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Supporting Information

Sulfonatocalix[4]arene-Based Light-Harvesting Amphiphilic Supramolecular Assemblies for Sensing Sulfites in Cells

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Compound 1

Compound **1**, **2**, **3**, **SC4A** and **SC4A-C6** were synthesized according to the reported references, respectively.¹⁻⁴

TPE-4Py

Compound **1** (100 mg, 0.08 mmol) was dissolved in H_2O (5 mL) and the solution was stirred at room temperature for 1h. After that, NH_4PF_6 saturated solution (20 mL) was added dropwise, and the reaction mixture was stirred overnight. After that, filtered and drip washing with H_2O (100 mL) to achieve **TPE-4Py** (85 mg, 81%) as a yellow solid.

¹H NMR (400 MHz, DMSO) δ 8.97 (d, J = 6.6 Hz, 8H), 8.45 (d, J = 6.6 Hz, 8H), 8.01 (d, J = 8.3 Hz, 8H), 7.38 (d, J = 8.3 Hz, 8H), 4.31 (s, 12H).

HRMS-ESI (m/z): $[M+]^{4+}$ calcd for $C_{24}H_{25}IN_2O_2S$: 175.0886; found: 175.0889.

<u>SP</u>

To a solution of compound **3** (200 mg, 0.82 mmol) in absolute ethyl alcohol, compound **2** (299 mg, 1.63 mmol) and piperidine (16.2 μ L, 0.16 mmol) were added, and the reaction was refluxed overnight. Cooling to room temperature to obtain a large amount of precipitate, filtered and recrystallized in absolute ethyl alcohol to afford **SP** as a dark green solid (261 mg, 59.9%).

¹H NMR (400 MHz, DMSO) δ 8.64 (s, 1H), 8.40 (d, J = 7.8 Hz, 1H), 8.28 (d, J = 8.5 Hz, 1H), 8.05 (s, 2H), 7.85 (t, J = 7.8 Hz, 1H), 7.76 (t, J = 7.7 Hz, 1H), 7.59 (d, J = 9.1 Hz, 1H), 6.89 (dd, J = 9.1, 2.2 Hz, 1H), 6.70 (d, J = 2.1 Hz, 1H), 4.80 (q, J = 7.1 Hz, 2H), 3.55 (q, J = 6.9 Hz, 4H), 1.49 (t, J = 7.2 Hz, 3H), 1.17 (t, J = 7.0 Hz, 6H).

HRMS-ESI (m/z): [M+H]⁺ calcd for C₂₄H₂₅IN₂O₂S: 405.1631; found: 405.1635.



Figure S1. Fluorescence spectra of **TPE-4Py** (10 μ M) in H₂O/THF mixtures with different H₂O fractions and in the presence of **SC4A-C6** at the H₂O content of 99% (λ_{ex} = 365 nm; slit: 2.5/5 nm).



Figure S2. UV-vis absorption spectrum of **TPE-4Py** (14 μ M) and **TPE-4Py/SC4A-C6** ([**SC4A-C6**] =12 μ M, [**TPE-4Py**] = 14 μ M) in PBS buffer solution.



Figure S3. (a) Optical transmittance of **TPE-4Py** at different concentrations in PBS buffer solution. (b) Dependence of the optical transmittance at 600 nm at different concentrations of **TPE-4Py**. (c) Optical transmittance of **TPE-4Py** at different concentrations in the presence of **SC4A-C6** (10 μ M) in PBS buffer solution.



Figure S4. Optical transmittance of SC4A-C6 on different concentrations in the presence of TPE-4Py (12 µM).



Figure S5. Zeta potential of TPE-4Py/SC4A-C6 ([SC4A-C6] =12 μ M, [TPE-4Py] = 14 μ M).



Figure S6. Fluorescence spectra of **TPE-4Py** (12 μ M) in the presence **SC4A** (14 μ M); **SC4A-C6** (14 μ M), SDBS [sodium dodecyl benzene sulfonate (14 μ M)] in PBS buffer solution (λ_{ex} = 365 nm; slit: 2.5/5 nm).



Figure S7. Quantum yields of (a) TPE-4Py, (b) TPE-4Py/SC4A-C6, (c) TPE-4Py/SC4A-C6/SP and (d) SP in PBS buffer solution (pH 7.4, 10 mM).



Figure S8. Normalized emission spectrum of TPE-4Py and absorption spectrum of SP.



Figure S9. Fluorescence spectra of **SC4A-C6/SP** ([**SC4A-C6**] =12 μ M, [**SP**] = 216 nM) and **TPE-4Py/SP** ([**TPE-4Py**] =12 μ M, [**SP**] = 216 nM) in PBS buffer solution (10 mM, pH 7.4). (a) (λ_{ex} = 533 nm slit: 2.5/5 nm); (b) (λ_{ex} = 365 nm; slit: 2.5/5 nm).

a)	(D)
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Fit Range Fitting Rang 149 to 1837 chans Override low limit	Fit Range Fitting Range 143 to 1143 chans Override low limit
$R(t) = B_1 e^{(-t/\tau_1)} + B_2 e^{(-t/\tau_2)} + B_3 e^{(-t/\tau_3)} + B_4 e^{(-t/\tau_4)}$	$R(t) = B_1 e^{(-t/\tau_1)} + B_2 e^{(-t/\tau_2)} + B_3 e^{(-t/\tau_3)} + B_4 e^{(-t/\tau_4)}$
Fix Value / ns Std. Dev / ns Fix Value Std. Dev Rel %	Fix Value / ns Std. Dev / ns Fix Value Std. Dev Rel %
τ ₁ 2.7907 0.14056 B ₁ 534.280 46.2585 36.45	Τ ₁ □ 1.1109 0.05137 B ₁ □ 483.192 17.0373 24.15
τ ₂ 5.6184 0.16782 B ₂ 462.783 48.7563 63.55	τ ₂ 3.3103 0.04832 B ₂ 509.214 18.6421 75.85
τ ₃ Β ₃	
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τ ₄ B ₄ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ Δ	A 🗌 0.833
\$\mathcal{L}_4\$ \$\mathcal{B}_4\$ \$\mathcal{L}_2\$ \end{formula} \$\mathcal{A}\$ \$\mathcal{L}_2\$ \end{formula} \$\mathcal{L}_2\$ \end{formula} \$\colored{t}_2\$ \$\mathcal{L}_2\$ \end{formula} \$\mathcal{L}_2\$ \end{formula}	A [] 0.833 x ² : 1.015
Tq Bq	A 0.833 X ² : 1.015 Copy Results To Clipboard

Figure S10. Fluorescence decay profiles of TPE-Py/SC4A-C6 (a), and TPE-Py/SC4A-C6/SP (b) in PBS buffer solution ([TPE-Py] = 12 μ M, [SC4A-C6] = 14 μ M, [SP] = 216 nM) (λ_{ex} = 365 nm, λ_{em} = 350–800 nm).



Figure S11. Fluorescence spectra of **TPE-4Py/SC4A-C6** (black) and **TPE-4Py/SC4A-C6/SP** (red) in PBS (10 mM, pH 7.4) ([**TPE-4Py**] = 12 μ M, [**SC4A-C6**] = 14 μ M, [**SP**] = 216 nM) (λ_{ex} = 365 nm; slits: 2.5/5 nm).



Figure S12. Fluorescence spectra of **SP** (a) and **SC4A-C6/SP** (b) at different concentrations in PBS buffer solution. (c) The fluorescence intensity of **SP** and **SC4A-C6/SP** at 658 nm in different concentration. (λ_{ex} = 533 nm; slit: 2.5/5 nm).



Figure S13. Fluorescent intensity ratio ($I_{546 \text{ nm}}/I_{643 \text{ nm}}$) changes of **TPE-4Py/SC4A-C6/SP** as a function of the concentration of NaHSO₃ in PBS buffer (pH 7.4, 10 mM). Each spectrum was recorded after 8 min (λ_{ex} = 365 nm; slits: 2.5/5 nm).



Figure S14. ¹H NMR spectra of SP and TPE-4Py/SC4A-C6/SP in the presence of HSO₃⁻ (DMSO-d₆/D₂O).



Figure S15. (a) Fluorescence spectra of **SP** (14 μ M) upon gradual addition of HSO₃⁻ in PBS buffer. (b) Fluorescence spectra of **SC4A-C6/SP** ([**SP**] = 14 μ M, [**SC4A-C6**] = 12 μ M) upon gradual addition of HSO₃⁻ in PBS buffer solution. (c) The fluorescence intensity of **SP** at 643 nm upon gradual addition of HSO₃⁻ in PBS buffer. (D) The fluorescence intensity of **SC4A-C6/SP** at 658 nm upon gradual addition of HSO₃⁻ in PBS buffer solution. (λ_{ex} = 533 nm; slit: 2.5/5 nm).



Figure S16. (a) Fluorescence spectra of **TPE-4Py/SC4A-C6/SP** at different pH values. (b) Fluorescence spectra of **TPE-4Py/SC4A-C6/SP** toward HSO₃⁻ (0.4 mM) at different pH values. (c) Fluorescence spectra of **TPE-4Py/SC4A-C6/SP** at different temperatures. (b) Fluorescence spectra of **TPE-4Py/SC4A-C6/SP** toward HSO₃⁻ (0.4 mM) at different temperatures. (**TPE-4Py**] = 12 μ M, [**SC4A-C6**] = 14 μ M, [**SP**] = 216 nM) (λ_{ex} = 365 nm; slit: 2.5/5 nm).



Figure S17. Cell viability revealed by CCK-8 assays with different concentration of TPE-4Py/SC4A-C6/SP based on TPE-4Py concentration.



Figure S18. ¹H-NMR spectrum of TPE-4Py (400 MHz, DMSO-d₆).



Figure S19. HRMS spectrum of TPE-4Py.



Figure S20. ¹H-NMR spectrum of SP (400 MHz, DMSO-*d*₆).



Figure S21. HRMS spectrum of SP.

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Author Contributions

Z. Liu and Prof. Y. Liu conceived and directed the study. X. Sun measured the fluorescence quantum yield. X. Dai performed the cytotoxicity test. J. Li performed the Zeta potential experiment. P. Li synthesized the SC4A-C6. Z. Liu wrote the main manuscript and prepared the Figures. Prof. Y. Liu supervised the work and edited the manuscript. All authors reviewed the manuscript. All authors have given approval to the final version of the manuscript.