

## Supplementary Information

### **Stereoselective total synthesis of (±)-vindeburnol and (±)-16- *epi*-vindeburnol**

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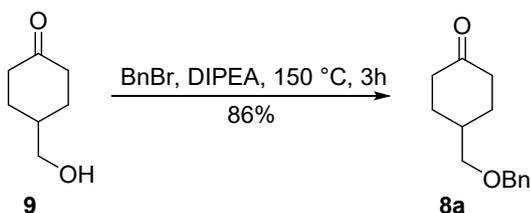
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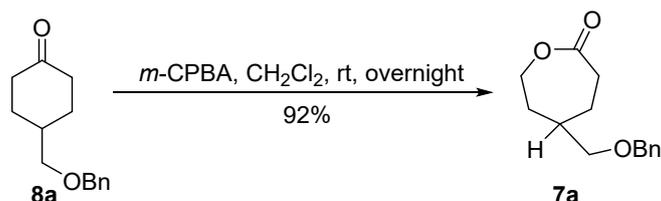
## 1. General Information

All commercially available reagents were used without further purification. Tetrahydrofuran was distilled from sodium/benzophenone ketyl. Dichloromethane was distilled from calcium hydride. DMSO was dried from the activated molecular sieve. Chromatography was conducted by using 200–300 mesh silica gel. All new compounds gave satisfactory spectroscopic analyses (IR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, HRMS). NMR spectra were recorded on a 400 MHz NMR spectrometer. Chemical shifts are reported relative to the residue peaks of the solvent ( $\text{CDCl}_3$ : 7.26 ppm for  $^1\text{H}$  and 77.0 ppm for  $^{13}\text{C}$ ) ( $\text{DMSO}-d_6$ : 2.50 ppm for  $^1\text{H}$  and 39.5 ppm for  $^{13}\text{C}$ ). Coupling constants ( $J$ ) are given in Hz and are uncorrected and multiplicities for coupled signals were denoted as: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br. = broad, apt. = apparent and dd = double doublet etc. Infrared (IR) spectra were recorded on a Perkin Elmer Spectrum Two FT-IR spectrometer. High-resolution mass spectra (HRMS) were recorded on a Bruker TOF Premier, by the ESI method. Melting points (m.p.) were recorded on an SRS-optic melting point apparatus. X-ray diffraction analysis was carried out by Dr. Meng Yang (Sichuan University). The data collections were done on a Bruker Xcalibur E using  $\text{MoK}\alpha$ -radiation. HPLC analysis was performed with Daicel Chiralpak OD column (4.6  $\times$  250 mm).

## 2. Experimental Procedures

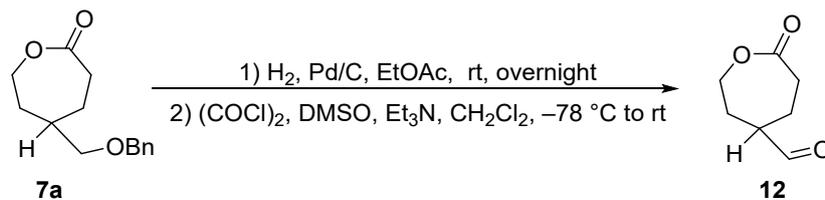


Commercially available 4-(hydroxymethyl)cyclohexan-1-one **9** (0.60 g, 4.68 mmol) was added to a solution of BnBr (0.67 mL, 5.61 mmol) in DIPEA (1.6 mL, 9.4 mmol). The resulting mixture was refluxed at 150 °C in sealed tube for 3 h and quenched by adding water at room temperature, extracted by EtOAc (3  $\times$  30 mL). The combined organic layer was washed with brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated in vacuo. The residue was purified by silica gel flash column chromatography [petroleum ether/EtOAc = 10:1,  $R_f$  (**8a**) = 0.4] to give **8a** (0.88 g, 86%) as a pale yellow oil.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.32 – 7.18 (m, 5H), 4.44 (s, 2H), 3.31 (d,  $J$  = 6.0 Hz, 2H), 2.34 – 2.21 (m, 4H), 2.09 – 1.95 (m, 3H), 1.43 – 1.32 (m, 2H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  211.9, 138.4, 128.5, 127.7, 127.6, 74.3, 73.2, 40.5, 36.6, 29.6. IR (neat) 1711, 1452, 1245, 1170, 1099, 919, 737, 698, 503  $\text{cm}^{-1}$ . HRMS (ESI)  $m/z$  calcd for  $\text{C}_{14}\text{H}_{19}\text{O}_2$  [ $\text{M} + \text{H}$ ] $^+$ : 219.1380, found: 219.1386.



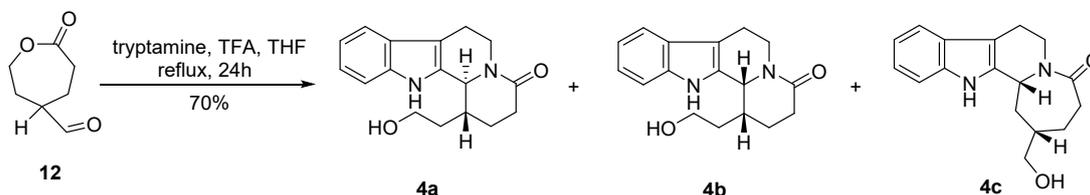
To a stirred solution of compound **8a** (0.90 g, 4.12 mmol) in  $\text{CH}_2\text{Cl}_2$  (20.0 mL) was added *m*-CPBA (0.92 g, 4.53 mmol). The reaction mixtures were stirred for overnight at room temperature. The saturated  $\text{K}_2\text{CO}_3$

aqueous solution (50 mL) was added and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). Then the organic layer was concentrated in vacuo and the crude products were purified directly by column chromatography on silica gel [petroleum ether/EtOAc = 3:1, R<sub>f</sub> (**7a**) = 0.2] to afford the desired products **7a** (0.89 g, 92%) as a pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.31 – 7.20 (m, 5H), 4.43 (s, 2H), 4.27 – 4.20 (m, 1H), 4.14 – 4.07 (m, 1H), 3.31 – 3.22 (m, 2H), 2.67 – 2.50 (m, 2H), 2.05 – 1.84 (m, 3H), 1.54 – 1.43 (m, 1H), 1.38 – 1.27 (m, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 175.7, 138.1, 128.3, 127.6, 127.4, 74.2, 73.0, 67.8, 40.6, 32.8, 32.3, 25.7. IR (neat) 1723, 1475, 1392, 1169, 1102, 1070, 1008, 736, 698, 586 cm<sup>-1</sup>. HRMS (ESI) m/z calcd for C<sub>14</sub>H<sub>18</sub>NaO<sub>3</sub> [M + Na]<sup>+</sup>: 257.1148, found: 257.1153.



A stirred solution of **7a** (0.89 g, 3.80 mmol) in EtOAc (12.0 mL) was treated with 5% Pd/C (0.18 g). After stirring at room temperature under H<sub>2</sub> for overnight, the reaction mixture was filtrated through a pad of celite. The solid was washed with EtOAc (3 x 30 mL) and concentrated in vacuo to give the alcohol intermediate as a colorless oil, which was directly used in next step without further purification.

A round bottom flask equipped with stirring bar was charged with oxalyl chloride (0.60 mL, 7.60 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (15.0 mL) and cooled to -78 °C. Dry dimethylsulfoxide (1.11 mL, 15.6 mmol) was diluted into 10 mL dry CH<sub>2</sub>Cl<sub>2</sub>, which was added slowly for 10 minutes. After stirring for 30 min at -78 °C, the alcohol intermediate was dissolved in 5.0 mL dry CH<sub>2</sub>Cl<sub>2</sub>, which was added into the reaction mixture over a period of 10 minutes. After stirring for additional 30 min at -78 °C, triethylamine (2.80 mL) was added and the reaction allowed to warm to room temperature. The reaction mixture was diluted with 20 mL H<sub>2</sub>O and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 30 mL). The combined organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuum. The residue was purified directly by column chromatography on silica gel [petroleum ether/EtOAc = 3:1, R<sub>f</sub> (**12**) = 0.3] to afford the corresponding product **12** (0.49 g, 90%) as a pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 9.77 (s, 1H), 4.39 – 4.35 (m, 1H), 4.03 – 3.98 (m, 1H), 2.66 – 2.49 (m, 5H), 2.13 – 2.04 (m, 1H), 1.63 – 1.53 (m, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 199.4, 170.9, 72.4, 45.6, 28.8, 27.3, 25.3. IR (neat) 2933, 1706, 1260, 1168, 971, 914, 803, 733 cm<sup>-1</sup>. HRMS (ESI) m/z calcd for C<sub>7</sub>H<sub>11</sub>O<sub>3</sub> [M + H]<sup>+</sup>: 143.0703, found: 143.0707.

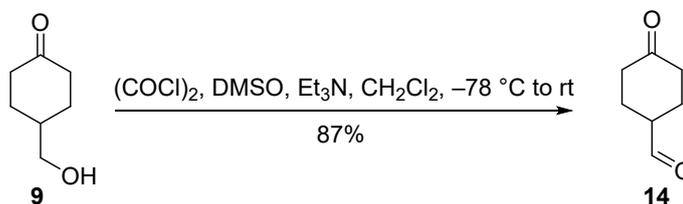


To a solution of **12** (0.47 g, 3.31 mmol) in THF (33.0 mL) was added tryptamine (0.69 g, 4.30 mmol) and TFA (0.30 mL). After refluxing for 24 h, water (10.0 mL) was added to the cooled reaction mixture and the pH was adjusted to 7.0 at 0 °C with saturated Na<sub>2</sub>CO<sub>3</sub> solution. The layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 30 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated in vacuo. The residue was purified by silica gel column chromatography [CH<sub>2</sub>Cl<sub>2</sub>/Acetone = 2:1, R<sub>f</sub> (**4a**) = 0.5, R<sub>f</sub> (**4b**) = 0.3, R<sub>f</sub> (**4c**) = 0.4] to afford **4a** (0.44 g, 47%) as a white solid, **4b** (0.11 g, 12%) as a white solid, **4c** (0.10 g, 11%) as a white solid.

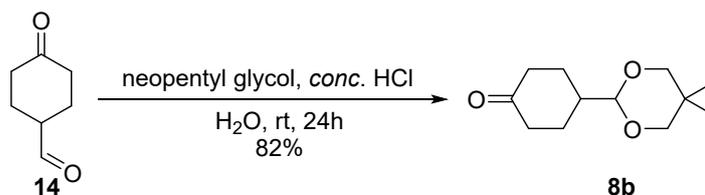
**4a:** mp = 189.8–192.5 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 10.83 (s, 1H), 7.35 (dd, *J* = 12.4, 8.0 Hz, 2H), 7.05 (td, *J* = 8.0, 1.2 Hz, 1H), 6.96 (td, *J* = 8.0, 1.2 Hz, 1H), 4.74 – 4.70 (m, 3H), 3.70 – 3.60 (m, 2H), 2.88 (td, *J* = 12.0, 4 Hz, 1H), 2.82 – 2.75 (m, 1H), 2.60 – 2.53 (m, 2H), 2.39 – 2.30 (m, 1H), 2.12 – 2.05 (m, 1H), 1.94 – 1.85 (m, 1H), 1.78 – 1.70 (m, 1H), 1.66 – 1.56 (m, 2H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 169.0, 135.8, 135.1, 126.8, 120.9, 118.7, 117.5, 111.4, 108.4, 59.2, 58.9, 42.1, 34.4, 31.3, 28.8, 22.7, 20.5. IR (KBr) 3258, 2931, 2866, 1610, 1471, 1409, 1355, 1303, 1234, 1056, 1028, 736, 677, 511 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>NaO<sub>2</sub> [M + Na]<sup>+</sup>: 307.1417, found: 307.1424;

**4b:** mp = 203.9–204.8 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 10.87 (s, 1H), 7.42 (d, *J* = 7.6 Hz, 1H), 7.34 (d, *J* = 8.0 Hz, 1H), 7.06 (t, *J* = 8.0 Hz, 1H), 6.97 (t, *J* = 7.6 Hz, 1H), 4.94 – 4.91 (m, 2H), 4.31 (t, *J* = 4.8 Hz, 1H), 3.31 – 3.23 (m, 2H), 2.70 (td, *J* = 14.8, 3.2 Hz, 2H), 2.60 – 2.53 (m, 2H), 2.38 – 2.29 (m, 2H), 1.98 – 1.87 (m, 2H), 1.28 – 1.18 (m, 1H), 1.00 – 0.92 (m, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 168.3, 136.4, 132.8, 126.3, 120.9, 118.5, 117.6, 111.1, 108.8, 58.9, 57.7, 38.6, 32.3, 27.7, 27.3, 22.2, 20.7. IR (KBr) 3190, 2931, 2850, 1612, 1491, 1435, 1294, 1226, 1080, 1209, 935, 871, 748, 688 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>NaO<sub>2</sub> [M + Na]<sup>+</sup>: 307.1417, found: 307.1424;

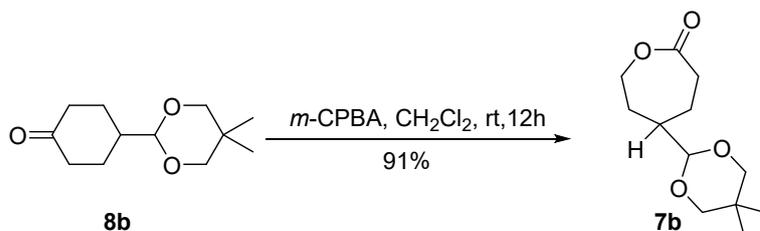
**4c:** mp = 228.6–230.3 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 10.99 (s, 1H), 7.41 (d, *J* = 7.6 Hz, 1H), 7.33 (d, *J* = 8 Hz, 1H), 7.05 (t, *J* = 7.6 Hz, 1H), 6.96 (t, *J* = 7.6 Hz, 1H), 5.11 (d, *J* = 9.6 Hz, 1H), 4.71 (dd, *J* = 12.8, 4.8 Hz, 1H), 4.61 (t, *J* = 5.6 Hz, 1H), 3.29 – 3.24 (m, 2H), 2.95 (t, *J* = 13.6 Hz, 1H), 2.85 (td, *J* = 12.4, 3.6 Hz, 1H), 2.77 (td, *J* = 16.0, 4 Hz, 1H), 2.58 – 2.52 (m, 1H), 2.36 (q, *J* = 6.8 Hz, 1H), 2.25 – 2.21 (m, 1H), 2.04 – 1.92 (m, 2H), 1.36 (q, *J* = 11.2 Hz, 1H), 1.13 (q, *J* = 13.2 Hz, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 174.5, 136.3, 134.2, 125.9, 120.9, 118.4, 117.7, 111.0, 107.7, 66.0, 52.5, 43.2, 36.8, 36.1, 34.8, 26.4, 20.6. IR (KBr) 3325, 2964, 2901, 1620, 1502, 1429, 1354, 1307, 1172, 1068, 1004, 933, 883, 852 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>NaO<sub>2</sub> [M + Na]<sup>+</sup>: 307.1417, found: 307.1425.



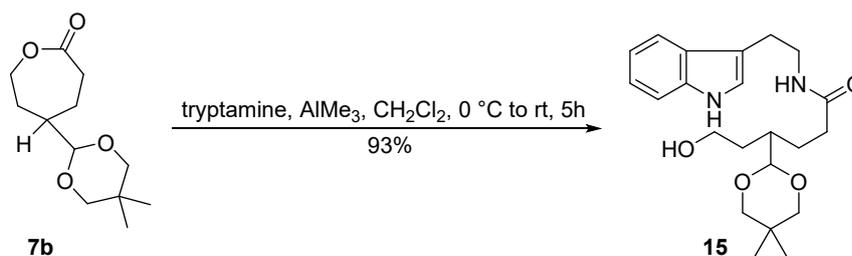
A round bottom flask equipped with stirring bar was charged with (1.30 mL, 15.60 mmol) oxalyl chloride in dry CH<sub>2</sub>Cl<sub>2</sub> (25.0 mL) and cooled to -78 °C. Dry dimethylsulfoxide (2.26 mL, 32.0 mmol) diluted with 10 mL dry CH<sub>2</sub>Cl<sub>2</sub>, was added slowly for 10 min. After stirring for another 30 min at -78 °C, **9** (1.0 g, 7.80 mmol) dissolved in 15 mL dry CH<sub>2</sub>Cl<sub>2</sub>, was added over a period of 20 minutes. After stirring for additional 30 min at -78 °C, trimethylamine (5.70 mL) was added and the reaction allowed to warm to room temperature. The reaction mixture was diluted with 30 mL H<sub>2</sub>O and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuum to give the crude product. The residue was purified by silica gel flash column chromatography [petroleum ether/EtOAc = 3:1, R<sub>f</sub> (**14**) = 0.3] to give **14** (0.86 g, 87%) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 9.71 (s, 1H), 2.67 – 2.59 (m, 1H), 2.45 – 2.28 (m, 4H), 2.21 – 2.13 (m, 2H), 1.97 – 1.87 (m, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 209.7, 202.5, 47.3, 39.5, 25.5. IR (neat) 2856, 1731, 1339, 1272, 1088, 905, 803, 701, 646 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>7</sub>H<sub>11</sub>O<sub>2</sub> [M + H]<sup>+</sup>: 127.0754, found: 127.0758.



To a stirred solution of compound **14** (0.87 g, 6.90 mmol) in H<sub>2</sub>O (14.0 mL) was added neopentyl glycol (0.72 g, 6.90 mmol) and *con.* HCl (4.0 mL). After at room temperature for 24 h, the precipitate was filtered, washed with ice-cold slightly alkaline water and Et<sub>2</sub>O. The aqueous layers were extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuum to give the crude product. The solid was dried under vacuum to give the residue as a pale yellow solid. The residue was purified by silica gel flash column chromatography [petroleum ether/EtOAc = 8:1, R<sub>f</sub> (**8b**) = 0.4] to give **8b** (1.20 g, 82%) as a white solid. mp = 75.1–75.7 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 4.29 (d, *J* = 4.8 Hz, 1H), 3.60 (t, *J* = 1.6 Hz, 1H), 3.57 (d, *J* = 1.6 Hz, 1H), 3.41 (s, 1H), 3.38 (s, 1H), 2.43 – 2.35 (m, 2H), 2.30 (td, *J* = 13.2, 6.0 Hz, 2H), 2.15 – 2.10 (m, 2H), 2.01 – 1.92 (m, 1H), 1.65 – 1.54 (m, 2H), 1.15 (s, 3H), 0.70 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 211.9, 103.7, 77.3, 40.6, 40.4, 30.3, 27.0, 22.9, 21.9. IR (neat) 2953, 2845, 1711, 1649, 1394, 1384, 1105, 925, 828, 793, 757, 656 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>12</sub>H<sub>21</sub>O<sub>3</sub> [M + H]<sup>+</sup>: 213.1485, found: 213.1489.



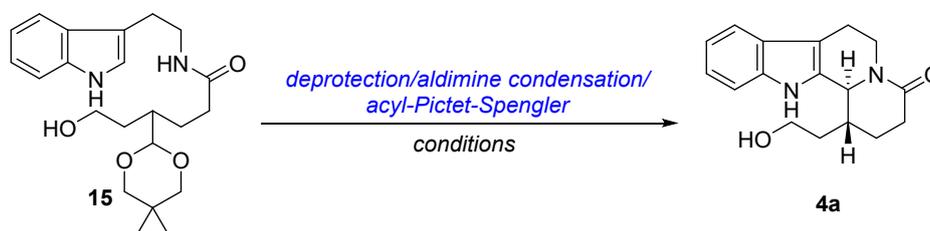
To a stirred solution of compound **8b** (0.50 g, 2.36 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (12.0 mL) was added *m*-CPBA (0.57 g, 2.83 mmol). After the reaction mixtures were stirred for 12 h at room temperature, the saturated K<sub>2</sub>CO<sub>3</sub> solution (30 mL) was added and the suspension was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). Then the combine organic layers were concentrated in vacuo and the crude products were purified directly by column chromatography on silica gel (petroleum ether/EtOAc = 3:1) to afford the desired product **7b** (0.49 g, 91%) as a white solid. mp = 122.9–125.6 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 4.32 (dd, *J* = 13.2, 6 Hz, 1H), 4.23 (d, *J* = 4.4 Hz, 1H), 4.12 (t, *J* = 10.4 Hz, 1H), 3.57 (d, *J* = 10.8 Hz, 2H), 3.37 (d, *J* = 10.8 Hz, 2H), 2.70 (dd, *J* = 14, 7.6 Hz, 1H), 2.54 (t, *J* = 13.6 Hz, 1H), 2.11 – 2.08 (m, 1H), 2.05 – 2.00 (m, 1H), 1.86 – 1.79 (m, 1H), 1.73 – 1.63 (m, 1H), 1.53 (q, *J* = 12.8 Hz, 1H), 1.11 (s, 3H), 0.68 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 175.9, 103.2, 77.2, 67.9, 44.8, 32.9, 30.2, 29.9, 23.4, 22.8, 21.8. IR (neat) 2958, 2853, 1729, 1472, 1393, 1311, 1137, 1078, 1018, 957, 929, 792, 646 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>12</sub>H<sub>20</sub>KO<sub>4</sub> [M + K]<sup>+</sup>: 267.0993, found: 267.0996.



To a stirred mixture of tryptamine (0.38 g, 2.41 mmol) and **7b** (0.50 g, 2.19 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (22.0 mL)

was added 2.0 M AlMe<sub>3</sub>-hexane (1.70 mL, 3.30 mmol) at 0 °C, and then the reaction mixture was stirred for 5 h at room temperature. The reaction was carefully quenched with H<sub>2</sub>O at 0 °C and the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 30 mL). The organic extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated to give a residue, which was chromatographed on silica gel [CH<sub>2</sub>Cl<sub>2</sub>/MeOH = 20:1, R<sub>f</sub> (**15**) = 0.5] to afford the desired product **15** (0.79 g, 93%) as a gel product. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.73 (s, 1H), 7.58 (d, *J* = 8 Hz, 1H), 7.37 (d, *J* = 8.4 Hz, 1H), 7.17 (t, *J* = 7.2 Hz, 1H), 7.09 (t, *J* = 7.6 Hz, 1H), 6.99 (s, 1H), 5.96 (s, 1H), 4.29 (d, *J* = 2.8 Hz, 1H), 3.69 – 3.61 (m, 2H), 3.58 – 3.49 (m, 4H), 3.36 – 3.29 (m, 2H), 2.93 (t, *J* = 6.8 Hz, 2H), 2.29 – 2.13 (m, 2H), 1.84 – 1.71 (m, 3H), 1.66 – 1.57 (m, 1H), 1.51 – 1.473 (m, 1H), 1.12 (s, 3H), 0.67 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 173.6, 136.6, 127.4, 122.5, 122.0, 119.3, 118.7, 112.7, 111.4, 104.1, 77.3, 77.2, 60.8, 39.8, 39.2, 34.5, 32.1, 30.2, 25.4, 25.3, 23.1, 21.8. IR (neat) 3286, 2950, 2893, 1636, 1534, 1456, 1361, 1100, 972, 925, 739, 701, 568 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>22</sub>H<sub>32</sub>N<sub>2</sub>NaO<sub>4</sub> [*M* + Na]<sup>+</sup>: 411.2254, found: 411.2262.

**Table 1** Exploration and optimization studies for synthesis of *trans*-fused indoloquinolizidine core **4a**.<sup>a</sup>



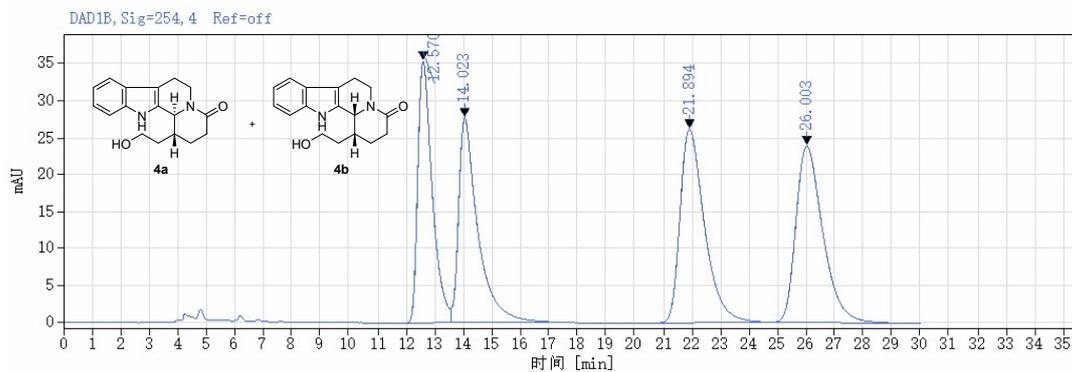
Entry	Acid	Sol.	Time (h)	Yield(%) <sup>b</sup>	<i>dr</i> <sup>c</sup>
1	TFA	DCM	16	46	5.5:1
2	TFA	MeCN	18	45	2.8:1
3	TFA	PhMe	24	54	4.6:1
4	TFA	THF	72	/	/
5	TFA	EtOH	72	/	/
6	TFA	Acetone	8	58	7.2:1
7	TsOH	Acetone	6	65	8.7:1
8	CSA	Acetone	12	81	12.5:1
9	MsOH	Acetone	24	60	10.0:1
10	AcOH	Acetone	48	/	/
11	Dichloroacetic acid	Acetone	36	83	16.5:1

<sup>a</sup>Unless noted otherwise, reactions were performed on scale of 100 mg of **15** at 25 °C. <sup>b</sup>Isolated yields. <sup>c</sup>Determined by HPLC analysis.

#### Procedure (entries 1 to 11)

Acid (10.0 equiv.) was added into a solution (4.90 mL) of **15** (0.10 g, 0.26 mmol).<sup>1</sup> After stirring at room temperature for various time, the pH was adjusted to 7.0 at 0 °C by adding saturated Na<sub>2</sub>CO<sub>3</sub> aqueous solution. The aqueous layer was extracted with DCM. The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated in vacuo. The residue was purified by silica gel flash column chromatography [(CH<sub>2</sub>Cl<sub>2</sub>/MeOH = 20:1, R<sub>f</sub> (**4a**) = 0.4, R<sub>f</sub> (**4b**) = 0.3)] to afford the product as a mixture of **4a** and **4b** compounds. Chiral HPLC conditions: Daicel Chiralcel OD column, 4.6 × 250 mm, 85:15 *n*-hexane/*i*-PrOH, 0.75 mL/min, UV detector at 254 nm. *t*<sub>R1</sub> = 12.5 min, *t*<sub>R2</sub> = 14.0 min, *t*<sub>R3</sub> = 21.9 min, *t*<sub>R4</sub> = 26.0 min. *dr* = 16.5:1 (**4a**:**4b**) (Table 1). In addition, **4a** and **4b** could be separated via column chromatography (CH<sub>2</sub>Cl<sub>2</sub>/MeOH = 30:1) to give **4a** as a white solid.

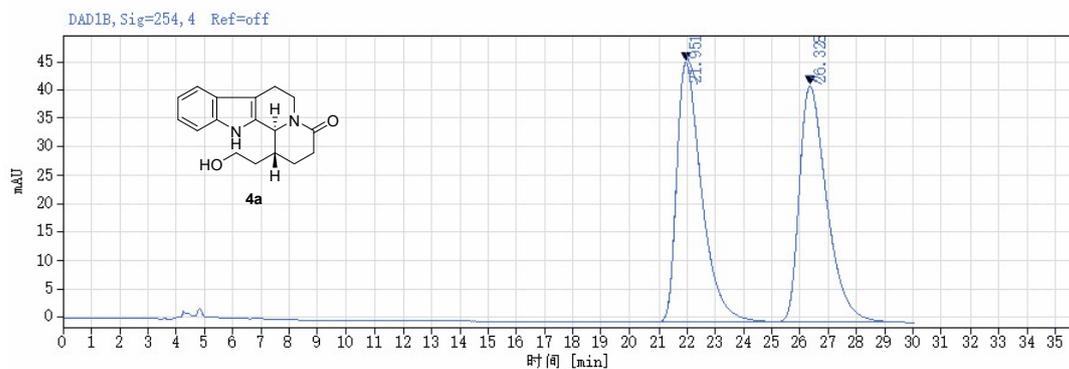
**Table 2. HPLC Spectrum of mixture of 4a and 4b**



RetTime [min]	Type	Width [min]	Area [mAU * s]	Height [mAU]	Area%
12.570	MM m	0.53	1250.48	35.48	22.03
14.023	MM m	0.67	1302.87	27.75	22.95
21.894	MM m	0.89	1557.91	26.23	27.44
26.003	MM m	0.95	1565.41	23.94	27.58
Totals			5676.67		

Chiral HPLC conditions: Daicel Chiralcel OD column, 4.6 × 250 mm, 85:15 *n*-hexane/*i*-PrOH, 0.75 mL/min, UV detector at 254 nm.  $t_{R1} = 12.5$  min,  $t_{R2} = 14.0$  min,  $t_{R3} = 21.9$  min,  $t_{R4} = 26.0$  min

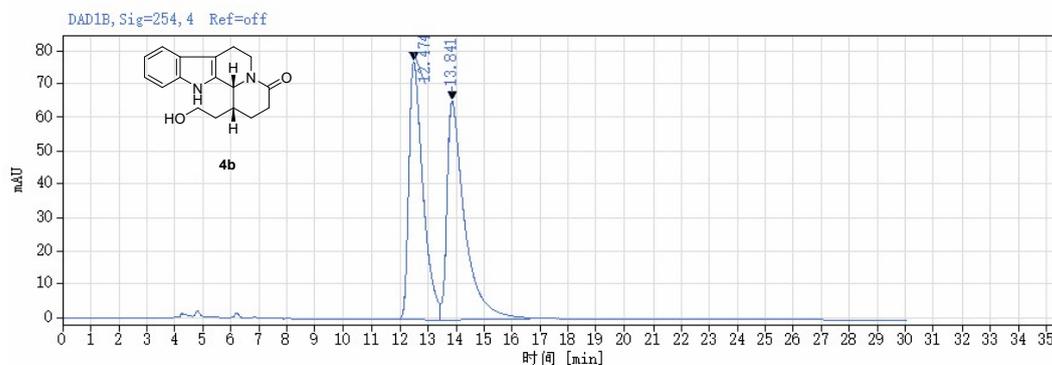
**Table 3. HPLC Spectrum of 4a**



RetTime [min]	Type	Width [min]	Area [mAU * s]	Height [mAU]	Area%
21.951	MM m	0.90	2786.11	45.80	50.02
26.328	MM m	1.00	2783.82	41.56	49.98
Totals			5569.92		

Chiral HPLC conditions: Daicel Chiralcel OD column, 4.6 × 250 mm, 85:15 *n*-hexane/*i*-PrOH, 0.75 mL/min, UV detector at 254 nm.  $t_{R3}$  (**4a**) = 21.9 min,  $t_{R4}$  (**4a**) = 26.3 min.

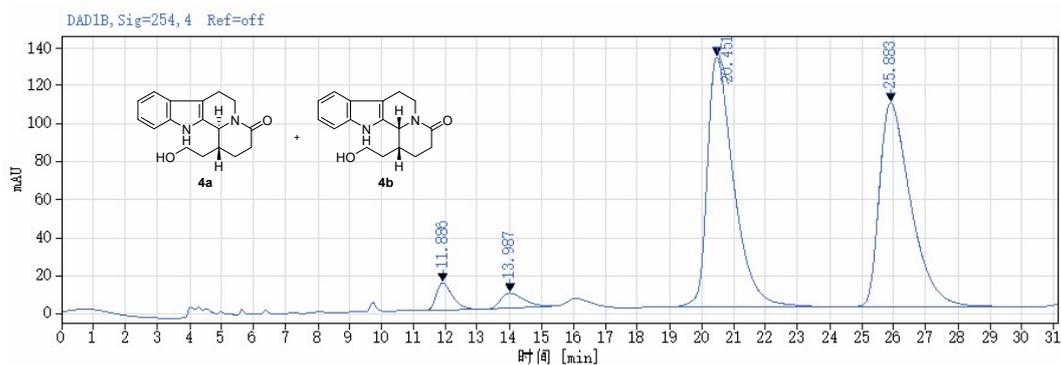
**Table 4. HPLC Spectrum of 4b**



RetTime [min]	Type	Width [min]	Area [mAU * s]	Height [mAU]	Area%
12.474	MM m	0.52	2699.01	77.42	48.49
13.841	MM m	0.62	2867.46	65.73	51.51
Totals			5566.47		

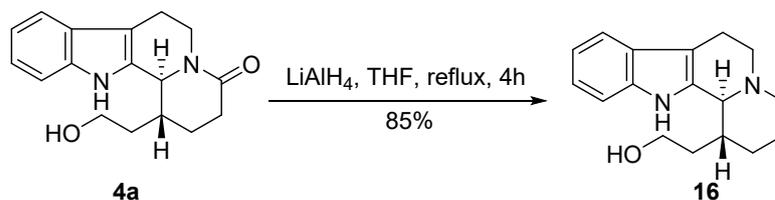
Chiral HPLC conditions: Daicel Chiralcel OD column, 4.6 × 250 mm, 85:15 *n*-hexane/*i*-PrOH, 0.75 mL/min, UV detector at 254 nm.  $t_{R1}$  (**4b**) = 12.4 min,  $t_{R2}$  (**4b**) = 13.8 min.

**Table 5. HPLC Spectrum of the products of acyl-Pictet-Spengler cascade reaction from 15 (entry 11)**

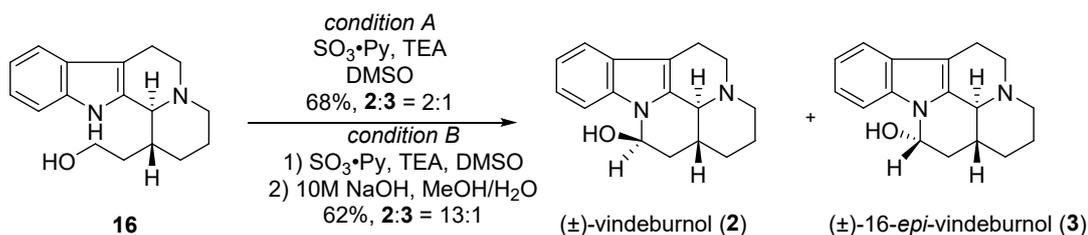


RetTime [min]	Type	Width [min]	Area [mAU * s]	Height [mAU]	Area%
11.886	MM m	1.89	505.83	14.42	3.23
13.987	MM m	2.15	393.80	7.82	2.51
20.451	MM m	4.94	7521.93	131.45	47.98
25.883	MM m	5.71	7526.991	10717	46.29
Totals			15678.55		

Chiral HPLC conditions: Daicel Chiralcel OD column, 4.6 × 250 mm, 85:15 *n*-hexane/*i*-PrOH, 0.75 mL/min, UV detector at 254 nm.  $t_{R1}$  (**4b**) = 11.9min,  $t_{R2}$  (**4b**) = 13.9 min,  $t_R$  (**4a**) = 20.4 min,  $t_R$  (**4a**) = 25.8 min.



To a stirred solution of compound **4a** (0.30 g, 1.05 mmol) in dry THF (20.0 mL) was added LiAlH<sub>4</sub> (0.20 g, 5.27 mmol). After the reaction mixture was refluxed for 4 h, the saturated Na<sub>2</sub>SO<sub>4</sub> solution (30 mL) was added and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 30 mL). Then the combined organic layers were concentrated in vacuo and the crude products were purified directly by column chromatography on silica gel [CH<sub>2</sub>Cl<sub>2</sub>/MeOH = 20:1, R<sub>f</sub> (**16**) = 0.2] to afford the desired product **16** (0.24 g, 85%) as a white solid. mp = 216.3 - 220.3 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 10.58 (s, 1H), 7.33 (q, *J* = 7.6 Hz, 2H), 7.00 (t, *J* = 7.2 Hz, 1H), 6.93 (t, *J* = 7.6 Hz, 1H), 4.51 (t, *J* = 4.8 Hz, 1H), 3.75 (d, *J* = 5.2 Hz, 1H), 3.60 – 3.47 (m, 2H), 3.20 – 3.13 (m, 1H), 2.88 – 2.83 (m, 1H), 2.81 – 2.72 (m, 2H), 2.63 – 2.57 (m, 1H), 2.54 – 2.50 (m, 1H), 2.25 – 2.19 (m, 1H), 1.99 – 1.88 (m, 1H), 1.68 – 1.60 (m, 1H), 1.56 – 1.43 (m, 3H), 1.37 – 1.28 (m, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 135.7, 134.9, 126.9, 120.2, 118.1, 117.3, 111.0, 106.5, 59.5, 58.9, 49.6, 49.2, 34.9, 31.8, 26.8, 20.6, 18.9. IR (KBr) 3188, 3059, 2920, 1454, 1361, 1321, 1255, 1107, 1056, 858, 736 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>17</sub>H<sub>23</sub>N<sub>2</sub>O [M + H]<sup>+</sup>: 271.1805, found: 271.1810.



#### Condition A

To a stirred solution of compound **16** (0.15 g, 0.55 mmol) in dry DMSO (5.50 mL) was added SO<sub>3</sub>·Py (0.35 g, 2.20 mmol) and triethylamine (0.60 mL). After at room temperature for 4h, the mixture was diluted by CH<sub>2</sub>Cl<sub>2</sub> (20 mL), and washed with H<sub>2</sub>O (3 × 10 mL). The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 30 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated in vacuo to give a brown solid as crude product. The residue was purified by column chromatography [CH<sub>2</sub>Cl<sub>2</sub>/Acetone/MeOH = 25:4:1, R<sub>f</sub> (**2**) = 0.3, R<sub>f</sub> (**3**) = 0.5] to afford (±)-vindeburnol (**2**) (67.1 mg, 45%) as a brown solid, (±)-16-*epi*-vindeburnol (**3**) (34.2 mg, 23%) as a brown solid.

(±)-vindeburnol (**2**): mp = 178.9–180.9 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 7.44 (d, *J* = 8 Hz, 1H), 7.36 (d, *J* = 7.6 Hz, 1H), 7.06 (t, *J* = 8 Hz, 1H), 6.99 (t, *J* = 7.6 Hz, 1H), 6.17 (d, *J* = 6.8 Hz, 1H), 5.91 (d, *J* = 6.8 Hz, 1H), 3.05 (dd, *J* = 11.2, 6.0 Hz, 1H), 2.96 (d, *J* = 10.8 Hz, 1H), 2.81 – 2.72 (m, 1H), 2.66 – 2.61 (m, 2H), 2.41 (td, *J* = 11.2, 4.4 Hz, 1H), 2.16 (td, *J* = 11.2, 2.8 Hz, 1H), 1.98 – 1.92 (m, 3H), 1.76 – 1.62 (m, 3H), 1.19 – 1.10 (m, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 136.0, 134.3, 128.2, 120.1, 119.2, 117.6, 111.3, 103.7, 73.8, 63.7, 54.6, 52.6, 37.4, 30.8, 29.5, 25.4, 21.2. IR (KBr) 2924, 2852, 1627, 1444, 1377, 1263, 1201, 1176, 1105, 1062, 902, 875, 854, 856 cm<sup>-1</sup>. HRMS (ESI) *m/z* calcd for C<sub>17</sub>H<sub>21</sub>N<sub>2</sub>O [M + H]<sup>+</sup>: 269.1648, found: 269.1652.

(±)-16-*epi*-vindeburnol (**3**): mp = 170.1–171.9 °C; <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 7.65 (d, *J* = 8 Hz, 1H), 7.37 (d, *J* = 8 Hz, 1H), 7.07 – 6.99 (m, 2H), 6.46 (d, *J* = 9.2 Hz, 1H), 5.54 – 5.48 (m, 1H), 3.02 (dd, *J* = 11.6,

6.0 Hz, 1H), 2.93 (d,  $J=10.4$  Hz, 1H), 2.78 – 2.61 (m, 3H), 2.44 (dd,  $J = 11.6, 4.8$  Hz, 1H), 2.25 – 2.17 (m, 2H), 1.77 – 1.58 (m, 4H), 1.49 – 1.40 (m, 1H), 1.18 (dd,  $J = 12.0, 4$  Hz, 1H).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta$  137.4, 135.4, 127.9, 120.4, 119.3, 117.6, 112.0, 104.0, 77.9, 63.2, 54.3, 52.0, 38.8, 36.0, 29.6, 25.2, 21.3. IR (KBr) 2924, 2854, 2808, 1732, 1458, 1321, 1258, 1186, 1112, 1082, 966, 740  $\text{cm}^{-1}$ . HRMS (ESI)  $m/z$  calcd for  $\text{C}_{17}\text{H}_{21}\text{N}_2\text{O}$   $[\text{M} + \text{H}]^+$ : 269.1648, found: 269.1654.

### Condition B

The residue from the Parikh-Doering oxidation, which was directly used in next step without further purification, was added to the solution of 10M NaOH (MeOH/ $\text{H}_2\text{O}$  = 1:1) for 36 h at room temperature. The mixture was diluted by  $\text{CH}_2\text{Cl}_2$ , and washed with  $\text{H}_2\text{O}$ . The aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers were washed with brine, dried over  $\text{Na}_2\text{SO}_4$ , filtered, and evaporated in vacuo to give a brown solid as crude product. The crude products were purified directly by flash column chromatography to afford the mixture solid [( $\pm$ )-vindeburnol (**2**): ( $\pm$ )-16-*epi*-vindeburnol (**3**) = 13:1, according to  $^1\text{H}$  NMR].

$^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.65 (d,  $J = 8$  Hz, 0.09H), 7.44 (d,  $J = 8$  Hz, 1.01H), 7.36 (d,  $J = 7.2$  Hz, 1.07H), 7.07 – 6.97 (m, 2.35H), 6.48 (d,  $J = 9.2$  Hz, 0.08H), 6.18 (d,  $J = 6.8$  Hz, 1.00H), 5.91 (d,  $J = 6.4$  Hz, 1.01H), 5.54 – 5.48 (m, 0.09H), 3.06 (dd,  $J = 11.2, 6$  Hz, 1.03H), 2.97 – 2.93 (m, 1.06H), 2.81 – 2.72 (m, 1.16H), 2.66 – 2.61 (m, 2.09H), 2.42 (qd,  $J = 11.2, 4.8$  Hz, 1.01H), 2.22 – 2.13 (m, 1.11H), 1.95 – 1.88 (m, 2.91H), 1.76 – 1.65 (m, 3.78H), 1.16 – 1.11 (m, 1.07H).  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ )  $\delta$  137.3, 136.0, 135.7, 134.2, 128.1, 127.9, 120.4, 120.0, 119.3, 119.1, 117.6, 117.3, 111.9, 111.3, 103.9, 103.6, 77.8, 73.8, 63.6, 63.1, 54.5, 54.2, 52.6, 52.0, 37.3, 35.9, 31.2, 30.8, 29.6, 29.4, 25.3, 25.2, 21.3, 21.1.

**Table 6.** Comparison of  $^1\text{H}$  NMR data of ( $\pm$ )-vindeburnol (**2**) with literature

Lit. <sup>2</sup> data (400 MHz, DMSO- $d$ ) $\delta$ /ppm	Our data (400 MHz, DMSO- $d$ ) $\delta$ /ppm	Error $\Delta$ $\delta$ /ppm
7.43 (dt, $J = 7.8, 0.8$ Hz, 1H)	7.44 (d, $J = 8$ Hz, 1H)	0.01
7.35 (d, $J = 7.5$ Hz, 1H)	7.36 (q, $J = 7.6$ Hz, 1H)	0.01
7.05 (br t, $J = 7.6$ Hz, 1H)	7.06 (t, $J = 8$ Hz, 1H)	0.01
7.00 (br t, $J = 7.4$ Hz, 1H)	6.99 (t, $J = 7.6$ Hz, 1H)	-0.01
6.14 (d, $J = 6.9$ Hz, 1H)	6.17 (d, $J = 6.8$ Hz, 1H)	0.03
5.9 (m, 1H)	5.91 (d, $J = 6.8$ Hz, 1H)	0
3.04 (dd, $J = 11.3, 5.8$ Hz, 1H)	3.05 (dd, $J = 11.2, 6.0$ Hz, 1H)	0.01
2.95 (m, 1H)	2.96 (d, $J = 10.8$ Hz, 1H)	0
2.77 (m, 1H)	2.78 – 2.72 (m, 1H)	0
2.64 (m, 2H)	2.66 – 2.61 (m, 2H)	0
2.41 (td, $J = 11.4, 4.4$ Hz, 1H)	2.41 (td, $J = 11.2, 4.4$ Hz, 1H)	0
2.16 (td, $J = 11.1, 2.6$ Hz, 1H)	2.16 (td, $J = 11.2, 2.8$ Hz, 1H)	0
1.98 – 1.89 (m, 3H)	1.98 – 1.92 (m, 3H)	0
1.78 – 1.61 (m, 3H)	1.76 – 1.62 (m, 3H)	0
1.15 (m, 1H)	1.19 – 1.10 (m, 1H)	0

**Table 7.** Comparison of  $^{13}\text{C}$  NMR data of ( $\pm$ )-vindeburnol (**2**) with literature

Lit. <sup>2</sup> data (125 MHz, DMSO- $d$ ) $\delta$ /ppm	Our data (100 MHz, DMSO- $d$ ) $\delta$ /ppm	Error $\Delta$ $\delta$ /ppm
136.0	136.0	0
134.3	134.3	0
128.2	128.2	0
120.0	120.1	0.1
119.2	119.2	0

117.6	117.6	0
111.3	111.3	0
103.7	103.7	0
73.8	73.8	0
63.7	63.7	0
54.6	54.6	0
52.6	52.6	0
37.4	37.4	0
30.8	30.8	0
29.5	29.5	0
25.4	25.4	0
21.2	21.2	0

**Table 8.** Comparison of  $^1\text{H}$  NMR data of ( $\pm$ )-16-epi-vindeburnol (**3**) with literature

Lit. <sup>2</sup> data (400 MHz, DMSO- <i>d</i> ) $\delta$ /ppm	Our data (400 MHz, DMSO- <i>d</i> ) $\delta$ /ppm	Error $\Delta$ $\delta$ /ppm
7.65 (m, 1H)	7.65 (d, $J = 8$ Hz, 1H)	0
7.37 (m, 1H)	7.37 (q, $J = 8$ Hz, 1H)	0
7.03 (m, 2H)	7.07 – 6.99 (m, 2H)	0
6.44 (d, $J = 9.0$ Hz, 1H)	6.46 (d, $J = 9.2$ Hz, 1H)	0.02
5.5 (td, $J = 9.0, 5.7$ Hz, 1H)	5.54 – 5.48 (m, 1H)	0
3.02 (dd, $J = 11.2, 5.6$ Hz, 1H)	3.02 (dd, $J = 11.6, 6.0$ Hz, 1H)	0
2.92 (dt, $J = 11.0, 2.7$ Hz, 1H)	2.93 (d, $J = 10.4$ Hz, 1H)	0.01
2.79 – 2.61 (m, 3H)	2.78 – 2.61 (m, 3H)	0
2.45 (td, $J = 11.2, 4.4$ Hz, 1H)	2.44 (dd, $J = 11.2, 4.4$ Hz, 1H)	-0.01
2.20 (m, 2H)	2.25 – 2.17 (m, 2H)	0
1.77 – 1.55 (m, 4H)	1.77 – 1.58 (m, 4H)	0
1.44 (m, 1H)	1.49 – 1.40 (m, 1H)	0
1.17 (m, 1H)	1.17 (dd, $J = 12, 4$ Hz, 1H)	0

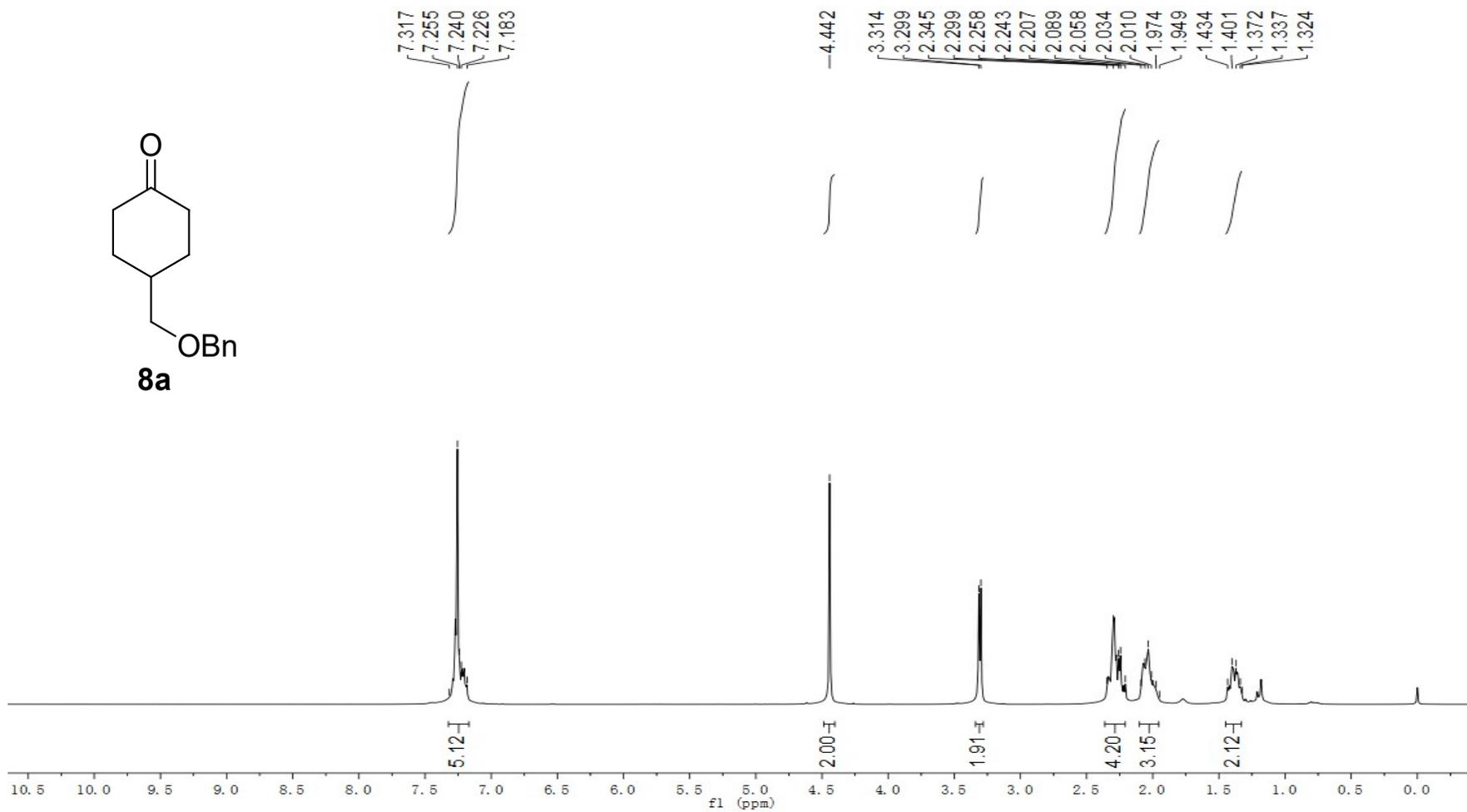
**Table 9.** Comparison of  $^{13}\text{C}$  NMR data of ( $\pm$ )-16-epi-vindeburnol (**3**) with literature

Lit. <sup>2</sup> data (125 MHz, DMSO- <i>d</i> ) $\delta$ /ppm	Our data (100 MHz, DMSO- <i>d</i> ) $\delta$ /ppm	Error $\Delta$ $\delta$ /ppm
137.4	137.4	0
135.4	135.4	0
127.9	127.9	0
120.3	120.4	0.1
119.3	119.3	0
117.6	117.6	0
111.9	112.0	0.1
103.9	104.0	0.1
77.8	77.9	0.1
63.2	63.2	0
54.3	54.3	0
52.0	52.0	0
38.8	38.8	0
36.0	36.0	0
29.6	29.6	0
25.2	25.2	0
21.3	21.3	0

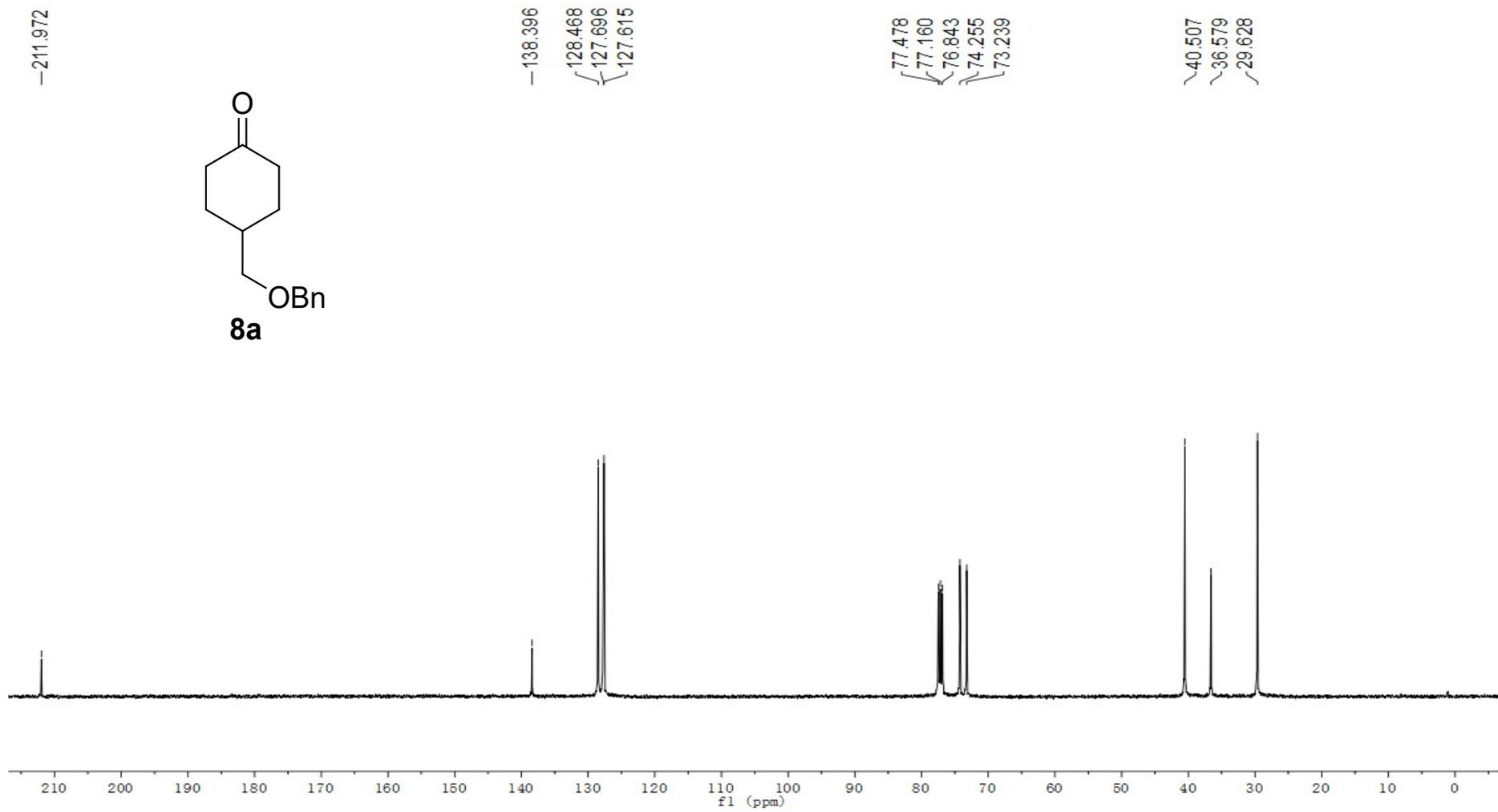
### 3. References

- 1 T. Yamashita, N. Kawai, H. Tokuyama and T. Fukuyama, *J. Am. Chem. Soc.*, 2005, **127**, 15038.
- 2 L. Salacz, C. Charpentier and N. Girard, *J. Org. Chem.* 2017, **82**, 2257.

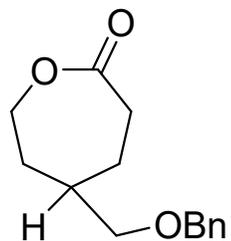
#### 4. NMR Spectra



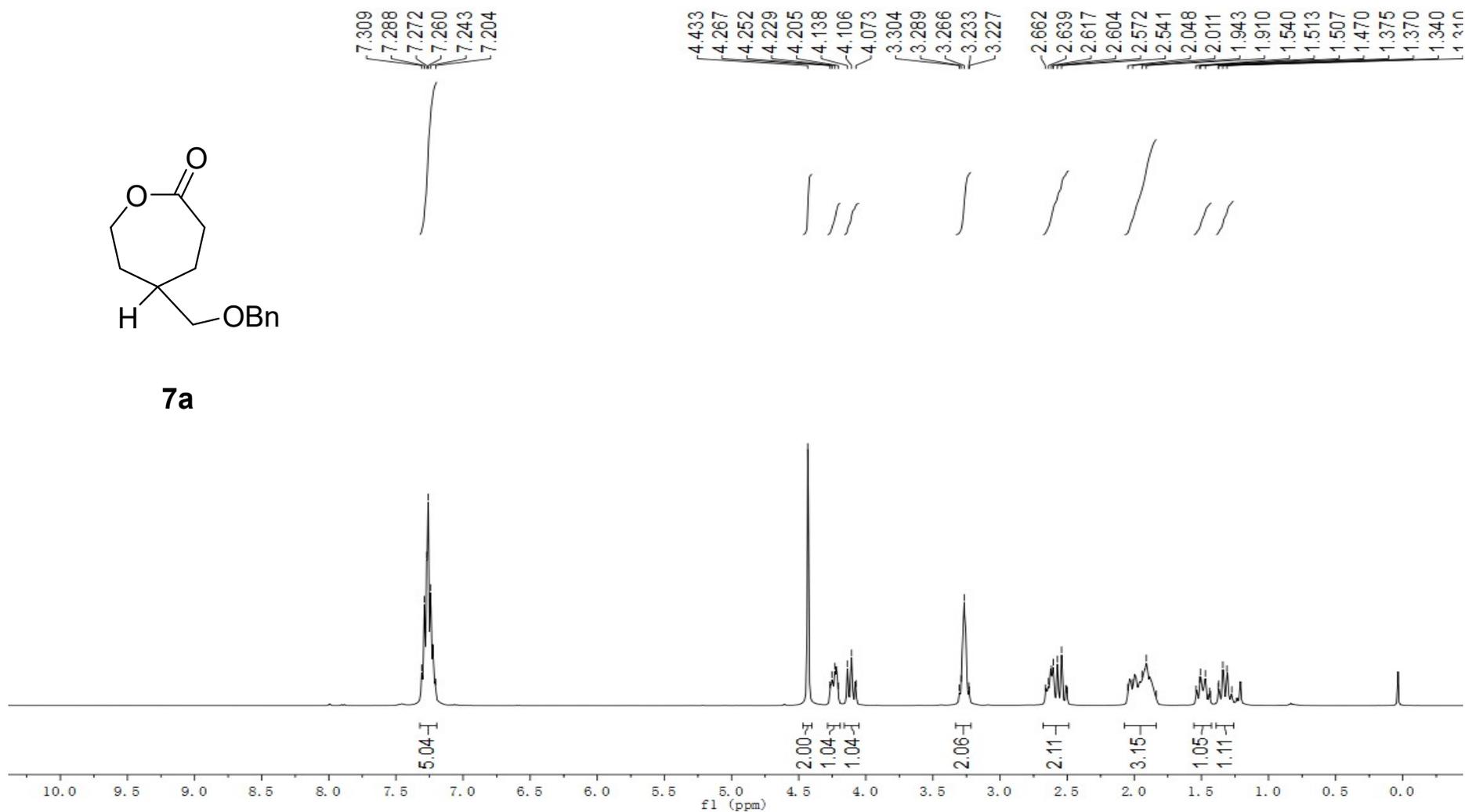
**<sup>1</sup>H-NMR Spectrum of 8a**



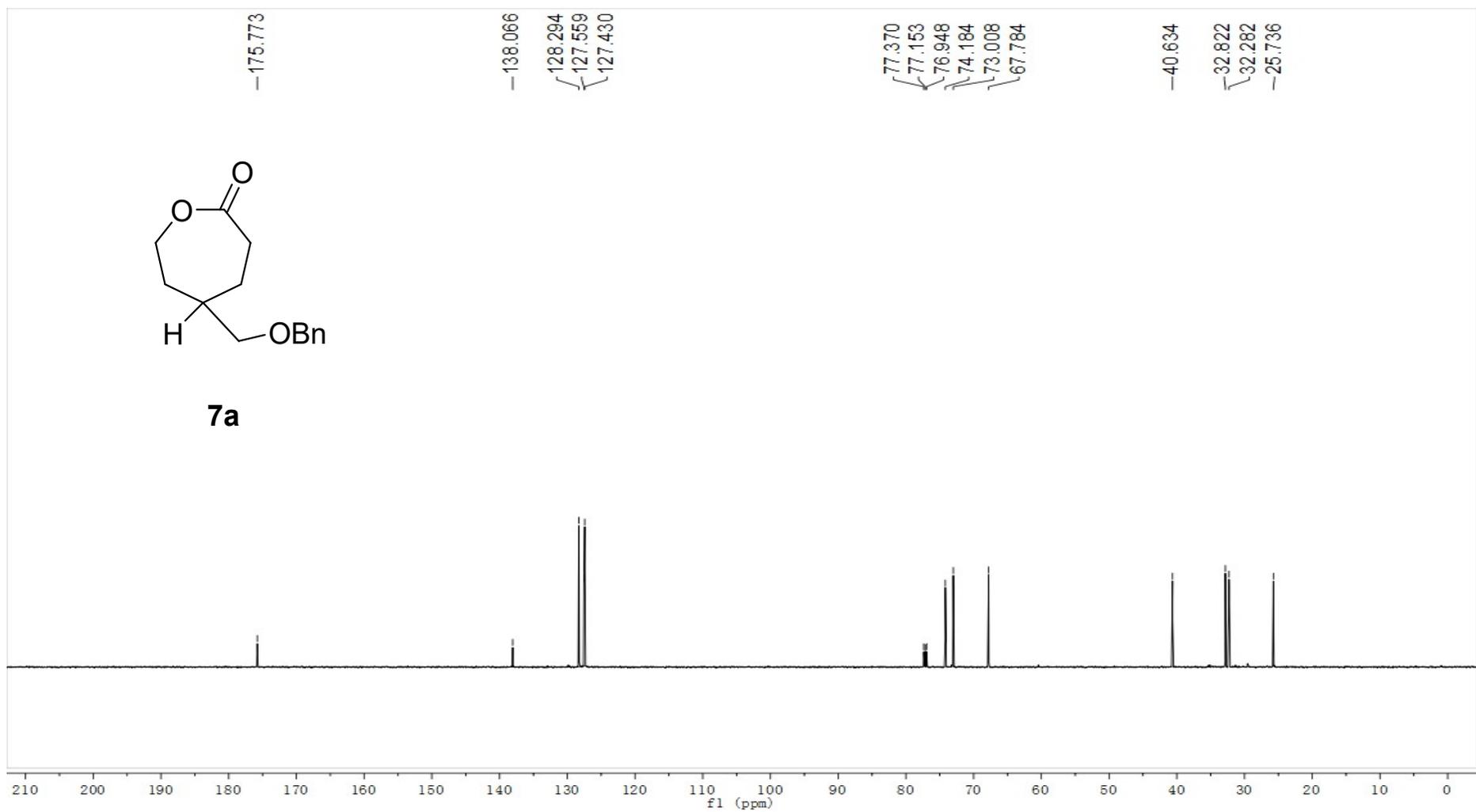
<sup>13</sup>C-NMR Spectrum of **8a**



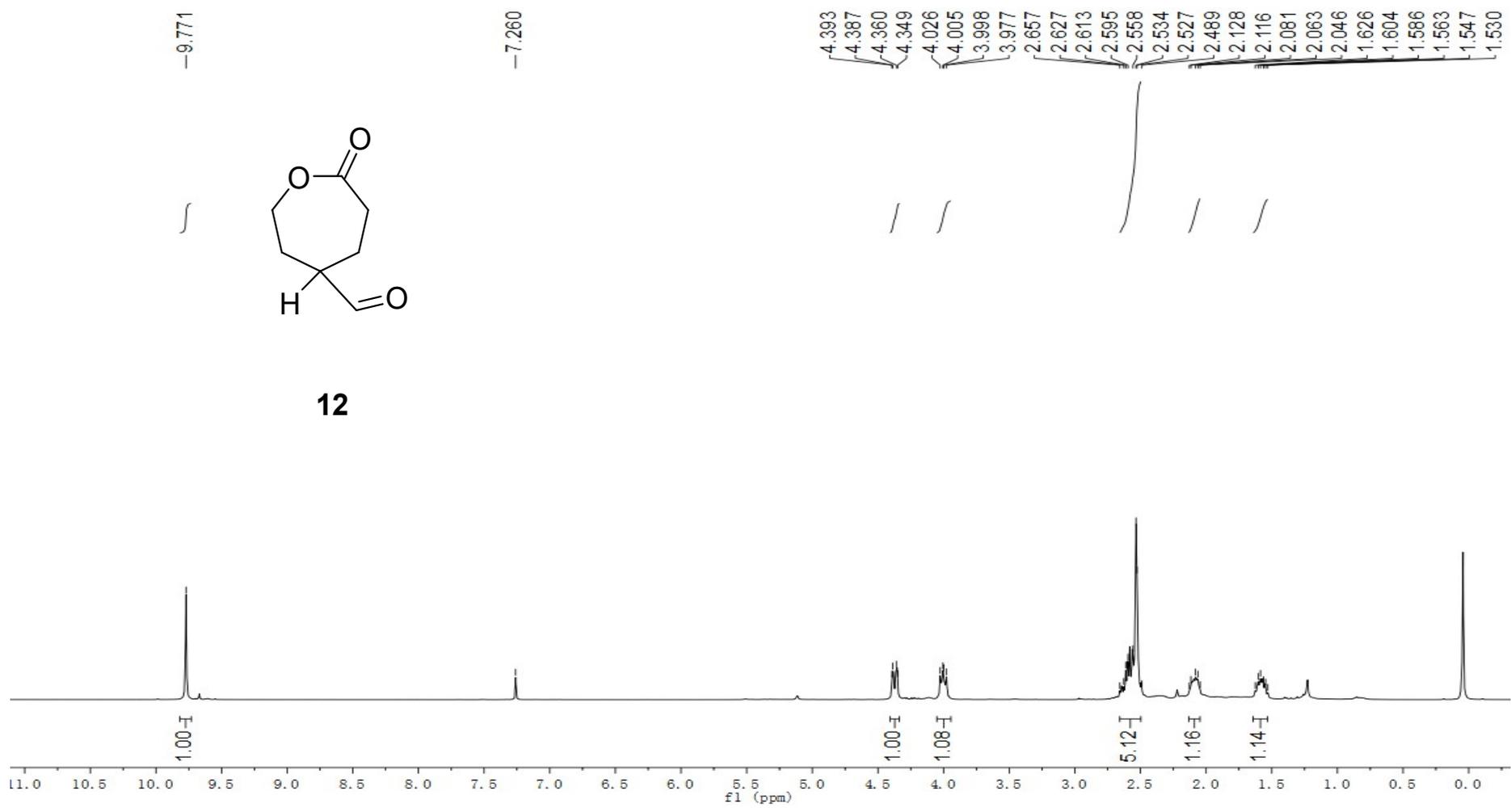
**7a**



**<sup>1</sup>H-NMR Spectrum of 7a**



<sup>13</sup>C-NMR Spectrum of **7a**



**<sup>1</sup>H-NMR Spectrum of 12**

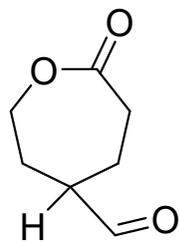
-199.391

-170.943

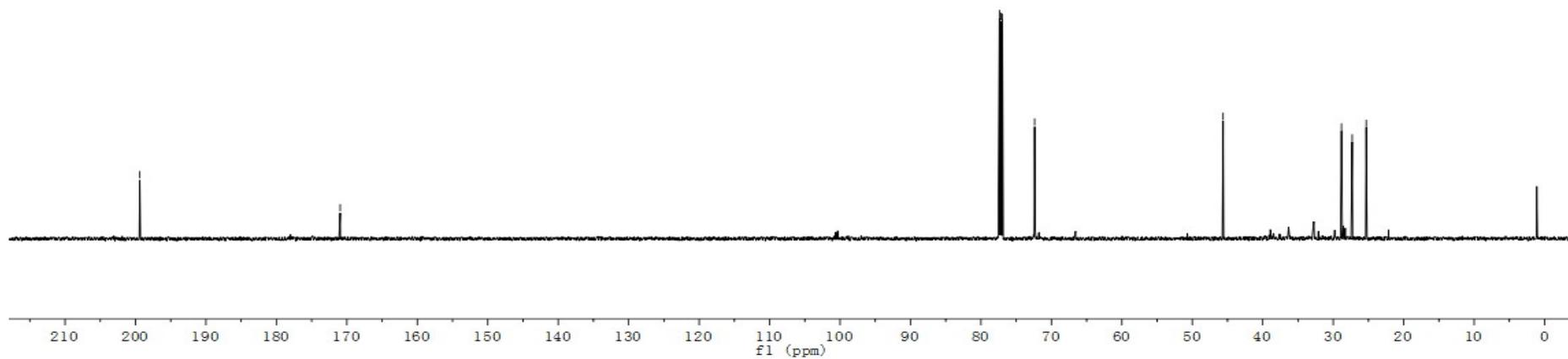
77.372  
77.159  
76.947  
72.378

-45.624

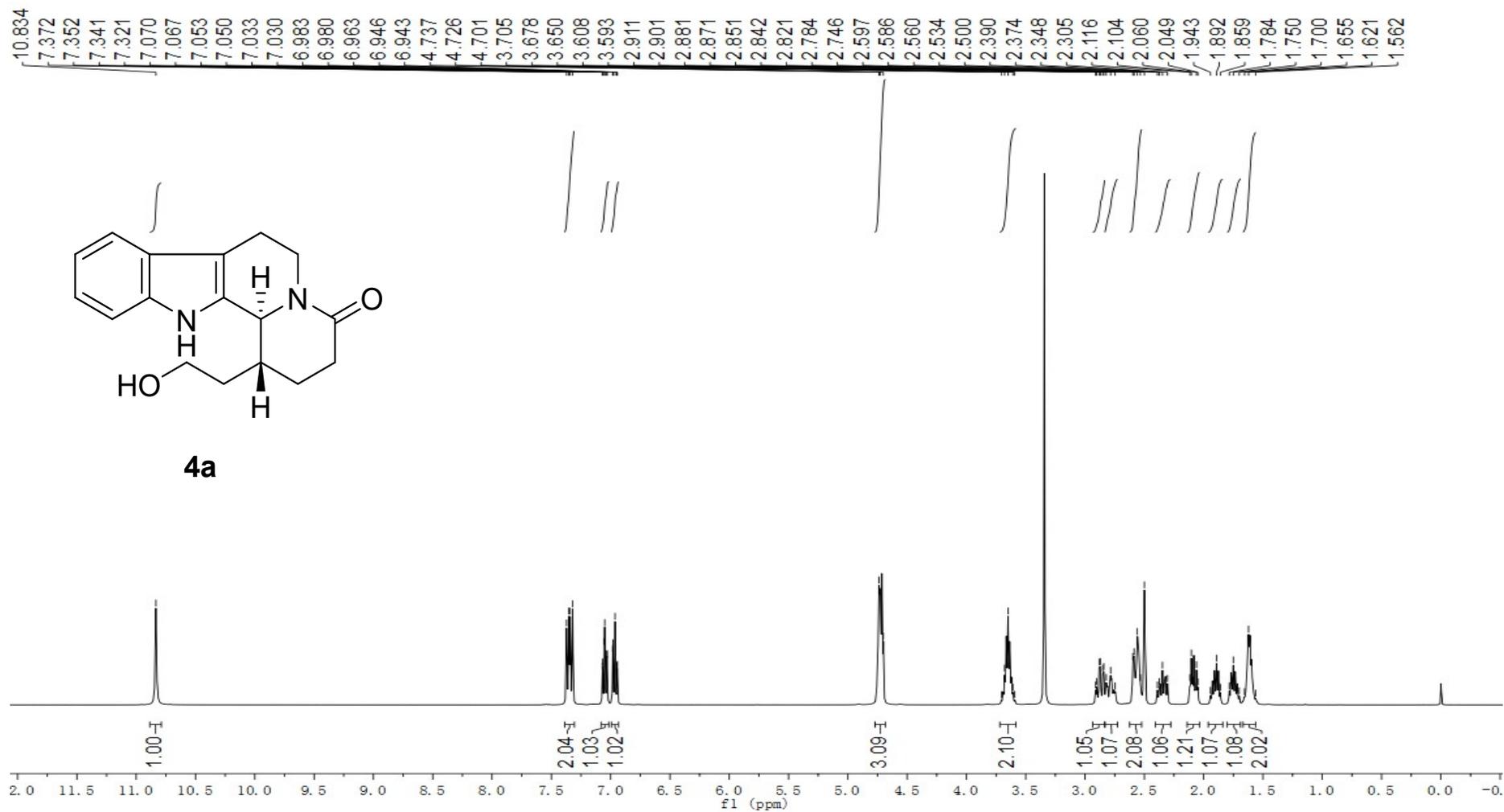
28.828  
27.306  
25.316

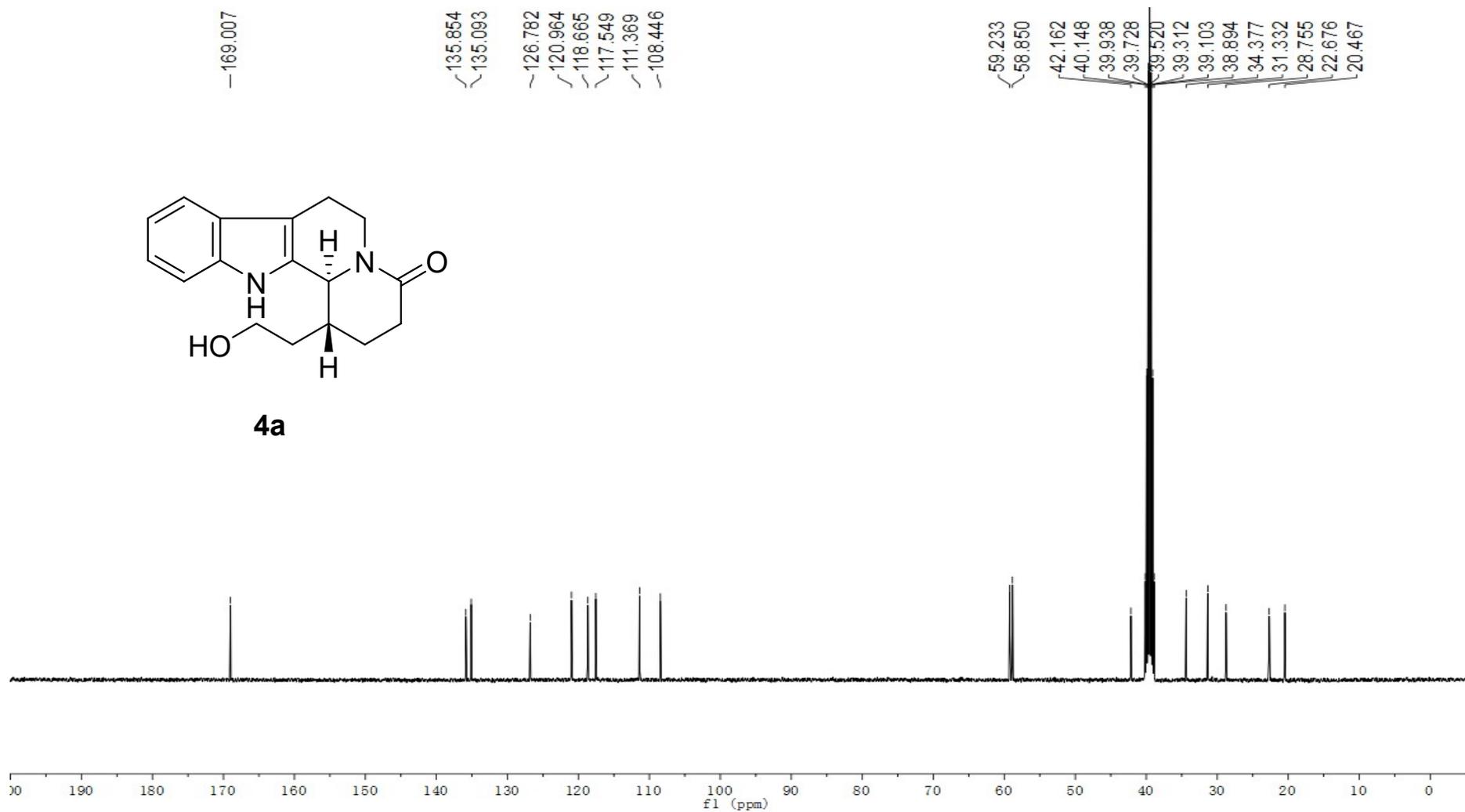


12

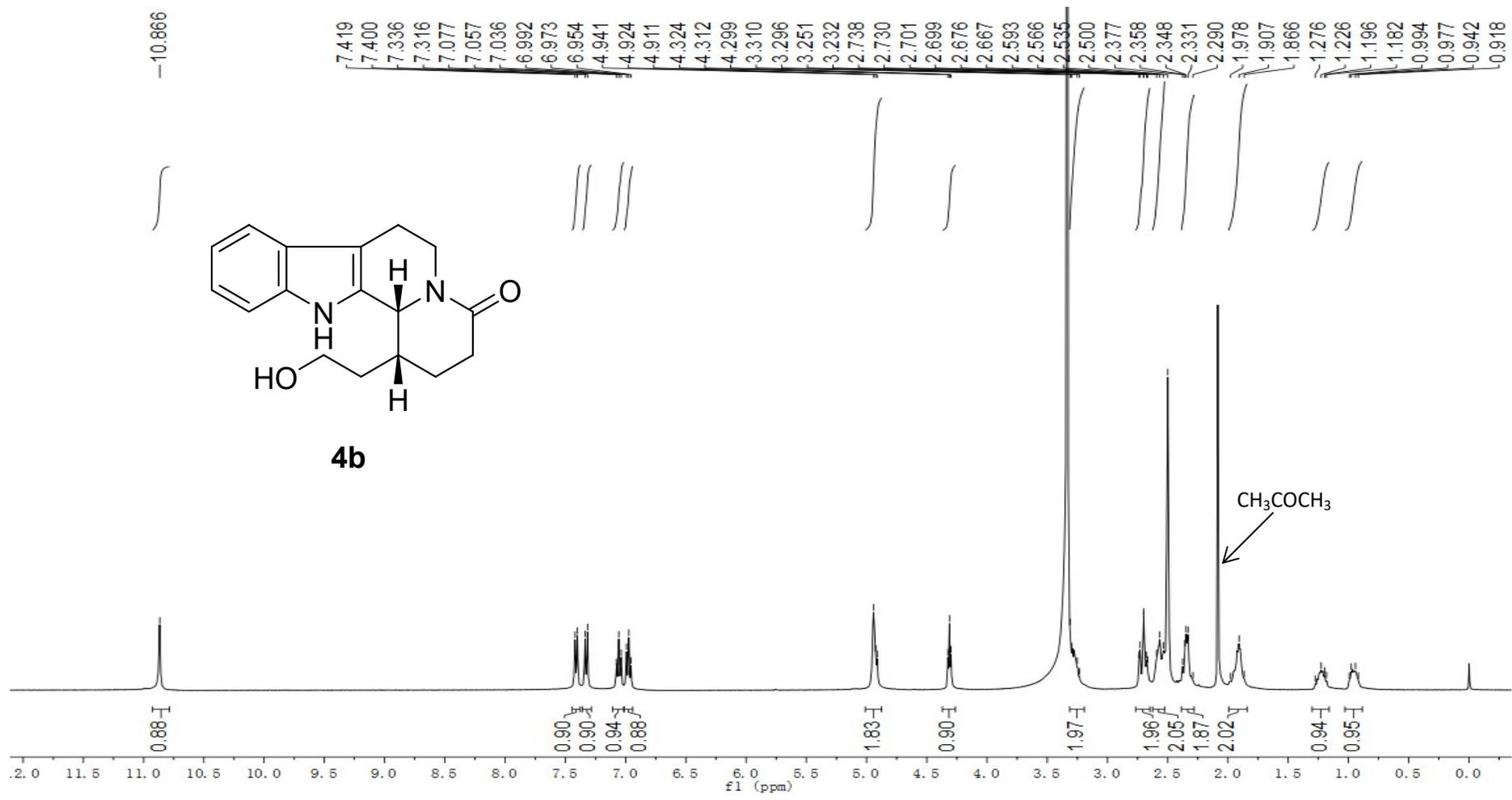


<sup>13</sup>C-NMR Spectrum of 12

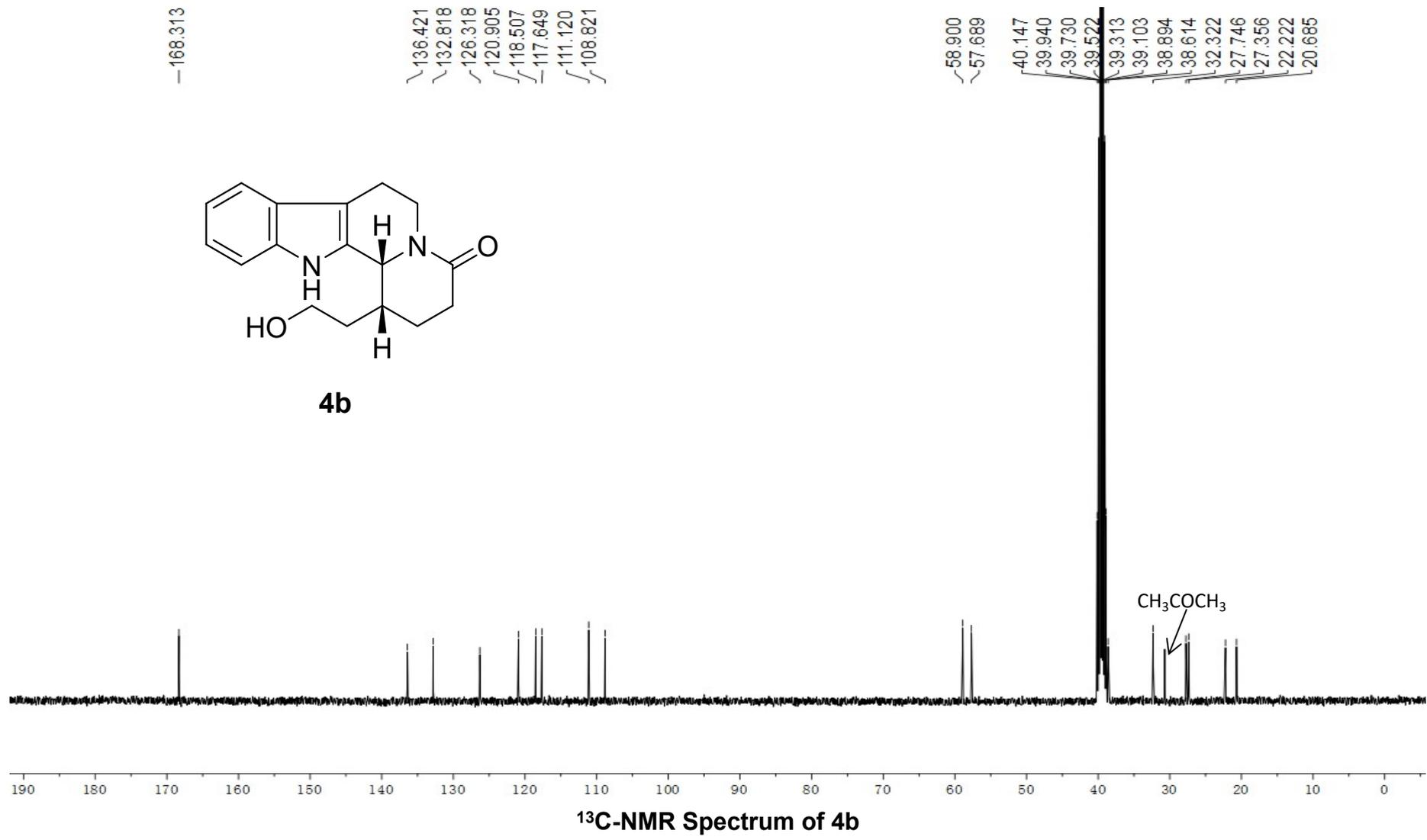


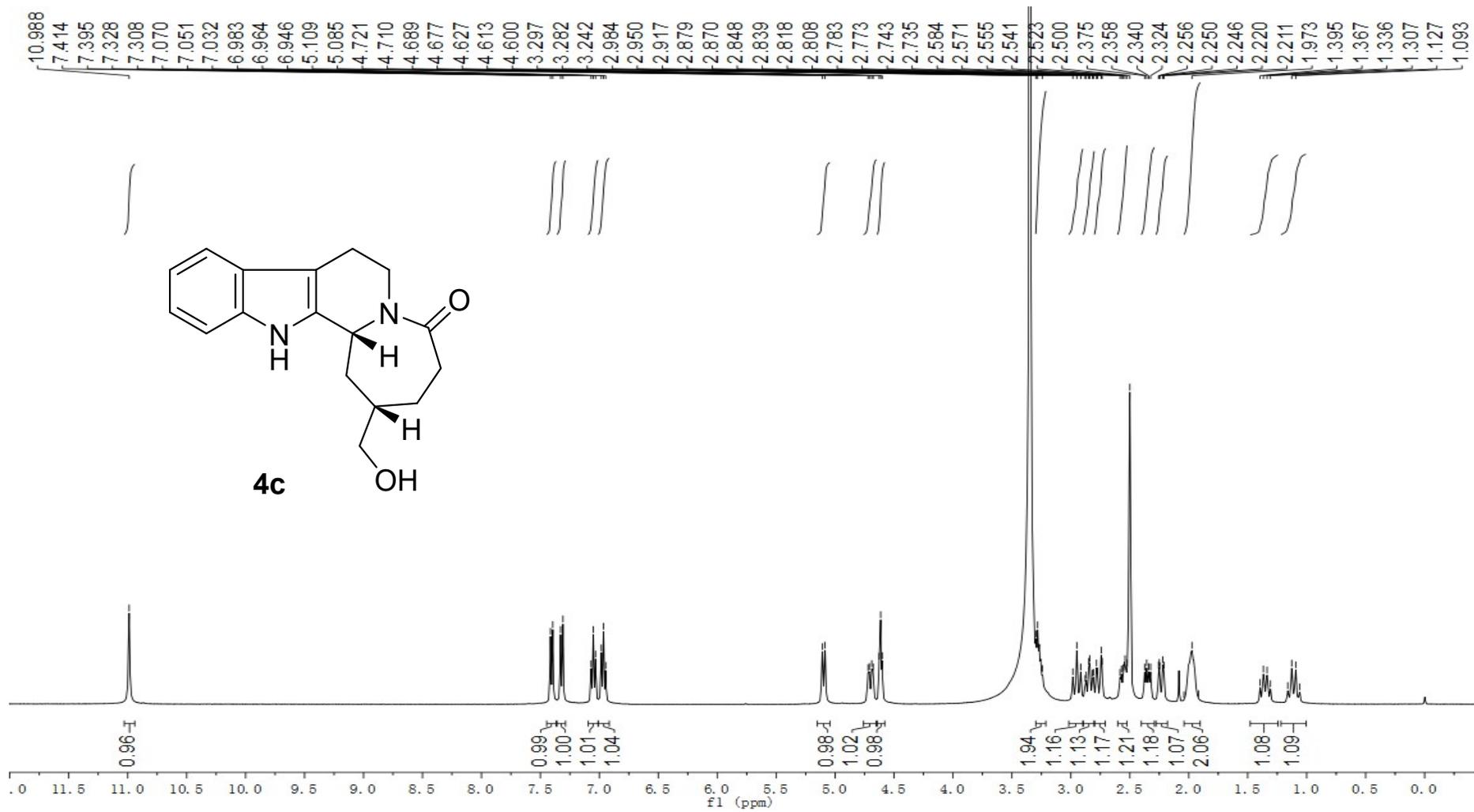


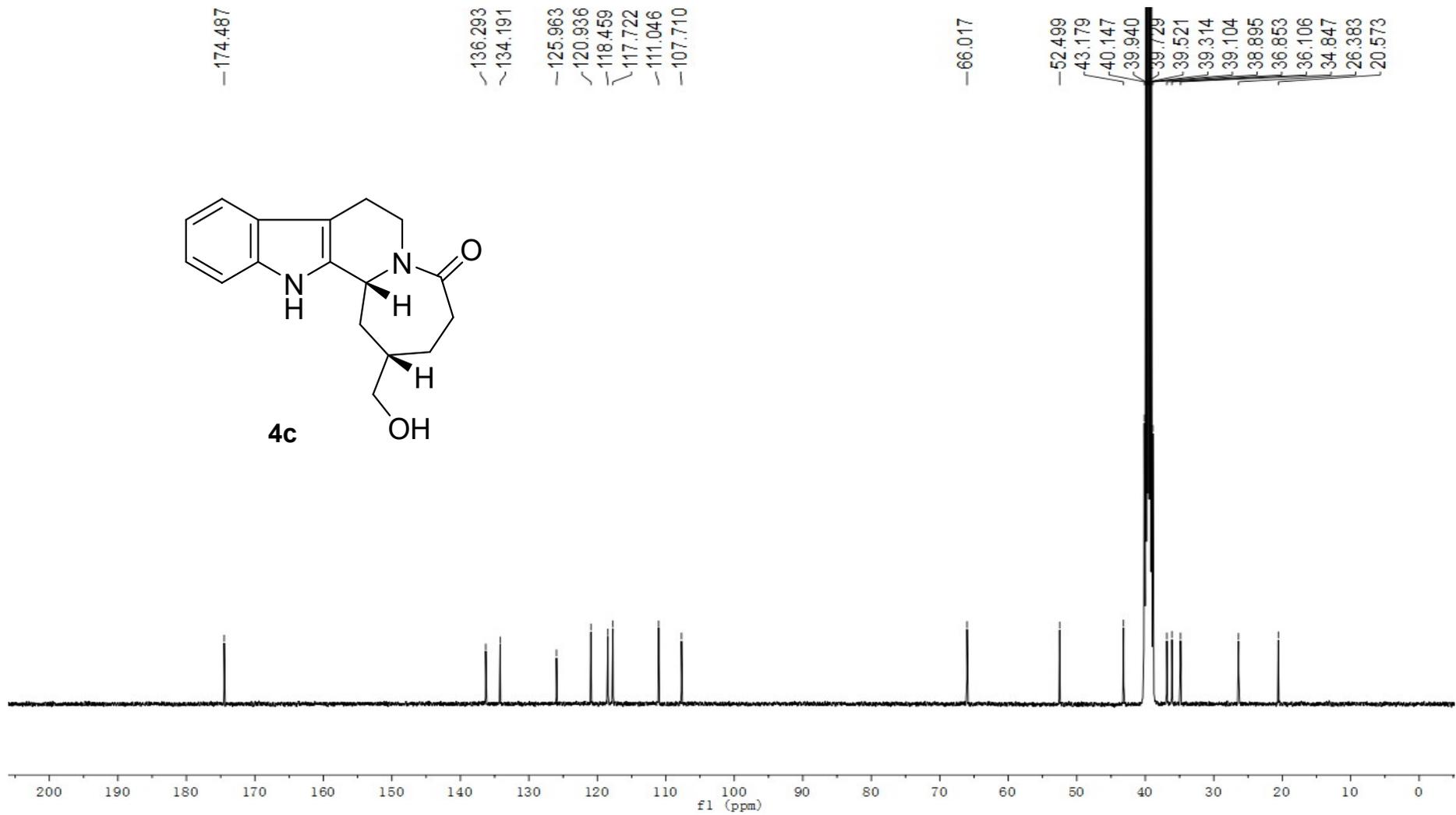
**<sup>13</sup>C-NMR Spectrum of 4a**



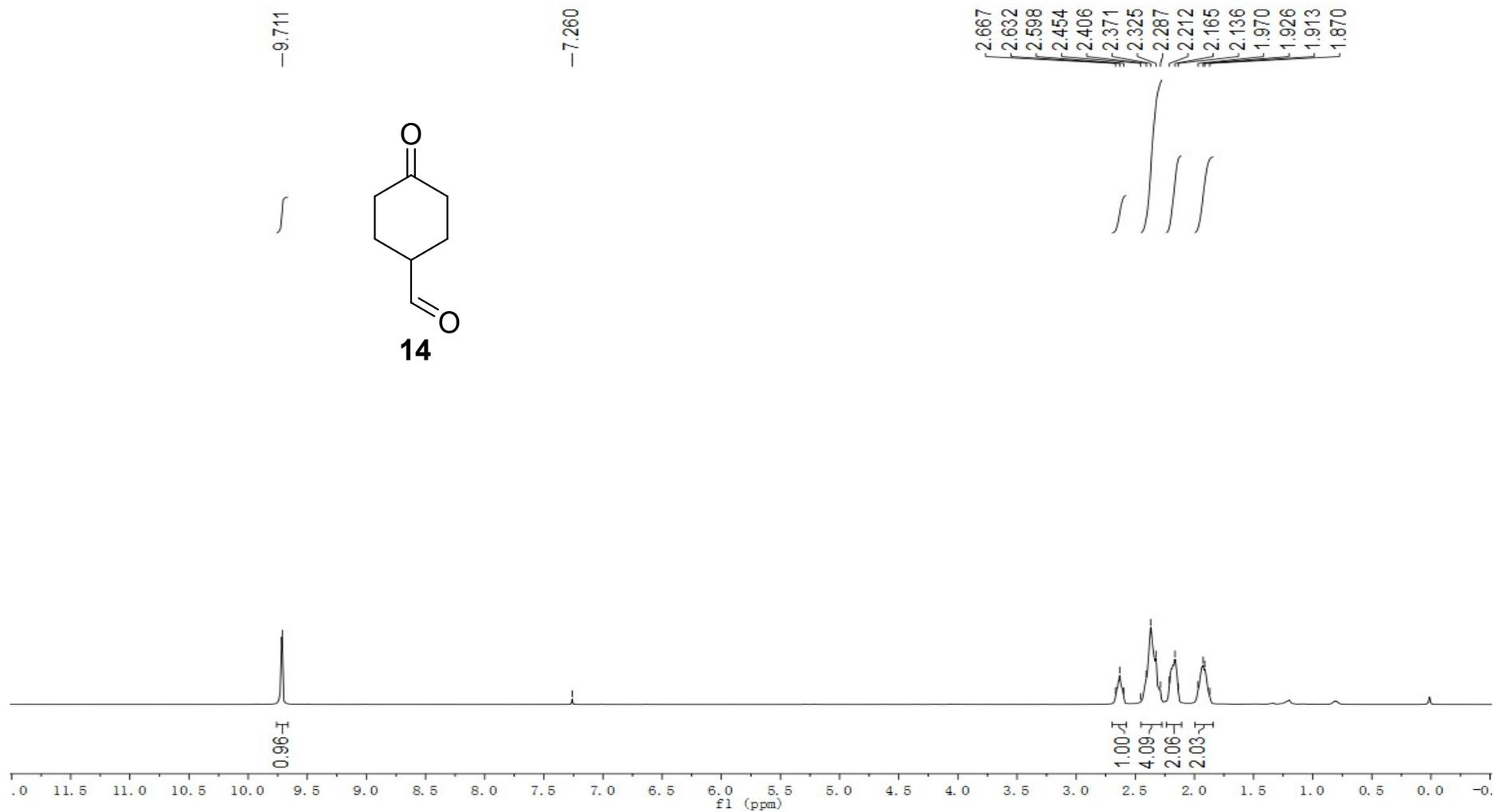
**<sup>1</sup>H-NMR Spectrum of 4b**





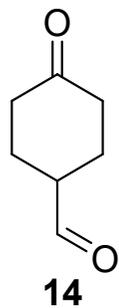


**<sup>13</sup>C-NMR Spectrum of 4c**



**<sup>1</sup>H-NMR Spectrum of 14**

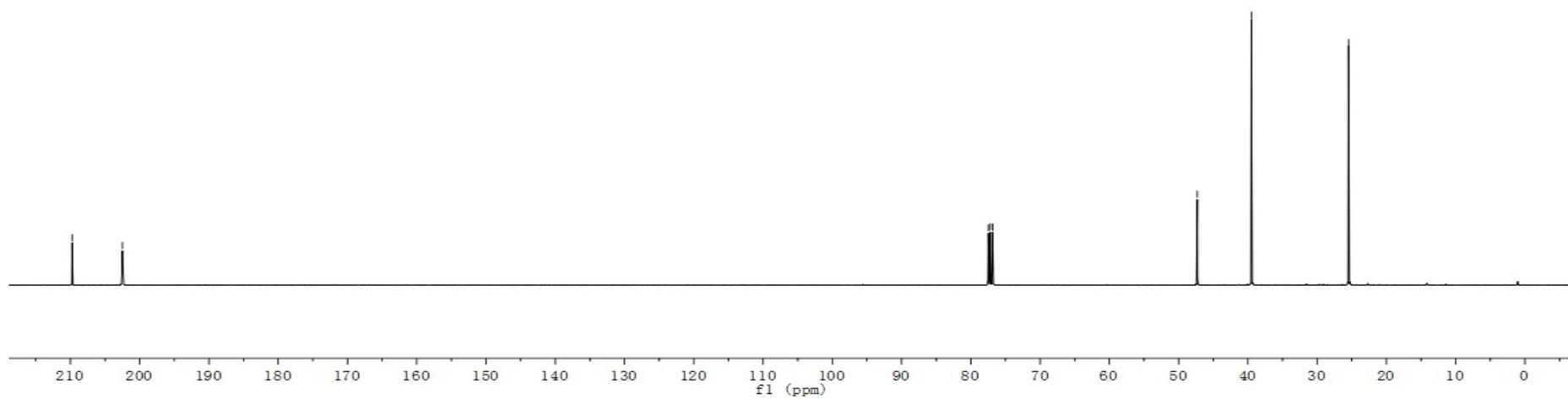
-209.749  
-202.492



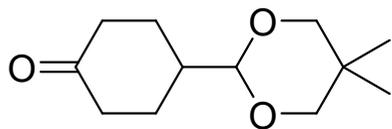
77.479  
77.160  
76.844

-47.305  
-39.490

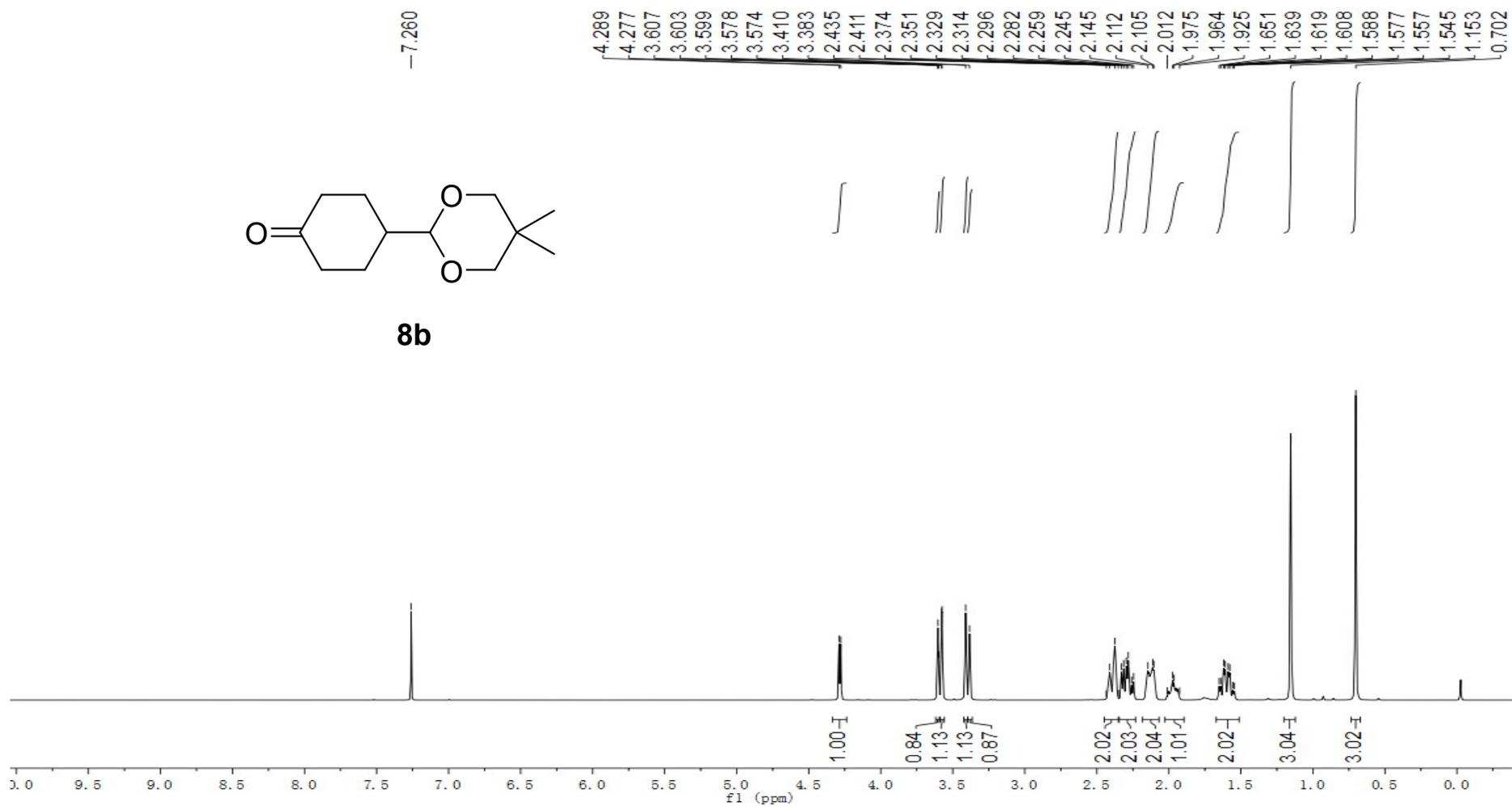
-25.474



<sup>13</sup>C-NMR Spectrum of 14

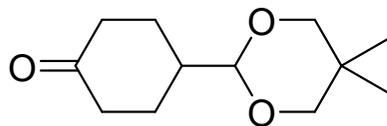


**8b**



**<sup>1</sup>H-NMR Spectrum of 8b**

-211.933



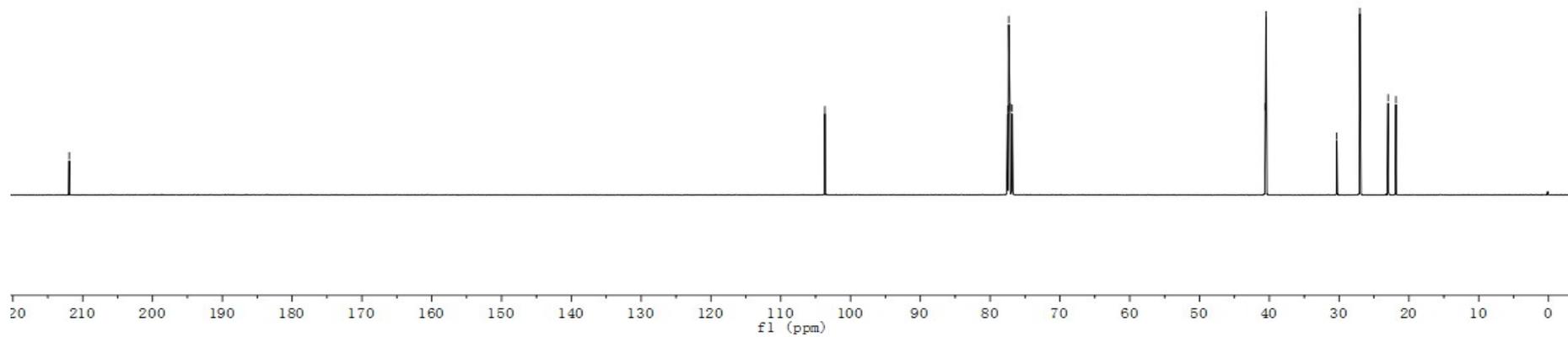
**8b**

-103.671

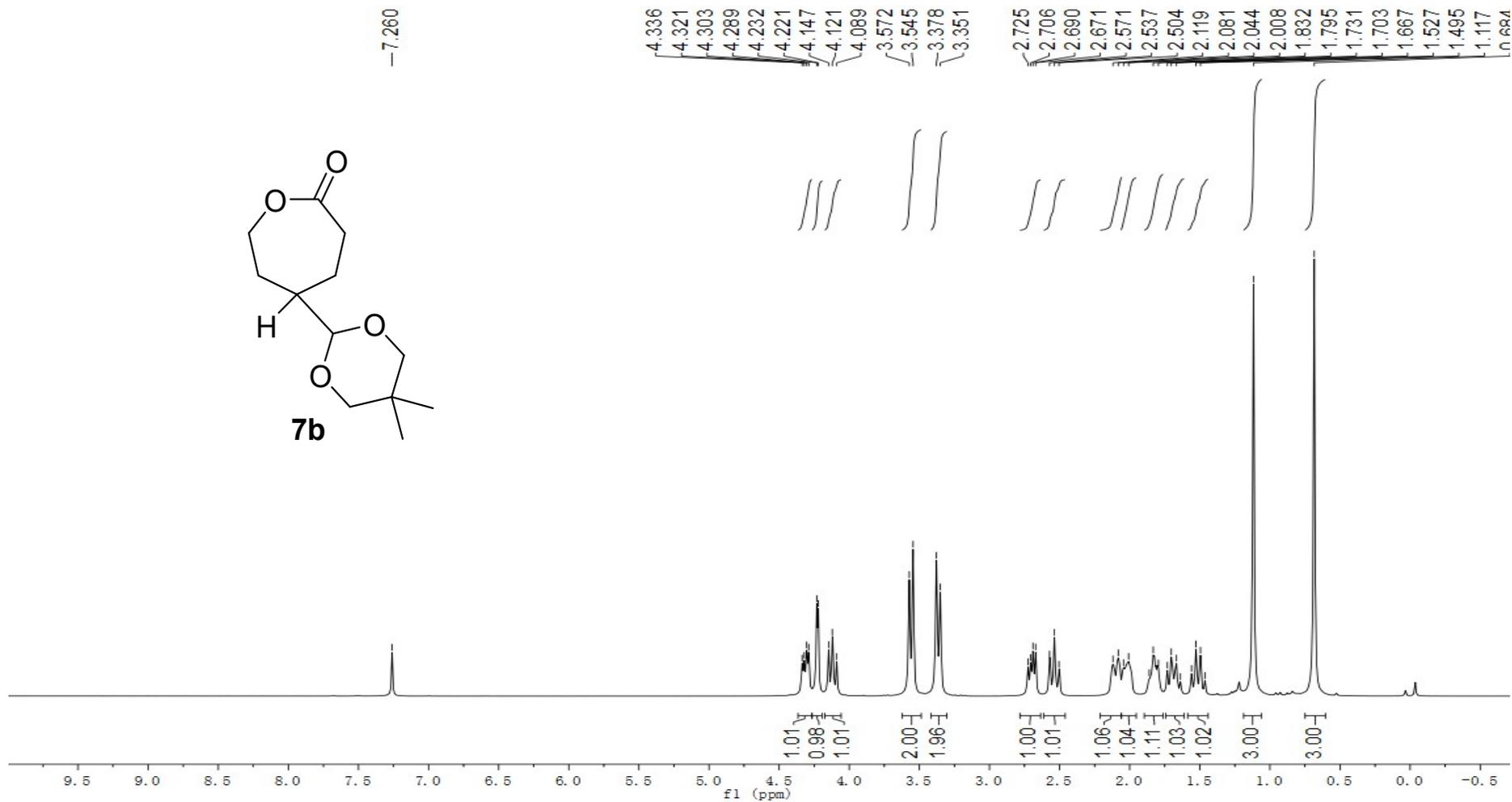
77.478  
77.328  
77.160  
76.842

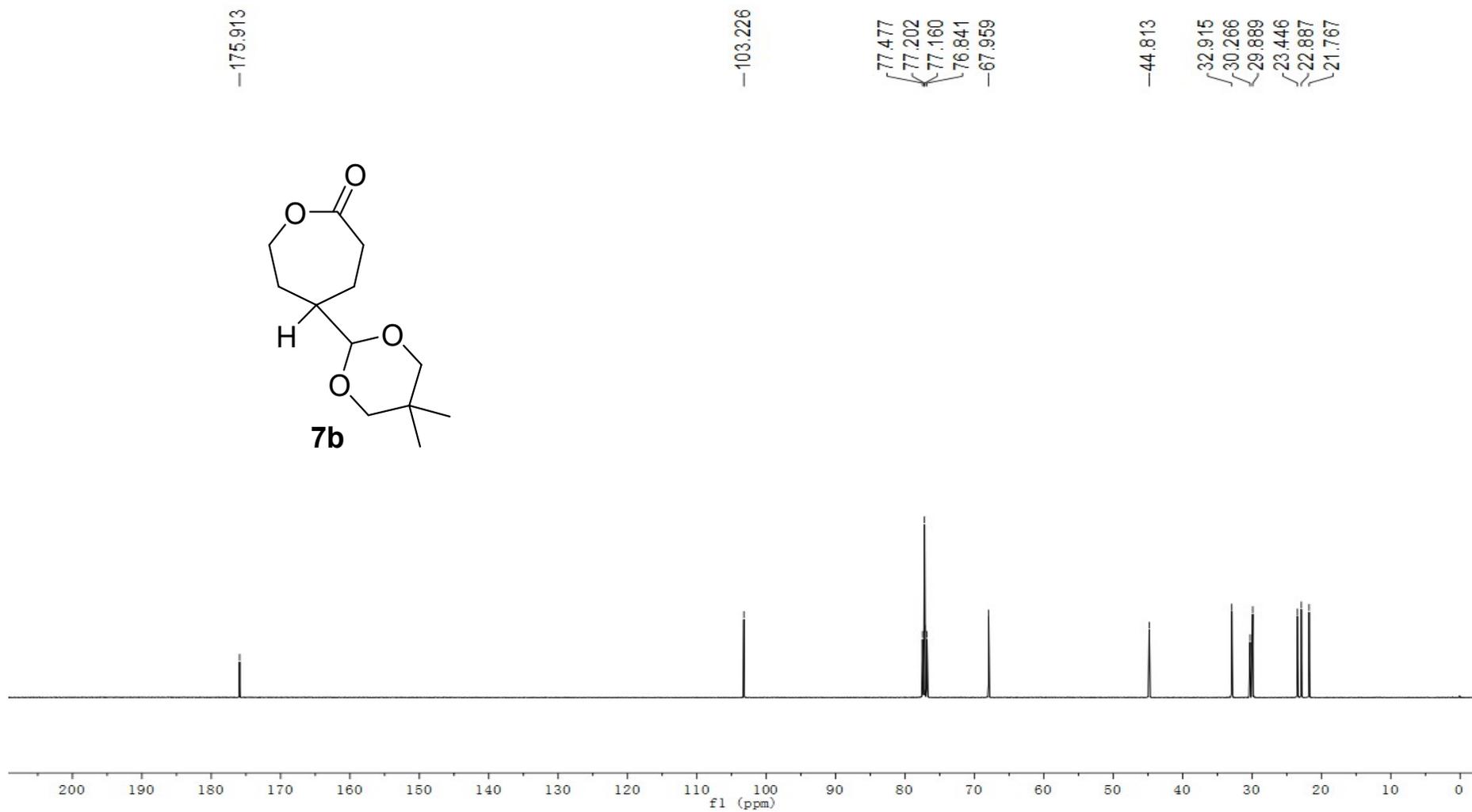
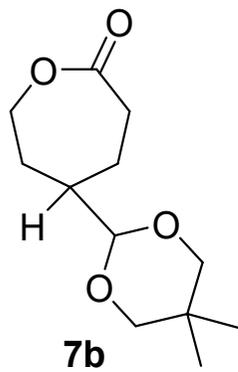
40.559  
40.467

30.335  
27.027  
22.980  
21.868

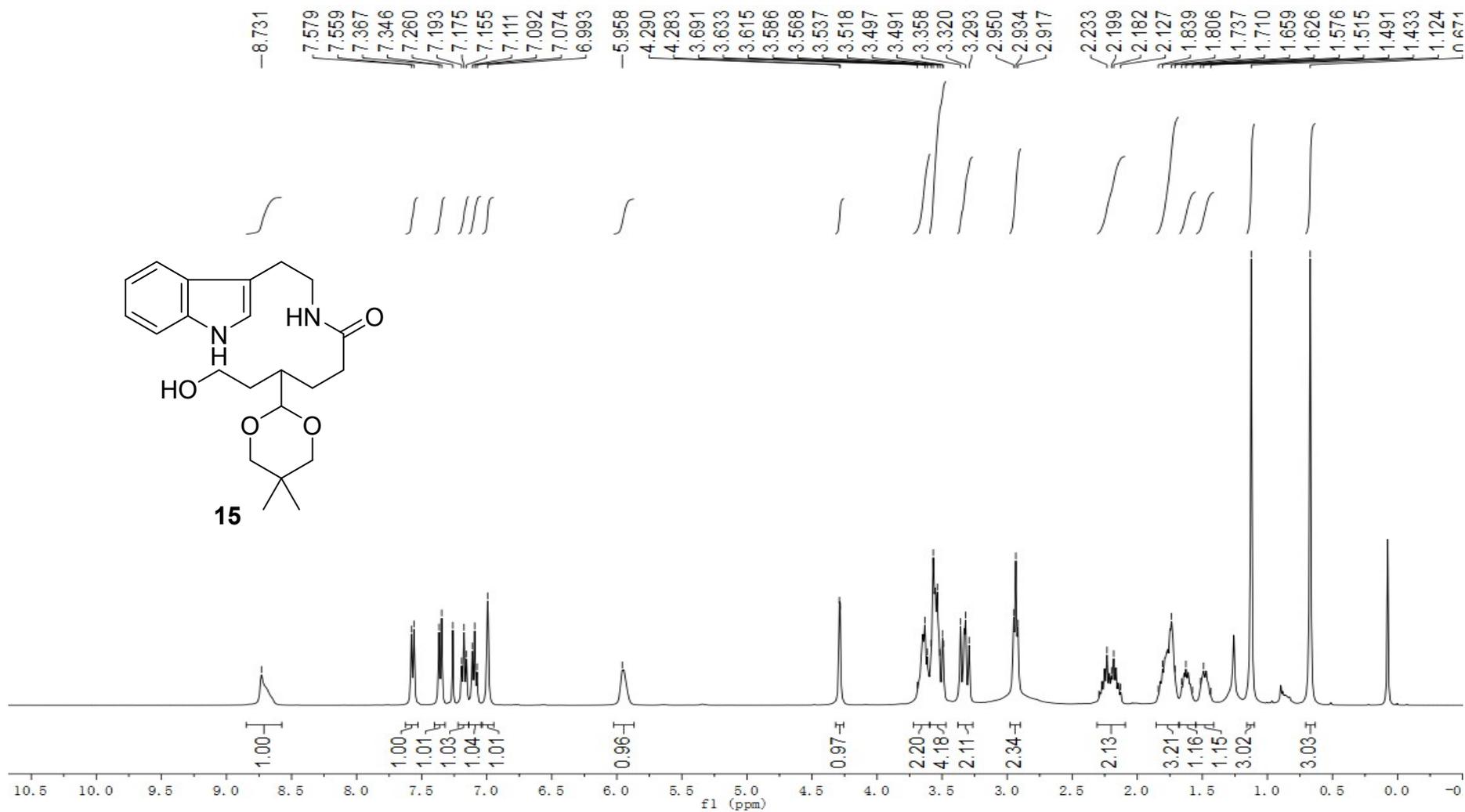


**<sup>13</sup>C-NMR Spectrum of 8b**

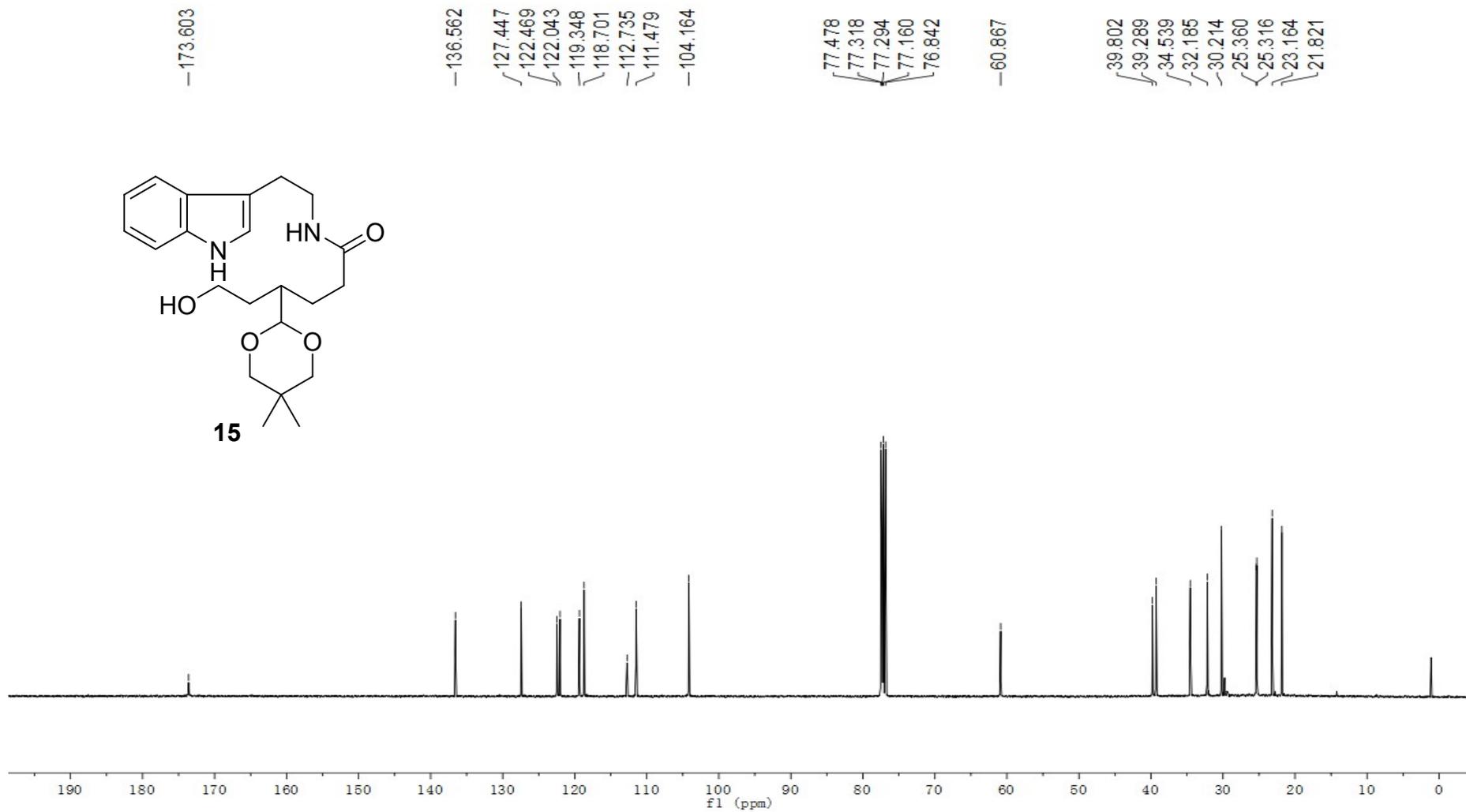
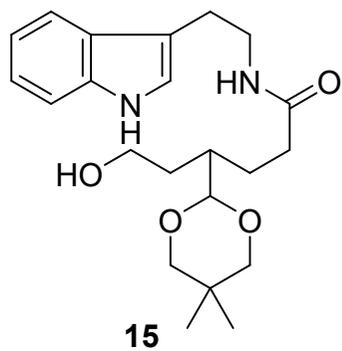




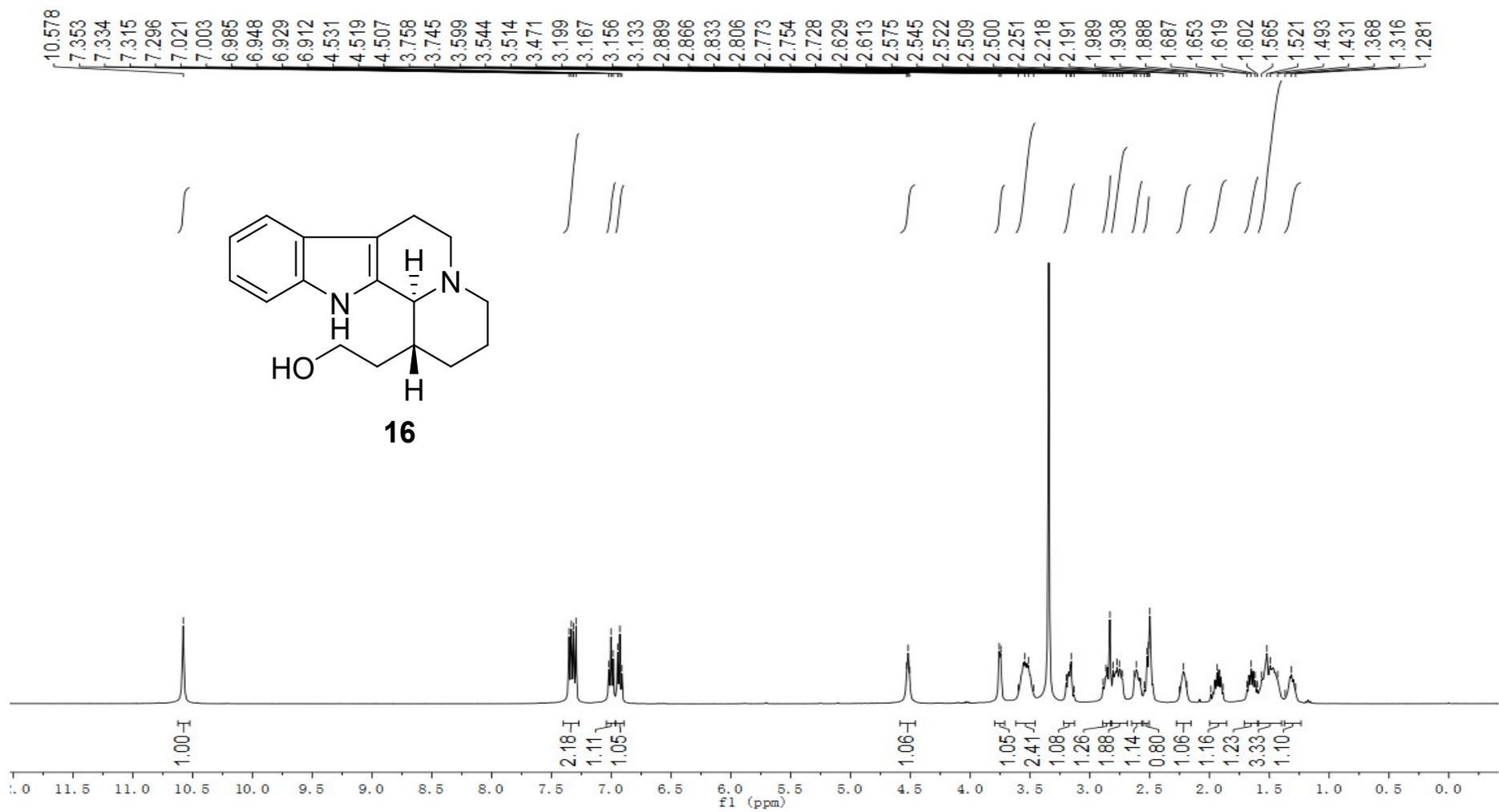
**<sup>13</sup>C-NMR Spectrum of 7b**



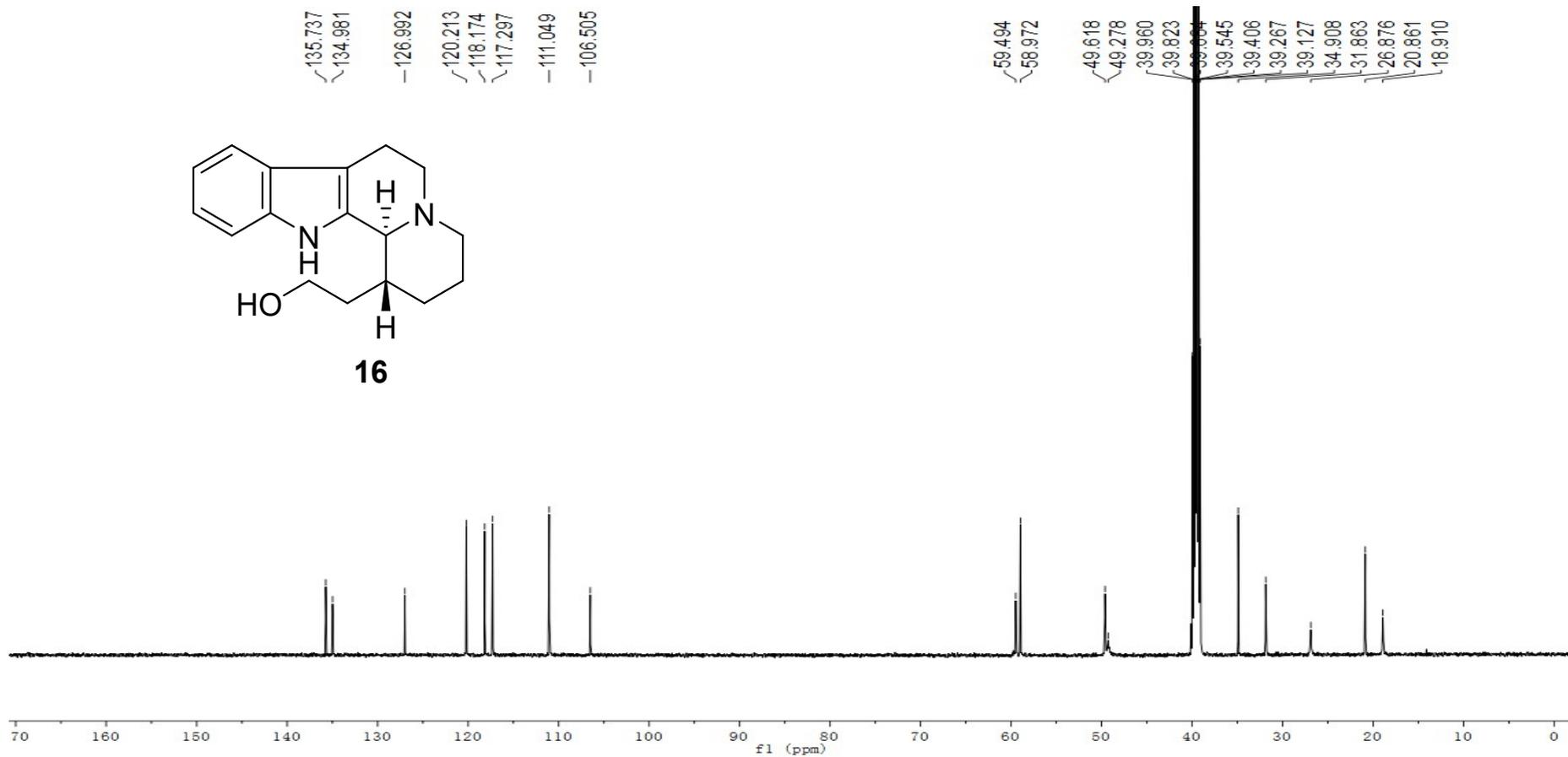
**<sup>1</sup>H-NMR Spectrum of 15**



**<sup>13</sup>C-NMR Spectrum of 15**

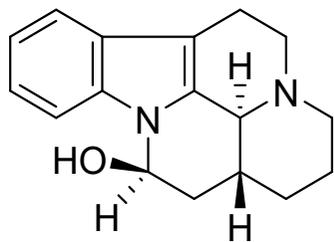


**1H-NMR Spectrum of 16**

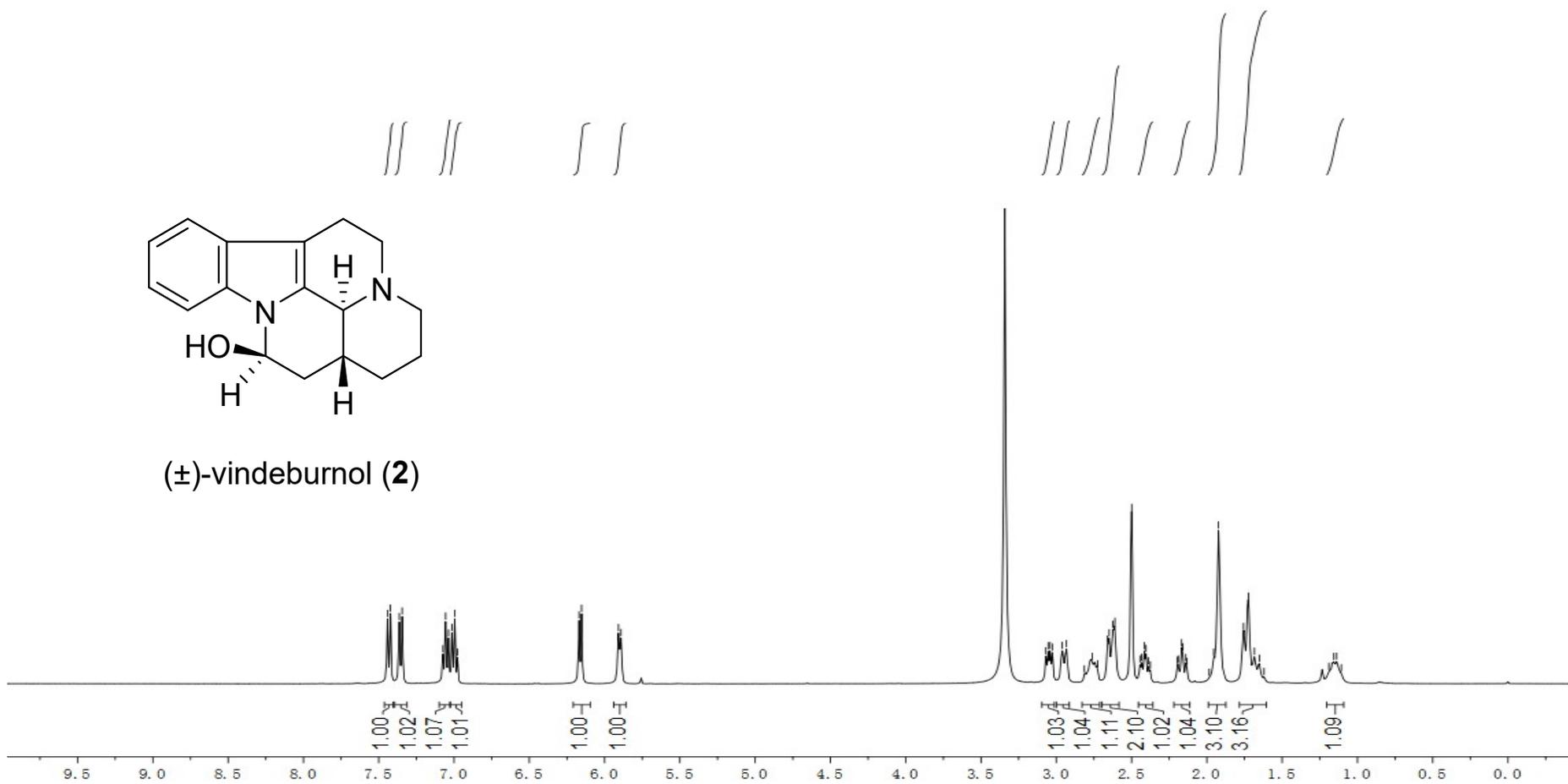


$^{13}\text{C-NMR}$  Spectrum of 16

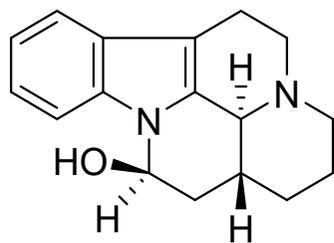
7.442  
7.422  
7.363  
7.344  
7.074  
7.056  
7.036  
7.014  
6.995  
6.977  
6.170  
6.153  
5.908  
5.891  
3.069  
3.054  
3.041  
3.026  
2.960  
2.933  
2.814  
2.762  
2.725  
2.660  
2.649  
2.623  
2.610  
2.499  
2.444  
2.433  
2.416  
2.404  
2.387  
2.376  
2.196  
2.190  
2.168  
2.161  
2.140  
2.133  
1.985  
1.955  
1.923  
1.757  
1.723  
1.684  
1.652  
1.620  
1.190  
1.157  
1.138  
1.107



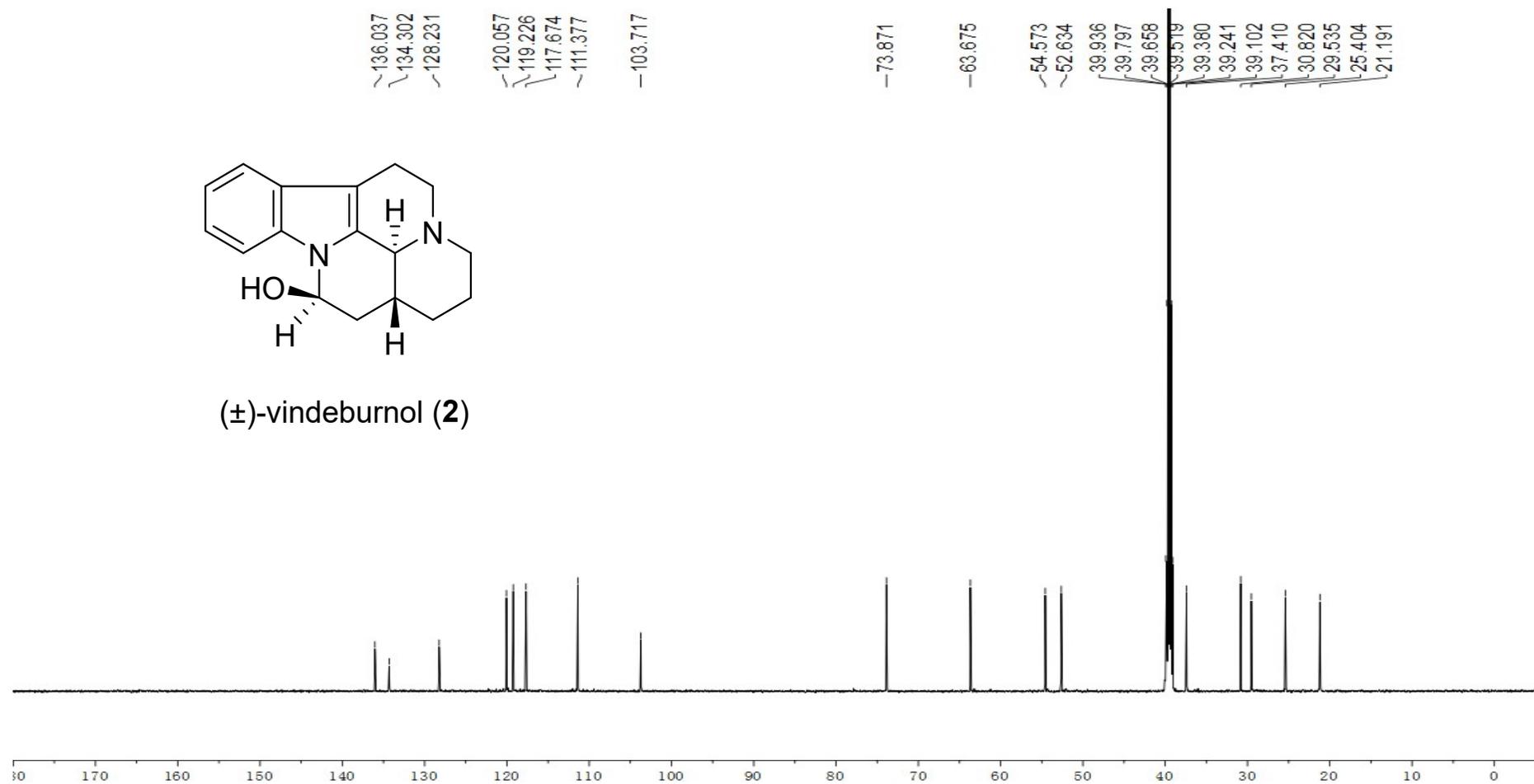
(±)-vindeburnol (**2**)



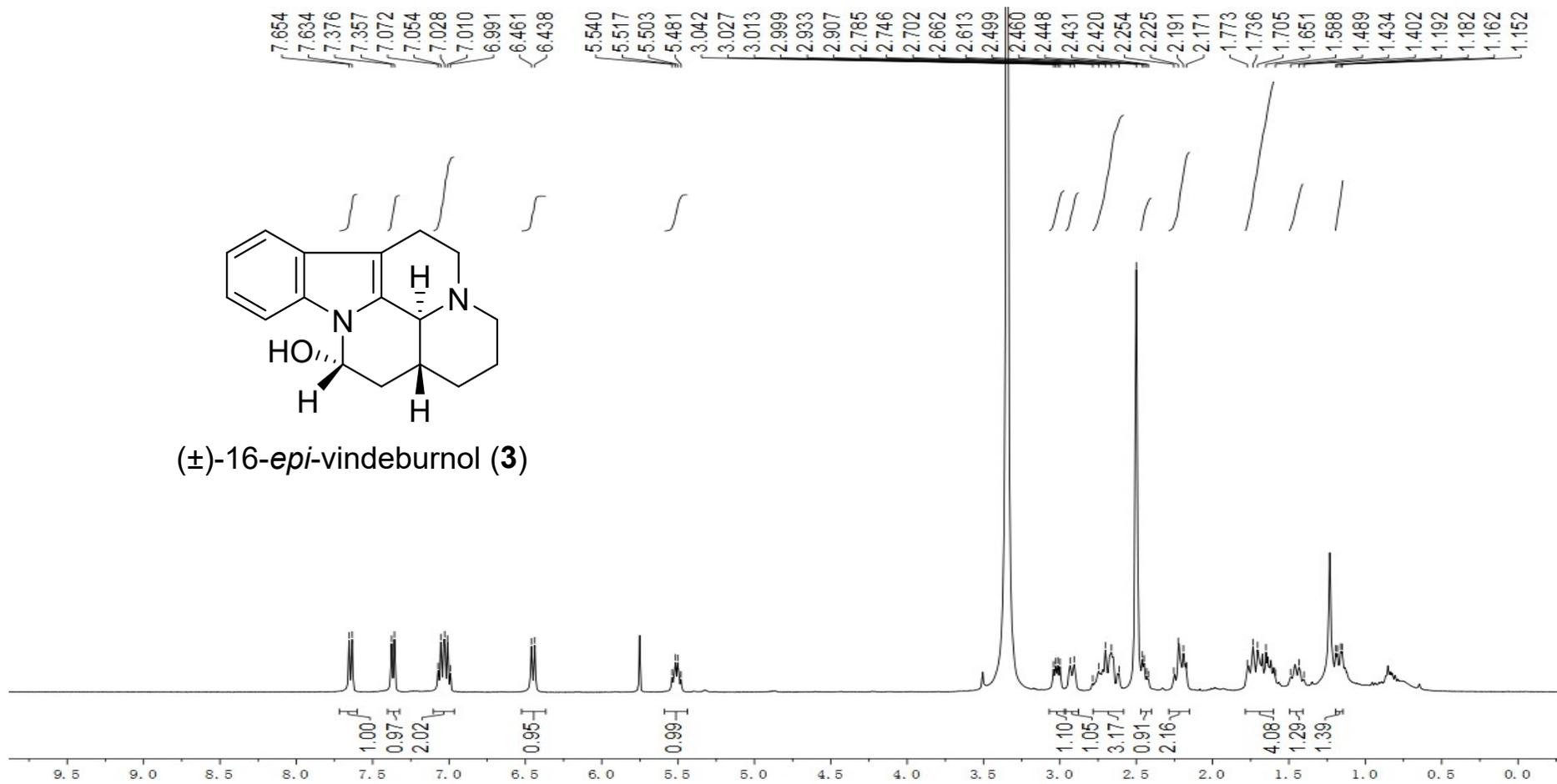
<sup>1</sup>H-NMR Spectrum of (±)-vindeburnol (**2**)



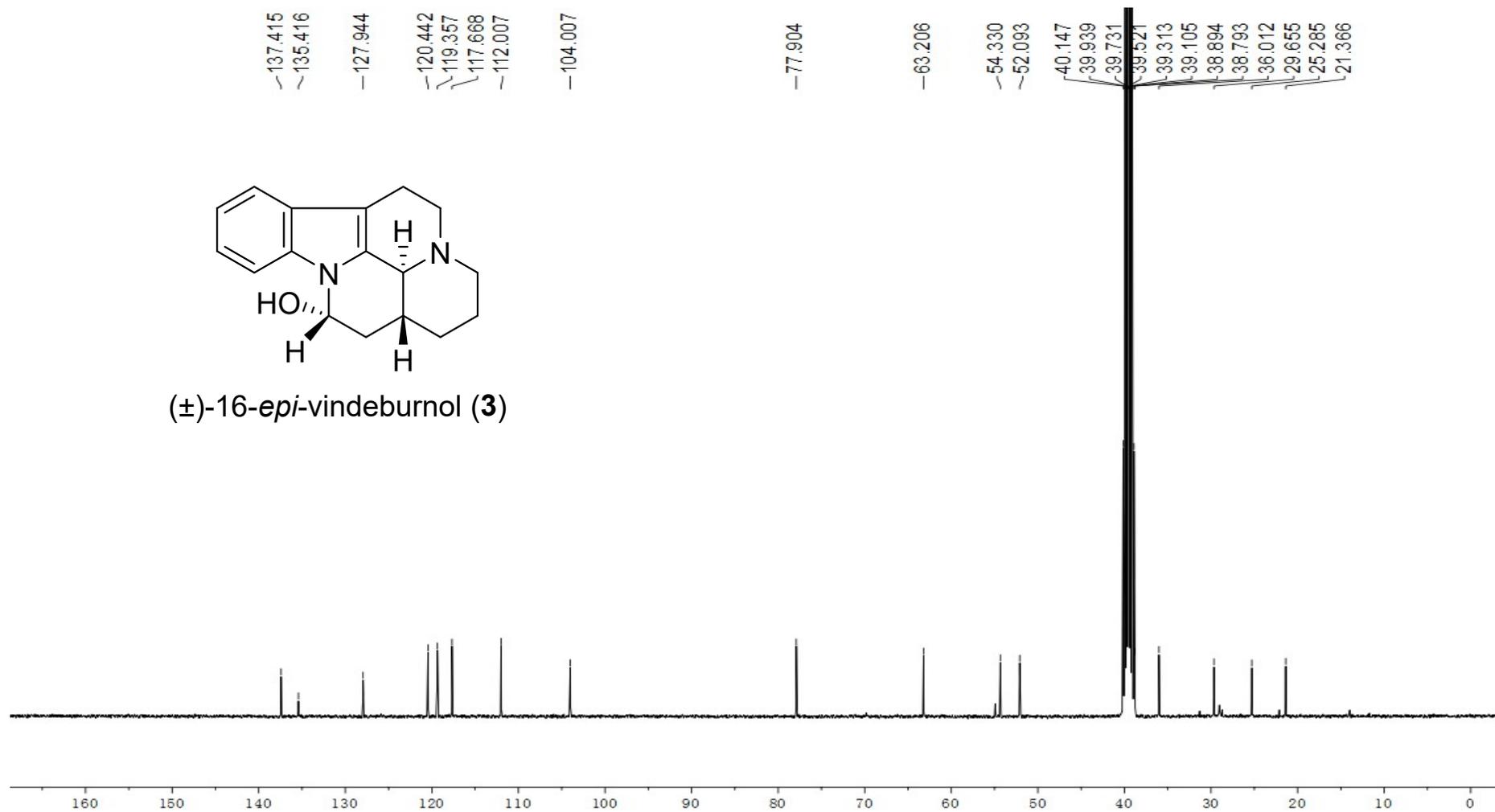
(±)-vindeburnol (**2**)

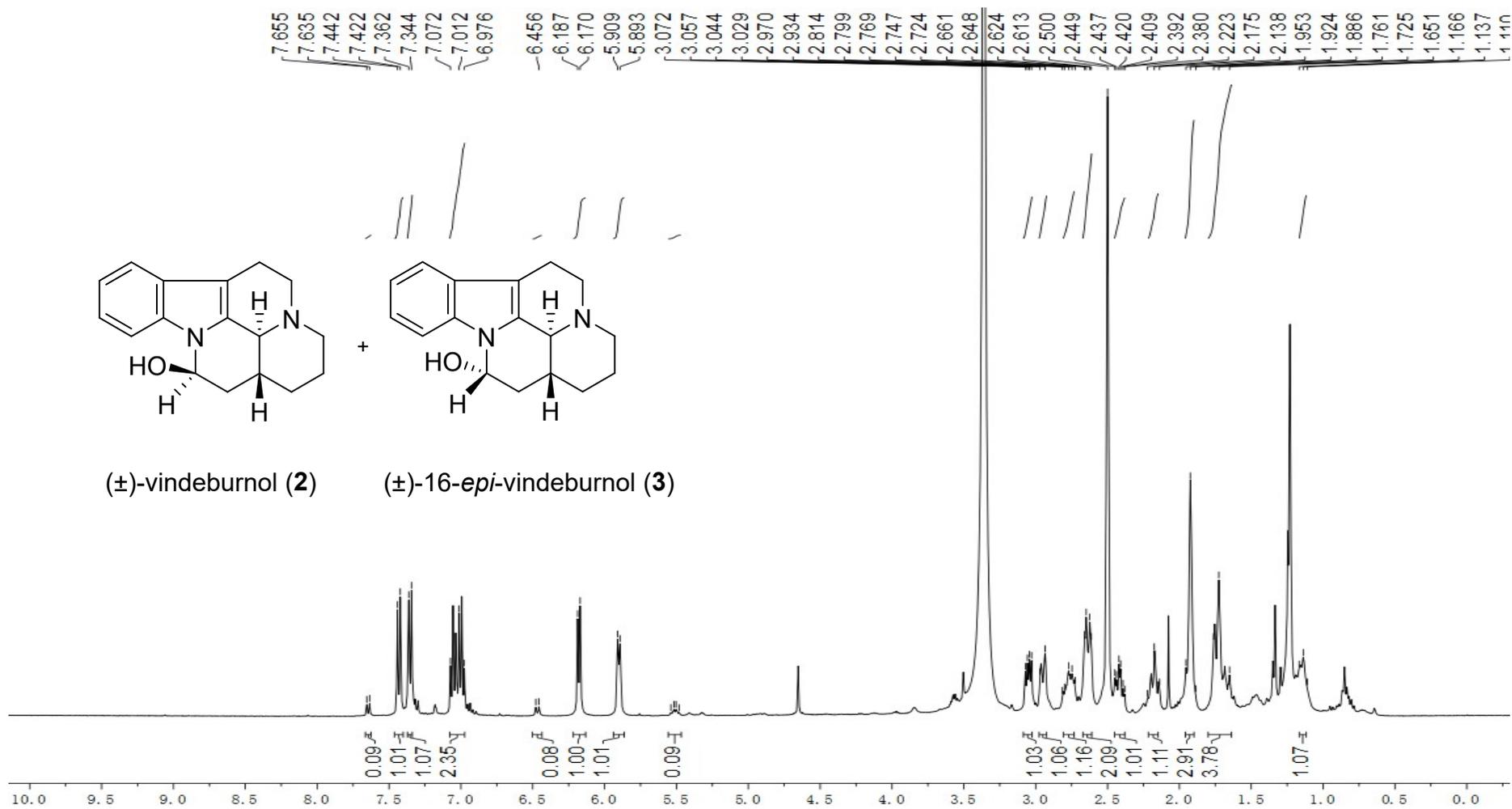


$^{13}\text{C}$ -NMR Spectrum of (±)-vindeburnol (**2**)

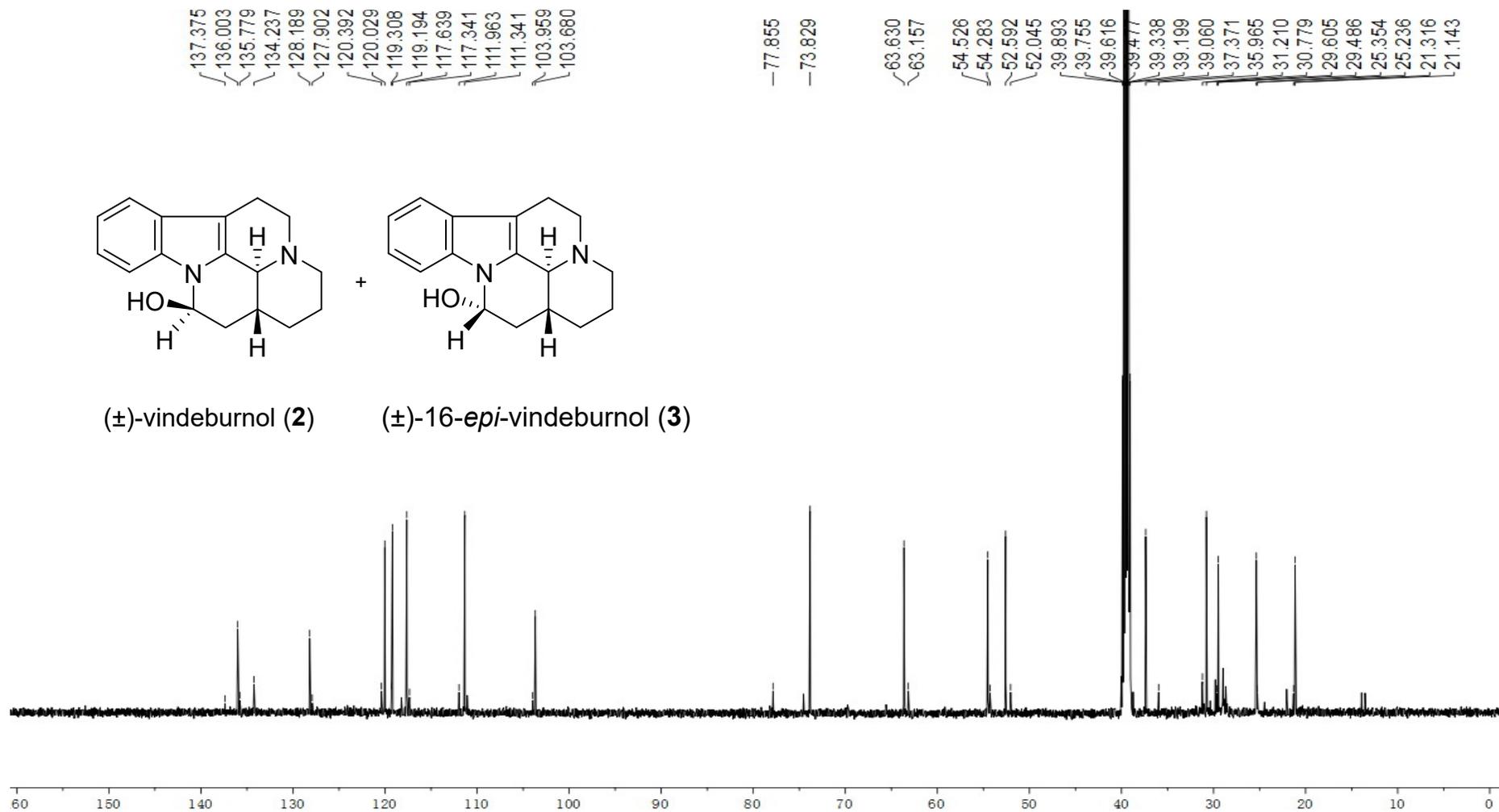


**<sup>1</sup>H-NMR Spectrum of Compound 16-epi-(±)-vindeburnol (3)**





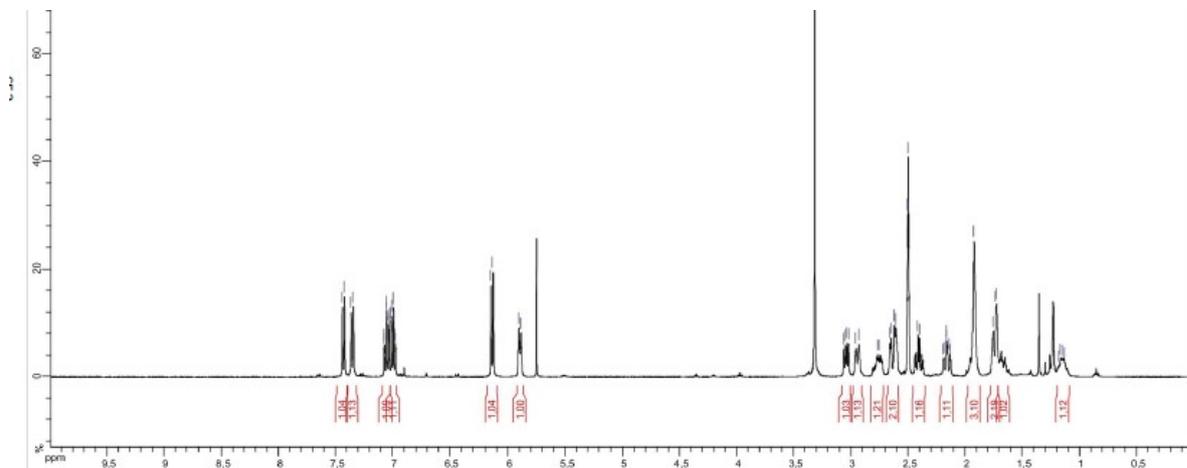
**<sup>1</sup>H-NMR Spectrum of Mixture of Compound (±)-vindeburnol (2)/16-*epi*-(±)-vindeburnol (3) (2:3 = 13:1)**



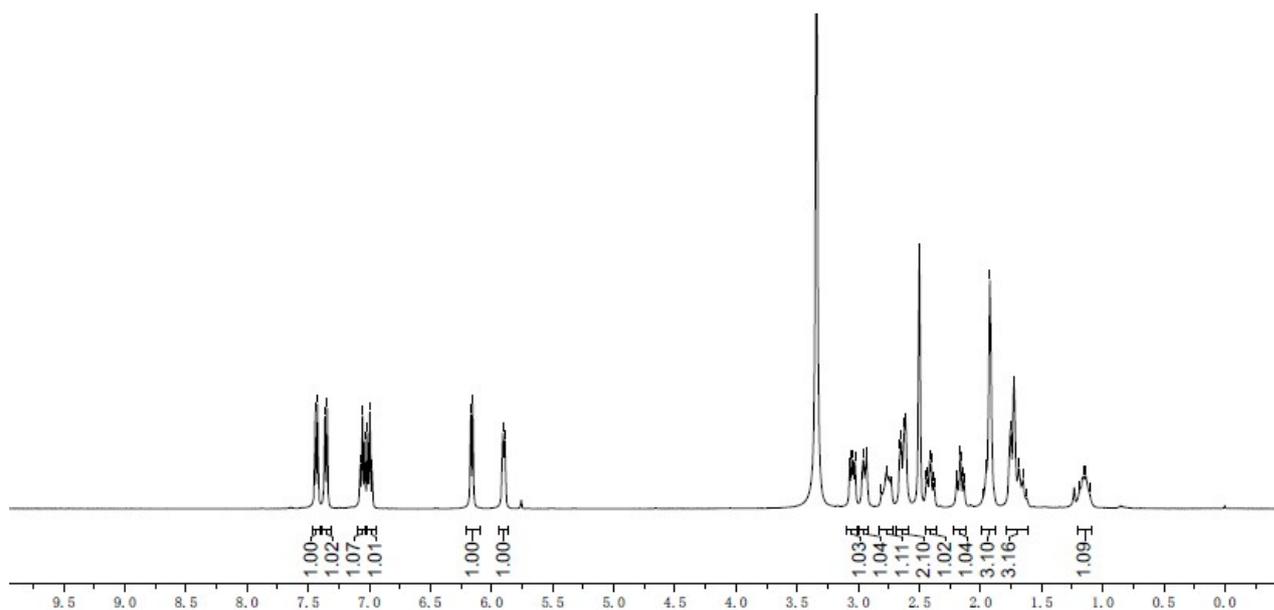
**$^{13}\text{C}$ -NMR Spectrum of Mixture of Compound (±)-vindeburnol (2) and 16-*epi*-(±)-vindeburnol (3) (2:3 = 13:1)**

**Comparison of  $^1\text{H}$  spectrums between the synthesized ( $\pm$ )-vindeburnol (2) by Prof. Girard and the synthesized 2 by Prof. Chen.**

**The synthesized ( $\pm$ )-vindeburnol by Prof. Girard**

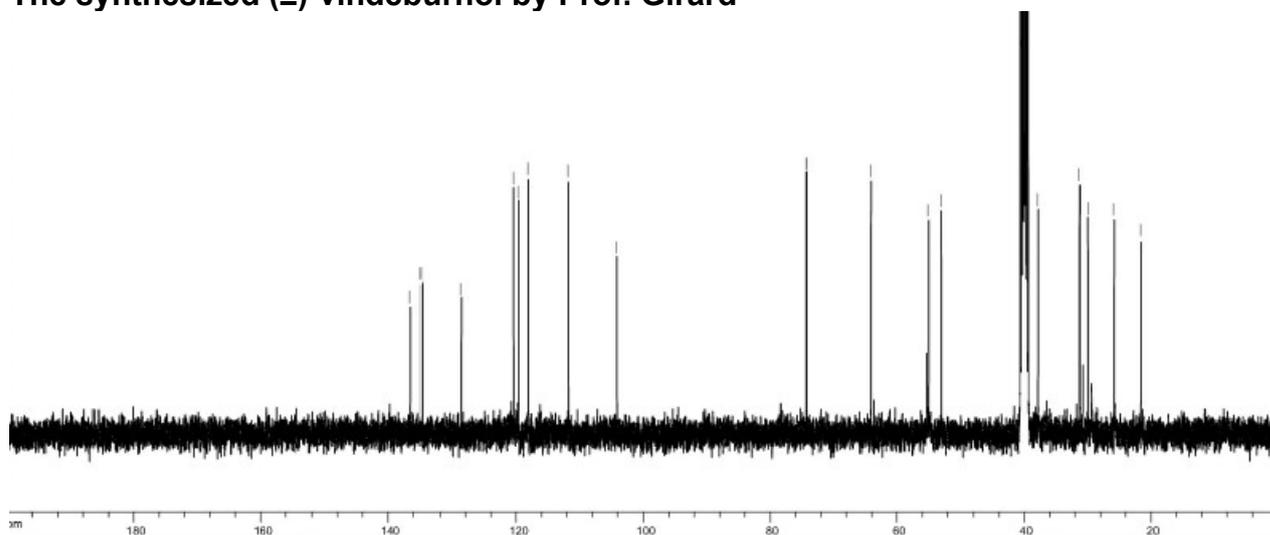


**The synthesized ( $\pm$ )-vindeburnol by Prof. Chen**

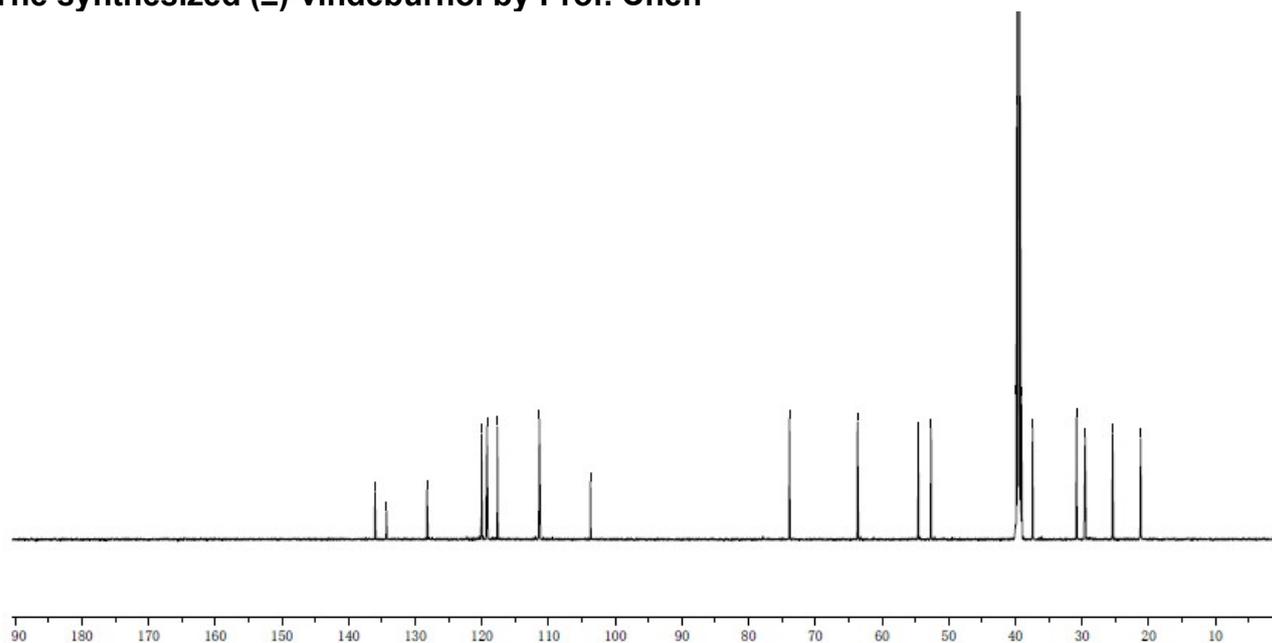


**Comparison of  $^{13}\text{C}$  spectrums between the synthesized ( $\pm$ )-vindeburnol (2) by Prof. Girard and the synthesized 2 by Prof. Chen.**

**The synthesized ( $\pm$ )-vindeburnol by Prof. Girard**

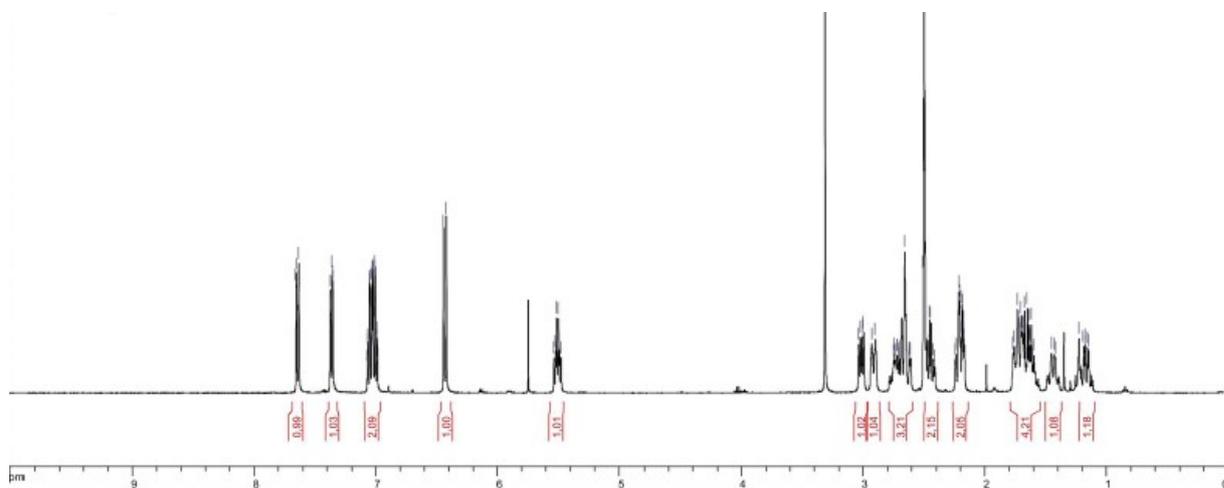


**The synthesized ( $\pm$ )-vindeburnol by Prof. Chen**

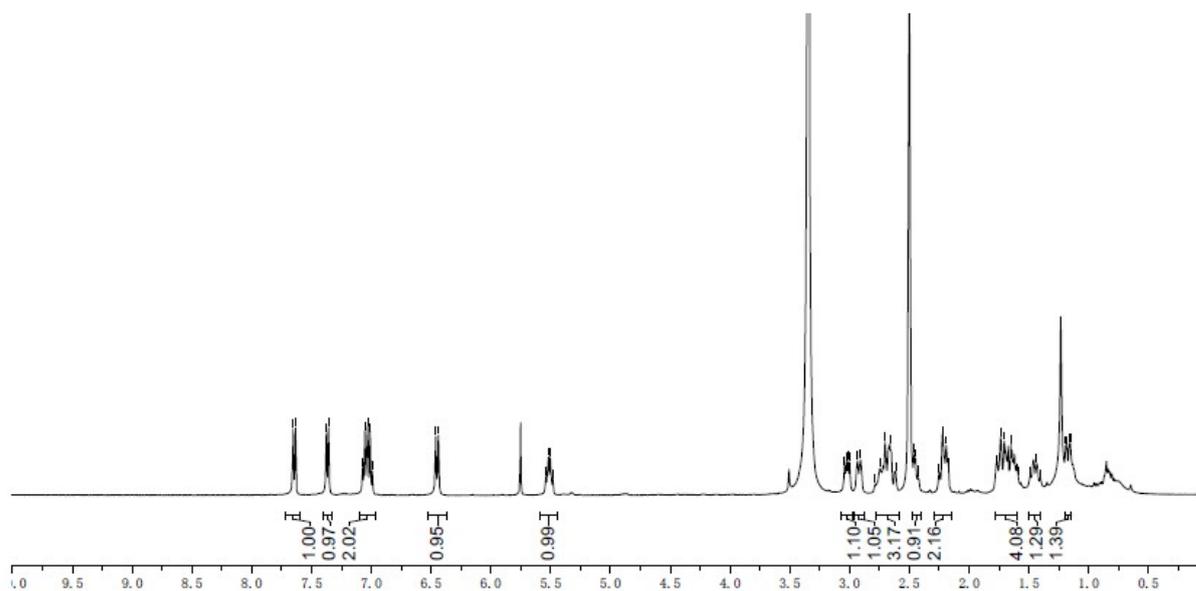


Comparison of  $^1\text{H}$  spectrums between the synthesized 16-*epi*-( $\pm$ )-vindeburnol (3) by Prof. Girard and the synthesized 3 by Prof. Chen.

The synthesized 16-*epi*-( $\pm$ )-vindeburnol (3) by Prof. Girard

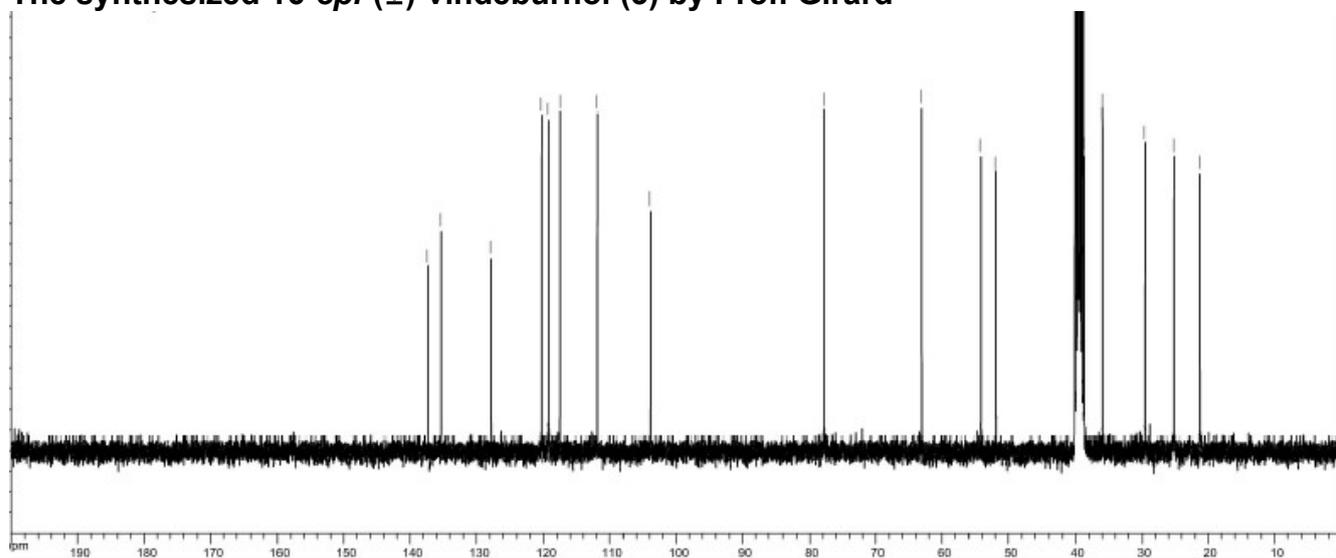


The synthesized 16-*epi*-( $\pm$ )-vindeburnol (3) by Prof. Chen

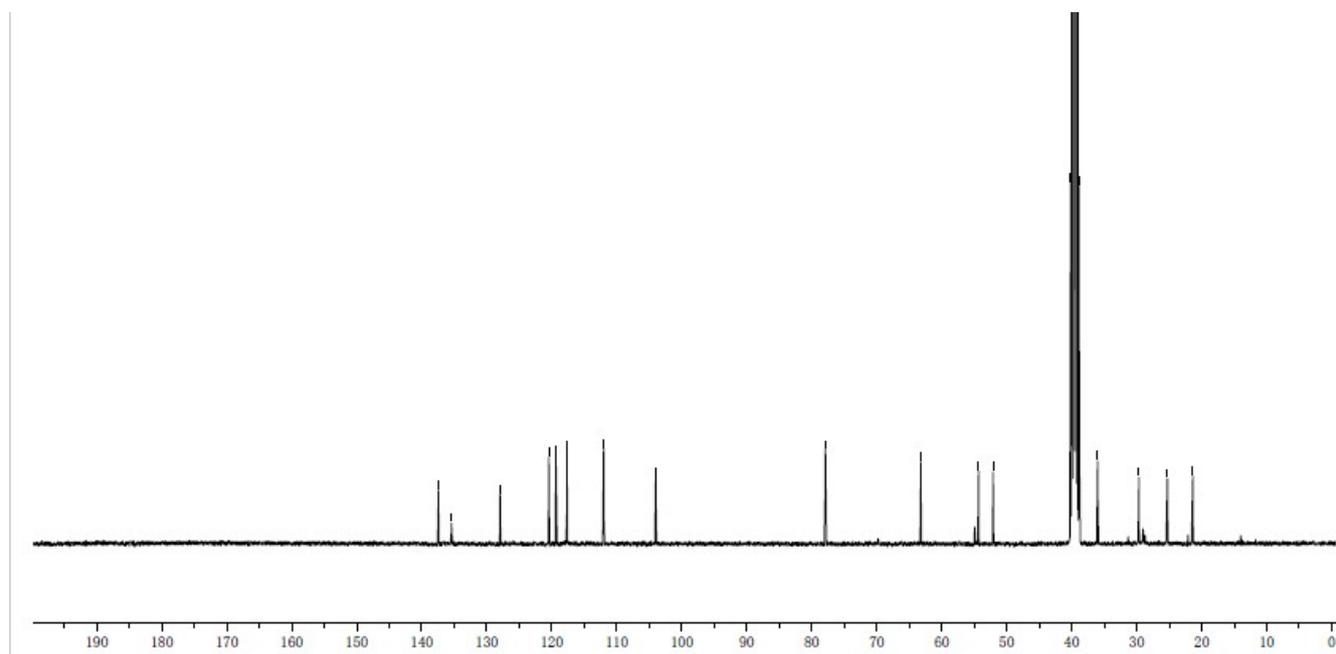


Comparison of  $^{13}\text{C}$  spectrums between the synthesized 16-*epi*-(±)-vindeburnol (3) by Prof. Girard and the synthesized 3 by Prof. Chen.

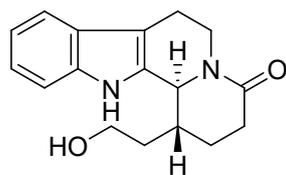
The synthesized 16-*epi*-(±)-vindeburnol (3) by Prof. Girard



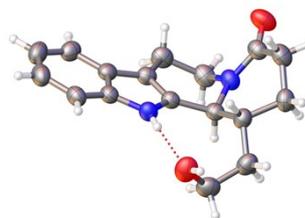
The synthesized 16-*epi*-(±)-vindeburnol (3) by Prof. Chen



## Crystallographic Data



**4a**

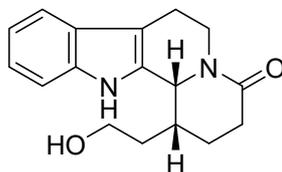


X-Ray of **4a**

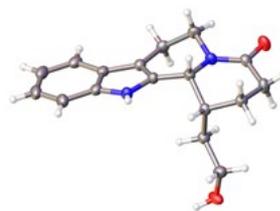
Figure 1. ORTEP of the molecular structure of **4a**  
CCDC 2103051 contains the supplementary crystallographic data for compound **4a**

Empirical formula	C <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> ·H <sub>2</sub> O
Formula weight	284.35
Temperature/K	297.1(5)
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /n
a/Å	10.1352(3)
b/Å	18.7726(5)
c/Å	15.8738(5)
α/°	90
β/°	103.950(3)
γ/°	90
Volume/Å <sup>3</sup>	2931.14(15)
Z	8
ρ <sub>calc</sub> /cm <sup>3</sup>	1.289
μ/mm <sup>-1</sup>	0.681
F(000)	1216.0
Crystal size/mm <sup>3</sup>	0.5 × 0.45 × 0.2
Radiation	CuKα (λ = 1.54184)
2θ range for data collection/°	7.424 to 142.708
Index ranges	-12 ≤ h ≤ 12, -22 ≤ k ≤ 13, -19 ≤ l ≤ 19
Reflections collected	13940
Independent reflections	5563 [R <sub>int</sub> = 0.0410, R <sub>sigma</sub> = 0.0357]
Data/restraints/parameters	5563/0/381
Goodness-of-fit on F <sup>2</sup>	1.028
Final R indexes [I ≥ 2σ (I)]	R <sub>1</sub> = 0.0565, wR <sub>2</sub> = 0.1488
Final R indexes [all data]	R <sub>1</sub> = 0.0654, wR <sub>2</sub> = 0.1611
Largest diff. peak/hole / e Å <sup>-3</sup>	0.24/-0.27

These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).



**4b**



X-Ray of **4b**

Figure 2. ORTEP of the molecular structure of **4b**

CCDC 2103479 contains the supplementary crystallographic data for compound **4b**.

Empirical formula	C <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub>
Formula weight	284.35
Temperature/K	150.00(10)
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c

a/Å	6.52889(15)
b/Å	26.9565(6)
c/Å	9.50589(19)
$\alpha$ /°	90
$\beta$ /°	104.117(2)
$\gamma$ /°	90
Volume/Å <sup>3</sup>	1622.47(6)
Z	4
$\rho_{\text{calc}}$ /cm <sup>3</sup>	1.164
$\mu$ /mm <sup>-1</sup>	0.615
F(000)	608.0
Crystal size/mm <sup>3</sup>	0.5 × 0.3 × 0.05
Radiation	CuK $\alpha$ ( $\lambda$ = 1.54184)
2 $\theta$ range for data collection/°	10.14 to 142.726
Index ranges	-7 ≤ h ≤ 8, -32 ≤ k ≤ 28, -10 ≤ l ≤ 11
Reflections collected	9153
Independent reflections	3096 [R <sub>int</sub> = 0.0461, R <sub>sigma</sub> = 0.0396]
Data/restraints/parameters	3096/0/195
Goodness-of-fit on F <sup>2</sup>	1.062
Final R indexes [I >= 2 $\sigma$ (I)]	R <sub>1</sub> = 0.0533, wR <sub>2</sub> = 0.1467
Final R indexes [all data]	R <sub>1</sub> = 0.0585, wR <sub>2</sub> = 0.1525
Largest diff. peak/hole / e Å <sup>-3</sup>	0.24/-0.26

These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

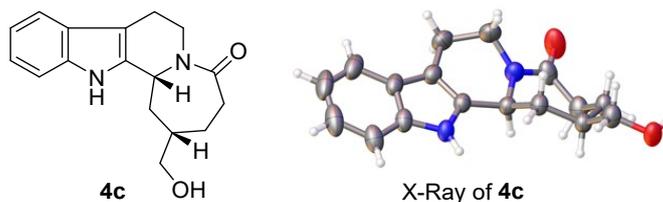


Figure 3. ORTEP of the molecular structure of **4c**  
CCDC 2103481 contains the supplementary crystallographic data for compound **4c**

Empirical formula	C <sub>17</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> ·
Formula weight	284.35
Temperature/K	297.4(2)
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /n
a/Å	10.0470(2)
b/Å	11.20580(19)
c/Å	13.1505(3)
$\alpha$ /°	90
$\beta$ /°	101.241(2)
$\gamma$ /°	90
Volume/Å <sup>3</sup>	1452.15(6)
Z	4
$\rho_{\text{calc}}$ /cm <sup>3</sup>	1.301
$\mu$ /mm <sup>-1</sup>	0.687
F(000)	608.0
Crystal size/mm <sup>3</sup>	0.4 × 0.3 × 0.3
Radiation	CuK $\alpha$ ( $\lambda$ = 1.54184)
2 $\theta$ range for data collection/°	10.178 to 142.792
Index ranges	-11 ≤ h ≤ 12, -13 ≤ k ≤ 8, -14 ≤ l ≤ 15
Reflections collected	7040
Independent reflections	2768 [R <sub>int</sub> = 0.0349, R <sub>sigma</sub> = 0.0338]

Data/restraints/parameters	2768/0/191
Goodness-of-fit on $F^2$	1.060
Final R indexes [ $I \geq 2\sigma(I)$ ]	$R_1 = 0.0648$ , $wR_2 = 0.1612$
Final R indexes [all data]	$R_1 = 0.0711$ , $wR_2 = 0.1710$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	0.26/-0.42

These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif)