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## An Efficient and Stable Tin-Based Perovskite Solar Cells Passivated by Aminoguanidine Hydrochloride

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**Figure S1.** Energy-dispersive X-ray (EDS) spectroscopy mapping of Sn, I, F, and Cl in (a) pristine and (b) NH<sub>2</sub>GACl incorporated perovskite film.



**Figure S2.** (a) XPS spectra of the perovskite films. (b) XPS spectra (Cl 2p) of the perovskite films with 0.02 mol, 0.05 mol and 0.10 mol  $NH_2GACI$ .



**Figure S3.** SEM images of the perovskite films with (a) 0.02 mol and (b) 0.10 mol NH<sub>2</sub>GACl-doped on ITO/PEDOT:PSS layers.



**Figure S4.** AFM images of the perovskite films on ITO/PEDOT:PSS layers with various proportions of  $NH_2GACl$  additive: (a) pristine, (b) 0.02 mol, (c) 0.05 mol and (d) 0.10 mol, respectively.



**Figure S5.** (a) X-ray diffraction patterns and (b) the corresponding full width at half maximum diffraction peaks of the perovskite films with varying concentrations of NH<sub>2</sub>GACl additive (0 mol, 0.02 mol, 0.05 mol and 0.10 mol).



**Figure S6.** (a, b) X-ray photoelectron spectra (Sn 3d) of the pristine and perovskite films modified with NH<sub>2</sub>GACl. (c) The contact angles between perovskite films (the pristine or NH<sub>2</sub>GACl-modified) and water.



**Figure S7.** 1H NMR spectra of  $NH_2GACl$ , perovskite and perovskite-  $NH_2GACl$  mixture solution in deuterated *N*, *N*-dimethylformamide-d<sub>7</sub> (DMF-d<sub>7</sub>).



Figure S8. (a) Ultraviolet photoelectron spectroscopy (UPS) patterns and (b) UV-vis absorbance spectra of the pristine, 0.02 mol, 0.05 mol and 0.10 mol NH<sub>2</sub>GACl-modified perovskite films.



Figure S9. A graphical representation of the device characteristics (20 devices).



**Figure S10.** Steady-state photocurrent (left) and stabilized PCE (right) of the perovskite devices measured at the maximum power point.



**Figure S11.** (a)  $V_{OC}$  and (b)  $J_{SC}$  of the perovskite solar cells without and with NH<sub>2</sub>GACl additive measured under various illumination levels.



**Figure S12**. Stability of the pristine and NH<sub>2</sub>GACl-modified perovskite solar cells in nitrogen at 25 °C without encapsulation. (a) Normalized  $V_{OC}$ . (b) Normalized  $J_{SC}$ . (c) Normalized FF.



**Figure S13**. Stability of the pristine and NH<sub>2</sub>GACl-modified perovskite solar cells under relative humidity (RH) 60% without encapsulation. (a) Normalized PCE, (b)Normalized  $V_{OC}$ , (c) Normalized  $J_{SC}$ , (d) Normalized FF.

Devices	$J_{\rm SC}$ (mA cm <sup>-2</sup> )	V <sub>OC</sub> (V)	FF (%)	PCE (%)
pristine	15.98	0.48	64.09	4.92
0.02 mol NH <sub>2</sub> GACl	18.88	0.50	67.89	6.42
0.05 mol NH <sub>2</sub> GACl	19.65	0.54	68.84	7.30
0.10 mol NH <sub>2</sub> GACl	14.52	0.45	63.48	4.19

**Table S1.** Photovoltaic parameters of the perovskite solar cells with differentconcentrations of  $NH_2GACI$  doped in precursor solution.

device active areas (cm <sup>2</sup> )	РСЕ (%)	Ref.
0.04	12.40	1
0.09	10.1	2
0.0222	9.6	3
0.04	9.0	4
0.04	7.3	This work

**Table S2** The different device active areas of Sn-perovskite solar cells in the previous

 literatures and this work.

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