Cu-MOFs/Hemin : A bionic enzyme with excellent dispersity for

the determination of hydrogen peroxide released from living cells

Hong Cui*a,1, Shuaishuai Cuia,1, Siyuan Zhanga, Qiuju Tiana, Yunfeng Liua, Ping Zhanga, Mingxiu

Wang^a, Jialing Zhang^a and Xiangjun Li^{*b}

^a School of Public Health, Shanxi Medical University, 56Xinjian South Road, Taiyuan, 030001, China

^b School of Chemical Sciences, University of Chinese Academy of Sciences, 19A Yuquan Road, Beijing,

100049, China

^{*} Corresponding authors. *E-mail addresses*: lixiangj@ucas.ac.cn (X. Li), cuihong19872007@126.com (H. Cui).

¹ co-first authors

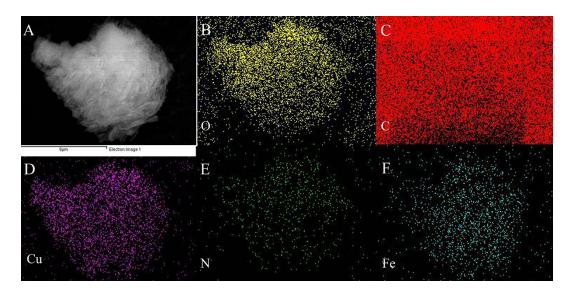


Fig. S1. The elemental mapping of Cu-MOFs/Hemin: (B) O, (C) C, (D) Cu, (E) N, and (F) Fe.

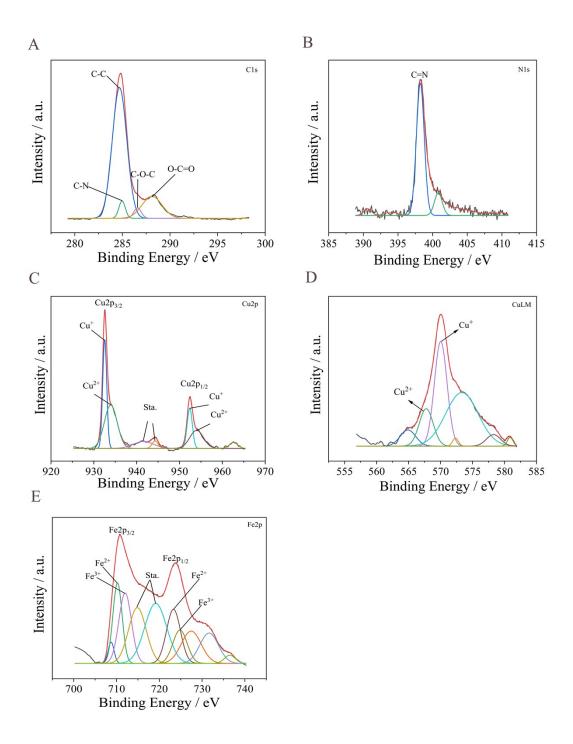


Fig. S2. The XPS spectra of Cu-MOFs/Hemin: (A) C1s, (B) N1s, (C) Cu2p, (D) CuLM, and (E) Fe2p.

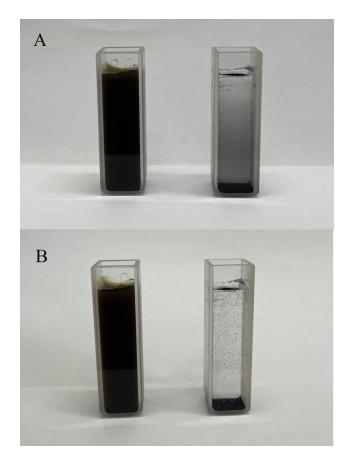


Fig. S3. (A) Images showing the aggregation and dispersion of Cu-MOFs/Hemin (lift) and Hemin (right) dissolved

in water, respectively. (B) Phenomenon after 30min placement.

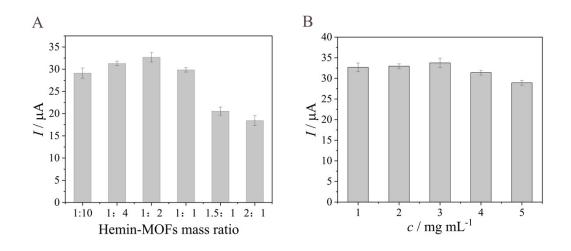


Fig. S4. (A) The comparison of reduction peak value of different mass ratio of Hemin and Cu-MOFs. (B) The histogram of reduction peak current with the Cu-MOFs/Hemin concentration.

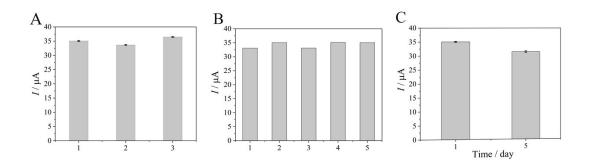


Fig. S5. (A) The current response of 3 different Cu-MOFs/Hemin/GCEs prepared under the same conditions in PBS containing 1.0 mM H₂O₂. (B) The current response of 6 repeated amperometric measurements using the same Cu-MOFs/Hemin/GCE in PBS containing 1.0 mM H₂O₂. (C) Stability test over 5 days in the air at room temperature.

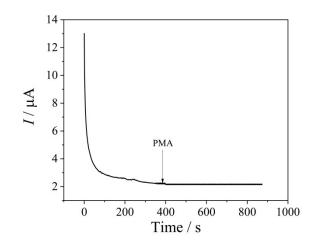


Fig. S6. Typical amperometric responses of the Cu-MOFs/Hemin/GCE for the reduction of H₂O₂ release from HepG2 cells induced by PMA with catalase.