

## Supporting Information

### Enhanced luminescence of Near-Infrared-Sensitized Upconversion

#### Nanoparticles via Ca<sup>2+</sup> doping for Nitric Oxide release platform

Jing Zhao,<sup>a,d</sup> Yanbing Hu,<sup>a,d</sup> Shao wei Lin<sup>a</sup>, U. Resch-Genger,<sup>b</sup> Rui Zhang,<sup>a</sup> Jiang Wen<sup>c</sup>, Xiangfei Kong,<sup>a</sup> Aimiiao Qin,<sup>a</sup> Jun Ou,<sup>\*a</sup>

<sup>a</sup>. Materials Science and Engineering College, Guilin University of Technology, Key Laboratory of New Processing Technology for Nonferrous Materials, Ministry of Education, Guangxi Key Laboratory of Optical and Electronic Materials and Devices, Guangxi Collaborative Innovation Center for Exploration of Hidden Nonferrous Metal Deposits and Development of New Materials, Guilin University of Technology, 541004 Guilin, China;

<sup>b</sup>. Federal Institute for Materials Research and Testing (BAM), 12489 Berlin, Germany;

<sup>c</sup>. Experimental Center of Medical Sciences, Guilin Medical University, 541002 Guilin, China

<sup>d</sup>. These authors contributed equally to this work and should be considered co-first authors.

\* Corresponding authors.

E-mail address: gloujun@126.com

## TABLE OF CONTENTS

<b>Figure S1</b> .....	<b>2</b>
<b>Figure S2</b> .....	<b>2</b>
<b>Figure S3</b> .....	<b>3</b>
<b>Figure S4</b> .....	<b>3</b>
<b>Figure S5</b> .....	<b>4</b>
<b>Figure S6</b> .....	<b>5</b>
<b>Table S1</b> .....	<b>6</b>
<b>Table S2</b> .....	<b>7</b>

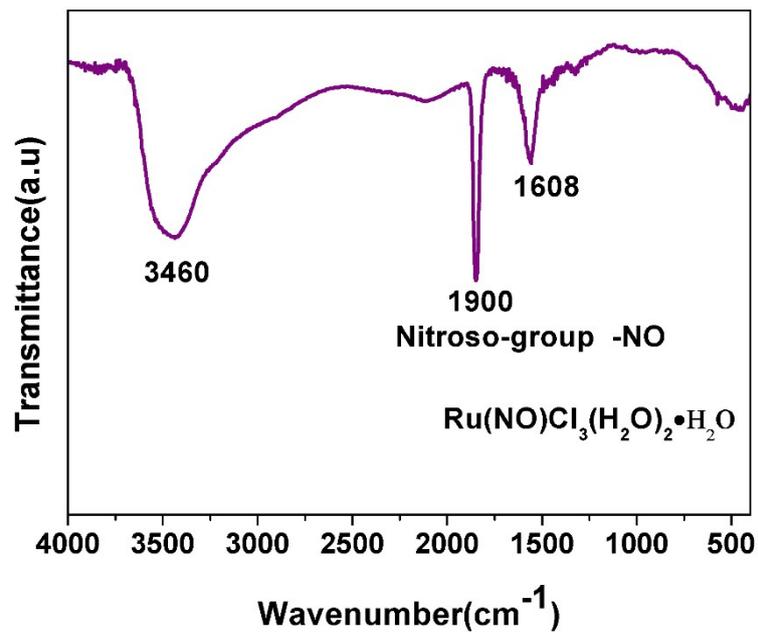


Figure. S 1. Infrared spectra of hydrated nitrosotrichlorosilane.

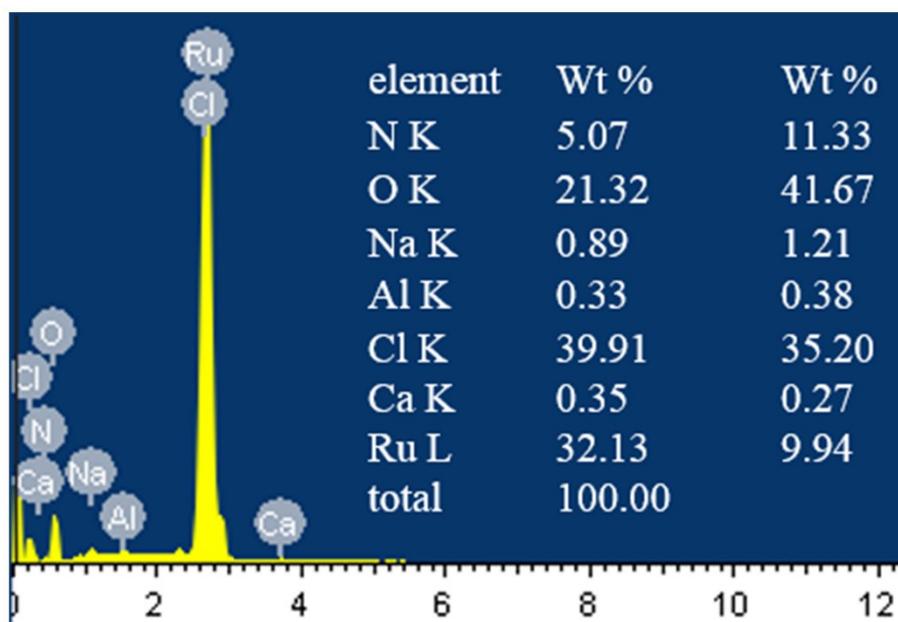
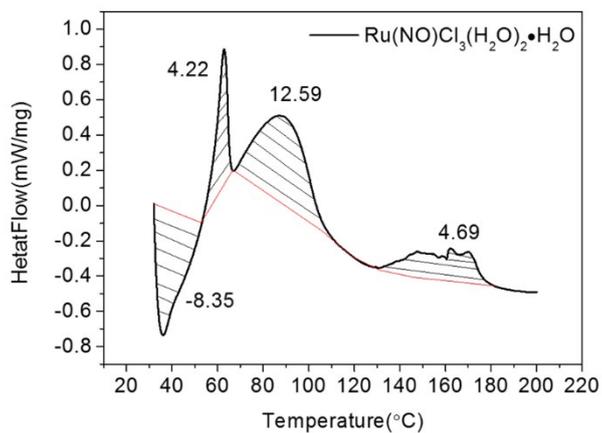
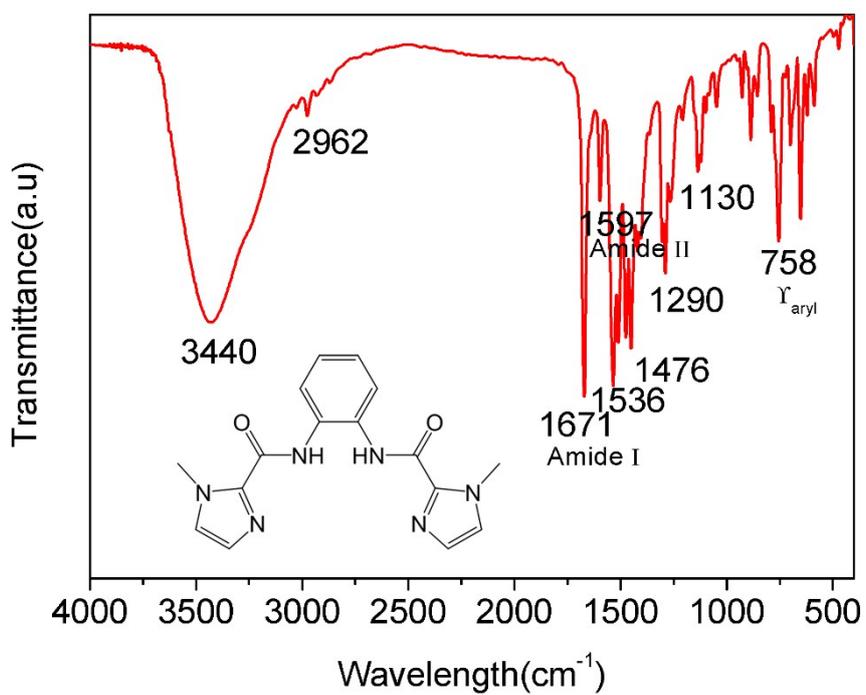


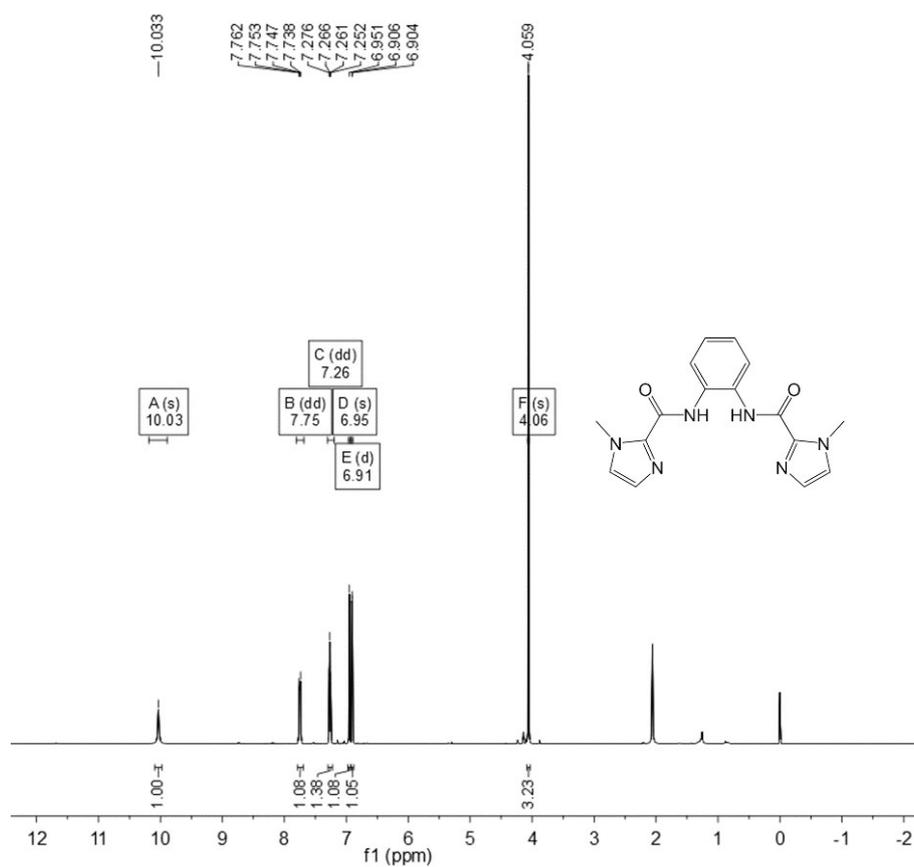
Figure. S 2 EDS energy spectrum and EDS element percentage of hydrated nitrosotrichlorosilane



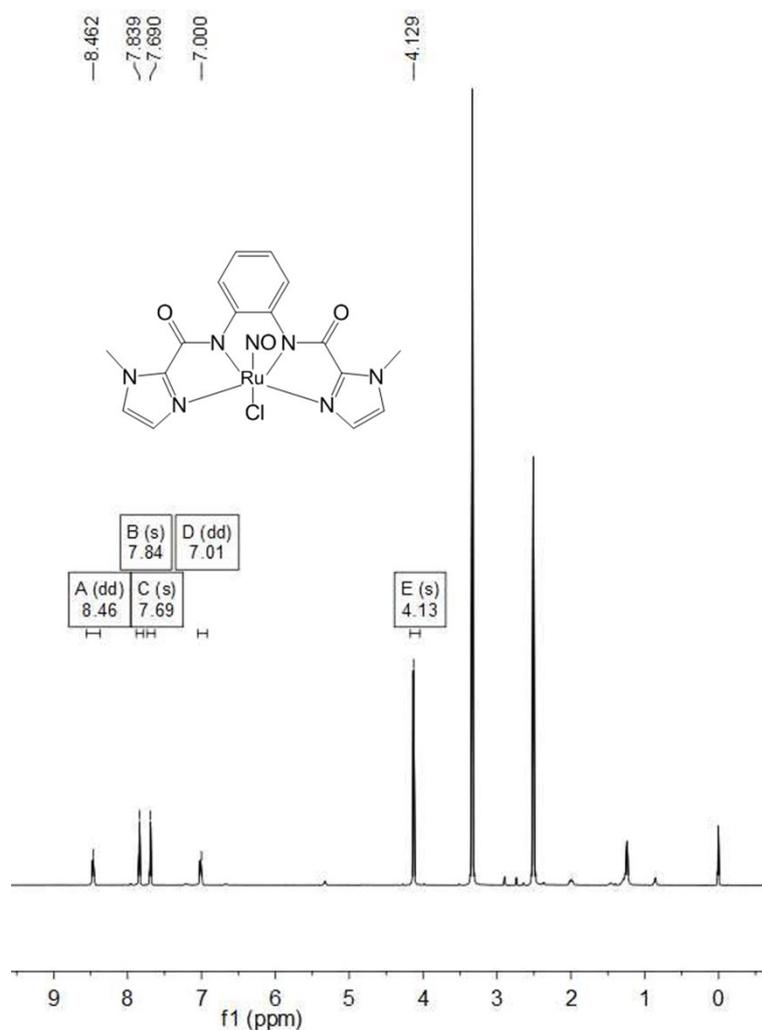
**Figure. S 3** Differential thermal analysis of hydrated nitrosotrichlorosilane



**Figure. S 4** FTIR of N,N'-(1,2-Phenylene)bis(1-methyl-1H-imidazole-2-carboxamide) ligand



**Figure. S 5**  $^1\text{H}$  NMR spectra of  $N,N'$ -(1,2-Phenylene)bis(1-methyl-1H-imidazole-2-carboxamide) ligand



**Figure. S 6**  $^1\text{H}$  NMR spectra of nitrosyl complex  $[(3)\text{Ru}(\text{NO})(\text{Cl})]$

**Table S 1** The table shows the main fluorescence bands corresponding to the fluorescence spectrum. (The enhancement multiples of fluorescence is integrated in this article.)

UCNPs	wavelength			
	345 nm	365 nm	451 nm	475 nm
$\text{NaYF}_4:\text{Yb/Tm}$	0.05087	0.05087	0.10173	1.32255
$\text{NaYF}_4:\text{Ca/Yb/Tm}$	1.93296	9.10524	13.02203	302.15167
$\text{NaYF}_4:\text{Ca/Yb/Tm@NaGdF}$	9.25785	53.86846	82.96455	722.21375

**Table S 2** Changes in pure NaYF<sub>4</sub>, 30% Ca doped-NaYF<sub>4</sub>, 30% Ca doped-NaYF<sub>4</sub> @ NaGdF<sub>4</sub> up-conversion luminous intensity (<sup>1</sup>G<sub>4</sub> → <sup>3</sup>H<sub>6</sub>) after integration.

wavelength	
UCNPs	475 nm( <sup>1</sup> G <sub>4</sub> → <sup>3</sup> H <sub>6</sub> )
NaYF <sub>4</sub> : Yb/Tm	37.185
NaYF <sub>4</sub> :Ca/Yb/Tm	4501.98795
NaYF <sub>4</sub> :Ca/Yb/Tm@NaGdF <sub>4</sub>	11265.19575