

## Supplementary Information

### A rapid and highly sensitive paper-based colorimetric device for the on-site screening of ammonia gas

Kawin Khachornsakkul<sup>a</sup>, Hung Kuen-Hau<sup>b</sup>, Jung-Jung Chang<sup>b</sup>, Wijitar Dungchai<sup>\*a</sup>, Chih-Hsin Chen<sup>\*b</sup>

a) *Department of Chemistry, Faculty of Science, King Mongkut's University of Technology Thonburi, Prachautid Road, Thungkru, Bangkok, 10140, Thailand.*

b) *Department of Chemistry, Tamkang University, New Taipei City 25137, Taiwan.*

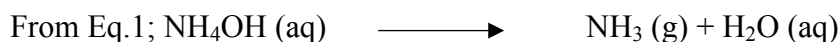
*Corresponding author: Asst. Prof. Dr. Wijitar Dungchai*

*E-mail: wijitar.dun@kmutt.ac.th*

*Corresponding author: Prof. Dr. Chih-Hsin Chen*

*E-mail: chc@mail.tku.edu.tw*

*Example calculation of 1.0 ppmv of NH<sub>3</sub> gas in this assay*



Mole of the NH<sub>4</sub>OH is equal to NH<sub>3</sub> gas according to Eq. 1.

NH<sub>3</sub> gas at 1.0 ppmv was prepared from 0.8 nM of aqueous NH<sub>3</sub> solution according to below calculation;

From Eq.2;  $\text{ppmv} = \mu\text{L of analyte gas} / \text{L of total air}$

In our experiment, we used the vial headspace, as a gas generation container, with 20.0 mL capacity.

$$\begin{aligned} \text{Then;} \quad \mu\text{L} &= 1.0 \text{ ppmv} \times 0.02 \text{ L} \\ &= 0.02 \mu\text{L} \end{aligned}$$

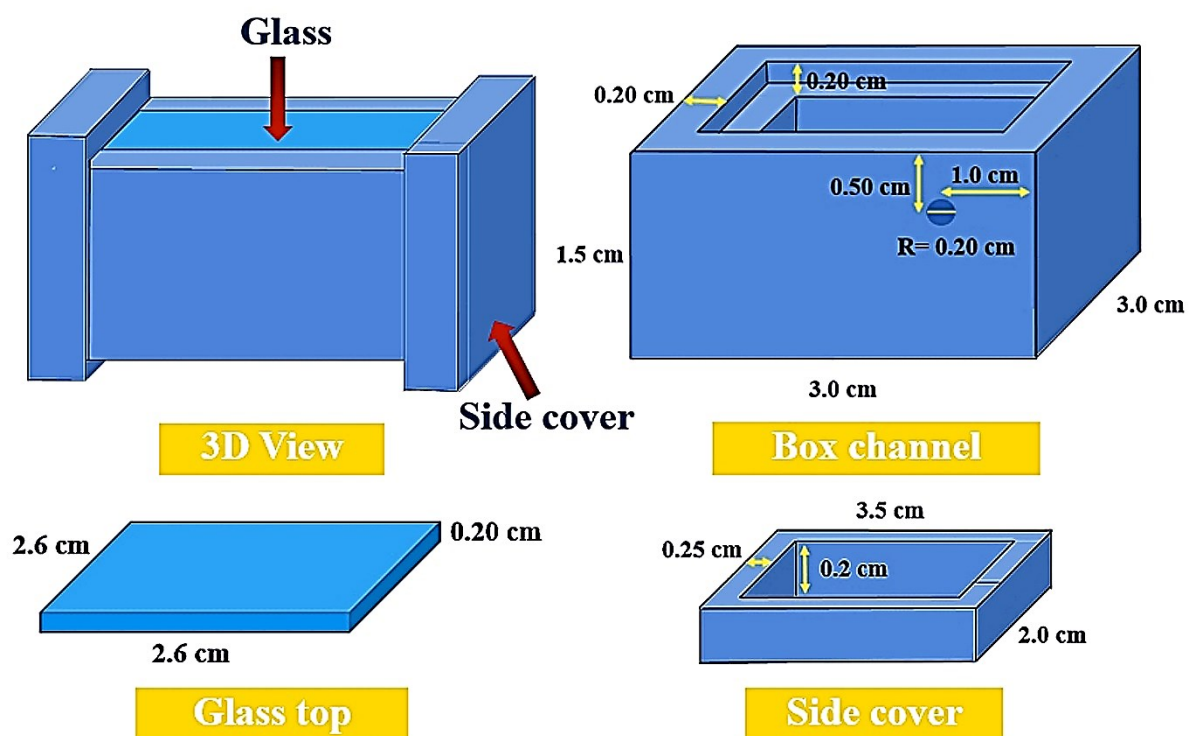
$$\text{Thus,} \quad \text{L} = 0.02 \times 10^{-6} \text{ L}$$

Next; we calculated the volume (V; L) of gas from aqueous solution by Eq. 3;

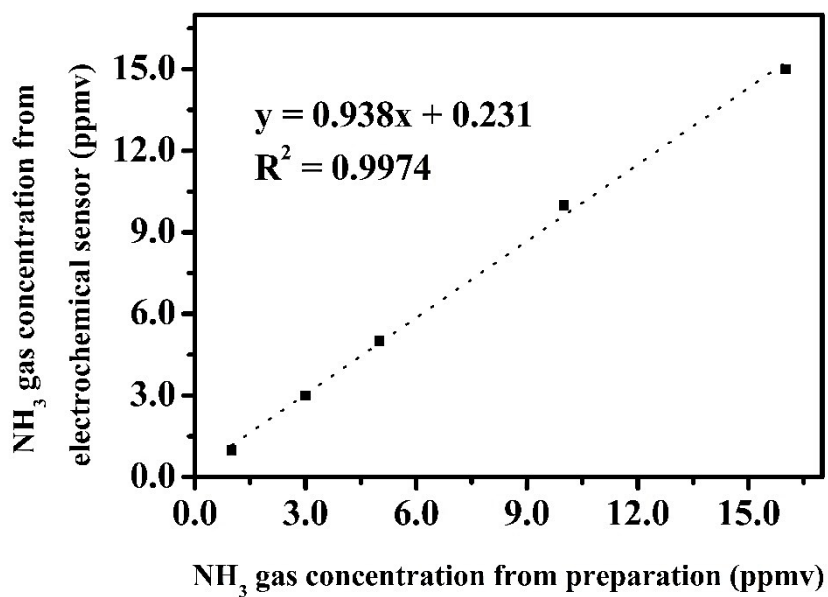
$$V = [\text{mol}_{\text{analyte}} \times 8.314 \text{ (L kPa/mol K)} \times 293.15 \text{ K}] / 101.325 \text{ kPa}$$

$$\begin{aligned} \text{Then;} \quad \text{mol}_{\text{analyte}} &= [0.02 \times 10^{-6} \text{ L} \times 101.325 \text{ kPa}] / [8.314 \text{ (L kPa/mol K)} \times 293.15 \text{ K}] \\ &= 0.83 \text{ nmol} \end{aligned}$$

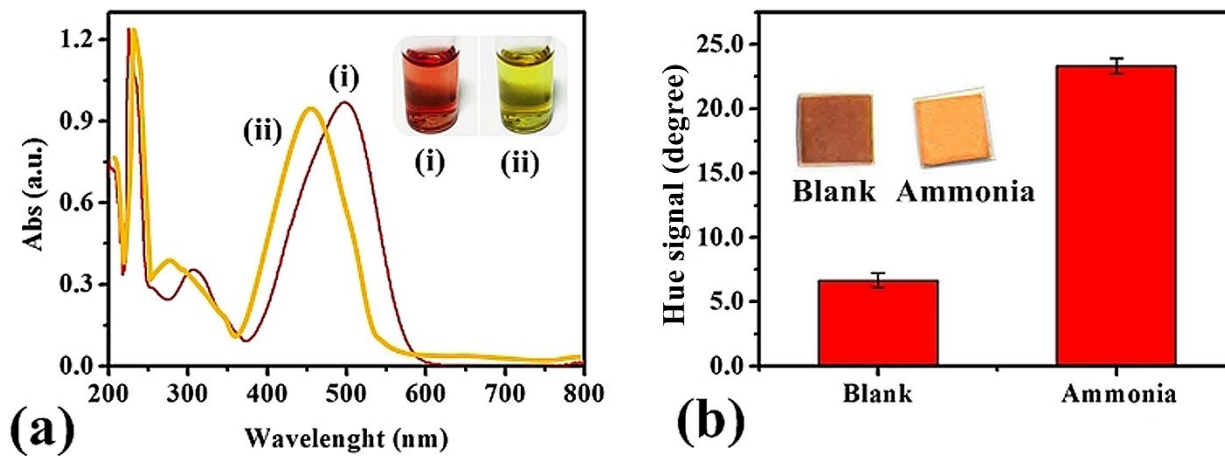
Therefore, we prepared the gaseous NH<sub>3</sub> at 1.0 ppmv by the introduction of 300.0 μL (0.30 mL) of aqueous NH<sub>3</sub> solution at 2.80 μM in the 20.0 mL of vial headspace and left to stand for 3 mins within a temperature as 25 °C and normal pressure, standard temperature pressure; STP.



**Scheme S1.** Demonstrated the fabricated device for NH<sub>3</sub> gas detection in the proposed method.



**Figure S1** The linear plot of NH<sub>3</sub> gas concentration between the proposed preparation and electrochemical sensor.



**Figure S2.** (a) The absorption spectra and corresponding images of methyl orange aqueous solution (i) without and (ii) containing 12.0 ppbv of  $\text{NH}_3$  gas. (b) comparison of the hue signal for (i) air and (ii) 12.0 ppbv of  $\text{NH}_3$  gas with the developed PADs in this method (n=3).