

## Supporting Information

### Particle Attachment Crystallization Facilitates the Occlusion of Micrometer-sized *Escherichia Coli* in Calcium Carbonate Crystals with Stable Fluorescence

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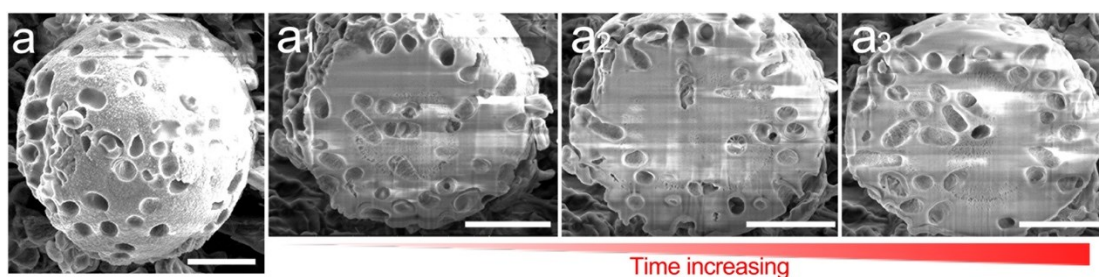


Figure S1. (a) Focused ion beam (FIB) etching of a single *E. coli*/vaterite crystal. Scale bar is 2  $\mu\text{m}$  in a, 2.5  $\mu\text{m}$  in a1-a3.

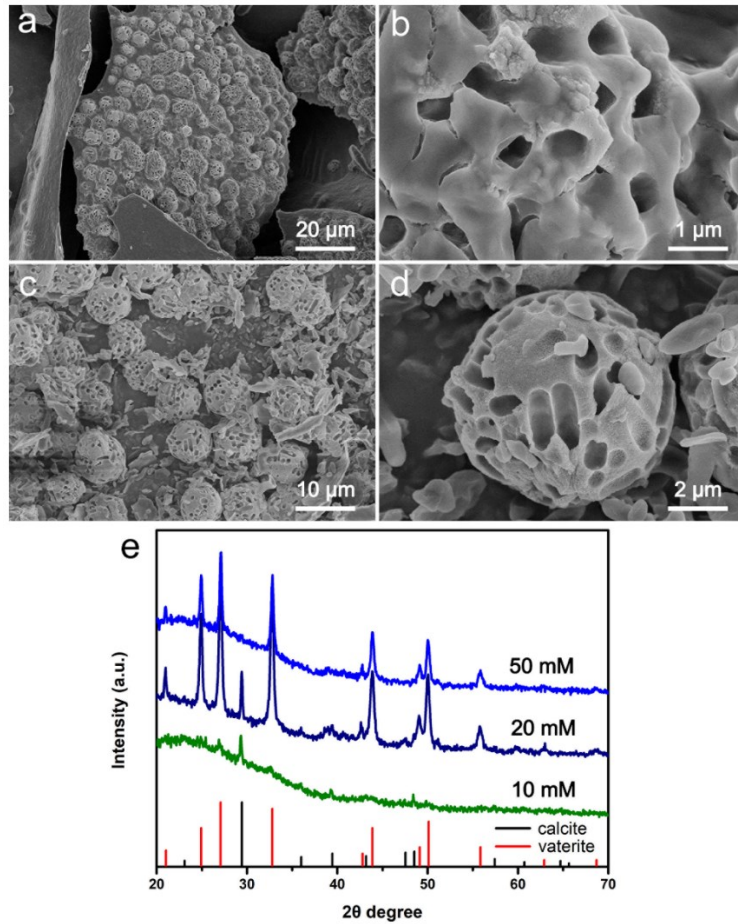


Figure S2. SEM images of composites under the concentration of (a-b) 10 mM and (c-d) 20 mM. (e) XRD patterns of composites with different concentrations.

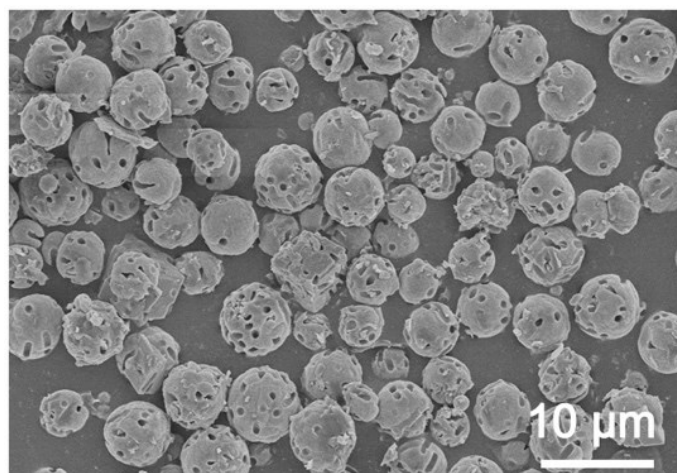


Figure S3. SEM image of *E. coli*/vaterite after mineralization for 10 min.

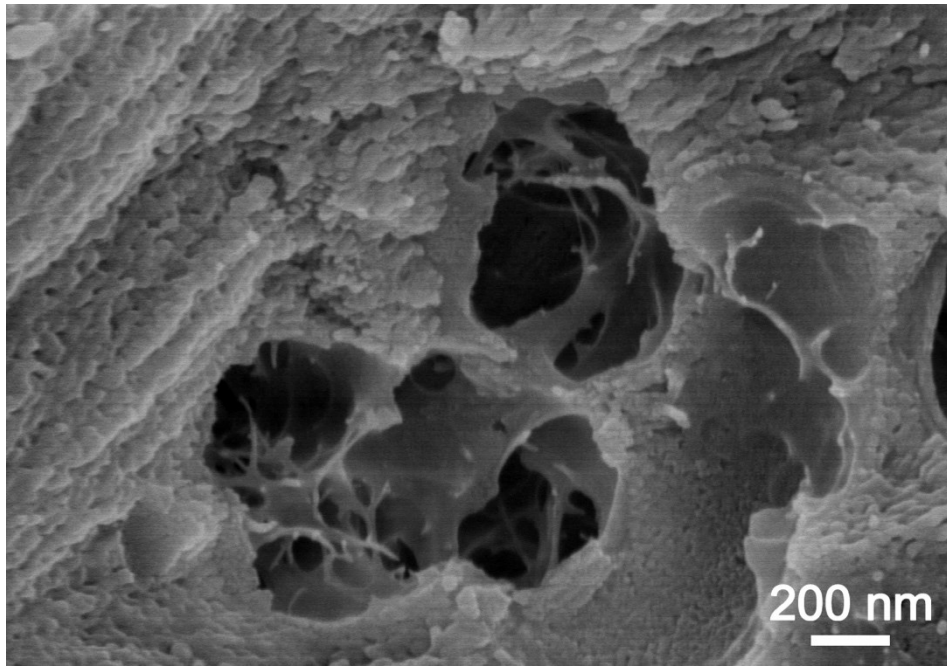


Figure S4. SEM image of fiber-like structure within the host crystal.

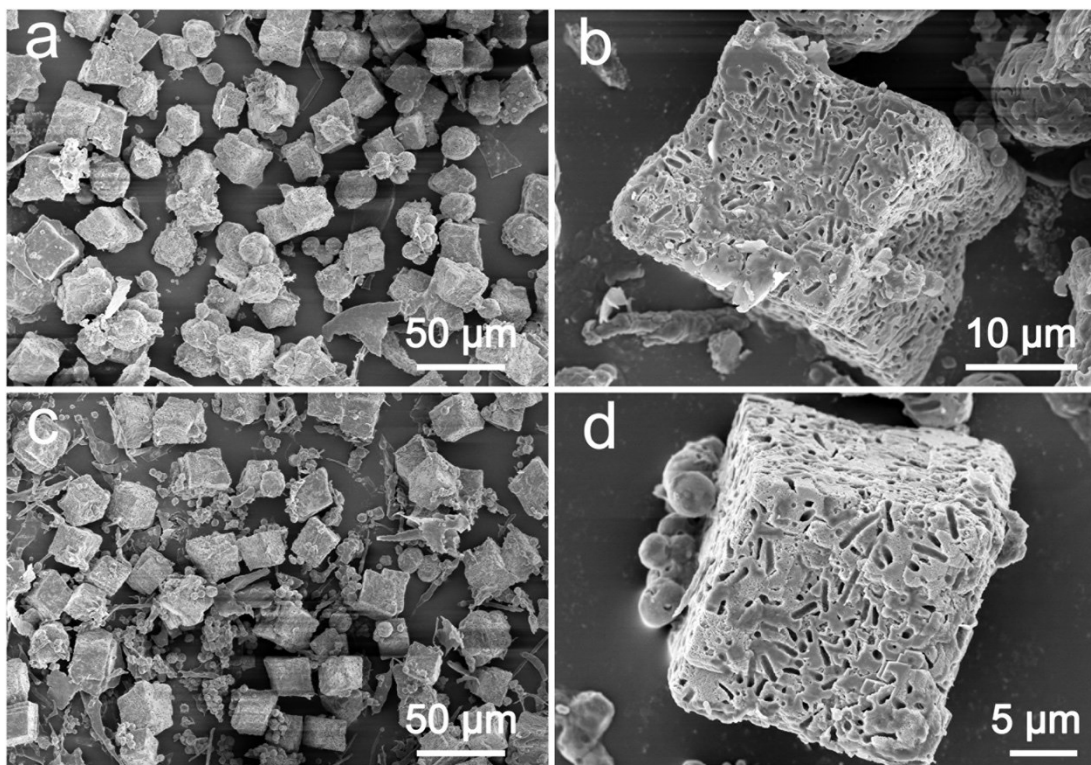


Figure S5. SEM images of *E. coli*/calcite composites after 5 min (a-b) and 10 min (c-d) of mineralization.

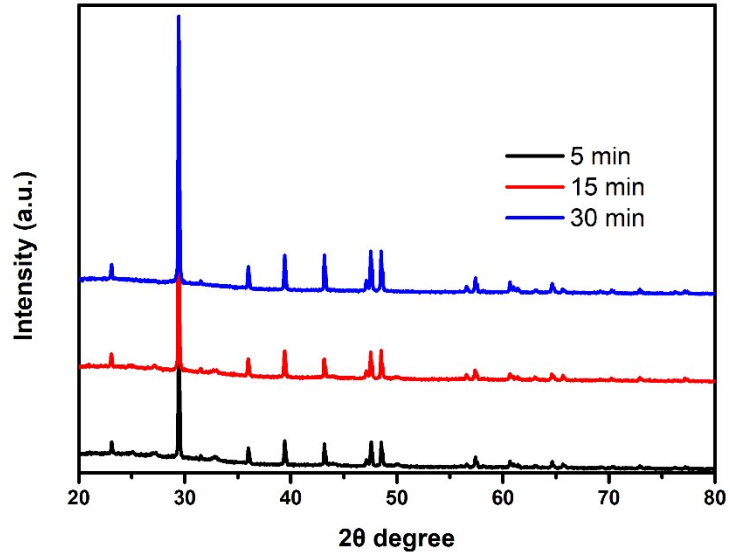


Figure S6. XRD patterns of *E. coli*/calcite composites.

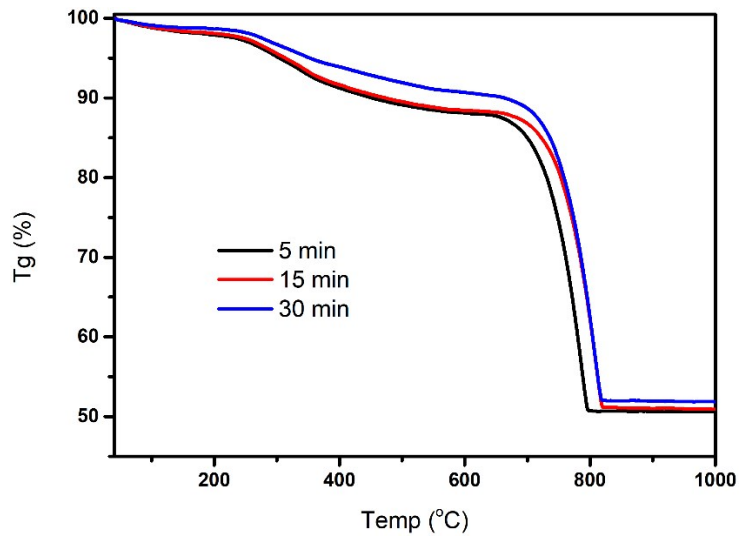


Figure S7. Thermogravimetric curves of *E. coli*/calcite composites.

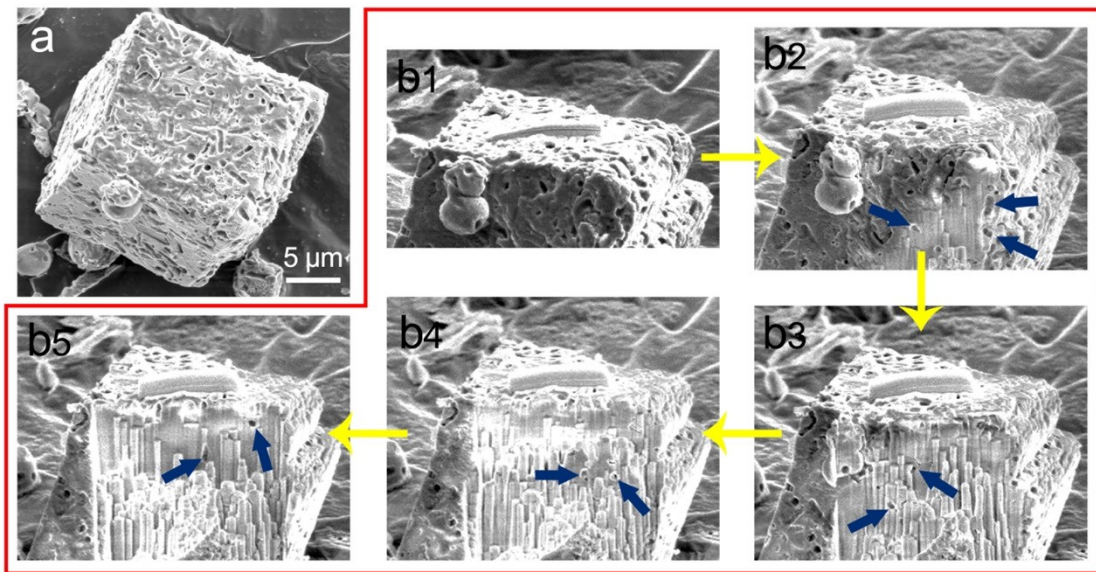


Figure S8. (a) SEM image of *E. coli*/calcite (top view). (b) Focused ion beam (FIB) etching of single calcite crystal (side view). The bacteria pores were marked by blue arrows.

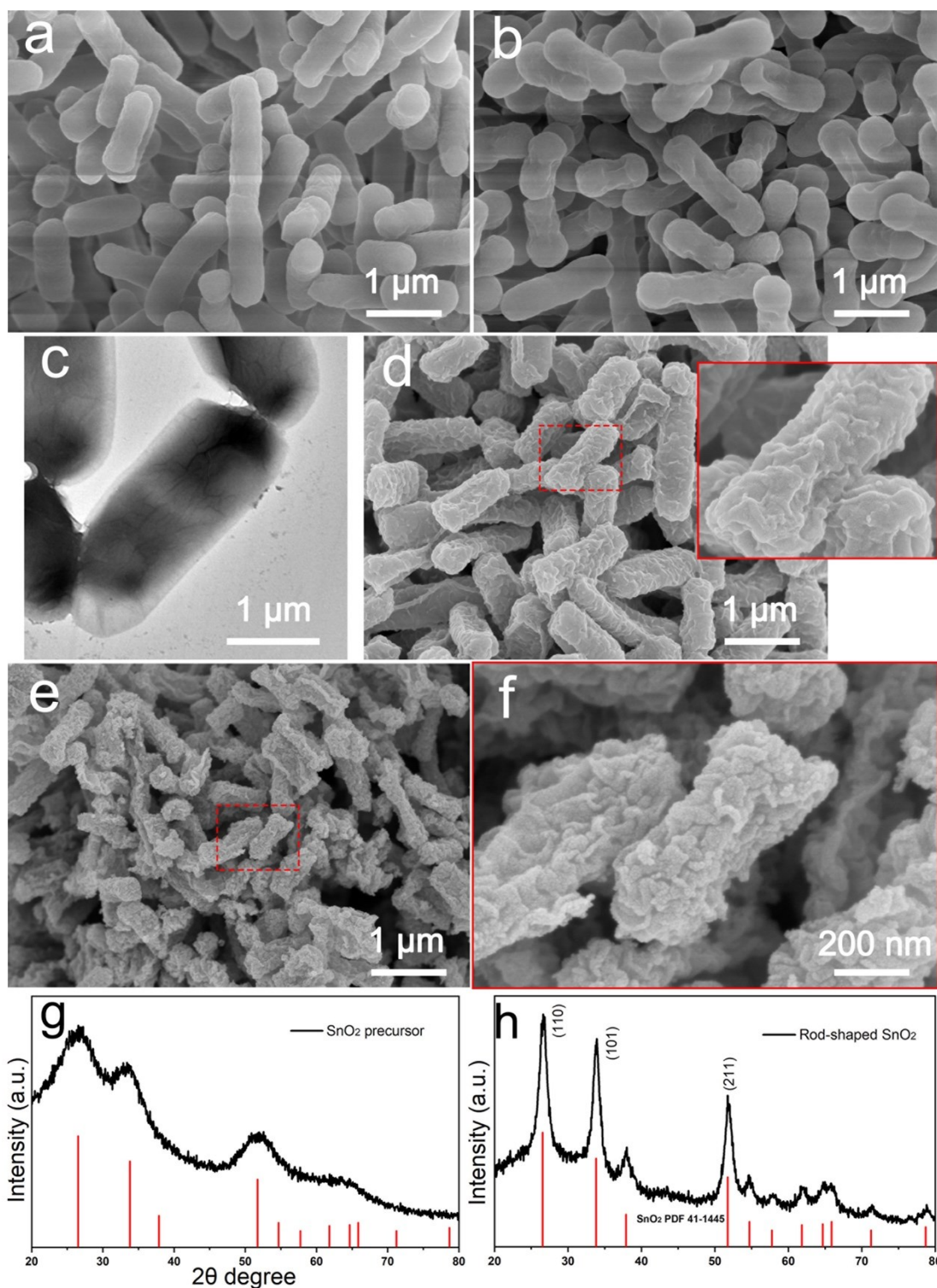


Figure S9. Morphology of various rod-shaped guest species with different surface structure and composition. (a) SEM image of native *E. coli*, (b) SEM image of 5R5 modified *E. coli*, (c) TEM image of *E. coli* with flagellum, (d) SEM image of  $\text{SnO}_2$  precursor deposited *E. coli*. (e-f) SEM images of rod-shaped  $\text{SnO}_2$ . (g) XRD pattern of  $\text{SnO}_2$  precursor, (h) XRD pattern of Rod-shaped  $\text{SnO}_2$ .

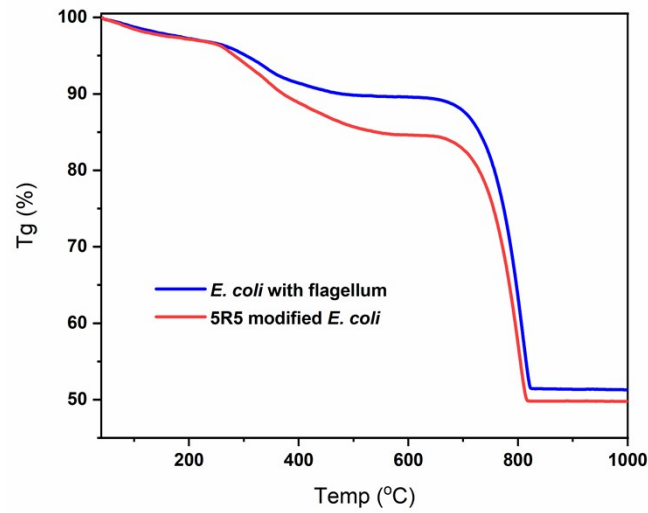


Figure S10. Thermogravimetric curves of guest/vaterite composites after 10 min of mineralization.

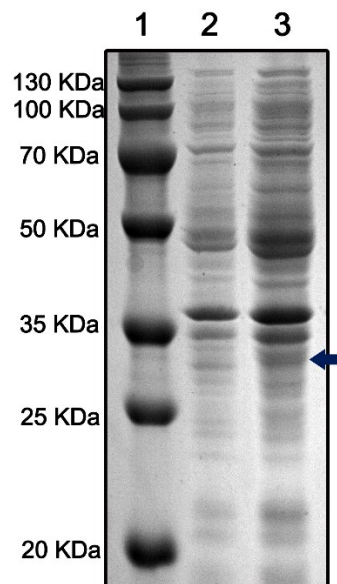


Figure S11. SDS-PAGE of GFP-modified *E. coli*. Lane 1, molecular weight marker; Lane 2, lysate of uninduced cells; Lane 3, lysate of IPTG-induced cells. The target band was marked by blue arrow.

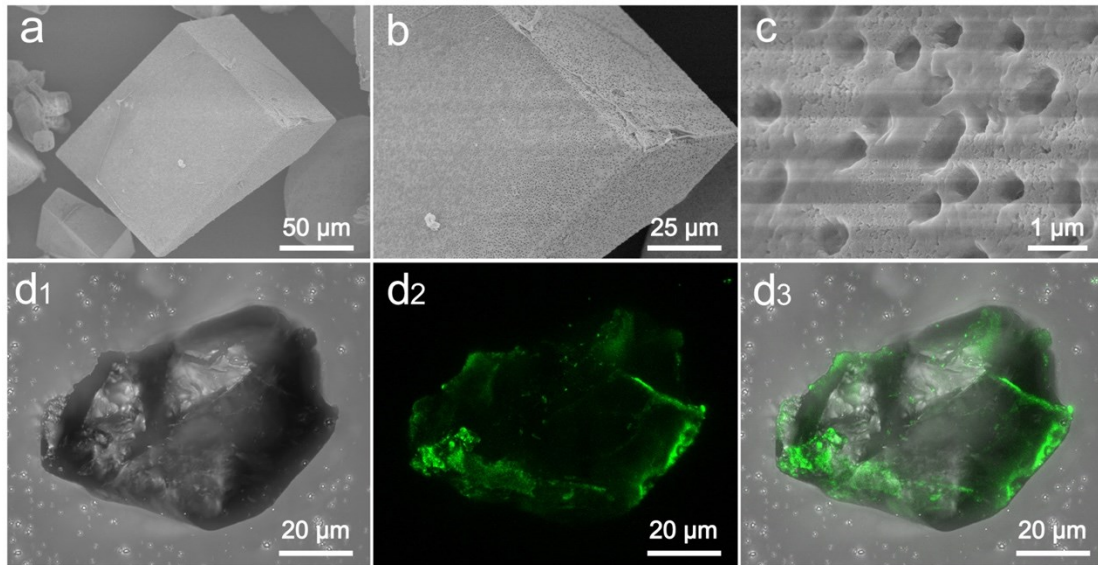


Figure S12. (a-c) SEM images of products synthesized by vapor diffusion method. (d1-d3) Confocal fluorescence images.

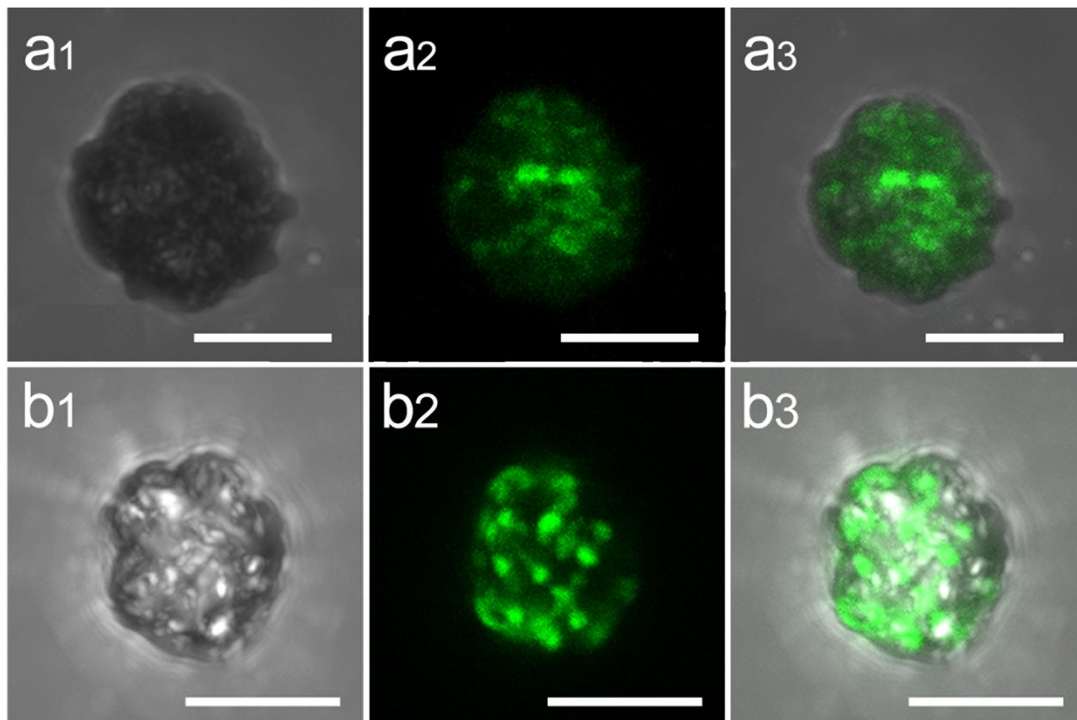


Figure S13. Confocal fluorescence images of GFP-*E. coli*/vaterite composite after (a) 3 months, (b) 6 months. Scale bar is 5 μm.



Table S1 FTIR data of  $\nu_2$  and  $\nu_4$  in *E. coli*/vaterite composites.

	$\nu_2$ (876 $\text{cm}^{-1}$ )	$\nu_4$ (745 $\text{cm}^{-1}$ )	$\nu_4$ (713 $\text{cm}^{-1}$ )
Transmittance (%)	45.34	86.6	93.19
Absorbance	0.343	0.0625	0.0306

Table S2 The mass loss during different temperature region.

	40-200 °C	200-600 °C	600-1000 °C	Total loss
<i>E. coli</i> /ACC	18.2 %	12.9 %	28.8 %	59.9 %
<i>E. coli</i> /vaterite	3.8 %	16.0 %	34.0 %	53.8 %

Table S3. The multi-exponential lifetime ( $T_i$ ) and preexponential ( $\alpha_i$ ) for lifetime of GFP-*E. coli*/CaCO<sub>3</sub> composites.

	$\alpha_1$	$T_1$	$\alpha_2$	$T_2$
GFP- <i>E. coli</i> /vaterite	3.435	1.222	0.0747	19.029
GFP- <i>E. coli</i> /calcite	3.630	1.479	0.0035	-0.0017