

## Electronic Supplementary Information

# Ready-to-use paraquat sensor using a graphene-screen printed electrode modified with a molecularly imprinted polymer coating on a platinum core

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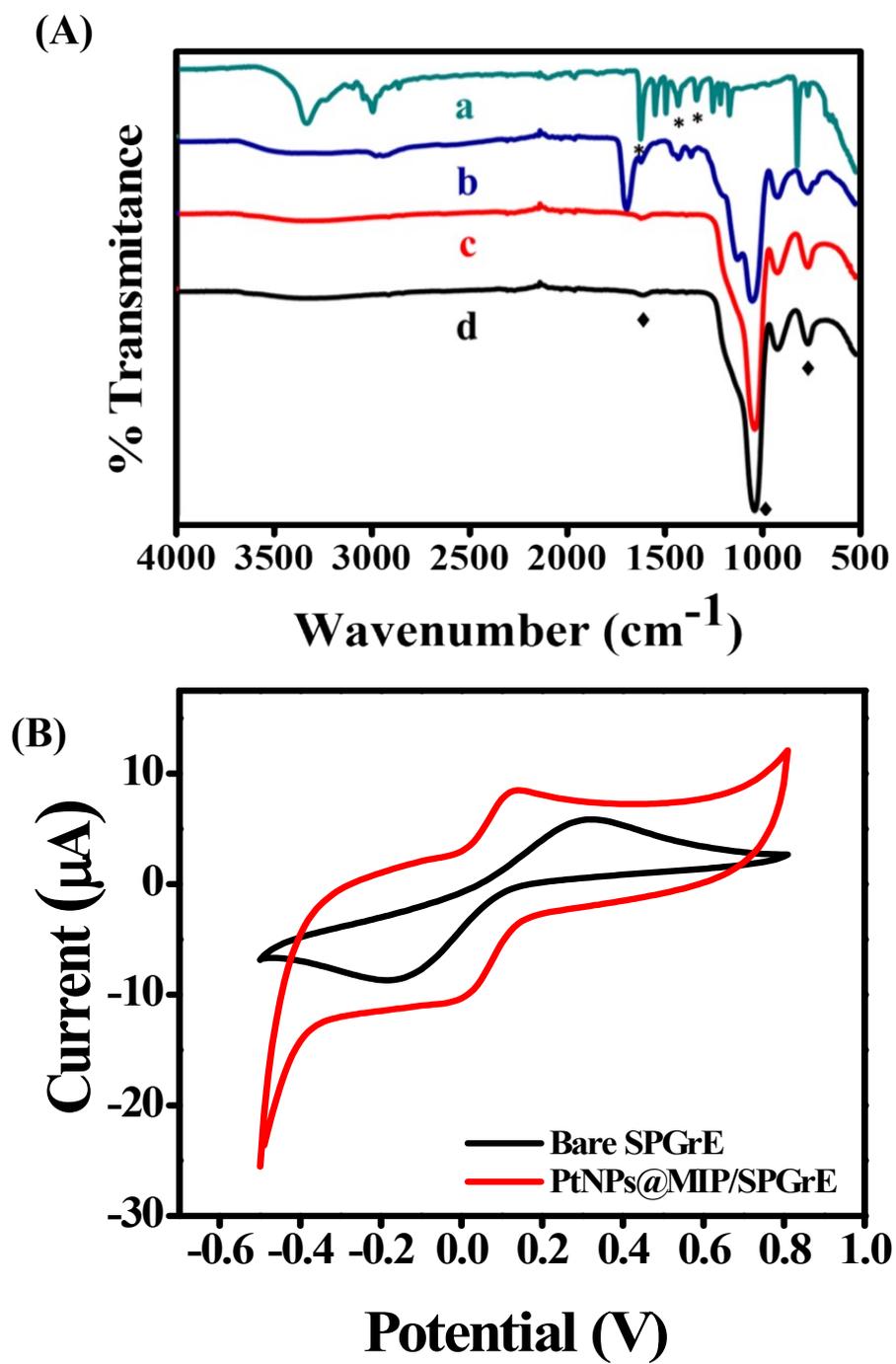
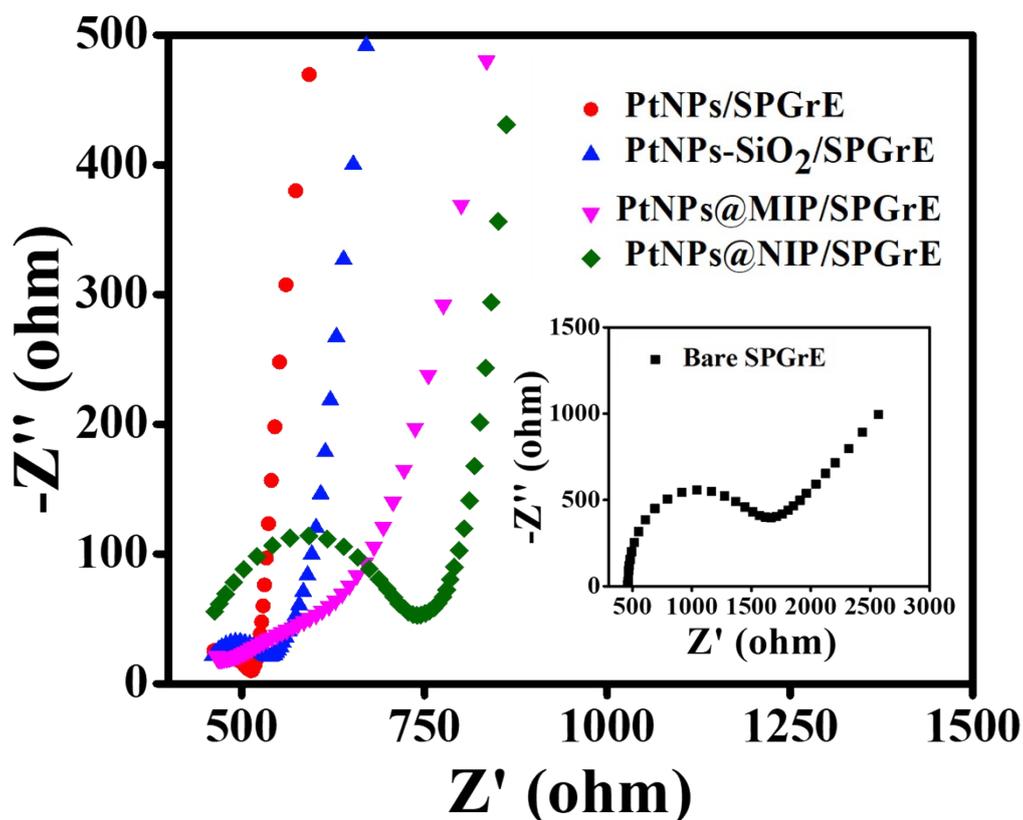


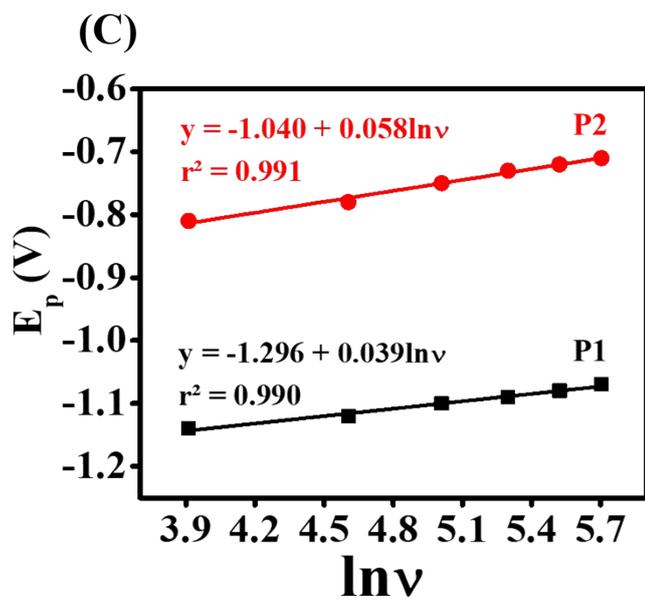
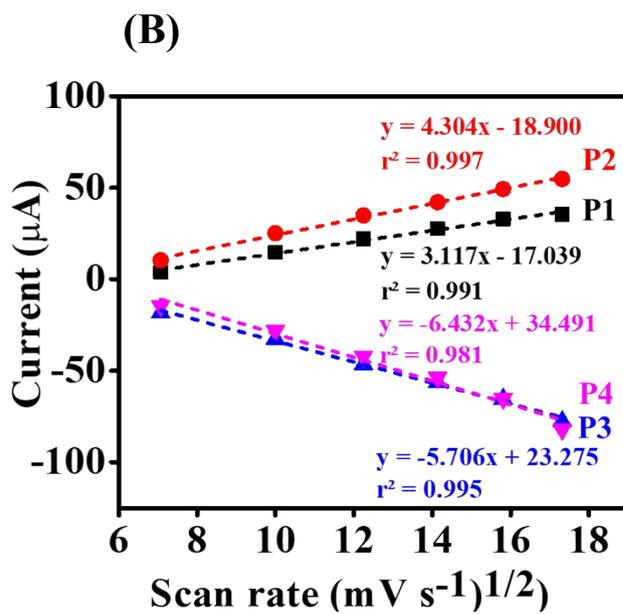
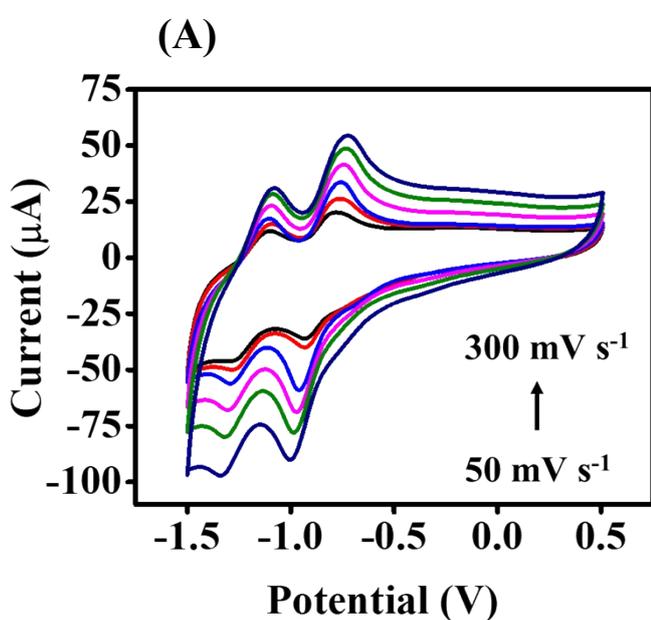
Fig. S1

(A) FT-  
IR  
spectra

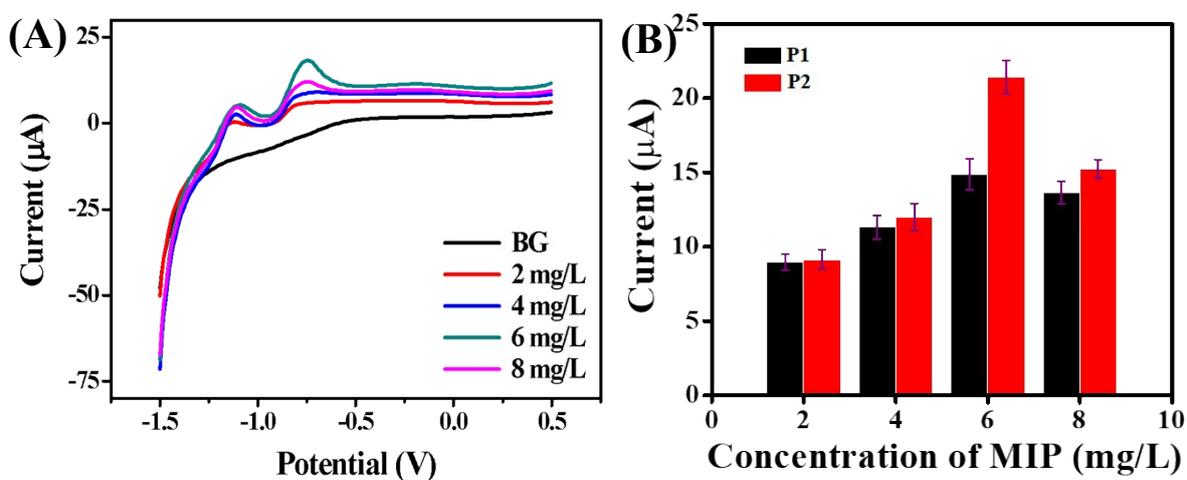
of a) paraquat, b) PtNPs@MIP before removal template, c) PtNPs@MIP after removal template, and d) PtNPs@NIP. (B) Cyclic voltammograms of 1.0 mM potassium ferricyanide on a bare and PtNPs@MIP/SPGrE in 0.1 M phosphate buffer (pH 7.0) supporting solution, the scan rate employed was 50 mV s<sup>-1</sup>.



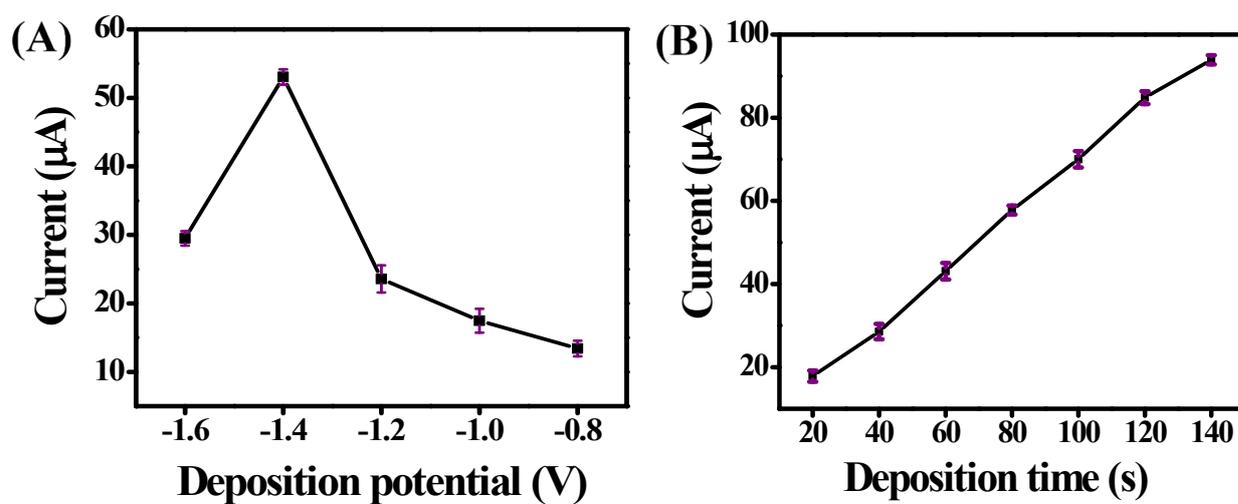
**Fig. S2** Nyquist plot observed for bare SPGrE (inset), PtNPs/SPGrE, PtNPs-SiO<sub>2</sub>@MIP/SPGrE, PtNPs@MIP/SPGrE and PtNPs@NIP/SPGrE in 0.1 M KCl containing 5 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup> over the frequency range 0.1 Hz to 100 kHz; amplitude = 10 mV. Inset: Randles equivalent circuit model; here  $R_s$ ,  $C_{dl}$ ,  $R_{ct}$  and  $Z_w$  are stand for the ohmic resistance of electrolyte ( $R_s$ ), double layer capacitance ( $C_{dl}$ ), electron transfer resistance ( $R_{ct}$ ) and Warberg impedance ( $Z_w$ ).



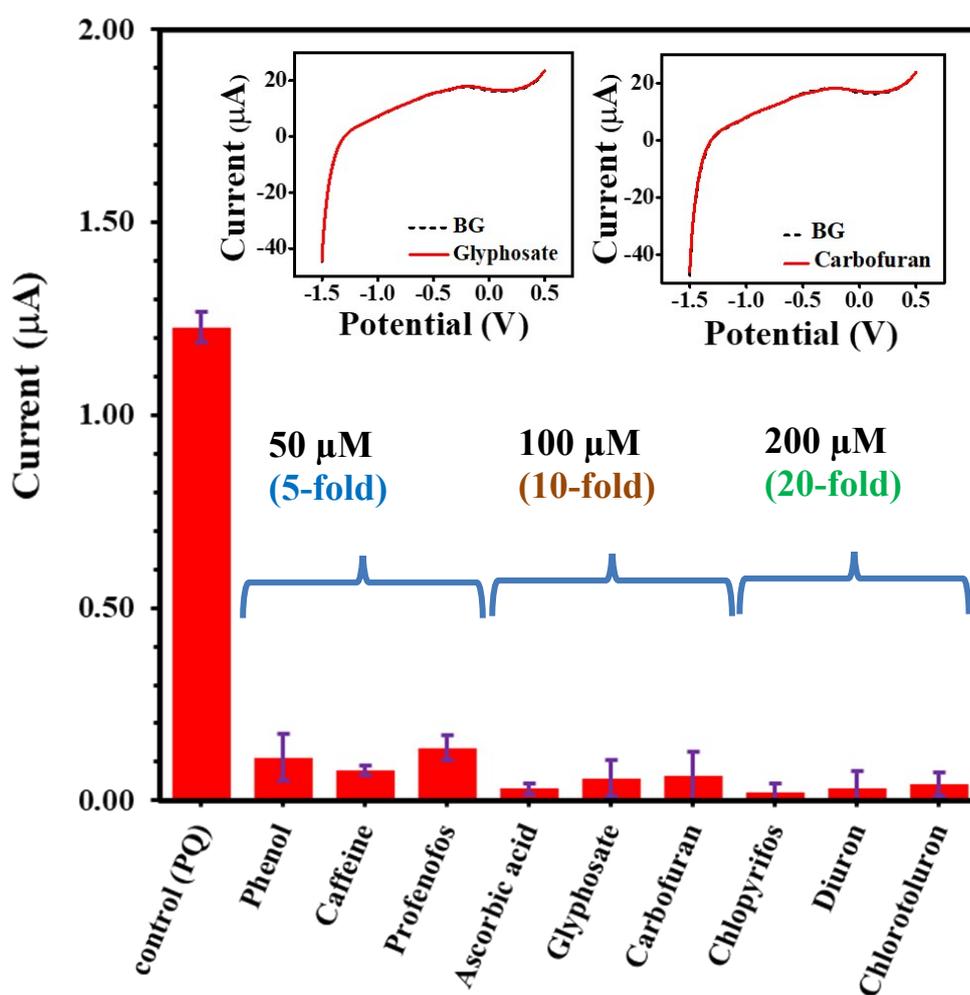
**Fig. S3** (A) Cyclic voltammograms of 500  $\mu\text{M}$  PQ on PtNPs@MIP/SPGrE in 0.1 M  $\text{K}_2\text{SO}_4$  solution at a scan rate ranging from 50 to 300  $\text{mV s}^{-1}$  and (B) Relationship between peak currents and square root of the scan rate.



**Fig. S4** (A) LSV signals of 500  $\mu\text{M}$  paraquat containing 0.1 M  $\text{K}_2\text{SO}_4$  solution on the PtNPs@MIP/SPGrE at a series of MIP loading values and (B) Effect of MIP loading on the current of paraquat determination.

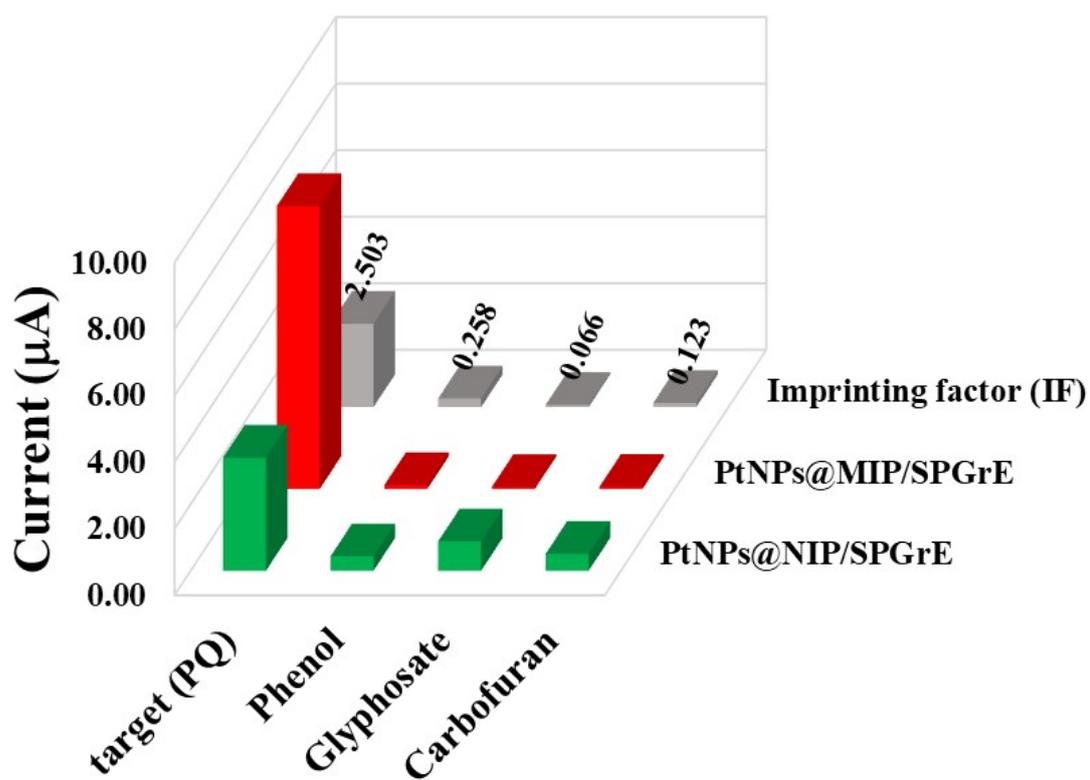


**Fig. S5** Effect of (A) deposition potential and (B) deposition time on the anodic stripping peak current of PQ (P2) at PtNPs@MIP/SPGrE. The paraquat concentration was 500  $\mu\text{M}$ ; the supporting electrolyte, 0.1 M  $\text{K}_2\text{SO}_4$  solution.

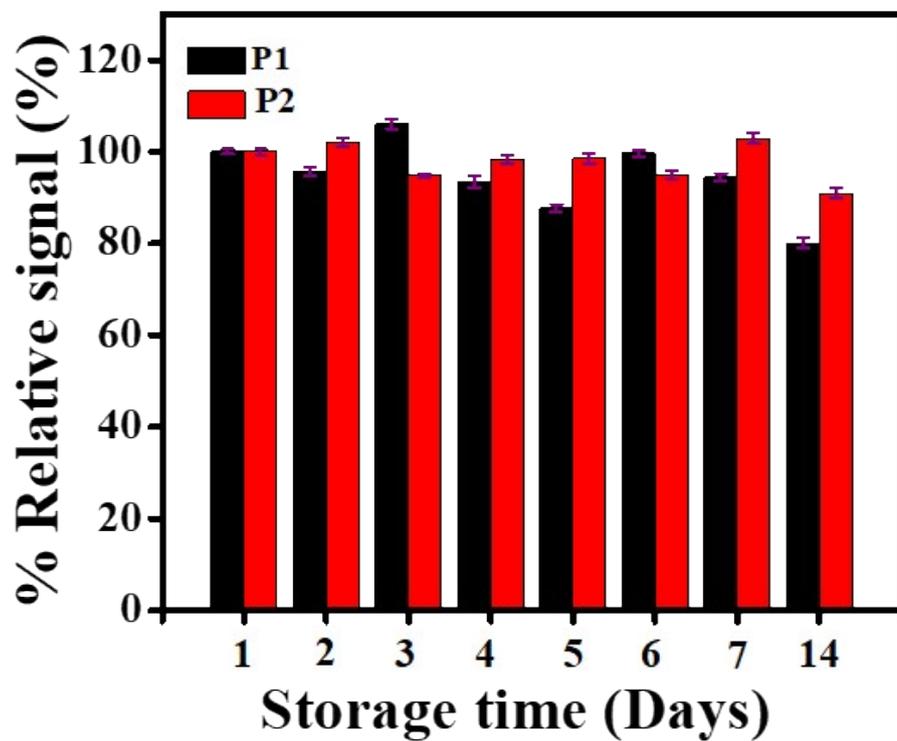


**Fig.S6** Current response of various species only, 10 µM PQ (control) and the currents of phenol, caffeine and profenofos at the concentration of 50 µM (5- fold), ascorbic acid, glyphosate and carbofuran at the concentration 100 µM (10- fold), and chlorpyrifos, diuron

and chlorotoluron at the concentration 200  $\mu\text{M}$  (20-fold). The insets are the voltammograms of 50  $\mu\text{M}$  glyphosate and carbofuran and the corresponding background, supporting electrolyte, 0.1 M  $\text{K}_2\text{SO}_4$  solution.



**Fig.S7** Selectivity pattern of PtNPs@MIP/SPGrE toward PQ, phenol, glyphosate and carbofuran at a concentration of 100  $\mu\text{M}$ .



**Fig.S8** stability during the storage period of 2 weeks in room temperature.