Supporting Information

Rational design of high performance nanotheranostics for NIR-II fluorescence/magnetic resonance imaging guided enhanced phototherapy

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1. Materials and apparatus

The Gd-DTPA-SA^[S1] and PEG-PHEMA-I ^[S2] were prepared according to previous protocols. Calcein AM/PI stain kit, NIH-3T3 normal cells and HeLa cells were purchased from Nanjing KeyGEN Biotech. Co. Ltd. Other Chemicals were purchased from Aladdin. All mice studies were approved by Animal Ethics Committee of Nanjing KeyGEN Biotech. Co. Ltd. and performed in Nanjing KeyGEN Biotech. Co. Ltd. The morphology and size of NPs were determined by a HT7700 transmission electron microscope (TEM) and a particle size analyzer (Brookhaven Instruments), respectively. Absorption and emission spectra were obtained using a UV3600 (Shimadzu) UV/vis/NIR spectrophotometer and а FLSP920 fluorescence spectrophotometer (Edinburgh), respectively. The MTT experiments were conducted using a PowerWave XS/XS2 microplate reader (BioTek).

2. Synthesis of DPPB



Fig. S1. Synthetic route of DPPB.

Compound 1 (0.1 mmol, 121 mg) and compound 2 (0.1mmol, 15 mg), $Pd(PPh_3)_4$ (15 mg), and diisopropylamine (30 mL) were mixed in a reaction bottle under N₂ atmosphere. The mixture was vigorously stirred at 85 °C for 48 h. After removal of the solvent, the crude product was settled with ether to obtain DPPB as a black-green solid.



Fig. S2. ¹H NMR spectrum of DPPB.



Fig. S3. GPC of DPPB, the number-average of DPPB was 14000.

3. Molar absorption coefficient of DPPB-Gd-I NPs



Fig. S4. (a) Absorption curves of DPPB-Gd-I NPs aqueous solution at different concentrations. (b) Linear absorbance versus concentration obtained from (a).

4. Photodynamic measurement of DPPB-Gd-I NPs



Fig. S5. Absorption spectra of the mixed aqueous solution of DPPB-Gd-I NPs and DPBF under 660 nm laser irradiation with different power densities (a) 0.25, (b) 0.50, (c) 0.75 W/cm². (d) The decline of normalized absorption of DPBF resulted from 660 nm light induced DPPB-Gd-I NPs.

5. Infrared thermal images of DPPB-Gd-I NPs



Fig. S6. (a) Infrared thermal images of DPPB-Gd-I NPs at various concentrations upon 660 nm laser illumination.

6. Photothermal measurement of DPPB-Gd NPs



Fig. S7. (a) Heating curves of DPPB-Gd NPs at various concentrations upon 660 nm laser illumination. (b) Heating curves of DPPB-Gd NPs upon 660 nm laser illumination with various powers.

7. Cellular uptake of NPs



Fig. S8. The cellular uptake of NPs toward HeLa cells evaluated by (a) confocal laser scanning microscopy and (b) flow cytometry.

Due to the fluorescence emission wavelength of polymer was too long, which was not suitable for confocal laser scanning microscopy. Fluorescein isothiocyanate (FITC) was used as a model dye and encapsulated by Gd-DTPA-SA and PEG-PHEMA-I to form water-soluble FITC NPs. HeLa cells after incubation with FITC NPs for 12 h showed intense green fluorescence assigned to nanoparticles, indicating the effective cellular internalization of the NPs, which was consistent with the result of flow cytometry.



8. Penetration depth measurement of DPPB-Gd-I NPs

Fig. S9. Penetration depth measurement of DPPB-Gd-I NPs in a simulated deeptissue setting (chicken-breast tissues).

9. Ex vivo NIR-II fluorescence imaging of tumor and major organs



Fig. S10. (a) Ex vivo NIR-II image of tumors and major organs. (b) Semi-quantitative biodistribution analysis based on fluorescence intensity of tumors and major organs.

10.H&E stained images of tumor



Fig. S11. H&E stained images of tumor. Scale bars: 80 µm.

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