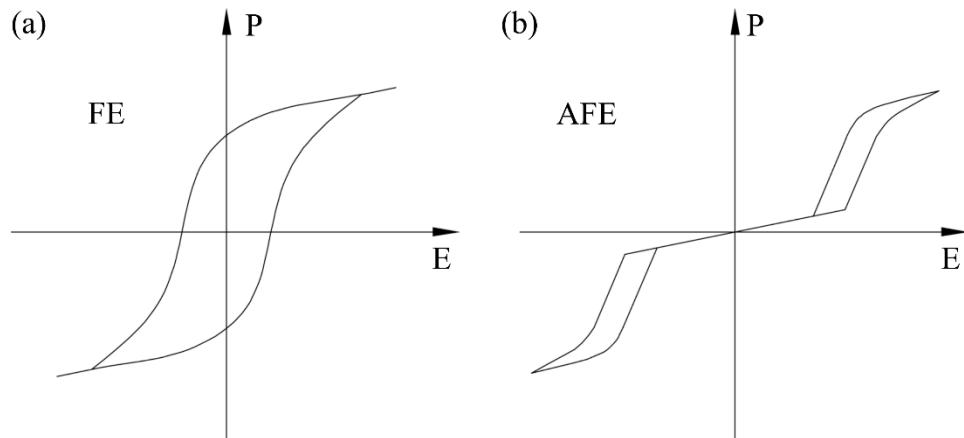


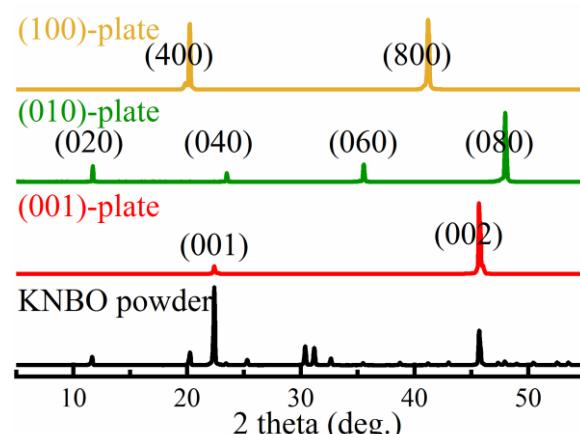
## Supporting Information

### Lead-free antiferroelectric borate crystal $K_3Nb_3B_2O_{12}$ with polar structure

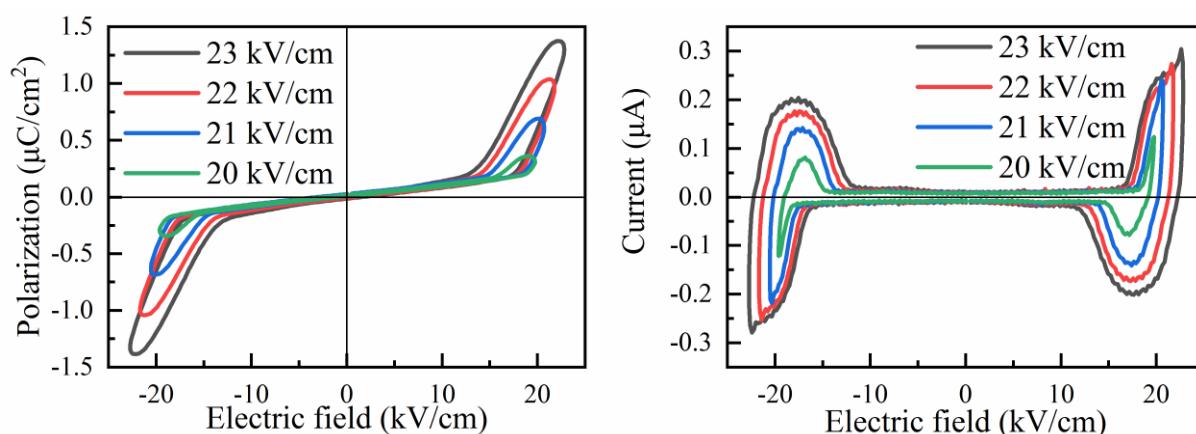
Pai Shan, Junjie Xiong, Zujian Wang\*, Chao He, Xiaoming Yang, Rongbing Su, Xifa Long\*



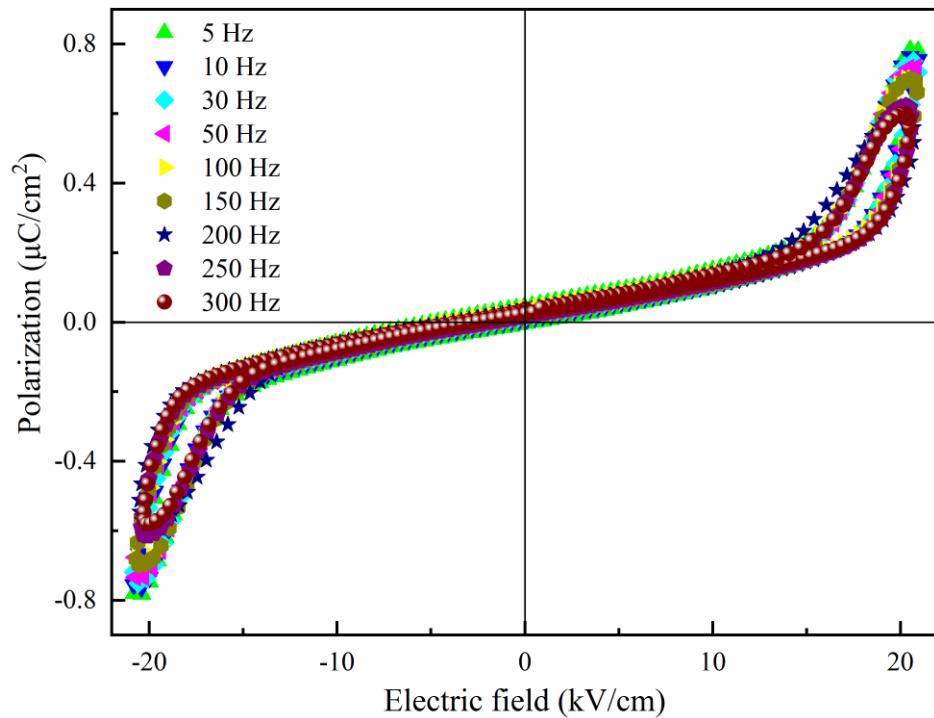
**Figure S1.** Schematic illustrations of (a) single  $P$ - $E$  hysteresis loop for FEs and (b) double  $P$ - $E$  hysteresis loop for AFEs.



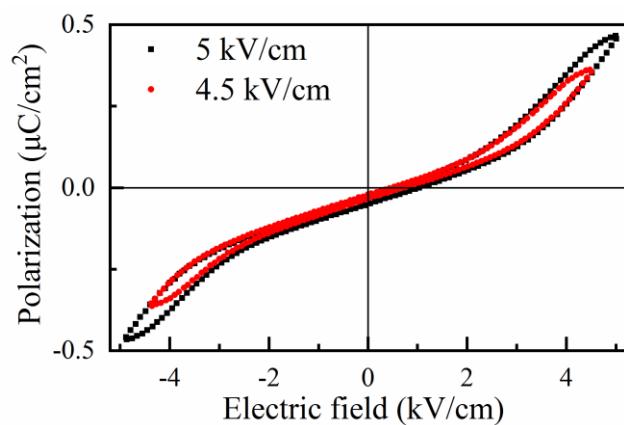
**Figure S2.** XRD patterns of KNBO powder, (001), (010) and (100) crystal plates.



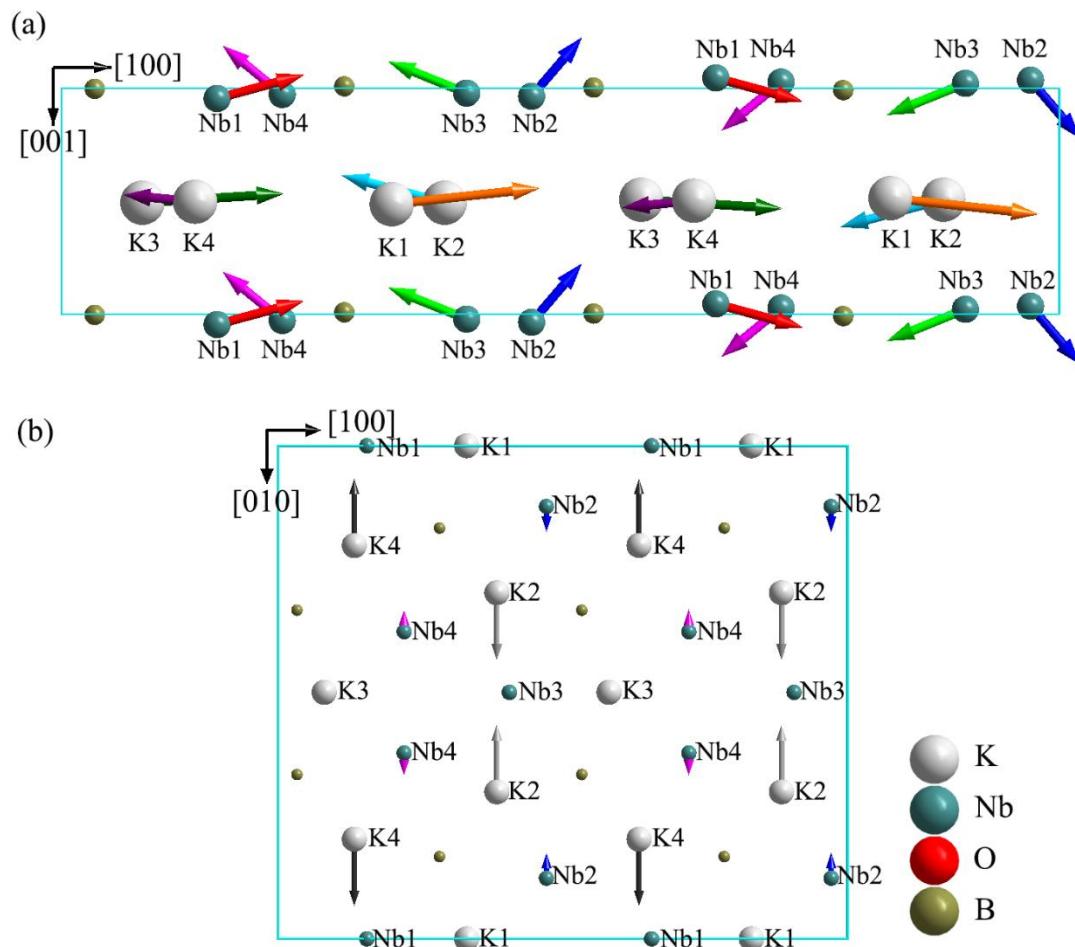
**Figure S3.** (a) Double  $P$ - $E$  hysteresis loops and (b) corresponding  $I$ - $E$  curves of KNBO crystal measured at different electric fields, by 50 Hz, at room temperature.



**Figure S4.** Double  $P$ - $E$  hysteresis loops of KNBO crystal measured with different frequencies, by 21 kV/cm, at room temperature.



**Figure S5.** Double  $P$ - $E$  hysteresis loops of KNBO crystal measured at 374 K, by 50 Hz.



**Figure S6.** Orientation of the dipole moment vectors for  $\text{NbO}_6$ ,  $\text{KO}_{13}$  and  $\text{BO}_3$  polyhedra in a unit cell of KNBO crystal, (a) in (010) plane and (b) along [010] direction. All O atoms are omitted for clarity. The numbers represent cations possessing different crystallographically independent sites.

**Table S1.** Crystal data and structure refinement for KNBO crystal at room temperature.

Formula sum	K <sub>3</sub> Nb <sub>3</sub> B <sub>2</sub> O <sub>12</sub>
Formula weight (g/mol)	609.65
Crystal color	colorless
Crystal system	orthorhombic
Space group	P2 <sub>1</sub> ma
a/Å	17.5219(7)
b/Å	15.1708(6)
c/Å	3.96882(11)
V/Å <sup>3</sup>	1055.00(7)
Z	4
Crystal size	0.18 × 0.13 × 0.09 mm <sup>3</sup>
F(000)	1144
Limiting indices	-4 ≤ h ≤ 4, -21 ≤ k ≤ 19, -18 ≤ l ≤ 18
GOF(F <sup>2</sup> )	1.096
Final R indices [F <sub>o</sub> <sup>2</sup> > 2σ(F <sub>c</sub> <sup>2</sup> ).] <sup>a)</sup>	R <sub>1</sub> = 0.0232, wR <sub>2</sub> = 0.0557
Final R indices (all data) <sup>a)</sup>	R <sub>1</sub> = 0.0316, wR <sub>2</sub> = 0.0616

<sup>a)</sup>  $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$  and  $wR_2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$  for  $F_o^2 > 2\sigma(F_c^2)$

**Table S2.** Calculation of dipole moments for NbO<sub>6</sub>, KO<sub>13</sub> and BO<sub>3</sub> polyhedra within a unit cell of KNBO crystal.

Cation of polar polyhedra	Dipole Moment [Debye]			
	[100]-component	[010]-component	[001]-component	total magnitude
Nb(1 <sup>a)</sup> -1 <sup>b)</sup> O <sub>6</sub>	1.016	0.000	-1.218	1.586
Nb(1-2)O <sub>6</sub>	1.015	0.000	1.218	1.586
Nb(2-1)O <sub>6</sub>	-0.259	0.788	-1.337	1.574
Nb(2-2)O <sub>6</sub>	-0.259	-0.788	-1.337	1.574
Nb(2-3)O <sub>6</sub>	-0.254	0.786	1.337	1.571
Nb(2-4)O <sub>6</sub>	-0.254	-0.786	1.337	1.571
Nb(3-1)O <sub>6</sub>	0.740	0.000	-1.300	1.496
Nb(3-2)O <sub>6</sub>	0.730	0.000	1.299	1.490
Nb(4-1)O <sub>6</sub>	-0.362	-0.683	-1.305	1.516
Nb(4-2)O <sub>6</sub>	-0.362	0.683	-1.305	1.516
Nb(4-3)O <sub>6</sub>	-0.362	-0.677	1.307	1.516
Nb(4-4)O <sub>6</sub>	-0.362	0.677	1.307	1.516
K(1-1)O <sub>13</sub>	2.362	0.000	1.129	2.618
K(1-2)O <sub>13</sub>	2.362	0.000	-1.129	2.618
K(2-1)O <sub>13</sub>	-1.216	2.064	1.424	2.787
K(2-2)O <sub>13</sub>	-1.216	-2.064	1.424	2.787
K(2-3)O <sub>13</sub>	-1.216	2.063	-1.425	2.787
K(2-4)O <sub>13</sub>	-1.216	-2.063	-1.425	2.787
K(3-1)O <sub>13</sub>	2.397	0.000	0.675	2.491
K(3-2)O <sub>13</sub>	2.397	0.000	-0.675	2.491
K(4-1)O <sub>13</sub>	-1.164	-2.060	0.498	2.418
K(4-2)O <sub>13</sub>	-1.164	2.060	0.498	2.418
K(4-3)O <sub>13</sub>	-1.164	-2.061	-0.498	2.419
K(4-4)O <sub>13</sub>	-1.164	2.061	-0.498	2.419
B(1-1)O <sub>3</sub>	-0.072	-0.048	-0.151	0.174
B(1-2)O <sub>3</sub>	-0.072	0.048	-0.151	0.174
B(1-3)O <sub>3</sub>	-0.071	-0.048	0.151	0.174
B(1-4)O <sub>3</sub>	-0.071	0.048	0.151	0.174
B(2-1)O <sub>3</sub>	-0.085	0.025	0.054	0.104
B(2-2)O <sub>3</sub>	-0.085	-0.025	0.054	0.104
B(2-3)O <sub>3</sub>	-0.085	0.023	-0.054	0.104
B(2-4)O <sub>3</sub>	-0.085	-0.023	-0.054	0.104

<sup>a)</sup> The first number represents cations possessing different crystallographically independent sites, these numbers obtain identical implication with Figure 3; <sup>b)</sup> The second number represents serial number of cations occupying in the same crystallographically independent sites.