# **Supporting Information**

### Solvent-assisted Crystallization of Two-dimensional

## **Ruddlesden-Popper Perovskite**

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#### 1. Materials

15  $\Omega \cdot \Box^{-1}$  ITO was purchased from Yingkou OPV Tech New Energy Co. PEDOT:PSS (Al 4083), BAI (99.5%), MAI (99.5%) and PC<sub>61</sub>BM (99.1%) were purchased from Xi'an Polymer Light Technology Corp. PbI<sub>2</sub> (99.9%) was purchased from Advanced Election Technology CO,. Ltd. BCP (95%) were purchased from Aladdin. PbCl<sub>2</sub> (99.999%), dimethyl sulfoxide (DMSO, hybridoma, 99.7%), N,Ndimethylformamide (DMF, for HPLC, 99.9%), Chlorobenzene (for HPLC, 99.9%) and isopropanol (99.5%) were purchased from Sigma-Aldrich.

#### 2. Device Fabrication

PEDOT:PSS was spin-coated on clean and UV-ozone-treated ITO at 5000 rpm for 30 s, and then annealed at 130°C for 20 min. Under N<sub>2</sub> atmosphere, the composition of BAI:MAI:PbCl<sub>2</sub>:PbI<sub>2</sub> with a stoichiometric ratio of 2:3:0.08:3.92 was dissolved in DMF to prepare a perovskite precursor solution with a concentration of 0.8 M.

Control method to deposit perovskite film: the 2D perovskite films were prepared by a one-step method with a spin speed of 5000 rpm for 30 s. Solventassisted method to deposit perovskite film: drop 160  $\mu$ L DMSO on the ITO with PEDOT:PSS, spin-coated at a low speed of 1000 rpm for 6 s, and dynamic spincoated 60  $\mu$ L the perovskite precursor solution at the 5th second, and then spin at a high speed of 5000 rpm for 10 s, and then anneal at 100°C for 15 min. PCBM (20 mg mL<sup>-1</sup> in CB) and BCP (0.5 mg mL<sup>-1</sup> in IPA) were spin-coated at 2000 rpm and 5000 rpm for 30 s, and then annealed at 75°C for 10 min and 5 min, respectively. Finally, 70 nm silver was deposited as a metal electrode in a vacuum environment.

#### 3. Characterization

Keithley 2460 (Keithley, America) was used to measure the J-V characteristics

of cells at AM 1.5G. The Incident photon-to-electron conversion efficiency (IPCE) is measured by using a computer-controlled xenon lamp combined with a monochromator (PEC-S20, Peccell). Photoluminescence (PL) spectra is measured by C5410 (Hamamatsu) at an excitation wavelength of 495 nm. The scanning electron microscopy (SEM) images were measured on a high-resolution field emission Nano SEM 450 (FEI, USA). The crystal structure and light absorption of 2D perovskite were measured using X-ray diffraction (XRD) (XRD-7000S Shimadzu, Japan) and UV-vis spectral (Lambda950), respectively. EIS/TPC/TPV/IMPS/IMVS are all measured using electrochemical workstation (Zennium Zahner, Germany).

### **Figures**



**Fig. S1**. Schematic diagram of devices with the structure of Glass/ITO/PEDOT:PSS/perovskite/PC<sub>61</sub>BM/BCP/Ag.



Fig. S3. UV-vis absorption characteristics of perovskite films prepared by different methods.



Fig. S4. Optical images for perovskite crystalline growth with different time on 100°C hot plate,
(a) control; (b) DMSO-assisted. UV-vis absorption spectra of perovskite films with time variation,
(c) control, (d) DMSO-assisted.



**Fig. S5**. Optical images for perovskite crystalline growth with different time on 100°C hot plate, (a) DMSO-assisted; (b) control-8:2 (DMF:DMSO); (c) control-7:3 (DMF:DMSO); (d) control-5:5 (DMF:DMSO).



Fig. S6. J-V curves of devices with different preparation methods, corresponding to Figure S3.

films prepared by different processes.					
		V <sub>oc</sub> [V]	J <sub>sc</sub> [mA/cm <sup>2</sup> ]	FF	PCE [%]
DMSO-assisted	best	1.05	17.17	0.69	12.42
	average	1.05±0.02	15.82±1.31	0.68±0.04	11.31±0.67
control	best	1.04	11.81	0.68	8.37
	average	1.04±0.03	10.18±1.66	0.61±0.06	6.40±1.07

 Table S1. Photovoltaic parameters of solar cells based on (BA)<sub>2</sub>(MA)<sub>3</sub>Pb<sub>4</sub>(I<sub>0.98</sub>Cl<sub>0.02</sub>)<sub>13</sub> perovskite

There are 30 devices taken into account to calculate the average parameters of devices.

SI 1. The density of defect states is calculated by the dark J-V curve of the device with the

structure of ITO/SnO<sub>2</sub>/2D PVK/PC<sub>61</sub>BM/BCP/ Ag. The trap-state density was determined from

$$N_{\rm t} = \frac{2\epsilon_0\epsilon_{\rm r}V_{\rm TFL}}{qL^2}$$

the following equation:

where  $N_{t}$  is the trap density,  $V_{TFL}$  is the trap-filling limit voltage, and  $\boldsymbol{q}$  is the elemental charge.