Electronic Supplementary Material (ESI) for Journal of Materials Chemistry B. This journal is © The Royal Society of Chemistry 2021

Supplementary information

Tumor Microenvironment-responsive multifunctional nanoplatform Based on MnFe₂O₄-PEG for Enhanced Magnetic Resonance Imaging-guided Hypoxic Cancer Radiotherapy

Zhenhu He, a,b Haixiong Yan, a,b Wenbin Zeng, d Kai Yang, e Pengfei Rong* a,b,c

- a. Department of Radiology, the Third Xiangya Hospital, Central South University, Changsha, Hunan 410013, China.
- b. Molecular Imaging Research Center, Central South University, Changsha, Hunan 410013, China.
- c. Key Laboratory of Biological Nanotechnology of National Health Commission, Changsha, Hunan 410008, China
- d. Xiangya School of Pharmaceutical Sciences, Central South University, Changsha, Hunan 410013, China.
- e. School of Radiation Medicine and Protection and School for Radiological and Interdisciplinary Sciences (RAD-X), Collaborative Innovation Center of Radiation Medicine of Jiangsu Higher Education Institutions Medical College of Soochow University, Suzhou, Jiangsu 215123, China.

Corresponding author E-mail: rongpengfei66@163.com

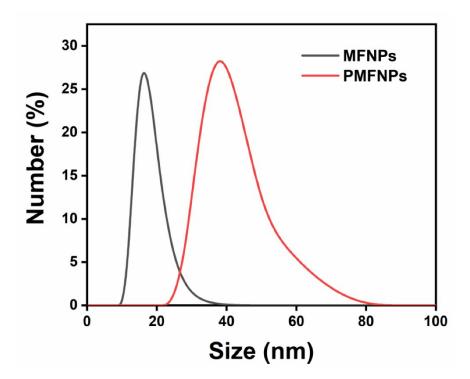


Figure S1. Dynamic light scattering (DLS) data of MnFe₂O₄ in n-Hexane and MnFe₂O₄-PEG in deionized water.

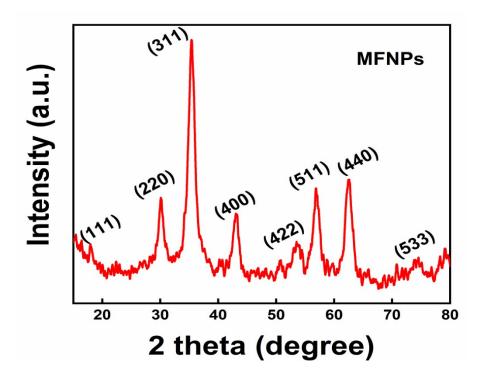


Figure S2. X-ray powder diffraction (XRD) patterns of MnFe₂O₄.

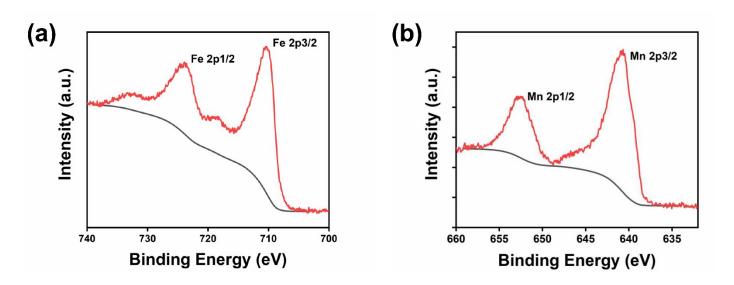


Figure S3. X-ray photoelectron spectroscopy (XPS) of MnFe₂O_{4:} (a) Fe 2p. (b) Mn 2p. The Fe 2p3/2 peak observed at 711.3 eV, and Mn 2p3/2 peak observed at 640.9 eV correspond to Fe³⁺ and Mn²⁺ species, respectively.

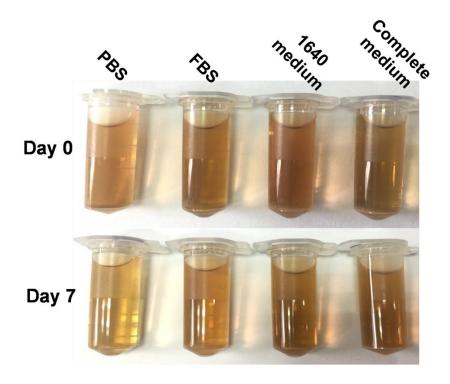


Figure S4. The stability of MnFe₂O₄-PEG in PBS, FBS, 1640 medium, and complete medium. The photos were taken on day 0 and day 7, respectively.

$$Mn^{2+} + H_2O_2 \rightarrow 2H^+ + MnO_2$$
 (1)

$$MnO_2+H_2O_2+2H^+ \rightarrow Mn^{2+}+2H_2O+O_2 \uparrow$$
 (2)

$$Fe^{3+} + H_2O_2 \rightarrow Fe^{2+} + HOO \cdot + H^+$$
 (3)

$$Fe^{3+} + HOO \cdot \rightarrow Fe^{2+} + H^+ + O_2 \uparrow \tag{4}$$

$$Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + OH^- + OH^-$$
 (5)

Figure S5. Reaction equations (1) and (2): Mechanism of catalyzing hydrogen peroxide to produce oxygen by Mn²⁺. Reaction equations (3), (4), and (5): Mechanism of catalyzing hydrogen peroxide to produce hydroxyl radical by Fe³⁺.

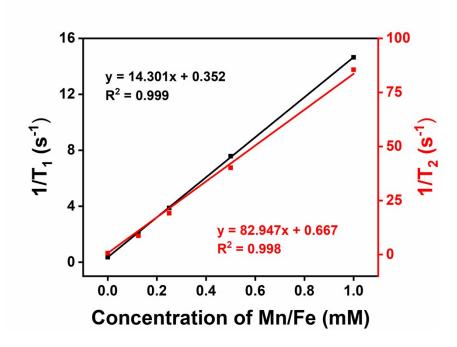


Figure S6. T_1 relaxation rate and T_2 relaxation rate of MnFe₂O₄-PEG nanoparticles measured as the function of Mn and Fe concentrations, respectively. $r_1 = 14.301 \text{ mM}^{-1}\text{s}^{-1}$, $r_2 = 82.947 \text{ mM}^{-1}\text{s}^{-1}$. Scanned by a Bruker Minispec analyzer (60 MHz).

$$Fe^{3+} + GSH \rightarrow Fe^{2+} + GSSG \tag{1}$$

$$Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + OH^- + OH^-$$
 (2)

$$MnO_2 + 2 GSH + 2 H^+ \rightarrow Mn^{2+} + GSSG + 2 H_2O$$
 (3) ²

$$Mn^{2+} + H_2O_2 \rightarrow MnO_2 + 2 H^+$$
 (4)

Figure S7. Reaction equations of GSH consumption.

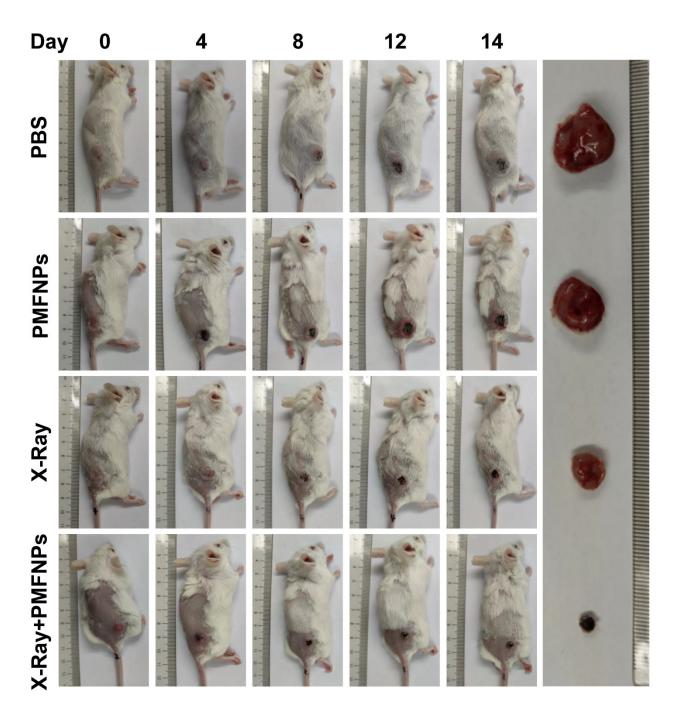


Figure S8. Digital photograph of mice bearing tumor. The tumor in the X-Ray + PMFNPs group has been necrotic and scabby.

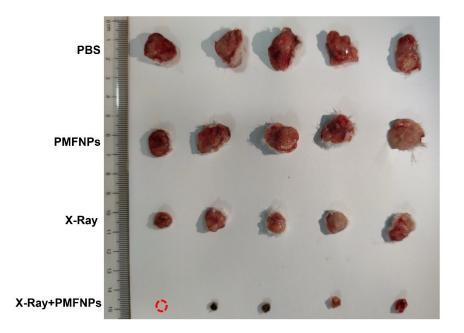


Figure S9. Digital photograph of the tumors.

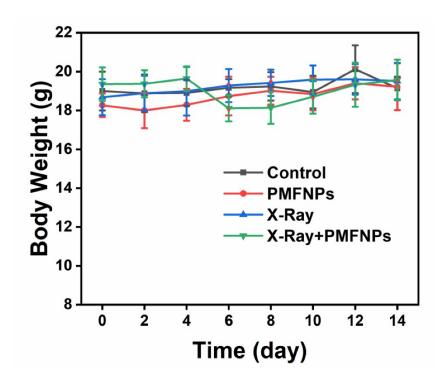


Figure S10. The body weight curves of mice after various treatments during the observation period (n = 5).

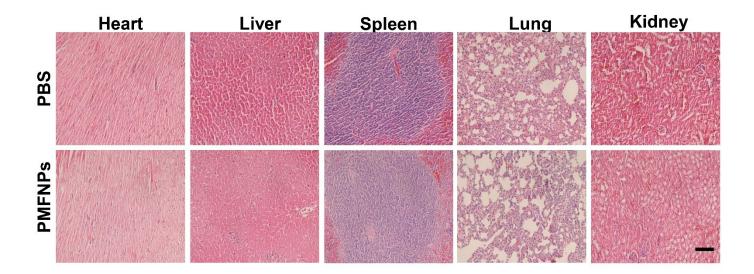


Figure S11. Hematoxylin-eosin (H&E) stainings images of main organs from different treatment groups (scale bar:100 μm).

Reference

- 1. X. Wan, L. Song, W. Pan, H. Zhong, N. Li and B. Tang, ACS Nano, 2020, 14, 11017-11028.
- 2. Z. Z. Dong, L. Lu, C. N. Ko, C. Yang, S. Li, M. Y. Lee, C. H. Leung and D. L. Ma, *Nanoscale*, 2017, 9, 4677-4682.