

**Supplementary Material**

**A novel imine linked covalent organic framework  
for rapid detection of methyl paraoxon**

Mengyao Li, Lulu Guo, Lili Chen<sup>\*1</sup>, Chunhua Lin and Li Wang

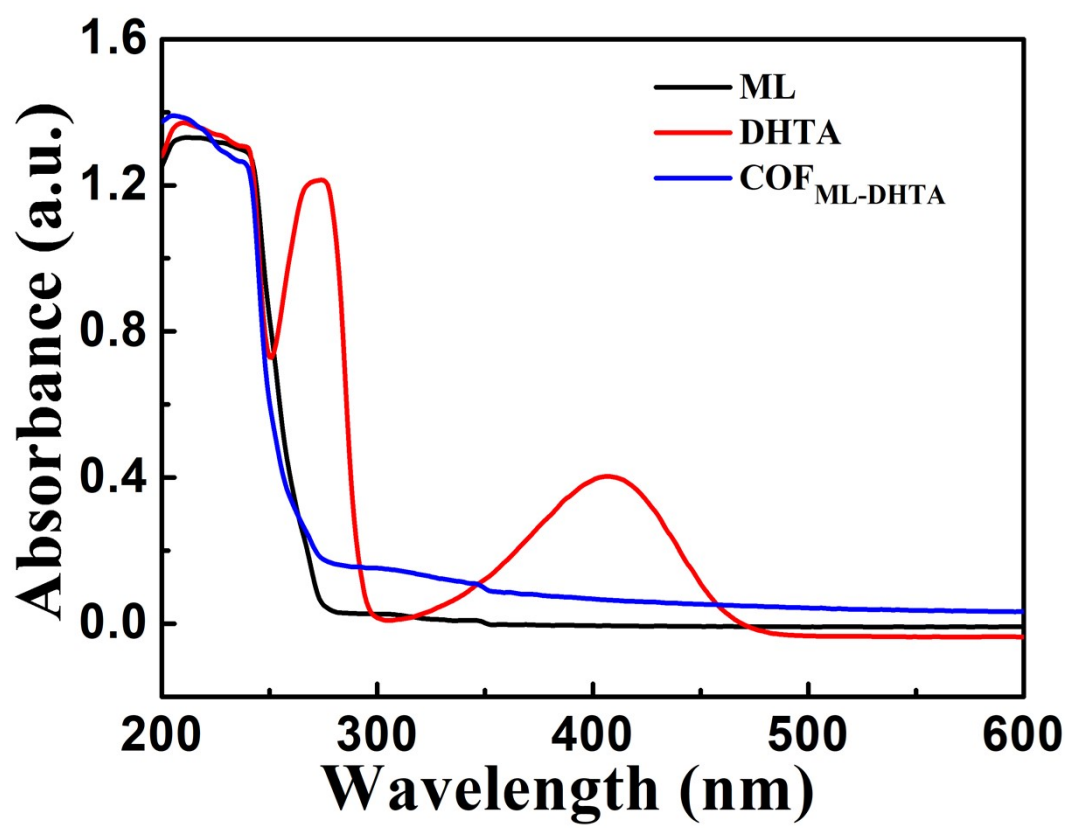
*Key Laboratory of Functional Small Organic Molecule, Ministry of Education, Key Laboratory of  
Chemical Biology, Jiangxi Province, College of Chemistry and Chemical Engineering, Jiangxi  
Normal University, 99 Ziyang Road, Nanchang 330022, China.*

---

<sup>1\*</sup> Corresponding author. E-mail: [chenlaura0797@163.com](mailto:chenlaura0797@163.com) (Lili Chen)



**Figure S1.** Water contact angle measurement of COF<sub>ML-DHTA</sub>



**Figure S2.** UV-vis absorption spectrum of ML, DHTA and COF<sub>ML-DHTA</sub>.

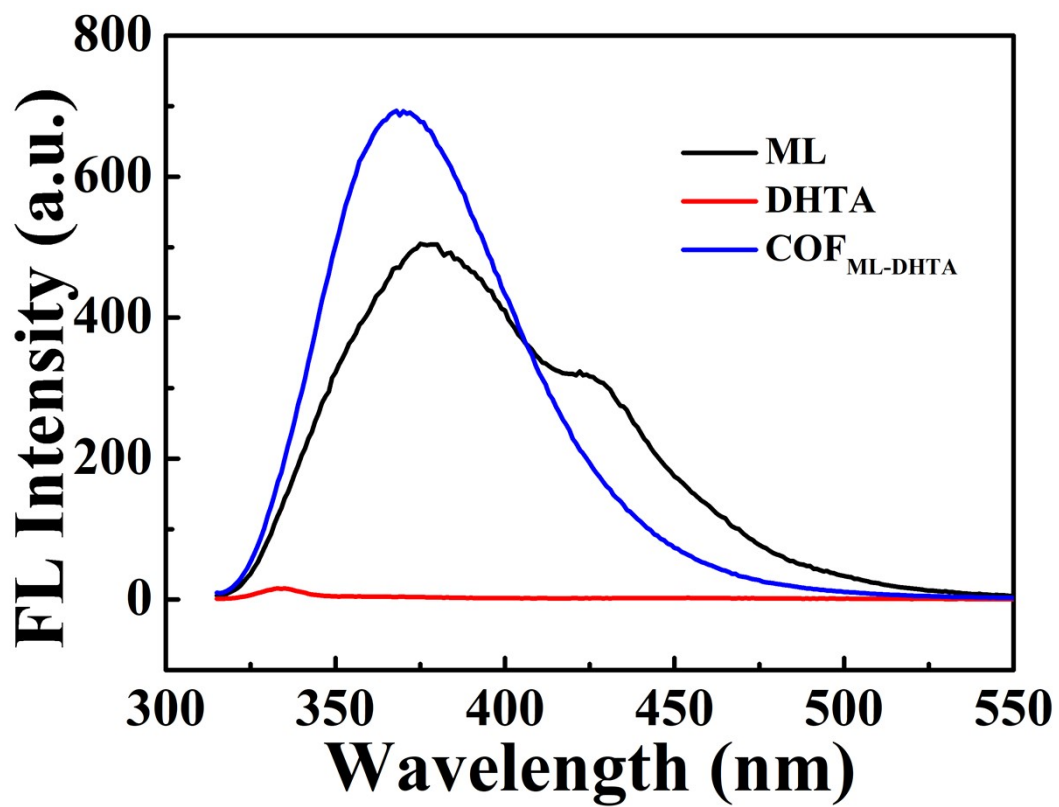
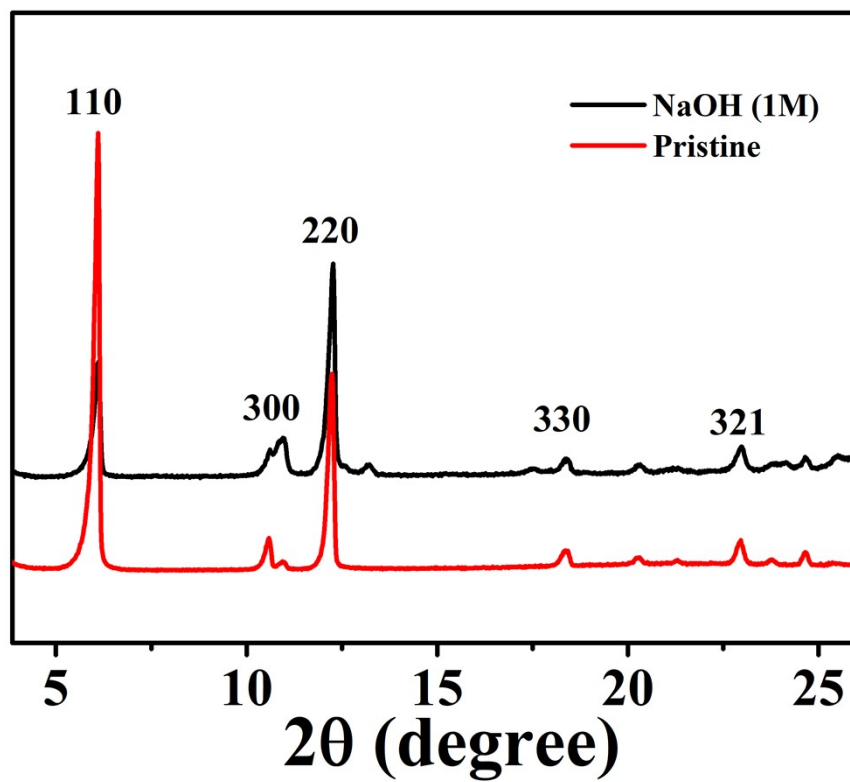
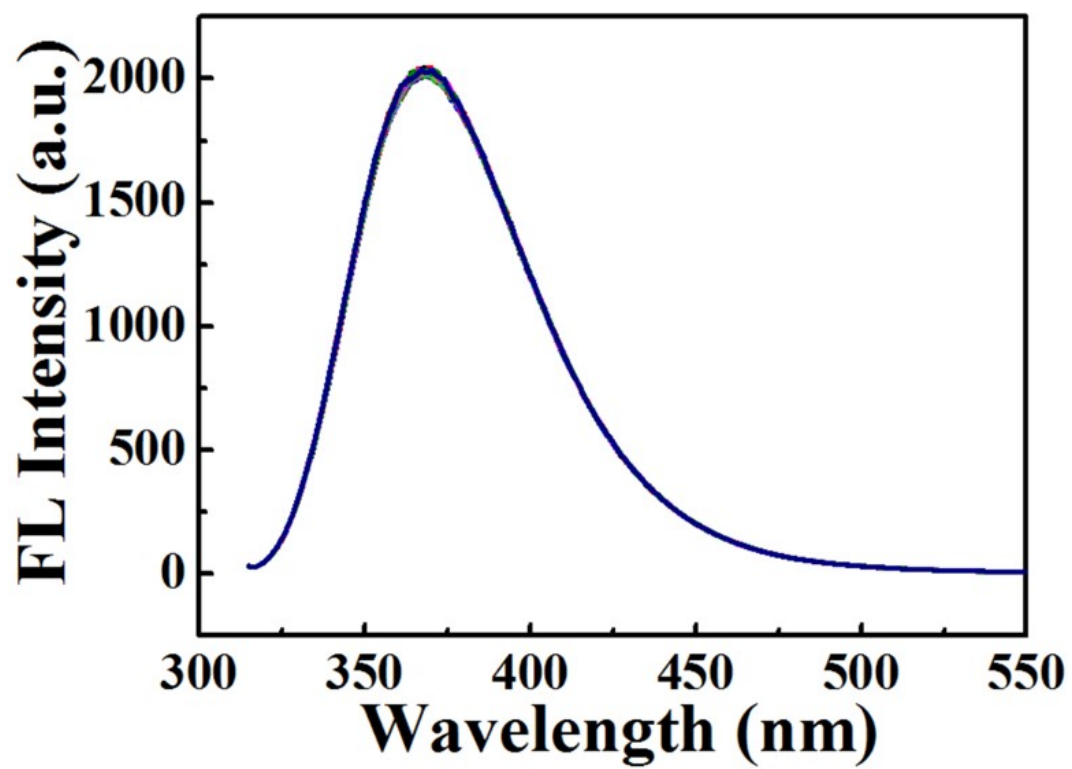


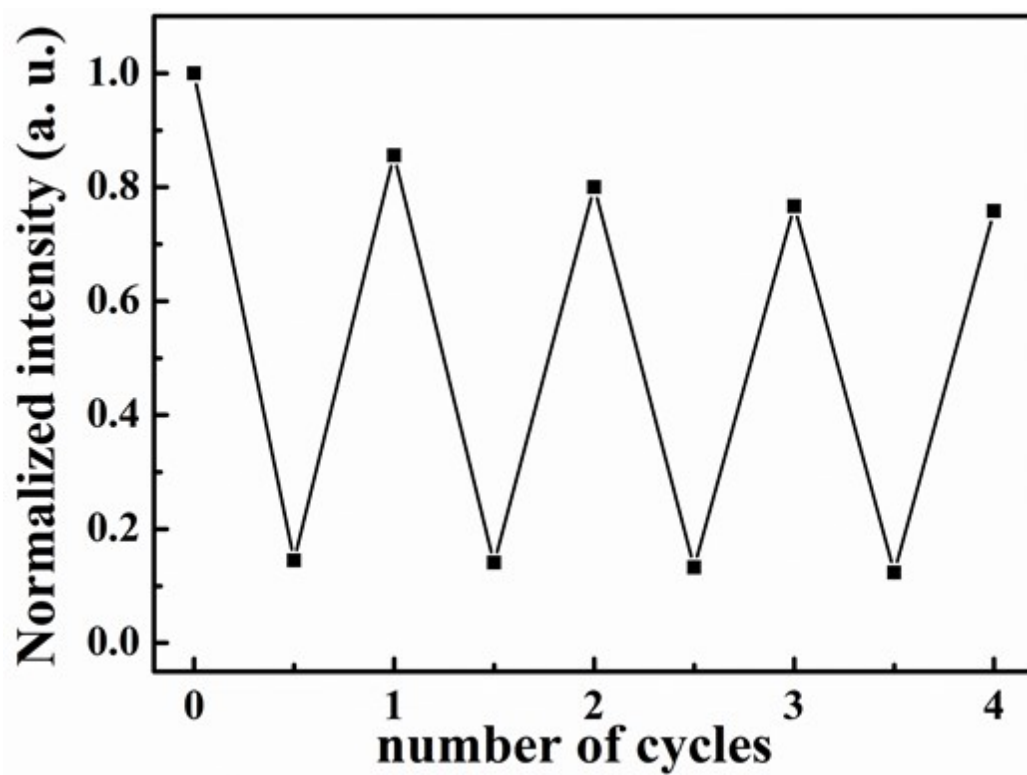
Figure S3. Fluorescence emission spectrum of ML, DHTA and COF<sub>ML-DHTA</sub>.



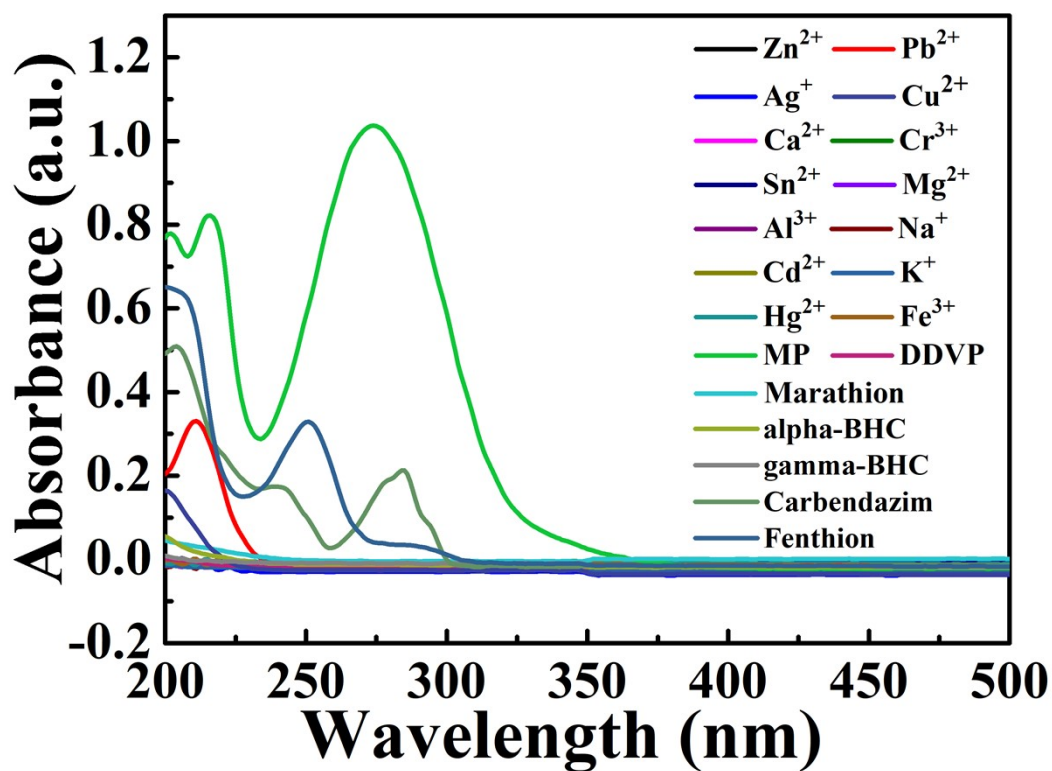
**Figure S4.** PXRD patterns of COF<sub>ML-DHTA</sub> and after treatment in 1M NaOH.



**Figure S5.** The fluorescence intensity of COF<sub>ML-DHTA</sub> material was measured continuously for 17min.



**Figure S6.** The reusability of COF<sub>ML-DHTA</sub> for four cycles



**Figure S7.** UV-vis spectra of Zn<sup>2+</sup>, Pb<sup>2+</sup>, Ag<sup>+</sup>, Cu<sup>2+</sup>, Ca<sup>2+</sup>, Cr<sup>3+</sup>, Sn<sup>2+</sup>, Mg<sup>2+</sup>, Al<sup>3+</sup>, Na<sup>+</sup>, K<sup>+</sup>, Cd<sup>2+</sup>, Hg<sup>2+</sup>, Fe<sup>3+</sup>, DDVP, malathion, fenthion, gamma-BHC, alpha-BHC and carbendazim.



**Table S1.** Crystallographic information of the modeled COF<sub>ML-DHTA</sub>

<b>Sample name</b>	<b>COF<sub>ML-DHTA</sub></b>
Stacking model	ABC
a [Å]	28.83444
b [Å]	28.83444
c [Å]	6.93499
$\alpha$ [deg]	90
$\beta$ [deg]	90
$\gamma$ [deg]	120
Interlayer distance (Å)	2.3 Å

**Table S2.** Atom sites for the modeled COF<sub>ML-DHTA</sub>

<b>Atom site label</b>	<b>Atom site fraction x</b>	<b>Atom site fraction y</b>	<b>Atom site fraction z</b>
N1	0.3028	0.6128	0.9306
N2	0.2494	0.5900	0.9361
C3	0.2240	0.6190	0.9394
N4	0.2567	0.6734	0.9339
N5	0.1640	0.5950	0.9533
C6	0.1228	0.5453	0.9689
C7	0.0600	0.5240	0.9844
C8	0.0225	0.4687	1.0006
C9	0.0344	0.5542	0.9846
H10	0.2392	0.6966	0.9368
H11	0.1325	0.5145	0.9729
H12	0.0375	0.4419	1.0020
H13	0.0583	0.5963	0.9733
C14	0.3333	0.6667	0.9282

**Table S3.** Comparison of analytical parameters of various probe.

<b>Probe/Material</b>	<b>Linear range</b>	<b>Detection limit</b>	<b>Reference</b>
AChE-SF/Pt	0.06–50 nM	0.02 nM	1
4-NP	1.0–1000.0 ppb	1.77 ppb	2
Su-TPE/PrS	0–16 nM (0–384 ng mL <sup>-1</sup> )	0.22 nM (5.28 ng mL <sup>-1</sup> )	3
AChE/CoPc/PVA-AWP/SPE	2.0–4000 nM	2.6nM	4
Fe <sub>3</sub> O <sub>4</sub> MNPs	-	10 nM	5
This work	0.57 ng mL <sup>-1</sup> to 30 µg mL <sup>-1</sup>	0.19 ng mL <sup>-1</sup>	

**Table S4.** Recovery tests of methyl paraoxon (MP) added in actual water samples by COF<sub>ML-DHTA</sub> material.

Samples	Spiked MP (ng mL <sup>-1</sup> )	Measured value (ng mL <sup>-1</sup> )	Recovery (%)
Lake water 1	0	Not detected	-----
Lake water 2	4	4.25±1.89	106.3
Lake water 3	8	8.11± 4.05	101.4
Lake water 4	12	12.23± 3.51	101.9