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Supplementary Information for

A new strategy for hyperconjugative antiaromatic compounds utilizing negative charges: a dibenzo[*b*,*f*]silepinyl dianion

Shotaro Ito,^a Youichi Ishii,^a Kazuya Ishimura,^b and Takuya Kuwabara^{*, c}

^aDepartment of Applied Chemistry, Faculty of Science and Engineering, Chuo University, 1-13-27, Kasuga, Bunkyo-ku, Tokyo 112-8551, Japan

^bX-Ability, Co., Ltd., Ishiwata Building 3rd Floor, 4-1-5 Hongo, Bunkyo-ku, Tokyo 113-0033, Japan ^cDepartment of Chemistry and Biochemistry, Graduate school of Humanities and Sciences, Ochanomizu University, 2-1-1, Otsuka, Bunkyo-ku, Tokyo 112-8610, Japan

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1. General considerations

All manipulations were carried out under an argon atmosphere by using standard Schlenk techniques or a glovebox unless otherwise stated. Et₂O, THF, DME, C₆D₆ and THF-*d*₈ were distilled over potassium mirror. Ph₂SiCl₂ was purchased from Tokyo Chemical Industry (TCI). 2,2'-Dibromostilbene was synthesized according to the literature.¹ ¹H (500 MHz), ¹³C{¹H}(126 MHz), ⁷Li{¹H} (194 MHz), and ²⁹Si{¹H} (99 MHz) NMR spectra were recorded on a JEOL ECZ-500 spectrometer at 20°C unless otherwise stated. Chemical shifts are reported in δ and referenced to residual ¹H and ¹³C signals of deuterated solvents as internal standards or to the ⁷Li{¹H} and the ²⁹Si NMR signal of LiCl in D₂O (δ = 0.00) and SiMe₄ in CDCl₃ (δ = 0.00), respectively, as external standards. Elemental analyses were performed on a Perkin Elmer 2400 series II CHN analyzer. UV-vis absorption spectra were were recorded on a SHIMADZU UV-1800 spectrophotometer.

X-ray diffraction analysis

Diffraction data for **1**, **2** and **3** were collected on a VariMax Saturn CCD diffractometer with graphitemonochromated Mo K α radiation ($\lambda = 0.71075$ Å) at –180 °C. Intensity data were corrected for Lorenzpolarization effects and for empirical absorption (REQAB).² All calculations were performed using the CrystalStructure³ crystallographic software package except for refinements, which were performed using SHELXL-2018/3.⁴ The positions of the non-hydrogen atoms were determined by SHELXT.⁵ All non-hydrogen atoms were refined on F_0^2 anisotropically by full-matrix least-square techniques. All hydrogen atoms were placed at the calculated positions with fixed isotropic parameters.

Theoretical studies

Theoretical calculations were performed by using the *Gaussian* 16 Rev.C⁶ for geometries and NICS values, and the Gaussian 09 Rev. E.01⁷ for the anisotropy of the current (induced) density (ACID) analysis. The NICS and ACID data as well as molecular orbitals for **1**, **2**, **2'**, and **4** were calculated using optimized structures. The optimized structures of **1** and **2** are in good agreements with their X-ray structures. All local minima were confirmed by the vibrational frequency calculations with zero imaginary frequency, which was performed at the same levels for the optimization.

NICS and ACID calculations for **1**, and optimization for **2** were performed at the B3LYP⁸ level of theory using 6-31G(d) basis sets.⁸ The NICS values, NBO calculations, and ACID data for **2'** and **4** were calculated at the B3LYP level of theory using 6-31+G(d) basis sets. The external magnetic field of ACID calculations was applied in the direction almost orthogonal to three-ring planes, and current density vectors were plotted onto the isosurface of 0.03.

2. Synthesis and analytical data for the new compounds

Synthesis of Diphenyldibenzosilepin 1.



^{*n*}BuLi (1.59 M in hexanes, 3.70 mL, 5.88 mmol) was added to a THF solution (100 mL) of dibromostilbene (0.998 g, 2.95 mmol) in THF at -80 °C. After 1 hour, Ph₂SiCl₂ (0.60 mL, 0.722 g, 2.85 mmol) was added dropwise, then the mixture was allowed to gradually warm to room temperature and stirred for 20 hours. The solvent was removed in vacuo, and materials insoluble in CH₂Cl₂ were filtered off. The filtrate was concentrated and purified by recrystallization from Et₂O to yield **1** as colorless crystals (0.890 g, 2.47 mmol, 84% yield). m.p. 203 °C; ¹H NMR (500 MHz, CDCl₃): δ = 7.45–7.40 (m, 8H, Ar), 7.38–7.33 (m, 8H, Ar), 7.30–7.27 (m, 2H, Ar), 6.79 (s, 2H, vinyl); ¹³C{¹H} NMR (126 MHz, CDCl₃): δ = 142.5 (4°, *C*_{α} or *C*_{θ}), 136.5 (3°), 136.0 (3°), 134.1 (4°, *C*_{α} or *C*_{θ}), 133.1 (3°, vinyl), 132.8 (4°, *C*_{*ipso} of Ph*), 130.2 (3°), 129.8 (3°), 129.7 (3°), 127.9 (3°), 127.2 (3°); ²⁹Si{¹H} NMR (99 MHz, CDCl₃): δ = -13.9; Elemental analysis calcd (%) for C₂₆H₂₀Si (**1**): C 86.62, H 5.59; found: C 86.24, H 5.55.</sub>

Synthesis of lithium salt of 5,5-diphenyldibenzo[*b*,*f*]silepinyl dianion 2.



Lithium (16.6 mg, 2.39 mmol) was added to a solution of **1** (71.2 mg, 0.197 mmol) in THF (2 mL) at room temperature. After 1 hour, lithium was removed and the solvent was removed in vacuo to yield analytically pure **2** as black-purple powder (145.1 mg, 0.197 mmol, quant.) without any purifications. m.p. > 108 °C (decomp.); ¹H NMR (500 MHz, THF-*d*₈): δ = 7.63–7.60 (m, 4H, *Ph*), 7.14–7.10 (m, 6H, *Ph*), 5.63 (dd, ³*J*_{H–H} = 7 Hz, ⁴*J*_{H–H} = 2 Hz, 2H, *fused benzene-d*), 5.34 (td, ³*J*_{H–H} = 7 Hz, ⁴*J*_{H–H} = 2 Hz, 2H, *fused benzene-b*), 4.99 (d, ³*J*_{H–H} = 8 Hz, 2H, *fused benzene-a*), 4.22 (t, ³*J*_{H–H} = 7 Hz, 2H, *fused benzene-c*), 2.88 (s, 2H, *vinyl*); ¹³C{¹H} NMR (126 MHz, THF-*d*₈): δ = 147.9 (*C*₆), 143.4 (*ipso-Ph*), 142.6 (*fused benzene-d*), 137.4 (3°, *Ph*), 127.9 (*fused benzene-b*), 126.5 (3°, *para-Ph*), 126.3 (3°, *Ph*), 120.4 (*fused benzene-a*), 103.8 (*C*_a), 97.4 (*fused benzene-c*), 78.4 (*vinyl*); ⁷Li{¹H} NMR (194 MHz, THF-*d*₈): δ = 0.21; ²⁹Si{¹H} NMR (99 MHz, THF-*d*₈): δ = -31.0. Elemental analysis calcd (%) for C₄₆H₆₀SiLi₂O₅ (**2**): C 75.18, H 8.23; found: C 74.91, H 8.32.

Reactivity of 2.



 O_2 quench: Compound 2 was generated by the above method from 1 (30.0 mg, 0.0830 mmol) in THF (2 mL). After removal of excess lithium, O_2 gas was introduced to the Schlenk flask. The solvent was removed in vacuo to obtain white solids. Materials insoluble in CH_2Cl_2 were filtered off and the solvent was removed in vacuo to yield 1 as white powder (30.4 mg, 0.0841 mmol, quant.).

Mel quench: Compound **2** was generated by the above method from **1** (30.0 mg, 0.0830 mmol) in THF (2 mL). After removal of excess lithium, CH₃I (0.1 mL, 228 mg, 1.61 mmol) was added dropwise. After 0.5 h, the solvent was removed in vacuo, and materials insoluble in CH_2CI_2 were filtered off. The solvent was removed in vacuo and crude product was purified by preparative TLC (Hexane:CH₂Cl₂ = 4:1) to yield **1** as a white powder (27.2 mg, 0.0753 mmol, 91% yield).

EtOH quench: Compound **2** was generated by the above method from **1** (30.4 mg, 0.0841 mmol). After removal of excess lithium, EtOH (0.1 mL, 78.9 mg, 1.71 mmol) was added dropwise. After 0.5 hour, the solvent was removed in vacuo, and materials insoluble in CH₂Cl₂ were filtered off. The solvent was removed in vacuo to yield **3** as white powder (29.5 mg, 0.0815 mmol, 97% yield). m.p. 159 °C; ¹H NMR (500 MHz, CDCl₃): δ = 7.43–7.40 (m, 6H, *Ar*), 7.36–7.32 (m, 8H, *Ar*), 7.21 (d, ³J_{H-H} = 7 Hz, 2H, *Ar*), 7.15 (t, ³J_{H-H} = 7 Hz, 2H, *Ar*), 3.20 (s, 4H, *methylene*); ¹³C{¹H} NMR (126 MHz, CDCl₃): δ = 150.5 (4°, *C*_α or *C*₆), 137.5 (3°), 136.2 (3°), 135.6 (4°), 133.1 (4°), 130.0 (3°), 129.5 (3°), 129.2 (3°), 127.9 (3°), 125.2 (3°), 36.8 (*methylene*); ²⁹Si{¹H} NMR (99 MHz, CDCl₃): δ = -12.7; Elemental analysis calcd (%) for C₂₆H₂₂Si (**3**): C 86.14, H 6.12; found: C 85.90, H 5.83.

 H_2O quench: Compound 2 was generated by the above method from 1 (30.1 mg, 0.0833 mmol). After removal of excess lithium, H_2O (0.1 mL, 100 mg, 5.55 mmol) was added dropwise. After 0.5 hour, MgSO₄ was added to the solution, then the solvent was removed in vacuo. MgSO₄ and materials insoluble in CH₂Cl₂ were filtered off. The solvent was removed in vacuo to yield mixture of 1 and 3 as white powder (29.9 mg, 92% NMR yield for 3). Compounds 1 and 3 could not be separated by column chromatography or recrystallization due to their similar chemical properties.

	1	2	3
Si-C1/ Si-C6	1.8682(19)/1.8680(18)	1.8424(14)/1.8470(14)	1.875(2)/1.880(3)
Si-C15/Si-C21	1.8709(19/1.8748(19)	1.8976(14)/1.8999(14)	1.870(3)/1.880(2)
C1-C2/C5-C6	1.416(2)/1.413(2)	1.4724(19)/1.4490(19)	1.408(3)/1.418(3)
C2-C3/C4-C5	1.476(3)/1.465(3)	1.390(2)/1.419(2)	1.503(3)/1.514(3)
C3-C4	1.341(3)	1.465(2)	1.519(4)
C1-C7/C6-C14	1.401(3)/1.402(3)	1.415(2)/1.4039(19)	1.405(3)/1.401(3)
C7-C8/C14-C13	1.389(3)/1.386(3)	1.382(2)/1.387(2)	1.388(3)/1.393(4)
C8-C9/C13-C12	1.391(3)/1.382(3)	1.420(2)/1.403(2)	1.382(4)/1.388(4)
C9-C10/C12-C11	1.379(3)/1.382(3)	1.358(2)/1.370(2)	1.384(4)/1.373(4)
C10-C2/C5-C11	1.403(3)/1.408(2)	1.457(2)/1.447(2)	1.394(3)/1.404(3)
HOMA values of Ring a/b	0.99/0.98	0.80/0.90	0.99/0.98

3. Comparison of Si–C and C–C bond lengths of 1, 2 and 3, and HOMA values



Table S1Comparison of Si–C and C–C bond lengths of 1, 2 and 3, and HOMA values for the six-memberedrings.

4. Comparison of ¹H NMR chemical shifts of 2 and some dianionic aromatic/antiaromatic compounds

Although ¹H NMR signals of the dibenzosilepin core underwent upfield-shifts upon reduction, it is not clear whether these shifts are the diagnostic for the antiaromaticity of **2** because both anionic charges and paratropic ring current induce upfield shifts. However, it is worth comparing the ¹H NMR data of **2** and other dianionic compounds having aromaticity or antiaromaticity (Table S2). The lithium salt of dibenzo[*a,e*]cyclooctatetraene dianion (**DBCOT**^{2–}),⁹ and the lithium salt of phenanthrene dianions (**Phen**^{2–}, **Me**₂**Phen**^{2–})¹⁰ were selected as representatives for negatively charged aromatic and antiaromatic compounds, respectively. The former compound shows ¹H NMR signals at common aromatic region [δ 6.98 (COT), 6.19 and 7.83 (fused benzene rings)] despite its negative charges. In contrast, ¹H NMR signals of the latter compounds appeared in high-field region ranging from δ –1.14 to 2.75 for **Phen**^{2–} and δ 1.23 to 4.32 for **Me**₂**Phen**^{2–} due to the paratropic ring current as well as the negative charges. Considering that the ¹H NMR signals for **2** recorded in THF-*d*₈ at room temperature ranged from δ 2.88 to 5.63, it is suggested that **2** has a weak antiaromatic character originating from its pseudo 16π -electron system involving negative hyperconjugation. The reason for the smaller upfield shifts in **2** compared to those in **Phen**^{2–} would originate from the hyperconjugation that is less effective than the conventional π -conjugation.

Table S2 ¹H NMR chemical shifts for dilithium salts of dianionic compounds having (anti)aromaticity recorded in THF- d_8 .

$\begin{bmatrix} H_A & H_B \\ H_B & H_C \\ H_C & H_C \\ H_E & H_D \end{bmatrix}^{2-1}$	$\begin{bmatrix} H_A H_B \\ H_C \end{bmatrix}^2$ DBCOT ²⁻	$- \left[\underbrace{H_{A}}_{H_{B}} + H_{C} \right]^{2-1}$ $H_{E} + H_{D}$ H_{C} H_{C} H_{C}	$ \begin{bmatrix} H_A & 2- \\ H_B & H_C \\ H_C & H_D \end{bmatrix} $ Me Me H _D
2	DBCOT2-	Dhon ²⁻	Me-Phen ²⁻
2	DBCOT	FIICH	
2.88 (H _A)	6.98 (H _A)	–1.14 (H _A)	1.23 (H _A)
4.22 (H _D)	6.19 (H _B or H _C)	0.62 (H _B)	2.40 (H _B)
4.99 (H _B)	7.83 (H_B or H_C)	0.80 (H _E)	3.62 (H _D)
5.34 (H _c)	-	1.69 (H _D)	4.32 (H _c)
5.63 (H _E)	-	2.75 (H _c)	-

5. VT-¹H NMR measurements of 2 in THF-d₈

VT-¹H NMR measurements in THF- d_8 were conducted. In the spectra recorded in THF- d_8 , the signals of the fused benzene rings shifted to a low-field region by ca. 0.34–0.40 ppm upon heating from -80 °C to 60 °C (Figure S2). This temperature dependence suggests that antiaromaticity of **2** become weaker as the temperature increases. Our interpretation for this phenomenon at this point is as follows; the dibenzosilepinide core is more twisted at higher temperatures to minimize destabilization caused by its antiaromaticity. In fact, planarity of an antiaromatic molecule largely affect its ¹H NMR chemical shifts. For instance, as mentioned in the previous page, the ¹H NMR signals of **Me₂Phen²⁻**, which has a highly twisted structure because of the steric repulsion between the Me groups, underwent notable downfield shifts compared to those of unsubstituted **Phen²⁻** (Table S2).⁹



Figure S1 VT-¹H NMR spectra of **2** in the temperature range of -80 to 60 °C in THF- d_8 .

6. Resonance structures, ¹³C NMR chemical shifts, and NPA charges of 2



Figure S2 Dominant resonance structures (top), ${}^{13}C{}^{1}H$ NMR chemical shifts of **2** in THF- d_8 (bottom, left) and the NPA charges of **2'** (bottom, right).

7. VT-⁷Li{¹H} NMR measurements of 2 in THF- d_8



Figure S3 VT-⁷Li NMR spectra of **2** in THF- d_8 .

8. NICS $_{\mbox{\scriptsize iso}}$ values for 1, 2', and a model compound 4

NICS_{iso} values¹¹ for **1**, **2'**, and dianion of dibenzothiepin **4**, which is a sulfur analog of **2'** as a model compound in which there is no contribution of hyperconjugation, were conducted (Table S3). Since these molecules are non-planar, NICS(+1) and NICS(-1) were defined as shown in Figure S4. As expected, the silepin and the flanking benzene rings of **1** are nonaromatic and aromatic, respectively. In stark contrast, the NICS values for both six- and seven-membered rings in **2'** were calculated to be positive, suggesting its antiaromatic character. It is noted that the NICS values of **4** much lower than **2'** are indicative of its nonaromatic nature. These results clearly indicate that the positive NICS values of **2'** originate from the negative hyperconjugation.

Table S3NICS(X) values for 1, 2' and 4.

1

	NICS(+1)				
	NIC.	5(0)			
	NICS	6(-1)			
	six-membered rings	seven-membered			
		ring			
1	-10.6/-10.6	-1.2			
	-7.7/-7.7	2.5			
	-10.7/-10.8	-1.2			
2′	7.0/8.0	2.1			
	13.6/12.8	5.1			
	8.2/7.5	4.2			
4	1.5	-4.6			
	4.1	-3.5			
	2.0	-3.6			





Figure S4 The positions for the dummy atoms for the NICS calculations.

2'

9. Important molecular orbitals to understand the electronic structures of 1 and 2

Figure S5 illustrates the frontier MOs for 1, 2, and 2'. Since compound 2 is a two-electron reduced form of 1, the HOMO and HOMO-1 of 2 essentially correspond to the LUMO and HOMO of 1, respectively. In contrast to the orbital overlap between the π -clouds and the $\sigma^*(Si-Ph)$ orbital confirmed in the HOMO-1 of 2 and 2', no such overlap was observed in the HOMO of 1, suggesting that the negative charges are essential to realize the hyperconjugation in this system.



Figure S5 Frontier MOs of 1, 2, and 2' (isovalue = 0.02).

10. X-ray crystallographic data for 1, 2 and 3

	1	2	3
CCDC	2100087	2100088	2100091
formula	$C_{26}H_{20}Si$	$C_{46}H_{60}Li_2O_5Si$	$C_{26}H_{22}Si$
fw	360.51	734.91	362.52
crystal size	$0.08 \times 0.07 \times 0.06$	$0.15 \times 0.09 \times 0.07$	$0.09 \times 0.05 \times 0.05$
crystal system	triclinic	monoclinic	monoclinic
space group	<i>P</i> -1	P2 ₁ /n	P2 ₁ /n
a [Å]	9.1512(14)	10.6771(12)	8.9514(18)
b [Å]	10.0552(16)	23.008(2)	19.204(4)
<i>c</i> [Å]	11.032(2)	16.8602(18)	11.397(2)
α [deg]	79.734(7)	90	90
β [deg]	81.052(8)	98.541(2)	98.546(4)
γ [deg]	74.730(8)	90	90
V [ų]	957.3(3)	4095.9(8)	1937.4(7)
Ζ	2	4	4
$ ho_{ m calcd}$ [g cm ⁻³]	1.251	1.192	1.243
F(000)	380	1584	768
μ [cm ⁻¹]	1.3	1.02	1.29
transmission	0.0544.4	0.0544 4	0.0544 4
factors range	0.8544– 1	0.8544– 1	0.8544 – 1
index range	-10 ≤ h ≤ 11	-10 ≤ h ≤ 13	-11 ≤ h ≤ 10
	-12 ≤ k ≤ 12	-29≤ k ≤ 29	10 ≤ h ≤ -24
	-12 ≤ l ≤ 13	-21 ≤ l ≤ 21	-24 ≤ h ≤ 24
no. reflections	7343	32961	15666
unique (R _{int})	3858 (0.0322)	9361 (0.0375)	4365 (0.0748)
<i>l</i> > 2 <i>σ</i> (l)	2719	7798	3034
no.			
parameters	244	520	245
$R_1 (I > 2\sigma(I))^a$	0.045	0.0494	0.0665
wR_2 (all data) ^b	0.1109	0.1285	0.1568
GOF °	1.009	1.053	1.048
max diff peak			
/ hole [e Å⁻³]	0.271/-0.359	0.285/-0.352	0.293/-0.358

Table S4X-ray crystallographic data for 1, 2 and 3.

^a $R1 = \Sigma ||F_o| - |F_c||/\Sigma |F_o|$. ^b $wR2 = [\Sigma \{w(F_o^2 - F_c^2)^2\}/\Sigma w(F_o^2)^2]^{1/2}$, $w = 1/[\mathbb{P}^2 F_o^2 + (aP)^2 + bP]$ (*a* and *b* are constants suggested by the refinement program; $P = [max(F_o^2, 0) + 2F_c^2]/3$). ^cGOF = $[\Sigma w(F_o^2 - F_c^2)^2/(N_{obs} - N_{params})]^{1/2}$.

11. Cartesian coordinates for 1, 2, 2' and 4

Table S5Cartesian coordinates for 1. $E_{total} = -1292.004537$ Hartree

Si 0.195113	-0.000118	0.00988
C -0.588265	-1.530735	0.77928
C -0.586987	1.531084	0.779403
C 0.214808	-2.586784	1.24507
Н 1.295776	-2.483581	1.212535
C -1.998066	1.686235	0.830124
C -1.999492	-1.684611	0.829964
C 0.217051	2.586366	1.24529
H 1.297921	2.482181	1.212757
C -0.097618	0.000051	-1.859661
C 2.999795	-0.00068	-0.690539
H 2.656981	-0.000183	-1.721793
C -0.329049	3.772099	1.736367
Н 0.322393	4.57005	2.083034
C 2.058099	-0.000867	0.353969
C -2.534874	-2.901085	1.301788
Н -3.615246	-3.025285	1.322554
C -0.189887	-1.204195	-2.581262
Н -0.136314	-2.153894	-2.05373
C -2.532353	2.903130	1.302093
Н -3.612615	3.028281	1.322915
C -1.718631	-3.929806	1.75825
Н -2.162408	-4.851303	2.126251
C 2.551012	-0.001533	1.674839
Н 1.855273	-0.001705	2.511553
C -2.980406	0.678491	0.409111
Н -3.932603	1.121927	0.11827

C 2.980961	-0.675979	0.409037
Н -3.933522	-1.118592	0.118128
C -1.715175	3.931076	1.75863
Н -2.158114	4.85294	2.126722
C 4.372646	-0.001123	-0.431366
Н 5.079027	-0.000969	-1.257804
C -0.332362	-3.772072	1.736034
H 0.318359	-4.570634	2.082642
C-0.188485	1.204399	-2.581263
Н -0.133816	2.154043	-2.053732
C -0.445426	0.000254	-4.662477
Н -0.581050	0.000328	-5.741163
C -0.360834	-1.207149	-3.9666910
Н -0.432112	-2.150777	-4.502072
C 3.920860	-0.001984	1.939804
Н 4.274306	-0.002494	2.967956
C 4.836011	-0.001777	0.884341

Table S6-1Cartesian coordinates for **2**.

E_{total} = -2468.881510 Hartree

Si -1.336557	-1.462434	0.185844	C -5.583085	-3.554610	-0.290742
C -1.520342	0.044772	1.274385	Н -6.557372	-4.027402	-0.390209
C -0.348284	0.906590	1.518825	С -5.076249	-2.761479	-1.321219
C 0.546476	1.301598	0.513889	Н -5.653101	-2.620240	-2.233055
H 1.270166	2.056864	0.841471	С -3.822774	-2.158000	-1.189178
C 0.505825	1.141267	-0.952710	Н -3.435202	-1.564313	-2.012477
H 1.015235	1.960481	-1.468786	Li -1.459906	1.989595	-0.148115
C 0.101747	0.098947	-1.836837	O -1.800231	3.826967	0.630056
C -0.587689	-1.137192	-1.496652	C -0.798360	4.631019	1.291111
C -2.548895	0.041534	2.250194	Н -0.521198	5.455636	0.619922
Н -3.429539	-0.572112	2.062658	Н 0.075918	4.000576	1.467924
C -2.471890	0.724416	3.459333	C -1.448607	5.142381	2.578561
Н -3.282006	0.680359	4.181562	H -1.034000	6.101435	2.904945
C -1.244006	1.366889	3.778002	H -1.310108	4.408164	3.378638
H -1.108382	1.807401	4.765854	C -2.930855	5.220480	2.182542
C -0.217646	1.433832	2.866649	Н -3.131446	6.136579	1.613513
H 0.706477	1.948022	3.133360	Н -3.608612	5.195827	3.041181
C 0.407233	0.270300	-3.241075	C -3.081880	3.987376	1.291344
H 0.866957	1.211005	-3.547647	Н -3.284925	3.086934	1.884034
C 0.161038	-0.695776	-4.195185	H -3.844891	4.094071	0.515903
H 0.427248	-0.497506	-5.233217	O -2.984126	2.092212	-1.467503
C -0.422260	-1.928527	-3.839366	C -4.294976	1.491481	-1.326787
Н -0.625654	-2.693782	-4.582773	Н -4.952339	2.192085	-0.798064
C -0.790316	-2.101254	-2.504163	Н -4.177800	0.591536	-0.718057
Н -1.301977	-3.025747	-2.237321	C -4.790936	1.186568	-2.752587
C -0.296466	-2.737272	1.179981	Н -5.290272	0.215123	-2.807832
C -0.213129	-2.672736	2.584541	Н -5.505144	1.949529	-3.083608
Н -0.699207	-1.853936	3.109177	C -3.506696	1.253003	-3.598700
C 0.484093	-3.635846	3.321412	Н -2.962880	0.303163	-3.576285
H 0.524441	-3.561470	4.406211	Н -3.697659	1.516220	-4.643888
C 1.123308	-4.692232	2.670069	C -2.700591	2.318607	-2.862726
C 0.351306	-3.817879	0.549759	Н -1.621067	2.223325	-2.995257
H 0.301316	-3.910343	-0.532796	Н -3.019275	3.334941	-3.140095
C -3.036157	-2.326855	-0.034280			

Table S6-2 Cartesian coordinates for 2.

Li 2.596637	0.379102	-0.250808	Н 4.291959	3.393877	0.318510	
O 3.556665	0.028708	1.534802	H 2.651615	3.543891	-0.361607	
C 3.975546	1.061177	2.452713	O 3.492117	-1.124760	-1.236625	
H 3.240318	1.875319	2.419369	C 2.862894	-2.213111	-1.975062	
H 4.945130	1.447053	2.119778	H 2.208033	-1.778525	-2.730669	
C 4.021182	0.407256	3.837351	Н 2.255797	-2.792090	-1.273086	
Н 3.827459	1.124832	4.640464	C 4.019869	-3.039722	-2.534839	
H 5.004495	-0.044633	4.015940	Н 4.363434	-2.628572	-3.492067	
C 2.941453	-0.678860	3.714506	Н 3.740242	-4.084946	-2.696795	
H 3.063408	-1.494372	4.433089	C 5.097589	-2.856872	-1.455517	
H 1.942107	-0.249738	3.843637	H 1.664539	-5.443123	3.241219	
C 3.125874	-1.147013	2.273193	C 1.053311	-4.781611	1.277665	
H 3.910033	-1.911881	2.191378	H 1.540107	-5.605551	0.759829	
H 2.208993	-1.521116	1.816591	C -3.568117	-3.141985	0.983603	
O 3.762800	1.922744	-1.032898	H -2.988104	-3.318266	1.886276	
C 3.966435	1.872855	-2.469166	C -4.822550	-3.744330	0.864150	
H 3.258940	1.155157	-2.889533	Н -5.202302	-4.368336	1.670082	
H 4.988764	1.520740	-2.658021	H 6.113024	-3.060480	-1.809723	
C 3.765211	3.304024	-2.970020	H 4.895833	-3.513684	-0.600994	
H 4.336649	3.511019	-3.879786	C 4.900739	-1.391976	-1.061875	
H 2.705869	3.487830	-3.183211	Н 5.162364	-1.164121	-0.02171	
C 4.219159	4.135931	-1.761174	Н 5.471274	-0.724475	-1.722705	
H 3.817127	5.153473	-1.757230				
Н 5.313270	4.200853	-1.728115				
C 3.695751	3.297576	-0.593773				

Table S7Cartesian coordinates for 2'.

E_{total} = -1291.964330 Hartree

Si	-0.11716629	-0.01426454	-0.08190402
С	0.96477815	-1.35028579	-0.81490225
С	0.19497687	1.72540178	-0.69131096
С	0.43132567	-2.64067433	-0.95435333
Η	-0.63955170	-2.77643889	-0.79670816
С	1.53585566	2.22477940	-1.01198361
С	2.39534576	-1.14317841	-1.04747321
С	-0.86845722	2.64545734	-0.61594774
Η	-1.86172427	2.26494114	-0.37225164
С	0.05937163	-0.03787417	1.82060838
С	-2.84761542	-1.02639160	0.41875359
Η	-2.53371110	-1.19433537	1.44677918
С	-0.73787086	4.02213761	-0.83518022
Η	-1.60307101	4.68312796	-0.80304271
С	-1.94100347	-0.45788722	-0.49167470
С	3.17823447	-2.35807488	-1.21459912
Η	4.26393692	-2.25606691	-1.28353921
С	1.15287869	-0.67611937	2.45196381
Η	1.89235921	-1.17850191	1.83163361
С	1.64190072	3.66250401	-1.18323140
Η	2.63454635	4.07048533	-1.38833769
Η	-4.83579411	-1.81687577	0.76857711
С	1.20042596	-3.77807731	-1.25366037
Η	0.73924557	-4.75933319	-1.35516902
С	-0.86714870	0.60977569	2.67961147
Η	-1.70366700	1.15619227	2.24695188
С	0.34921615	-0.07419919	4.66927204
Η	0.45847362	-0.09186281	5.75341270
С	1.30399751	-0.69064092	3.83957535
С	-0.73663015	0.58000546	4.07012680
Η	-1.48054141	1.08100818	4.69167652

Η	2.16362518	-1.19560493	4.28126347
С	-3.67239428	-0.66833568	-2.22143230
Н	-3.98171430	-0.54044623	-3.25943288
С	-4.56196068	-1.21402559	-1.29044437
Η	-5.56789882	-1.50547869	-1.59681541
С	2.60969843	-3.60746335	-1.33336075
Η	3.25595142	-4.47595631	-1.48635237
С	-2.38398668	-0.28507744	-1.82004542
Η	-1.70729396	0.15998376	-2.54702817
С	2.71964684	1.47220640	-1.14973515
Η	3.59551566	2.10577044	-1.32299557
С	3.07488024	0.08929026	-1.13130333
Н	4.15317242	-0.04480342	-1.26918430
С	0.56757933	4.51972064	-1.11302769
Н	0.72385066	5.59046156	-1.26606297
С	-4.15060817	-1.38906712	0.03596814

Table S8Cartesian coordinates for 4.

E_{total} = -937.461840 Hartree

С	2.481596	-1.441241	0.248612
Н	2.239814	-2.507023	0.250506
С	-1.464149	-0.545765	-0.076898
С	0.718876	1.895656	-0.151554
Η	1.137417	2.907996	-0.196830
С	-1.697538	0.899835	-0.051485
С	-0.718878	1.895656	-0.151553
Η	-1.137420	2.907995	-0.196832
С	1.464150	-0.545765	-0.076899
С	1.697536	0.899836	-0.051480
С	-2.481593	-1.441241	0.248621
Η	-2.239809	-2.507023	0.250521
С	-3.101596	1.260813	0.056709
Η	-3.364001	2.315274	-0.074878
С	3.801969	-1.031202	0.528369
Н	4.572799	-1.753006	0.798578
С	3.101593	1.260816	0.056720
Н	3.363997	2.315278	-0.074861
С	-4.096907	0.351886	0.360612
Н	-5.126702	0.705784	0.467894
С	-3.801966	-1.031202	0.528377
Н	-4.572795	-1.753005	0.798592
С	4.096906	0.351888	0.360615
Η	5.126700	0.705788	0.467900
S	0.000000	-1.251118	-0.841448



12. ¹H, ¹³C{¹H}, ²⁹Si{¹H}, ⁷Li{¹H} NMR spectra for the new compounds

Figure S6 ¹H NMR spectrum of **1** in CDCl₃.



Figure S7 ${}^{13}C{}^{1}H$ NMR spectrum of 1 in CDCl₃.



Figure S8 29 Si{¹H} NMR spectrum of **1** in CDCl₃.



Figure S9 ¹H NMR spectrum of **2** in THF- d_8 .



Figure S10 ${}^{13}C{}^{1}H$ NMR spectrum of **2** in THF- d_8 .



Figure S11 ²⁹Si{¹H} NMR spectrum of **2** in THF- d_8 .



Figure S12 ⁷Li{¹H} NMR spectrum of **2** in THF- d_8 .



Figure S13 ¹H NMR spectrum of **3** in CDCl₃.



Figure S14 ${}^{13}C{}^{1}H$ NMR spectrum of 3 in CDCl₃.



Figure S15 ²⁹Si{¹H} NMR spectrum of **3** in CDCl₃.

13. References

- 1 L. G. Mercier, W. E. Piers and M. Parvez, Angew. Chem., Int. Ed., 2009, 48, 6108-6111.
- 2 R. A. Jacobson, Private Communication to Rigaku Corp.; Rigaku Corp.; Tokyo Japan, 1998.
- 3 Crystal Structure 4.0: Single Crystal Structure Analysis Package; Rigaku Corp.; Tokyo, Japan 2000-2010.
- 4 G. Sheldrick, Acta Crystallogr. Sec. C, 2015, 71, 3-8.
- 5 A. Altomare, M. C. Burla, M. Camalli, G. L. Cascarano, C. Giacovazzo, A. Guagliardi, A. G. G. Moliterni, G. Polidori and R. Spagna, *J. Appl. Cryst.*, 1999, **32**, 115-119.
- M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, Williams, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman and D. J. Fox, *Gaussian 16 Rev. C.01*, Gaussian, Inc., Wallingford, CT, 2016.
- M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. J. A. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman and D. J. Fox, *Gaussian 09 Rev. E.01*, Gaussian, Inc., Wallingford CT, 2016.
- 8 (a) C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B*, 1988, 37, 785-789; (b) A. D. Becke, *J. Chem. Phys.*, 1993, 98, 5648-5652.
- 9 Y. Zhu, Z. Zhou, Z. Wei and M. A. Petrukhina, Organometallics, 2020, 39, 4688-4695.
- 10 R. Frim, A. Mannschreck and M. Rabinovitz, Angew. Chem., Int. Ed. Eng., 1990, 29, 919-921.
- P. v. R. Schleyer, C. Maerker, A. Dransfeld, H. Jiao and N. J. R. v. E. Hommes, J. Am. Chem. Soc., 1996, 118, 6317-6318.