Supplementary materials

Ultrahigh SERS Activity of TiO₂@Ag Nanostructure Leveraged for

Accurately Detecting CTCs in Peripheral Blood

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Fig. S1 X-ray diffraction (XRD) spectra of TiO₂, TiO₂@Ag, and Ag.



Fig. S2 Element distributions mapping images of Ti, O, and Ag in $TiO_2@Ag$ nanostructures.



Fig. S3 Energy-dispersive spectroscopy (EDS) characterization of $TiO_2@Ag$.



Fig. S4 XPS spectra of survey spectrum (a), Ag 3d (b), Ti 2p (c), and O 1s (d) of TiO_2 and $TiO_2@Ag$ samples, respectively.



Fig. S5 UV-vis spectroscopy of TiO₂@Ag, and TiO₂.



Fig. S6 Raman spectra of R6G molecule (5×10^{-5} M) adsorbed on TiO₂@Ag NPs and pure SERS signal of R6G molecules (5×10^{-2} M).

Enhancement factor (EF) calculation of TiO₂@Ag NPs was based on equation (1):

$$EF = (I_{SERS} / N_{ads}) / (I_{bulk} / N_{bulk}) (1)$$

Where N_{ads} and N_{bulk} imply the number of R6G molecules adsorbed on the TiO₂@Ag NPs substrate and R6G molecules in normal Raman substrate, respectively. I_{SERS} and I_{bulk} are the vibration peak (1356 cm⁻¹) intensity of R6G molecules on TiO₂@Ag NPs substrate, and normal Raman spectrum of R6G molecules, respectively.

During the SERS experiment, 100 μ L of ethanol aqueous R6G solution (5 × 10⁻² M) was dried onto 0.4 cm × 0.4 cm⁻² silicon wafer. N _{Raman} was lied on equation (2).

$$N_{bulk} = 100 \ \mu L \times 5 \times 10^{-2} \ mol/L \times 8 \times 6.02 \times 10^{23} \ mol^{-1} \times 1.3 \ \mu m^2 \ / \ 0.16 \ cm^{-2} \ (2)$$

where *d* is the diameter of the light spot estimated as $d = 1.22 \lambda$ /NA, λ is incident wavelength 532 nm, the numerical aperture of the objective lens N_A=0.5, the laser spot is ~ 1.3 µm². On the basis of equation (2), N_{bulk} was estimated as ~1.95× 10¹².

In addition, N_{ads} is synergistically determined by laser spot illuminating the TiO₂@Ag-R6G substrate, and the density of R6G molecules, N_{ads} is concluded as:

$$N_{ads} = \sigma \times 1.3 \ \mu m^2 \times 6.02 \times 10^{23} \ mol^{-1}(3)$$

Where σ is the density of R6G molecule absorbed onto TiO₂@Ag substrate, which is approximated to ~0.5 nM cm⁻²(3). N_{ads} is concluded as ~ 3.91 × 10⁶. I_{SERS} is the Raman peak intensity at 1356 cm⁻¹ of R6G molecules on TiO₂@Ag NPs, while I_{bulk} represents the Raman peak intensity at 1356 cm⁻¹ of pure R6G molecule (fig S6). I_{SERS} = ~ 135065 and I_{bulk} = ~ 882. Substituting these values into equation (1), EF of TiO₂@Ag NPs was concluded to be ~ 7.61× 10⁷.



Fig. S7 SERS signal of R6G (5×10^{-7} M) molecule adsorbed on TiO₂@Ag nanostructure, and Ag NPs.



Fig. S8 FI-TR spectra of FA, rBSA, $TiO_2@Ag-rBSA$, and $TiO_2@Ag-rBSA-FA$.



Fig. S9 SERS spectra of R6G (5×10^{-3} M) molecule collected from six TiO₂@Ag-rBSA-FA SERS bioprobes; Excitation wavelength: 532 nm; Laser power: 0.1 mW; Lens: $50 \times$ objective; Accumulations: 10 s.



Fig. S10 Cell viability of A549 and MCF-7 cells co-incubated with $TiO_2@Ag-R6G-$ rBSA-FA bioprobe for 24 h with different $TiO_2@Ag$ concentrations.



Fig. S11 FCS analysis of MCF-7 and A549 cells incubated with 20 μL FA-PEG-FITC (red) and PBS (blue).



Fig. S12 HeLa cell count of bright field under laser scanning confocal microscope. Scale bar: 25 μ m.



Fig. S13 Detection sensitivity of the $TiO_2@Ag-R6G-rBSA-FA$ SERS bioprobe for different numbers of HeLa cells in the rabbit blood.



Fig. S14 TiO₂@Ag-R6G-rBSA-FA SERS bioprobe was utilized for CTCs detection in peripheral blood of four liver cancer patients.