Supplementary material

A Sensitive Metal-Organic Frameworks Nanosensor with Cation-Introduced Chirality for Enantioselective Recognition and Determination of Quinine and Quinidine in Human Urine

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Figure S1 Solid-state circular dichroism (CD) spectra of (a) 1 (b) N-benzylquininium chloride and

(c) **1-**C.



Figure S2 Powder X-ray diffraction (PXRD) patterns of 1, 1-Eu, 1-C, and 1-C-Eu.



Figure S3 Fourier transform infrared (FT-IR) spectra of 1, 1-Eu, 1-C, and 1-C-Eu.



Figure S4 N₂ adsorption (solid symbols) and desorption (hollow symbols) isotherms of (a) 1, (b) 1-

Eu, (c) 1-C, and (d) 1-C-Eu.



Figure S5 Scanning electron microscope (SEM) images of (a) 1, (b) 1-Eu, (c) 1-C, and (d) 1-C-Eu.



Figure S6 Energy dispersive analysis by X-rays (EDX) spectra of (a) 1 and (b) 1-C-Eu.



Figure S7 Thermogravimetric analyses curves of (a) 1, (b) 1-Eu, (c) 1-C, and (d) 1-C-Eu.



Figure S8 Solid-state photoluminescent spectra of (a) H_2 bpdc and (b) 1. (Red line: the excitation spectra, black line: the emission spectra)



Figure S9 Solid-state photoluminescent spectra of **1**-Eu. (Red line: the excitation spectra, black line: the emission spectra) The inserts are the photograph and CIE chromaticity diagram of **1**-Eu excited with 365 nm laboratory UV light.



Figure S10 Fluorescence lifetime patterns of ${}^{5}D_{0}$ in the solid powder of (a) 1-Eu and (b) 1-C-Eu.



Figure S11 Time-response emission intensities at 613 nm ($\lambda_{ex} = 313$ nm) of the DMA suspensions of **1**-C-Eu upon the addition of (a) Quinine and (b) Quinidine (0.2mg·mL⁻¹).



Figure S12 PXRD patterns of 1-C-Eu after immersed in different solvents for 4 hrs: (a) simulated. (b) As-synthesized 1. (c) As-synthesized 1-C-Eu. (d) 1-C-Eu immersed in the mixed-solvent of DMA and aqueous solutions (v/v = 2/1). (e) 1-C-Eu immersed in aqueous solutions.



Figure S13 Emission spectra of 1-Eu-C at 613 nm ($\lambda_{ex} = 313$ nm) immersed in the mixed-solvent

of DMA and aqueous solutions of different pH (v/v =2/1).



Figure S14 PXRD patterns of 1-C-Eu after immersed in the mixed-solvent of DMA and aqueous

solutions of different pH for 4 hrs.



Figure S15 Time dependence of (a) emission spectra and (b) emission intensities at 613 nm of 1-C-Eu in the mixed-solvent of DMA and aqueous solutions ($\lambda_{ex} = 313$ nm).



Figure S16 PXRD patterns of 1-C-Eu after soaking in different solutions for about 2 hrs: (a) simulated. (b) As-synthesized 1-C-Eu. (c) 1-C-Eu after soaked in the DMA solutions of Quinine (10⁻² M). (d) 1-C-Eu after soaked in the DMA solutions of Quinidine (10⁻² M).



Figure S17 Luminescence lifetime patterns of 5D_0 in 1-Eu-C with the presence of (a) Quinine and

(b) Quinidine in DMA solutions.



Figure S18 Excitation spectra of 1-Eu-C suspensions with different concentrations of (a) Quinine

and (b) Quinidine ($\lambda_{em} = 613$ nm).



Figure S19 UV-Vis absorption spectra of 1, 1-Eu, 1-C-Eu, N-benzylquininium chloride, H₂BPDC,

Quinine and Quinidine in DMA.



Figure S20 Emission spectra of 1-Eu dispersed in in DMA solutions of (a) Quinine and (b) Quinidine at different concentrations ranging from 0 to 0.25 mg·mL⁻¹ ($\lambda_{ex} = 313$ nm). (c) The fitting curve of I₀/I versus concentration of Quinine and Quinidine.



Figure S21 Liquid ¹H NMR spectra of the analytes and *N*-benzylquininium chloride. (a) Chemical shift between 12 and 0 ppm. (b) Chemical shift between 9 and 4 ppm. (c) Chemical shift between 6.65 and 6.55 ppm.

Table S1 Weight percentages of all elements in 1 and 1-C-Eu determined by Energy dispersive

analysis by X-rays (EDX).

Element	Weight percentages	
	1	1-C-Eu
С	62.49	57.31
N	3.07	1.59
0	31.50	28.69
Cd	2.94	1.42
Eu		10.99

Table S2 Responses of luminescence lifetimes of 1-C-Eu towards Quinine and Quinidine of various

concentrations in DMA solutions.

Analytes	Concentrations	τ (μs)
Quinine	$0 \text{ mg} \cdot \text{mL}^{-1}$	706.07
	0.1 mg·mL ⁻¹	667.29
	0.2 mg·mL ⁻¹	642.46
Quinidine	0 mg·mL ⁻¹	700.23
	0.1 mg·mL ⁻¹	672.77
	0.2 mg·mL ⁻¹	650.60