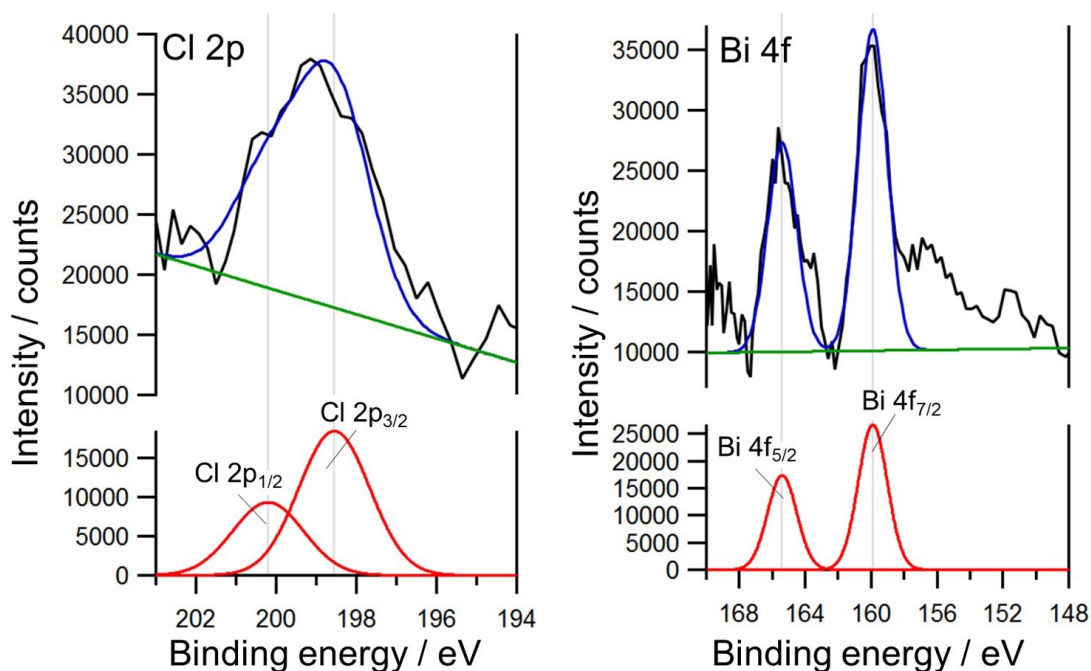


Fig. S8 EUPS spectra of $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**). (left) Cl 2p (Cl $2p_{1/2}$ area: 20578; Cl $2p_{3/2}$ area: 40659; total area: 61237) and (right) Bi 4f (Bi $4f_{5/2}$ area: 38422; Bi $4f_{7/2}$ area: 58970; total area: 97392).

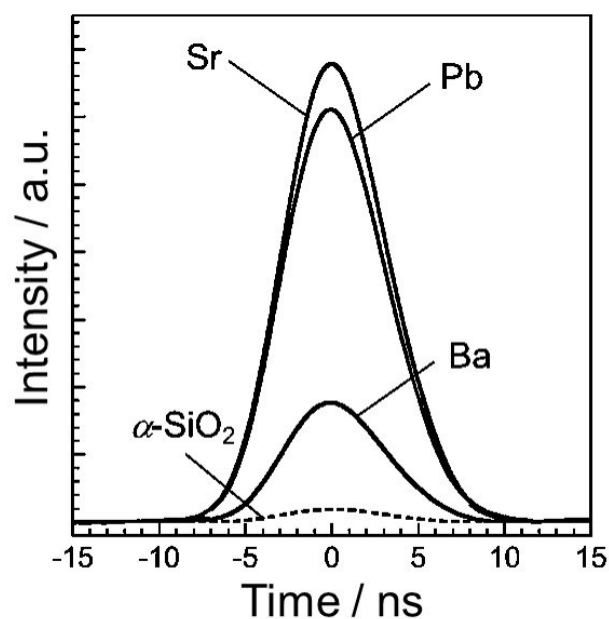


Fig. S9 SHG intensities for BaBi₅Ti₃O₁₄Cl (**Ba-3**). Bi₅MTi₃O₁₄Cl ($M = \text{Sr, Pb}$) and α -SiO₂ are references.

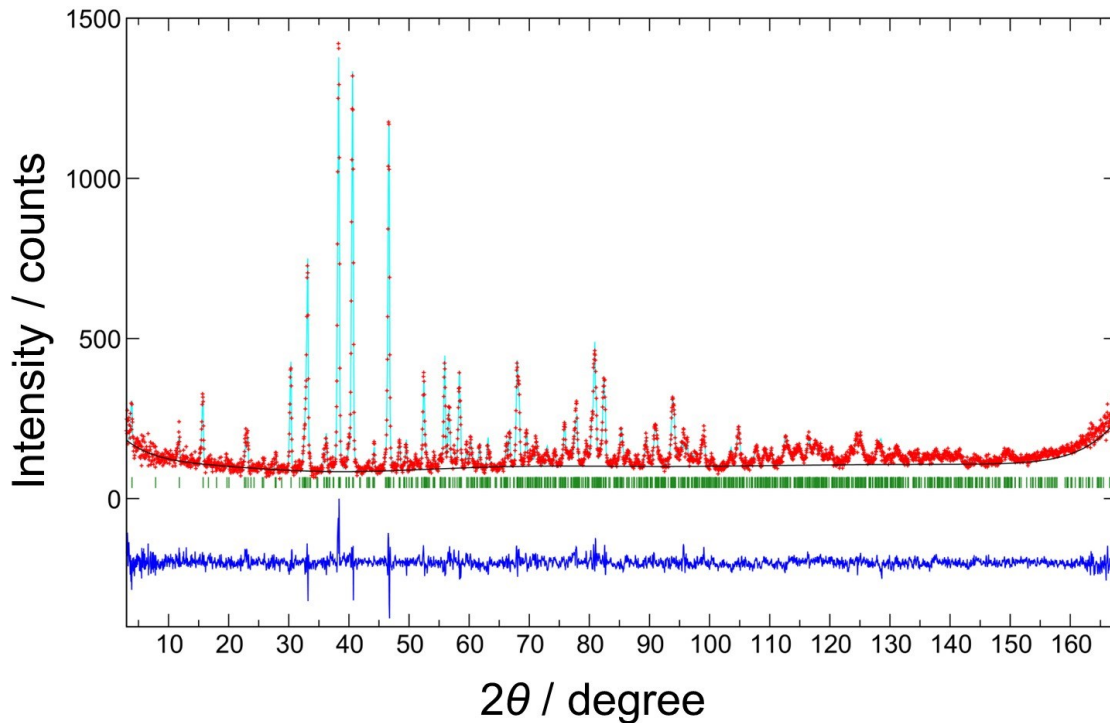


Fig. S10 Final Rietveld plot for BaBi₅Ti₃O₁₄Cl (**Ba-3**) using the NPD pattern collected at room temperature; $R_p = 0.0754$, $R_{wp} = 0.0966$, $R_B = 0.0713$, $R_F = 0.0310$, $\chi^2 = 1.171$.

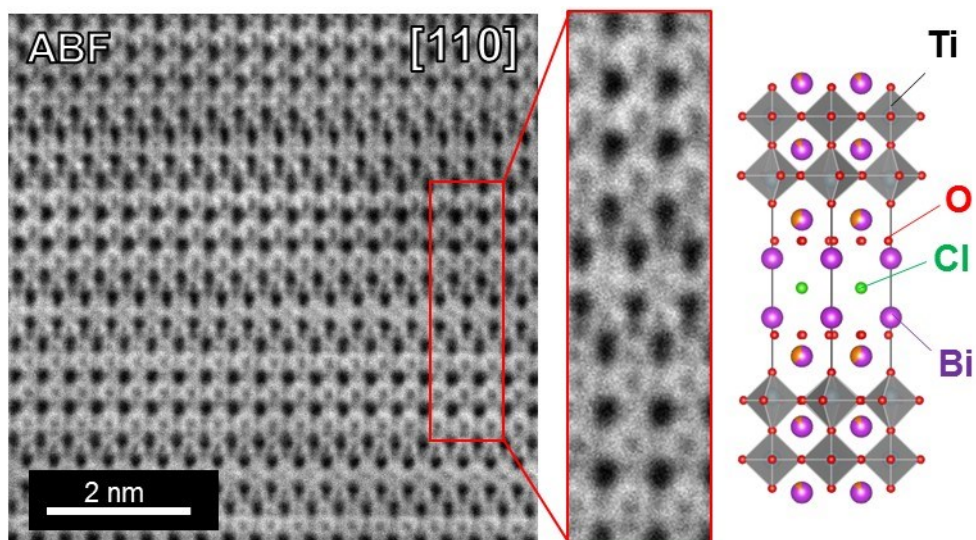


Fig. S11 ABF-STEM images of $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$ (**Ba-3**) in the direction of $[110]$.

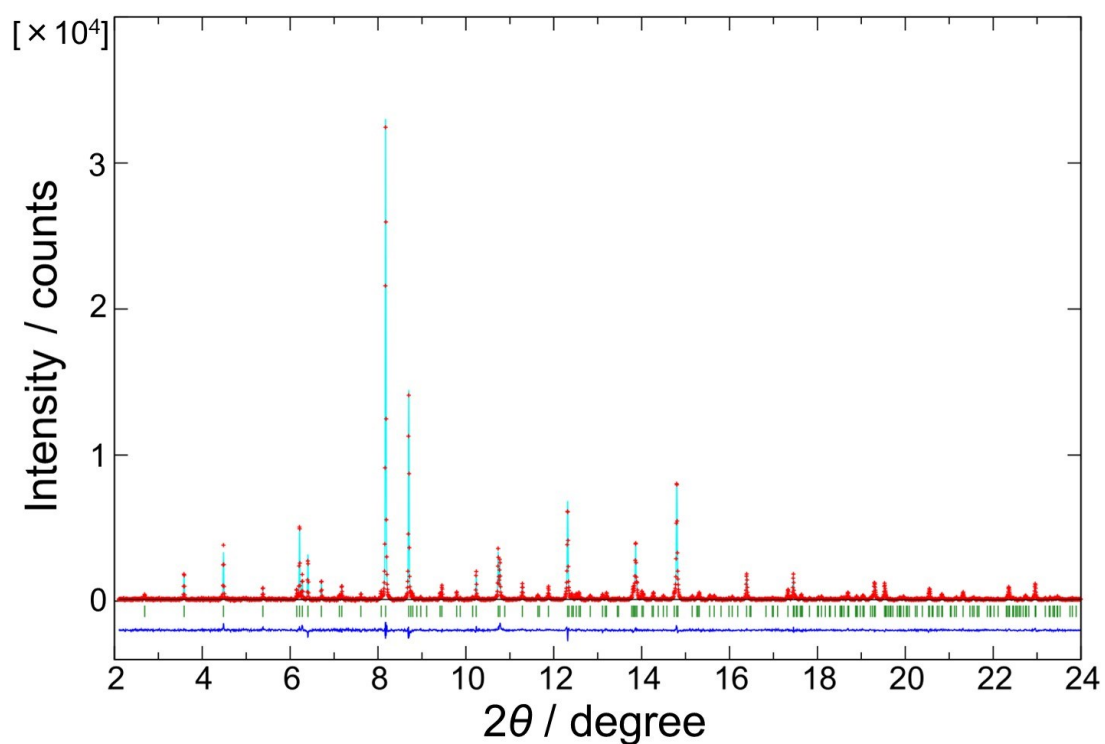


Fig. S12 Final Rietveld plot for $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ (**Ba-4**) using the SXRD pattern collected at room temperature; $R_p = 0.1184$, $R_{wp} = 0.1761$, $R_B = 0.0620$, $R_F = 0.0526$, $\chi^2 = 2.884$.

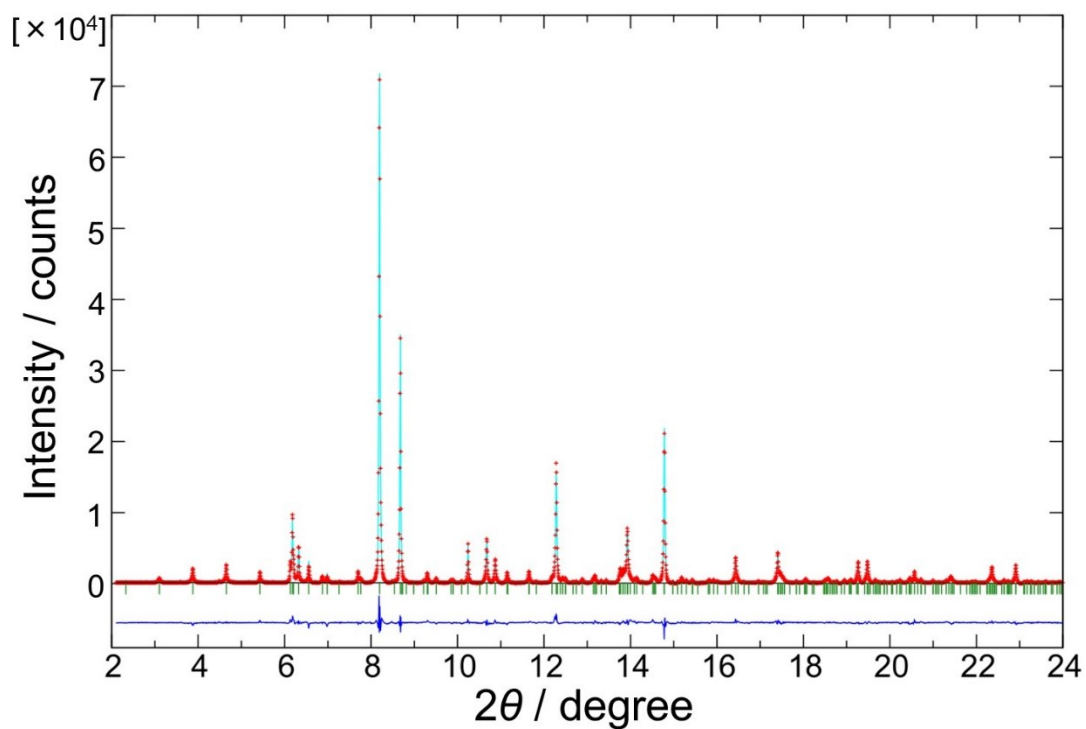


Fig. S13 Final Rietveld plot for $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**) using the SXRD pattern collected at room temperature; $R_p = 0.0944$, $R_{wp} = 0.1437$, $R_B = 0.0468$, $R_F = 0.0365$, $\chi^2 = 3.609$.

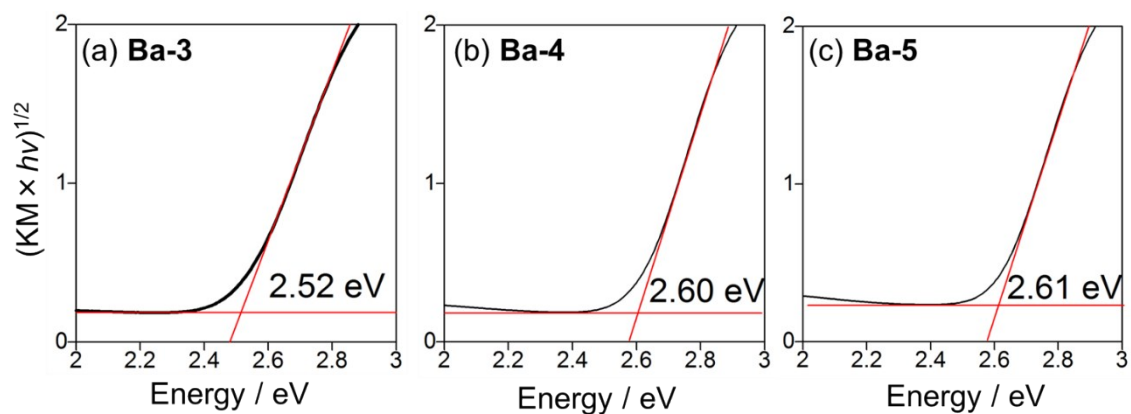


Fig. S14 Tauc plots of (a) $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$ (**Ba-3**), (b) $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ (**Ba-4**) and (c) $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**).

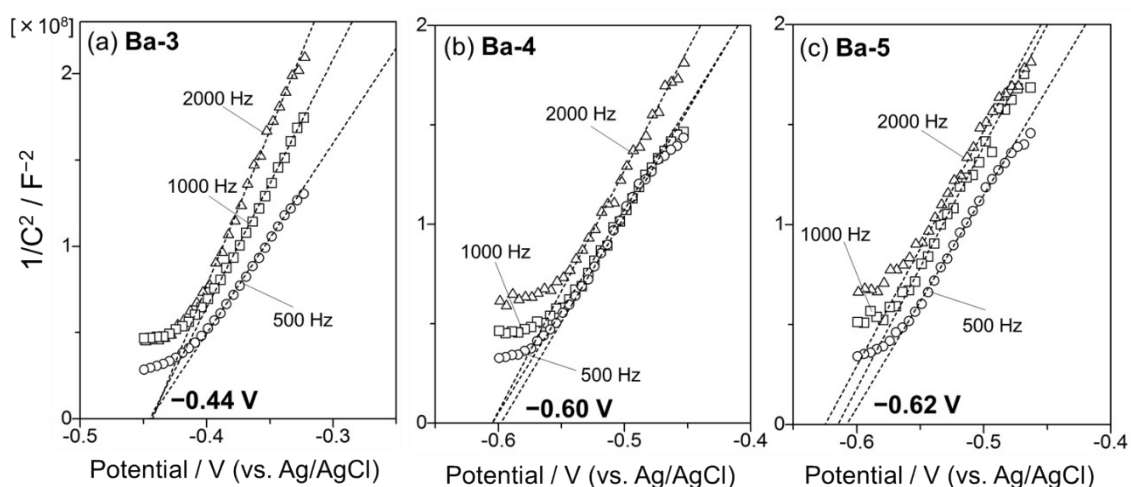


Fig. S15 Mott-Schottky plots of (a) $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$ (**Ba-3**), (b) $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ (**Ba-4**) and (c) $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**) collected in an aqueous Na_2SO_4 solution (0.1 M, pH 2.0, adjusted by 0.5 M H_2SO_4) with the amplitude and frequency were set to 10 mV and 500, 1000, 2000 Hz, respectively.

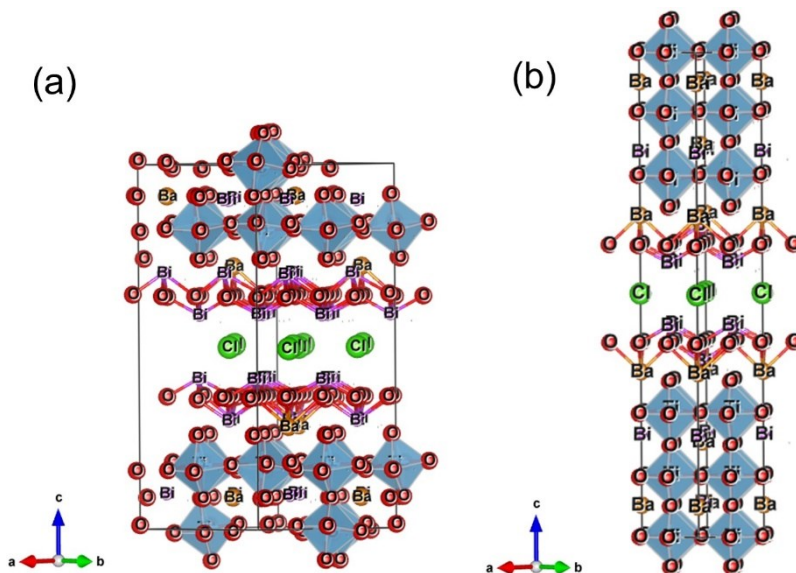


Fig. S16 Structural models employed for the DFT calculation; (a) 192-atom supercell of $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$ (**Ba-3**) and (b) 68-atom supercell of $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**).

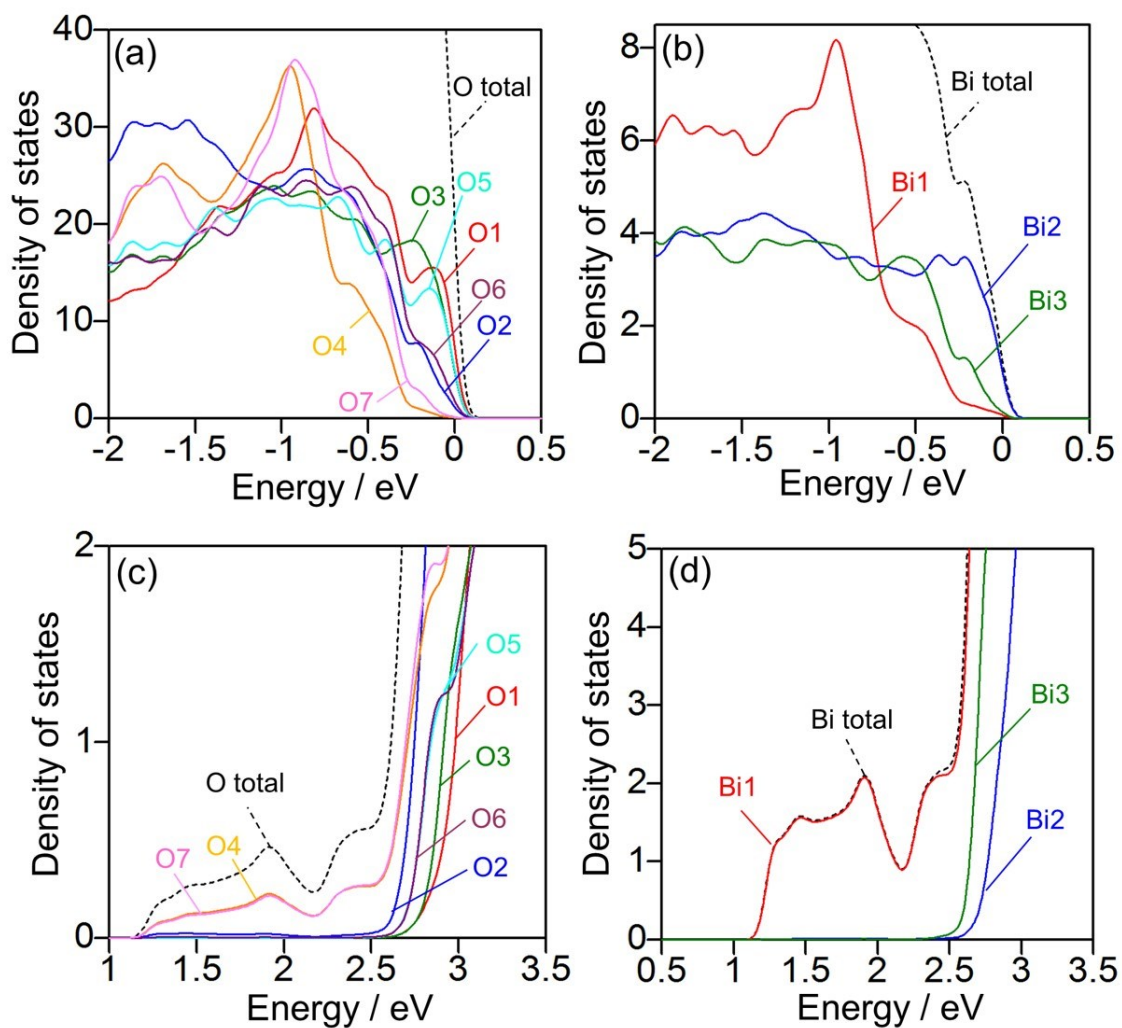


Fig. S17 PDOS around the VBM from (a) each O site and (b) Bi site, and around the CBM from (c) each O site and (d) Bi site in BaBi₅Ti₃O₁₄Cl (**Ba-3**). See Fig. 5 for the detailed labelling.

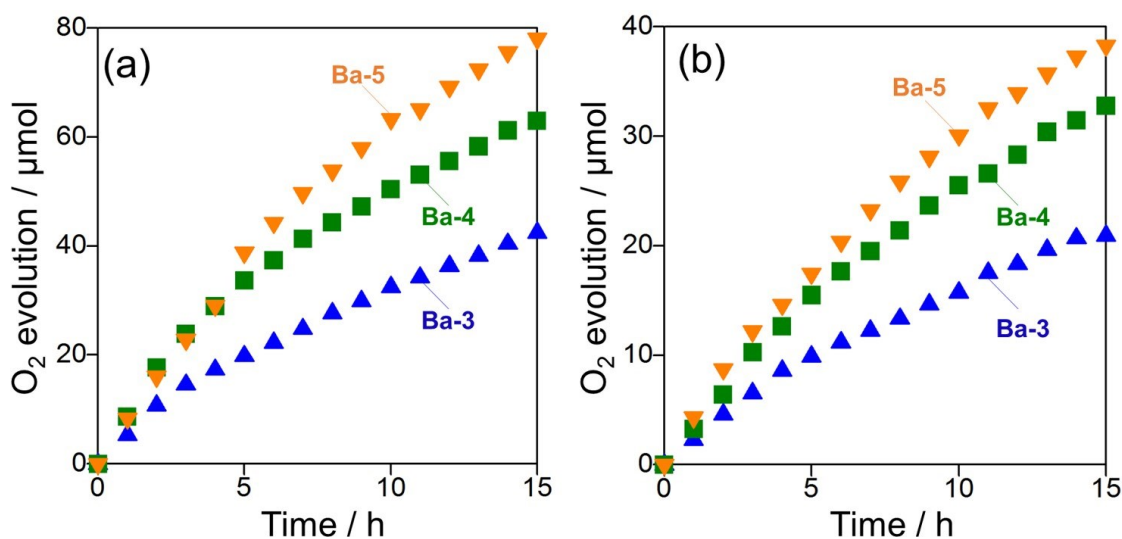


Fig. S18 Time courses of photocatalytic O₂ evolution on 100 mg of BaBi₅Ti₃O₁₄Cl (**Ba-3**), Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**) and Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**) photocatalysts under visible light (400 < λ < 800 nm) in aqueous solutions of (a) AgNO₃ (10 mM, 120 mL) and (b) Fe(ClO₄)₃ (5 mM, 120 mL, pH 2.3, adjusted by HClO₄).

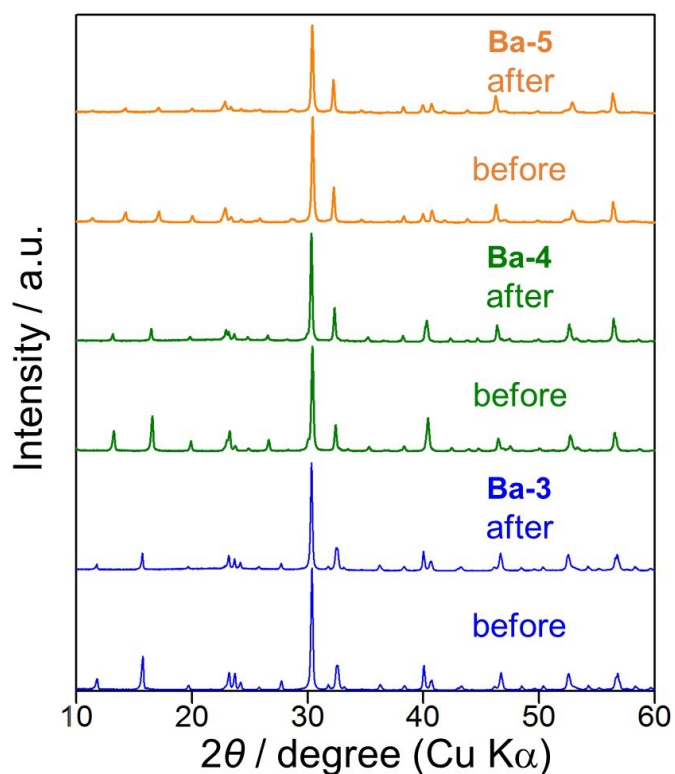


Fig. S19 XRD patterns of BaBi₅Ti₃O₁₄Cl (**Ba-3**), Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**) and Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**) before and after photocatalytic O₂ evolution in an Fe(ClO₄)₃ aqueous solution under visible light.

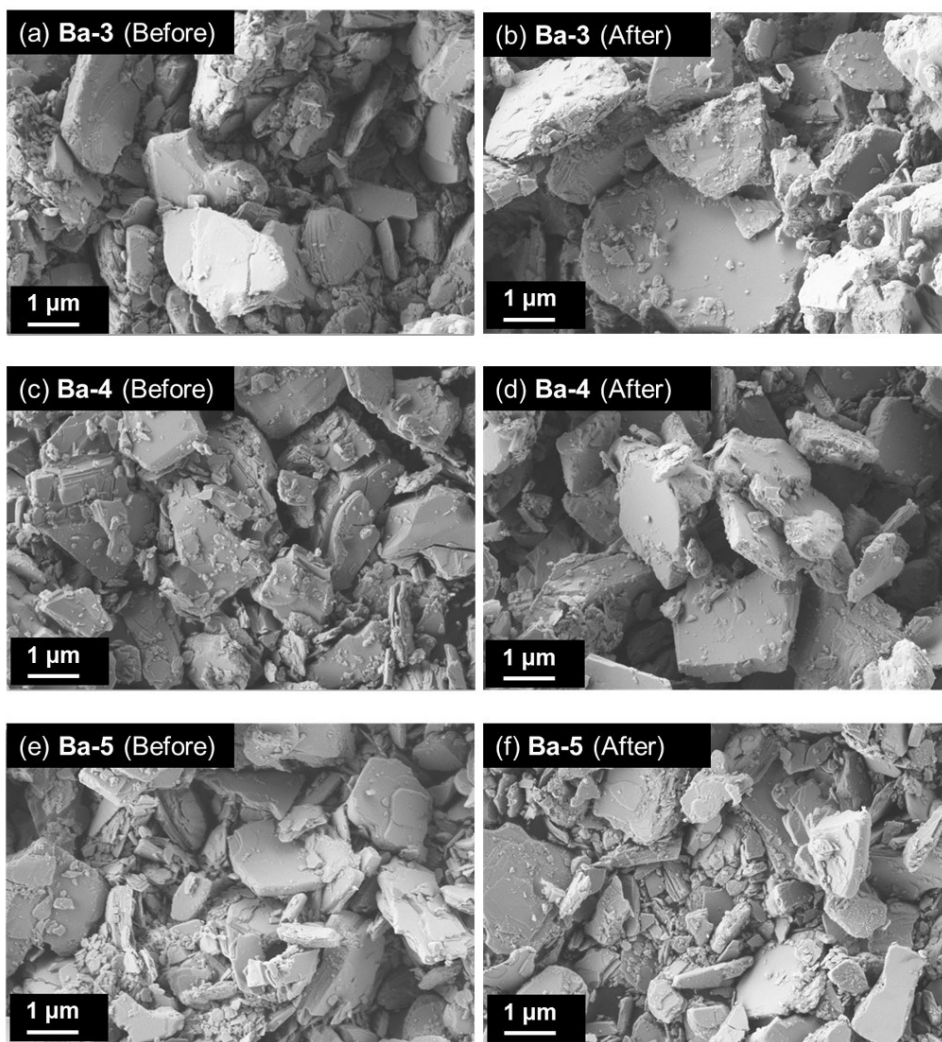


Fig. S20 SEM images of $\text{BaBi}_5\text{Ti}_3\text{O}_{14}\text{Cl}$ (**Ba-3**), $\text{Ba}_2\text{Bi}_5\text{Ti}_4\text{O}_{17}\text{Cl}$ (**Ba-4**) and $\text{Ba}_3\text{Bi}_5\text{Ti}_5\text{O}_{20}\text{Cl}$ (**Ba-5**) before and after photocatalytic O_2 evolution in an $\text{Fe}(\text{ClO}_4)_3$ aqueous solution under visible light.

Table S1 Final refined structure parameters for BaBi₅Ti₃O₁₄Cl (**Ba-3**)^a

Atom	Site	x	y	z	g	100U / Å ²	BVS
Bi/Ba1	4c	0.031(3)	-0.000(3)	0.41350(9)	Bi: 1	1.13(5)	3.1 ^e
Bi/Ba2	4c	0 ^b	0.497(3)	0.0956(10)	Bi: 0.874(2) Ba: 0.126(2)	2.56(7)	2.4 ^e
Bi/Ba3	4c	0.038(14)	0.495(2)	0.3008(11)	Bi: 0.626(2) Ba: 0.374(2)	2.55(7)	2.2 ^e
Ti1	2a	0.018(8)	0	0	1	0.3(2) ^c	3.9
Ti2	4c	0.009(6)	-0.00(11)	0.1843(4)	1	0.3(2) ^c	4.2
Cl1	2b	0.026(4)	1/2	1/2	1	1.8(14)	-
O1	4c	-0.014(3)	-0.035(3)	0.0896(5)	1	2.54(7) ^d	-
O2	4c	0.050(4)	0.023(4)	0.2623(4)	1	2.54(7) ^d	-
O3	4c	0.293(3)	0.218(4)	0.004(11)	1	2.54(7) ^d	-
O4	4c	0.281(5)	0.245(9)	0.3646(9)	1	2.54(7) ^d	-
O5	4c	0.319(4)	0.222(4)	0.1688(8)	1	2.54(7) ^d	-
O6	4c	0.775(4)	0.734(6)	0.1773(7)	1	2.54(7) ^d	-
O7	4c	0.783(5)	0.744(8)	0.363(10)	1	2.54(7) ^d	-

^aSpace group *P2an* (#30), orthorhombic, *a* = 5.50525(6) Å, *b* = 5.48057(5) Å, *c* = 22.4870(2) Å, *V* = 678.49(12) Å³. ^bFixed to define origin of polar axis. ^cConstrained to be equivalent to Ti1. ^dConstrained to be equivalent to O1. ^eCalculated by using bond valence parameter of Bi³⁺.

Table S2 Bond lengths of BaBi₅Ti₃O₁₄Cl (**Ba-3**)

Bond	Bond length / Å
Ti1–O1	2.032(13) × 2
Ti1–O3	1.93(4)×2, 1.98(4) × 2
Ti2–O1	2.140(16)
Ti2–O2	1.770(14)
Ti2–O5	1.89(6), 2.13(5)
Ti2–O6	1.95(6), 1.95(6)
Bi1–O4	2.22(4), 2.25(4)
Bi1–O7	2.23(1), 2.26(1)
Bi1–Cl1	3.349(17), 3.360(14), 3.361(14), 3.392(18)
Bi2–O1	2.57(3), 2.685(18), 2.842(18), 2.92(3)
Bi2–O3	2.64(3), 2.78(3), 3.04(3), 3.17(3)
Bi2–O5	2.27(3), 2.84(3)
Bi2–O6	2.57(3), 2.84(3)
Bi3–O2	2.82(3), 2.847(14), 2.902(14), 2.95(3)
Bi3–O4	2.39(4), 2.41(4)
Bi3–O7	2.42(4), 2.41(4)

Table S3 Final refined structure parameters for Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**)^a

Atom	Sit	x	y	z	g	100U / Å ²	BVS
Bi/Ba1	2h	1/2	1/2	0.4267(12)	Bi: 1	1.01(6)	3.5 ^c
Bi/Ba2	2g	0	0	Bi2: 0.3385(2) Ba2: 0.3133(6)	Bi: 0.670(3) Ba: 0.330(3)	1.7(10)	2.1 ^c
Bi/Ba3	2g	0	0	Bi3: 0.1669(3) Ba3: 0.1505(5)	Bi: 0.574(3) Ba: 0.426(3)	2.4(11)	2.5 ^c
Bi/Ba4	1a	0	0	0	Bi: 0.512(3) Ba: 0.488(3)	3.5(2)	2.1 ^c
Ti1	2h	1/2	1/2	0.2364(5)	1	1.1(3)	4.2
Ti2	2h	1/2	1/2	0.0769(6)	1	0.9(3)	3.7
Cl1	1b	0	0	1/2	1	2.4(7)	-
O1	4i	0	1/2	0.391(12)	1	1.2(3) ^b	-
O2	2h	1/2	1/2	0.298(2)	1	1.2(3) ^b	-
O3	4i	0	1/2	0.220(11)	1	1.2(3) ^b	-
O4	2h	1/2	1/2	0.153(2)	1	1.2(3) ^b	-
O5	4i	0	1/2	0.070(11)	1	1.2(3) ^b	-
O6	1c	1/2	1/2	0	1	1.2(3) ^b	-

^aSpace group *P4/mmm* (#123), tetragonal, $a = b = 3.91036(4)$ Å, $c = 26.8125(4)$ Å, $V = 409.988(9)$ Å³. ^bConstrained to be equivalent to O1. ^cCalculated by using bond valence parameter of Bi³⁺.

Table S4 Final refined structure parameters for Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**)^a

Atom	Site	x	y	z	g	100U / Å ²	BVS
Bi/Ba1	2h	1/2	1/2	0.43608(7)	Bi: 1	2.15(6)	3.2 ^d
Bi/Ba2	2g	0	0	Bi2: 0.3611(11)	Bi: 0.573(4)	2.20(10)	2.4 ^d
				Ba2: 0.3413(3)	Ba: 0.427(4)		
Bi/Ba3	2g	0	0	Bi3: 0.2042(2)	Bi: 0.489(4)	2.03(9)	2.0 ^d
				Ba3: 0.2167(3)	Ba: 0.511(4)		
Bi/Ba4	2g	0	0	Bi4: 0.0629(2)	Bi: 0.438(4)	2.96(9)	2.0 ^d
				Ba4: 0.0753(2)	Ba: 0.562(4)		
Ti1	2h	1/2	1/2	0.2725(3)	1	0.6(2) ^b	4.4
Ti2	2h	1/2	1/2	0.1343(3)	1	0.6(2) ^b	4.1
Ti3	1c	1/2	1/2	0	1	0.6(2) ^b	4.3
Cl1	1b	0	0	1/2	1	1.8(5)	-
O1	4i	0	1/2	0.4029(7)	1	0.8(2) ^c	-
O2	2h	1/2	1/2	0.323(10)	1	0.8(2) ^c	-
O3	4i	0	1/2	0.2566(7)	1	0.8(2) ^c	-
O4	2h	1/2	1/2	0.191(11)	1	0.8(2) ^c	-
O5	4i	0	1/2	0.1261(7)	1	0.8(2) ^c	-
O6	2h	1/2	1/2	0.061(11)	1	0.8(2) ^c	-
O7	2f	0	1/2	0	1	0.8(2) ^c	-

^aSpace group *P4/mmm* (#123), tetragonal, $a = b = 3.92036(4)$ Å, $c = 30.9881(7)$ Å, $V = 476.26(12)$ Å³. ^bConstrained to be equivalent to Ti1. ^cConstrained to be equivalent to O1.

^dCalculated by using bond valence parameter of Bi³⁺.

Table S5 Bond lengths of Ba₂Bi₅Ti₄O₁₇Cl (**Ba-4**) and Ba₃Bi₅Ti₅O₂₀Cl (**Ba-5**)

Bond	Bond length / Å	
	Ba-4	Ba-5
Ti1–O2	1.66(5)	1.57(4)
Ti1–O3	2.004(8) × 4	2.001(6) × 4
Ti1–O4	2.24(5)	2.53(4)
Ti2–O4	2.04(5)	1.76(4)
Ti2–O5	1.963(4) × 4	1.977(3)
Ti2–O6	2.061(16)	2.26(4)
Ti3–O6	-	1.90(4)
Ti3–O7	-	1.960
Bi1–Cl1	3.3920(19) × 4	3.4071(13) × 4
Bi1–O1	2.176(15) × 4	2.214(10) × 4
Bi2–O1	2.411(19) × 4	2.348(12) × 4
Ba2–O1	2.86(3) × 4	2.736(16) × 4
Bi2–O2	2.966(16) × 4	3.011(13) × 4
Ba2–O2	2.794(7) × 4	2.828(7) × 4
Bi3–O3	2.418(18) × 4	2.546(14) × 4
Ba3–O3	2.70(3) × 4	2.317(12) × 4
Bi3–O4	2.791(6) × 4	2.802(5) × 4
Ba3–O4	2.7657(10) × 4	2.885(10) × 4
Bi3–O5	3.25(3) × 4	3.113(17) × 4
Ba3–O5	2.91(3) × 4	3.424(18) × 4
Bi4–O5	2.72(3) × 4	2.771(15) × 4
Bi4–O6	2.76504(3) × 4	2.7726(6) × 4
Bi4–O7	-	2.765(4) × 4
Bi4–O5	-	2.516(14) × 4
Bi4–O6	-	2.806(6) × 4
Bi4–O7	-	3.046(6) × 4

Table S6 Comparison between calculated and experimental band gaps of **Ba-3** and **Ba-5**

Material	Band gap / eV	
	DFT (GGA-PBE)	Experimental
Ba-3	1.20 (-52.4%) ^a	2.52
Ba-5	0.55 (-78.9%) ^a	2.61

^aError to the experimental value

Table S7 Comparison between calculated and experimental lattice parameters of **Ba-3** and **Ba-5**

Ba-3	<i>a</i> / Å	<i>b</i> / Å	<i>c</i> / Å
DFT	11.27 (+2.4%) ^a	11.17 (+1.9%) ^a	22.66 (+0.8%) ^a
Experimental	11.01	10.96	22.49

Ba-5	<i>a</i> / Å	<i>b</i> / Å	<i>c</i> / Å
DFT	5.61 (1.3%) ^a	5.61 (1.3%) ^a	31.52 (1.7%) ^a
Experimental	5.54	5.54	30.99

^aError to the experimental value