Supporting Materials

Nitrogen, Copper-doped Carbon Quantum Dots with Intrinsic Peroxidase-like Activity for Double-signals Detection of Phenol

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Fig. S1 FTIR spectrum of the N, Cu-CQDs.



Fig. S2 The UV-vis absorption spectra of -AAP + H_2O_2 + phenol in the presence (black line) and absence (red line) of the N, Cu-CQDs.



Fig. S3 The UV-vis absorption spectra of four different systems.



Fig. S4 The UV-vis absorption spectra of the -AAP + H_2O_2 +phenol in the presence of the Cu-free CQDs (black line) or the N, Cu-CQDs (red line).



Fig. S5 (A) Fluorescence emission spectrum of the N, Cu-CQDs (black line) and UVvis absorption spectrum of the N, Cu-CQDs/4-AAP/H₂O₂ /phenol solution (red line). (B) Time-resolved decays of the N, Cu-CQDs before (black line) and after (red line) addition of H₂O₂, 4-AAP and phenol.



Fig. S6 Dependency of catalytic activity of N, Cu-CQDs on synthetic temperature.



Fig. S7 Dependency of catalytic activity of N, Cu-CQDs on reaction time.



Fig. S8 Time-dependent absorbance of the N, Cu-CQDs/4-AAP/H₂O₂ system with the addition of phenol at room temperature.



Fig. S9 Absorption response of N, Cu-CQDs/phenol/4-AAP/H₂O₂ system at different pH values.



Fig. S10 4-AAP-dependent absorbance changes at 505 nm during the assay for peroxidase activity of the N, Cu-CQDs.



Fig. S11 The H_2O_2 -dependent absorbance changes at 505 nm during the assay for peroxidase activity of the N, Cu-CQDs.



Fig. S12 The UV-vis absorption spectra of the N, Cu-CQDs/4-AAP/H₂O₂/phenol (black line, photograph a), N, Cu-CQDs/4-AAP/H₂O₂/phenol/K₂Cr₂O₇ (red line, photograph b) and N, Cu-CQDs/4-AAP/H₂O₂/phenol/TBA (blue line, photograph c).