Electronic Supplementary Information (ESI)

Injectable hydrogel platform with biodegradable molybdenum polyoxometalate and R848 for combinational photothermalimmunotherapy of cancer

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## 1. Calculation of photothermal conversion efficiency

The photothermal conversion efficiency of three samples was be calculated by using the following formula according to the methods of these articles<sup>1, 2</sup>.

$$\eta = \frac{hS(T_{max} - T_{max, water})}{I(1 - 10^{-A_{808}})}$$
$$hS = \sum mC_p / \tau_s$$
$$\tau_s = (-t) / ln\theta$$
$$\theta = (T_{amb} - T) / (T_{amb} - T_{max})$$
$$h \text{ is the heat transfer coefficient;}$$

S is the surface area of the photothermal test vessel;

 $\tau_s$  is the time constant of samples;

m is the mass of samples (1.0 g);

 $C_p$  is the specific heat capacity of water ( $C_p = 4.2 \text{ J} \cdot \text{mol}^{-1}$ );

 $A_{808}$  is the absorbance at 808 nm of different samples.

According to the first heating-cooling process in the photothermal cycle,  $\tau_s$  is obtained by the linear relationship between the cooling period and natural logarithm of driving force temperature ( $T_{max, water} = 22.8$  °C).

For POM@GG hydrogel,  $T_{amb}$  is the temperature of the surroundings (21.0 °C),  $T_{max}$  is the equilibrium temperature of POM@GG hydrogel (42.6 °C), and the value of  $\tau_s$  is 281.69 (Figure S5), hS is calculated from the above equation (hS = 0.01491). The deviation of  $T_{max}$  and  $T_{max, water}$  is 21.6 °C. I is 0.45 W where the area of light spot was 1.5 cm<sup>2</sup>, and A is the absorbance of POM@GG hydrogel at 808 nm ( $A_{808} = 1.41$ ). The photothermal conversion efficiency  $\eta$  is obtained from the first formula ( $\eta = 63.1\%$ ).

## 2. Supplementary Figures



**Figure S1.** Powder X-ray diffraction (PXRD) analysis of ox-POM using Cu K $\alpha$  in the range of 10–60° with a scanning rate of 10°/min.



Figure S2. Color change of POM before and after reduction.



**Figure S3.** Size distribution of POM clusters as a function of time in the reduction process measured by the dynamic light scattering (DLS).



**Figure S4.** (a) Temperature rises of 300  $\mu$ g/mL POM solution under different GSH concentrations and (b) the temperature rises in different concentration POM solutions.



**Figure S5.** The image of POM@GG hydrogel before and after the NIR laser irradiation (808 nm, 0.3 W/cm<sup>2</sup>, 10 min).



**Figure S6.** Calculation of photothermal conversion efficiency using the first photothermal cycle of POM@GG.



Figure S7. The color change of POM (300  $\mu$ g/mL, 1 mL) after being dispersed in the H<sub>2</sub>O<sub>2</sub> solution (10 mM, 1 mL).



**Figure S8.** (a) The residual tumor weights after the treatment and (b) body weights of mice during the treatment. G1: untreated; G2: R848@GG; G3: POM@GG; G4: POM@GG with NIR; G5: R848/POM@GG with NIR.



**Figure S9.** H&E staining images of major organs (heart, liver, spleen, lung, and kidney) in different groups after the treatment.

## References

- 1. C. Sun, L. Wen, J. Zeng, Y. Wang, Q. Sun, L. Deng, C. Zhao and Z. Li, *Biomaterials*, 2016, **91**, 81-89.
- W. Ren, Y. Yan, L. Zeng, Z. Shi, A. Gong, P. Schaaf, D. Wang, J. Zhao, B. Zou, H. Yu, G. Chen, E. M. B. Brown and A. Wu, *Adv. Healthcare Mater.*, 2015, 4, 1526-1536.