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Fig. S1 XRD patterns of S1 and S2 samples



Fig. S2 Raman spectrum of 50TeO₂-25WO₃-25Li₂O base glass



Fig. S3 (a) and (d) PL emission spectra, (b) and (e) color coordinates, (c) and (f) digital images for corresponding positions of the laser beam for bulk S₁ and S₂ samples, respectively



Fig. S4 (a) and (d) PL emission spectra, (b) and (e) CIE color coordinates, (c) and (f) digital images of glass samples for related excitation power densities for bulk S₁ and S₂ samples, respectively



Fig. S5 PL spectra and calculated UCQY values of (a) powder, (b) bulk forms of S₂ sample recorded at 77W/cm² power density

Sample	TeO ₂	WO ₃	Li ₂ O*	Yb ₂ O ₃	Ho ₂ O ₃	Tm ₂ O ₃				
Batch Composition (mol %)										
S_1	50	25	25	2	0.05					
S_2	50	25	25	2	0.05	0.05				
Batch Composition (wt %)										
S ₁	50.915	37.004	4.730	7.327	0.25					
S_2	50.906	36.997	4.729	7.325	0.25	0.18				
Final Composition (wt %)										
S ₁	50.09	37.36	4.73	7.82	-					
S_2	50.03	37.48	4.73	7.76	-	-				

Table S1 Chemical compositions of S_1 and S_2 samples

*: As Li could not be detected by XRF analysis, the final composition of Li_2O in S_1 and S_2 samples are considered the same as the starting composition.

-: Ho₂O₃ and Tm₂O₃ could not be detected due to their smaller quantities.

Sample	475 nm		545 nm		659 nm	
	Powder	Bulk	Powder	Bulk	Powder	Bulk
S ₁	-	-	1.92	2.33	1.73	2.23
S ₂	0.93	1.55	0.58	1.23	0.73	1.65
-: Not av	ailable					

Table S2 Calculated UCQY values (%) of powder and bulk forms of $S_1 \, and \, S_2$ samples under $77 W/cm^2$ power density for different wavelengths