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## Supporting Information for

## Mixed-dimensional self-assembly organic-inorganic perovskite microcrystals for stable and efficient photodetectors

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**Fig. S1** (a) Schematic of the synthesis procedure for the  $(PEA)_2(MA)_{n-1}Pb_nBr_{3n+1}$  (n = 1, 3, 5,  $\infty$ ) perovskite microcrystals by an anti-solvent vapor-assisted method (ASV). The entire setup was sealed with tin foil and stood for 24 h at room temperature in air. (b), (c) Top-view SEM images and (d), (e) the corresponding size distributions of the cubes in the  $(PEA)_2(MA)_2Pb_3Br_{10}$  and  $(PEA)_2(MA)_4Pb_5Br_{16}$  perovskite microcrystals, respectively.



**Fig. S2** (a), (b) Optical microscope images, (c), (d) fluorescence microscope images, and (e), (f) top-view SEM images of the (PEA)<sub>2</sub>PbBr<sub>4</sub> and MAPbBr<sub>3</sub> perovskite microcrystals, respectively.



**Fig. S3** (a) SEM image, (b) EDS spectrum, and (c)-(f) the corresponding EDS elemental mappings of the (PEA)<sub>2</sub>PbBr<sub>4</sub> microcrystals.



Fig. S4 (a) SEM image, (b) EDS spectrum, and (c)-(f) the corresponding EDS elemental mappings of the  $(PEA)_2(MA)_2Pb_3Br_{10}$  perovskite microcrystals.



Fig. S5 (a) SEM image, (b) EDS spectrum, and (c)-(f) the corresponding EDS elemental mappings of the  $(PEA)_2(MA)_4Pb_5Br_{16}$  perovskite microcrystals.



**Fig. S6** (a) SEM image, (b) EDS spectrum, and (c)-(f) the corresponding EDS elemental mappings of the MAPbBr<sub>3</sub> microcrystals.



**Fig. S7** (a) Optical microscope image, (b) micro-area absorption spectrum, and (c) micro-area PL spectrum of the (PEA)<sub>2</sub>(MA)<sub>4</sub>Pb<sub>5</sub>Br<sub>16</sub> perovskite microcrystal photodetector. The center position in (a) surrounded by six bright spots is the test area.



**Fig. S8** (a), (b), (c) Optical microscope images and (d), (e), (f) fluorescence microscope images of the photodetectors based on the (PEA)<sub>2</sub>PbBr<sub>4</sub>, (PEA)<sub>2</sub>(MA)<sub>2</sub>Pb<sub>3</sub>Br<sub>10</sub>, and MAPbBr<sub>3</sub> perovskite microcrystals, respectively.



**Fig. S9** (a), (b), (c) Optical microscope images, (d), (e), (f) micro-area absorption spectra, and (g), (h), and (i) micro-area PL spectra of the (PEA)<sub>2</sub>PbBr<sub>4</sub>, (PEA)<sub>2</sub>(MA)<sub>2</sub>Pb<sub>3</sub>Br<sub>10</sub>, and MAPbBr<sub>3</sub> perovskite microcrystal photodetectors, respectively. The center positions surrounded by six bright spots in (a), (b), (c) are the test areas. The spots in (c) are invisible since the MAPbBr<sub>3</sub> microcube is much thicker.



**Fig. S10** Device stability of the mixed-dimensional  $(PEA)_2(MA)_4Pb_5Br_{16}$  perovskite microcrystal photodetector. The following characteristics were investigated after the photodetector was placed in an atmospheric environment with a humidity of 60% for 8 days. (a) *I-V* characteristics of the photodetector in the dark and under 405-nm light illumination at various power intensities. (b) Photocurrent as a function of light intensity at 15 V. (c) Variations of responsivity and gain with light intensity at 15 V. (d) Variations of responsivity and gain with voltage under a light intensity of 2.47 nW/cm<sup>2</sup>. (e) Specific detectivity as a function of light intensity at 15 V. (f) Time-resolved response of the photodetector illuminated by 405-nm light pulse at 5 s intervals at 15 V.



**Fig. S11** Device characteristics of the pristine  $(PEA)_2PbBr_4$  perovskite microcrystal photodetector and after being placed in an atmospheric environment with a humidity of 60% for 8 days. (a), (b) *I-V* characteristics in the dark and under 405-nm light illumination at various power intensities. (c), (d) Variations of photocurrent and specific detectivity with light intensity. (e), (f) Variations of responsivity and gain with voltage under a light intensity of 2.47 nW/cm<sup>2</sup>. (g), (h) Variations of responsivity and gain with light intensity at 15 V. (i), (j) Time-resolved response of the photodetector illuminated by 405-nm light pulses at 5 s intervals at 15 V.



**Fig. S12** Device characteristics of the pristine  $(PEA)_2(MA)_2Pb_3Br_{10}$  perovskite microcrystal photodetector and after being placed in an atmospheric environment with a humidity of 60% for 8 days. (a), (b) *I-V* characteristics in the dark and under 405-nm light illumination at various power intensities. (c), (d) Variations of photocurrent and specific detectivity with light intensity at 15 V. (e), (f) Variations of responsivity and gain with voltage under a light intensity of 2.47 nW/cm<sup>2</sup>. (g), (h) Variations of responsivity and gain with light intensity at 15 V. (i), (j) Time-resolved response of the photodetector illuminated by 405-nm light pulses at 5 s intervals at 15 V.



**Fig. S13** Device charateristics of the pristine MAPbBr<sub>3</sub> perovskite photodetector and after being placed in an atmospheric environment with a humidity of 60% for 8 days. (a), (b) *I-V* characteristics in the dark and under 405-nm light illumination at various power intensities. (c), (d) Variations of photocurrent and specific detectivity with light intensity at 15 V. (e), (f) Variations of responsivity and gain with voltage under a light intensity of 2.47 nW/cm<sup>2</sup>. The insets in (e), (f) show the magnified curves of responsivity and gain versus voltage after aging. (g), (h) Variations of responsivity and gain with light intensity at 15 V. (i), (j) Time-resolved response of the photodetector illuminated by 520-nm light pulses at 5 s intervals at 15 V.

n	Domovaluito	PbBr <sub>2</sub>	MABr	PEABr	
	rerovskite	concentration	concentration	concentration	
1	(PEA) <sub>2</sub> PbBr <sub>4</sub>	1 mmol/ml	0	2 mmol/ml	
		(367 mg/ml)	0	(404 mg/ml)	
3	(PEA) <sub>2</sub> (MA) <sub>2</sub> Pb <sub>3</sub> Br <sub>10</sub>	1 mmol/ml 0.67 mmol/ml		0.67 mmol/ml	
		(367 mg/ml)	(74.7 mg/ml)	(134.7 mg/ml)	
=	(PEA) <sub>2</sub> (MA) <sub>4</sub> Pb <sub>5</sub> Br <sub>16</sub>	1 mmol/ml	0.8 mmol/ml	0.4 mmol/ml	
5		(367 mg/ml) (89.6 mg/ml) (8		(80.8 mg/ml)	
ø	MAPbBr <sub>3</sub>	1 mmol/ml	1 mmol/ml	0	
		(367 mg/ml)	(112 mg/ml)		

**Table S1** Amount of each starting material for preparing the  $(PEA)_2(MA)_{n-1}Pb_nBr_{3n+1}$ (n = 1, 3, 5,  $\infty$ ) perovskite microcrystals according to stoichiometric ratios.

**Table S2** Device performance parameters of the pristine photodetectors based on the  $(PEA)_2(MA)_{n-1}Pb_nBr_{3n+1}$  (n = 1, 3, 5  $\infty$ ) perovskite microcrystals and after being placed in an atmospheric environment with a humidity of 60% for 8 days at 15 V.

n	Perovskite (days)	Rise Time	Decay Time	Responsivity (A/W)	Detectivity (Jones)	Gain
1	(PEA) <sub>2</sub> PbBr <sub>4</sub> (p)	179.5	150.9	5.70	3.50×10 <sup>13</sup>	17.43
1	$(PEA)_2PbBr_4$ (8)	185.2	120.1	11.22	5.09×10 <sup>13</sup>	34.33
3	$(PEA)_2(MA)_2Pb_3Br_{10}(p)$	149.9	369.6	351.78	2.56×10 <sup>14</sup>	1.16×10 <sup>3</sup>
3	(PEA) <sub>2</sub> (MA) <sub>2</sub> Pb <sub>3</sub> Br <sub>10</sub> (8)	129.1	188.1	620.81	5.74×10 <sup>14</sup>	1.90×10 <sup>3</sup>
5	(PEA) <sub>2</sub> (MA) <sub>4</sub> Pb <sub>5</sub> Br <sub>16</sub> (p)	358.4	589.8	182.63	2.51×10 <sup>14</sup>	5.58×10 <sup>2</sup>
5	(PEA) <sub>2</sub> (MA) <sub>4</sub> Pb <sub>5</sub> Br <sub>16</sub> (8)	66.8	66.7	364.34	4.21×10 <sup>14</sup>	1.11×10 <sup>3</sup>
8	MAPbBr <sub>3</sub> (p)	66.4	132.4	1.40×10 <sup>6</sup>	4.92×10 <sup>16</sup>	3.34×10 <sup>6</sup>
ø	MAPbBr <sub>3</sub> (8)	135.0	150.7	-	-	-

\*p represents pristine; 8 represents 8 days.