

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry C.

## Electronic Supplementary Information

### **Tert-butyl Substituted Hetero-donor TADF compounds for Efficient Solution-Processed Non-doped Blue OLEDs**

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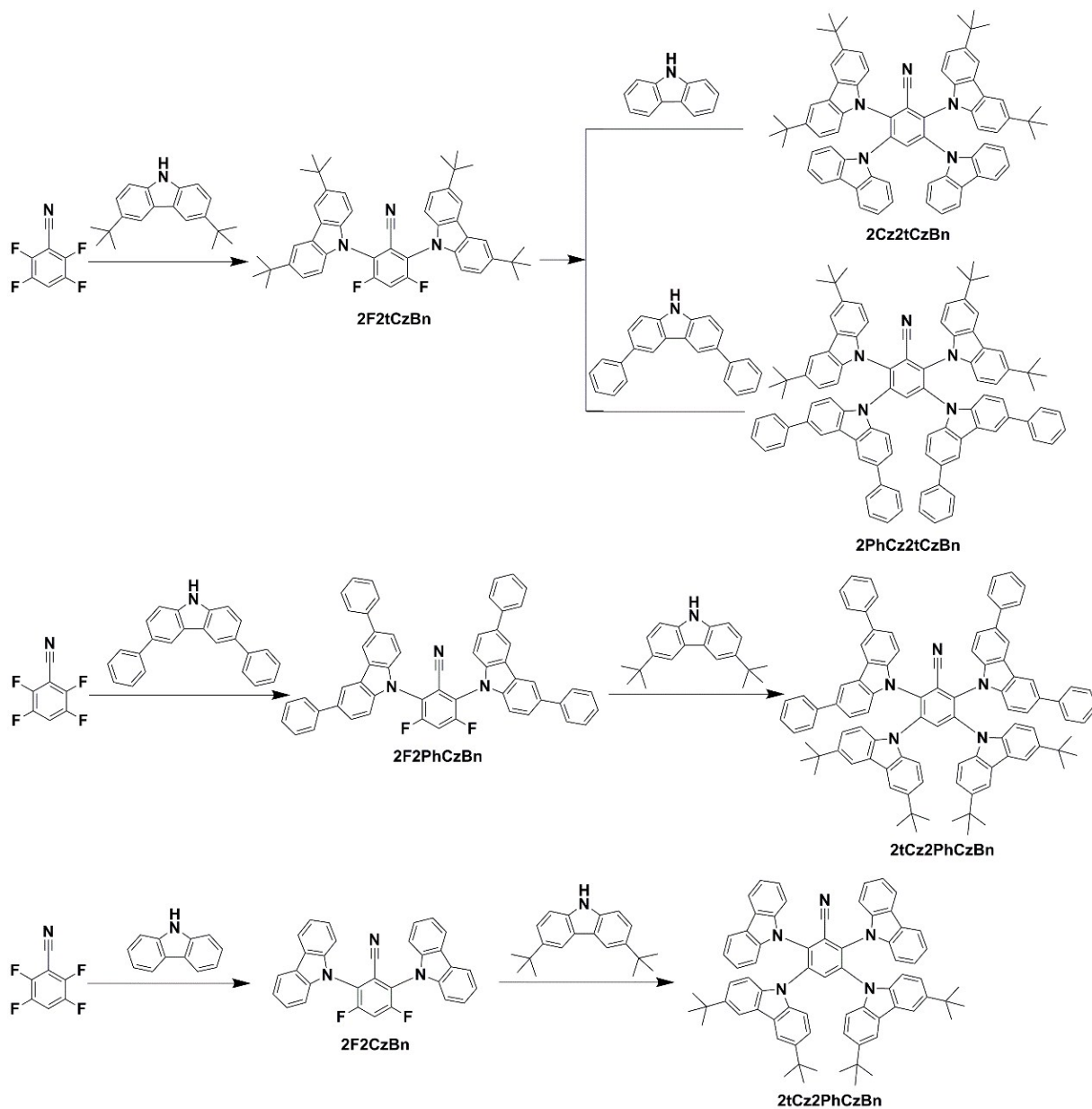
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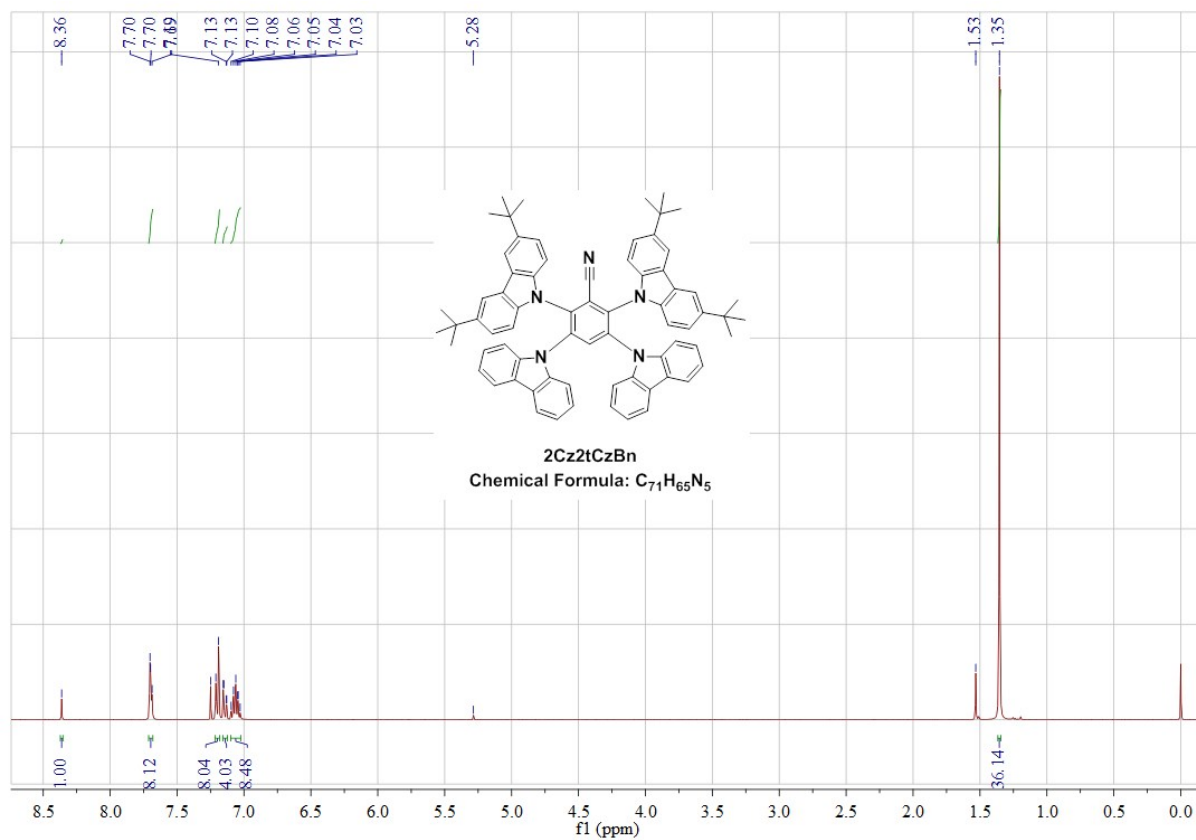
**2Cz2tCzBn:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.36 (s, 1H), 7.71-7.68 (m, 8H), 7.20 (d,  $J = 8.2$  Hz, 8H), 7.14 (dd,  $J = 8.6, 1.7$  Hz, 4H), 7.10-7.02 (m, 8H), 1.35 (s, 36H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  144.12, 139.33, 138.77, 137.46, 137.11, 135.72, 125.43, 124.50, 123.88, 123.14, 120.68, 119.86, 118.10, 116.16, 113.43, 109.59, 109.29, 34.59, 31.83. MS (APCI) calcd. for  $\text{C}_{71}\text{H}_{65}\text{N}_5$ :  $m/z = 987.52$ , found: 1005.6  $[\text{M}+18]^+$ .

**2PhCz2tCzBn:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.50 (s, 1H), 7.96 (s, 4H), 7.70 (s, 4H), 7.60 (d,  $J = 7.6$  Hz, 8H), 7.46 (t,  $J = 7.6$  Hz, 8H), 7.38-7.32 (m, 6H), 7.31 (s, 4H), 7.28 (d,  $J = 3.6$  Hz, 2H), 7.25 (s, 4H), 7.19 (dd,  $J = 8.7, 1.0$  Hz, 4H), 1.35 (s, 36H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  144.30, 141.52, 139.29, 138.82, 137.50, 136.79, 135.71, 134.44, 128.77, 127.16, 126.77, 125.12, 124.69, 124.57, 123.28, 118.37, 116.23, 109.74, 109.71, 34.62, 31.84. MS (APCI) calcd. for  $\text{C}_{95}\text{H}_{81}\text{N}_5$ :  $m/z = 1292.65$ , found: 1310.7  $[\text{M}+18]^+$ .

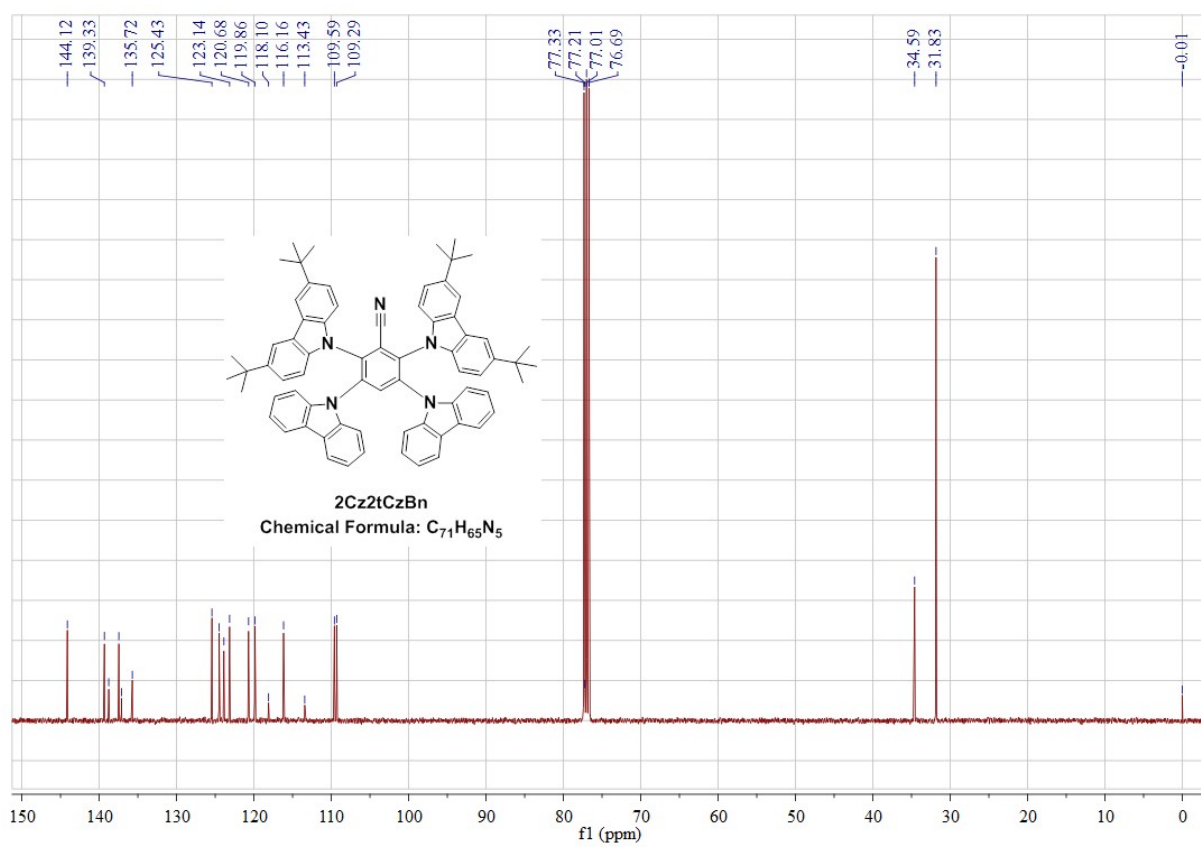
**2tCz2PhCzBn:**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.49 (s, 1H), 7.98 (s, 4H), 7.71 (d,  $J = 1.5$  Hz, 4H), 7.61 (d,  $J = 7.4$  Hz, 8H), 7.47 (t,  $J = 7.6$  Hz, 8H), 7.41-7.33 (m, 13H), 7.23 (s, 1H), 7.21 (s, 2H), 7.14 (dd,  $J = 8.6, 1.7$  Hz, 4H), 1.35 (s, 36H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  144.13, 141.63, 139.01, 137.82, 137.34, 134.74, 128.74, 127.24, 126.73, 125.26, 125.02, 124.22, 123.26, 118.58, 116.12, 110.51, 108.93, 34.61, 31.82. MS (APCI) calcd. for  $\text{C}_{95}\text{H}_{81}\text{N}_5$ :  $m/z = 1292.65$ , found: 1310.7  $[\text{M}+18]^+$ .



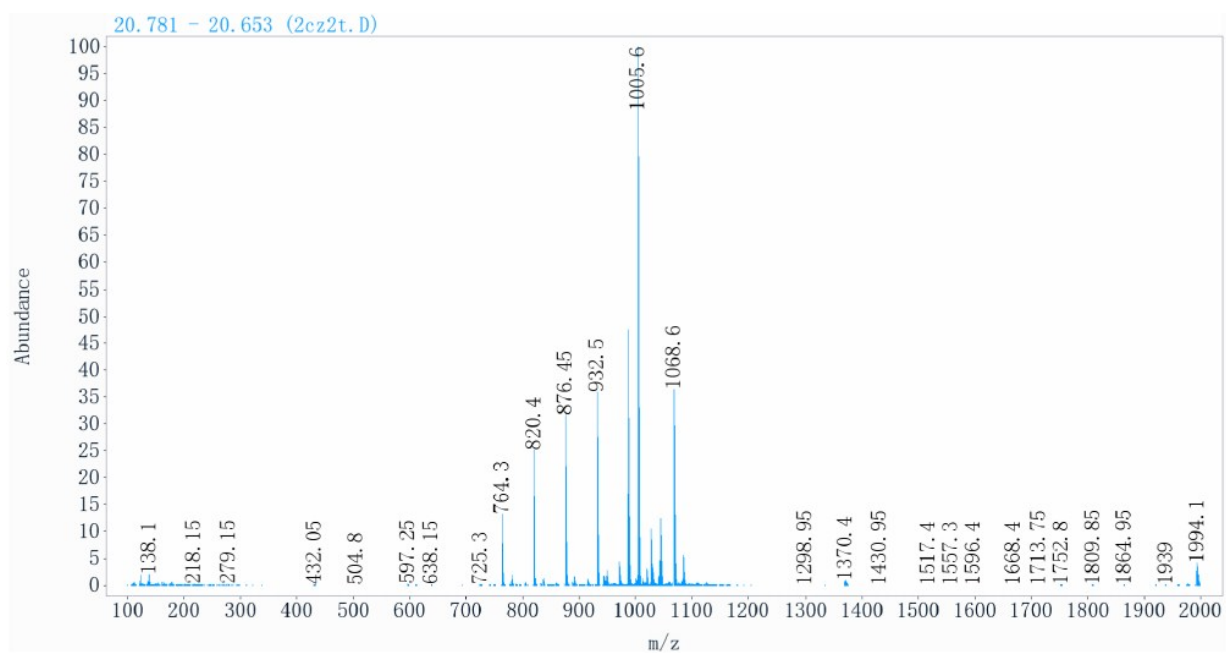
**Scheme S1.** Synthetic routes of **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn**, respectively.



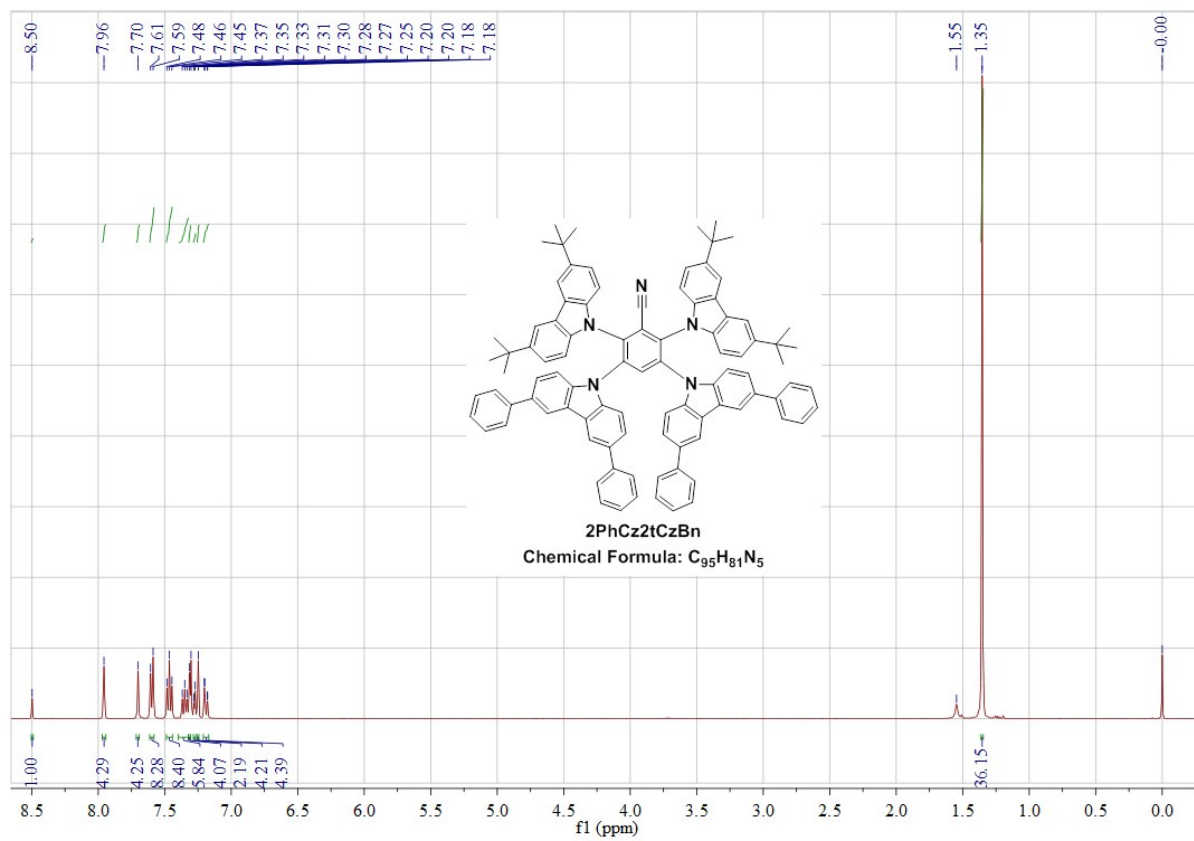
**Figure S1.** <sup>1</sup>H NMR spectrum of 2Cz2tCzBn.



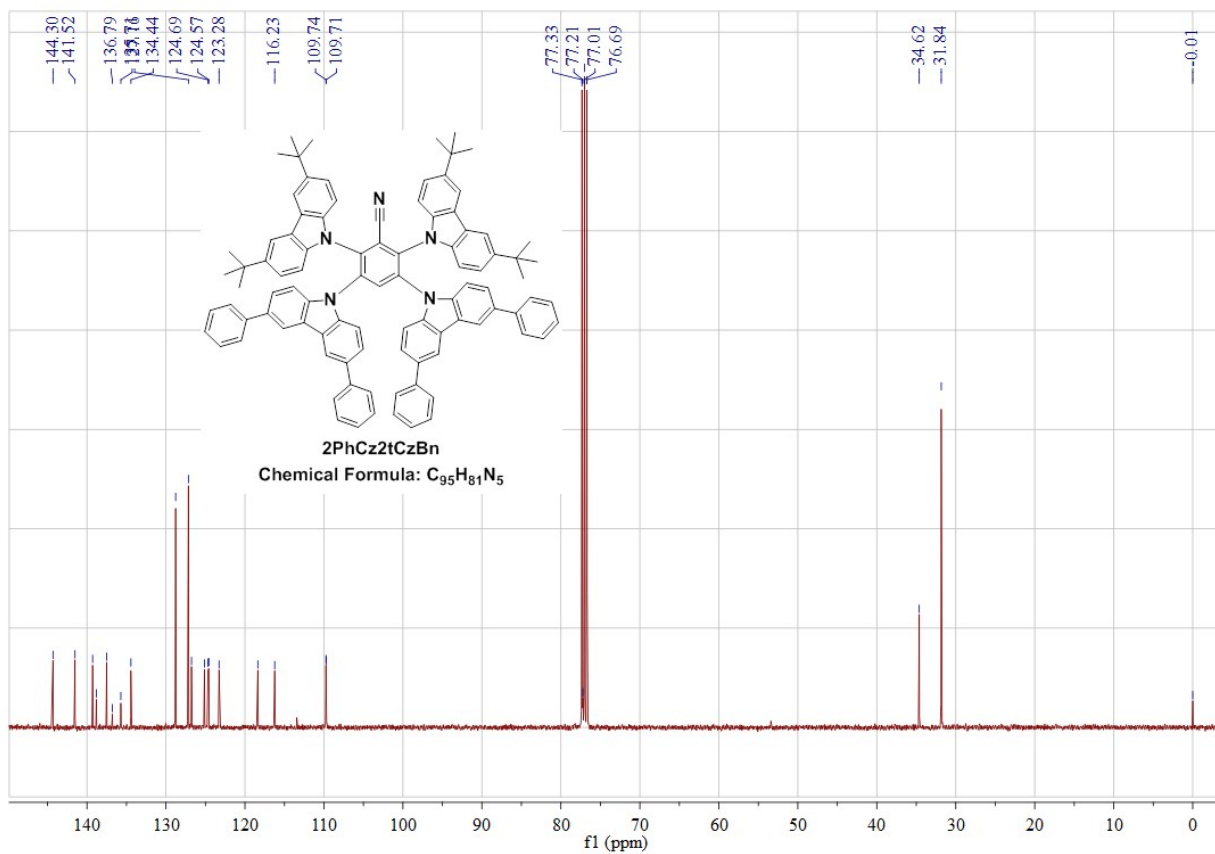
**Figure S2.**  $^{13}C$  NMR spectrum of 2Cz2tCzBn.



**Figure S3.** TOF-MS spectrum of 2Cz2tCzBn.

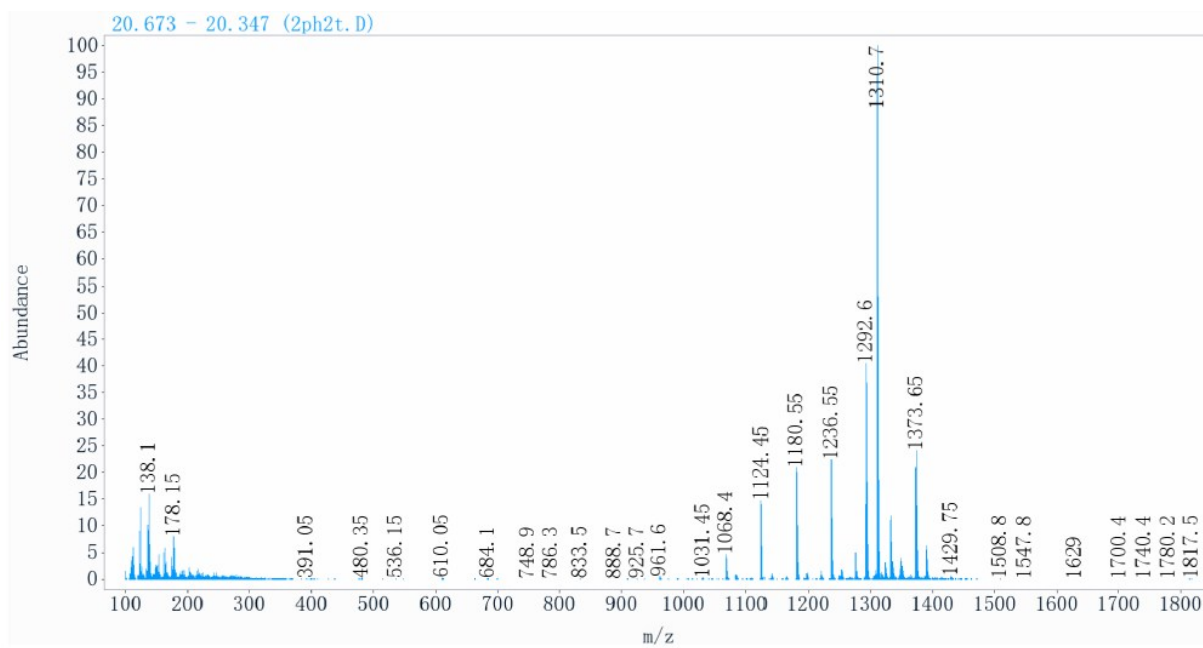


**Figure S4.**  $^1H$  NMR spectrum of 2PhCz2tCzBn.

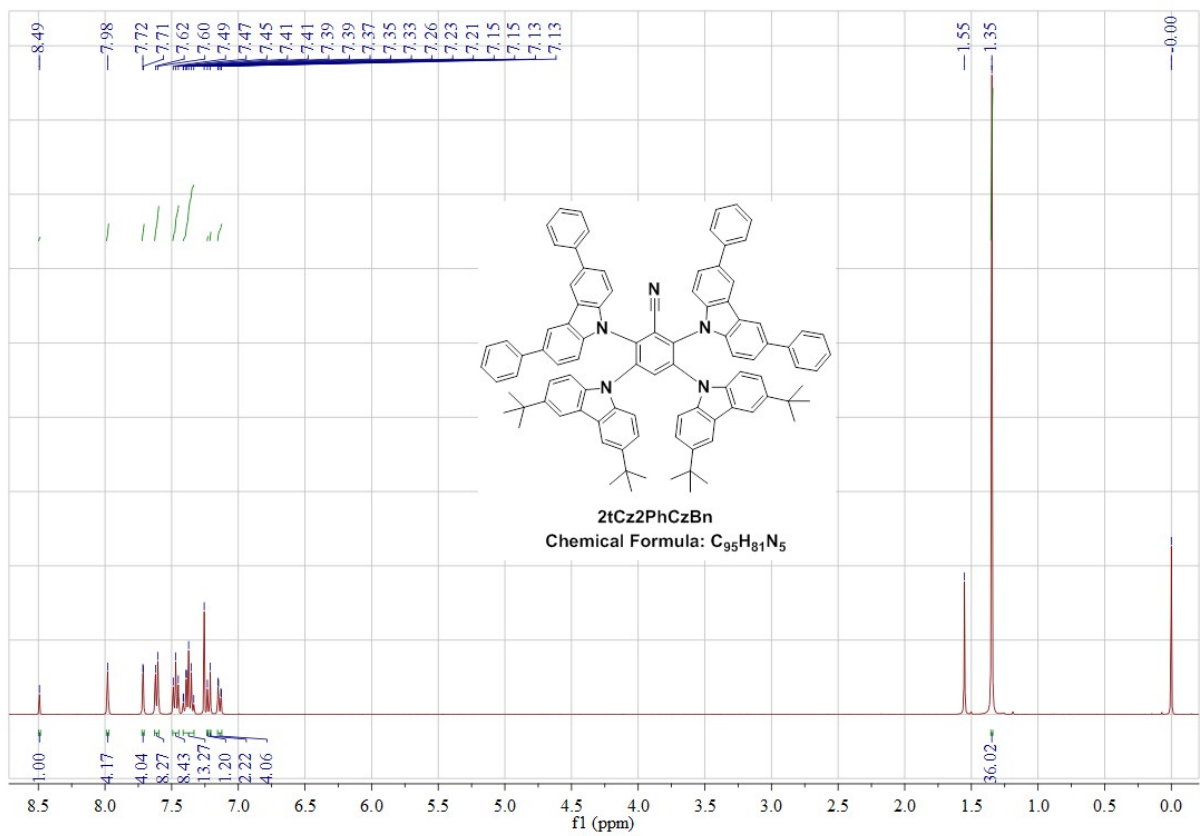


**Figure S5.** <sup>13</sup>C NMR spectrum of 2PhCz2tCzBn.

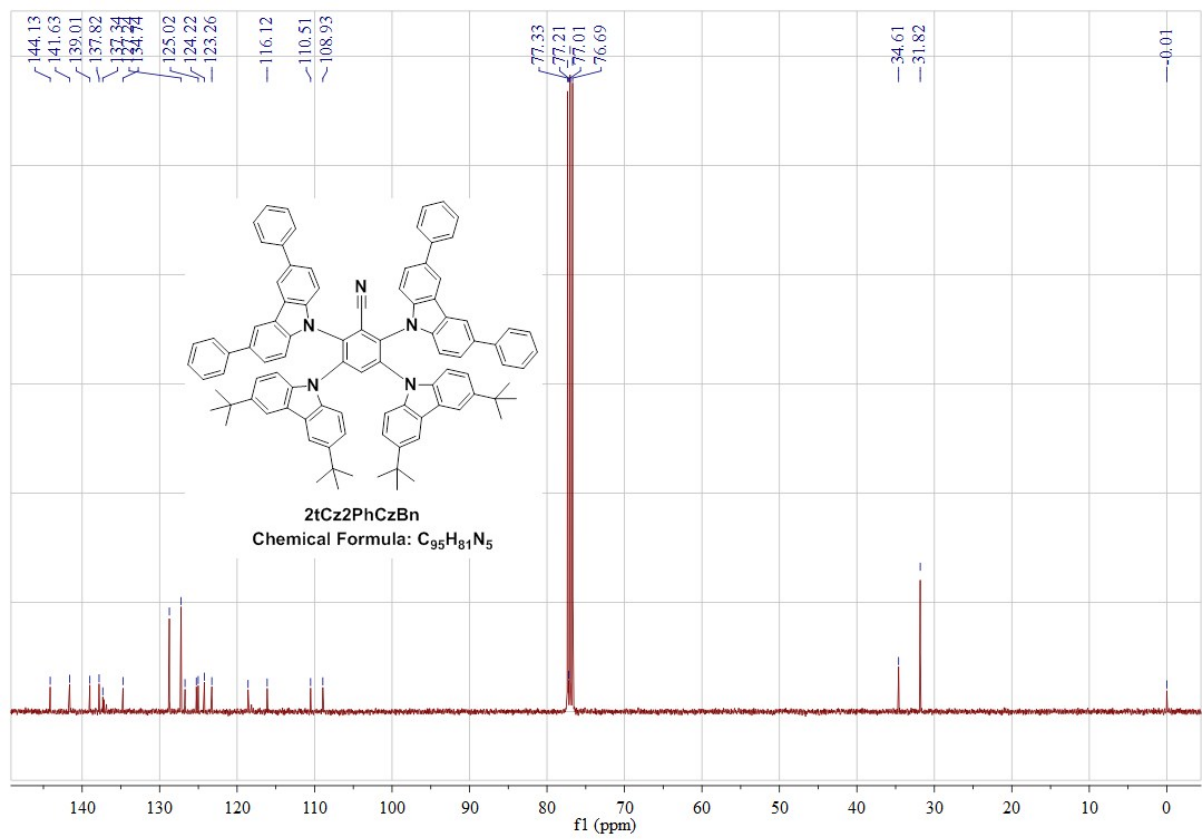




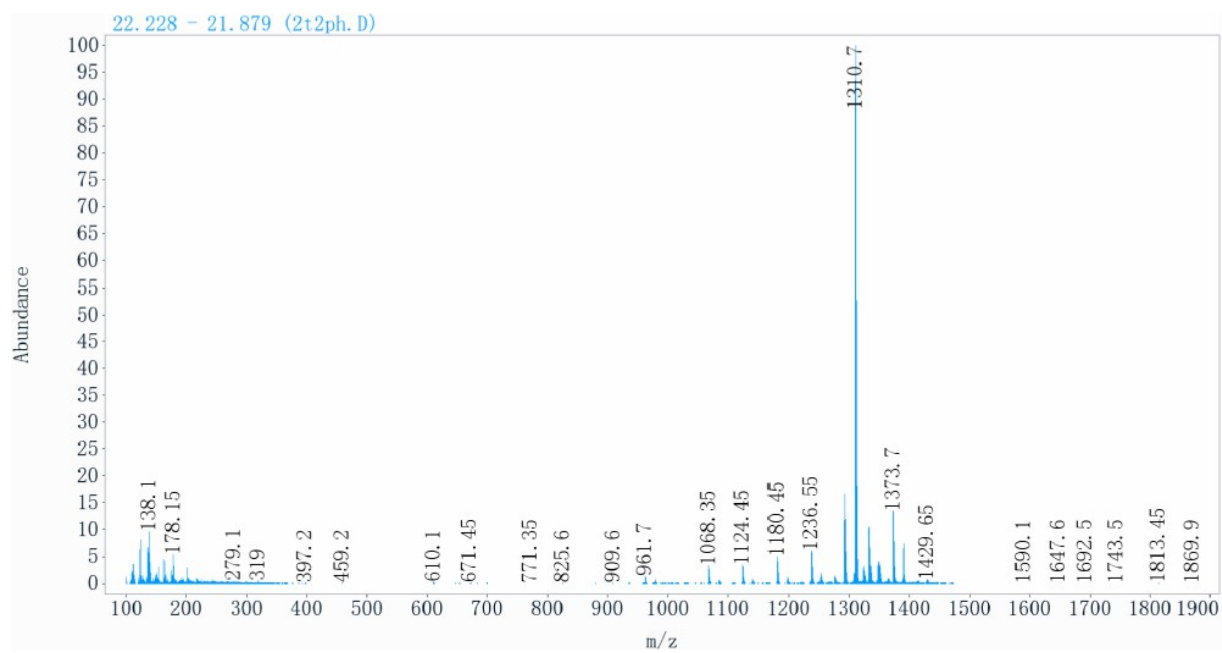
**Figure S6.** TOF-MS spectrum of 2PhCz2tCzBn.



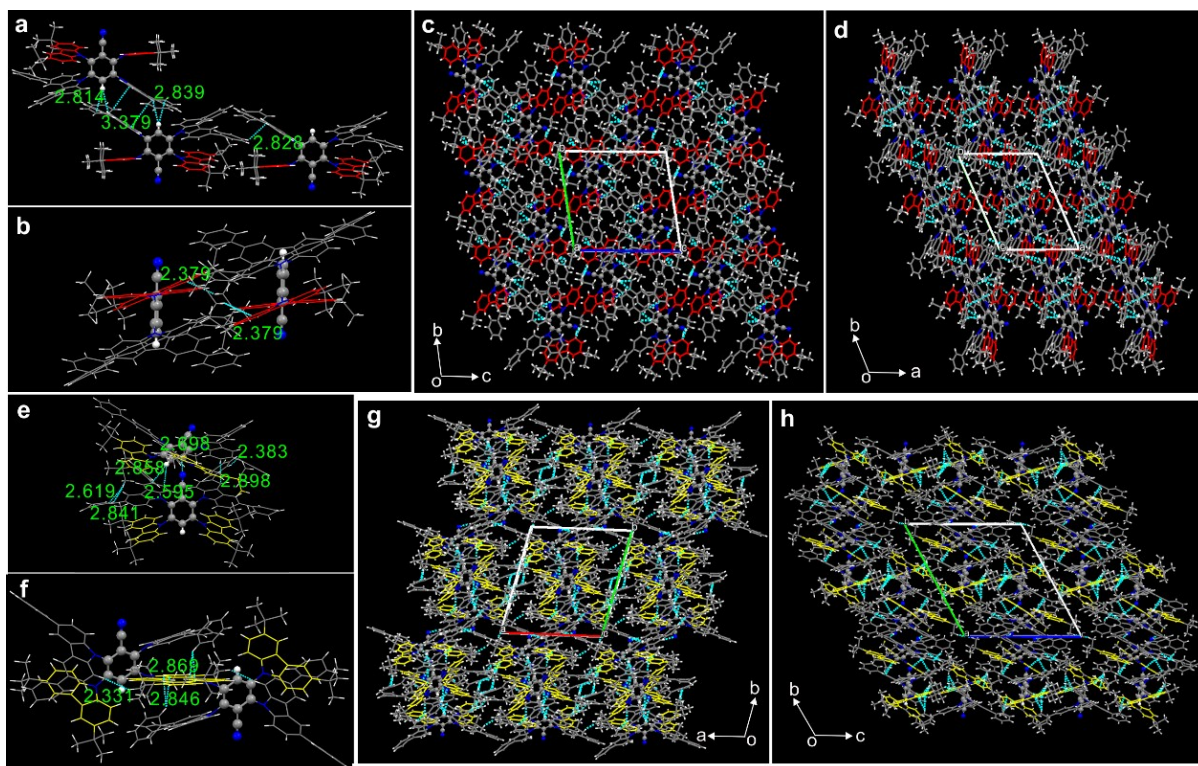
**Figure S7.**  $^1H$  NMR spectrum of 2tCz2PhCzBn.



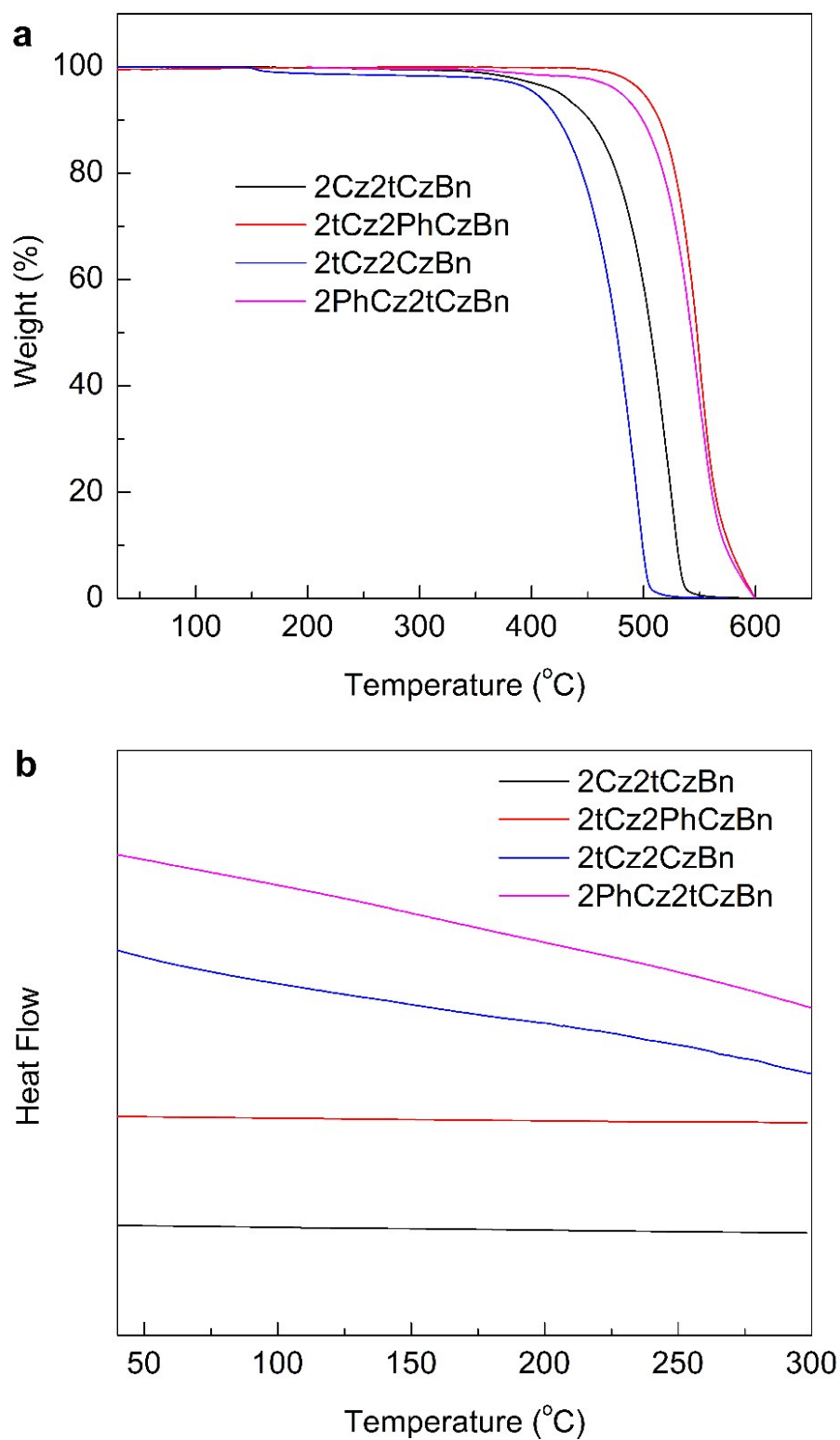
**Figure S8.**  $^{13}C$  NMR spectrum of 2tCz2PhCzBn.



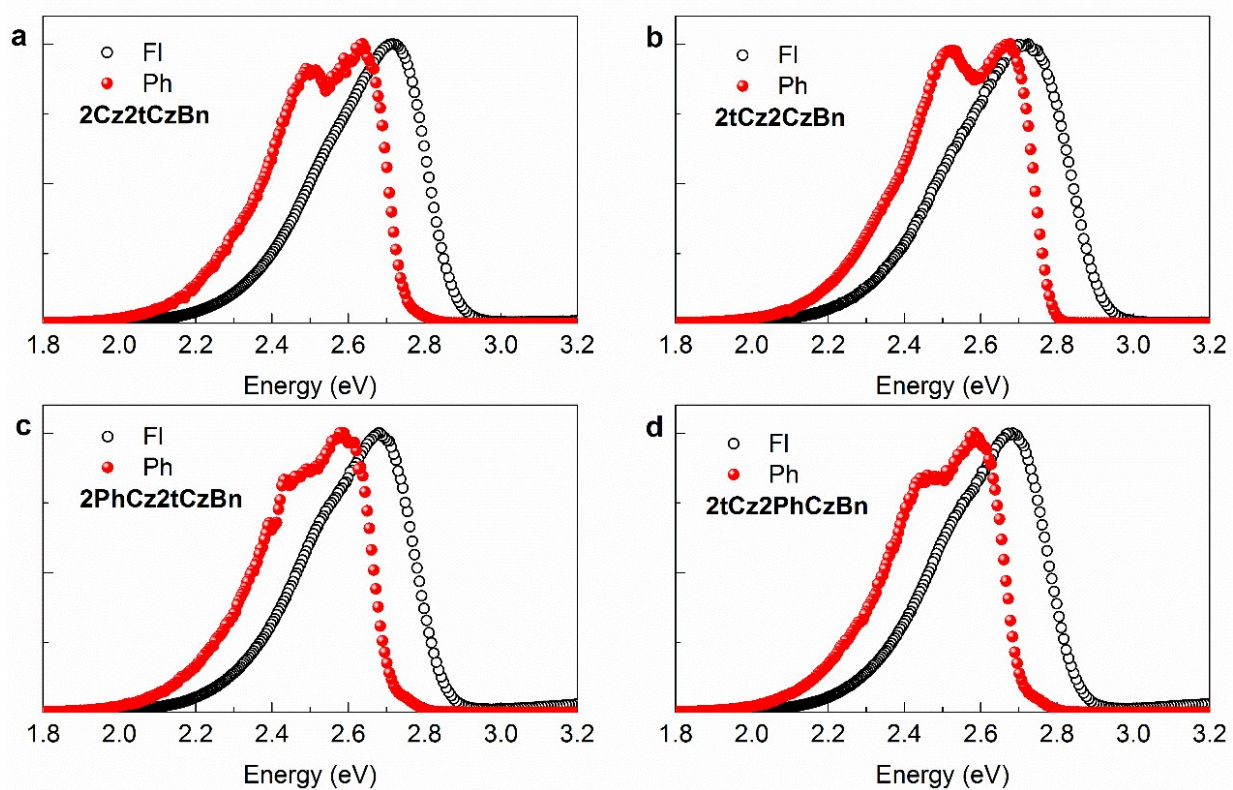
**Figure S9.** TOF-MS spectrum of 2tCz2PhCzBn.



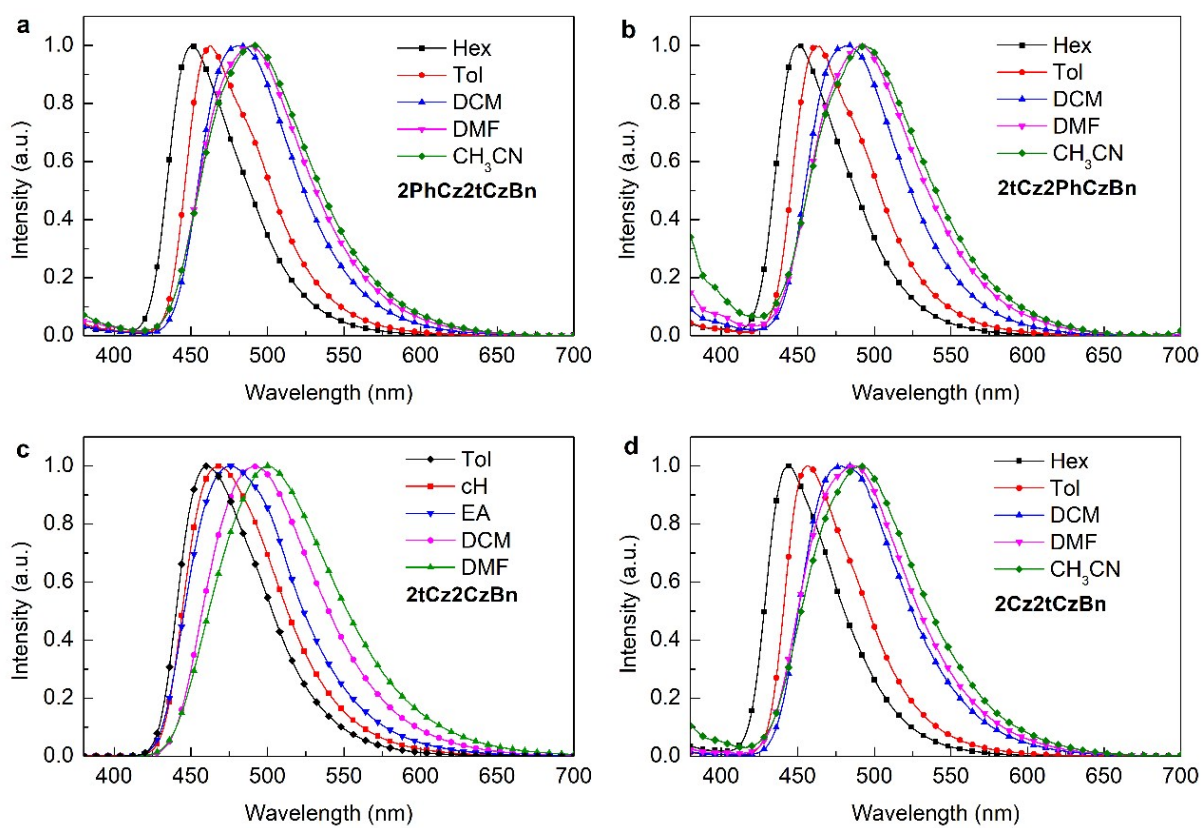
**Figure S10.** Single-crystal structures and molecular packing with selected intermolecular distances of (a-d) **2PhCz2tCzBn** and (e-h) **2tCz2PhCzBn**.



**Figure S11.** (a) TGA and (b) DSC curves of **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn**.

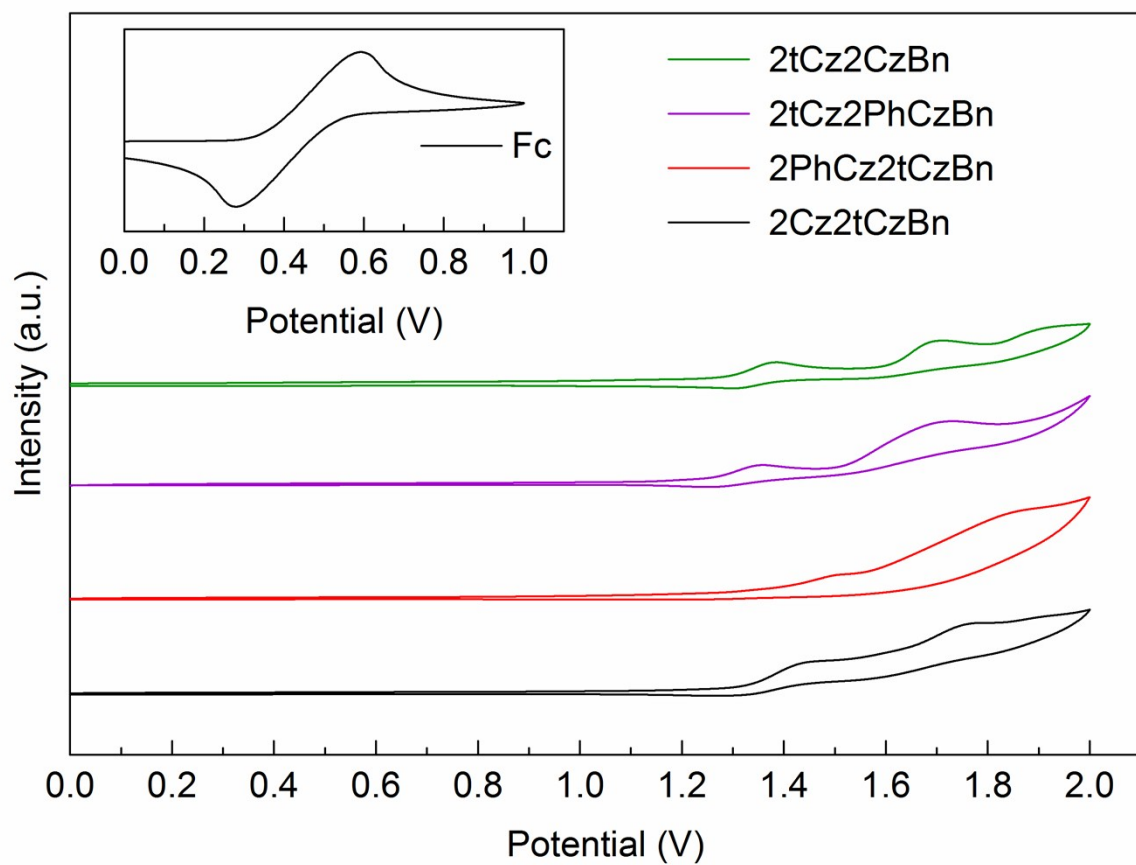


**Figure S 12.** Room-temperature fluorescence and low-temperature phosphorescence (at 77 K) spectra of **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn**. The  $\Delta E_{ST}$  values are estimated from the difference in the threshold energies of two spectra.

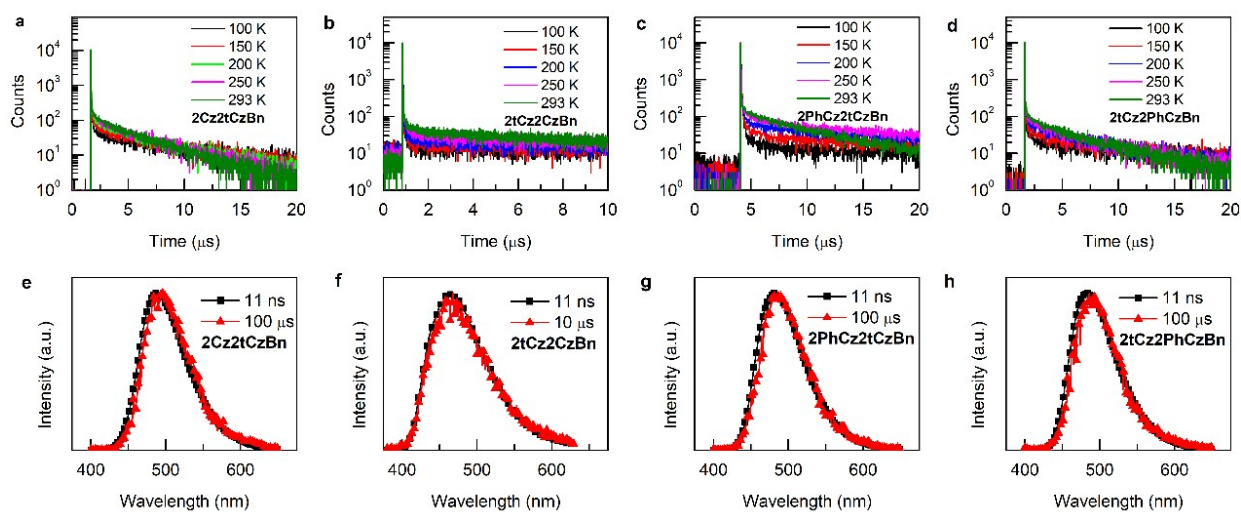


**Figure S13.** PL spectra of (a) **2PhCz2tCzBn**, (b) **2tCz2PhCzBn**, (c) **2tCz2CzBn**, and (d) **2Cz2tCzBn** in different solvents.





**Figure S14.** Cyclic voltammograms of **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn** in DCM solution ( $10^{-5}$  M).



**Figure S15.** (a-d) Temperature-dependent transient PL decay curves of (a) **2Cz2tCzBn**, (b) **2tCz2CzBn**, (c) **2PhCz2tCzBn**, and (d) **2tCz2PhCzBn** in neat films at temperatures ranging from 100 K to 293 K. (e-h) Prompt and delayed emission spectra of (e) **2Cz2tCzBn**, (f) **2tCz2CzBn**, (g) **2PhCz2tCzBn**, and (h) **2tCz2PhCzBn**.

**Table S1.** Photophysical characteristics of four compounds in neat films.

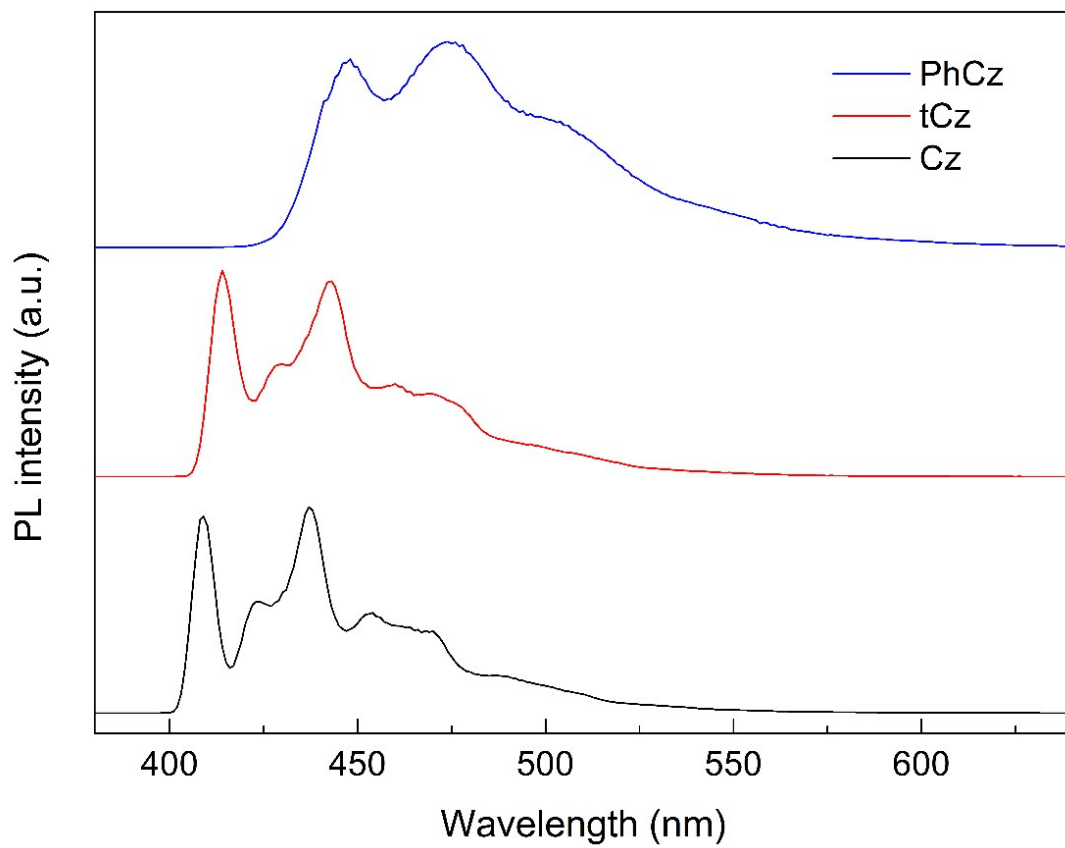
Films	$\tau_p$	$\tau_d$	PLQY/ $\Phi_p/\Phi_d$	$k_r$	$k_{ISC}$	$k_{RISC}$
	[ns]	[ $\mu$ s]	[%]	[ $10^7 \text{ s}^{-1}$ ]	[ $10^7 \text{ s}^{-1}$ ]	[ $10^5 \text{ s}^{-1}$ ]
<b>2Cz2tCzBn</b>	26	3.9	78/42/36	1.6	1.8	4.7
<b>2tCz2CzBn</b>	10	15.7	66/31/35	2.0	3.1	3.4
<b>2PhCz2tCzBn</b>	16	5.7	55/27/28	1.6	3.2	3.3
<b>2tCz2PhCzBn</b>	16	5.6	53/28/25	1.7	3.0	4.2

<sup>1</sup> Radiative rate constants of  $S_1$ ,  $k_r = \Phi_p/\tau_p + \Phi_d/\tau_d$ .

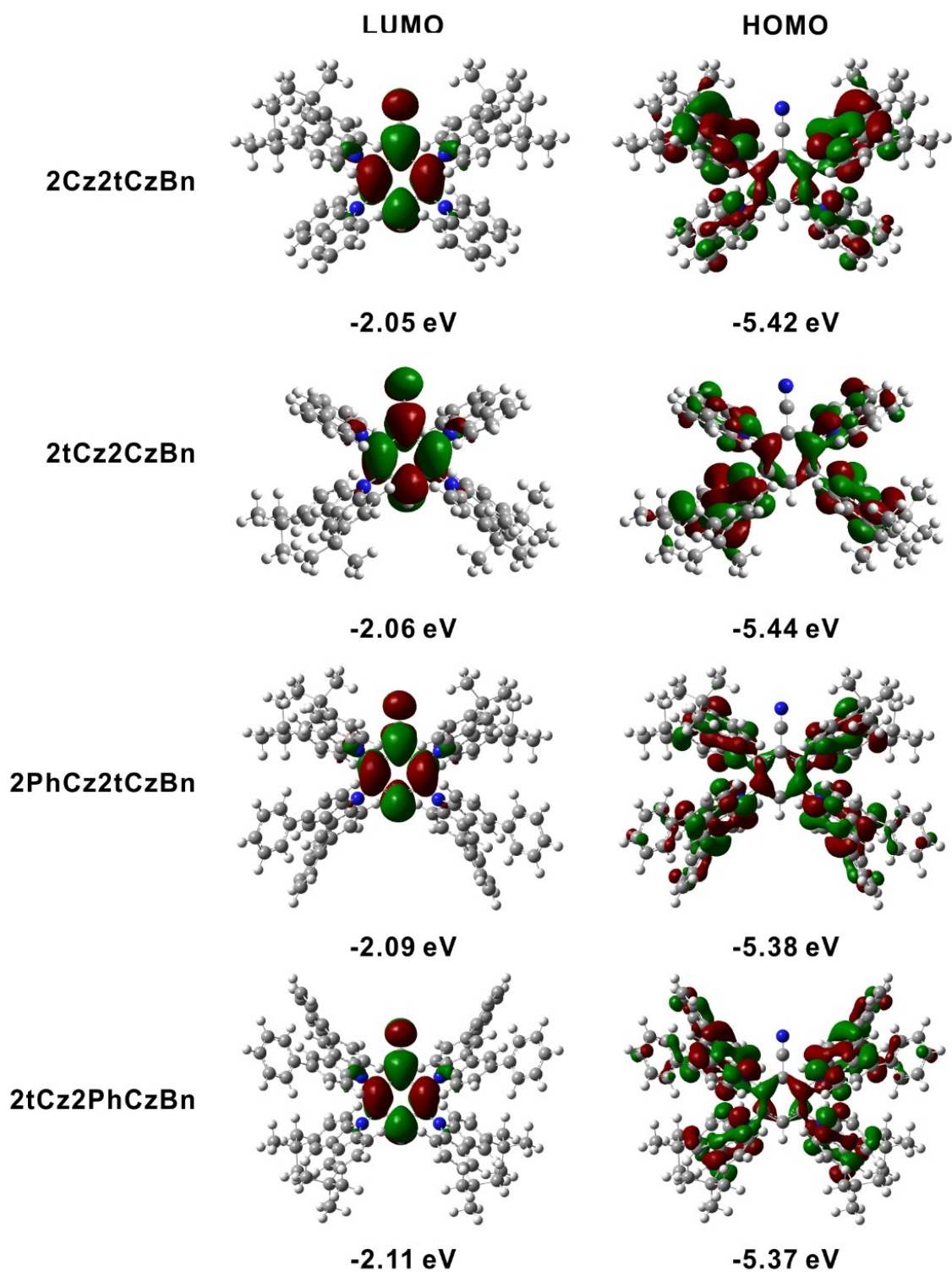
<sup>2</sup> Nonradiative rate constants of  $S_1$ ,  $k_{nr} = k_r(1 - \Phi_{PL})/\Phi_{PL}$ .

<sup>3</sup> Rate constants for ISC ( $S_1 \rightarrow T_1$ ),  $k_{ISC} = k_p - k_r - k_{nr}$ .

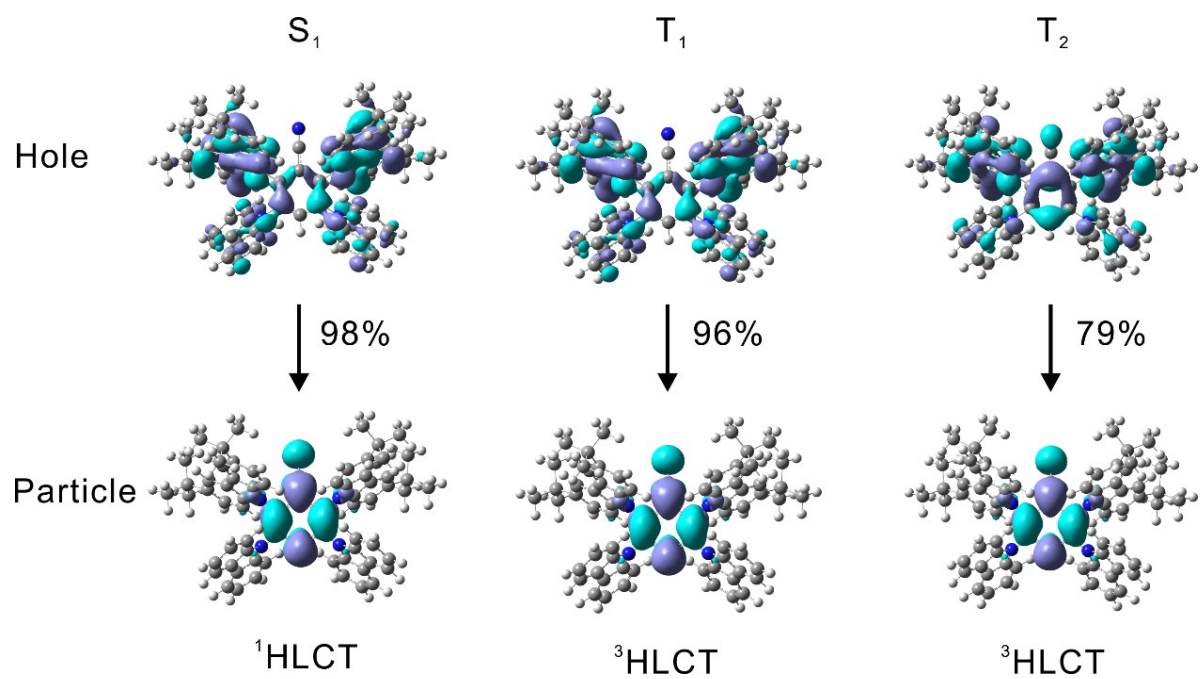
<sup>4</sup> Rate constants for RISC ( $T_1 \rightarrow S_1$ ),  $k_{RISC} = k_p k_d / k_{ISC} \cdot \Phi_d / \Phi_p$ .



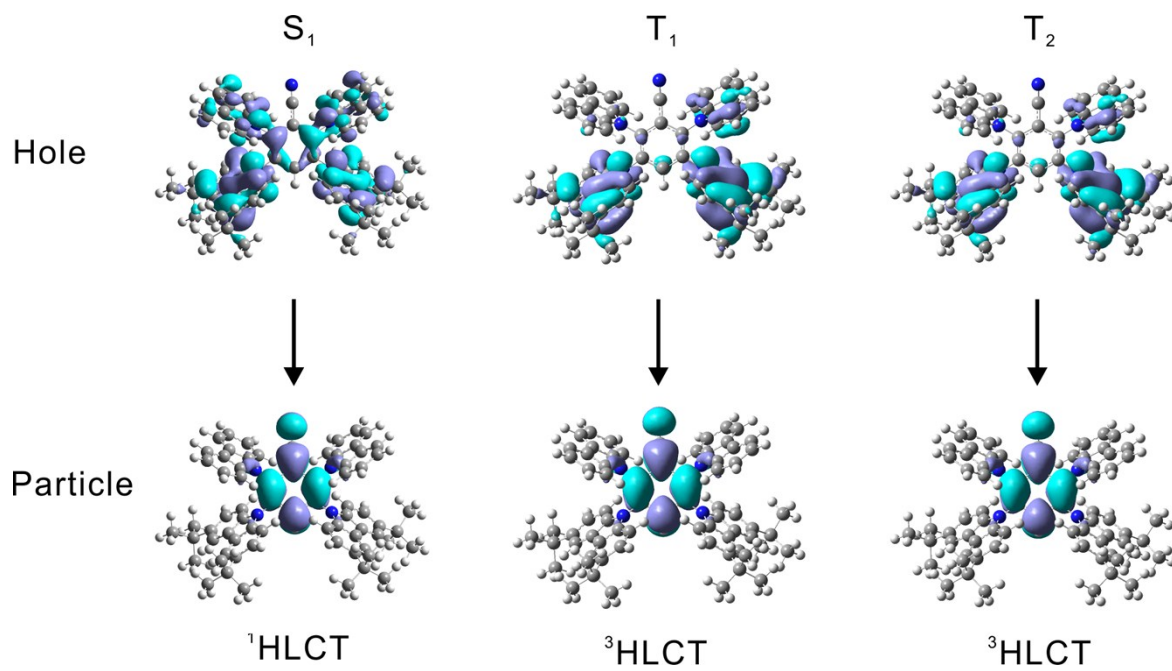
**Figure S16.** Phosphorescence spectra of Cz, tCz and PhCz in toluene at 77 K.



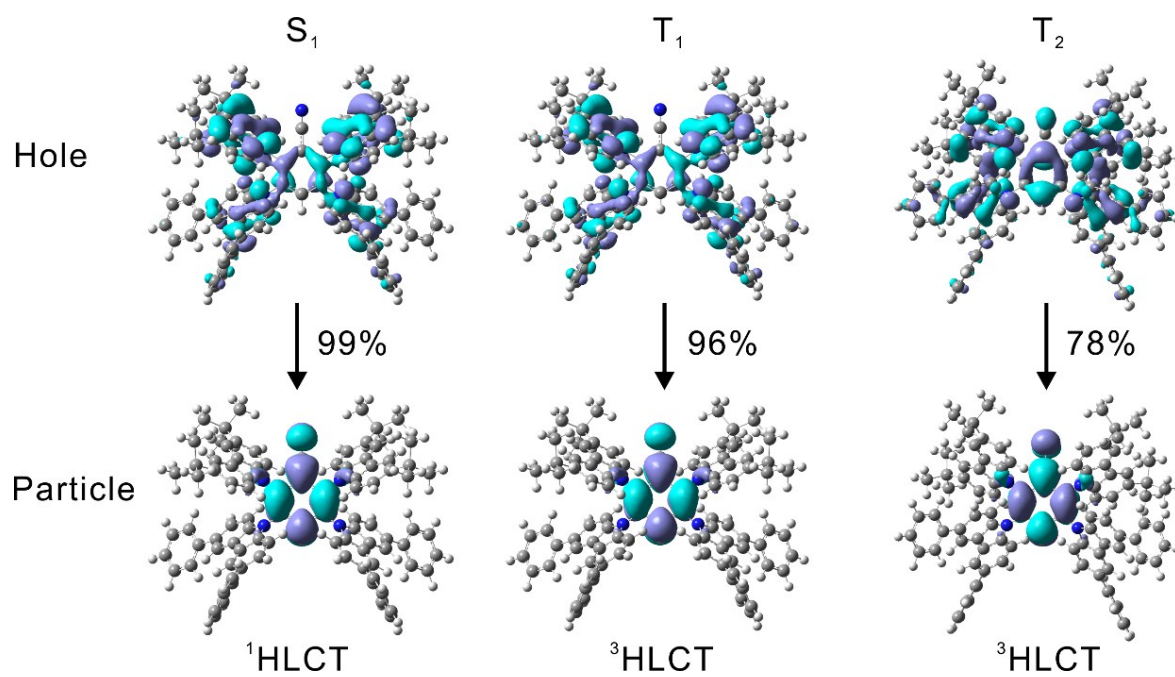
**Figure S17.** Calculated LUMO and HOMO distributions and energy levels of **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn**.



**Figure S18.** Natural transition orbital (NTO) analysis for **2Cz2tCzBn**.

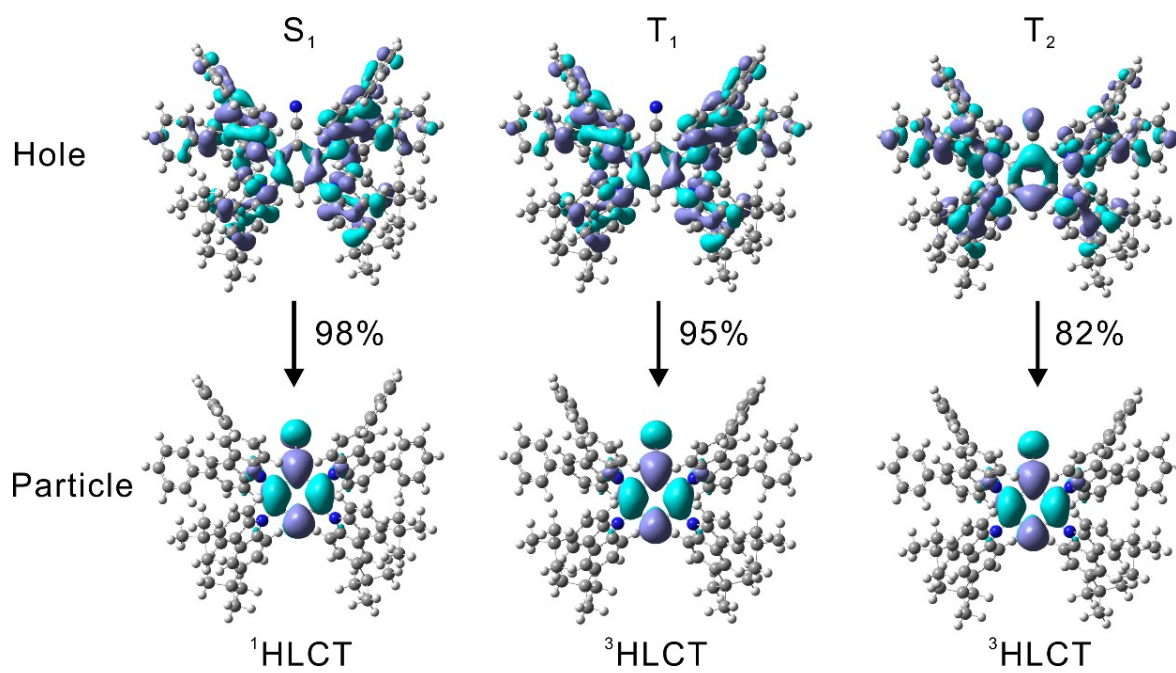


**Figure S19.** Natural transition orbital (NTO) analysis for **2tCz2CzBn**.

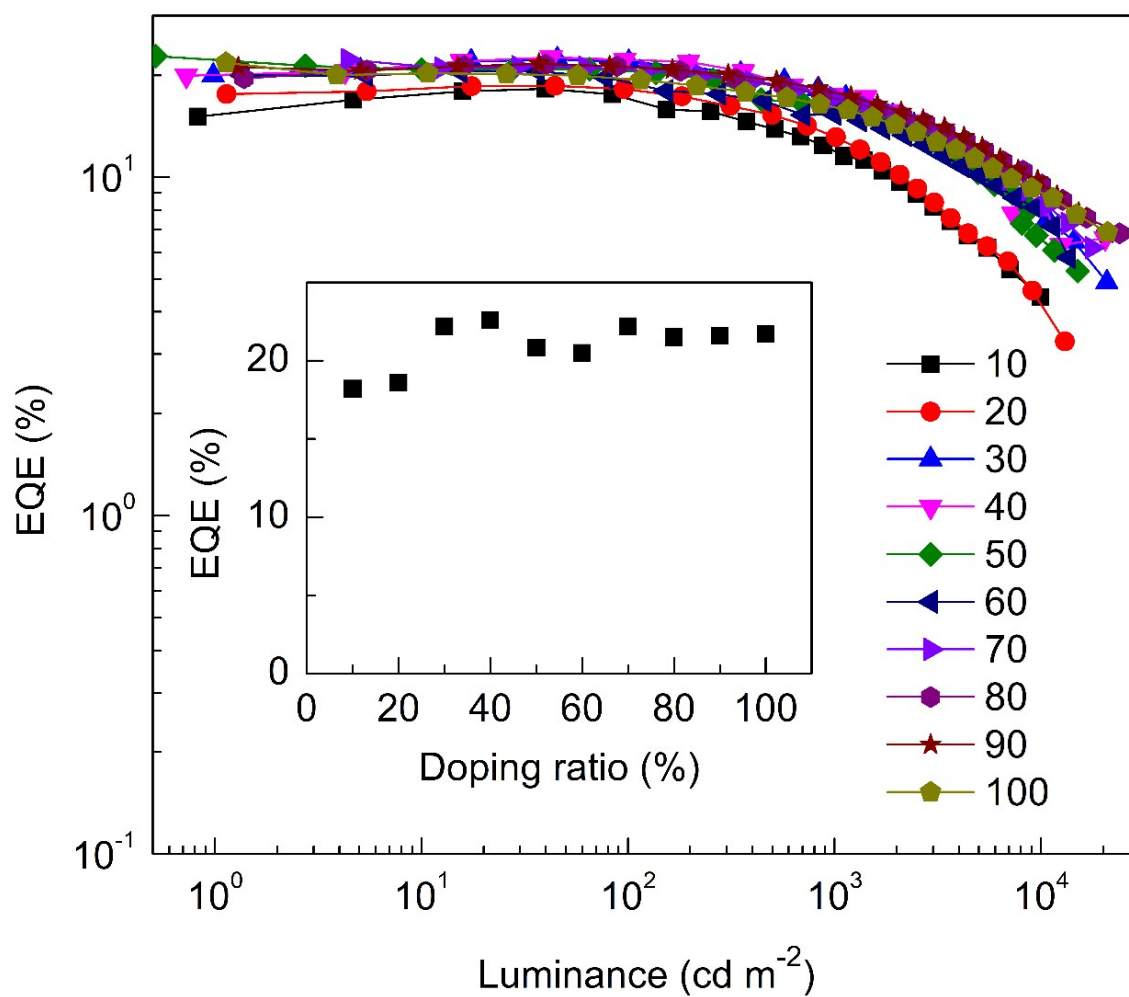


**Figure S20.** Natural transition orbital (NTO) analysis for **2PhCz2tCzBn**.

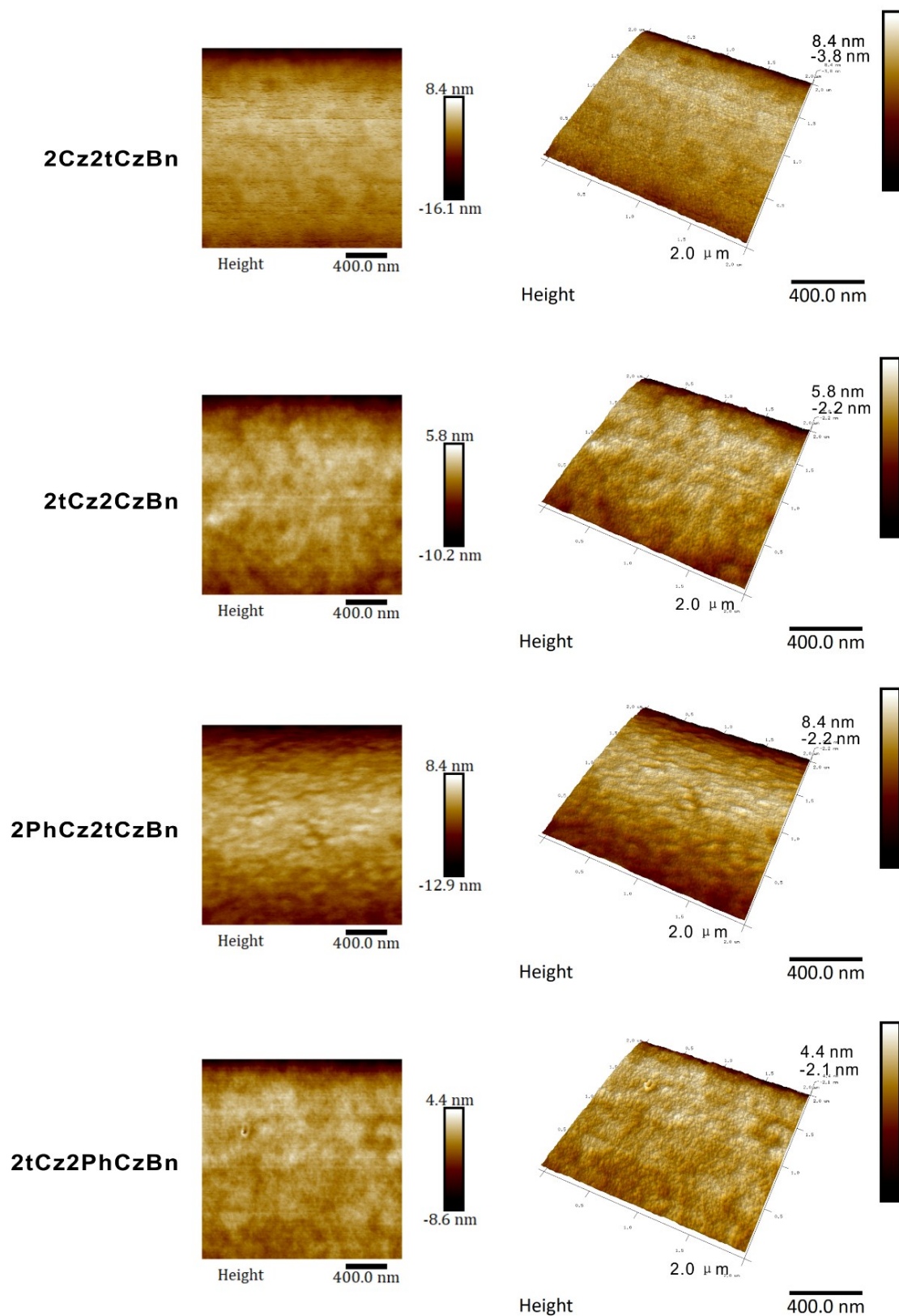




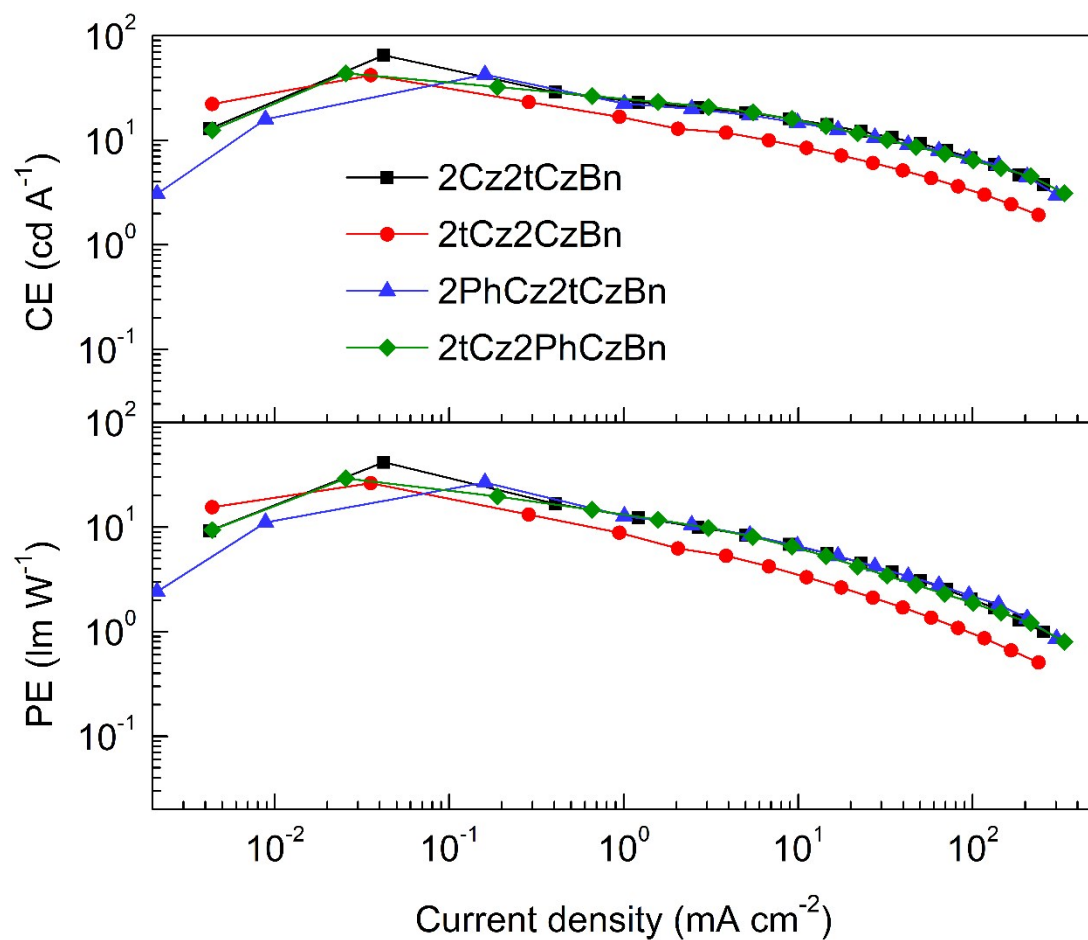
**Figure S21.** Natural transition orbital (NTO) analysis for **2tCz2PhCzBn**.



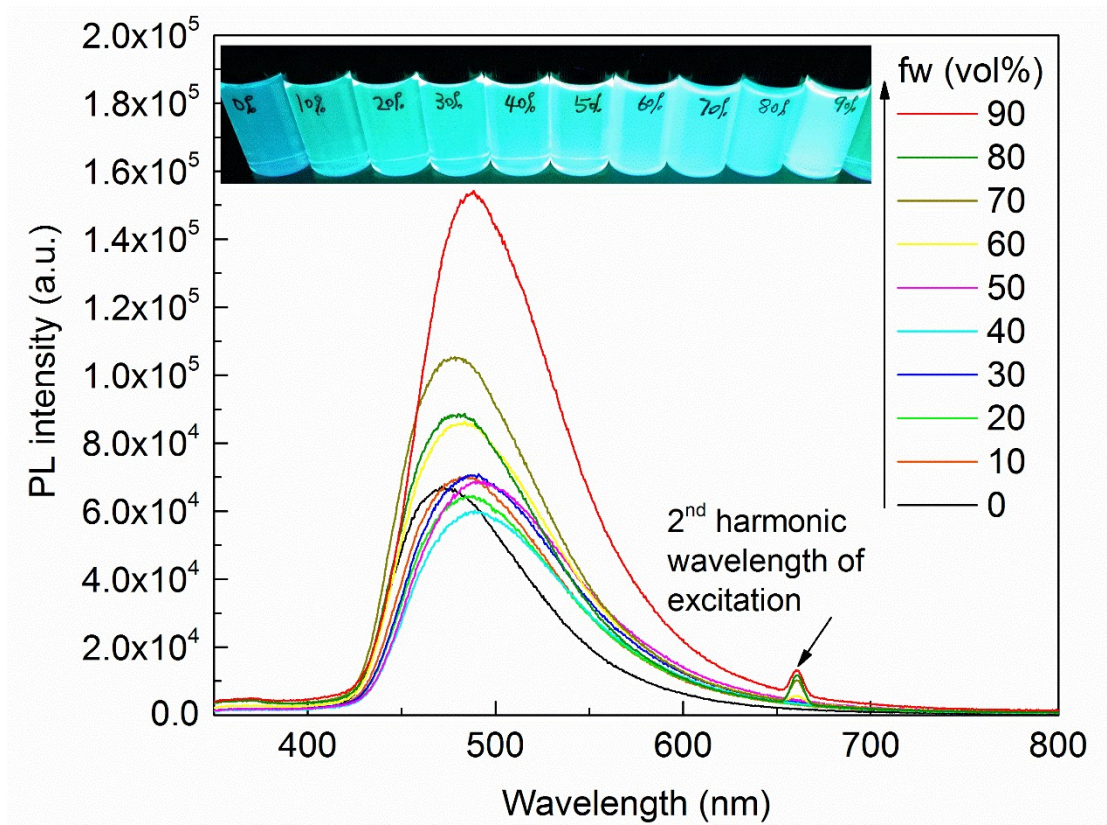
**Figure S22.** EQE evolution of blue OLEDs based on **2Cz2tCzBn** with varied doping concentrations from 10 to 100 wt%. Device structure: ITO/HATCN (10 nm)/TAPC (40 nm)/mCP (10 nm)/mCP:**2Cz2tCzBn** (x wt%, 30 nm)/PO-T2T (50 nm)/LiF (1 nm)/Al (100 nm).



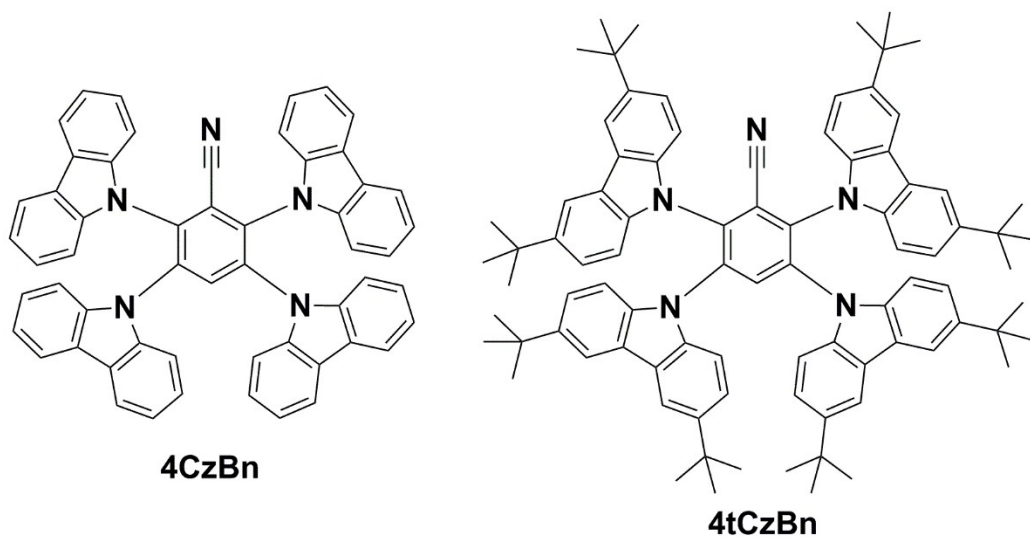
**Figure S23.** AFM images of solution-processed TADF thin films on PVK substrates for (a) **2Cz2tCzBn**, (b) **2tCz2CzBn**, (c) **2PhCz2tCzBn**, and (d) **2tCz2PhCzBn**, respectively.



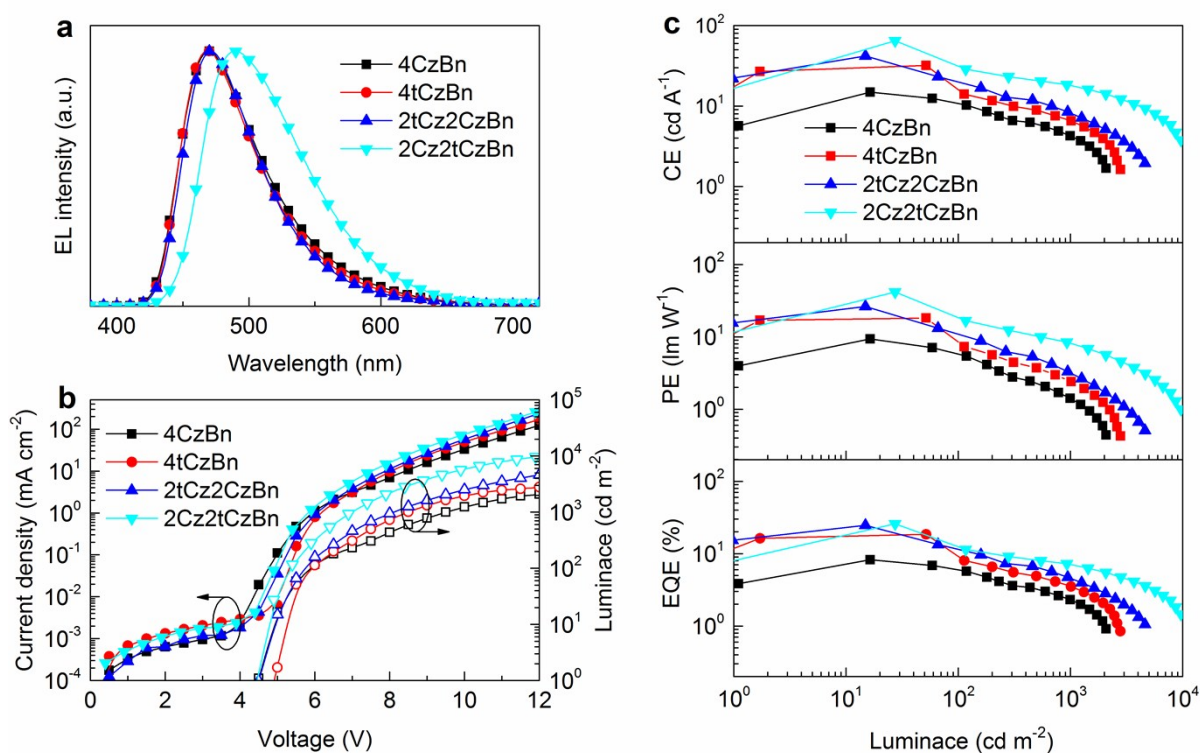
**Figure S24.** Current efficiency (CE) and power efficiency (PE) as a function of current density for solution-processed non-doped blue OLEDs using **2Cz2tCzBn**, **2tCz2CzBn**, **2PhCz2tCzBn**, and **2tCz2PhCzBn**, respectively.



**Figure S25.** PL emission of 2Cz2tCzBn in in THF/water mixtures with different water fractions ( $f_w$ ).

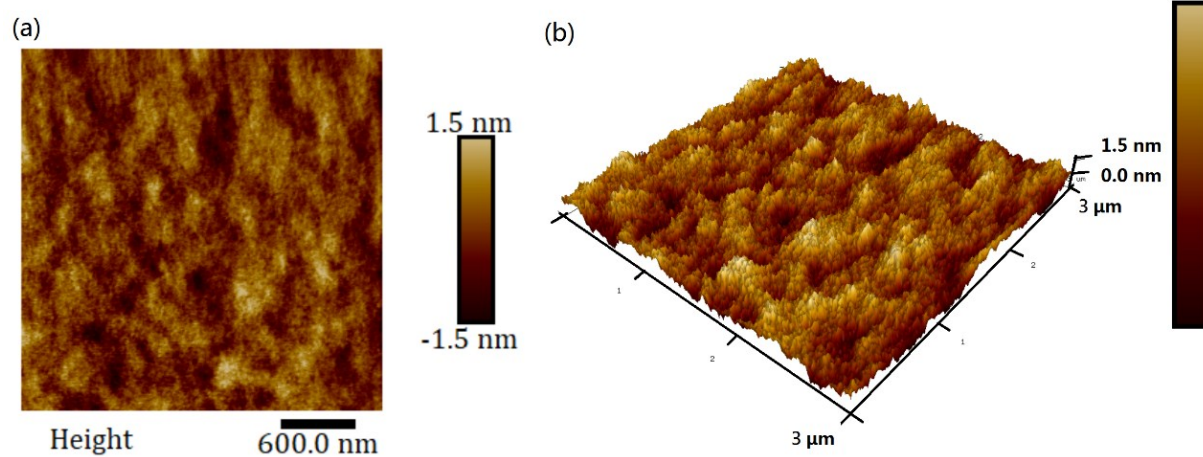


**Scheme S2.** Chemical structures of 4CzBn and 4tCzBn.



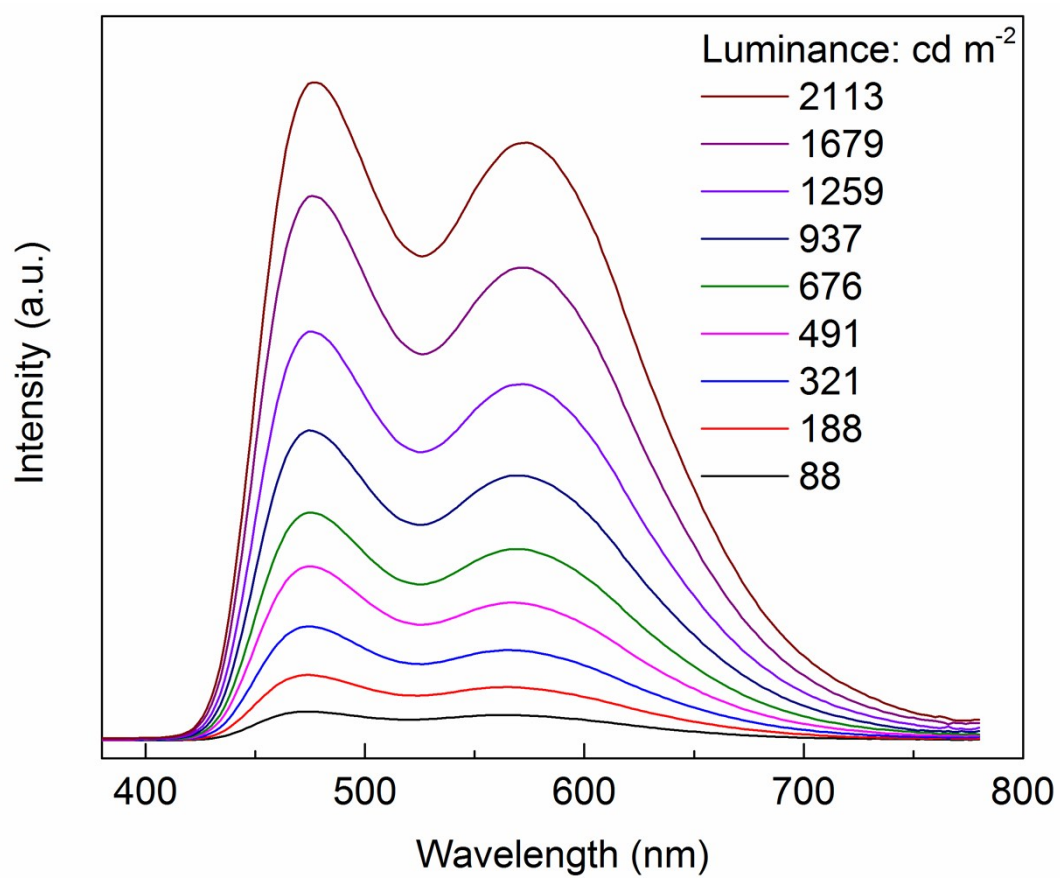
**Figure S26.** (a) EL spectra, (b) current density (J)-voltage (V)-luminance (L) characteristics, and (c) efficiency versus luminance relationships of the nondoped devices.

Two molecules 4CzBn and 4tCzBn have been synthesized (see **Scheme S2**). The EQEs of non-doped devices with 4CzBn and 4tCzBn are 8.2% and 18.4%, respectively. These results fully prove that the tert-butyl substitution is the main factor for the suppressed ACQ.



**Figure S27.** AFM images of solution-processed **2tCz2CzBn:3DMAC-BP** thin film on PVK substrate.





**Figure S28.** EL spectra of white OLED at different luminances.

**Table S2.** Comparison of the representative solution-processed white TADF emitters reported in the literature.

<b>EML</b>	<b>CIE</b> <b>(x, y)</b>	<b>CE<sub>max</sub></b> <b>[cd A<sup>-1</sup>]</b>	<b>PE<sub>max</sub></b> <b>[lm W<sup>-1</sup>]</b>	<b>EQE<sub>max</sub></b> <b>[%]</b>	<b>Refs</b>
<b>2tCz2CzBn:3DMAC-BP</b>	(0.34, 0.40)	67.0	35.1	27.3	This work
<b>DMAC-TRZ:Ir(dpm)PQ<sub>2</sub>:PO-01-TB</b>	(0.35, 0.44)	48.7	44.5	17.4	39
<b>PDTPT-1</b>	(0.31, 0.39)	38.8	20.3	14.2	40
<b>DCzDCN:SimCP2:TXO-TPA</b>	(0.35, 0.39)	36.50	37.31	13.39	41
<b>tBuCN-Flpic-mCP:(m- CF<sub>3</sub>DPQ)<sub>2</sub>Ir(pic)</b>	(0.35, 0.35)	43.5	15.7	20.6	5
<b>G2:Ir(bt)<sub>2</sub>(acac)</b>	(0.32, 0.33)	17.69	7.88	10.1	41