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Electronic Supplementary Information

Super-assembled Silica Nanoprobes for Intracellular Zn Ion Sensing and Reperfusion Injury Treatment through in-situ MOF Crystallization

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Table S1. Surface properties of MSN-NH₂ and 2MI-P@MSN samples.

Table S2. Comparison with the recent work on Zn^{2+} probe. References

References



Figure S1. The synthesis process of PZn probe through a mild Schiff base reaction.



Figure S2. ¹H NMR spectrum of PZn probe.



Figure S3. (**A**) TEM image of P@MSN@ZIF8 incubated with zinc ion for 3 hours. (**B**) SEM image of P@MSN@ZIF8. (**C**) Size distribution of P@MSN@ZIF8.



Figure S4. (**A**) Fluorescence spectra of PZn treated with various metal ions $(Zn^{2+}, Ag^+, Al^{3+}, Ba^{2+}, Ca^{2+}, Co^{2+}, Cr^{3+}, Cu^{2+}, Fe^{3+}, Fe^{2+}, Hg^{2+}, K^+, Li^+, Mn^{2+}, Na^+, Ni^{2+}, Pb^{2+})$ and blank group. (**B**) Fluorescence ratio of PZn at 466 nm treated with multiple metal ions. F₀ was the fluorescence intensity in Zn²⁺ solution and F was the fluorescence emission intensity of the PZn in various metal ions solution, respectively. (**C**) The visual fluorescence color of PZn under ultraviolet light after the addition of various metal ions.



Figure S5. Fluorescence intensity of PZn at different stoichiometric ratios of Zn^{2+} and PZn. The total concentration was 1×10^{-4} M.



Figure S6. The relative fluorescence intensity of the PZn after adding 2MI at different concentration ratios and various times.



Figure S7. In vitro cytotoxicity assay of 2MI-P@MSN on SH-SY5Y cells. (A-C) The cell viability of SH-SY5Y cells incubated with various concentrations of 2MI-P@MSN for 3, 6, 12 hours, respectively. *P < 0.05 and **P < 0.01.



Figure S8. The cell viability of SH-SY5Y cells incubated with P@MSN and 2MI-P@MSN and treated with OGD/R for 6 hours. The control was the cell viability of SH-SY5Y cells incubated with 0 and 25 μ g/mL of P@MSN and 2MI-P@MSN. The 6 h was the cell viability of SH-SY5Y cells treated with OGD 3 hours and reperfusion for 6 hours. *P < 0.05 and **P < 0.01.

	MSN-NH ₂	2MI-P@MSN
$S_{BET}(m^2/g)$	897.398	267.023
$V_t(cm^3/g)$	0.947	0.632
D _{BJH} (nm)	2.11	1.96

Table S1. Surface properties of MSN-NH₂ and 2MI-P@MSN samples. The S_{BET} was the Brunauer–Emmett–Teller (BET) surface area, V_t was the pore volume, D_{BJH} was the Barrett-Joyner-Halenda (BJH) pore size distribution.

No.	Emission	Detection	Stoichiometric	Solvent	Function	Reference
	Wavelength	Limit (LOD)	Ratio			
1	410 nm	$5.2 \times 10^{-7} \mathrm{M}$	1:3	DMF/H2O	Zn ²⁺ detection	[S1]
				(6:4, v/v)		
2	450 nm	$3.5 \times 10^{-7} \mathrm{M}$	1:1	Ethanol	Zn^{2+} detection	[S2]
3	365 nm、555 nm	$5.0 \times 10^{-8} M$	1:1	Ethanol	Zn^{2+} detection	[S3]
4	405 nm	$4.2 \times 10^{-7} \mathrm{M}$	1:1	HEPES	Zn ²⁺ detection	[S4]
				buffer		
				with 0.01%		
				Triton X-100		
5	560 nm	$1.7 \times 10^{-8} \mathrm{M}$	1:1	CH ₃ CN/H ₂ O	Zn ²⁺ detection	[S5]
				(1:1, v/v)		
6	535 nm	$3.8 \times 10^{-8} \mathrm{M}$	1:1	MeOH/H ₂ O	Zn ²⁺ detection	[S 6]
				(4:1, v/v)		
7	466 nm	$1.6 \times 10^{-7} \mathrm{M}$	1:1	H ₂ O	Zn ²⁺ detection, cell	This work
					protection	

Table S2. Comparison with the recent work on Zn^{2+} probe.

The performance of the Zn^{2+} ion probe proposed in this work was compared with other reported methods in literatures. According to Table S2, relatively low LOD was achieved by 2MI-P@MSN in this work. Moreover, 2MI-P@MSN showed high selectivity and sensitivity towards detection of Zn^{2+} ion in aqueous solutions and living cells. In addition to intracellular zinc ion imaging, 2MI-P@MSN also protected cells from reperfusion injury. In summary, the proposed method in this work realized the integration of detection effects and protection effects into a multi-functional material framework and obtained satisfactory results.

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