

Supporting Information

Highly dispersed MoS_x nanodot-modified TiO₂ photocatalyst: Vitamin C-mediated synthesis and improved H₂ evolution activity

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SI-1 The AQE calculation

The apparent quantum efficiency (AQE) of the prepared photocatalyst is calculated via the following equation:

$$\begin{aligned} \text{AQE}(\%) &= \frac{\text{number of reacted electrons}}{\text{number of incident photons}} \times 100\% \\ &= \frac{\text{number of evolved H}_2 \text{ molecules} \times 2}{\text{number of incident photons}} \times 100\% \end{aligned}$$

The average power of the UV light (four 3-W 365 nm) was 22.4 mW/cm². Hence, the AQE of the TiO₂@C/MoS_x(0.7 wt%) photocatalyst can be calculated to be 3.94%.

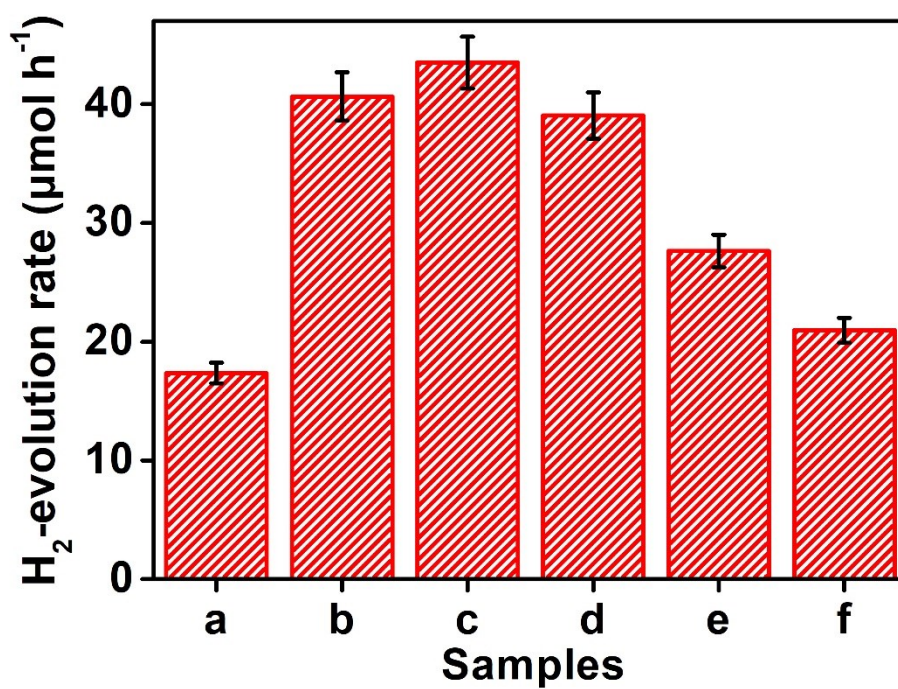


Fig. S1. The influence of VC amount on the photocatalytic H₂-evolution activities of TiO₂@C/MoS_x(1.0%) photocatalyst: (a) 0 mg, (b) 5 mg, (c) 9 mg, (d) 15 mg, (e) 20 mg, (f) 50 mg.

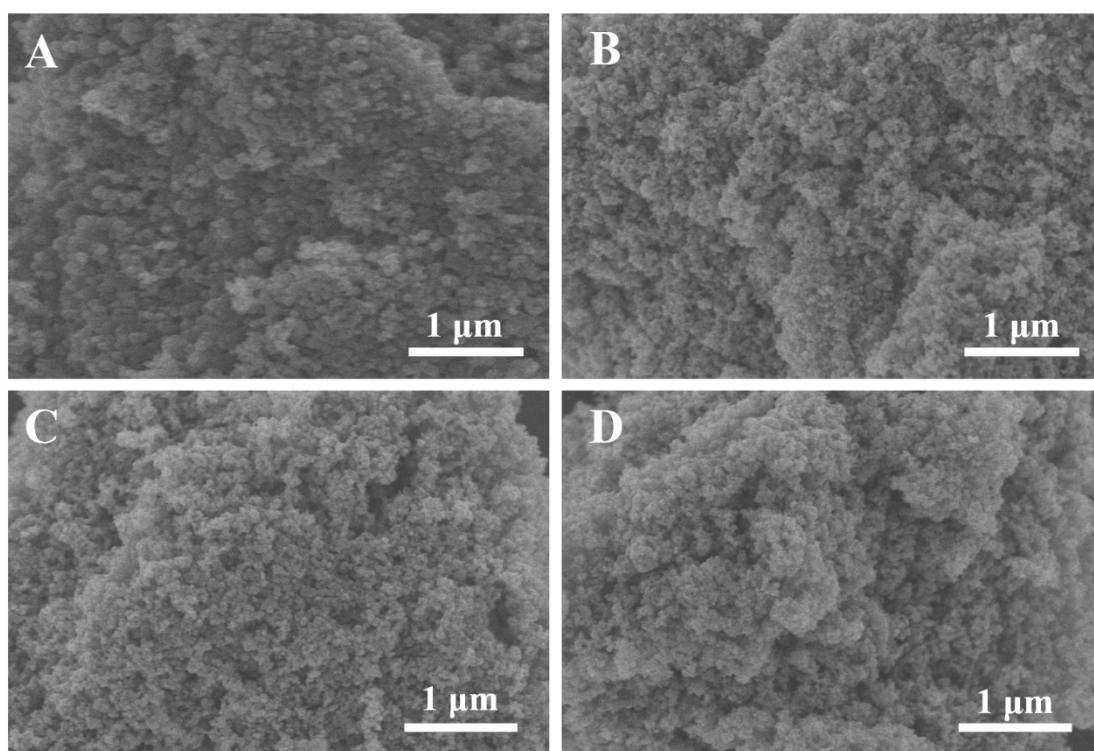


Fig. S2. FESEM images of different photocatalysts: (a) TiO_2 , (b) $\text{TiO}_2@\text{C}$, (c) $\text{TiO}_2@\text{C}/\text{MoS}_x(0.7\%)$, (d) $\text{TiO}_2/\text{MoS}_x(0.7\%)$.

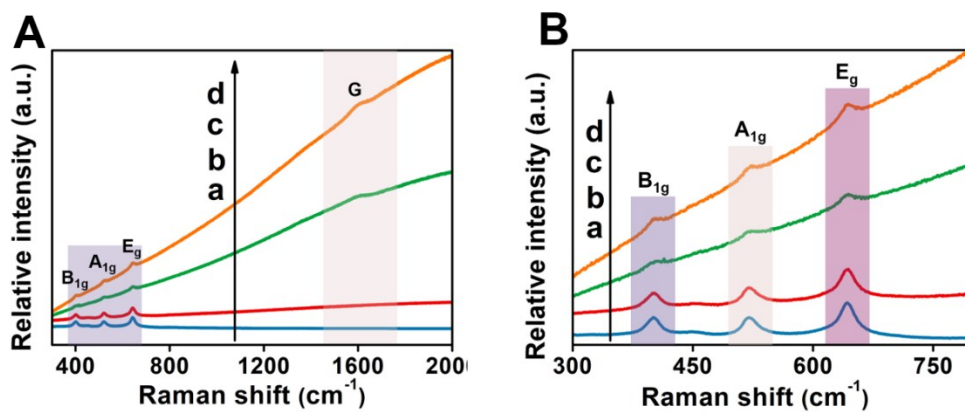


Fig. S3. (A) Raman spectra, (B) The enlarged Raman spectra of different samples: (a) TiO₂, (b) TiO₂/MoS_x(0.7%), (c) TiO₂@C/MoS_x(0.7%), (d) TiO₂@C.

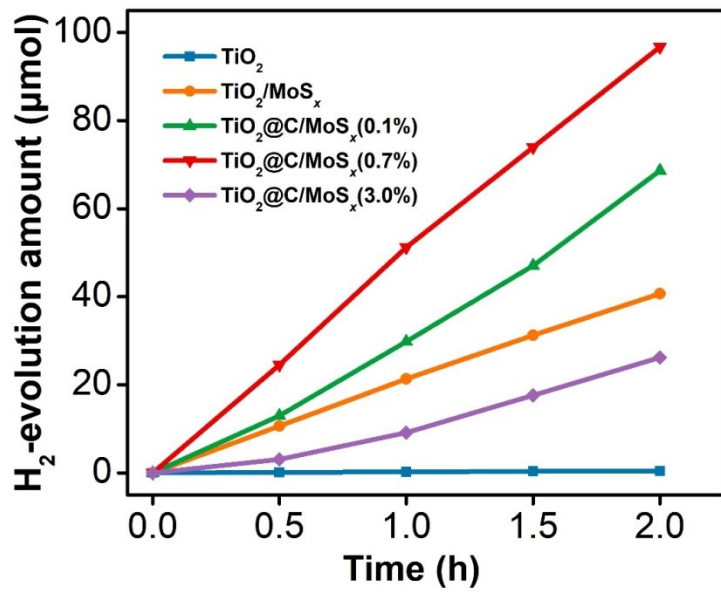


Fig. S4. Photocatalytic H₂-evolution activities of typical TiO₂ and MoS_x-modified photocatalysts.