Supporting information

for

Excellent Thermoelectric Performance Achieved in Bi₂Te₃/Bi₂S₃@Bi Nanocomposites

Yi-Xin Zhang^{a§}, Yu-Ke Zhu^{a§}, Dong-Sheng Song^b, Jing Feng^a, and Zhen-Hua Ge^{a*}

 Faculty of Materials Science and Engineering, Kunming University of Science and Technology, Kunming 650093, China.

2.Ernst Ruska-Centre for Microscopy and Spectroscopy with Electrons and Peter Grünberg Institute,

Forschungszentrum Jülich, 52425 Jülich, Germany.

Experimental Details

Chemicals used in this work were purchased from Alfa Aesar without any further purification. The nano powders were synthesized by a two-step hydrothermal method as described in our previous work [1]. The Bi₂S₃ nanowire is used as a template. At first step, as template materials, [001] oriented single crystal Bi₂S₃ nanowires with diameters of 50 to 200 nm and lengths of 500 nm to 10 μ m. The preparation of Bi₂S₃@Bi core-shell nanowires was based on a topical experiment. Initially, 1 g NaOH was added into 20 ml of deionized (DI) water, followed by stirring for 10 min. Then, 10 ml hydrazine was added to the solution, which was again stirred for 20 min. Finally, 0.5 mmol Bi₂S₃ nanowire powder was added to the solution, and subjected to a further 30 min of stirring. These stirring steps throughout the process ensured thorough mixing of the solution. The resulting solution was then transferred into a Teflon-lined stainless-steel autoclave (100 ml capacity) together with an additional 30 ml of DI water. The autoclave was heated up to 180°C for 1 h to obtain Bi₂S₃@Bi core-shell nanowires. For the preparation of Bi₂Te₃ nanowires the process is the same with the exception that Te

powder was added together with the resulting solution in the Teflon-lined stainless steel autoclave. The sealed autoclave was heated to 180°C for 1 h. The final black solid product was filtered and then washed with DI water and ethanol three times prior to drying under vacuum at room temperature.

The powders were checked by X-ray diffraction (XRD; Bruker, Germany). Transmission electron microscopy (TEM) observations were performed by using a FEI Tecnai F20 microscope operated at 200 kV. The as-prepared Bi₂Te₃ powders and Bi₂S₃@Bi nanowires with a weight ratio of 4:1 were mixed and densified by spark plasma sintering at 673 K for 5 min under the pressure of 50 MPa using an SPS system (Sumitomo SPS1050, Japan). The phase structure of the sintered samples was analyzed by XRD. The morphologies of fractography were observed by field emission scanning electron microscopy (FESEM; Zeiss Merlin, Germany). The TE properties were evaluated along the sample section perpendicular to the pressing direction of SPS. The Seebeck coefficient and electrical resistivity were measured from 323 to 523 K in a helium atmosphere by using the Seebeck coefficient/electric resistance measuring system (ZEM-3, Ulvac-Riko, Japan). The thermal conductivity κ was calculated via the relation $\kappa = DC_p d$. Among that the thermal diffusivity coefficient (D) of the specimens was captured through laser flash method (NETZSCH, LFA457, Germany), the mass density (d) was obtained via the Archimedes method, the C_p value was obtained by the Kopp's law. The Hall coefficients (R_H) of the samples were measured at the range of 323-523 K using a physical properties measurement system (PPMS-9T, Quantum Design Inc., USA).



Fig. S1 XRD patterns of Bi₂S₃@Bi powders, Bi₂Te₃ powders, and composite bulk.



Fig. S2 (a) The FESEM image of Bi_2Te_3 bulk and (c) $Bi_2S_3@Bi$ nanowires in the bulk sample and the EDS line scanning results along with the yellow line to the (b) Bi_2Te_3 bulk and (d) $Bi_2S_3@Bi$ nanowires in the bulk.



Fig. S3 TEM images of the Bi₂S₃@Bi nanowires, which reveals the size uniformity of the Bi₂S₃@Bi nanowires and the thickness of Bi shell.



Fig. S4 Lattice thermal conductivity and carrier thermal conductivity of the bulk composite

References

[1] Z.H. Ge, G.S. Nolas, Cryst. Growth Des. 14 (2014) 533-536.