

Electronic Supplementary Information

**Structure–Property Relationship in
Contrasting Aggregation-Induced Enhancement/Quenching of
Emission in Rigid Aromatic Molecules**

Yosuke Tani* and Takuji Ogawa

*Department of Chemistry, Graduate School of Science, Osaka University
Machikaneyama 1-1, Toyonaka, Osaka 560-0043, Japan
E-mail: y-tani@chem.sci.osaka-u.ac.jp*

Table S1. Summary of photophysical properties for **1–6** solutions and crystals

		Φ^a	τ^b / ns	error ^b / ns	χ^2 ^b	k_r^c / ns ⁻¹	k_{nr}^c / ns ⁻¹
1	in DCM	0.009	0.066	$\pm 3.1 \times 10^{-4}$	1.30	0.14	15
	in Toluene	0.008	0.073	$\pm 3.9 \times 10^{-4}$	2.35	0.11	14
	in EtOAc	0.003	0.039	$\pm 5.7 \times 10^{-4}$	2.87	0.077	26
	in MeCN	0.007	0.082	$\pm 3.6 \times 10^{-4}$	2.47	0.085	12
	Crystal	0.21	1.36	$\pm 6.4 \times 10^{-4}$	0.99	0.15	0.58
2	in DCM	0.049	0.33	$\pm 9.7 \times 10^{-4}$	6.23	0.15	2.9
	Crystal	0.12	1.53 ^d	$\pm 3.3 \times 10^{-2}$	1.93	0.078	0.58
3	in DCM	0.004	≤ 0.02	—	—	≥ 0.18	≥ 45
	Crystal	0.11	0.41 ^d	$\pm 4.8 \times 10^{-3}$	1.38	0.27	2.2
4	in DCM	0.32	0.86	$\pm 4.3 \times 10^{-4}$	1.03	0.37	0.79
	Crystal	0.59	1.76	$\pm 8.7 \times 10^{-3}$	1.03	0.34	0.23
5	in DCM	0.90	2.37	$\pm 2.0 \times 10^{-2}$	1.04	0.38	0.042
	Crystal	0.20	1.87	$\pm 1.7 \times 10^{-3}$	1.17	0.11	0.43
6	in DCM	0.98	2.56	$\pm 8.4 \times 10^{-3}$	0.97	0.38	0.0078
	Crystal	0.45	2.39	$\pm 6.1 \times 10^{-3}$	1.01	0.19	0.23

^aRelative quantum yield for solutions using quinine sulfate (0.60 in 0.05 M H₂SO₄) as a standard, or absolute quantum yield for crystals measured using an integrating sphere. ^bFluorescent lifetime τ , its error, and χ^2 obtained by a single exponential fitting of the decay curves upon excitation at 365 or 368 nm.

^cCalculated according to $k_r = \Phi / \tau$ and $k_{nr} = (1 - \Phi) / \tau$. ^dArea-weighted averaged lifetimes obtained by a biexponential fitting.

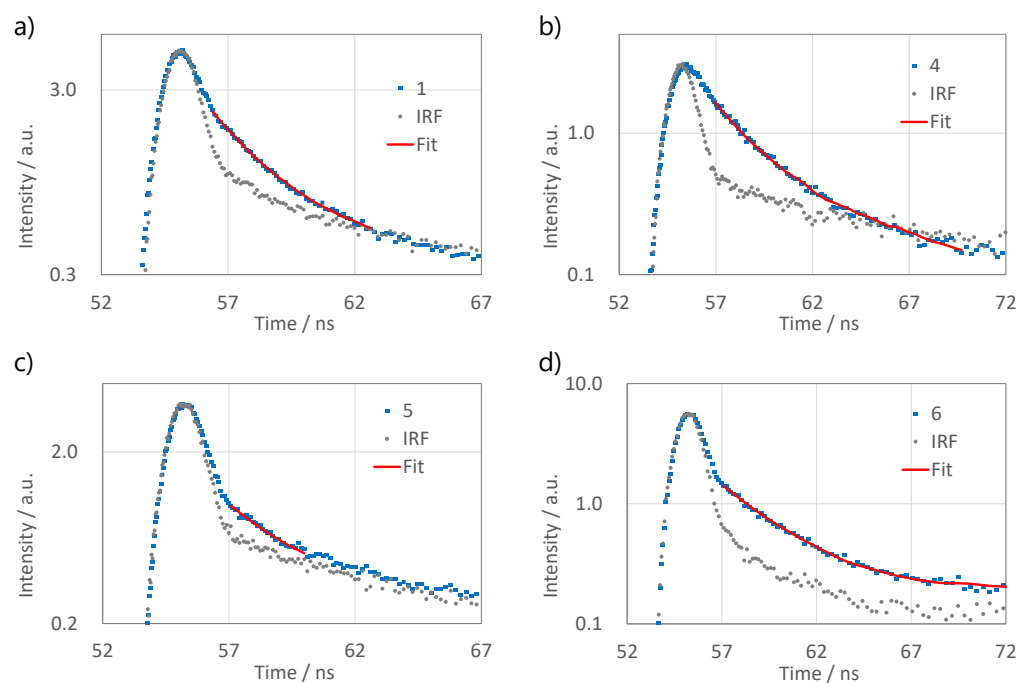


Figure S1. Fluorescence decay curves (blue dots) of **1** and **4–6** crystals in semi-log scale. Red lines denote the fitting curves.

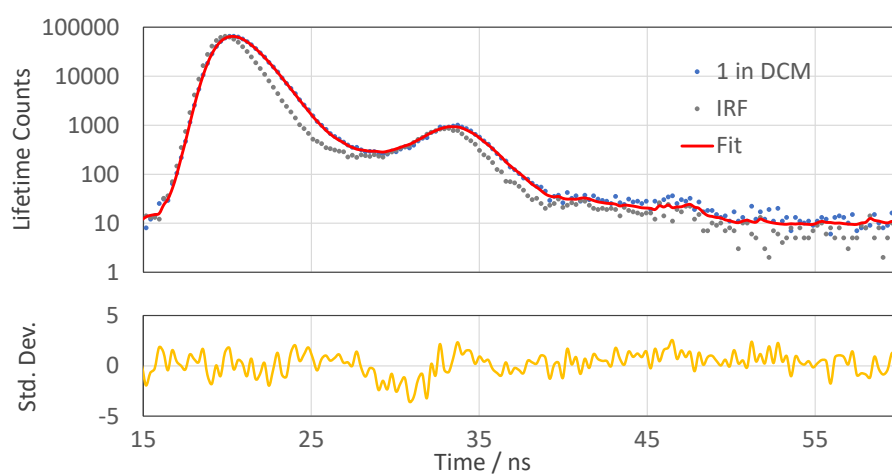


Figure S2. Top: Decay curve (blue dots) of the fluorescence intensity at 430 nm for **1** in dichloromethane in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

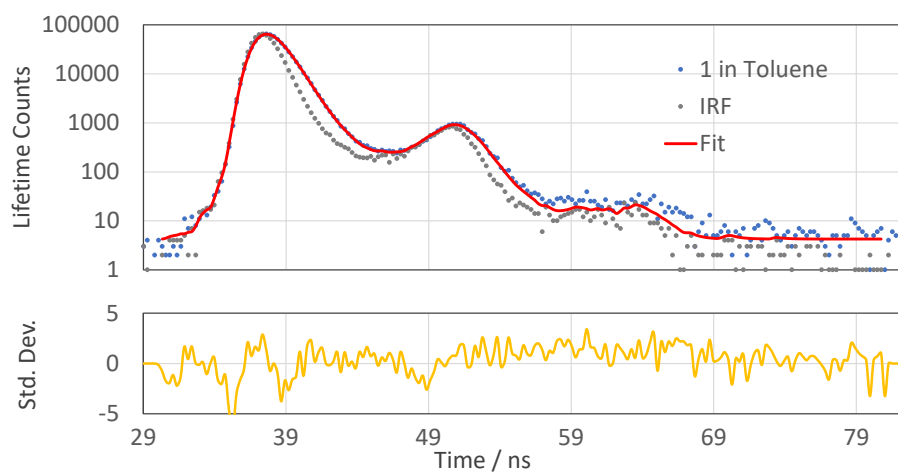


Figure S3. Top: Decay curve (blue dots) of the fluorescence intensity at 460 nm for **1** in toluene in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

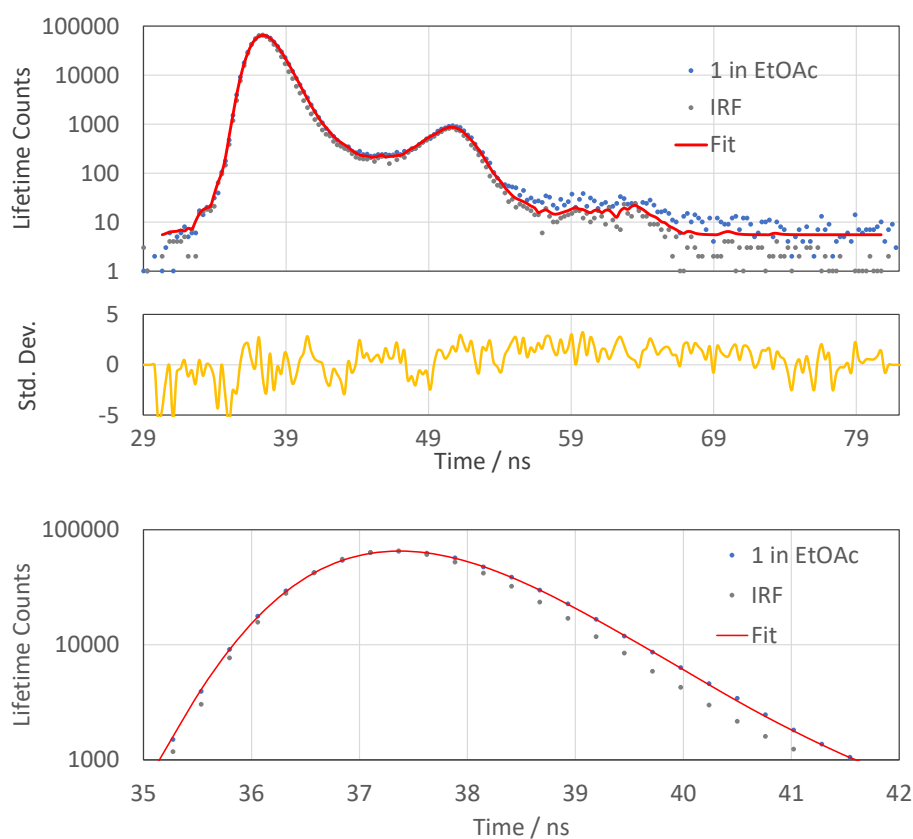


Figure S4. Top: Decay curve (blue dots) of the fluorescence intensity at 460 nm for **1** in ethyl acetate in semi-log scale. Red lines denote the fitting curves. Middle: Residue of the fit. Bottom: Enlarged graph near the peak top of the decay curve.

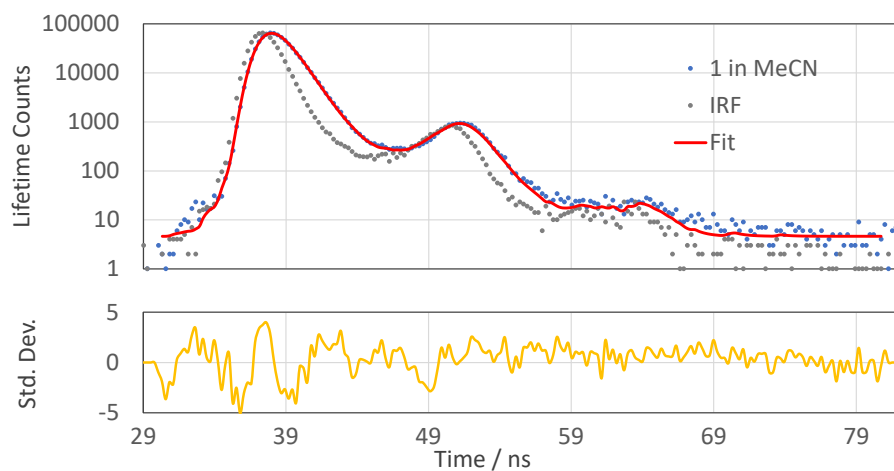


Figure S5. Top: Decay curve (blue dots) of the fluorescence intensity at 460 nm for **1** in acetonitrile in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

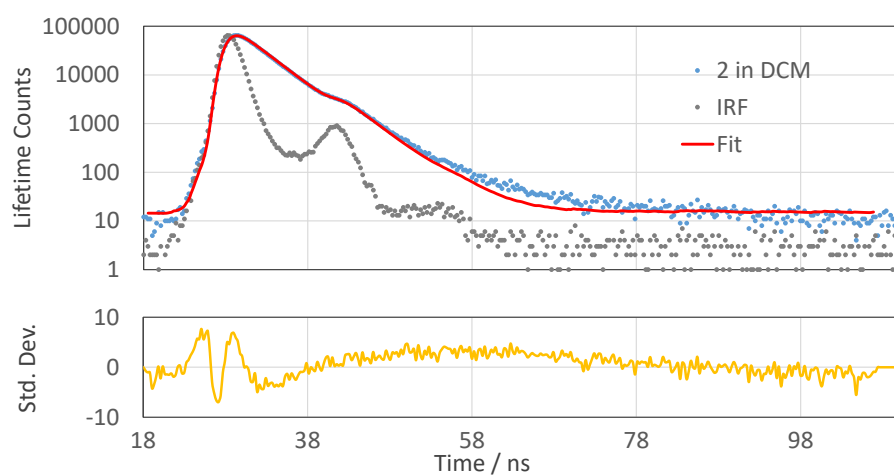


Figure S6. Top: Decay curve (blue dots) of the fluorescence intensity at 420 nm for **2** in dichloromethane in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

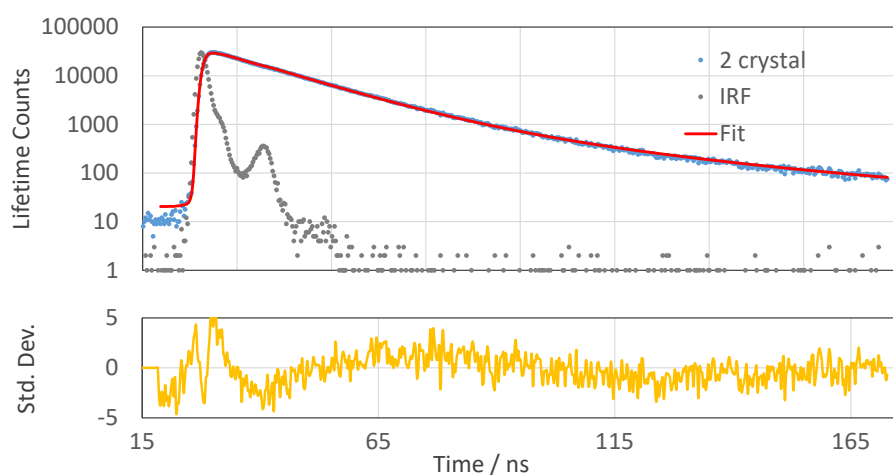


Figure S7. Top: Decay curve (blue dots) of the fluorescence intensity at 450 nm for **2** crystal in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

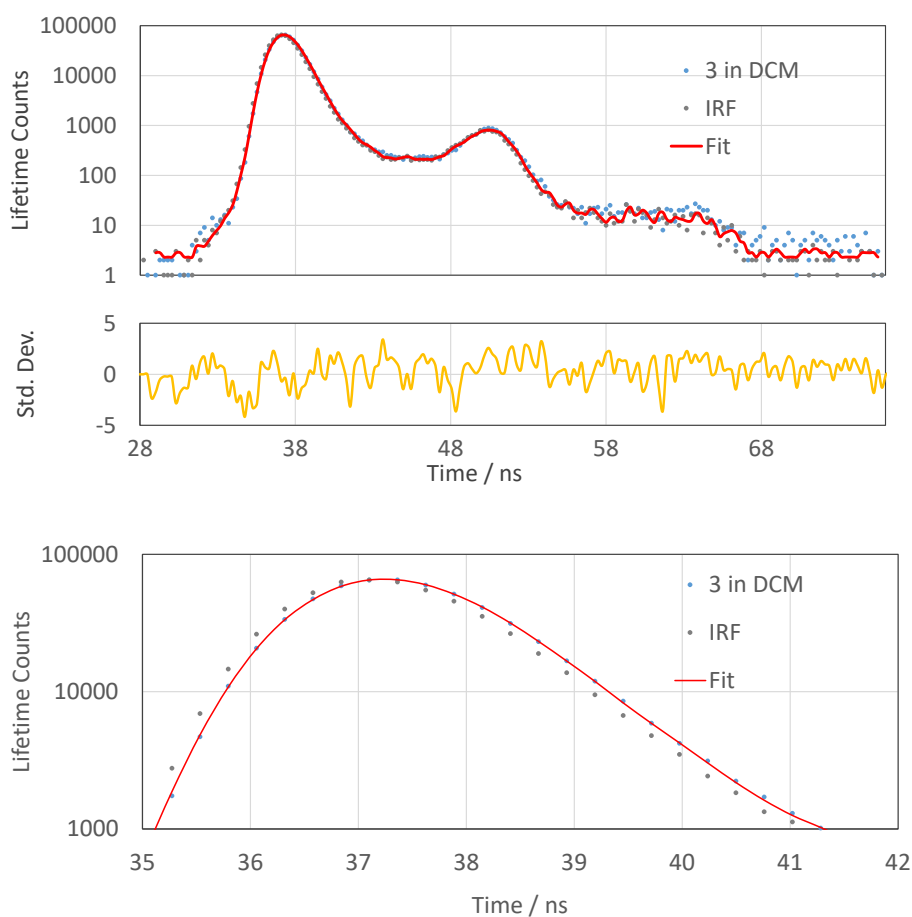


Figure S8. Top: Decay curve (blue dots) of the fluorescence intensity at 460 nm for **3** in dichloromethane (2×10^{-5} M) in semi-log scale. Red lines denote the fitting curves. Middle: Residue of the fit. Bottom: Enlarged graph near the peak top of the decay curve. The fit provided $\tau = 4.9 \pm 5.8$ ps.

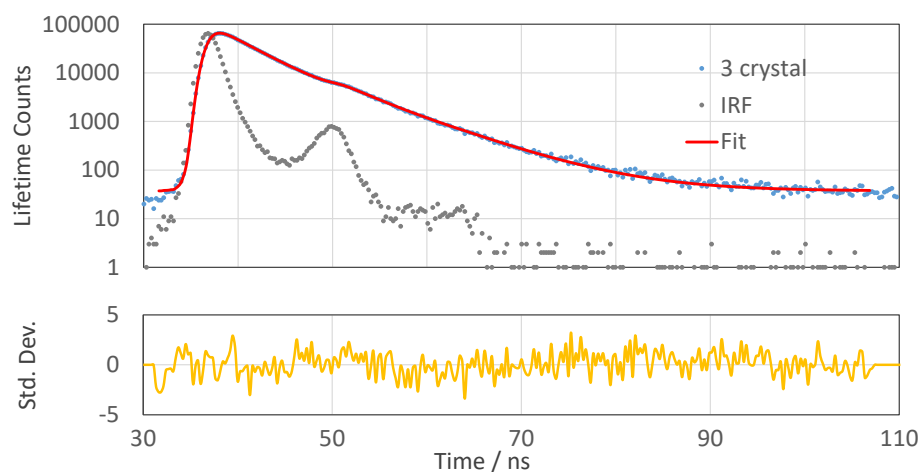


Figure S9. Top: Decay curve (blue dots) of the fluorescence intensity at 450 nm for **3** crystal in semi-log scale. Red lines denote the fitting curves. Bottom: Residue of the fit.

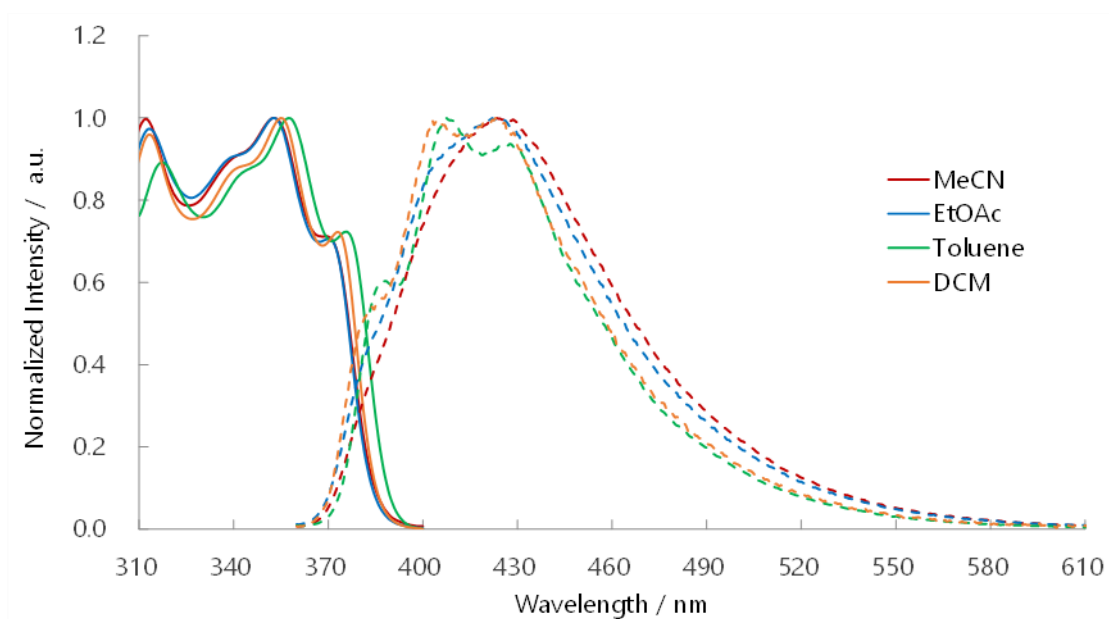


Figure S10. Normalized absorption (solid lines) and fluorescence (broken lines) spectra of **1** in acetonitrile (red), ethyl acetate (blue), toluene (green) and dichloromethane (orange).