## **Supporting Information**

# High-performance visible laser rewritable black paper

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### **Table of Contents**

1. Experimental section	S3
<b>2.</b> Synthesis Methods	S5
<b>3.</b> Supplementary Figures	S6
4.	
References	S10

#### 1. Experimental section

#### 1.1. Materials and reagents

Resorcine (99 %), 2-(4-diethylamino-2-hydroxybenzoyl) benzoic acid (98 %), trifluoroacetic acid (TFA, 99 %), methanesulfonic acid (CH<sub>3</sub>SO<sub>3</sub>H, 99 %), phenylhydrazine (98 %), 3-Methyl-2-butanone (98 %), 1,3-Propanesultone (98 %) and salicylaldehyde (98 %) were purchased from Energy Chemical (Shanghai, China). Acidochromic dyes and pH-sensitive dye: 2-Anilino-6-dibutylamino-3-methylfluoran (ODB-2) and 3diethylamino-7-(2'-chlorophenylamino) fluorane (FH 101) were purchased from Deba Chemical (Shanghai, China). Methyl orange (MO) was purchased from Guangfu Fine Chemical Research Institute (Tianjin, China). Anhydrous sodium sulfade  $(Na_2SO_4)$ , sodium bicarbonate (NaHCO<sub>3</sub>) and hydrochloric acid (HCl) were purchased from Beijing Chemical Works (Beijing, China). Solvents: methanol (MeOH, HPLC) was purchased from Yuwang Group (Shandong, China). Ethyl acetate (EtOAc), ethanol (EtOH), dimethyl sulfoxide (DMSO) and dichloromethane ( $CH_2Cl_2$ ) were purchased from Beijing Chemical. Wahaha purified  $H_2O$  (Hangzhou, China) was used for all experiments. PEG 20000 (molecular weight: 17,000–22,000) was purchased from Guangfu Fine Chemical Research Institute (Tianjin, China). Cellulose filter paper (Whatman–Xinhua, grade 91, Hangzhou, China) was selected as the paper substrate.

#### **1.2.** Instruments and characterizations

Absorption spectra were measured using Analitik Jena Specord<sup>®</sup>210 plus UV/VIS spectrophotometer. Reflective spectra were tested via integrating sphere on Analitik Jena Specord<sup>®</sup>210 plus UV-VIS spectrophotometer, using barium sulfate (BaSO<sub>4</sub>) as background, slit was 2 cm. CIE L\*, a\*, b\* was measured by X-rite spectrodensitometer. <sup>1</sup>H NMR (400, 500 MHz) spectra was recorded on a Bruker AVANCE400 at room temperature. Blue light source for imaging with a mask here used was Wota F25 fishing lamp (Ningbo, China). Laser automatic printing was realized by laser marking machine Xiai XYC-UV / MGL, which was purchased from Changchun New Industries Optoelectronics Tech. Co. (Changchun, China). Photographs were captured using the Nikon D7100 camera.

#### 1.3. Preparation of black VLRP

The preparation process of black **VLRP** referred to our previous work<sup>S1</sup>. Filter paper was firstly coated with a layer of 10 wt% PEG20000 aqueous solution and then dried by heating at 80 °C. Later, it was soaked into a MeOH solution of the acidochromic dyes (5 mM ODB-2 & 0.7 mM Rhodol), P-1 (11.4 mM) and 15 wt% PEG20000 mixture.

S4

#### 2. Synthesis Methods

#### 2.1. Synthesis of P-1

Compound P-1 was reported previously<sup>S1,S2</sup>. <sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>)  $\delta$  11.03 (s, 1H), 8.60 (d, *J* = 16.3 Hz, 1H), 8.28 (d, *J* = 7.8 Hz, 1H), 8.03 (d, *J* = 7.6 Hz, 1H), 7.94-7.81 (m, 2H), 7.62 (m, 2H), 7.48 (t, *J* = 7.5 Hz, 1H), 7.04 (d, *J* = 8.2 Hz, 1H), 6.99 (t, *J* = 7.5 Hz, 1H), 4.81 (t, *J* = 7.3, 2H), 2.64 (t, *J* = 6.0 Hz, 2H), 2.18 (m, 2H), 1.77 (s, 6H).

#### 2.2. Synthesis of Rhodol

The synthesis of Rhodol was according to the literature<sup>S3</sup>. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)  $\delta$  10.08 (s, 1H), 7.97 (d, *J* = 7.5 Hz, 1H), 7.78 (t, *J* = 7.4 Hz, 1H), 7.70 (t, *J* = 7.4 Hz, 1H), 7.26 (d, *J* = 7.6 Hz, 1H), 6.66 (s, 1H), 6.52 (s, 2H), 6.46 (s, 1H), 6.44 (s, 2H), 3.37-3.32 (m, 4H), 1.09 (t, *J* = 6.9 Hz, 6H).



#### 3. Supplementary Figures

**Fig. S1** Plots of the absorbance at two complementary absorption peaks of (a) ODB-2 and (b) FH101 after adding excessive  $CH_3SO_3H$  versus increasing concentration and thus its molar absorption coefficients ( $\epsilon$ ) at two complementary absorption peaks were obtained respectively according to LambertBeer law.



**Fig. S2** Normalized UV–vis spectra of Rhodol & H<sup>+</sup> (MeOH, DMSO) and ODB-2 & H<sup>+</sup> (MeOH).



**Fig. S3** Schematic illustrations of adding acid in as-prepared paper (The as-prepared paper was test after enough time fumigation by acid, and the paper has the darkest colour and colour-changed no longer with fumigation time increased).



Fig. S4 Normalized UV–vis spectra of MO &  $H^+$  and ODB-2 &  $H^+$  in MeOH.



**Fig. S5** (a) Photographs of papers loaded with different ratios of MO to ODB-2 (2.5 mM) before and after addition HCl. (b) L\*, a\*, b\* measured by spectrodensitometer and photographs of papers loaded with different ratios of MO to ODB-2 (2.5 mM) after addition HCl. (c, d) Reflectance change at 605 nm of dyes & P-1 based paper with different ratio of P-1 and dyes before and after irradiation, which has different ratio of ODB-2 to MO: c) 4 / 1; d) 3 / 1. (e) L\*, a\*, b\* measured by spectrodensitometer and photographs of P-1 & dyes (n/n 2.5/1) based paper after irradiation with different ratio of ODB-2 to MO: 4/1 (left);

3/1 (right).

#### 4. References

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