# Naphthalene-Functionalized Resorcinarenes as Potent and Highly- Selective, Fluorescent Self-Quenching, Sensors for Kynurenic Acid

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## SUPPORTING INFORMATION

#### CONTENTS

Α.	GENERAL INFORMATION	2
B.	SYNTHESIS	2
C.	NMR SPECTROSCOPY	4
D.	ISOTHERMAL CALORIMETRY	8
E.	FLUORESCENCE SPECTROSCOPY	9
F.	UV VISIBLE SPECTROSCOPY	12
G.	COMPUTATION	15
H.	REFERENCES	24

### A. GENERAL INFORMATION

Kynurenic acid (KynA), tryptophan (Trp) and solvents used for syntheses were purchased from Sigma Aldrich or Oakwood Chemicals (Estill SC, USA). The receptors (**1-3**) were synthesized according to reported procedures.<sup>1,2</sup> All <sup>1</sup>H NMR, ITC, fluorescence titration and UV-visible spectroscopy experiments were carried out in 70% DI water/buffer, 30% DMSO at 298 K on a Bruker Avance 400, Microcal VP-ITC 2000, Horiba Jobin-Yvon Fluorolog 3, and Agilent Cary-100 spectrophotometers respectively.

## **B. SYNTHESIS**

The sodium sulfonated resorcinarene (1),<sup>1</sup> and N-alkyl ammonium resorcinarene salts  $(2,3)^2$  were synthesized according to reported procedures. *N*-Napthyl ammonium resorcinarene chloride (4) was synthesized according to the following procedure:





The resorcinarene **4a** was synthesized by reported procedure.<sup>3</sup> To a solution of **4a** (5.0 g, 6.9 mmol), excess formaldehyde solution 36% (28 mL) and ethanol (60 ml), 1-napthylmethylamine (4.3 mL, 29 mmol) in ethanol (15 mL) is slowly added and stirred at room temperature for 24 hours. The precipitate (**4b**) that separated is filtered, dried and used in the next step without further purification. To a solution of **4b** (0.5 g, 0.4 mmol), 1.5 mL of concentrated hydrochloric acid and 2 mL of H<sub>2</sub>O in 25 mL of isopropanol is added. The mixture is heated to boiling, refluxing for 4 hours. Water and formaldehyde are removed by azeotropic distillation with chloroform. The remaining isopropanol is evaporated, and the crude product triturated with diethyl ether to give receptor **4** without further purification (0.23 g, 42 %).



298 K.



**Figure S3**. <sup>13</sup>C NMR of *N*-naphthyl ammonium resorcinarene chloride **4** in  $90\%D_2O/10\%DMSO$  at 298 K.

**Receptor 4**: Melting point > 300°C. ESI-TOF-HRMS (Positive ion mode, sprayed from MeOH): m/zFound 699.34303 [M-4Cl-2H]<sup>+</sup>; calc. 699.3429, m/z found 1433.66197 [M-3Cl-2H]<sup>+</sup>; calc. 1433.6551. <sup>1</sup>H NMR (400 MHz in D<sub>2</sub>O/[D<sub>6</sub>]DMSO 9/1)  $\delta_{ppm}$ : 1.56 (m, 8H,CH<sub>2</sub>), 2.43 (q, J=7.66Hz, 8H, CH<sub>2</sub>), 3.72 (t, J=6.4Hz, 8HCH<sub>2</sub>), 3.85 (s, 8HCH<sub>2</sub>), 3.96 (s, 8HCH<sub>2</sub>), 4.41 (t, J=8.21Hz, 4H, CH) 6.60-7.7 (m, Ar, 32H): <sup>13</sup>C NMR: (100 MHz, 298 K in CD<sub>3</sub>OD)  $\delta_{ppm}$  = 30.9, 31.6, 35.9, 42.3, 47.2, 63.1, 110.4, 122.5, 126.6, 127.0, 127.3, 128.2, 129.1, 130.8, 131.7, 134.5, 151.6.

#### C. NMR SPECTROSCOPY

#### i) Proton NMR Experiment

For sample preparation, stock solutions of the receptors **1-4** (2 mM), KynA (2 mM), and Trp (2 mM) were prepared in 70%  $D_2O$  and 30% DMSO. For the pure **1-4**, KynA and Trp samples, 250  $\mu$ L of the stock solution was transferred to an NMR tube and diluted with 250  $\mu$ L of pure solvent mixture to make a 1 mM sample concentration.

For a 1:1 (**1-4**·KynA/Trp) mixture, 250  $\mu$ L of **1-4** and 250  $\mu$ L of KynA/Trp was pipetted into a clean NMR tube making 1:1 equimolar sample with 1 mM concentration of each component in the mixture.



**Figure S4.** <sup>1</sup>H NMR spectra (70%  $D_2O$ , 30%  $D_6$ -DMSO, 298 K) of sulfonated resorcinarene (1), KynA and equimolar mixtures of KynA and 1.



**Figure S5.** <sup>1</sup>H NMR spectra (70%  $D_2O$ , 30%  $D_6$ -DMSO, 298 K) of C3-hydroxyl resorcinarene salt (2), KynA, and equimolar mixtures of KynA and 2.



**Figure S6.** <sup>1</sup>H NMR spectra (70%  $D_2O$ , 30%  $D_6$ -DMSO, 298 K) of cyclohexyl resorcinarene salt (**3**), KynA, and equimolar mixtures of KynA and **3** 

### iii) 2D NOESY and DOSY NMR Experiment

2D NOESY NMR experiments were performed on a 500 MHz Bruker Avance III spectrometer at a temperature of 298 K. The mixing time was set to 200  $\mu$ s. The 90° pulse was determined to be 8.7  $\mu$ s. Spectra were recorded with 4k × 1k complex data points using 16 or 20 scans per t1 increment and 16 dummy scans at 25 and 70 °C. Diffusion ordered NMR spectroscopy (DOSY) measurements were performed using a Bruker Avance 500 MHz spectrometer equipped with a Great 1/10 pulsed gradient unit and a direct probe at 298 K. A LED29 pulse sequence (ledbpgp2s) was used for the diffusion experiments with a sine-shape pulsed gradient duration  $\delta$  (P30) of 1.0–1.2 ms incremented from 0.68 to 32.4 G cm<sup>-1</sup> in 16 steps. The pulsed gradient separation  $\Delta$  (D20) was 50 ms, the spoil gradient (P19) was set to 1100  $\mu$ s, and the eddy current delay (D21) was 5 ms. The reported diffusion coefficients were obtained using the T1/T2 relaxation module in TopSpin 4.1.4 software.





#### ii) Proton NMR Competition Experiment

NMR competition titration experiments were carried out in 70%  $D_2O$ , 30%  $D_6$ -DMSO at 298 K on a Bruker Avance 400 MHz spectrometer. 250 ul of 2mM pure KynA and Trp were charged into the sample tube. Due to the limited solubility of KynA, only a fraction of the KynA dissolved. Careful integration reveals a 1:8 equivalent KynA:Trp mixture which was recalculated to give a final concentration ratio of 0.125:1. 5µl aliquots of 20mM receptor **4** was titrated in the sample and the <sup>1</sup>H NMR recorded. Up to 6 equivalents (0.2, 0.4, 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0, 2.4, 3.0, 6.0 equivalents) of **5** was added to KynA and Trp.



Figure S8. Competitive NMR titration in  $70\%D_2O$ ,  $30\% D_6[DMSO]$  at 298 K. (I) a) receptor **4**, b) 1:1 equivalent of **4** and Trp, c) pure Trp, d) addition of 1 equivalent of KynA into a 1:1 mixture of **4** and Trp, e) pure KynA. (II) a) receptor **4**, b) 1:1 equivalent of **4** and KynA, c) pure KynA, d) addition of 1 equivalent of Trp into a 1:1 mixture of **4** and KynA, e) pure Trp. Dashed lines give an indication of the signal changes in ppm.

#### D. ISOTHERMAL CALORIMETRY

A VP-ITC instrument by MicroCal was used to determine the molar enthalpy ( $\Delta$ H) of complexation. Subsequent fitting of the data to a 1:1 binding model using Origin software provides association constant (K), change in enthalpy ( $\Delta$ H) and entropy ( $\Delta$ S). The ITC experiment was carried out by filling the sample cell with the guest (0.25 mM), filling the syringe with the receptor (5.0 mM), and titrating via computer-automated injector at 298 K. Blank titrations into plain solvent were also performed and subtracted from the corresponding titration to remove any effect from the heats of dilution from the titrant.



Figure S9: ITC traces of the titration of receptors with KynA (**a**-**e**) and Trp (**f**) at 298 K. (a) KynA@**1**, (c) KynA@**3**, (d) KynA@**4** in 70% H<sub>2</sub>O/30% DMSO were fitted to a one set of site binding model. (b) KynA@**2** all in 70% H<sub>2</sub>O/30% DMSO was fitted to two set of sites binding model. (e) KynA@**4** and (f) Trp@**4** in 70% Tris buffer pH 7.4/30% DMSO were fitted to one set of site binding model.

### E. FLUORESCENCE SPECTROSCOPY

Fluorescence titration experiments were carried out in 70%  $H_2O$ , 30% DMSO at 298 K. 2.0 ml of 0.1 mM of KynA was pipetted into the sample cuvette. 5  $\mu$ l aliquots of 20.0 mM receptor **4** at 0.2, 0.4, 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0, 2.4, 3.0, 6.0 equivalents were titrated in the sample and emission spectrum recorded.

In the reverse titration, 2.0 ml of 0.1 mM receptor **4** was pipetted into the sample cuvette. 5µl aliquots of 10.0 mM KynA (at constant sonication) at 0.2, 0.4, 0.6, 0.8, 1.0, 1.2, 1.4, 1.6, 1.8, 2.0, 2.4, 3.0, 6.0 equivalents were titrated in the sample and emission spectrum recorded.



#### **REVERSE TITRATION**

**Figure S10.** Steady state emission spectra ( $\lambda_{exc}$  = 475nm) of receptor **4** in H<sub>2</sub>O/DMSO (70%/30%, 298 K) at different equivalents of KynA: 1) 0.00, 0.30, 0.60, 0.90, 1.19, 1.49, 1.78, 2.08, 2.37, 2.66, 2.96, and 3.25.



**Figure S11.** Steady state emission spectra ( $\lambda_{exc}$  = 350nm) of 3mM receptor **4** in H<sub>2</sub>O/DMSO (70%/30%, 298 K)



**Figure S12.** a) Stern-Volmer plot of first 14 titration points to reveal positive curvature, b) full stern-volmer plot of extended titration points.



**Figure S13.** Modified Stern-Volmer regression analysis plot of Fo/F against increasing concentrations (logarithmic) of receptor 4.



Figure S14. Plot of Kapp vs [receptor 4] showing a linear plot with a specific slope.



Figure S15. Fitting of percentage quenching of receptor 4 with concentration of KynA.

### F. UV VISIBLE SPECTROSCOPY



UV Titration of Trp into receptor 4

**Figure S16.** UV-Vis titration of increasing concentrations of Tryptophan (Trp, mM) into a solution of receptor **4** in 50:50  $H_2O/DMSO$  system, monitoring absorbance of receptor **4**.



Figure S17. Output bindfits from absorbance of receptor 4 at 510nm, 514nm, 515nm and 520nm.

Table ST. Output III qualit	Table	S1.	Output	fit	qualit
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	510nm	514nm	515nm	520nm	Total
Root Mean Square (RMS)	0.020	0.020	0.020	0.020	0.020
Covariance	0.014	0.0134	0.014	0.014	0.014



**Figure S18.** UV-Vis titration of increasing concentration of 60mM KynA in 50:40:10 DMSO/H<sub>2</sub>O/Human Serum, into a solution of 3mM receptor **4** in 90:10 H<sub>2</sub>O/DMSO system.



Absorbance response curve at 514nm

**Figure S19.** Linear response plot of absorbance at 514nm vs equivalence of host/guest in 10% HSA samples.



**Figure S20.** a) UV-Vis titration of increasing concentration of 50:40:10 DMSO/H<sub>2</sub>O/Human Serum (HS), into a blank solution of 90:10 H<sub>2</sub>O/DMSO system. b) UV-Vis titration of increasing concentration of 50:40:10 DMSO/H<sub>2</sub>O/Human Serum, into a solution of 3mM receptor **4** in 90:10 H<sub>2</sub>O/DMSO system in the absence of KynA.

### G. ADDITIONAL COMPUTATIONAL DETAIL

The computational part of this study was performed with the Gaussian 16 Rev. C suite of programs.<sup>4</sup> The Symmetry Adapted Perturbation Theory (SAPT) analysis were performed using the Psi4 package.<sup>5</sup> The structural optimizations confirmed by frequency calculations were performed at the B3LYP-D3(BJ) level with Grimme's dispersion correction, <sup>6</sup> and the Becke-Johnson damping functions employed. The def2-SVP basis set was used throughout. The solvent optimization using SCRF=(PCM,Solvent=Water) solvation model leads to very similar geometry of the gas phase optimized geometry of the receptor **4** with O---N and O---O distances slightly longer by ca. 0.01-0.02Å, respectively. Cl<sup>-</sup> counterions are also slightly more distant from the respective protons by 0.1Å from H-N and 0.03Å from H-O compared to gas-phase data. It is not unexpected since the dielectric continuum favors more charge separation (the dipole moment of solvent-optimized structure increases to 14.1D). The xyz coordinate files of both structures are shown below along with the host-guest, KynA@**4**, gas-phase optimized structure. The gas-phase thermodynamic results lead to the following values

Thermodynamic calculations at 298°K					
	KynA@ <b>4</b>	Receptor 4	KynA	∆G°, kcal/mol	T∆S, kcal/mol
G° <i>,</i> a.u.	-6419.605084	-5754.391504	-665.178367	-22.10	
S, cal/mol K 386.818 343.449 27.129 4.84					

The SAPT approach defines the interaction energy as a following sum of the electrostatic (elst) exchange (exch), induction (ind), exchange-induction (ex-ind), dispersion (disp), and exchange-dispersion counterpart (ex-disp).

$$E_{SAPT} = E_{elst} + E_{exch} + E_{ind,r} + E_{ex-ind,r} + E_{disp} + E_{ex-disp} + \delta_{HF}$$
(S1)

where "r" in the subscripts refers to the response-inclusive treatment of induction and the last term stands for higher order induction and residual effects. In this work we employ a SAPT variant based on the zero<sup>th</sup>-order wavefunction obtained within the Hartree-Fock approximation. Evaluation of energy terms utilizes the density fitting and Laplace transformation techniques.<sup>7</sup> The induction term (Tables 2 and 3 in the main manuscript) has been combined with its exchange and residual contributions, while the dispersion has been combined with its exchange counterpart.

Another way of confirming the stability of the optimized host-guest complex, is by evaluating the interaction energy of host-guest complex as a difference of the total energies of the supermolecule (KynA@4) and its constituents,

$$E_{int} = E(KynA@\mathbf{4}) - E(KynA; DCBS) - E(\mathbf{4}; DCBS)$$
(S2)

where DCBS stands for Dimer-Centered Basis Set to make sure that the interaction energy is free from the effects of the basis set superposition error (BSSE).<sup>8</sup> E<sub>int</sub> obtained from B3LYP-D3 total energies amounts to -35.8 kcal/mol. This result is not inconsistent with the perturbation theory estimate given that SAPT was based on the Hartree-Fock description of the monomer wavefunctions. We can also offer a gas-phase estimate of the free energy as  $\Delta G^{\circ}(298^{\circ}K) = -22.1$  kcal/mol out of which, 13.5 kcal/mol of BSSE should be subtracted.

#### 

Gas-	-phase Recept	cor 4		
С	0.00000000	0.00	000000	0.00000000
С	-0.23789400	2.52	278700	-1.23116600
С	0.70789800	0.28	3533300	-1.18132200
С	-0.80912400	1.00	887200	0.53198400
С	-0.91907700	2.29	026200	-0.02365300
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Н	-1.36560100	0.79	9494700	1.44465400
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Н	-1.37015600	4.34	214000	0.29619200
С	-3.19598700	3.30	989700	-0.00079000
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С	-3.48135000	4.01	688200	-1.18264000
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С	-5.48628200	2.39	085300	-0.02359700
С	-4.70227400	3.82	2747600	-1.84293300
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Н	0.25038600	-4.92006300	-2.67945000
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н	0 13234600	-2 43570100	2 51474500
ц	-5 74634700	-3 32976100	2 51385800
и П	-0 76014500	3 11101600	2.51370000
и Ц	-6 63007400	2 5/007500	2.51/0600
л П	-5 50/56/00	6 22807300	-2 83830000
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U U	-5 09460500	-6 52465700	-3 96691000
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С	-0.07331600	-4.04652100	-6.63987900
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Cl	-3.97853100	7.34639500	-2.42347200
Cl	4.03854600	0.78131300	-2.42108200
176			
Wate	er solvent or	stimized Rece	entor 4
C			
C	2 52747000	0 24353000	-1 22276600
C	0 29925000	-0 72094800	-1 17005600
č	0.20025000	0.72094000	1.17000000

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С	0.99975200	0.82283800	0.53003700
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C	-3.76861300	1.76020500	-1.82164200
H	-1.83646400	2.50854700	1.43648/00
С	-3.26/52200	4.69205800	2.1361/300
Η	-2.33653900	4.27739400	2.54907200
Η	-4.09936800	4.09086400	2.53262700
С	1.46912900	6.55453800	2.13412900
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Η	2.39981700	2.23221200	2.54815100
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О Ц	-0 21965500	-2 46405600	_1 90990300
п	-0.21905500	-2.40403000	1 76049500
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н	4.24019900	1.08077500	-1.75989600
H	5.75250100	3.0052/600	-1.90691500
H	2.204/9200	7.46265000	-1.76101400
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Н	-4.1/858800	5.42924400	-1./5/99800
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Η	-0.82267000	8.03934600	-3.64074800
Η			
С	-2.04821600	6.75385700	-3.85268300
	-2.04821600 4.35095300	6.75385700 4.99692500	-3.85268300 -3.20721600
Η	-2.04821600 4.35095300 3.53024500	6.75385700 4.99692500 5.33050300	-3.85268300 -3.20721600 -3.85418900
H H	-2.04821600 4.35095300 3.53024500 4.81641500	6.75385700 4.99692500 5.33050300 4.10590100	-3.85268300 -3.20721600 -3.85418900 -3.64087100
H H C	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600	6.75385700 4.99692500 5.33050300 4.10590100 -1.06230100	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700
H H C H	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200	$\begin{array}{c} 6.75385700\\ 4.99692500\\ 5.33050300\\ 4.10590100\\ -1.06230100\\ -0.24120000 \end{array}$	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800
Н Н С Н Н	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200 0.88230100	$\begin{array}{c} 6.75385700\\ 4.99692500\\ 5.33050300\\ 4.10590100\\ -1.06230100\\ -0.24120000\\ -1.52673000 \end{array}$	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800 -3.64269700
H H C H H C	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200 0.88230100 -4.29029000	$\begin{array}{r} 6.75385700\\ 4.99692500\\ 5.33050300\\ 4.10590100\\ -1.06230100\\ -0.24120000\\ -1.52673000\\ 1.51249600 \end{array}$	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800 -3.64269700 -3.20548000
H H C H H C H	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200 0.88230100 -4.29029000 -3.47000900	$\begin{array}{c} 6.75385700\\ 4.99692500\\ 5.33050300\\ 4.10590100\\ -1.06230100\\ -0.24120000\\ -1.52673000\\ 1.51249600\\ 1.17818700 \end{array}$	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800 -3.64269700 -3.20548000 -3.85260800
Н Н С Н С Н Н	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200 0.88230100 -4.29029000 -3.47000900 -4.75543300	$\begin{array}{c} 6.75385700\\ 4.99692500\\ 5.33050300\\ 4.10590100\\ -1.06230100\\ -0.24120000\\ -1.52673000\\ 1.51249600\\ 1.17818700\\ 2.40357700 \end{array}$	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800 -3.64269700 -3.20548000 -3.85260800 -3.63932100
H H C H C H H N	-2.04821600 4.35095300 3.53024500 4.81641500 1.77346600 2.10786200 0.88230100 -4.29029000 -3.47000900 -4.75543300 -2.79811900	6.75385700 4.99692500 5.33050300 4.10590100 -1.06230100 -0.24120000 -1.52673000 1.51249600 1.17818700 2.40357700 8.60644600	-3.85268300 -3.20721600 -3.85418900 -3.64087100 -3.20824700 -3.85429800 -3.64269700 -3.20548000 -3.85260800 -3.63932100 -3.21897600

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C	5 71330200	6 62060300	-4 57810600
ц	6 19729500	5 80896700	-5 13396200
и П	-2 36849800	9 46504000	-2 76989400
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H	-4.13446800	9.6/8/0500	-4.41902500
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C	-5.65452800	-0.11096600	-4.5/511200
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Н	4.70047300	5.72152100	-6.95101800
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KynA	A@4 from gas-	-phase optimi	ization
С	0.0000000	0.00000000	0.0000000
С	1.00094600	-2.36758700	-1.14665300
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C	-0 04473300	-1 66648000	-1 76989200
U U	1 5603/500	_0 262/7700	1 13262200
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C	6 78136800	0 45210100	1 01794300
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C	5 56654800	3 80682200	-1 44040400
C	7 17705100	2 09156000	0 00100500
C	5 27459400	2.00130900	-0.00100300
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C	6.68345500	3.02656700	-1./851/300
H	4.69040600	2.2183/000	1.41215300
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С	2.63553900	-2.64352700	2.23900300
Н	2.71014700	-1.63105900	2.66280300
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Н	4.32014700	0.85199200	-1.28722200

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