

Supporting Information

Cp*Rh(III)/Boron Hybrid Catalysis for Directed C–H Addition to β -Substituted α,β -Unsaturated Carboxylic Acids.

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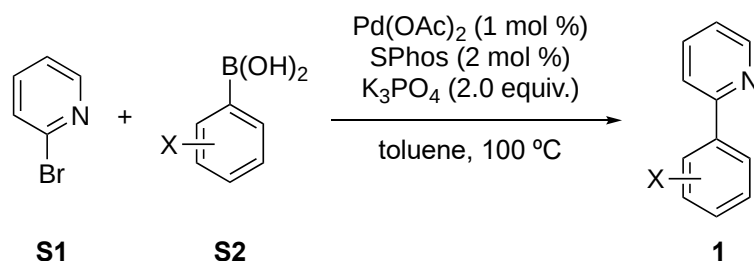
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General

Reported melting points were uncorrected. Infrared (IR) spectra were recorded on a JASCO FT/IR-5300 spectrophotometer and absorbance bands are reported in wave numbers (cm^{-1}). NMR spectra were recorded on JEOL JNM-ECS400 spectrometers operating at 391.78 MHz for ^1H NMR and 98.52 MHz for ^{13}C NMR, JEOL JNM-ECX400 spectrometers, operating at 395.88 MHz for ^1H NMR and 99.55 MHz for ^{13}C NMR, JEOL JNM-ECZ400 spectrometers, operating at 399.78 MHz for ^1H NMR and 100.53 MHz for ^{13}C NMR and 368.62 MHz for ^{19}F NMR, and JNM-ECA500 spectrometers, operating at 500.16 MHz for ^1H NMR and 125.77 MHz for ^{13}C NMR. Chemical shifts were reported in the scale relative to TMS (0.00 ppm for ^1H NMR), CHCl_3 (7.26 ppm for ^1H NMR), CDCl_3 (77.0 ppm for ^{13}C NMR), and benzotrifluoride (-62.75 ppm for ^{19}F NMR) as an internal reference, respectively. ESI mass spectra were measured on JEOL JMS-T100LCP spectrometer. Silica gel column chromatography was performed with Kanto Silica gel 60 N (40-50 mesh) or Wakogel® C-200. Dichloromethane (CH_2Cl_2) and toluene were purified by Glass Contour solvent purification system. 1,1,1,3,3,3-Hexafluoropropan-2-ol (HFIP) were distilled from CaH_2 , purged with argon for over 30 min, and stored over activated molecular sieves 4A under argon atmosphere before use. Commercially available 1,2-dichloroethane (DCE, KANTO CHEMICAL Co.,INC., dehydrated -Super-), 1,4-dioxane (FUJIFILM Wako Pure Chemical Corporation, super dehydrated grade), and dimethylformamide (DMF, FUJIFILM Wako Pure Chemical Corporation, super dehydrated grade) were used without further manipulation unless otherwise stated. Carboxylic acids **2a**, **2c**, **2d**, and **2e** were synthesized according to the literature.^{1,2} Carboxylic acid **2b** was commercially available and distilled under reduced pressure before use. All other reagents were commercially available and used as received.

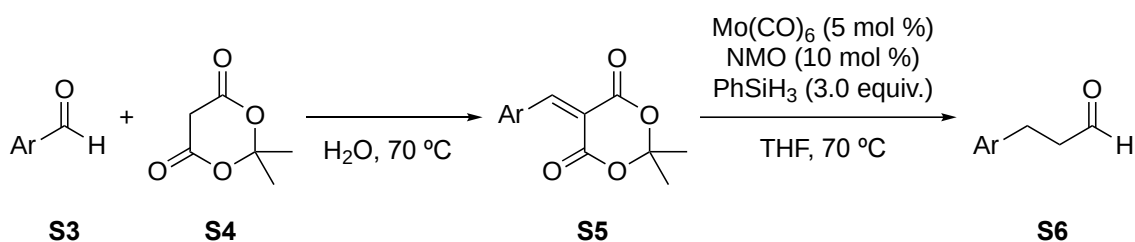
General procedure for the synthesis of 2-arylpyridine derivatives



To a flamed dried test tube were added 2-bromopyridine **S1**, arylboronic acid **S2** (2.0 equiv.), Pd(OAc)_2 (1 mol %), SPhos (2 mol %), K_3PO_4 (2.0 equiv.), and toluene (0.4 M). The resulting mixture was heated at 100 °C. After 2-bromopyridine was disappeared (confirmed by TLC

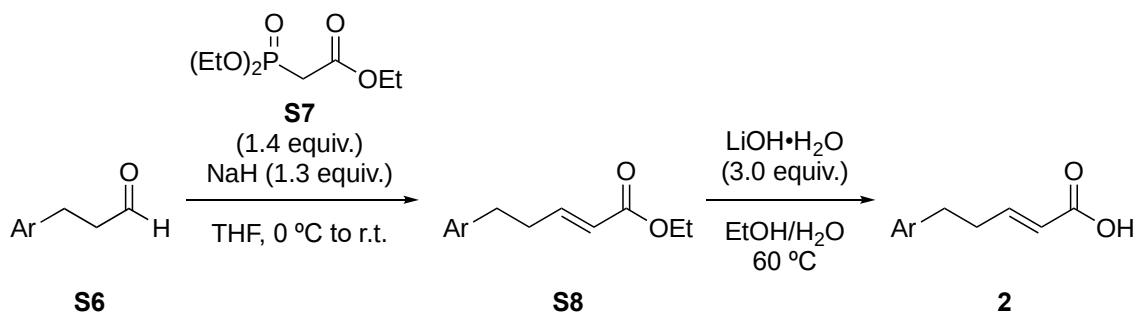
analysis), the reaction mixture was cooled to room temperature and H₂O was added. The mixture was extracted with AcOEt, and the organic layer was washed with H₂O and brine. After drying with Na₂SO₄, filtration, and evaporation, the crude mixture was purified by SiO₂ column chromatography (hexane/AcOEt) to afford the corresponding product **1**. The spectroscopic data of **1** were matched with those in previously reports.³⁻⁸

General procedure for the synthesis of α,β -unsaturated carboxylic acids



S6 was prepared from corresponding aldehyde **S3** according to the literature.⁹ To a recovery flask were added aldehyde **S3** (1.0 equiv.), Meldrum's acid **S4** (1.1 equiv.), and H₂O (0.5 M). The mixture was heated at 70 °C for 4 h. After cooling to room temperature, the precipitates were collected by filtration and washed by MeOH to give the corresponding product **S5**.

To a flame dried recovery flask were added **S5** (1.0 equiv.), Mo(CO)₆ (5 mol %), NMO (10 mol %), and THF (0.2 M). To the resulting solution was added PhSiH₃ (3.0 equiv.) and the reaction mixture was heated at 70 °C overnight. After cooling to room temperature, H₂O was added. The mixture was extracted with AcOEt, and the organic layer was washed with sat. NaHCO₃ aq, H₂O, and brine. After drying with Na₂SO₄, filtration, and evaporation, the crude mixture was purified by SiO₂ column chromatography (hexane/AcOEt) to afford the corresponding aldehyde **S6**.



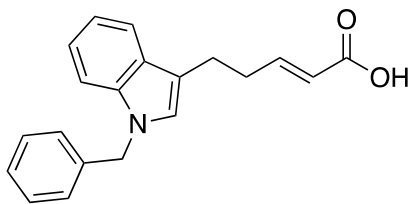
To a solution of NaH (1.3 equiv.) and THF (0.2 M) was added diethylphosphonoacetic acid ethyl ester **S7** (1.4 equiv.) at 0 °C. After 30 min., aldehyde **S6** (1.0 equiv.) in THF was slowly added. After stirring at room temperature for 1 h, the reaction mixture was quenched by sat.

NH₄Cl aq and extracted with AcOEt. The organic layer was washed with H₂O and brine, dried with Na₂SO₄. After filtration and evaporation, the crude mixture was purified by SiO₂ column chromatography (hexane/AcOEt) to afford **S8**.

To a recovery flask were added **S8** (1.0 equiv.), LiOH•H₂O (3.0 equiv.), and EtOH/H₂O (1:1, 0.2 M). The resulting mixture was heated at 60 °C until **S8** disappeared (confirmed by TLC). After cooling to room temperature, the reaction mixture was quenched by 10% HCl aq, and extracted with AcOEt. The organic layer was washed with brine, dried with Na₂SO₄. After filtration and evaporation, the crude mixture was purified by SiO₂ column chromatography (hexane/AcOEt) or recrystallization (CH₂Cl₂/hexane) to afford **2**.

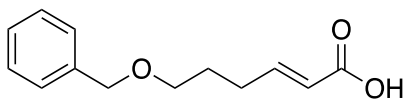
Characterization data of new α,β -unsaturated carboxylic acids

(*E*)-5-(1-benzyl-1*H*-indol-3-yl)pent-2-enoic acid (**2f**)



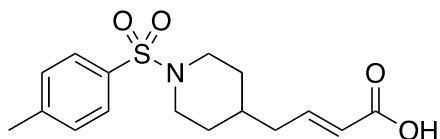
673.4 mg; a yellow solid; mp 88-90 °C; IR (KBr) ν 3028, 1687, 1646, 1468, 1285, 737, 698 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 2.61-2.70 (m, 2H), 2.95 (t, J = 7.4 Hz, 2H), 5.28 (s, 2H), 5.86 (dt, J = 15.6, 1.5 Hz, 1H), 6.92 (s, 1H), 7.07-7.21 (m, 5H), 7.24-7.32 (m, 4H), 7.59 (d, J = 7.6 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ 23.7, 32.8, 49.8, 109.7, 114.1, 118.9, 119.0, 121.0, 121.8, 125.6, 126.7, 127.5, 127.8, 128.7, 136.7, 137.6, 151.8, 171.9; HRMS (ESI): m/z calculated for C₂₀H₁₈O₂N [M-H]: 304.1343, found: 304.1347.

(*E*)-6-(benzyloxy)hex-2-enoic acid (**2g**)



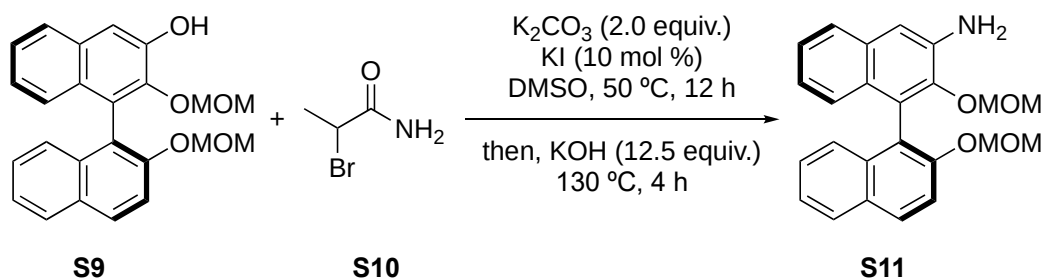
650.5 mg; a colorless oil; IR (KBr) ν 2861, 1696, 1651, 1421, 1288, 1104, 983, 739, 698 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 1.79 (tt, J = 7.1, 6.3 Hz, 2H), 2.32-2.40 (m, 2H), 3.50 (t, J = 6.3 Hz, 2H), 4.50 (s, 2H), 5.82-5.87 (m, 1H), 7.09 (dt, J = 15.6, 7.2 Hz, 1H), 7.25-7.38 (m, 5H); ¹³C NMR (CDCl₃, 100 MHz) δ 28.0, 29.0, 69.1, 73.0, 121.0, 127.62, 127.65, 128.4, 138.3, 151.5, 171.9; HRMS (ESI): m/z calculated for C₁₃H₁₅O₃ [M-H]: 219.1027, found: 219.1028.

(*E*)-4-(1-tosylpiperidin-4-yl)but-2-enoic acid (**2h**)

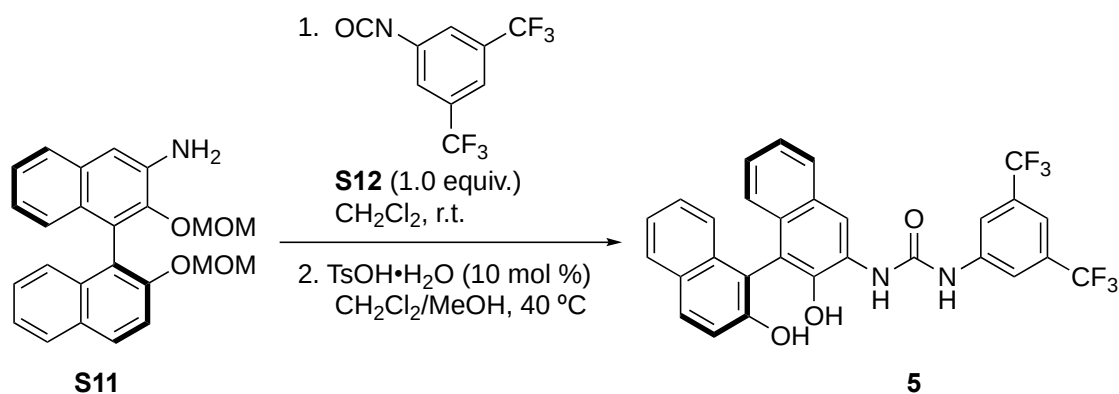


809.2 mg; a colorless solid; mp 136-138 °C; IR (KBr) ν 2924, 1693, 1649, 1424, 1338, 1162, 1095, 931, 734 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.30-1.42 (m, 3H), 1.73 (d, J = 9.9 Hz, 2H), 2.13-2.27 (m, 4H), 2.44 (s, 3H), 3.78 (d, J = 11.9 Hz, 2H), 5.81 (dt, J = 15.5, 1.5 Hz, 1H), 6.94 (dt, J = 15.5, 7.6 Hz, 1H), 7.32 (d, J = 8.2 Hz, 2H), 7.63 (d, J = 8.2 Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 21.5, 31.3, 34.7, 38.7, 46.2, 122.3, 127.7, 129.6, 133.2, 143.4, 149.1, 170.9; HRMS (ESI): m/z calculated for $\text{C}_{16}\text{H}_{20}\text{O}_4\text{NS}$ [$\text{M}-\text{H}$]: 322.1119, found: 322.1122.

Synthesis of BINOL-urea ligand **5**



S11 was prepared from **S9**¹⁰ according to the literature.¹¹ To a 100 mL recovery flask were added **S9** (3.9 g, 10.1 mmol), 2-bromopropionamide **S10** (3.1 g, 20.2 mmol, 2.0 equiv.), K_2CO_3 (2.8 g, 0.2 mmol, 2.0 equiv.), KI (166 mg, 1.0 mmol, 0.1 equiv.), and DMSO (40 mL, 0.25 M). The reaction mixture was stirred at 50 °C for 2 h. To the resulting mixture was added KOH (7.1 g, 126 mmol, 12.5 equiv.), and the mixture was heated at 150 °C for 4 h. After the mixture was cooled to room temperature, H_2O and AcOEt were added. The organic layer was separated, and the aqueous layer was extracted with AcOEt three times. The combined organic layers were washed with H_2O and brine, and dried over Na_2SO_4 . After filtration and evaporation, the obtained crude mixture was purified by silica gel column chromatography (hexane/AcOEt) to give corresponding product **S11** (ca. 6.7 mmol, ca. 66%) with inseparable byproducts. **S11** was used for the next step without further purification.



To a 100 mL recovery flask were added **S11**, 3,5-bis(trifluoromethyl)phenyl isocyanate (1.1 mL, 6.7 mmol, 1.0 equiv.), and CH_2Cl_2 (33 mL, 0.2 M). The resulting mixture was stirred at room temperature for 1 h. After evaporation, $\text{TsOH}\cdot\text{H}_2\text{O}$ (ca. 10 mol %) and $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (1/1, 67 mL, 0.1 M) were added to the crude mixture. After stirring at 40 °C for overnight, the reaction mixture was quenched by the addition of *sat.* NaHCO_3 *aq.* The aqueous layer was extracted with AcOEt three times. The combined organic layers were washed with H_2O and brine, dried over Na_2SO_4 , and evaporated in *vacuo*. Product **5** was obtained as a yellow solid (2.1 g, 3.73 mmol, 37% from **S9**) after purification by silica gel column chromatography (hexane/ AcOEt). mp 138-140 °C; IR (KBr) ν 3374, 1545, 1474, 1385, 1349, 1279, 1179, 1133, 884 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 5.18 (s, 1H), 6.20-6.50 (brs, 1H), 6.71 (s, 1H), 6.94-7.04 (m, 3H), 7.08 (d, $J = 8.2$ Hz, 1H), 7.22-7.37 (m, 4H), 7.47-7.64 (m, 4H), 7.87 (dd, $J = 8.5, 2.6$ Hz, 2H), 8.55 (s, 1H); ^{13}C NMR (100 MHz, CDCl_3) δ 111.2, 114.2, 115.9, 116.7, 118.3, 118.5, 118.9, 121.7, 123.4, 123.7, 124.1, 124.4, 126.0, 126.4, 127.3, 127.6, 128.8, 129.1, 129.3, 131.3, 131.5 (q, $J_{\text{CF}} = 33.7$ Hz), 133.5, 139.1, 141.9, 152.1, 152.8; ^{19}F NMR (400 MHz, CDCl_3) δ -63.0; HRMS (ESI): m/z calculated for $\text{C}_{29}\text{H}_{18}\text{O}_3\text{N}_2\text{F}_6\text{Na}$ [$\text{M}+\text{Na}^+$]: 579.1114, found: 579.1105.

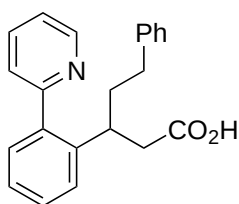
General procedure for $\text{Cp}^*\text{Rh(III)/BH}_3\cdot\text{SMe}_2$ -catalyzed C–H addition to α,β -unsaturated carboxylic acids

To a flame dried test tube were added carboxylic acid **2** (0.30 mmol) and toluene (1.5 mL). $\text{BH}_3\cdot\text{SMe}_2$ (1.0 M in CH_2Cl_2 , 60 μL , 20 mol %) was added to the reaction mixture at 0 °C. After the reaction mixture was stirred at room temperature for 1 h, the solvent was removed under vacuum. To the resulting residue were added $[\text{Cp}^*\text{RhCl}_2]_2$ (4.6 mg, 2.5 mol %), AgSbF_6 (10.3 mg, 10 mol %), **5** (33.4 mg, 20 mol %), DMF (0.6 mL), and **1** (1.0 equiv.) in a glove box. The test tube was capped and heated at 50 °C for 15 h with stirring. After the mixture was cooled to

room temperature, acetic acid was added and an insoluble solids were removed by filtration through a short pad of SiO₂. After evaporation, obtained crude mixture was purified by SiO₂ column chromatography (CH₂Cl₂/MeOH) to give **3**.

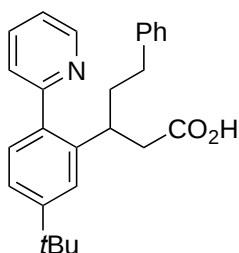
Characterization data of **3**

5-phenyl-3-(2-(pyridin-2-yl)phenyl)pentanoic acid (**3aa**)



81.4 mg (82%) in 0.30 mmol scale; 253.8 mg (77%) in 1.0 mmol scale; a colorless solid; mp 130–132 °C; IR (KBr) ν 2915, 2475, 1946, 1710, 1593, 1427, 1220, 1185, 768 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 1.78-1.92 (m, 2H), 2.11 (ddd, J = 14.9, 8.9, 6.7 Hz, 1H), 2.24 (ddd, J = 14.9, 8.7, 6.0 Hz, 1H), 2.96-3.04 (m, 2H), 3.63 (tt, J = 9.7, 6.3 Hz, 1H), 6.78-6.84 (m, 2H), 7.02-7.10 (m, 3H), 7.32-7.40 (m, 3H), 7.46-7.50 (m, 2H), 7.58 (d, J = 7.8 Hz, 1H), 7.93 (ddd, J = 7.8, 7.6, 1.6 Hz, 1H), 8.52 (d, J = 5.0 Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz) δ 33.0, 34.3, 39.4, 43.4, 122.5, 125.0, 125.7, 126.9, 127.0, 127.9, 128.2, 130.0, 130.1, 137.8, 138.9, 141.1, 141.5, 146.3, 158.1, 173.7; HRMS (ESI): m/z calculated for C₂₂H₂₂O₂N [M+H⁺]: 332.1645, found: 332.1642.

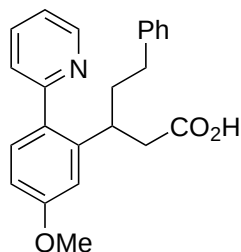
3-(5-(*tert*-butyl)-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ba**)



92.9 mg (80%); a yellow solid; mp 51–53 °C; IR (KBr) ν 3481, 2961, 1722, 1608, 1468, 792, 659 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 1.36 (s, 9H), 1.74-1.95 (m, 2H), 2.08-2.27 (m, 2H), 2.98 (d, J = 8.4 Hz, 2H), 3.55 (tt, J = 16.4, 8.4 Hz, 1H), 6.80-6.85 (m, 2H), 7.02-7.12 (m, 3H), 7.29 (d, J = 8.2 Hz, 1H), 7.33-7.42 (m, 2H), 7.46 (d, J = 1.9 Hz, 1H), 7.55 (d, J = 7.9 Hz, 1H), 7.94 (ddd, J = 7.9, 7.8, 1.6 Hz, 1H), 8.48 (d, J = 4.3 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz) δ 31.2, 33.0, 34.4, 34.8, 39.6, 43.4, 122.6, 123.6, 124.1, 125.3, 125.6, 127.9, 128.2, 129.8, 134.2,

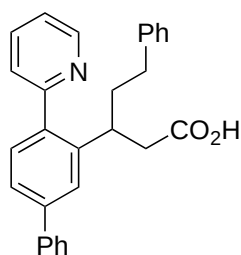
139.6, 141.1, 141.2, 145.5, 153.4, 157.4, 174.4; HRMS (ESI): m/z calculated for $C_{26}H_{30}O_2N$ $[M+H]^+$: 388.2271, found: 388.2272.

3-(5-methoxy-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ca**)



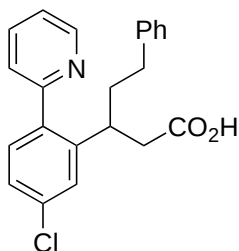
65.4 mg (60%); a yellow solid; mp 144–146 °C; IR (KBr) ν 2932, 1712, 1607, 1467, 1289, 1235, 753, 660 cm^{-1} ; 1H NMR ($CDCl_3$, 400 MHz) δ 1.74–1.93 (m, 2H), 2.12 (ddd, J = 14.9, 8.7, 6.9 Hz, 1H), 2.26 (ddd, J = 14.9, 8.7, 6.2 Hz, 1H), 2.89–3.02 (m, 2H), 3.54 (tt, J = 9.9, 5.9 Hz, 1H), 3.84 (s, 3H), 6.80–6.89 (m, 3H), 6.96 (d, J = 2.5 Hz, 1H), 7.03–7.11 (m, 3H), 7.30 (d, J = 8.6 Hz, 1H), 7.37 (dd, J = 7.0, 5.8 Hz, 1H), 7.54 (d, J = 7.9 Hz, 1H), 7.93 (ddd, J = 7.8, 7.8, 1.5 Hz, 1H), 8.46 (d, J = 4.5 Hz, 1H); ^{13}C NMR ($CDCl_3$, 100 MHz) δ 32.9, 34.3, 39.4, 43.5, 55.3, 112.4, 112.6, 122.4, 125.3, 125.7, 127.9, 128.1, 129.5, 131.6, 139.8, 141.0, 143.4, 145.2, 157.1, 161.1, 174.3; HRMS (ESI): m/z calculated for $C_{23}H_{24}O_3N$ $[M+H]^+$: 362.1751, found: 362.1751.

5-phenyl-3-(4-(pyridin-2-yl)-[1,1'-biphenyl]-3-yl)pentanoic acid (**3da**)



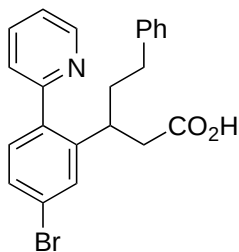
80.4 mg (66%); a yellow solid; mp 59–61 °C; IR (KBr) ν 3436, 2922, 1712, 1598, 1468, 1287, 764, 698 cm^{-1} ; 1H NMR ($CDCl_3$, 400 MHz) δ 1.82–1.98 (m, 2H), 2.17 (ddd, J = 14.8, 8.2, 8.2 Hz, 1H), 2.29 (ddd, J = 14.8, 6.6, 6.6 Hz, 1H), 2.98–3.10 (m, 2H), 3.65 (tt, J = 9.0, 7.1 Hz, 1H), 6.83 (d, J = 6.6 Hz, 2H), 7.01–7.11 (m, 3H), 7.34–7.49 (m, 5H), 7.51–7.67 (m, 5H), 7.92 (dd, J = 7.8, 7.7 Hz, 1H), 8.52 (d, J = 4.3 Hz, 1H); ^{13}C NMR ($CDCl_3$, 100 MHz) δ 33.0, 34.3, 39.4, 43.4, 122.6, 125.1, 125.68, 125.74, 127.2, 127.8, 127.9, 128.2, 128.8, 130.6, 136.4, 139.2, 140.2, 141.0, 142.0, 143.0, 146.1, 157.5, 173.9; One of the aromatic signals was missing probably due to overlapping; HRMS (ESI): m/z calculated for $C_{28}H_{26}O_2N$ $[M+H]^+$: 408.1958, found: 408.1957.

3-(5-chloro-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ea**)



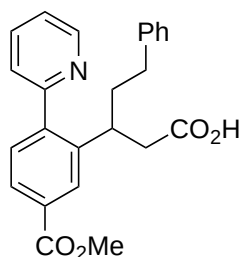
53.6 mg (49%); a yellow solid; mp 46–48 °C; IR (KBr) ν 2926, 2469, 1709, 1600, 1467, 1274, 756, 699 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.76-1.97 (m, 2H), 2.11-2.22 (m, 1H), 2.25-2.34 (m, 1H), 2.87-3.03 (m, 2H), 3.53-3.65 (m, 1H), 6.84 (dd, $J = 7.7, 1.8$ Hz, 2H), 7.03-7.13 (m, 3H), 7.29-7.35 (m, 2H), 7.37-7.42 (m, 1H), 7.44 (s, 1H), 7.51 (dd, $J = 7.9, 0.9$ Hz, 1H), 7.87-7.96 (m, 1H), 8.51-8.56 (m, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 32.9, 34.4, 39.1, 43.0, 122.8, 125.0, 125.8, 127.2, 127.9, 128.2, 131.4, 136.1, 136.3, 138.9, 140.8, 143.6, 146.7, 157.0, 173.4; One of the aromatic signals was missing probably due to overlapping; HRMS (ESI): m/z calculated for $\text{C}_{22}\text{H}_{21}\text{O}_2\text{NCl}$ $[\text{M}+\text{H}^+]$: 366.1255, found: 366.1257.

3-(5-bromo-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3fa**)



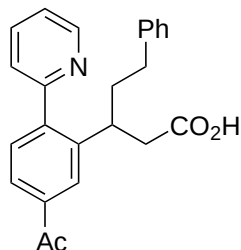
52.5 mg (43%); an orange solid; mp 98–100 °C; IR (KBr) ν 2926, 2484, 1712, 1598, 1466, 1245, 1026, 788 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.75-1.96 (m, 2H), 2.15 (ddd, $J = 14.9, 8.6, 6.8$ Hz, 1H), 2.29 (ddd, $J = 14.9, 8.5, 6.4$ Hz, 1H), 2.88-3.04 (m, 2H), 3.52-3.63 (m, 1H), 6.83 (dd, $J = 7.5, 1.7$ Hz, 2H), 7.03-7.12 (m, 3H), 7.22-7.27 (m, 1H), 7.40 (dd, $J = 5.7, 5.4$ Hz, 1H), 7.46-7.54 (m, 2H), 7.59 (d, $J = 1.3$ Hz, 1H), 7.88-7.96 (m, 1H), 8.53 (ddd, $J = 5.0, 1.7, 0.8$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 32.9, 34.4, 39.1, 42.9, 122.8, 124.4, 124.9, 125.8, 127.9, 128.2, 130.2, 131.6, 136.9, 138.8, 140.8, 143.9, 146.8, 157.1, 173.4; One of the aromatic signals was missing probably due to overlapping; HRMS (ESI): m/z calculated for $\text{C}_{22}\text{H}_{21}\text{O}_2\text{NBr}$ $[\text{M}+\text{H}^+]$: 410.0750, found: 410.0751.

3-(5-(methoxycarbonyl)-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ga**)



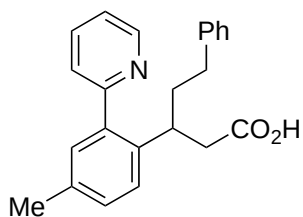
73.0 mg (62%); an orange solid; mp 152–153 °C; IR (KBr) ν 2921, 1718, 1596, 1431, 1292, 1251, 1173, 760 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.81–1.99 (m, 2H), 2.08–2.20 (m, 1H), 2.21–2.31 (m, 1H), 3.02 (d, J = 7.9 Hz, 2H), 3.64 (tt, J = 7.4, 7.3 Hz, 1H), 3.96 (s, 3H), 6.83 (d, J = 7.3 Hz, 2H), 7.03–7.12 (m, 3H), 7.39–7.48 (m, 2H), 7.56 (d, J = 7.9 Hz, 1H), 7.94 (dd, J = 7.9, 7.9 Hz, 1H), 8.01 (dd, J = 8.0, 1.6 Hz, 1H), 8.15 (d, J = 1.6 Hz, 1H), 8.56 (d, J = 5.0 Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 33.0, 34.6, 38.9, 42.7, 52.3, 123.0, 125.1, 125.7, 127.87, 127.91, 128.2, 130.2, 131.39, 131.45, 138.7, 140.9, 142.2, 142.3, 146.9, 157.2, 166.4, 173.9; HRMS (ESI): m/z calculated for $\text{C}_{24}\text{H}_{24}\text{O}_4\text{N}$ [$\text{M}+\text{H}^+$]: 390.1700, found: 390.1699.

3-(5-acetyl-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ha**)



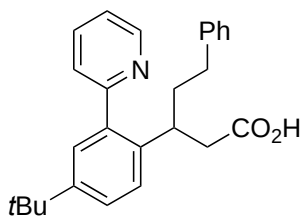
88.3 mg (79%); a yellow solid; mp 152–154 °C; IR (KBr) ν 2929, 2513, 1715, 1675, 1594, 1284, 1227, 1175, 699 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.81–2.01 (m, 2H), 2.17 (ddd, J = 14.8, 8.5, 6.9 Hz, 1H), 2.27 (ddd, 14.8, 8.8, 6.3 Hz, 1H), 2.65 (s, 3H), 2.92–3.03 (m, 2H), 3.57–3.70 (m, 1H), 6.84 (d, J = 7.3 Hz, 2H), 7.03–7.13 (m, 3H), 7.36–7.43 (m, 1H), 7.46 (d, J = 8.1 Hz, 1H), 7.53 (d, J = 7.9 Hz, 1H), 7.90 (d, J = 7.8 Hz, 2H), 8.05 (s, 1H), 8.57 (d, J = 5.0 Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 26.7, 33.0, 34.5, 39.0, 42.8, 123.0, 125.0, 125.8, 126.77, 126.84, 127.9, 128.2, 130.5, 138.0, 138.6, 140.9, 142.3, 142.5, 147.1, 157.2, 173.6, 197.5; HRMS (ESI): m/z calculated for $\text{C}_{24}\text{H}_{24}\text{O}_3\text{N}$ [$\text{M}+\text{H}^+$]: 374.1751, found: 374.1751.

3-(4-methyl-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ia**)



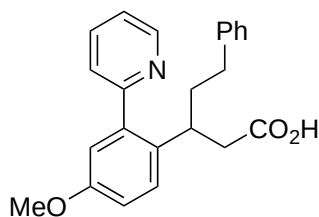
54.3 mg (52%); a yellow solid; mp 123–125 °C; IR (KBr) ν 2929, 2449, 1736, 1704, 1597, 1234, 1008 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.72–1.93 (m, 2H), 2.10 (ddd, $J = 14.4, 9.0, 7.2$ Hz, 1H), 2.23 (ddd, $J = 14.4, 8.5, 5.8$ Hz, 1H), 2.38 (s, 3H), 2.90–2.99 (m, 2H), 3.50 (tt, $J = 15.3, 6.3$ Hz, 1H), 6.83 (d, $J = 6.7$ Hz, 2H), 7.02–7.11 (m, 3H), 7.17 (s, 1H), 7.30 (d, $J = 8.1$ Hz, 1H), 7.36 (d, $J = 7.6$ Hz, 1H), 7.41 (dd, $J = 7.0, 5.6$ Hz, 1H), 7.58 (d, $J = 8.1$ Hz, 1H), 7.95 (td, $J = 7.7, 1.6$ Hz, 1H), 8.50 (d, $J = 4.9$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 20.9, 32.9, 33.9, 39.4, 43.5, 122.7, 125.3, 125.7, 126.9, 127.9, 128.1, 130.6, 131.2, 136.6, 136.9, 138.4, 139.6, 141.1, 145.6, 157.6, 174.3; HRMS (ESI): m/z calculated for $\text{C}_{23}\text{H}_{24}\text{O}_2\text{N}$ $[\text{M}+\text{H}^+]$: 346.1802, found: 346.1802.

3-(4-(*tert*-butyl)-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ja**)



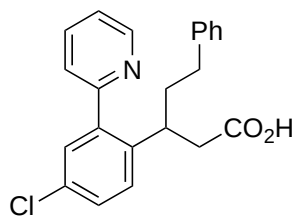
62.0 mg (53%); a yellow solid; mp 156–158 °C; IR (KBr) ν 2962, 1711, 1595, 1469, 1363, 1254, 757 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.33 (s, 9H), 1.76–1.93 (m, 2H), 2.14 (ddd, $J = 15.0, 8.5, 6.5$ Hz, 1H), 2.26 (ddd, $J = 15.0, 8.4, 6.6$ Hz, 1H), 2.93–2.99 (m, 2H), 3.49 (tt, $J = 15.4, 6.1$ Hz, 1H), 6.84 (dd, $J = 8.0, 1.9$ Hz, 2H), 7.02–7.11 (m, 3H), 7.32 (d, $J = 2.1$ Hz, 1H), 7.39 (d, $J = 8.3$ Hz, 1H), 7.43 (ddd, $J = 7.6, 5.2, 1.1$ Hz, 1H), 7.51 (dd, $J = 8.3, 2.1$ Hz, 1H), 7.59 (d, $J = 8.0$ Hz, 1H), 7.98 (td, $J = 7.8, 1.7$ Hz, 1H), 8.50 (d, $J = 4.3$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 31.2, 33.0, 33.9, 34.5, 39.3, 43.4, 122.7, 125.4, 125.7, 126.7, 126.9, 127.7, 127.9, 128.2, 136.3, 138.4, 139.8, 141.2, 145.4, 149.7, 157.9, 174.4; HRMS (ESI): m/z calculated for $\text{C}_{26}\text{H}_{30}\text{O}_2\text{N}$ $[\text{M}+\text{H}^+]$: 388.2271, found: 388.2271.

3-(4-methoxy-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ka**)



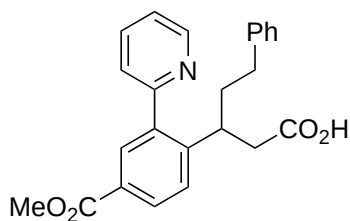
89.0 mg (82%); a yellow solid; mp 48–50 °C; IR (KBr) ν 2940, 1712, 1608, 1469, 1299, 1226, 1034, 660 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.71–1.92 (m, 2H), 2.11 (ddd, $J = 15.0, 9.1, 6.7$ Hz, 1H), 2.25 (ddd, $J = 15.0, 9.1, 6.0$ Hz, 1H), 2.84–2.97 (m, 2H), 3.46 (tt, $J = 9.9, 6.0$ Hz, 1H), 3.82 (s, 3H), 6.84 (dd, $J = 8.0, 1.8$ Hz, 2H), 6.87 (d, $J = 2.7$ Hz, 1H), 7.00–7.11 (m, 4H), 7.33–7.41 (m, 2H), 7.54 (dd, $J = 7.8, 0.6$ Hz, 1H), 7.91 (td, $J = 7.9, 1.6$ Hz, 1H), 8.51 (dd, $J = 5.1, 0.8$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 33.0, 33.8, 39.4, 43.4, 55.4, 115.2, 115.9, 122.7, 125.1, 125.6, 127.9, 128.1, 133.3, 138.5, 139.2, 141.2, 146.0, 157.6, 158.0, 174.1; One of the aromatic signals was missing probably due to overlapping; HRMS (ESI): m/z calculated for $\text{C}_{23}\text{H}_{24}\text{O}_3\text{N}$ $[\text{M}+\text{H}^+]$: 362.1751, found: 362.1750.

3-(4-chloro-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3la**)



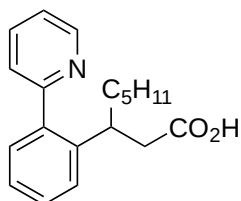
87.6 mg (80%); a yellow solid; mp 145.5–146 °C; IR (KBr) ν 2914, 2469, 1721, 1598, 1472, 1210, 1180, 1007, 698 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.72–1.84 (m, 1H), 1.84–1.96 (m, 1H), 2.16 (ddd, $J = 14.9, 8.6, 8.6$ Hz, 1H), 2.26 (ddd, $J = 14.9, 9.3, 6.1$ Hz, 1H), 2.88 (d, $J = 7.3$ Hz, 2H), 3.45–3.57 (m, 1H), 6.85 (d, $J = 7.2$ Hz, 2H), 7.03–7.13 (m, 3H), 7.31–7.51 (m, 5H), 7.86 (t, $J = 7.7$ Hz, 1H), 8.55 (d, $J = 4.8$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 32.9, 34.2, 39.0, 42.8, 123.0, 125.0, 125.7, 127.9, 128.2, 128.4, 129.8, 130.0, 132.4, 138.8, 139.6, 140.1, 140.9, 146.9, 156.7, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{22}\text{H}_{21}\text{O}_2\text{NCl}$ $[\text{M}+\text{H}^+]$: 366.1255, found: 366.1258.

3-(4-(methoxycarbonyl)-2-(pyridin-2-yl)phenyl)-5-phenylpentanoic acid (**3ma**)



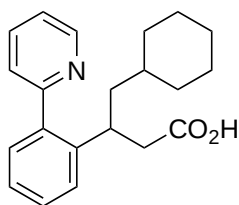
81.9 mg (70%); a yellow solid; mp 120.5–121.5 °C; IR (KBr) ν 2923, 2483, 1721, 1597, 1436, 1314, 1252, 1113, 1010, 769 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.77-1.99 (m, 2H), 2.15 (ddd, $J = 14.2, 9.1, 6.6$ Hz, 1H), 2.24 (ddd, $J = 14.2, 9.1, 6.1$ Hz, 1H), 2.96 (d, $J = 8.0$ Hz, 2H), 3.58-3.71 (m, 1H), 3.92 (s, 3H), 6.84 (d, $J = 6.2$ Hz, 2H), 7.03-7.12 (m, 3H), 7.39 (dd, $J = 7.4, 5.0$ Hz, 1H), 7.56 (dd, $J = 9.2, 9.2$ Hz, 2H), 7.91 (t, $J = 7.7$ Hz, 1H), 8.06 (s, 1H), 8.12 (d, $J = 8.2$ Hz, 1H), 8.56 (d, $J = 4.9$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 32.9, 34.8, 39.0, 42.7, 52.2, 122.9, 125.1, 125.7, 127.2, 127.9, 128.2, 128.7, 130.7, 131.3, 138.3, 138.8, 140.8, 146.8, 147.0, 157.1, 166.3, 173.6; HRMS (ESI): m/z calculated for $\text{C}_{24}\text{H}_{24}\text{O}_4\text{N}$ $[\text{M}+\text{H}^+]$: 390.1700, found: 390.1702.

3-(2-(pyridin-2-yl)phenyl)octanoic acid (**3ab**)



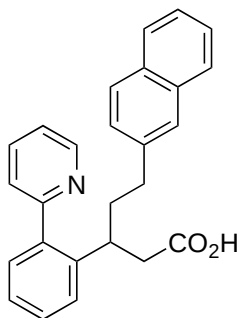
52.9 mg (59%); an orange solid; mp 115–115.5 °C; IR (KBr) ν 2955, 2922, 2856, 2488, 1709, 1594, 1428, 1219, 1004, 768 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 0.68 (t, $J = 7.2$ Hz, 3H), 0.73-1.12 (m, 6H), 1.48 (dt, $J = 7.4, 7.2$ Hz, 2H), 2.96 (d, $J = 7.9$ Hz, 2H), 3.48-3.67 (m, 1H), 7.28-7.37 (m, 2H), 7.40-7.48 (m, 3H), 7.62 (d, $J = 7.9$ Hz, 1H), 7.97 (dd, $J = 7.9, 7.5$ Hz, 1H), 8.62 (d, $J = 4.5$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 13.8, 22.2, 26.5, 31.2, 34.5, 37.9, 43.6, 122.5, 125.0, 126.7, 126.9, 129.8, 130.0, 137.7, 139.0, 141.8, 146.3, 158.4, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{19}\text{H}_{24}\text{O}_2\text{N}$ $[\text{M}+\text{H}^+]$: 298.1802, found: 298.1799.

4-cyclohexyl-3-(2-(pyridin-2-yl)phenyl)butanoic acid (**3ac**)



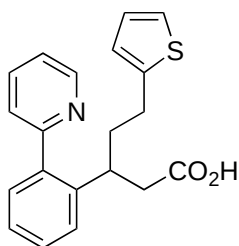
77.4 mg (80%); a yellow solid; mp 164–166 °C; IR (KBr) ν 2920, 2849, 2522, 1704, 1604, 1449, 1248, 1002, 760 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 0.43–0.57 (m, 1H), 0.62–0.74 (m, 2H), 0.76–1.09 (m, 5H), 1.22–1.57 (m, 7H), 2.87 (dd, $J = 17.0, 5.4$ Hz, 1H), 2.96 (dd, $J = 17.0, 11.0$ Hz, 1H), 3.60 (tt, $J = 10.8, 5.4$ Hz, 1H), 7.31–7.39 (m, 2H), 7.42–7.52 (m, 2H), 7.53–7.60 (m, 1H), 7.68 (d, $J = 8.1$ Hz, 1H), 8.08 (ddd, $J = 7.6, 7.6, 1.3$ Hz, 1H), 8.71 (d, $J = 3.6$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 25.8, 26.0, 26.3, 31.9, 32.5, 33.7, 34.7, 43.6, 46.0, 123.0, 125.5, 126.7, 126.8, 127.1, 129.8, 130.5, 139.9, 142.1, 145.6, 157.8, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{21}\text{H}_{16}\text{O}_2\text{N}$ [$\text{M}+\text{H}^+$]: 324.1958, found: 324.1957.

5-(naphthalen-2-yl)-3-(2-(pyridin-2-yl)phenyl)pentanoic acid (**3ad**)



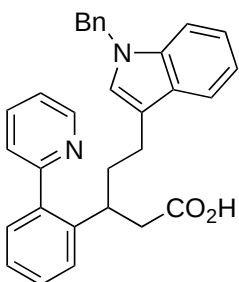
86.1 mg (75%); a yellow solid; mp 145–146 °C; IR (KBr) ν 2925, 2491, 1712, 1598, 1427, 1223, 758 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.92 (ddt, $J = 14.4, 7.9, 7.3$ Hz, 1H), 2.05 (ddt, $J = 14.4, 7.2, 6.9$ Hz, 1H), 2.37 (dt, $J = 14.4, 7.3$ Hz, 1H), 2.51 (dt, $J = 14.4, 7.2$ Hz, 1H), 2.93–3.08 (m, 2H), 3.49–3.61 (m, 1H), 7.00 (dd, $J = 8.4, 1.6$ Hz, 1H), 7.06–7.12 (m, 1H), 7.23 (s, 1H), 7.29–7.36 (m, 3H), 7.38–7.45 (m, 2H), 7.51 (d, $J = 3.1$ Hz, 2H), 7.53–7.59 (m, 2H), 7.59–7.64 (m, 1H), 7.69–7.76 (m, 1H), 8.32–8.38 (m, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 32.8, 33.7, 38.6, 43.2, 122.2, 125.0, 125.1, 125.8, 126.0, 126.6, 126.8, 126.9, 127.3, 127.4, 127.7, 130.0, 130.2, 131.8, 133.3, 137.2, 138.4, 138.8, 141.4, 145.5, 157.2, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{26}\text{H}_{24}\text{O}_2\text{N}$ [$\text{M}+\text{H}^+$]: 382.1802, found: 382.1799.

3-(2-(pyridin-2-yl)phenyl)-5-(thiophen-2-yl)pentanoic acid (**3ae**)



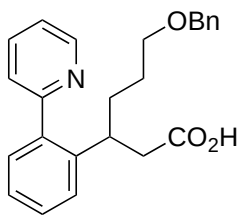
52.0 mg (51%); a colorless solid; mp 150-151 °C; IR (KBr) ν 2916, 2467, 1708, 1593, 1428, 1286, 1223, 1188, 769, 702 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.82-2.01 (m, 2H), 2.36 (dt, J = 14.7, 7.0 Hz, 1H), 2.48 (dt, J = 14.7, 6.6 Hz, 1H), 2.97 (d, J = 8.3 Hz, 2H), 3.59-3.71 (m, 1H), 6.41-6.45 (m, 1H), 6.71 (dd, J = 5.1, 3.4 Hz, 1H), 6.95 (dd, J = 5.2, 1.2 Hz, 1H), 7.31-7.42 (m, 3H), 7.44-7.51 (m, 2H), 7.55 (d, J = 7.8 Hz, 1H), 7.88-7.96 (m, 1H), 8.54 (ddd, J = 5.2, 1.7, 0.9 Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 27.5, 34.4, 39.7, 43.6, 123.0, 123.2, 124.3, 125.6, 126.8, 127.3, 130.4, 130.6, 137.7, 139.6, 141.4, 144.3, 146.3, 158.0, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{20}\text{H}_{19}\text{O}_2\text{NSNa}$ [$\text{M}+\text{Na}^+$]: 360.1029, found: 360.1025.

5-(1-benzyl-1*H*-indol-3-yl)-3-(2-(pyridin-2-yl)phenyl)pentanoic acid (**3af**)



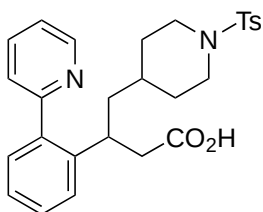
89.9 mg (65%); a yellow solid; mp 80-82 °C; IR (KBr) ν 2925, 1728, 1596, 1468, 1332, 1243, 740 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 1.84-2.04 (m, 2H), 2.27 (dt, J = 14.9, 7.6 Hz, 1H), 2.40 (dt, J = 14.9, 6.7 Hz, 1H), 2.92-3.03 (m, 2H), 3.56-3.68 (m, 1H), 5.09 (s, 2H), 6.53 (s, 1H), 6.96-7.03 (m, 3H), 7.07-7.34 (m, 10H), 7.45-7.51 (m, 2H), 7.58-7.66 (m, 1H), 8.36 (d, J = 5.2 Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 22.2, 34.1, 37.8, 43.1, 49.6, 109.5, 114.3, 118.69, 118.73, 121.6, 122.4, 125.0, 125.1, 126.7, 126.8, 127.0, 127.5, 127.7, 128.6, 129.9, 130.3, 136.4, 137.0, 137.5, 139.3, 141.8, 145.3, 157.2, 174.2; HRMS (ESI): m/z calculated for $\text{C}_{31}\text{H}_{29}\text{O}_2\text{N}_2$ [$\text{M}+\text{H}^+$]: 461.2224, found: 461.2220.

6-(benzyloxy)-3-(2-(pyridin-2-yl)phenyl)hexanoic acid (**3ag**)



77.1 mg (68%); a yellow solid; mp 110-112 °C; IR (KBr) ν 2938, 2860, 2488, 1704, 1428, 1362, 1222, 1094, 769 cm^{-1} ; ^1H NMR (CDCl_3 , 500 MHz) δ 1.05-1.23 (m, 2H), 1.50-1.59 (m, 1H), 1.60-1.68 (m, 1H), 2.94 (d, J = 8.0 Hz, 2H), 3.11-3.17 (m, 2H), 3.50-3.58 (m, 1H), 4.27 (d, J = 12.0 Hz, 1H), 4.30 (d, J = 12.0 Hz, 1H), 7.18-7.22 (m, 2H), 7.23-7.27 (m, 1H), 7.27-7.34 (m, 4H), 7.38-7.47 (m, 3H), 7.58 (dd, J = 8.0, 1.1 Hz, 1H), 7.90-7.96 (m, 1H), 8.58 (d, J = 4.6 Hz, 1H); ^{13}C NMR (CDCl_3 , 125 MHz) δ 27.3, 34.5, 34.9, 43.5, 69.7, 72.8, 122.7, 125.2, 126.8, 126.9, 127.50, 127.52, 128.3, 130.0, 130.2, 137.4, 138.2, 139.4, 141.5, 146.1, 158.0, 173.8; HRMS (ESI): m/z calculated for $\text{C}_{24}\text{H}_{26}\text{O}_3\text{N}$ [$\text{M}+\text{H}^+$]: 376.1907, found: 376.1902.

3-(2-(pyridin-2-yl)phenyl)-4-(1-(tosylpiperidin-4-yl)butyl)butanoic acid (**3ah**)



98.0 mg (68%); a yellow solid; mp 86-88 °C; IR (KBr) ν 2914, 1731, 1597, 1468, 1336, 1248, 1163, 934, 726 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 0.63-0.76 (m, 1H), 0.81-0.93 (m, 2H), 1.05 (dddd, J = 12.0, 12.0, 12.0, 4.4 Hz, 1H), 1.24-1.40 (m, 2H), 1.43-1.54 (m, 1H), 1.86 (dt, J = 11.6, 7.6 Hz, 1H), 2.01 (ddd, J = 11.7, 11.4, 2.2 Hz, 1H), 2.41 (s, 3H), 2.75-2.94 (m, 2H), 3.40-3.67 (m, 3H), 7.25-7.46 (m, 7H), 7.48-7.58 (m, 3H), 7.88-7.98 (m, 1H), 8.62 (d, J = 5.0 Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 21.4, 30.7, 31.7, 32.2, 43.3, 44.3, 45.7, 45.9, 122.8, 124.9, 126.8, 126.9, 127.5, 129.5, 130.0, 130.1, 133.3, 137.6, 138.9, 141.3, 143.3, 146.7, 158.1, 173.7; One of the aliphatic signals was missing probably due to overlapping; HRMS (ESI): m/z calculated for $\text{C}_{27}\text{H}_{31}\text{O}_4\text{N}_2\text{S}$ [$\text{M}+\text{H}^+$]: 479.1999, found: 479.1996.

A preparative scale synthesis of **3aa**

To a flame dried 20mL flask equipped with a stirring bar were added **2a** (176 mg, 1.0 mmol, azeotroped with toluene before use) and toluene (5 mL). $\text{BH}_3\cdot\text{SMe}_2$ (1.0 M in CH_2Cl_2 , 200 μL ,

20 mol %) was added to the reaction mixture at 0 °C. After the reaction mixture was stirred at room temperature for 1 h, the solvent was removed under vacuum, and the mixture was brought into a glove box. To another screw-capped test tube were added **1a** (142 μ L, 1.0 mmol, azeotroped with toluene before use), [Cp*RhCl₂]₂ (15.5. mg, 2.5 mol %), AgSbF₆ (34.3 mg, 10 mol %), **5** (111 mg, 20 mol %), and the above-prepared mixture dissolved in DMF (2 mL). The test tube was capped and heated at 50 °C for 15 h with stirring. After the mixture was cooled to room temperature, acetic acid was added and an insoluble solids were removed by filtration through a short pad of SiO₂. After evaporation, obtained crude mixture was purified by SiO₂ column chromatography (CH₂Cl₂/MeOH = 100/1; 40/1; 30/1) to give **3aa** (253.8 mg, 77%).

Computational study on acyloxyborane intermediate

We prepared a initial structure of acyloxyborane intermediate **IV** (Fig. 4) without hydrogen bonding between the urea and carbonyl moiety. A conformation search was run using Grimme's crest 2.11.1 and xtb 6.4.1 programs¹² (GFN2-xTB, iMTD-GC, 10 kcal/mol energy window, and default settings for others), and 432 conformers were generated. There were 42 conformations within a 3.0 kcal/mol window, all of which showed similar hydrogen bonding between the urea, carbonyl, and phenolic oxygen. As a representative structure, we selected the most stable conformation and its geometry was further optimized at the M06-2X/def2-SVP+SMD(DMF) level of theory¹³ using Gaussian 16 Rev.C.01.¹⁴ The optimized structure has no imaginary frequencies in the vibrational calculation at the same level. The cartesian coordinates of the initial structure for the conformation search and the optimized structure are shown below.

Initial structure

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C	-1.83537466	1.14319839	-2.03985144
C	-2.85469293	1.89312901	-2.57113621
C	-5.24695024	2.18825284	-3.03666493
C	-6.52827972	1.71288538	-3.00820680
C	-6.78319554	0.41781142	-2.53533195
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C	-4.19200985	-4.91763556	-1.07487600
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C	-5.21099519	-3.27389769	-4.79935048
C	-5.62040290	-4.57630020	-4.47870262

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Optimized structure

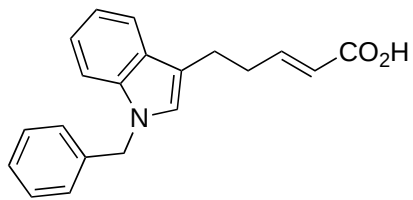
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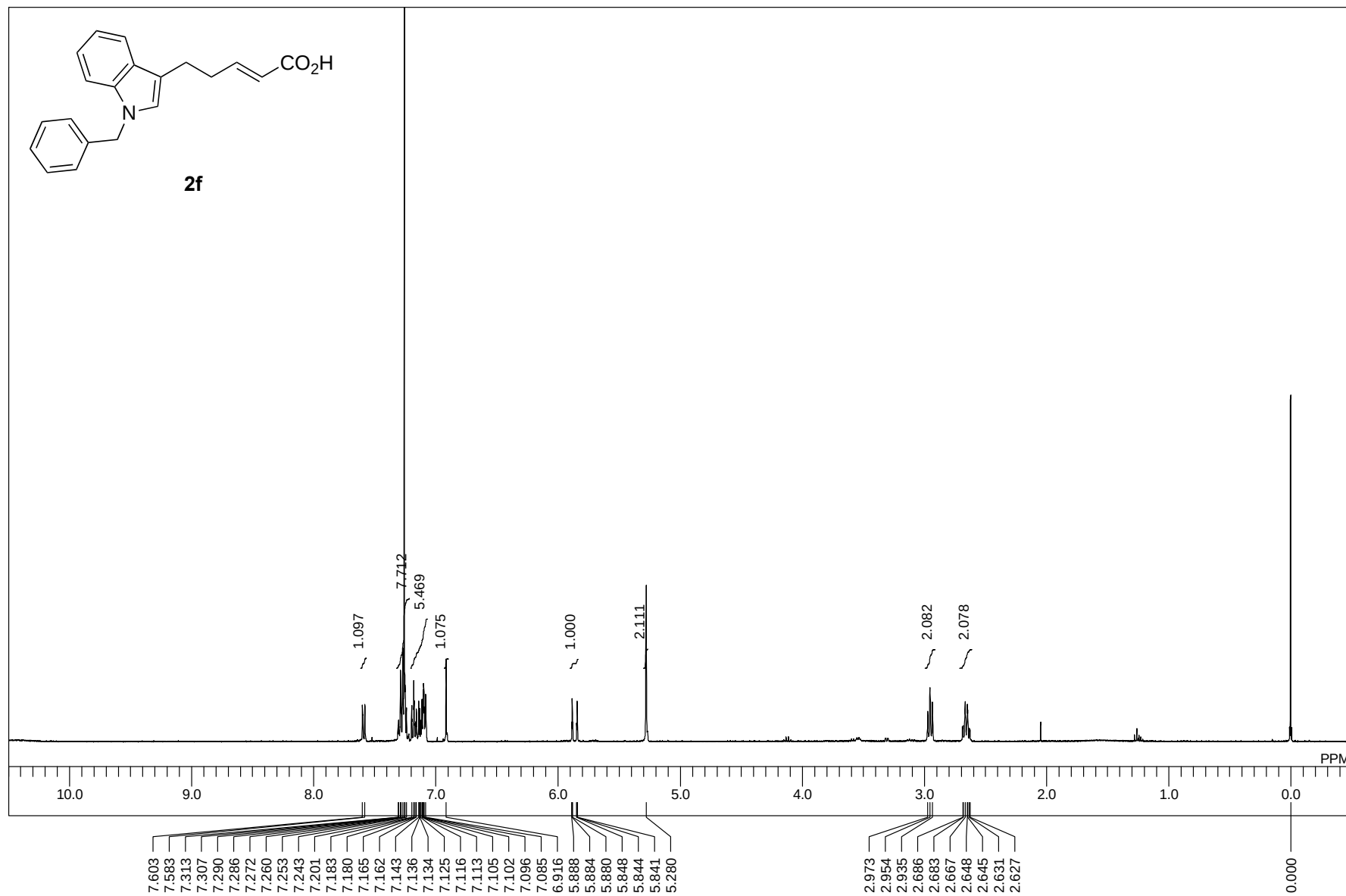
References

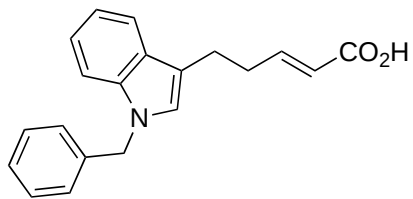
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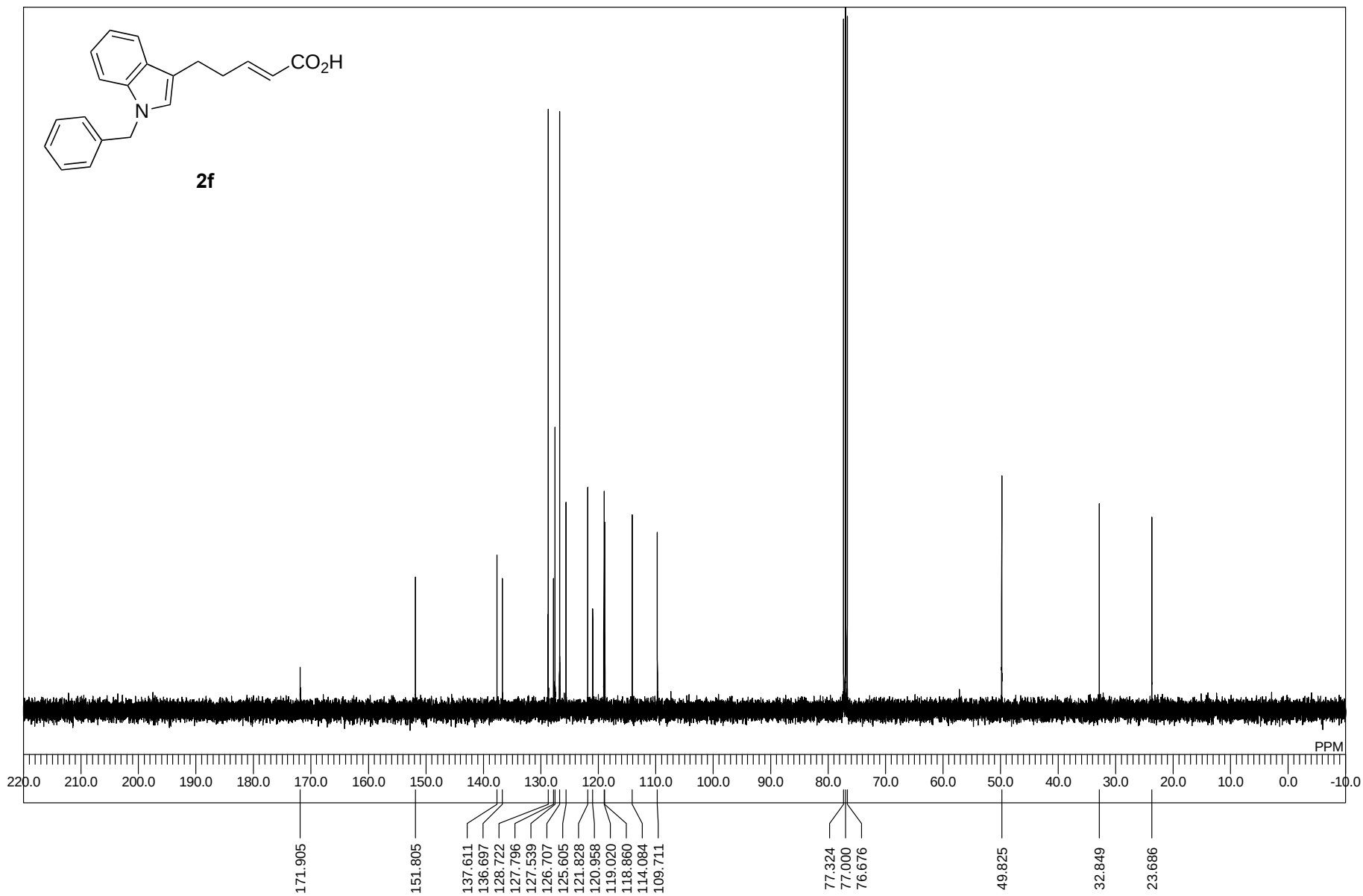


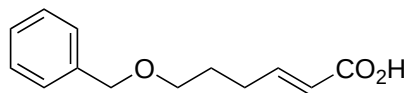
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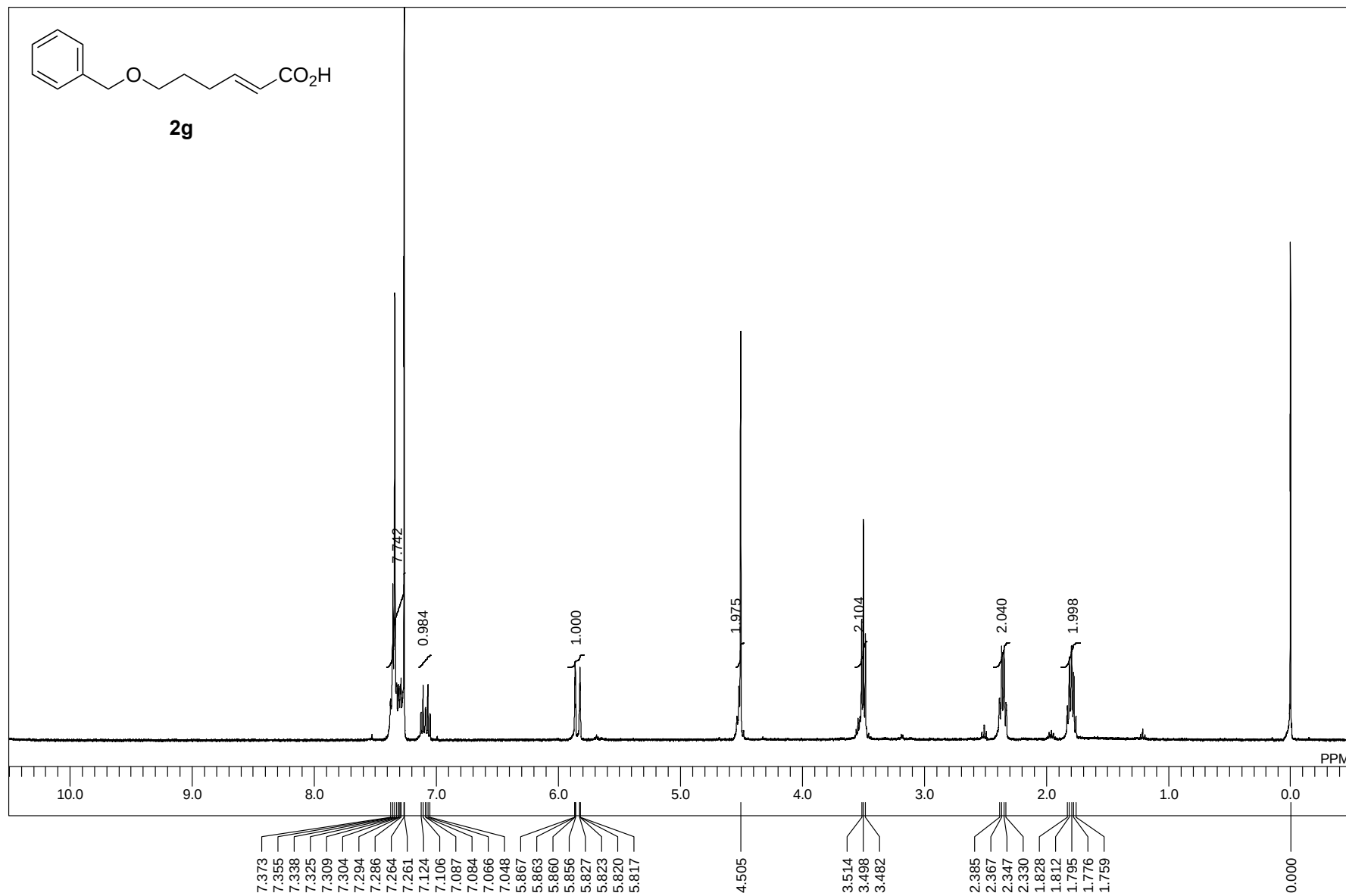


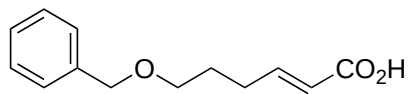
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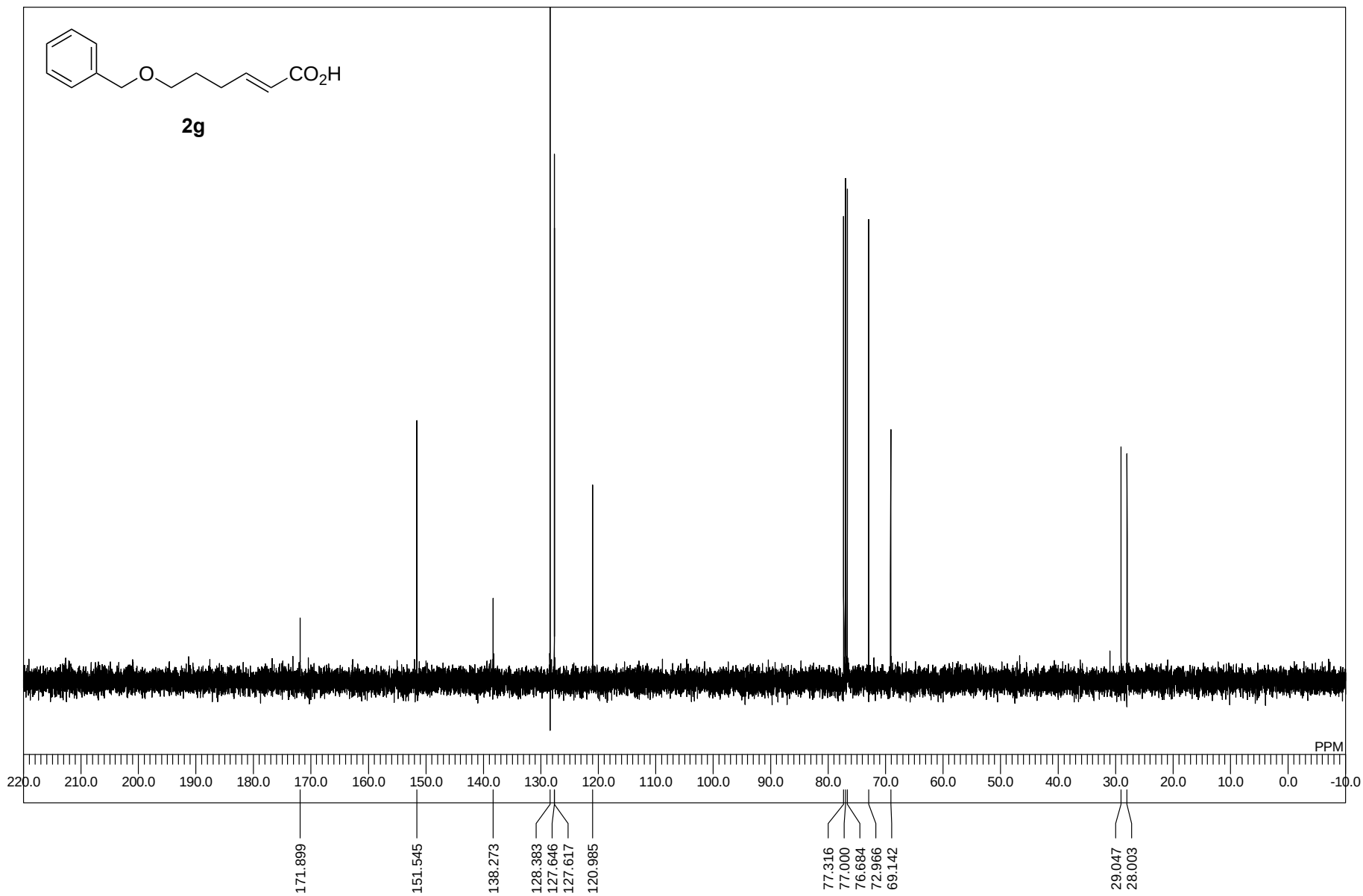


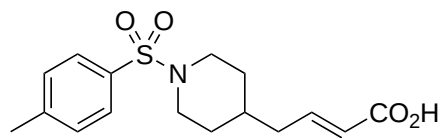
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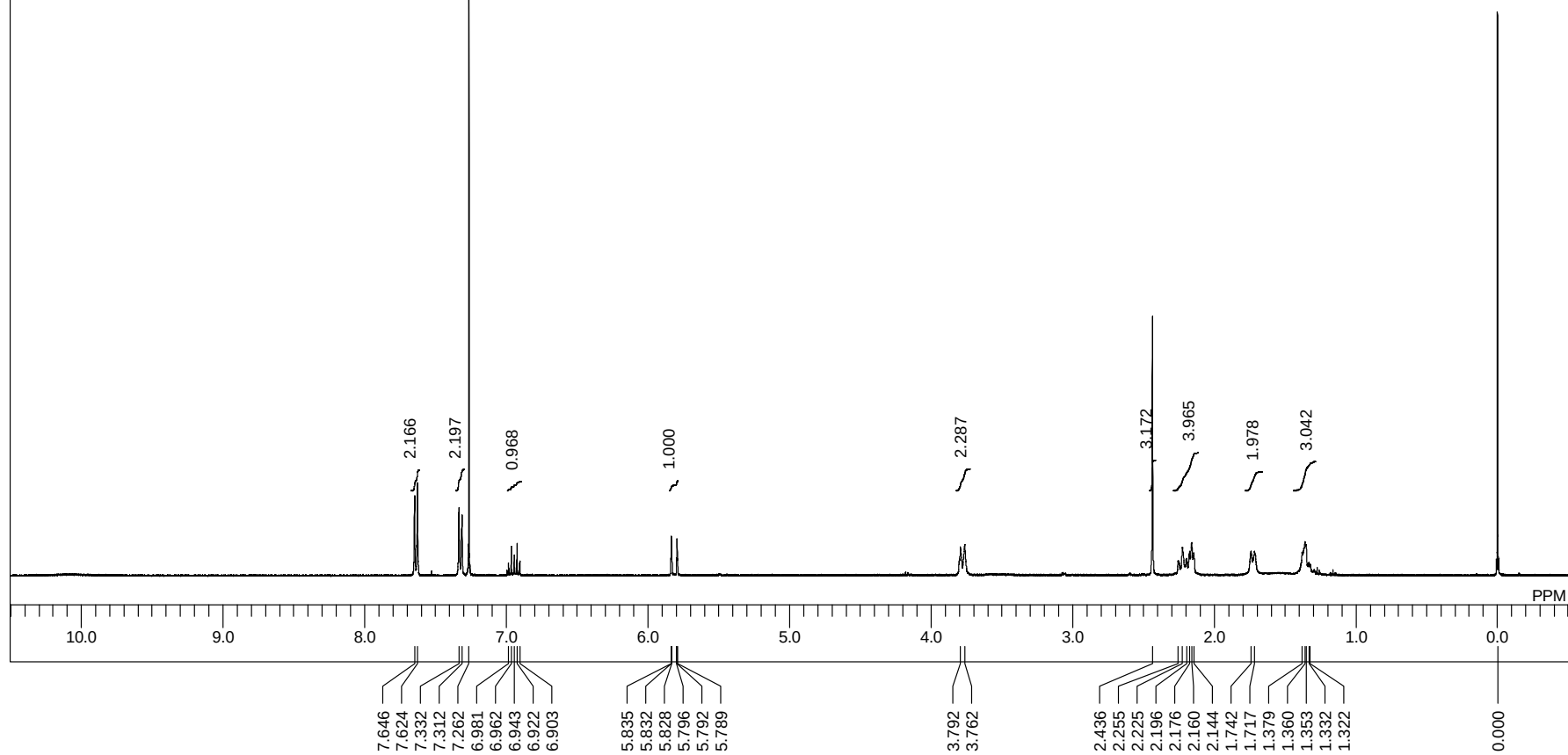


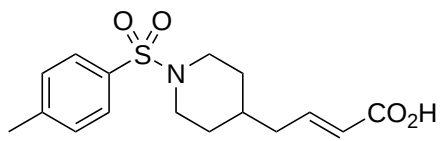
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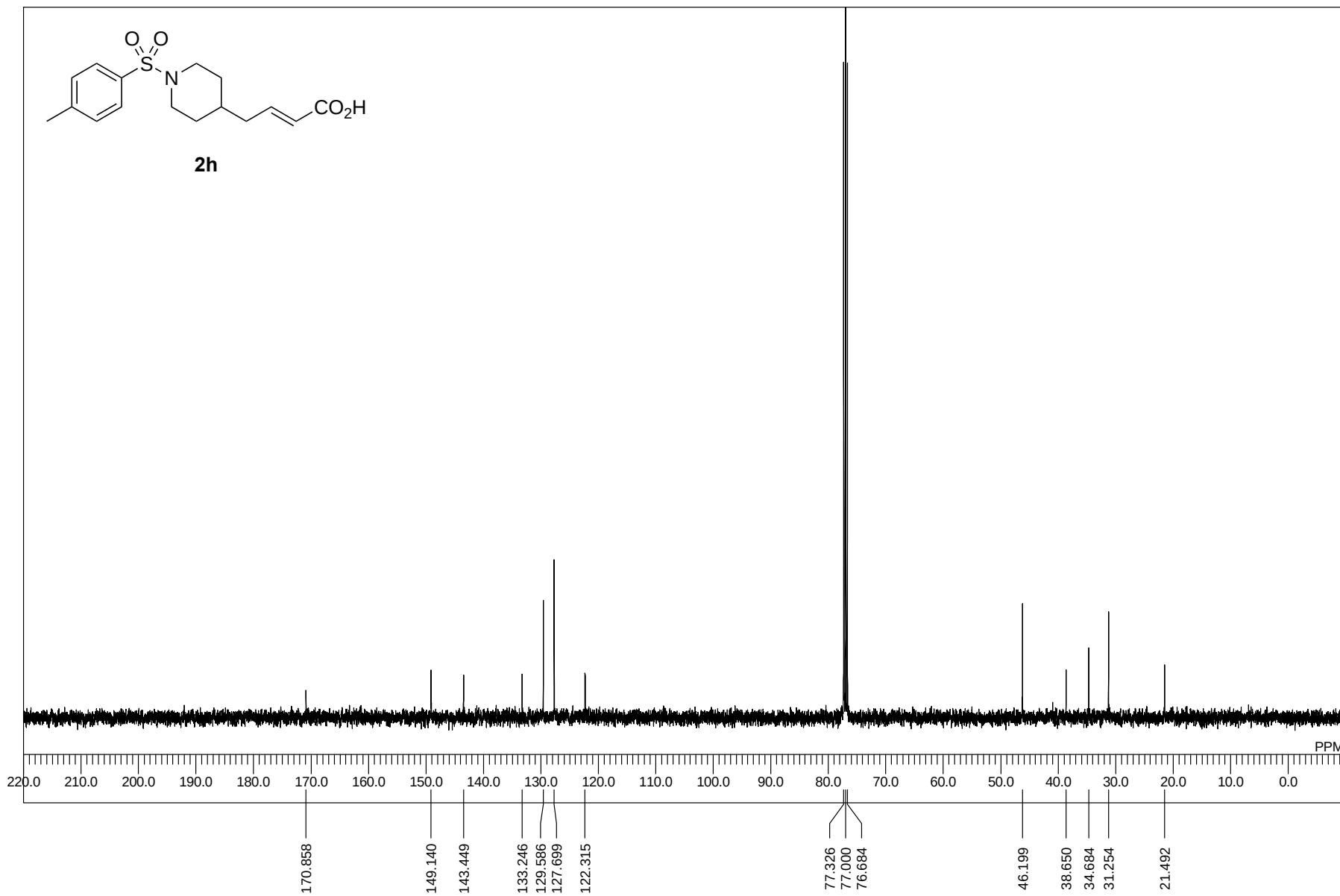


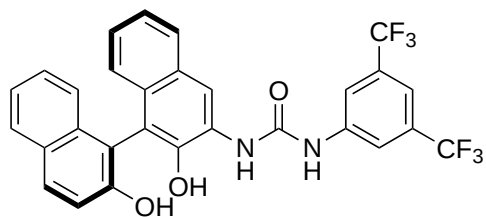
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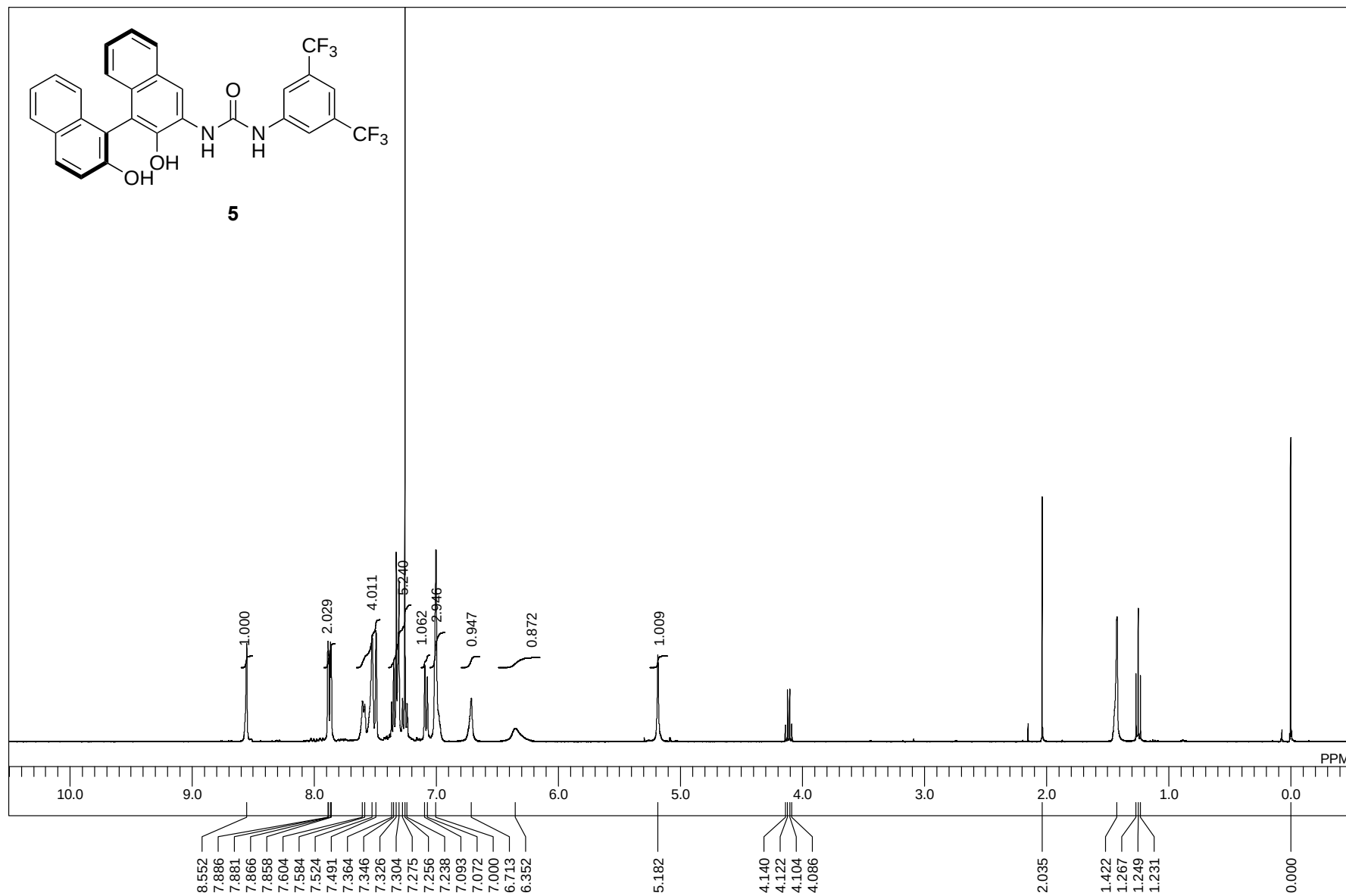


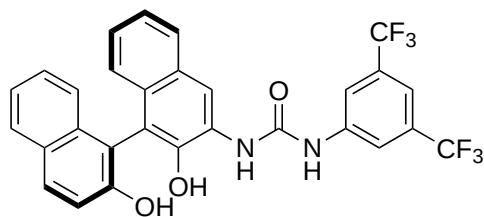
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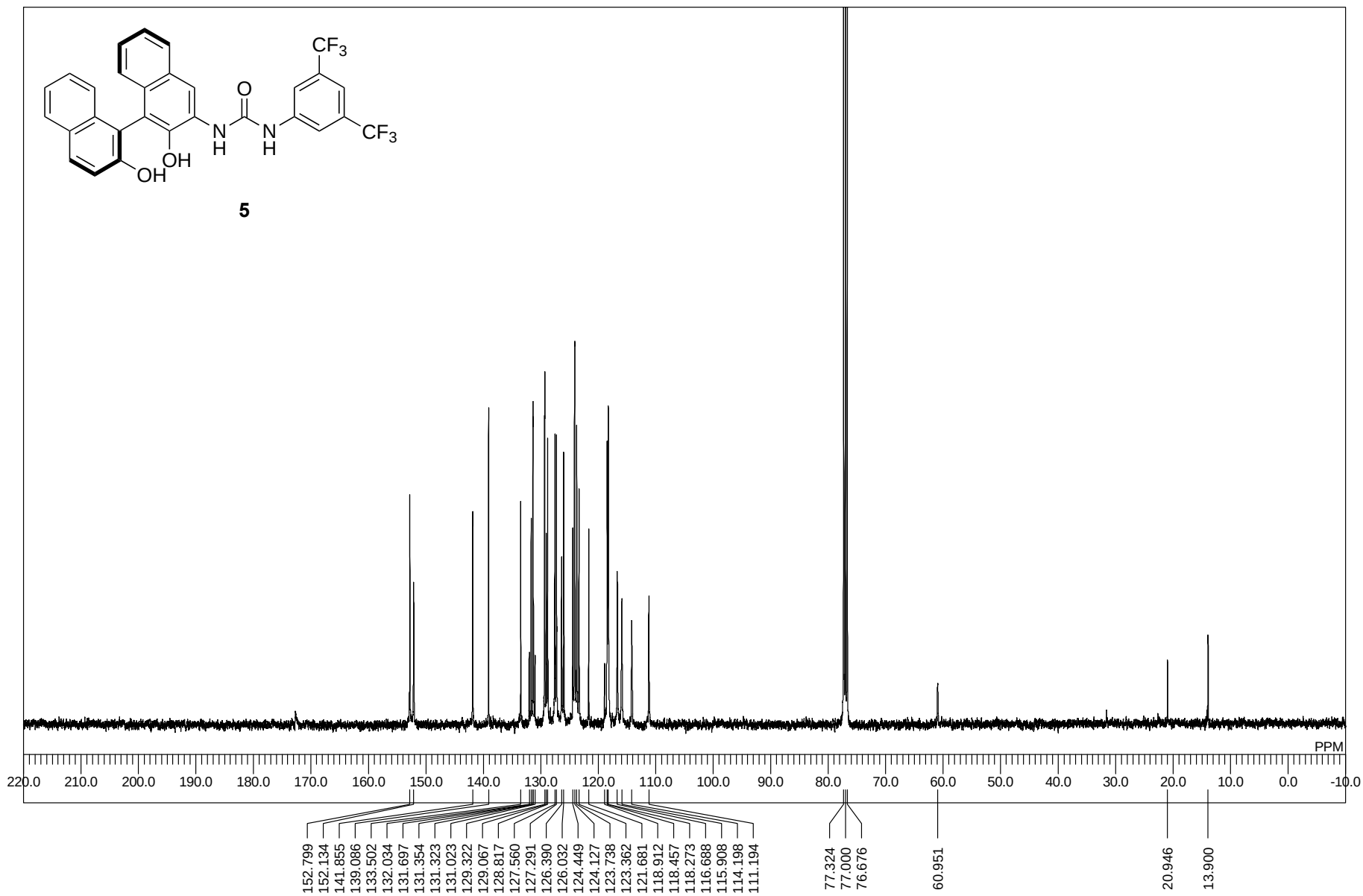


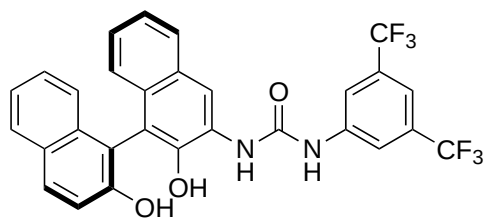
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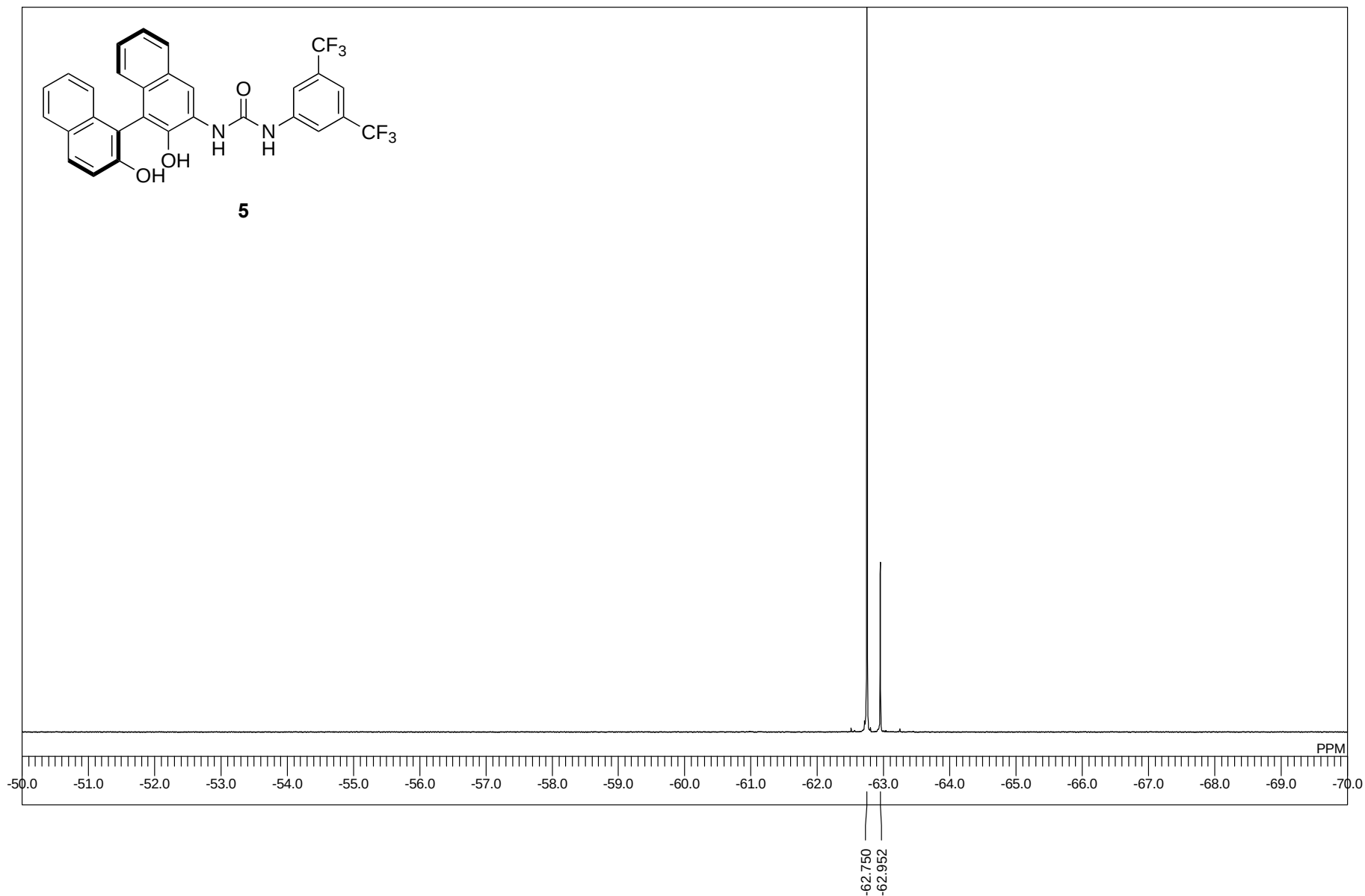


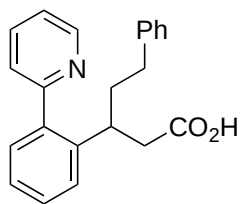
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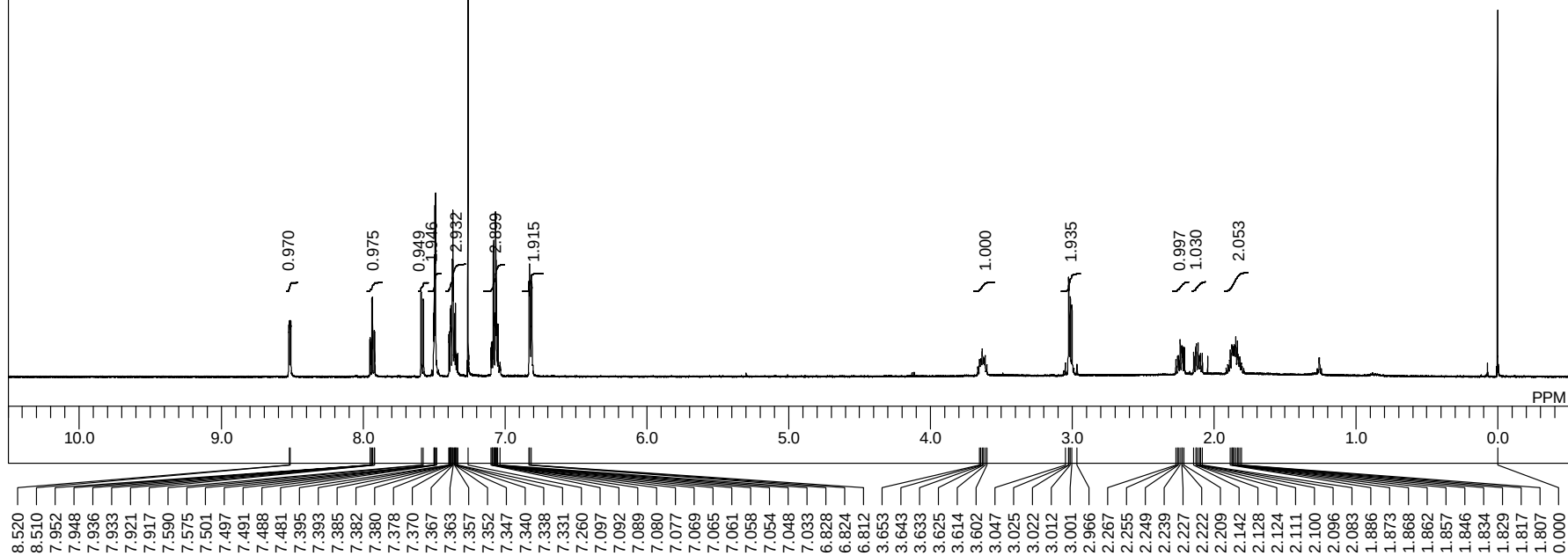


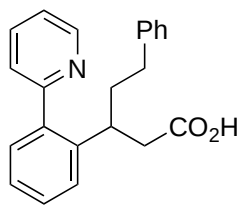
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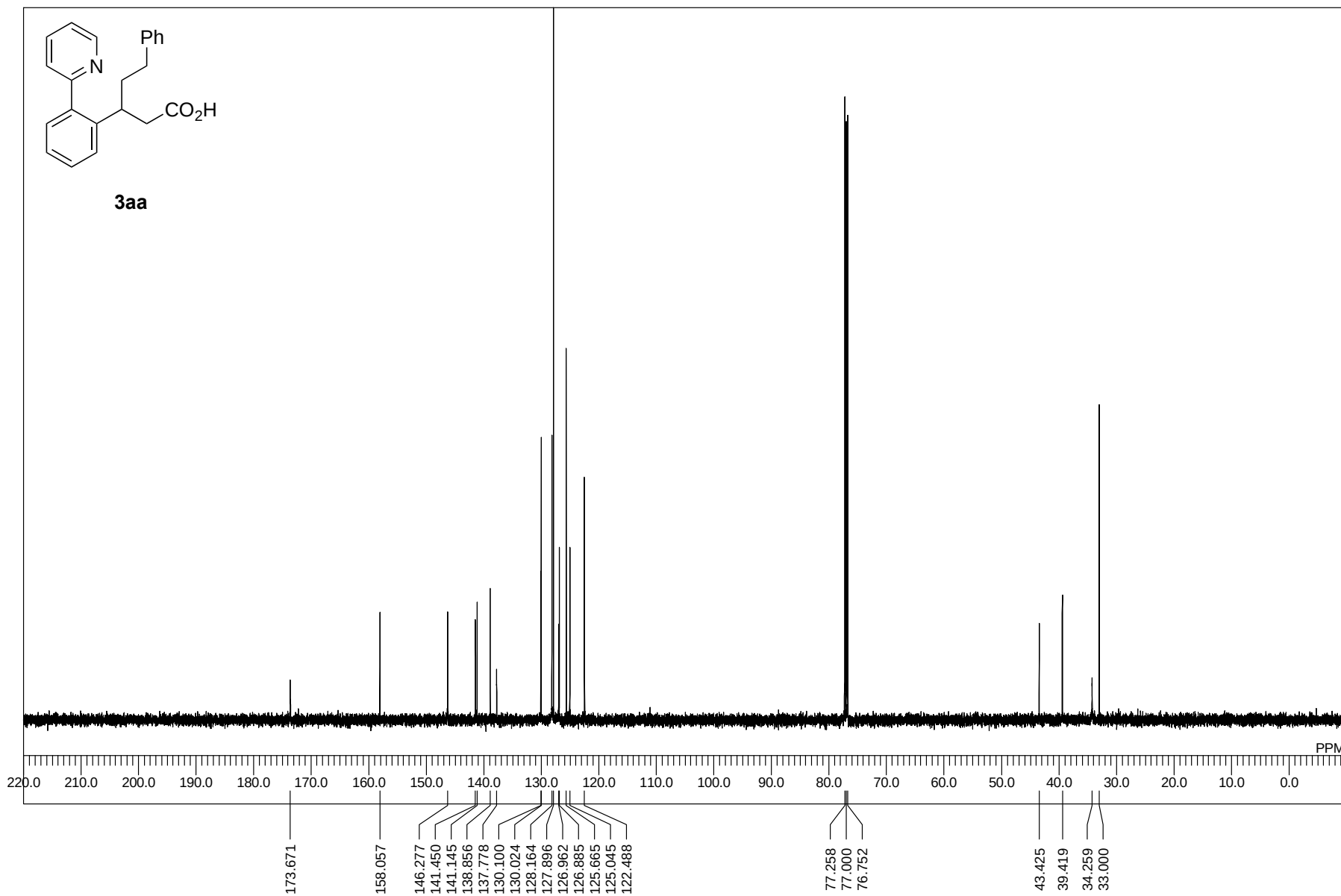


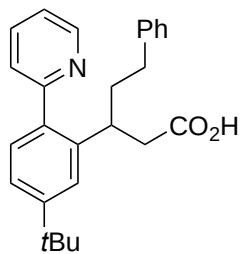
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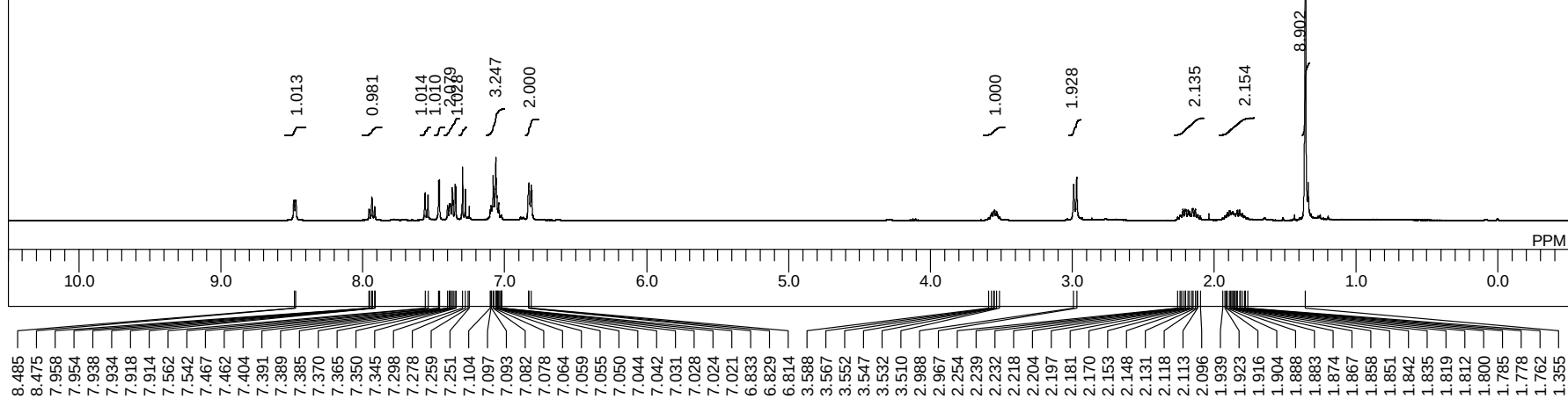


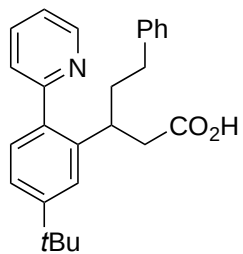
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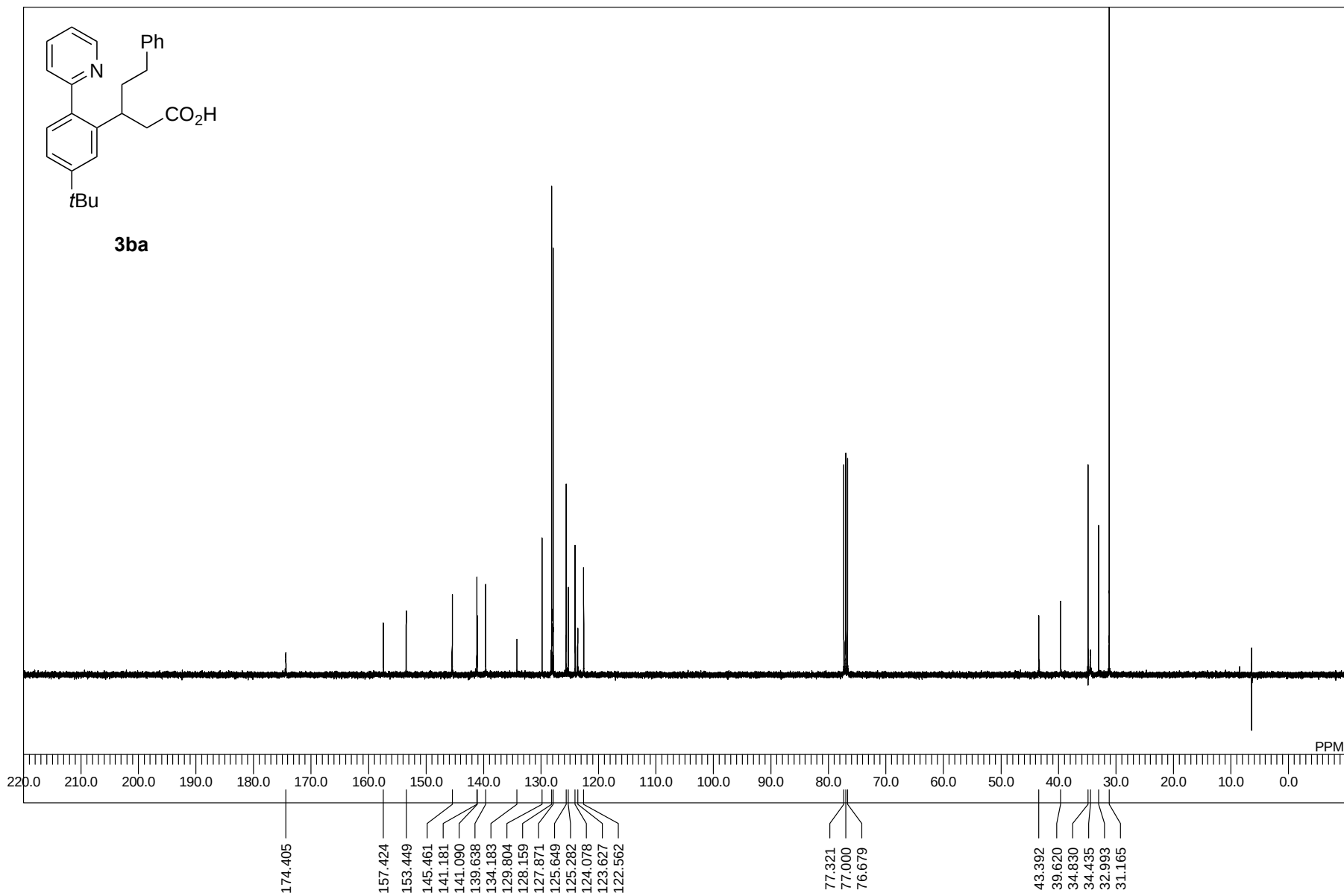


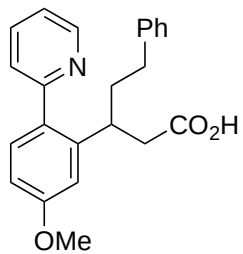
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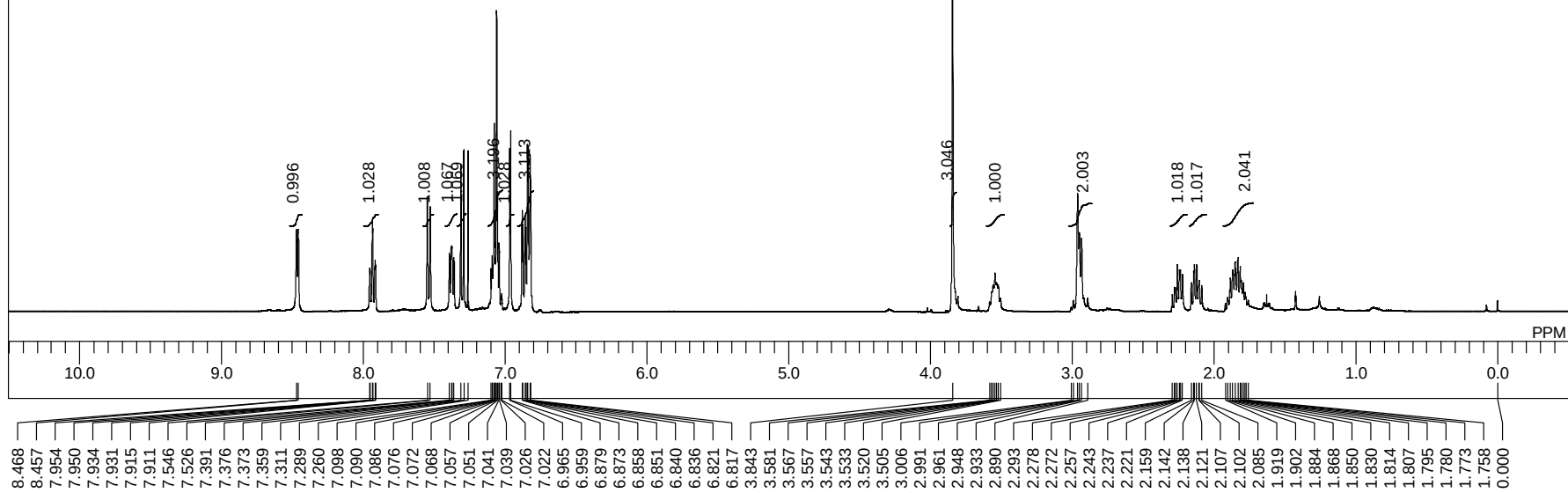


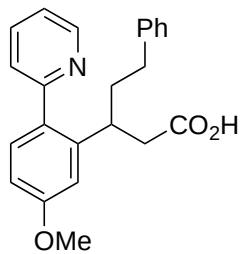
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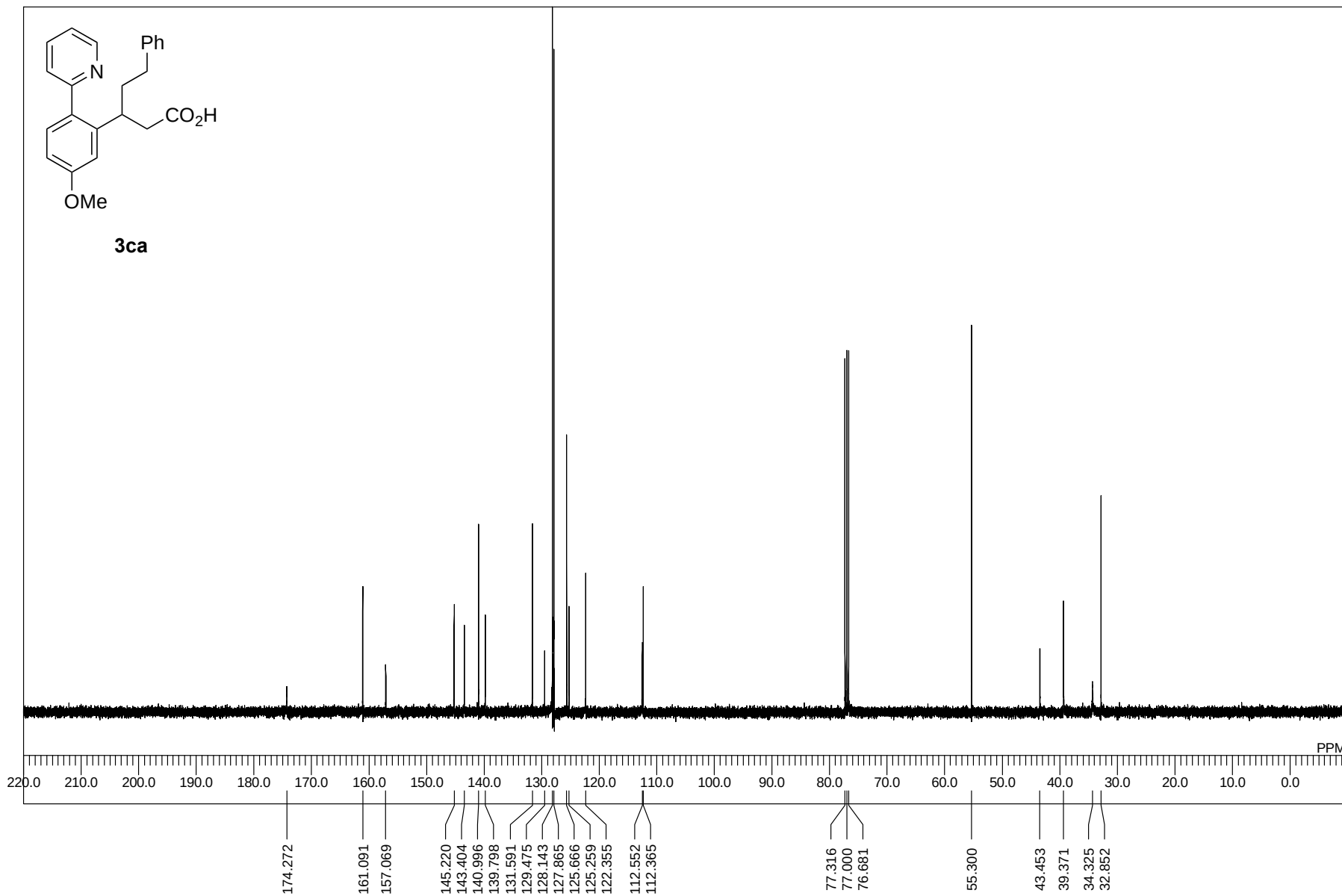


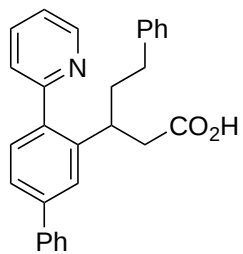
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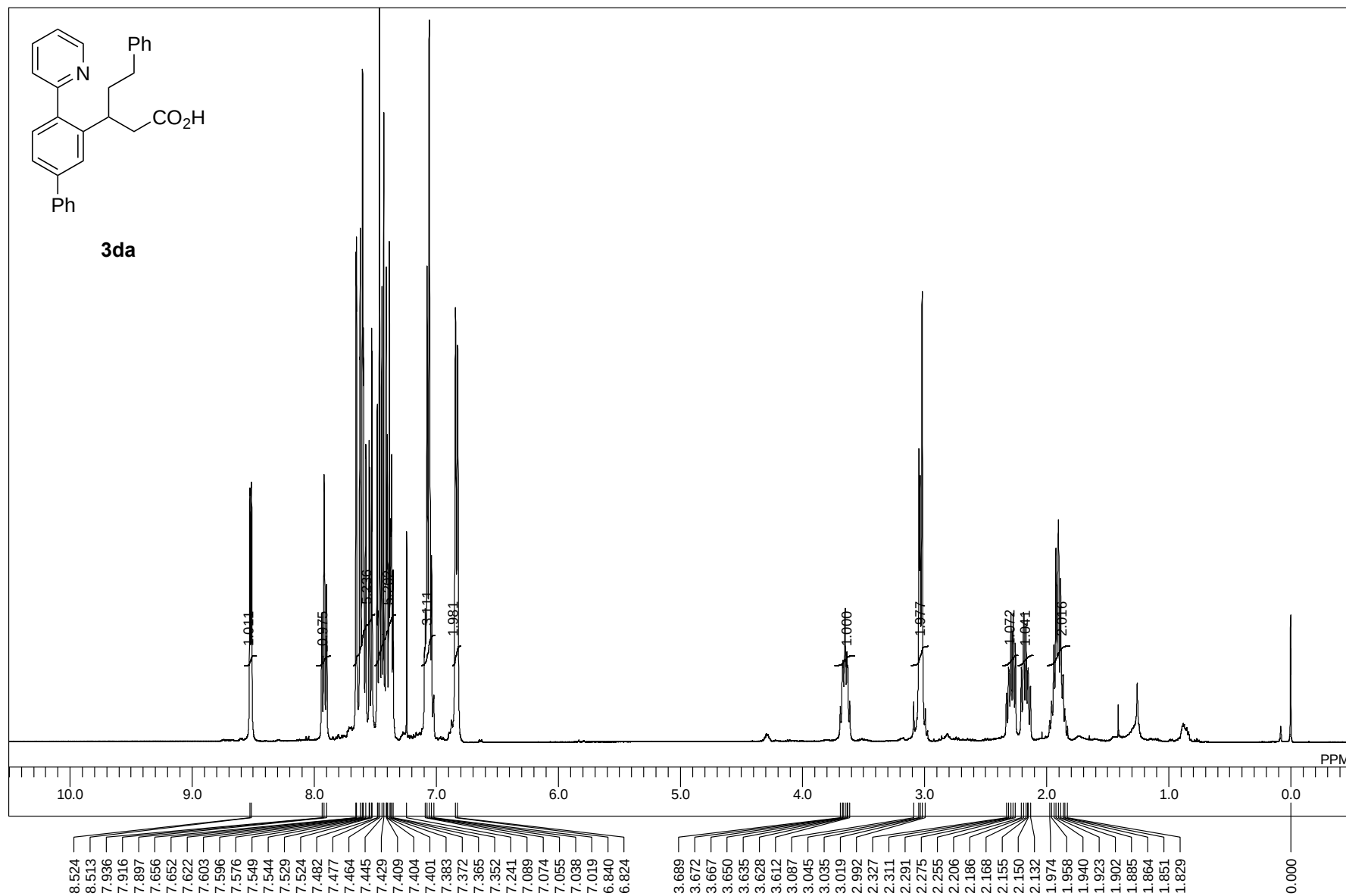


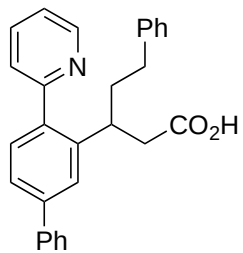
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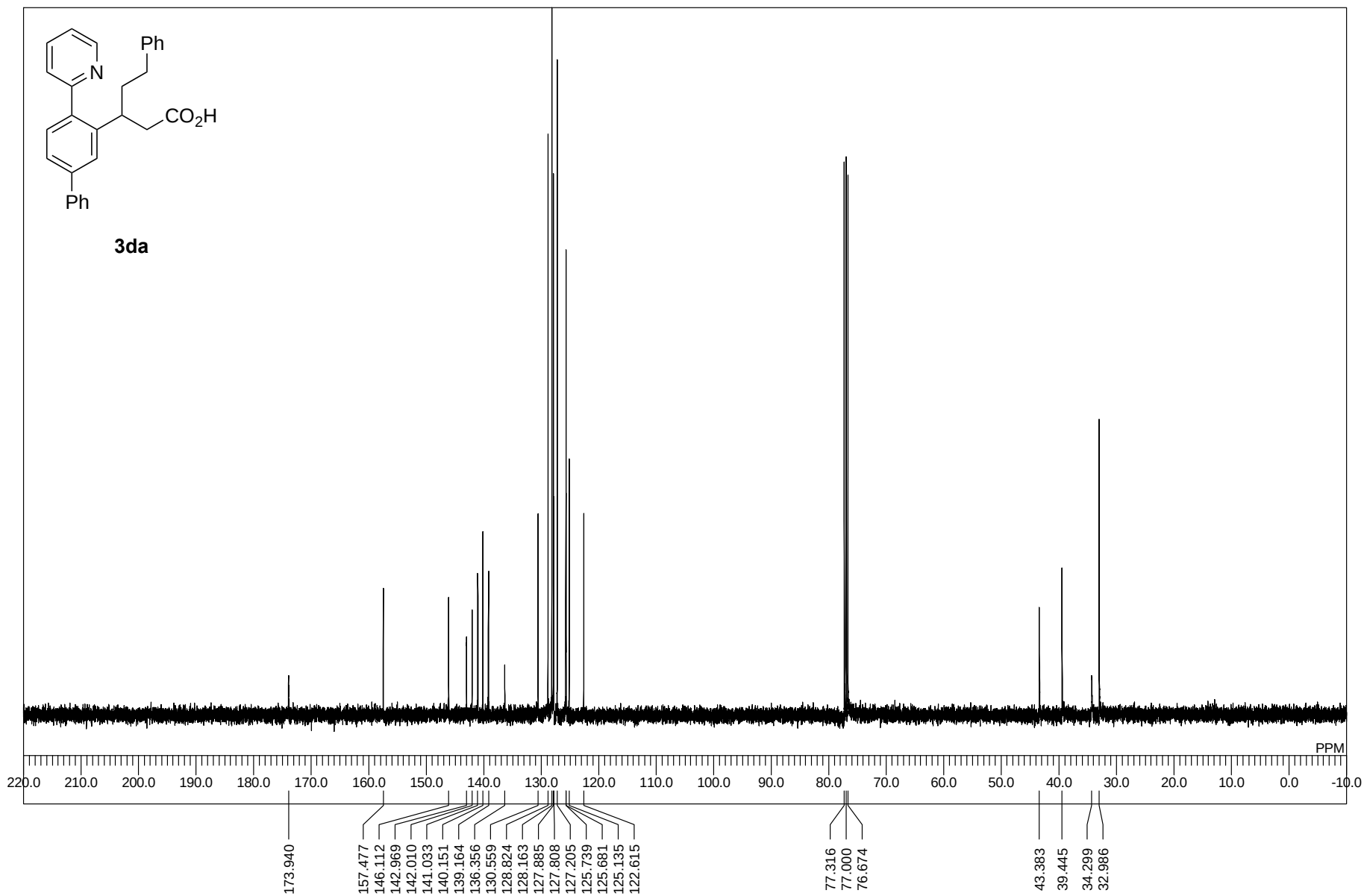


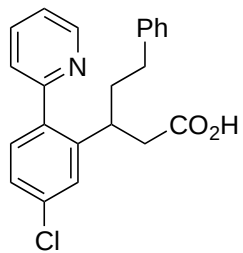
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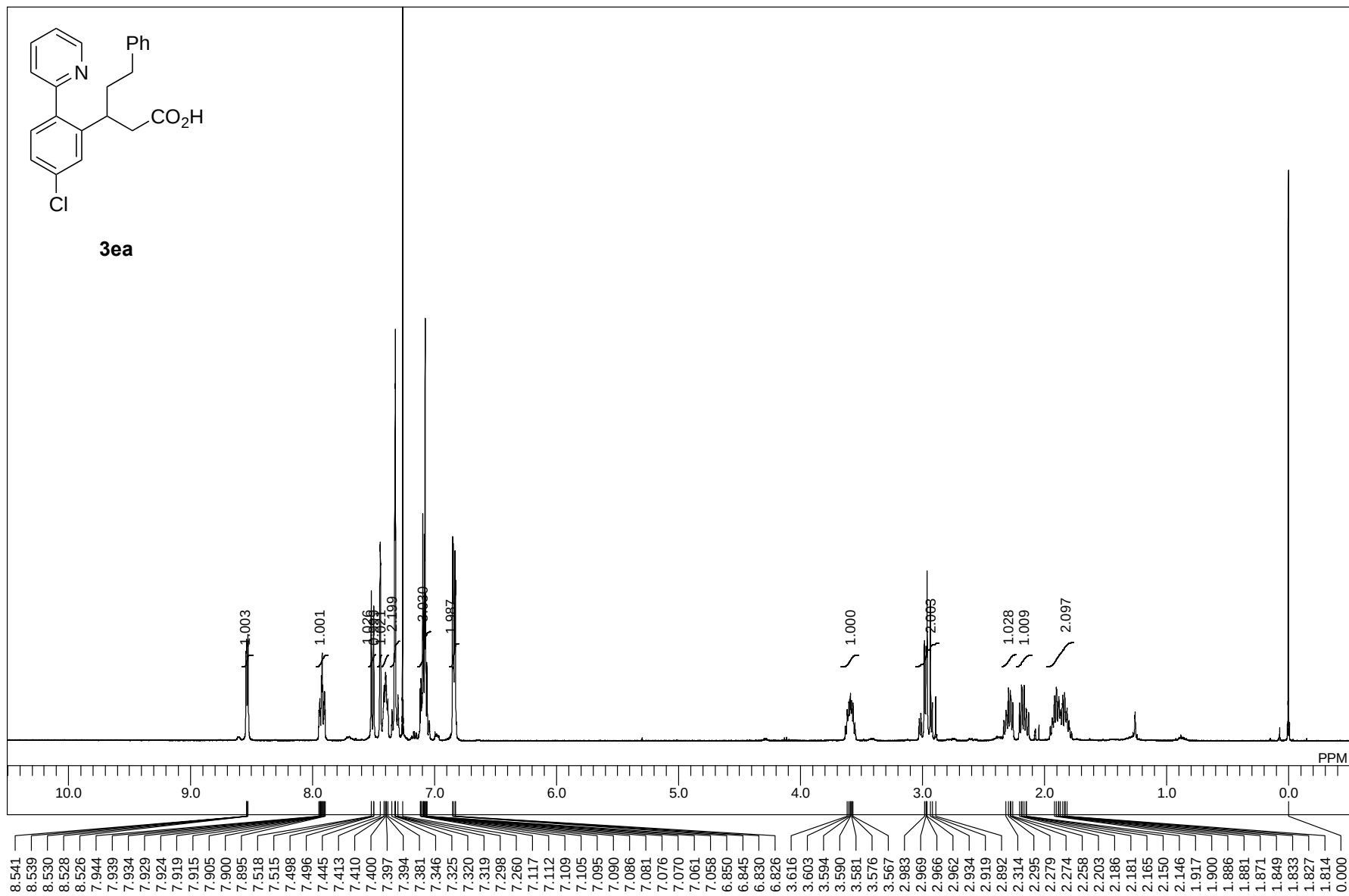


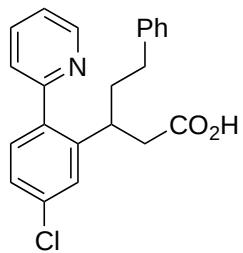
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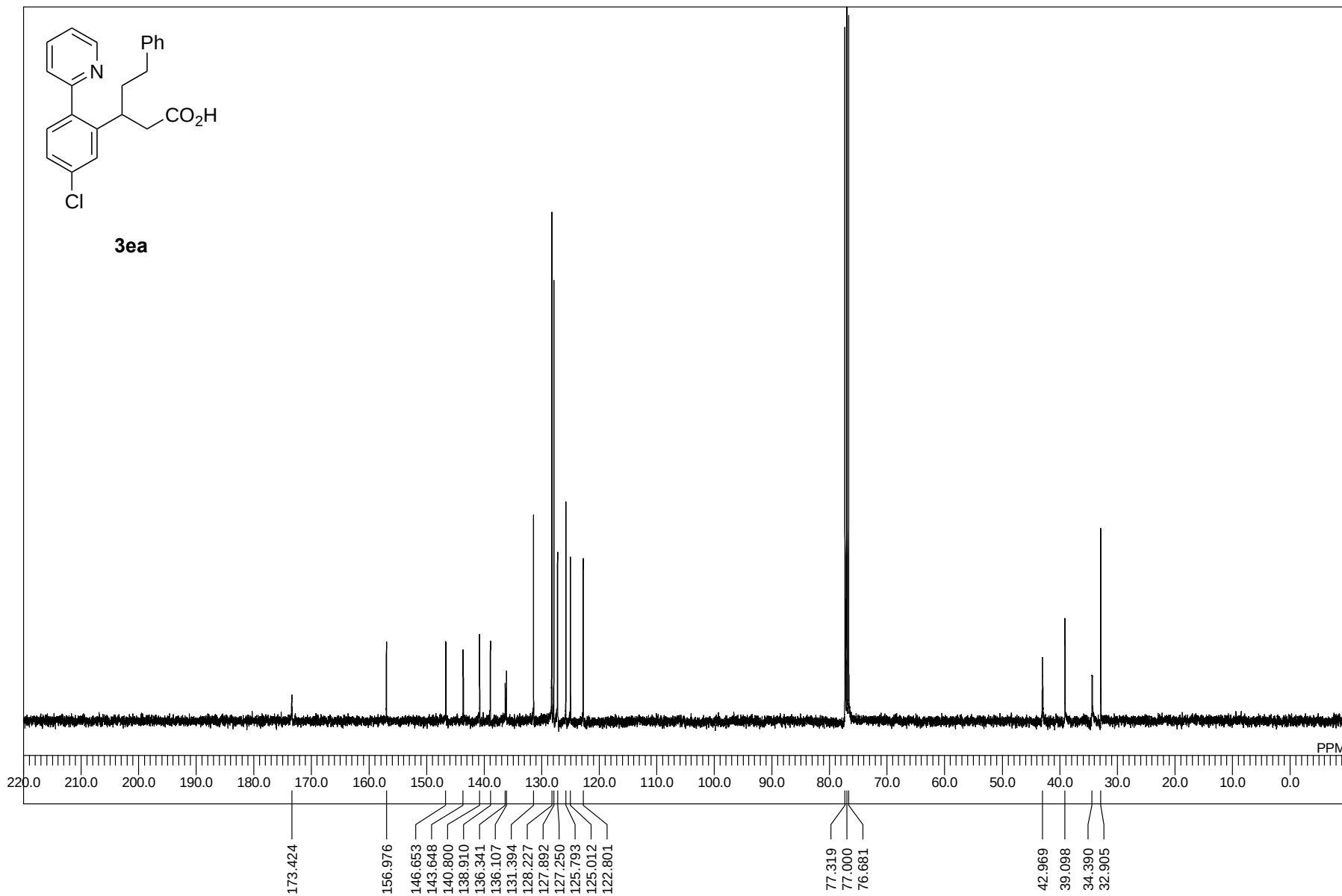


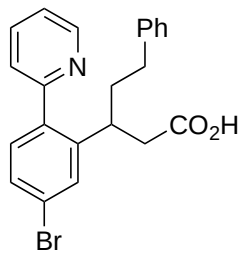
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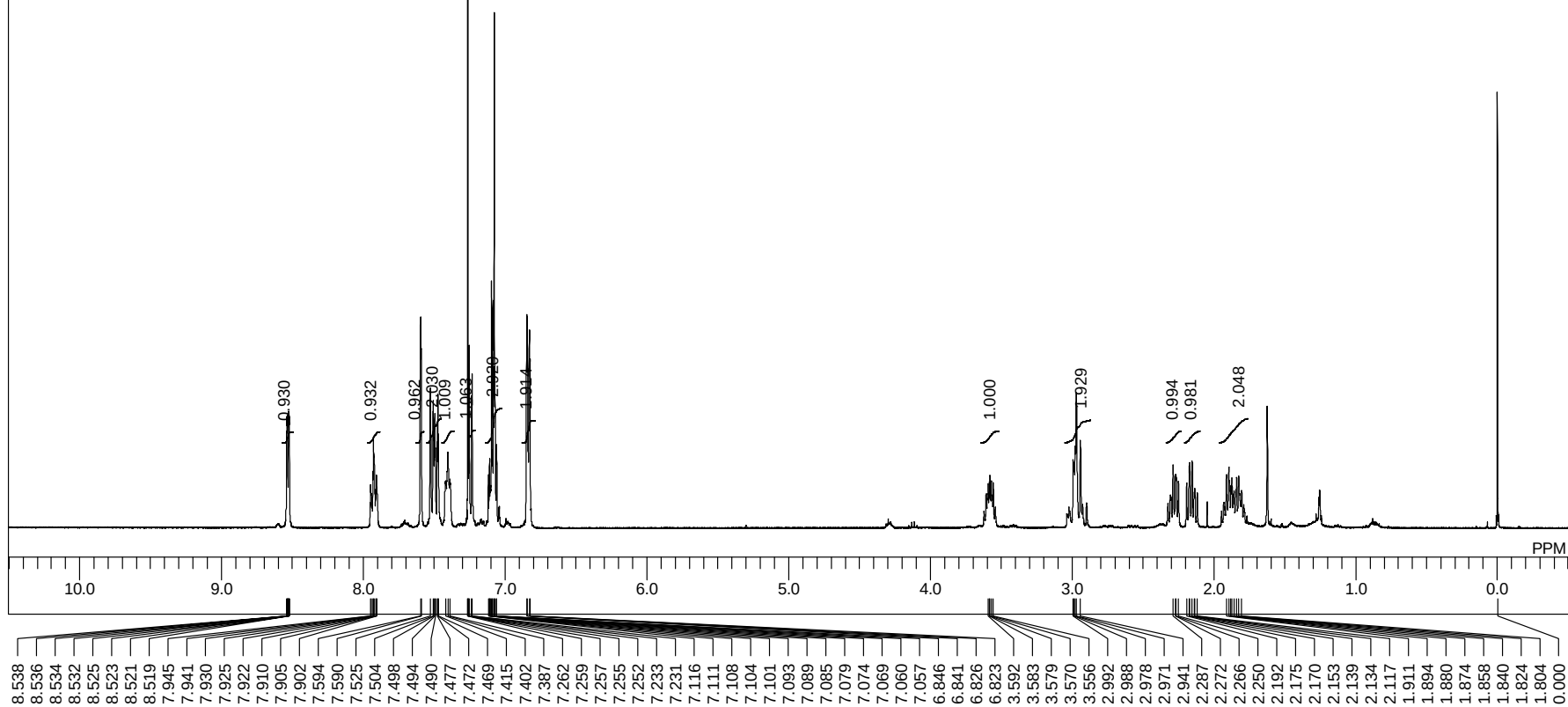


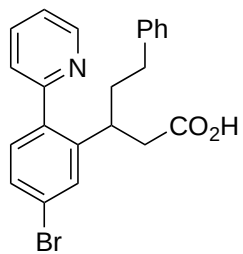
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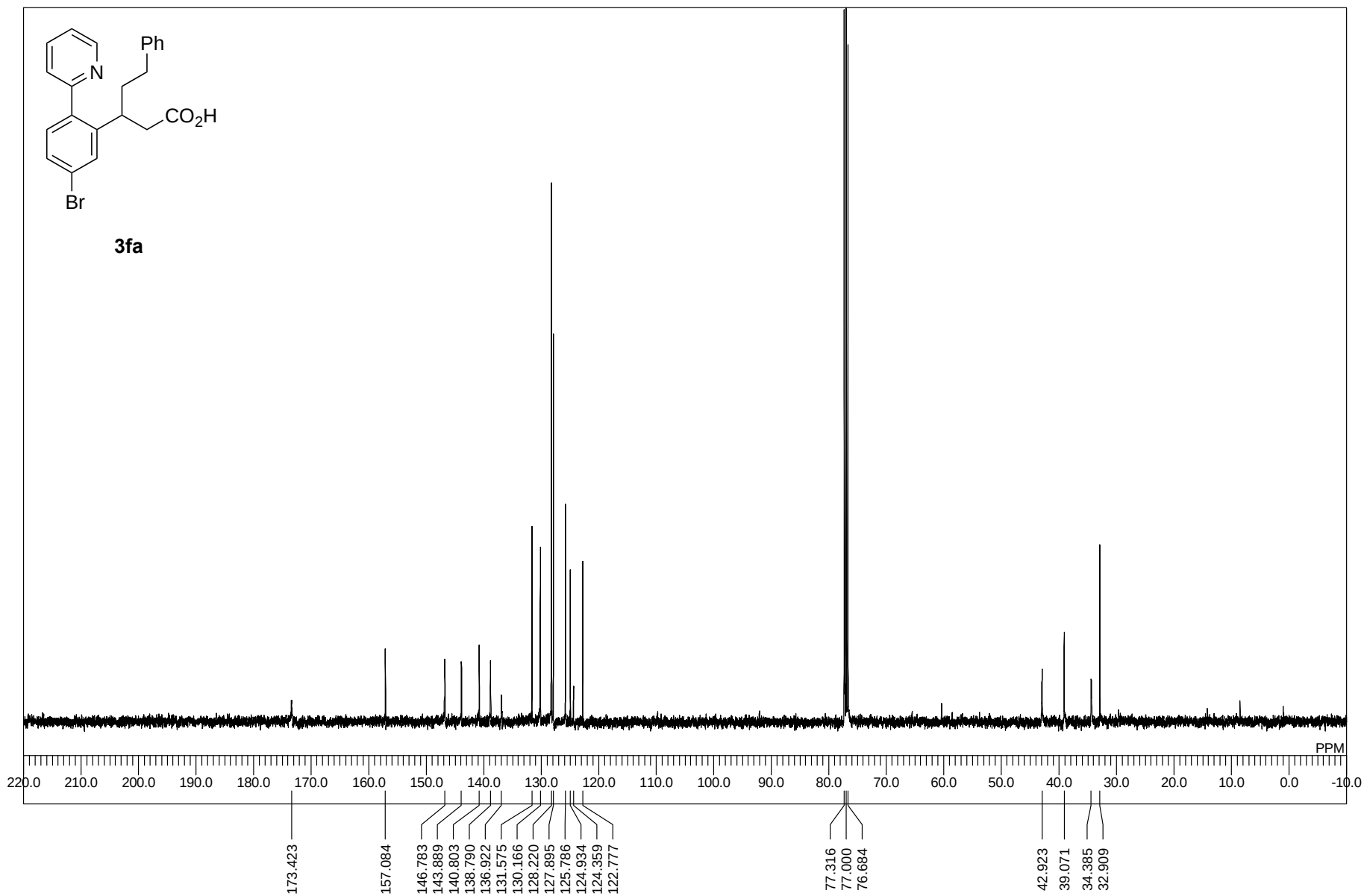


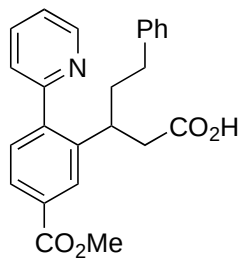
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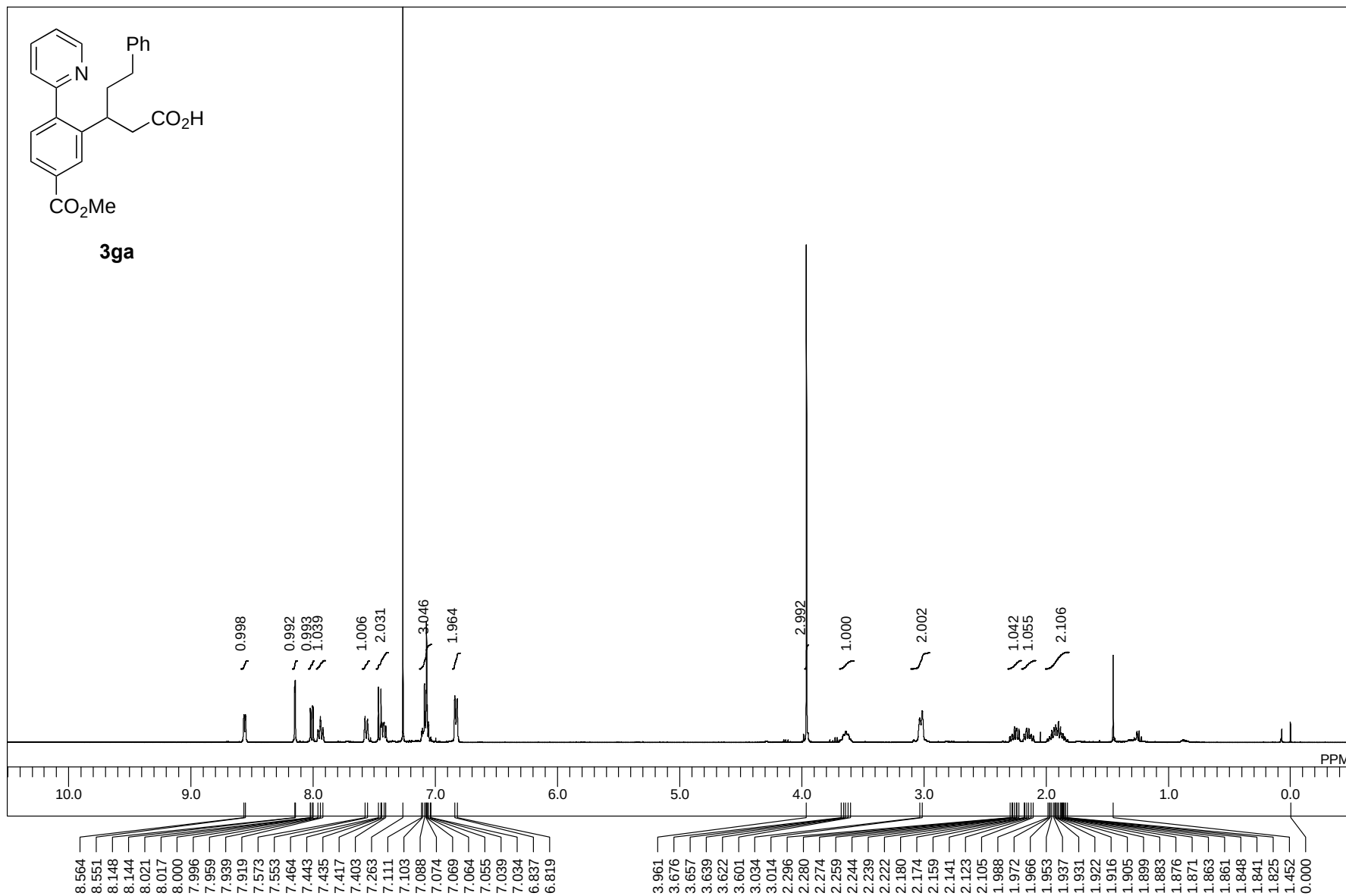


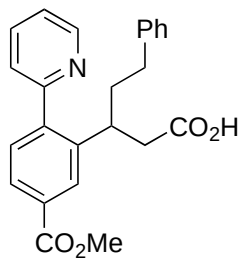
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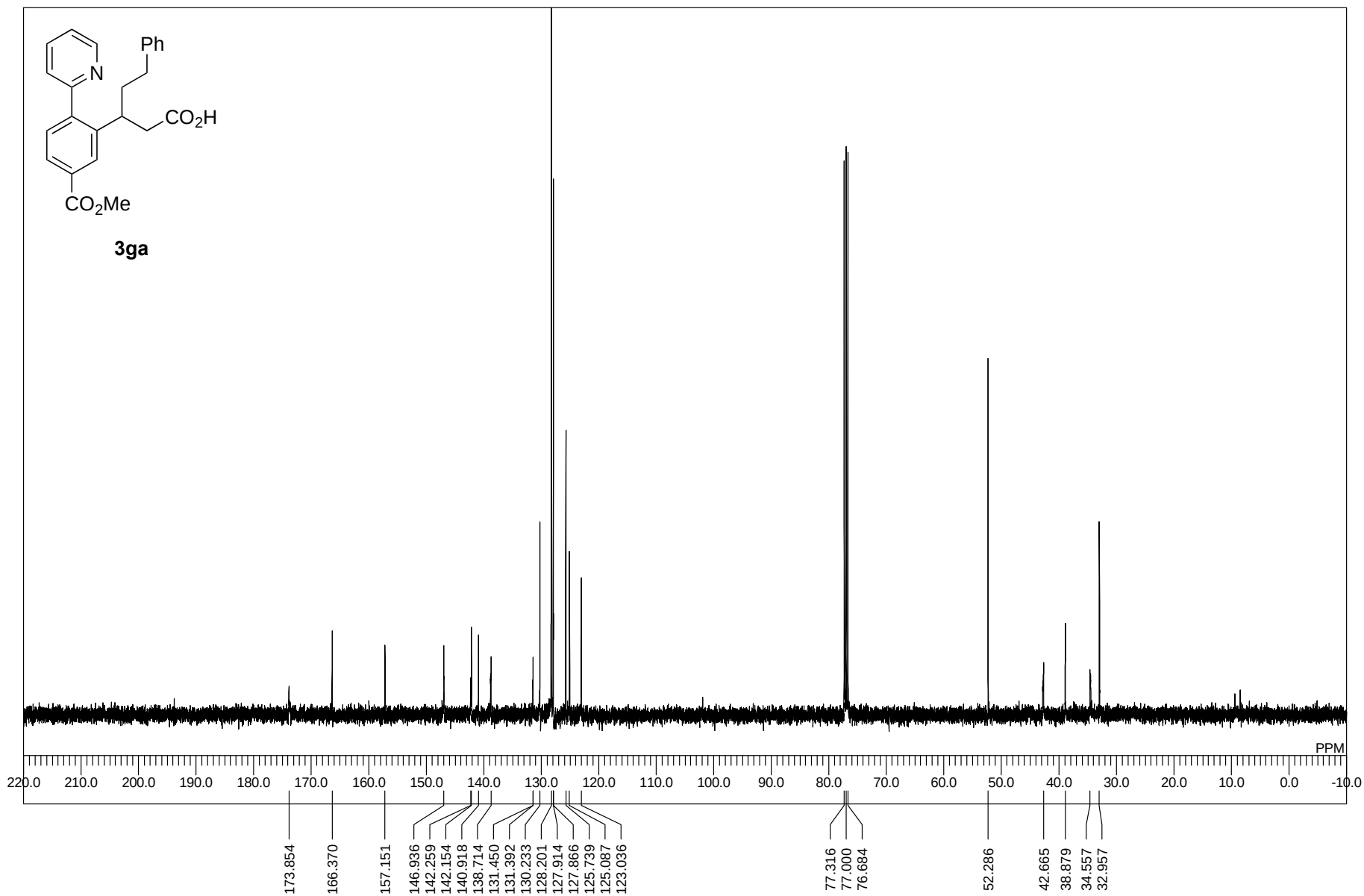


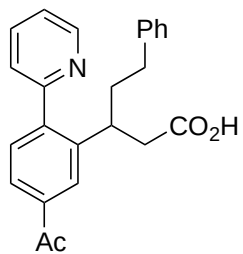
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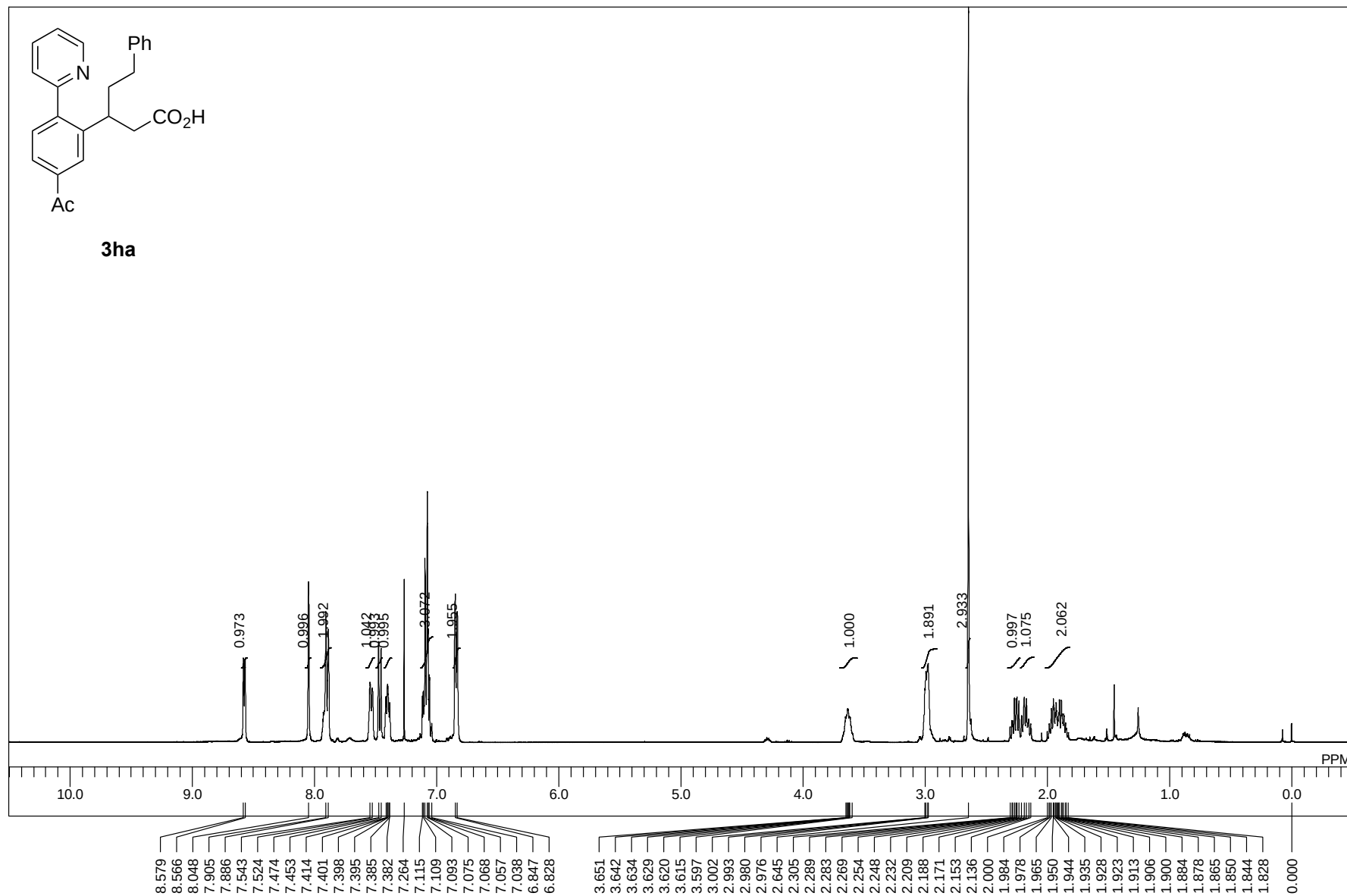


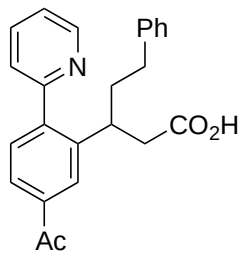
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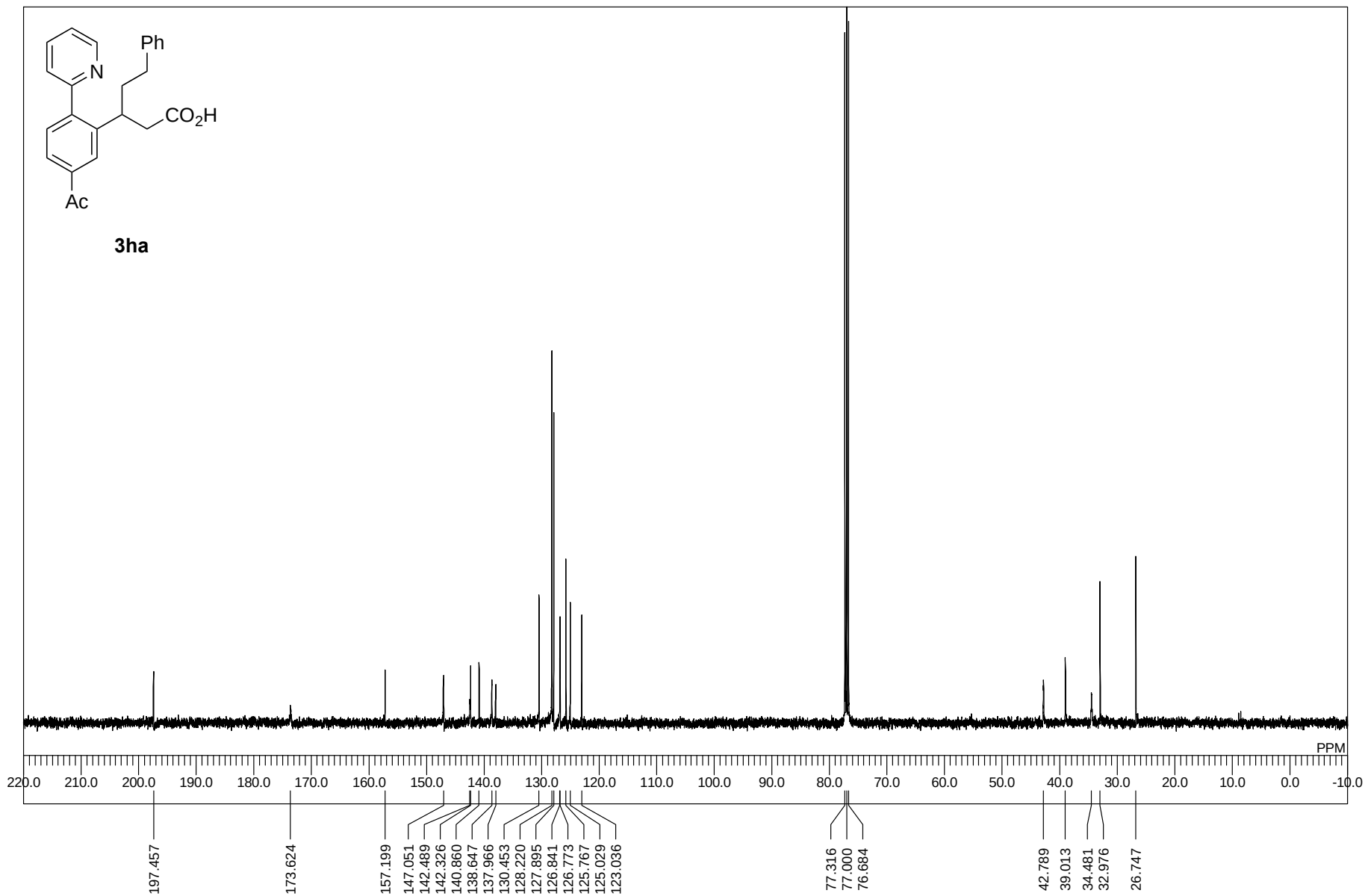


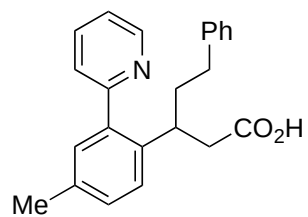
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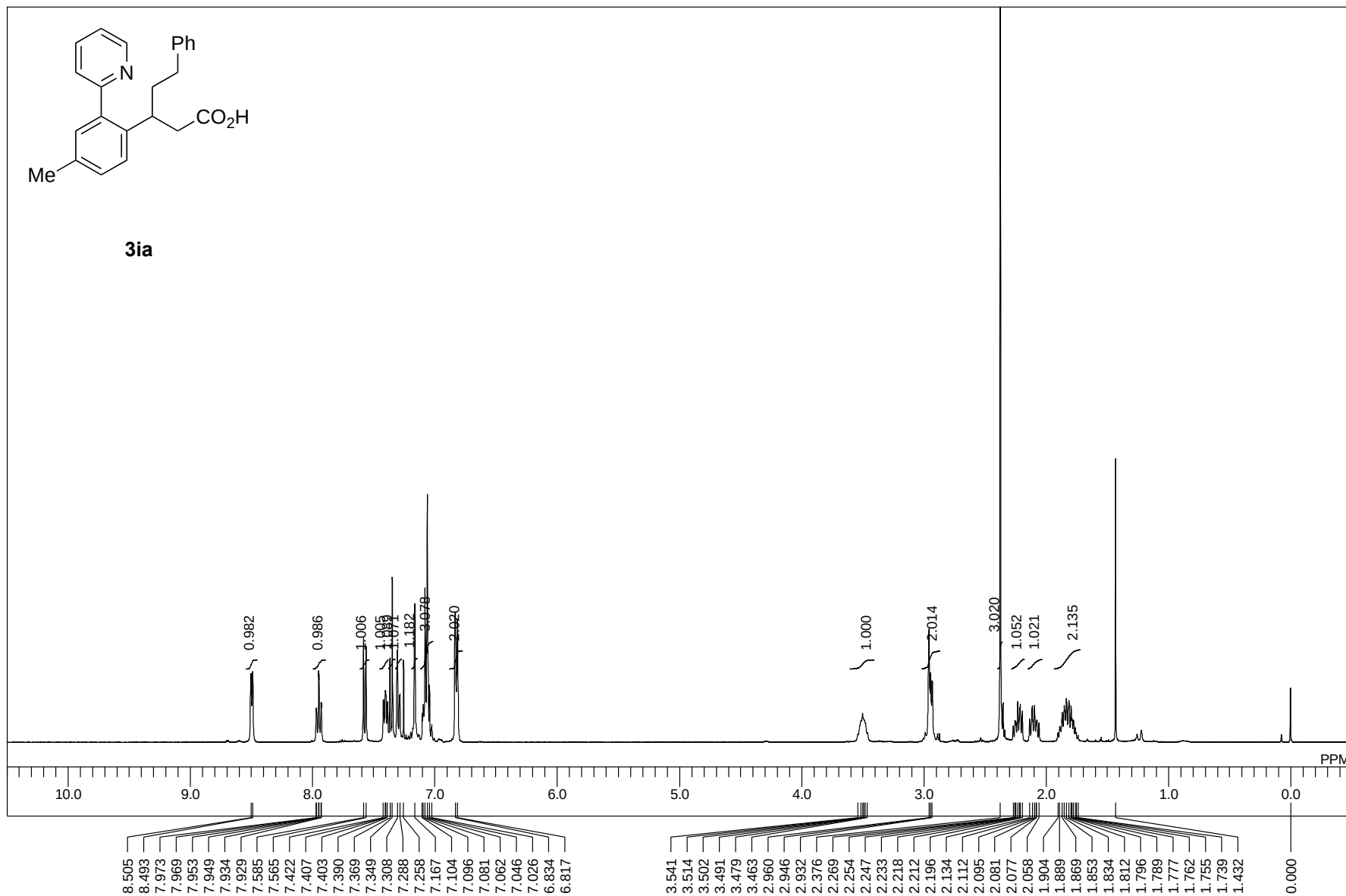


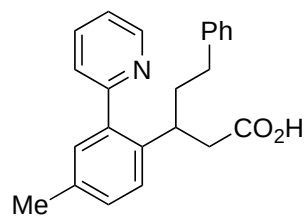
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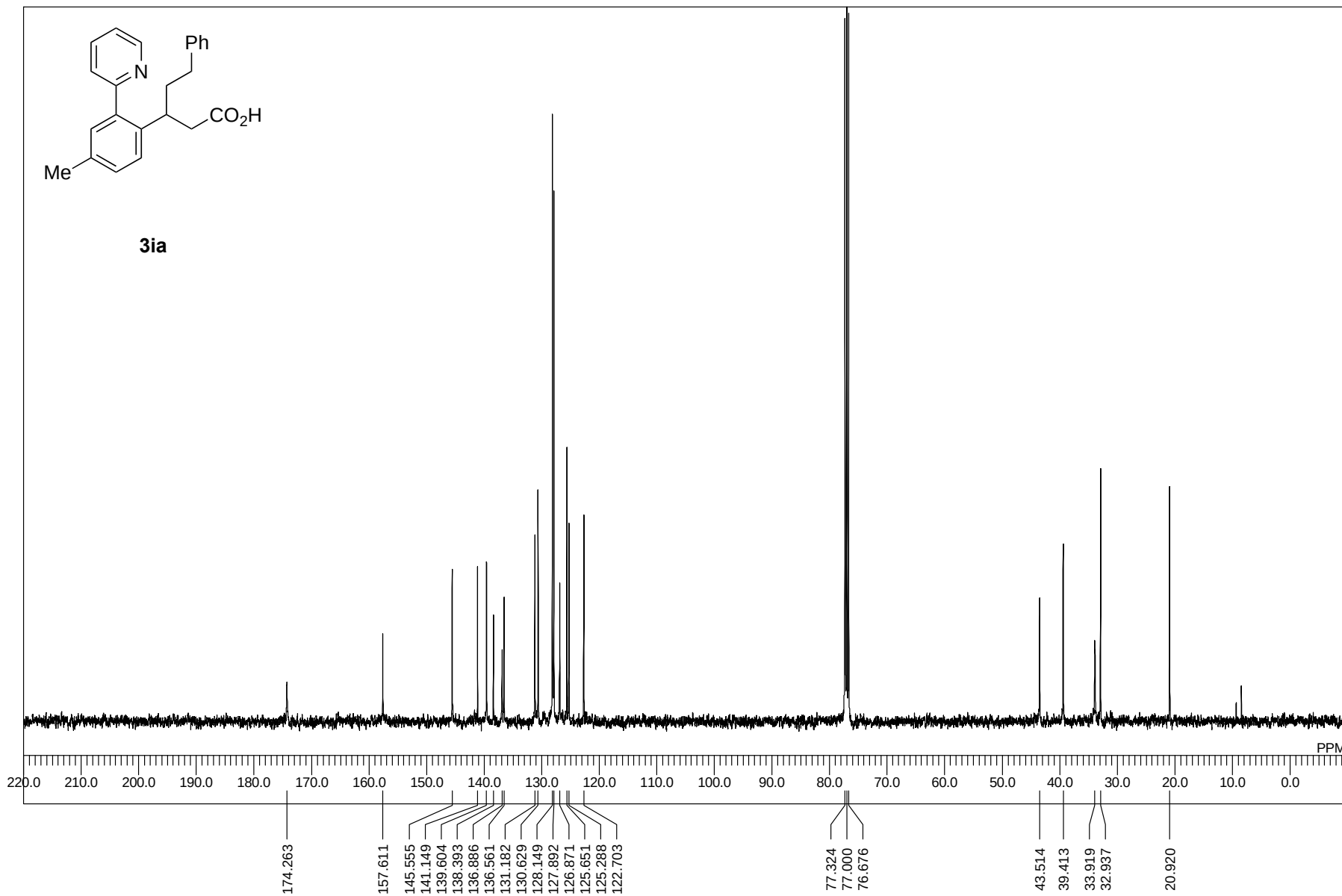


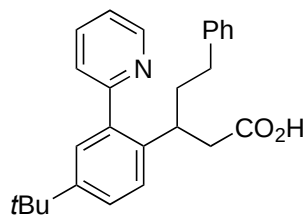
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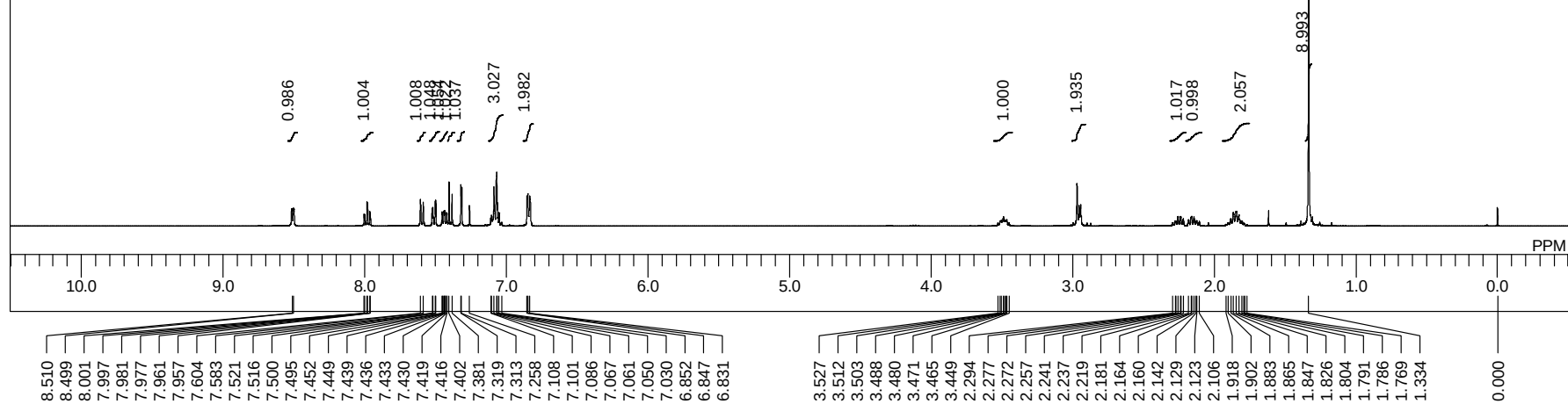


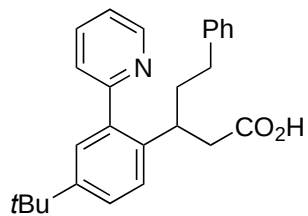
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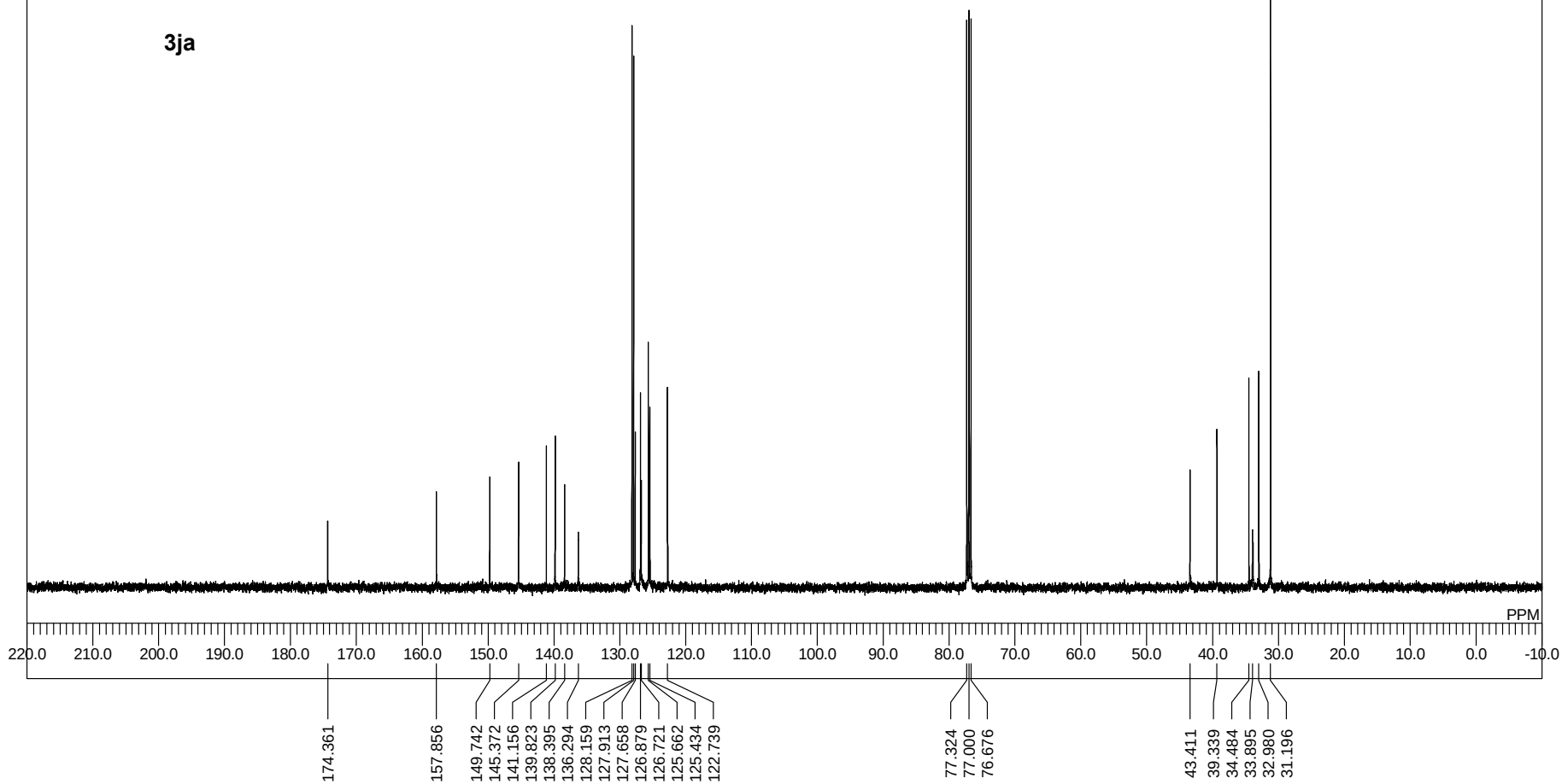


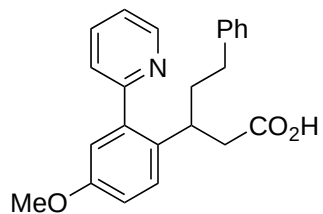
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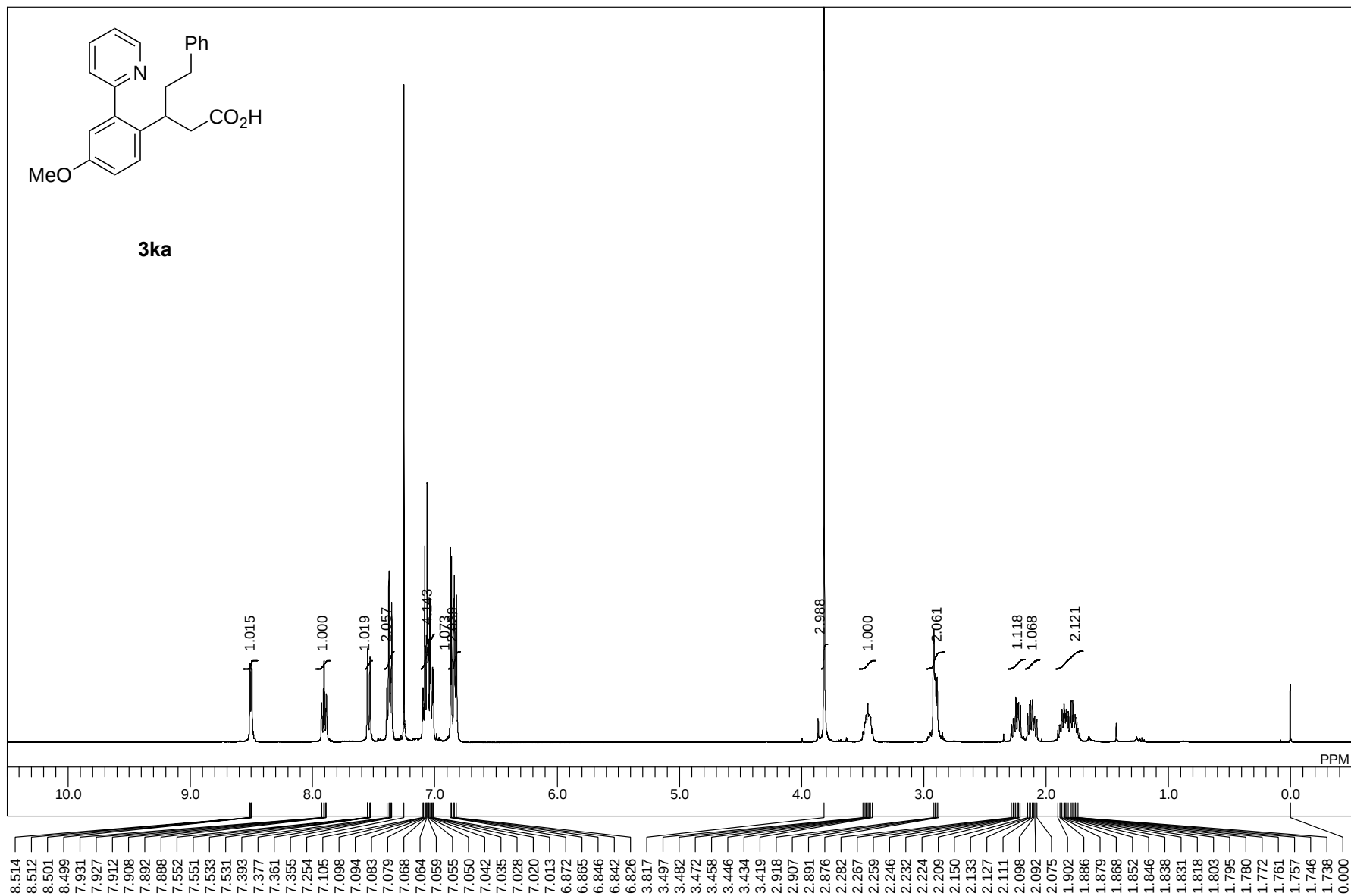


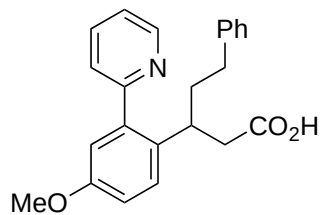
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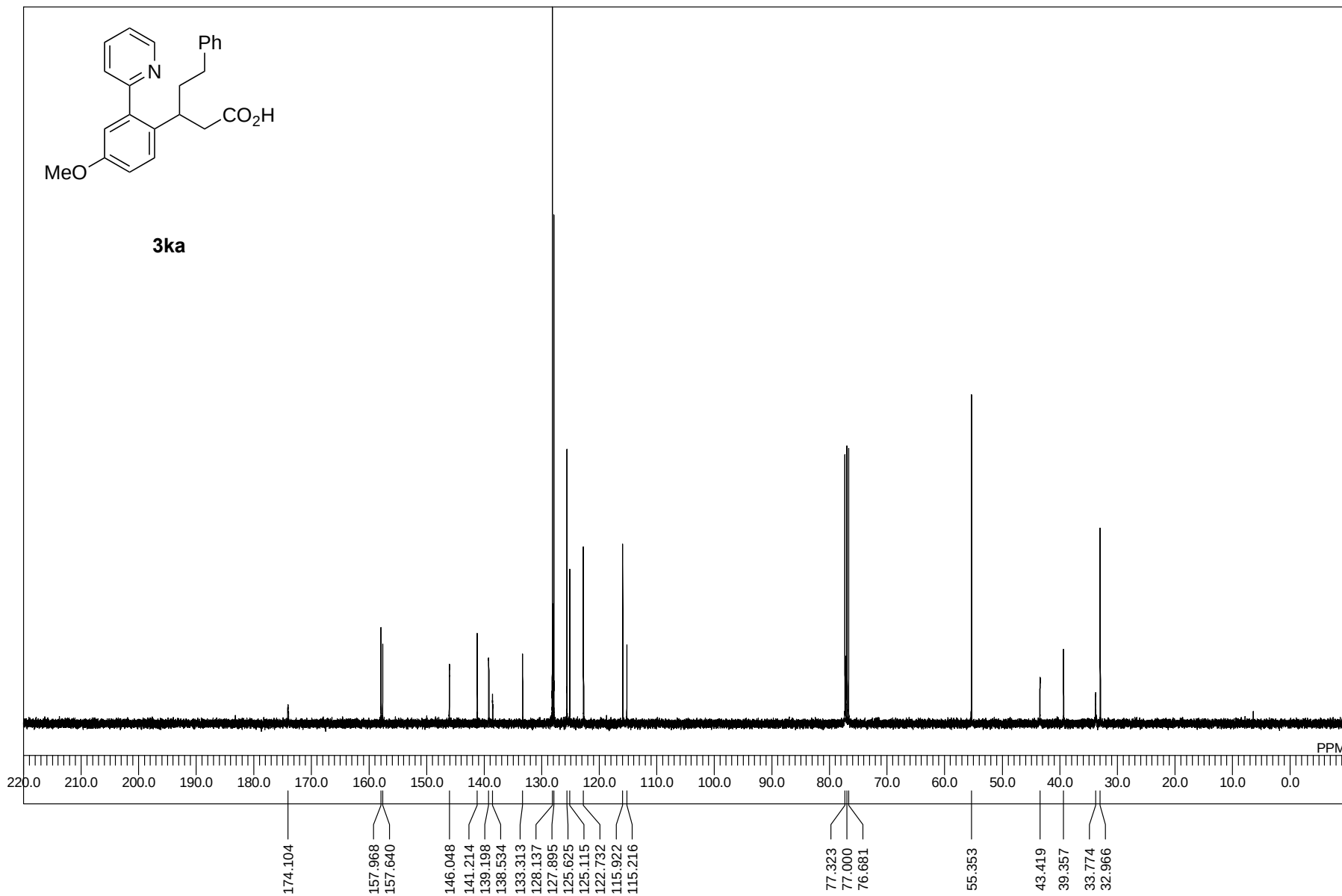


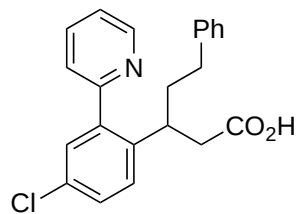
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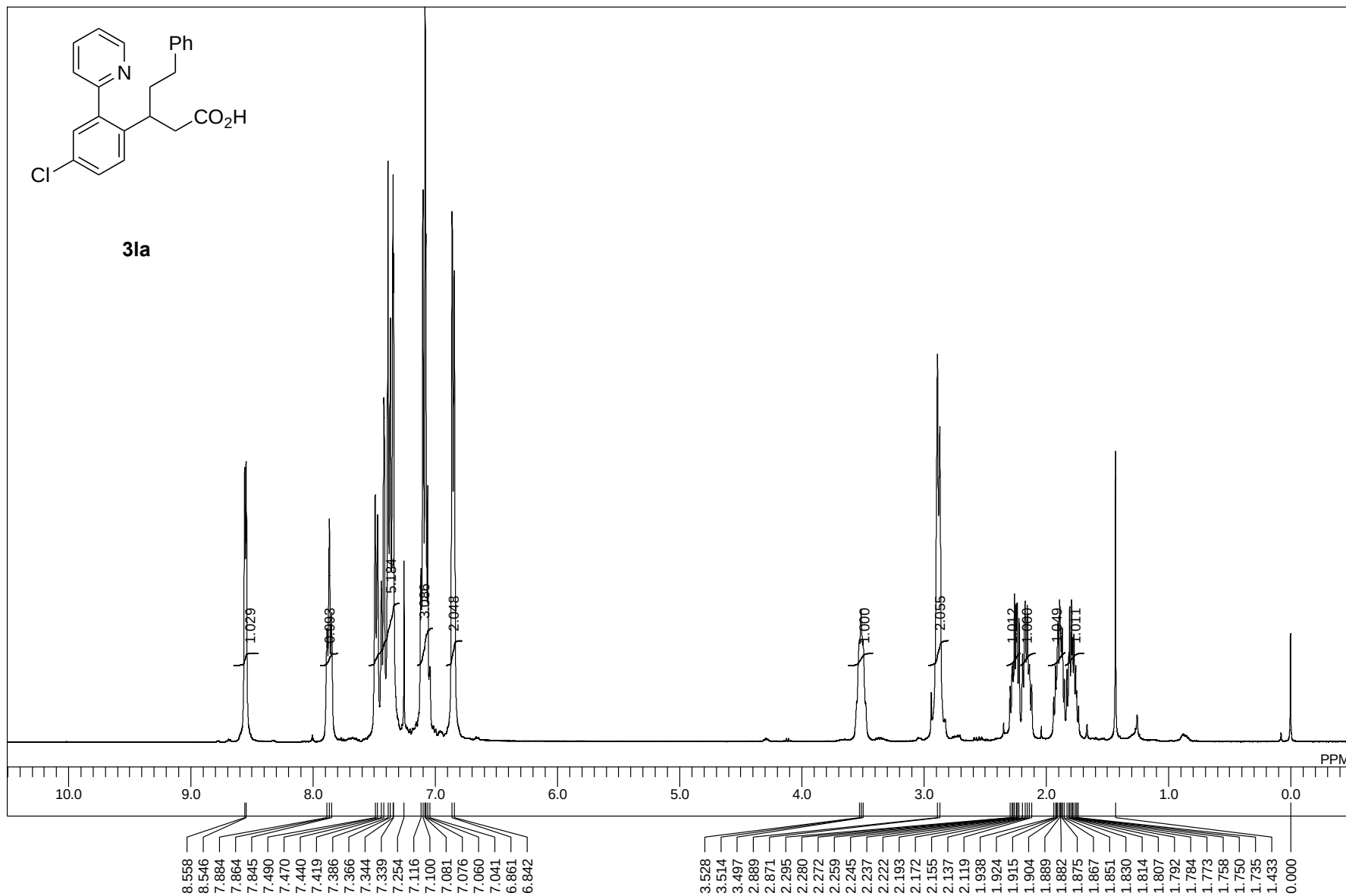


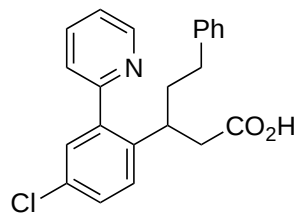
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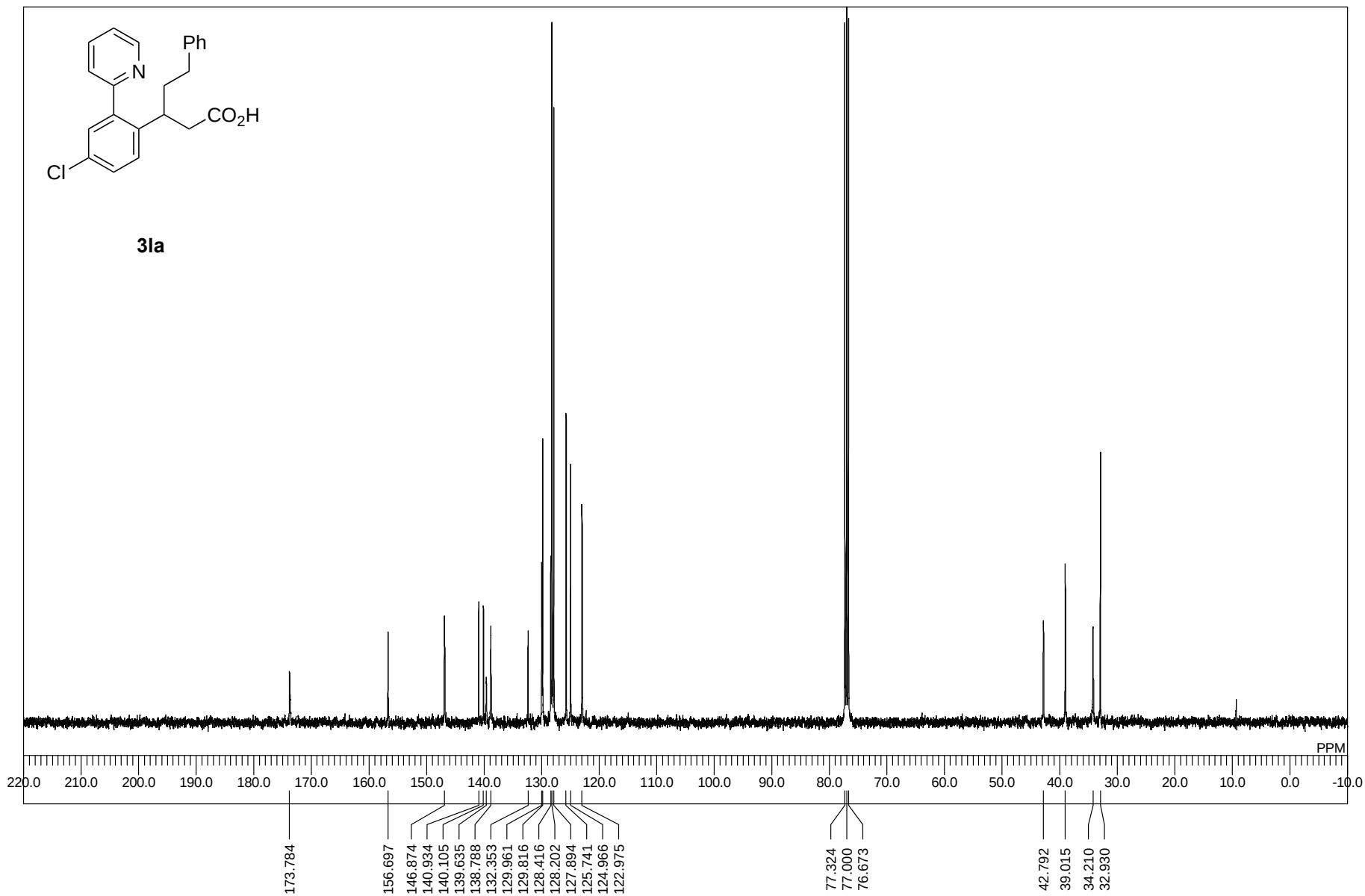


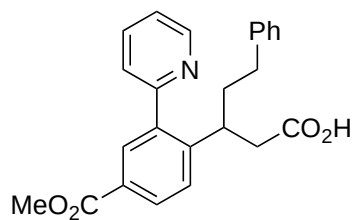
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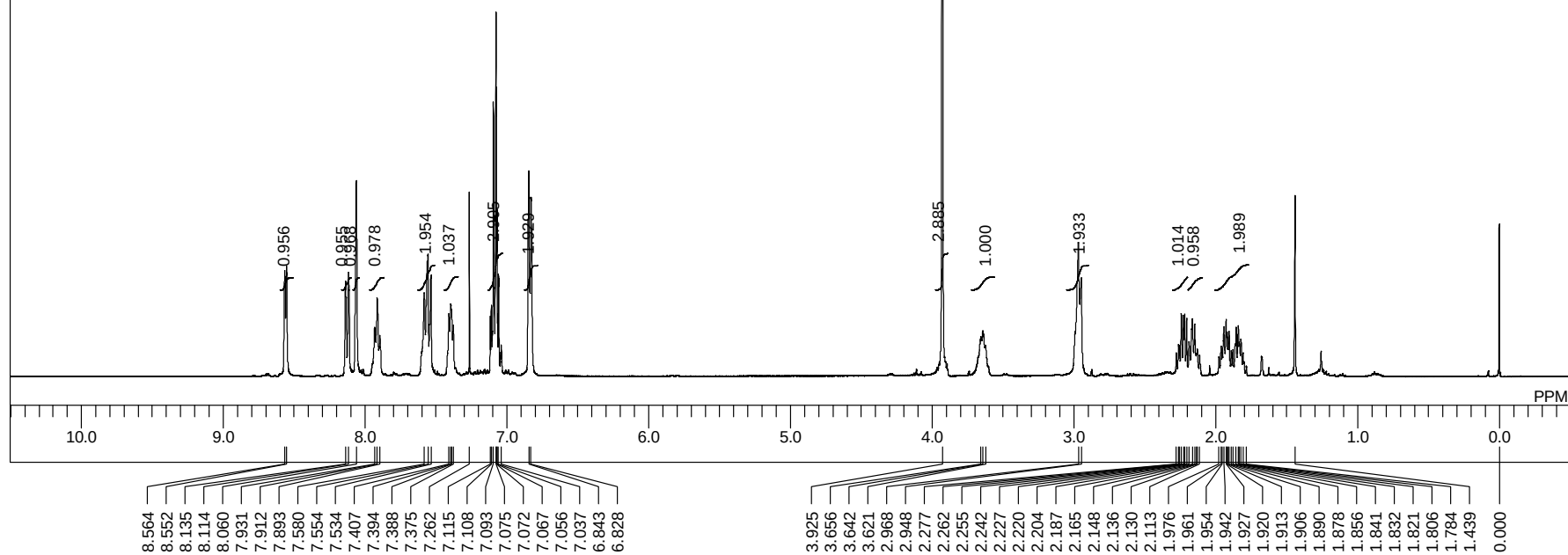


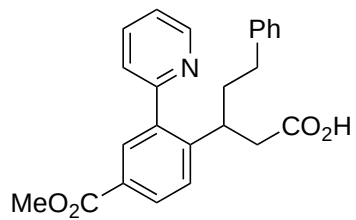
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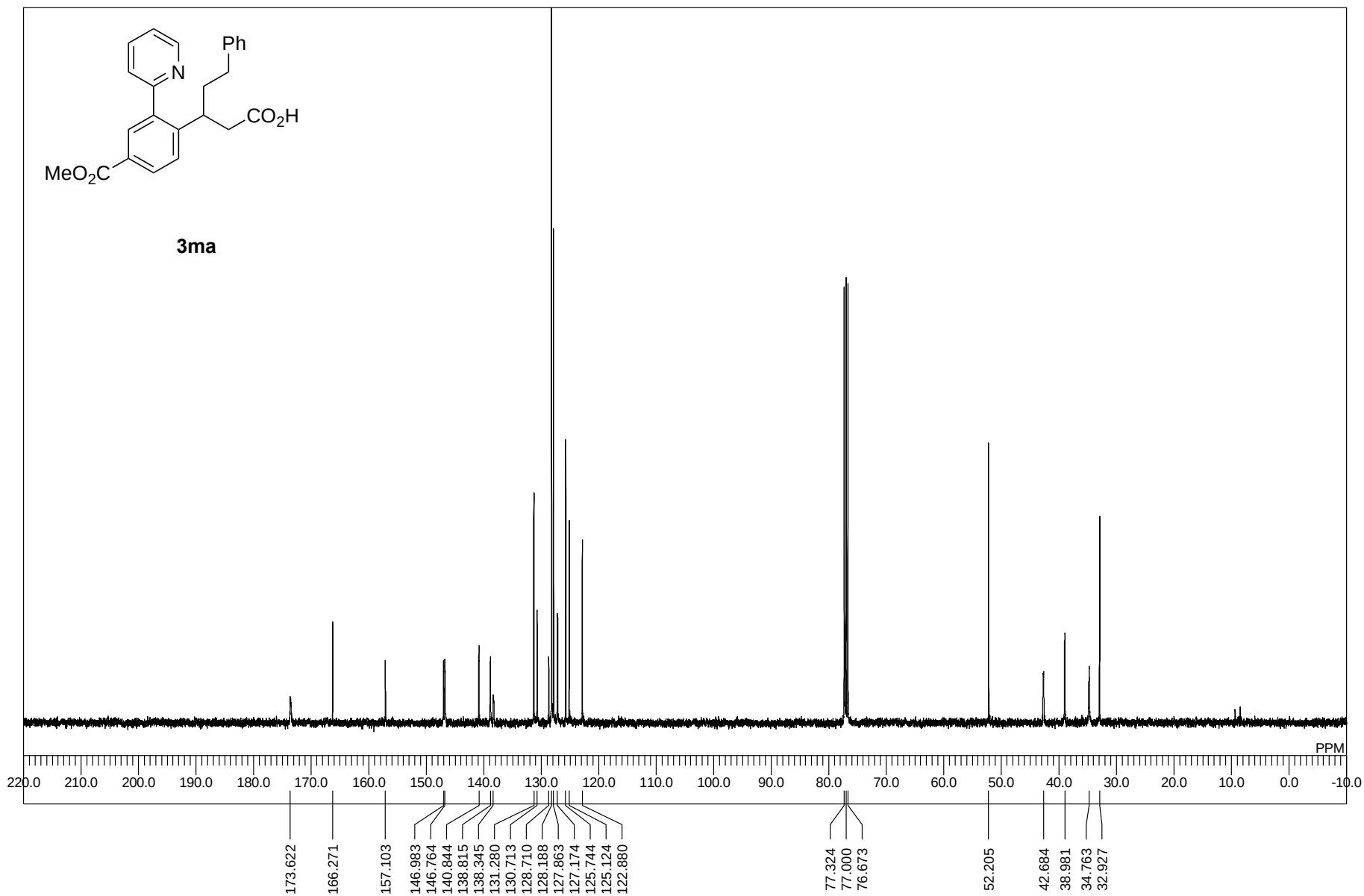


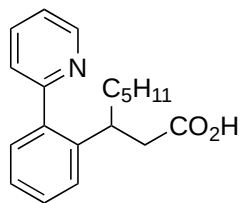
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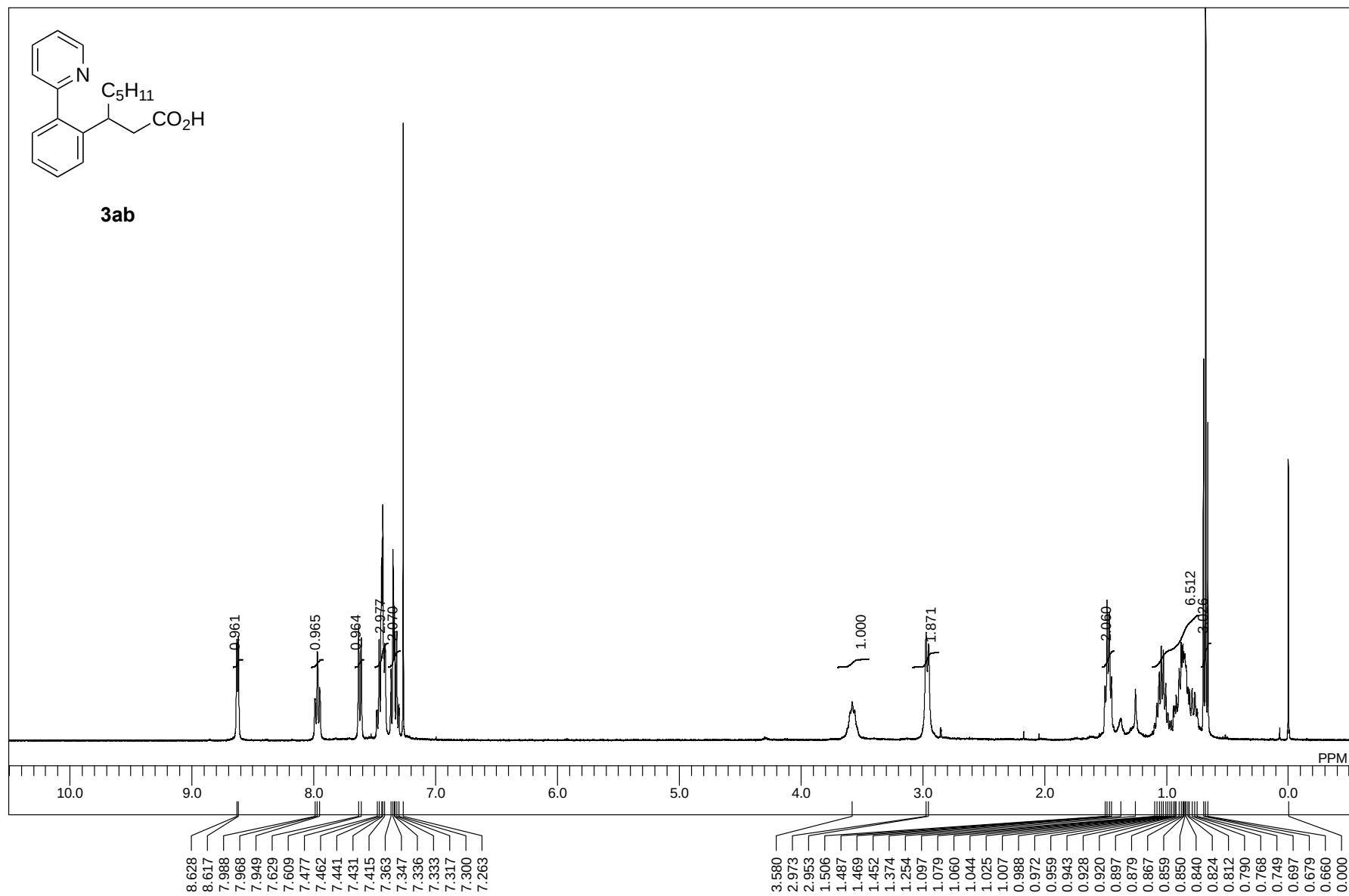


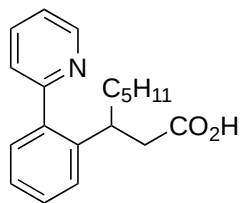
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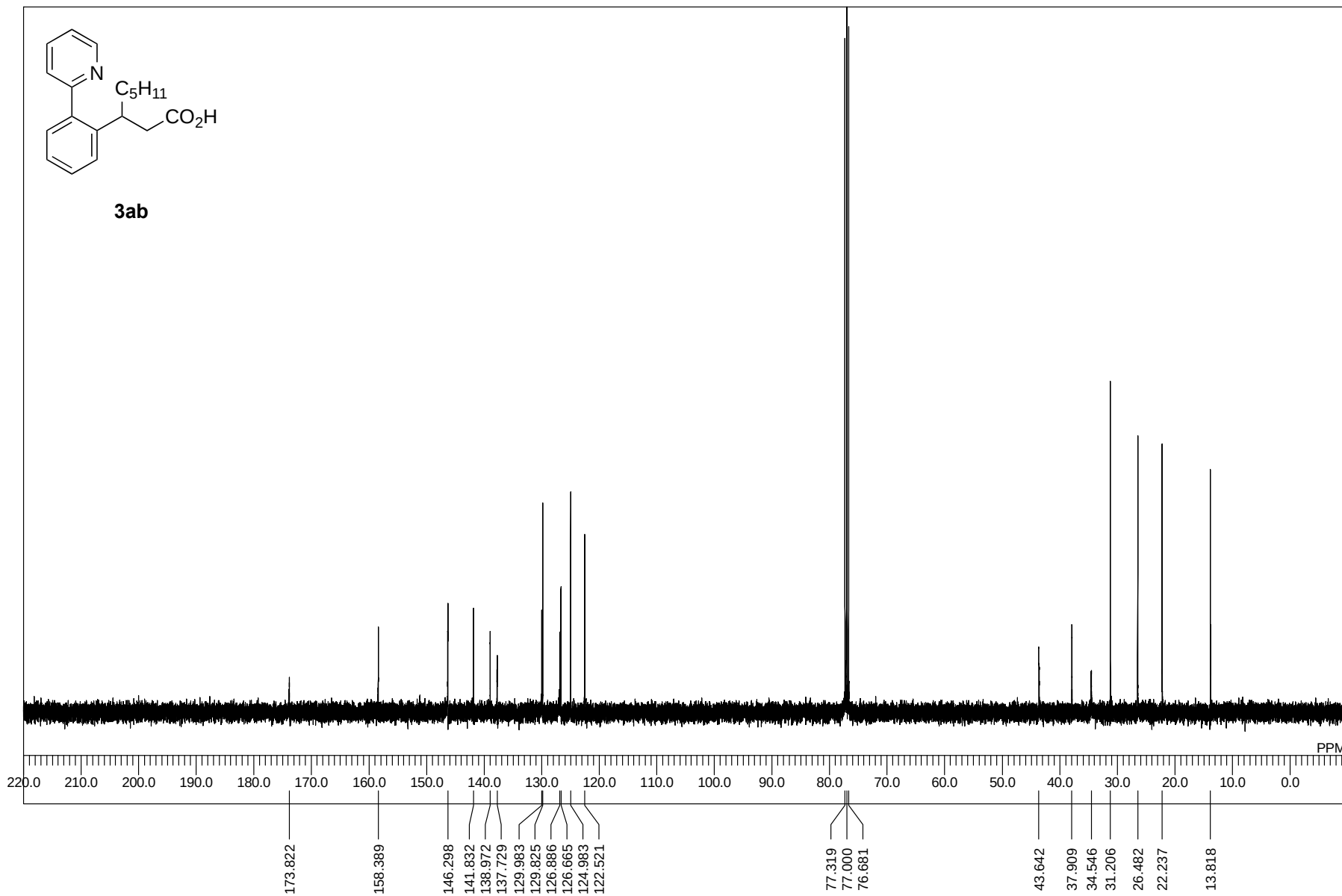


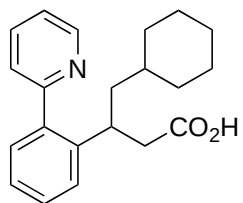
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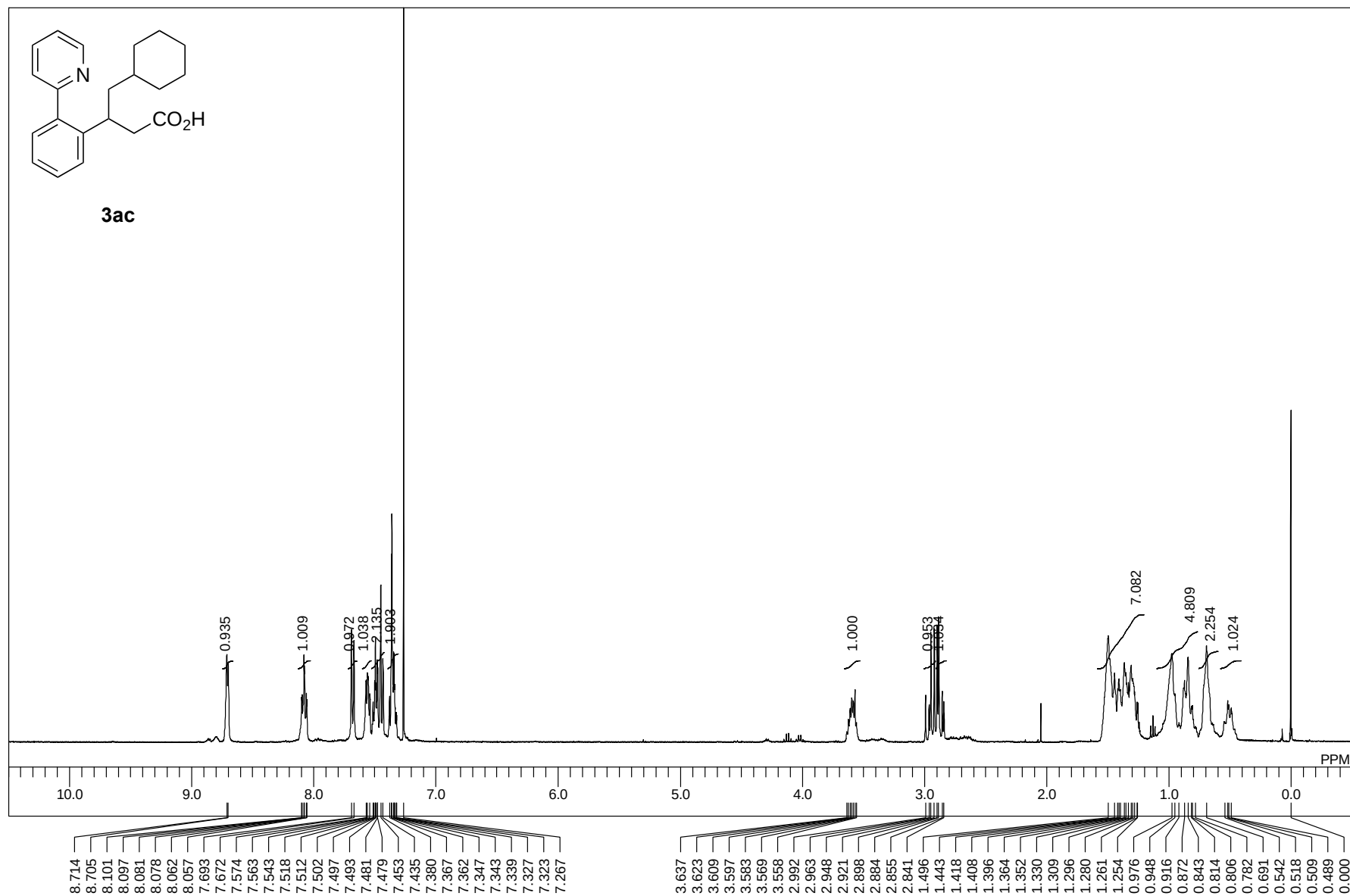


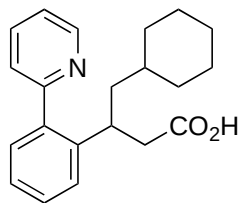
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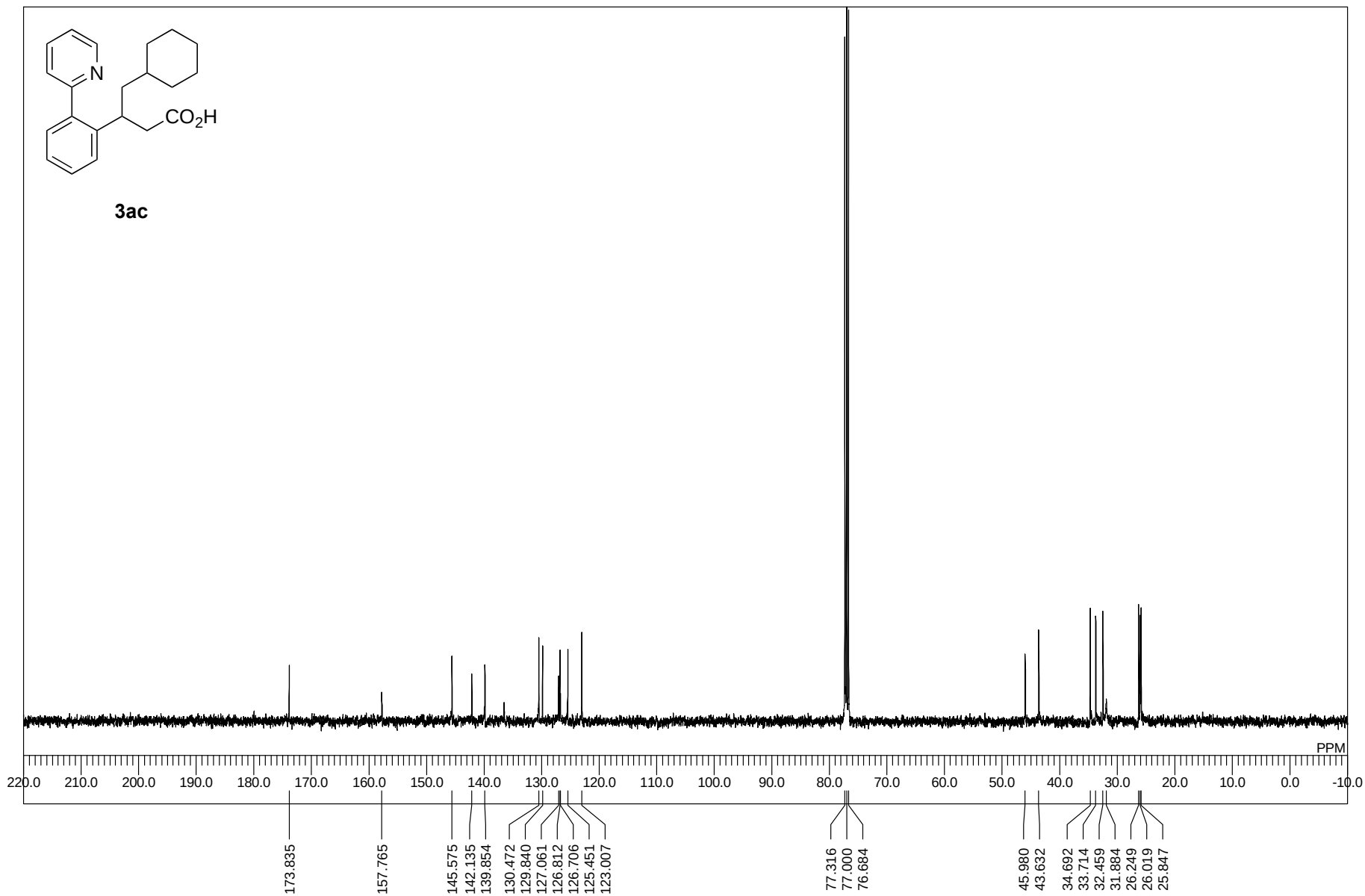


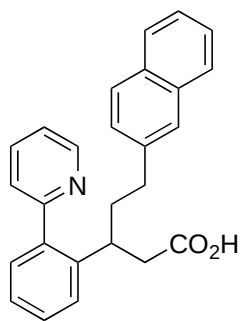
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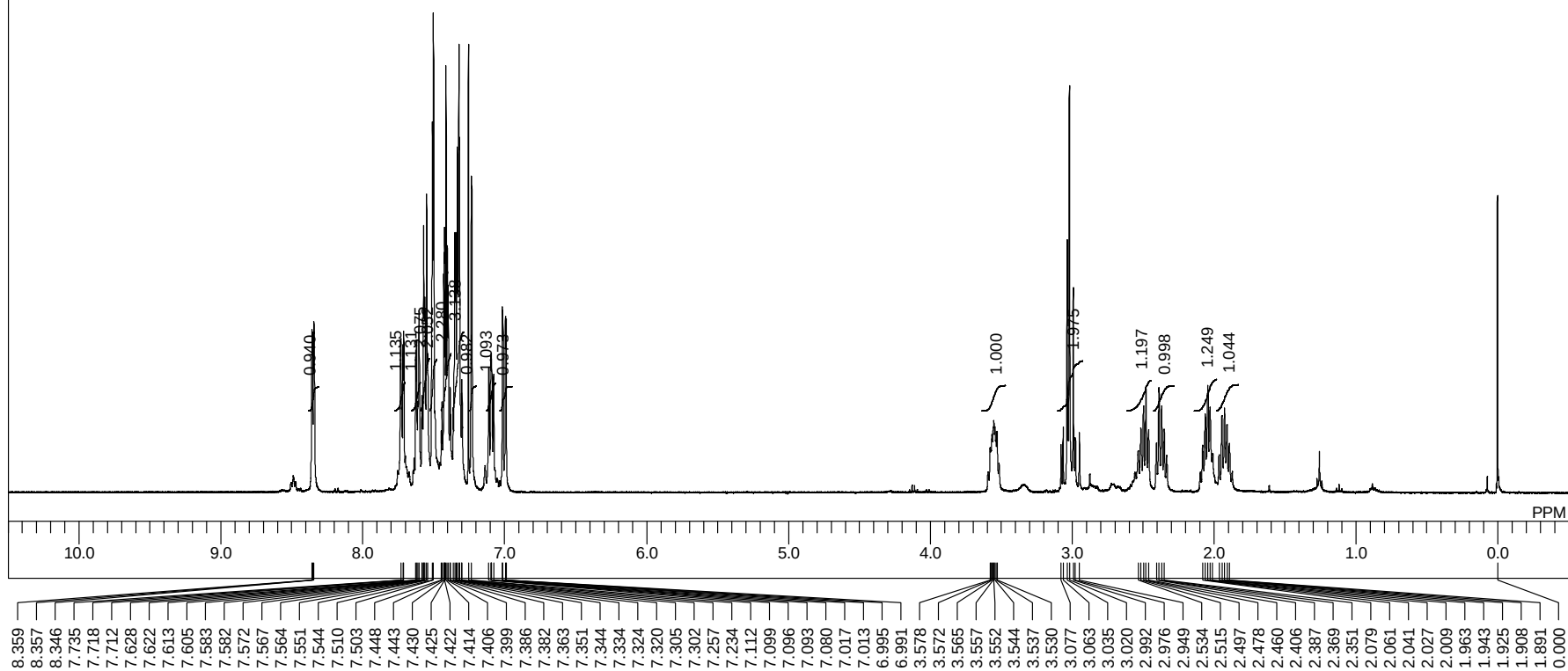


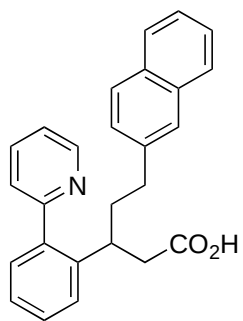
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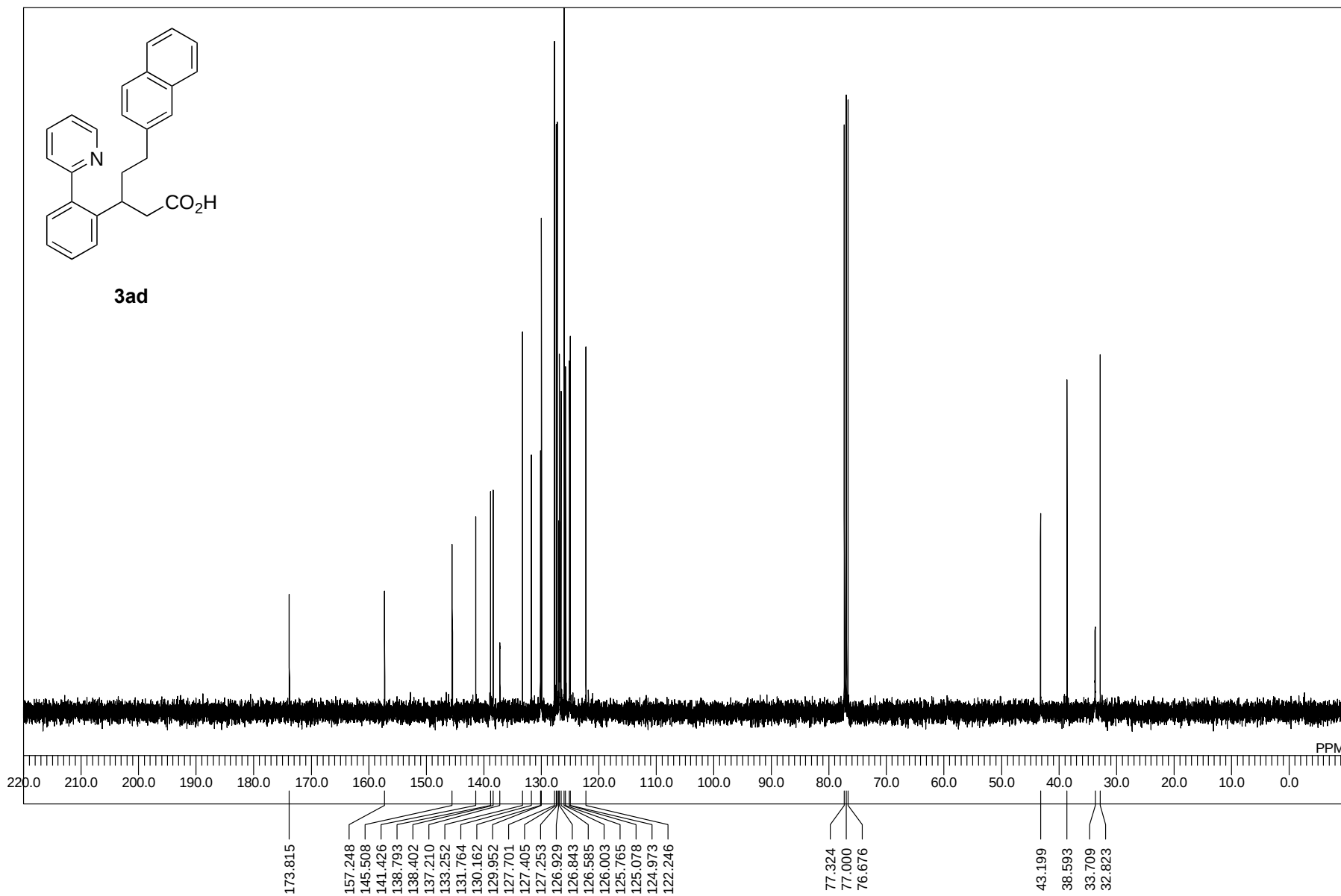


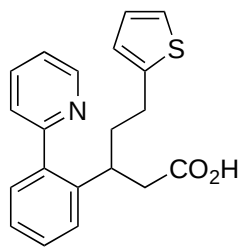
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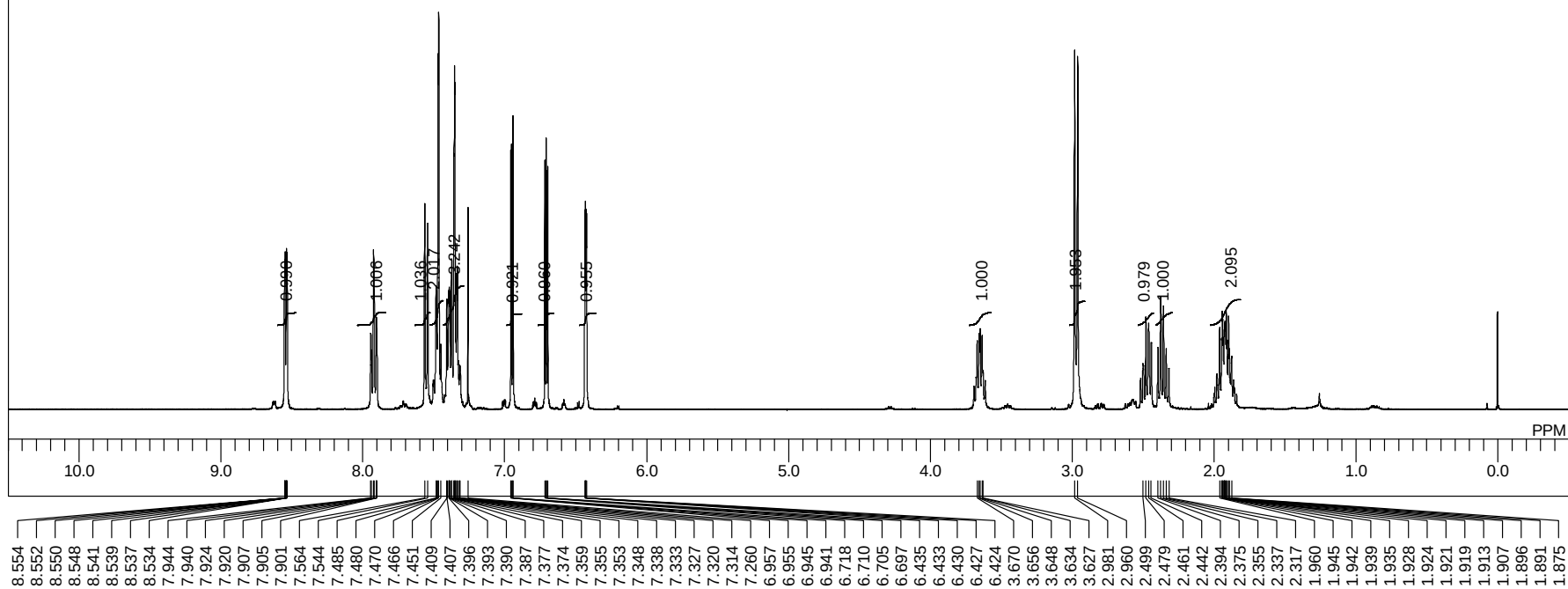


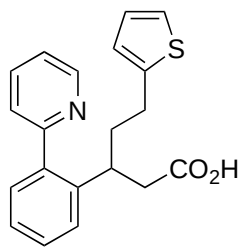
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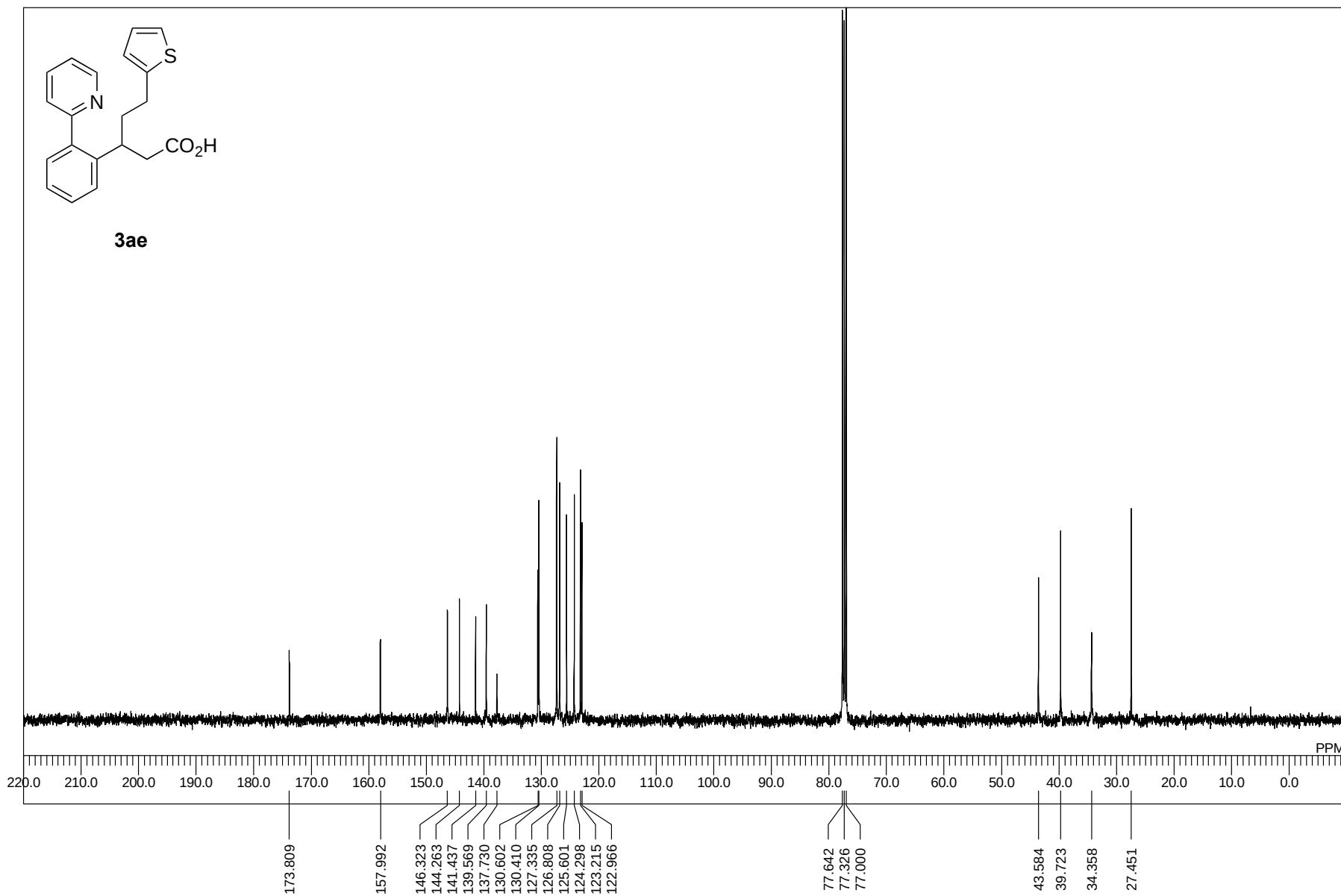


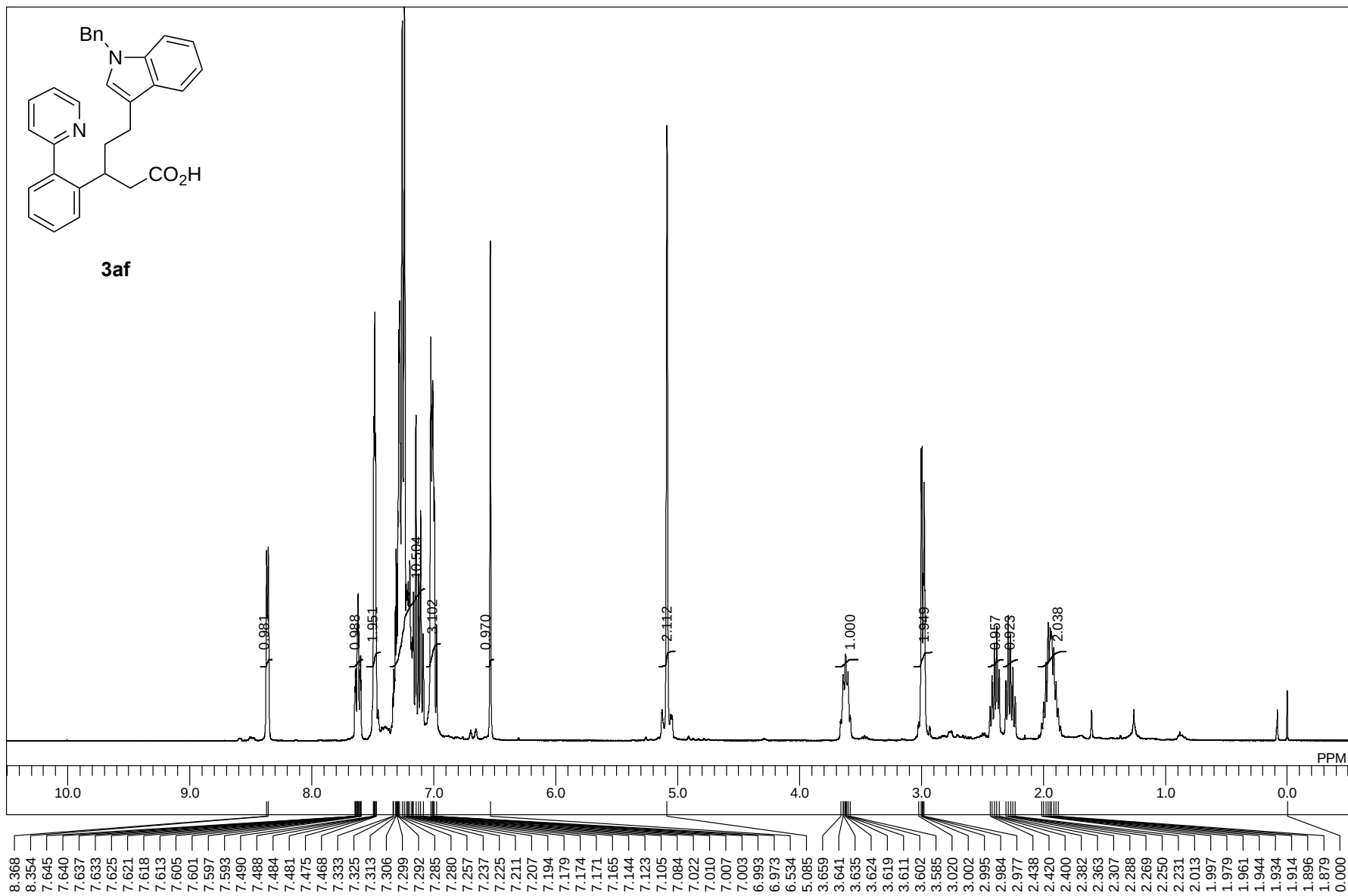
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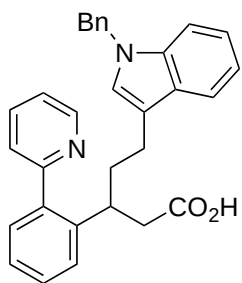




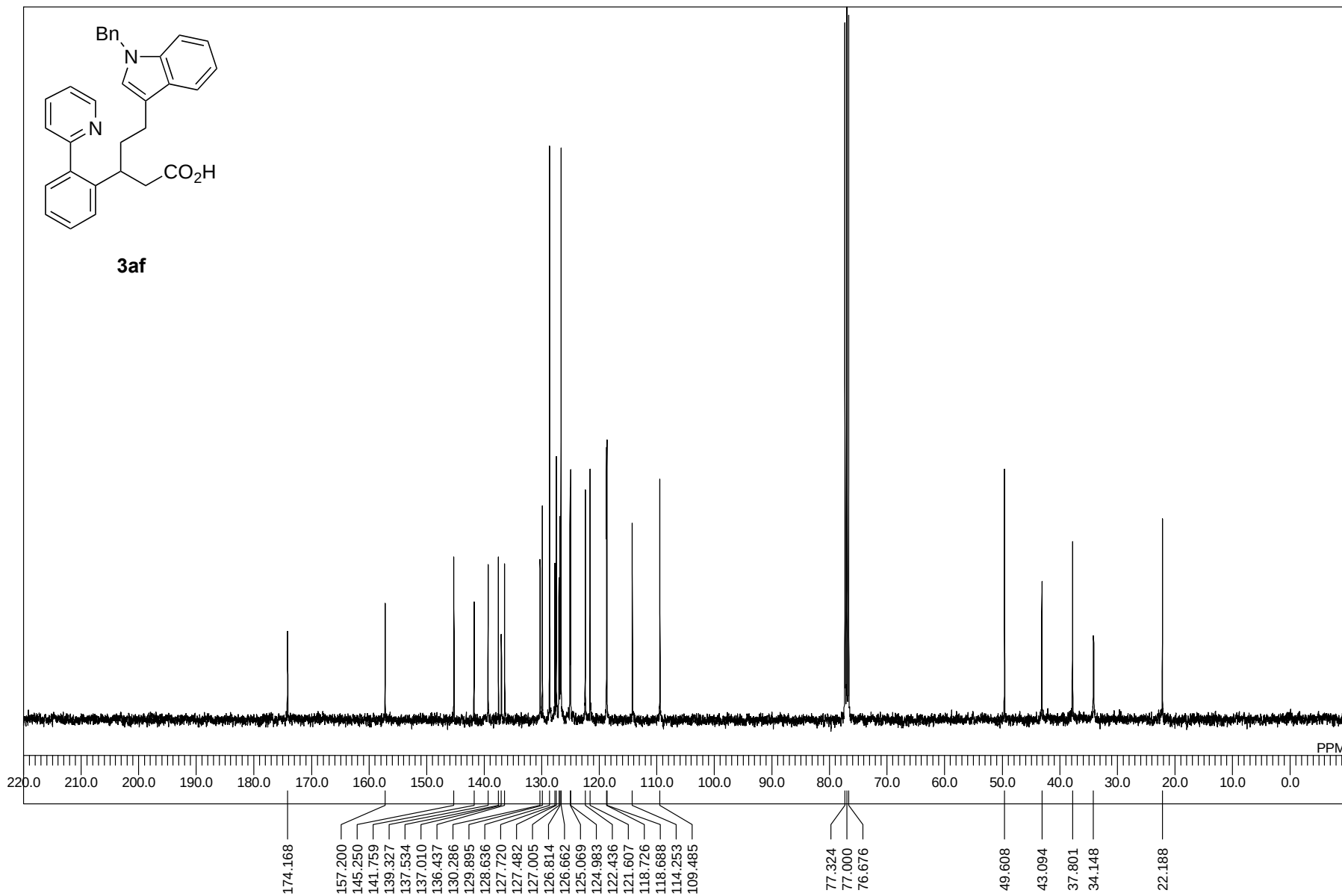
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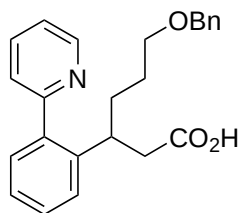




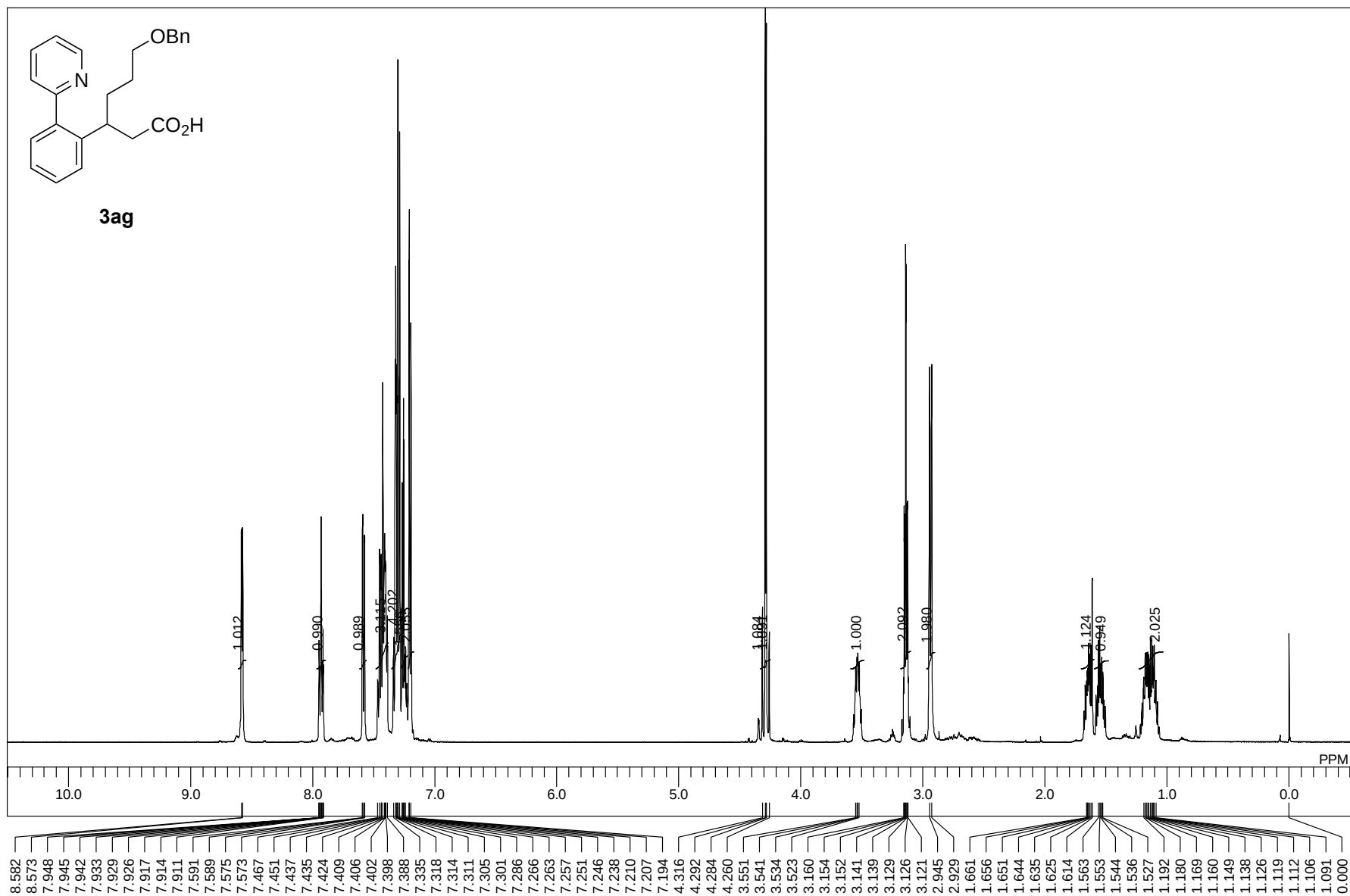


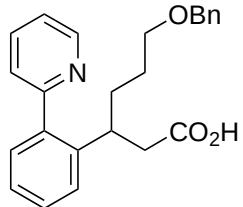
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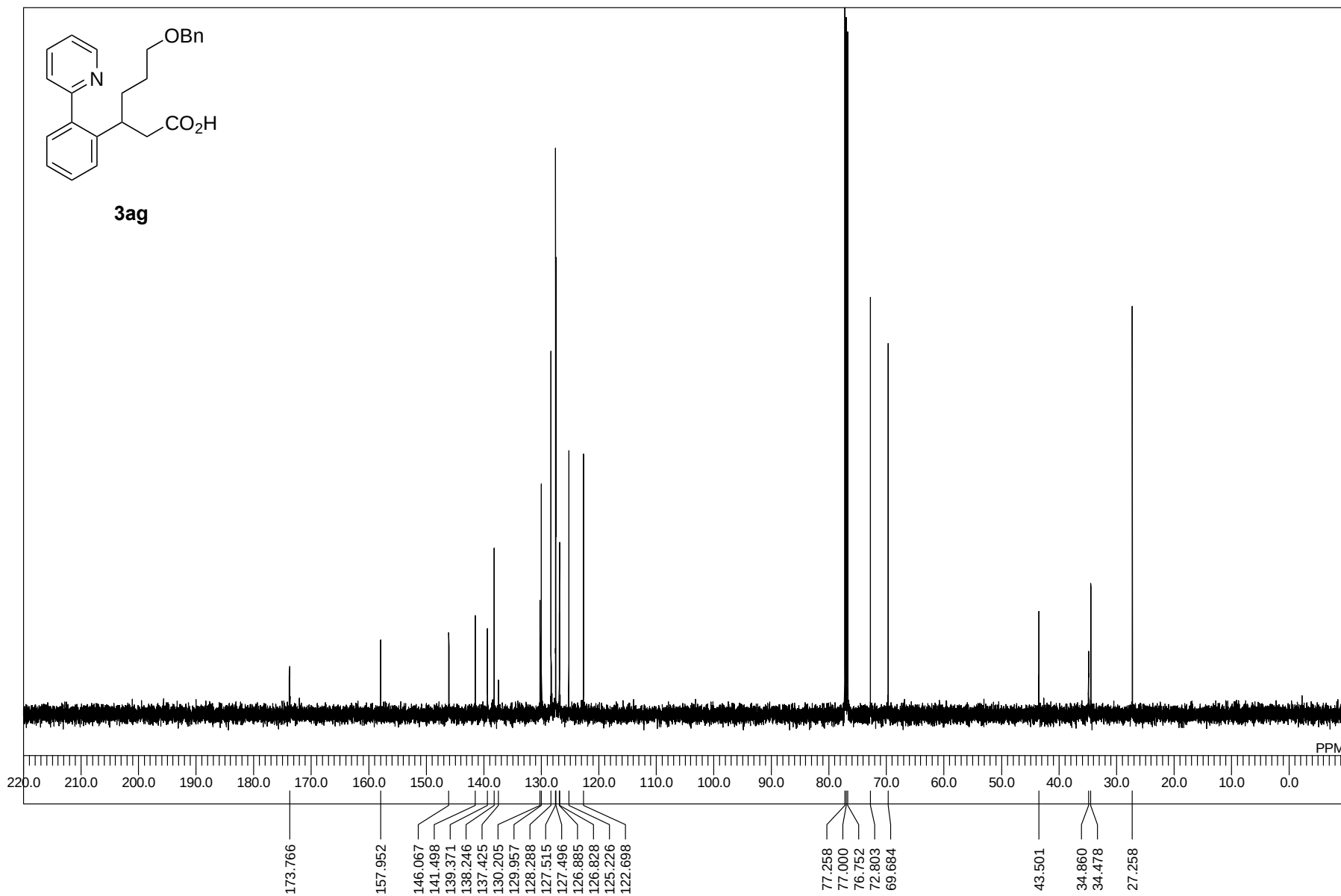


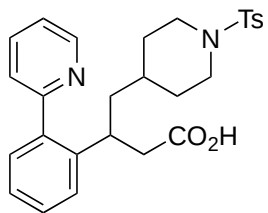
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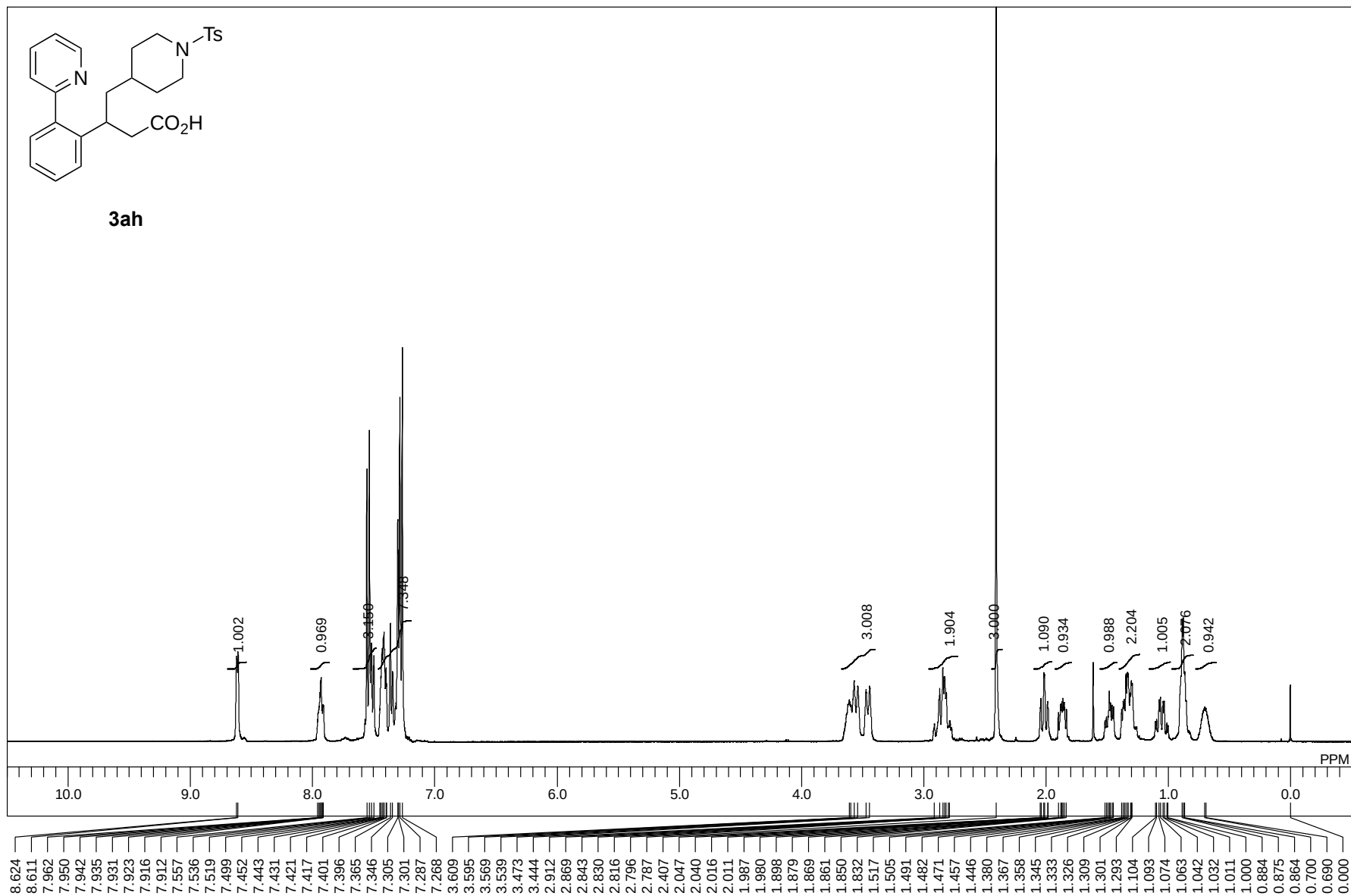


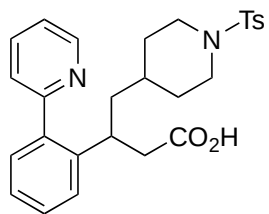
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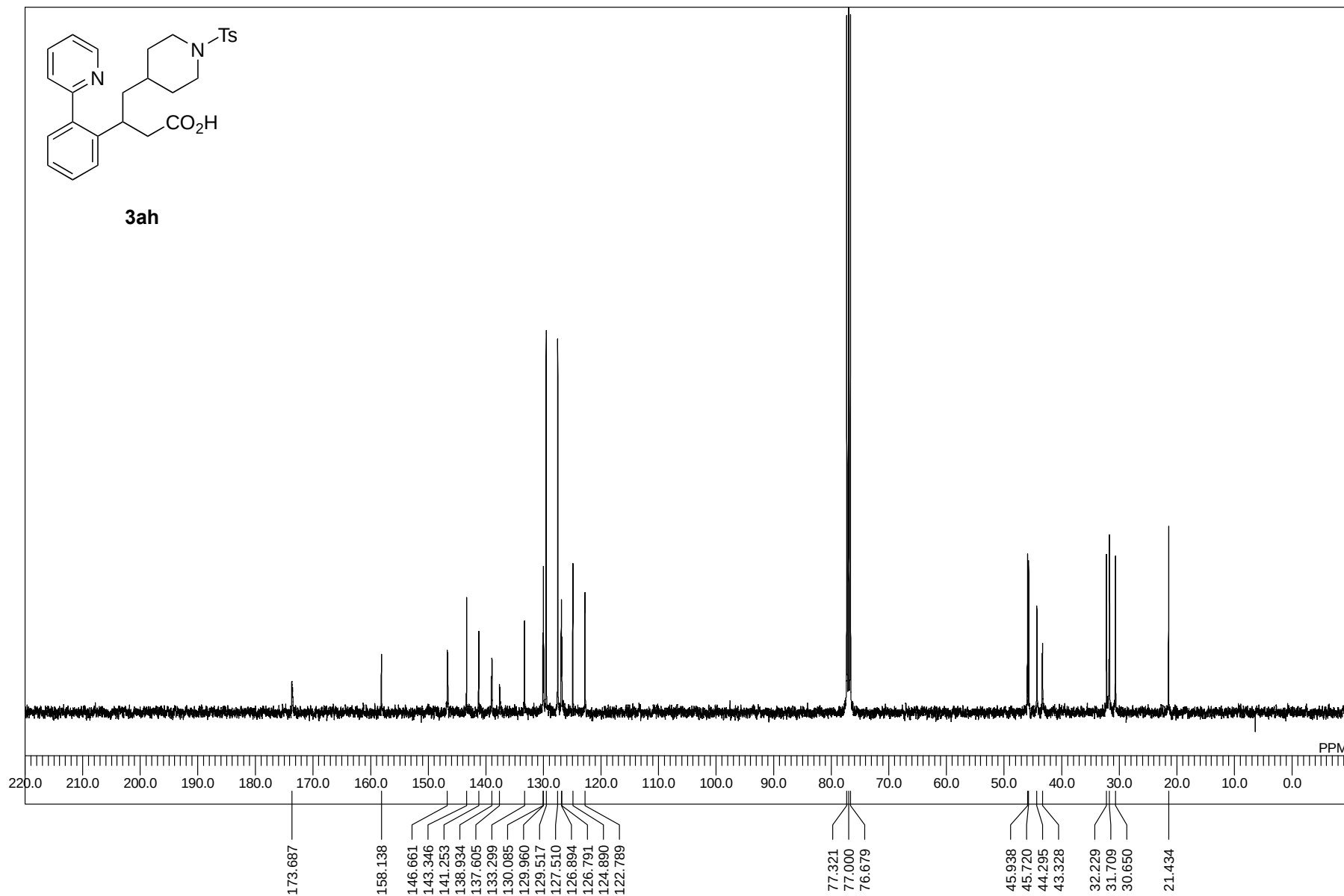


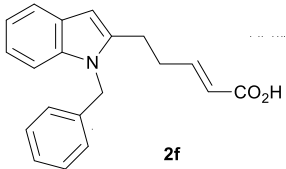
3ah





3ah





202060

Sample No. : C:\Xcalibur\...0222\202000_13147_pn

Instrument : Exactive

Neg

Operator name :

Mobile phase solvent : MeOH

Date : 2/22/2021 5:56:06 PM

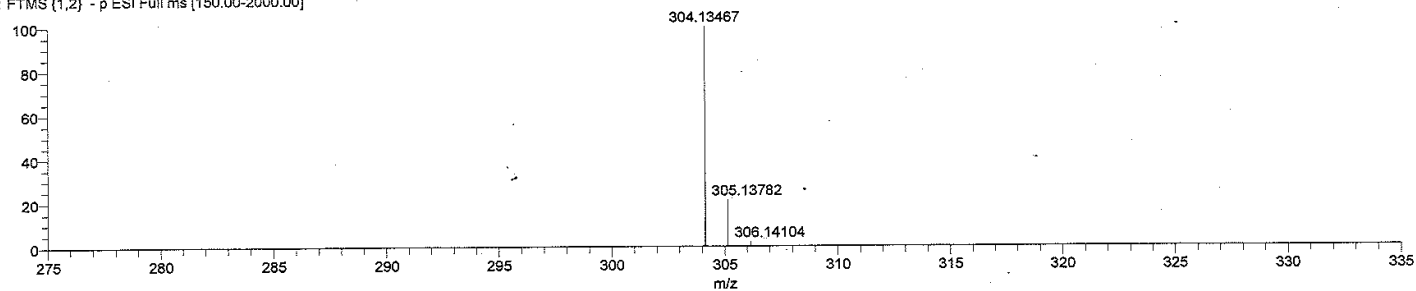
Sample solvent : submitting solution

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

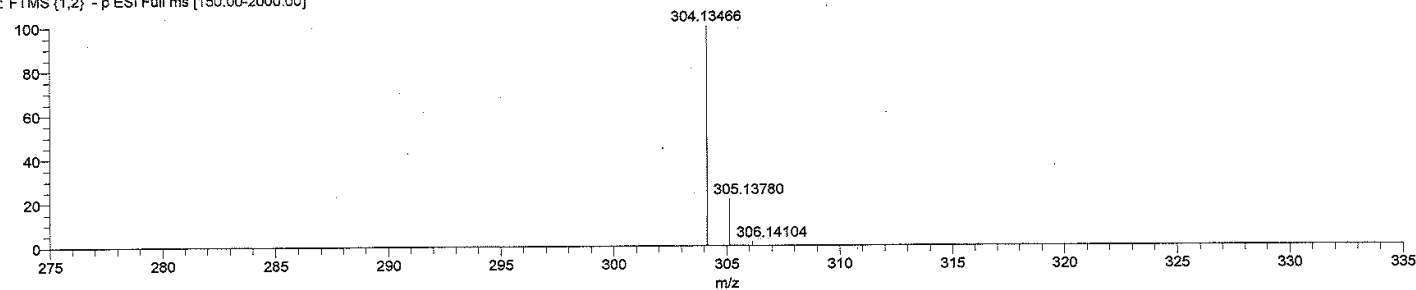
202060_13147_pn #20-23 RT: 0.31-0.33 AV: 2 NL: 7.92E6

T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]



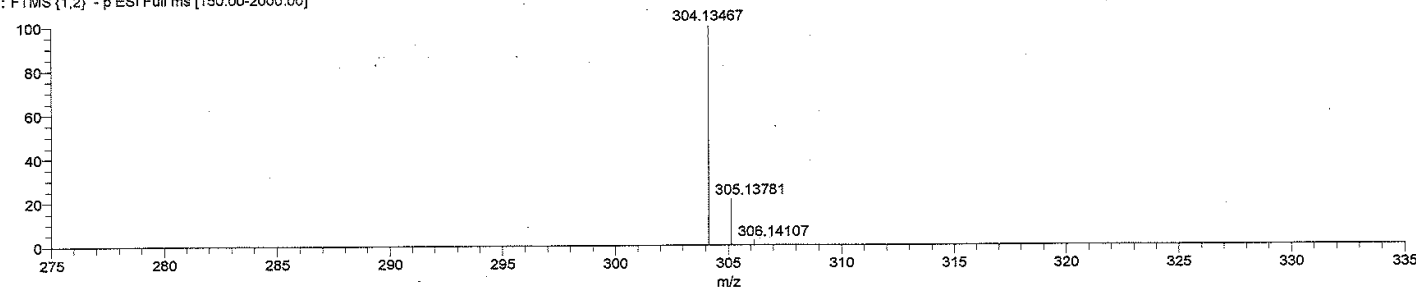
202060_13147_pn #23-27 RT: 0.35-0.38 AV: 2 NL: 3.44E7

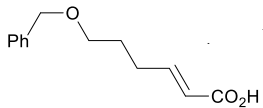
T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]



202060_13147_pn #27-31 RT: 0.40-0.42 AV: 2 NL: 4.95E7

T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]





2g

202059

Sample No. : C:\Xcalibur\...0222\202009_13118_pn

Instrument : Exactive

Neg

Operator name :

Mobile phase solvent : MeOH

Date : 2/22/2021 5:51:03 PM

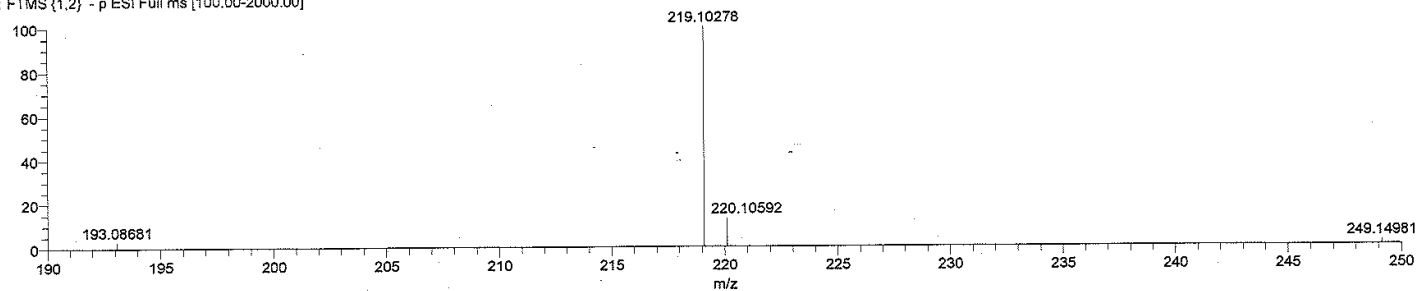
Sample solvent : submitting solution

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30_100.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

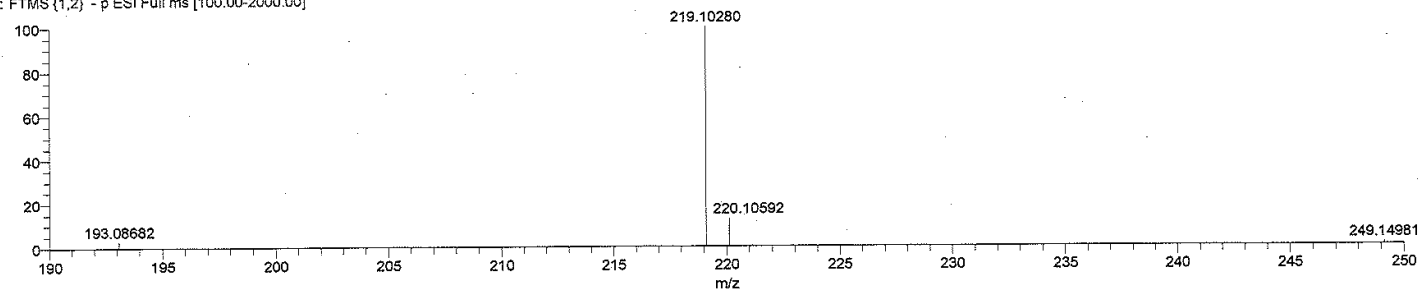
202059_13118_pn #21-24 RT: 0.33-0.35 AV: 2 NL: 2.47E7

T: FTMS (1,2) - p ESI Full ms [100.00-2000.00]



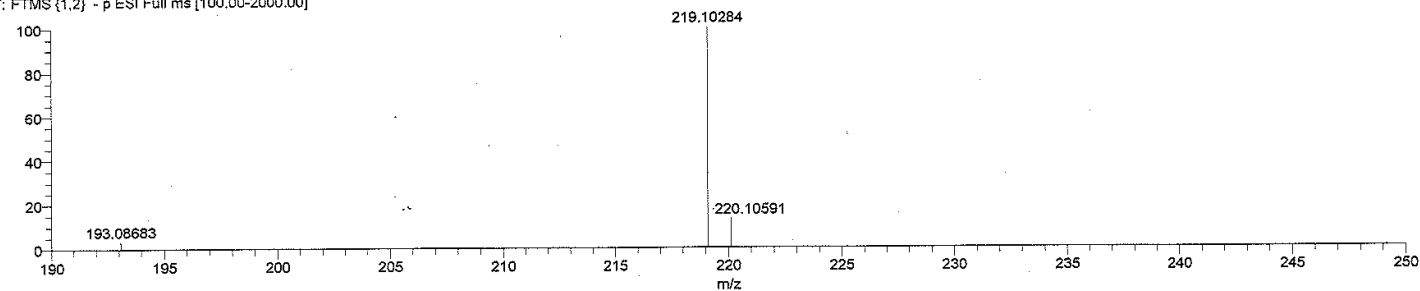
202059_13118_pn #24-27 RT: 0.35-0.37 AV: 2 NL: 4.40E7

T: FTMS (1,2) - p ESI Full ms [100.00-2000.00]



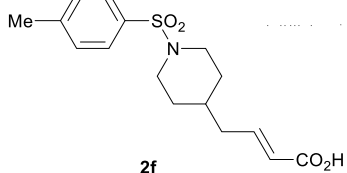
202059_13118_pn #27-31 RT: 0.40-0.42 AV: 2 NL: 7.46E7

T: FTMS (1,2) - p ESI Full ms [100.00-2000.00]



3h

202061



Sample No. : C:\Xcalibur\...0222\202001_13168_pn

Instrument : Exactive

Neg

Operator name :

Mobile phase solvent : MeOH

Date : 2/22/2021 6:01:08 PM

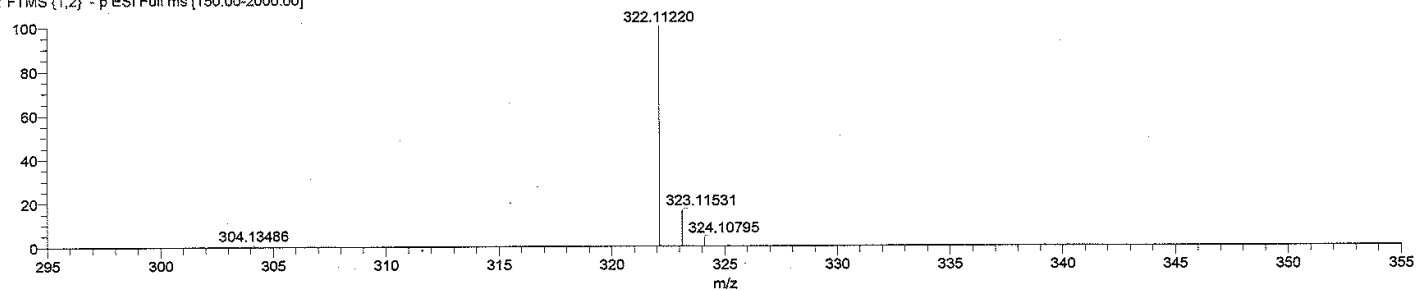
Sample solvent : submitting solution

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

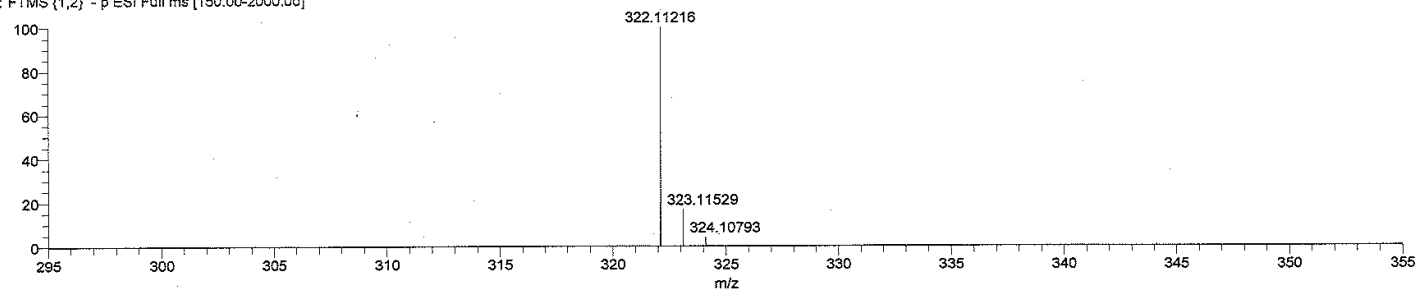
202061_13168_pn#19-22 RT: 0.31-0.34 AV: 2 NL: 3.49E6

T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]



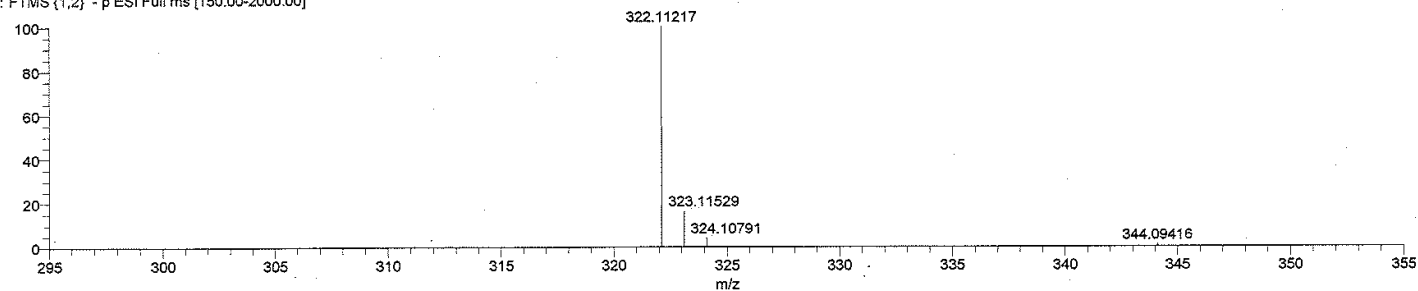
202061_13168_pn#22-27 RT: 0.34-0.38 AV: 3 NL: 1.22E7

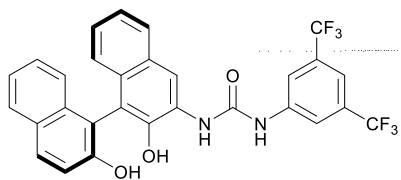
T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]



202061_13168_pn#27-30 RT: 0.41-0.43 AV: 2 NL: 2.21E7

T: FTMS (1,2) - p ESI Full ms [150.00-2000.00]





5

Sample No. : C:\Xcalibur\...201419_BINOL_urea_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

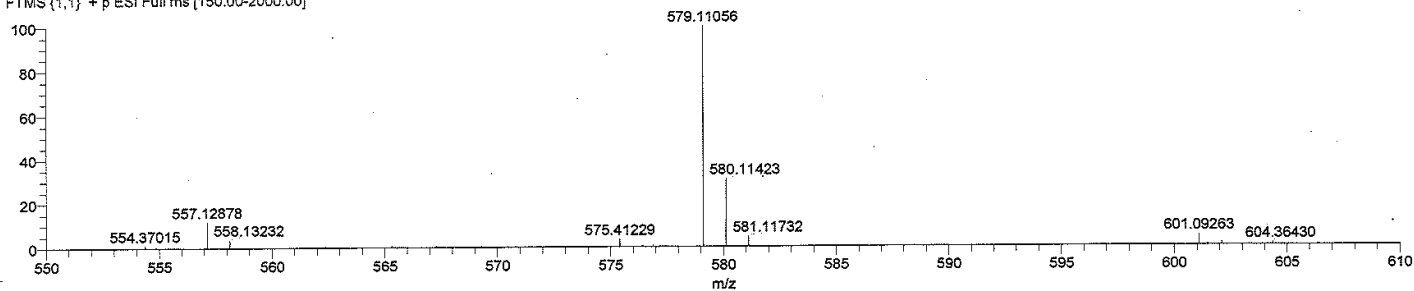
Date : 12/2/2020 12:20:13 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

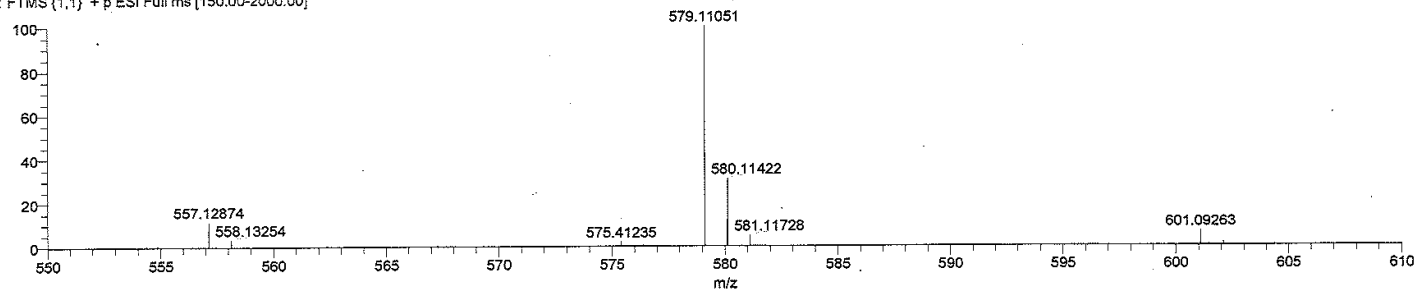
201419_BINOL_urea_pn #16-21 RT: 0.26-0.31 AV: 3 NL: 6.77E5

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



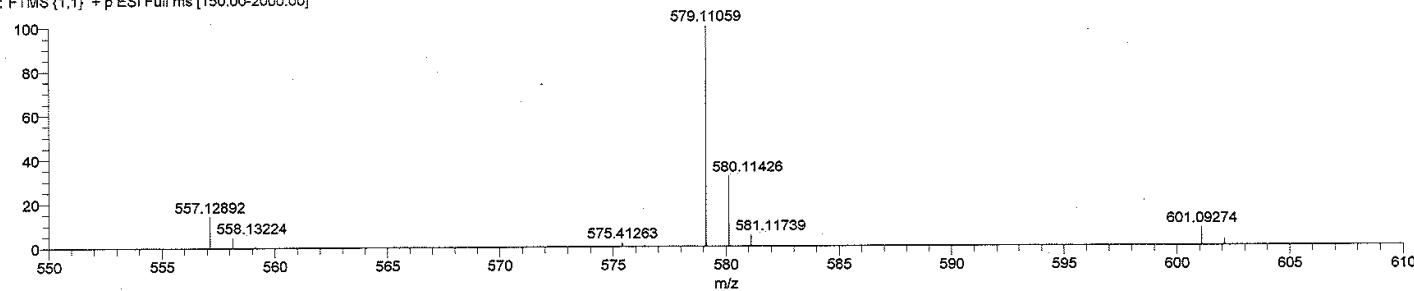
201419_BINOL_urea_pn #21-25 RT: 0.31-0.35 AV: 3 NL: 5.20E5

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]

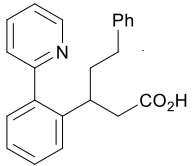


201419_BINOL_urea_pn #26-29 RT: 0.35-0.41 AV: 3 NL: 2.91E5

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



300



3aa

Sample No. : C:\Xcalibur\...0820\200000_RT12170_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hirose tomohiro

Sample solvent : submitting solution

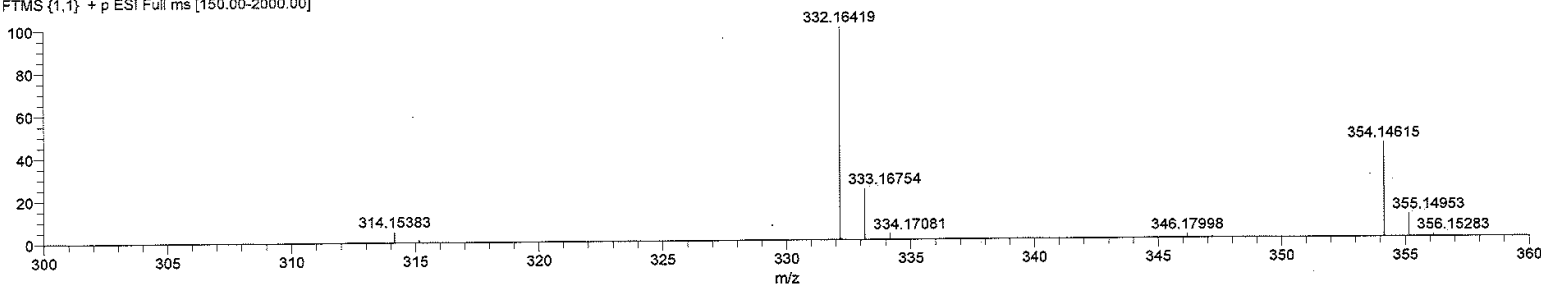
Date : 8/20/2020 9:36:49 AM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

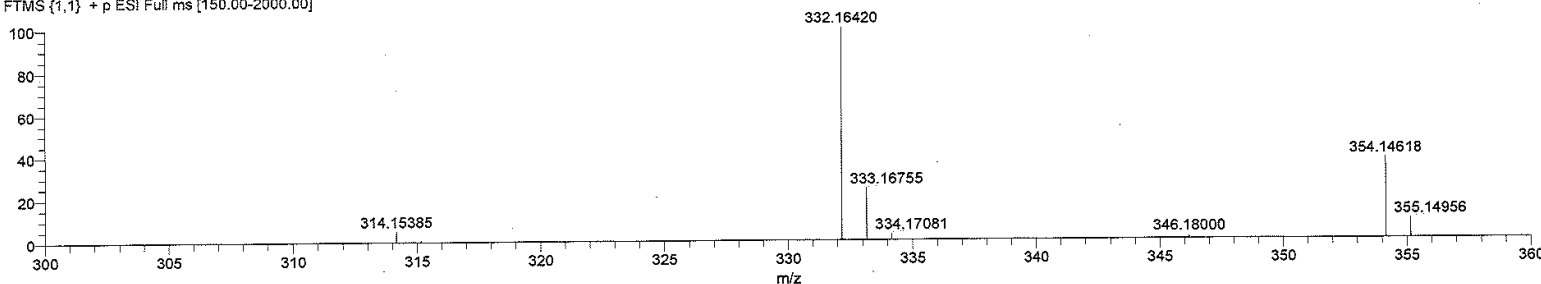
200600_RT12170_pn #21-25 RT: 0.25-0.29 AV: 3 NL: 1.20E8

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



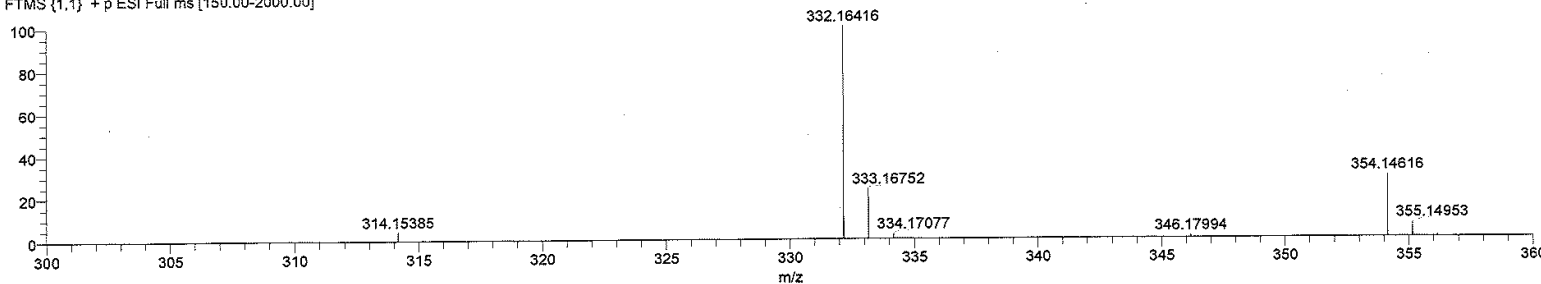
200600_RT12170_pn #25-30 RT: 0.29-0.34 AV: 3 NL: 1.30E8

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



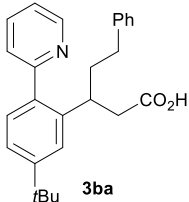
200600_RT12170_pn #30-34 RT: 0.36-0.38 AV: 2 NL: 1.46E8

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



3ba

201251



Sample No. : C:\Xcalibur\11119\201251_RT13110_a_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

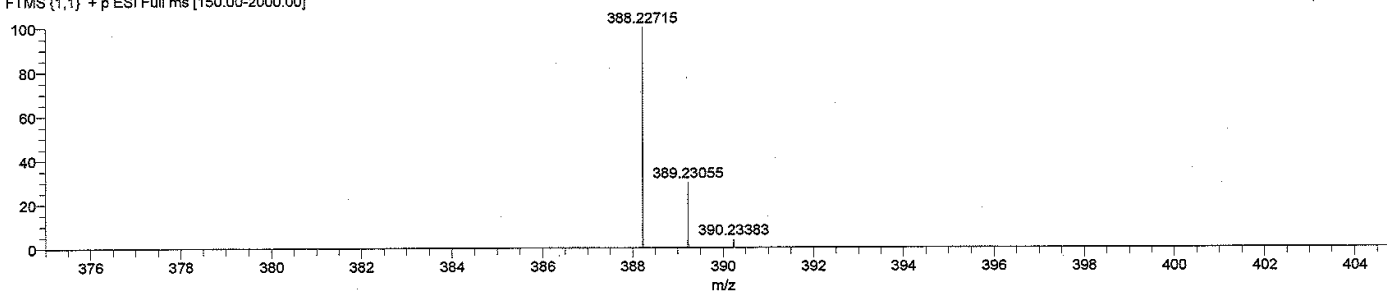
Date : 11/19/2020 2:35:01 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

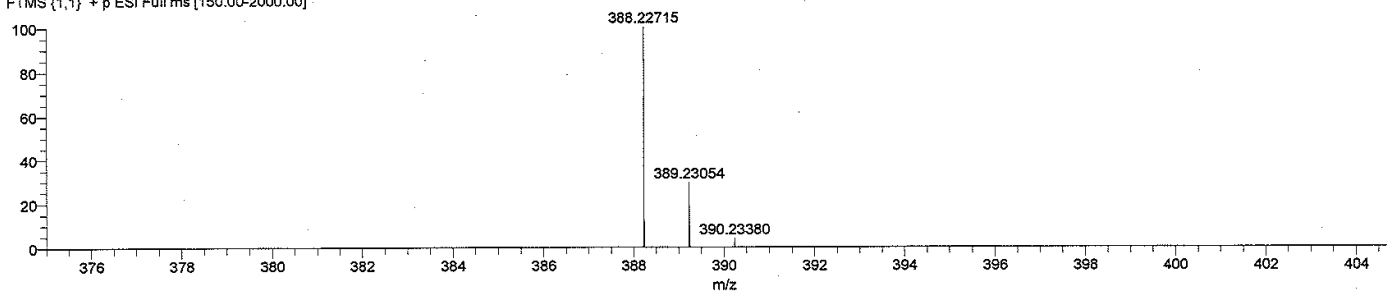
201251_RT13110_a_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 2.10E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



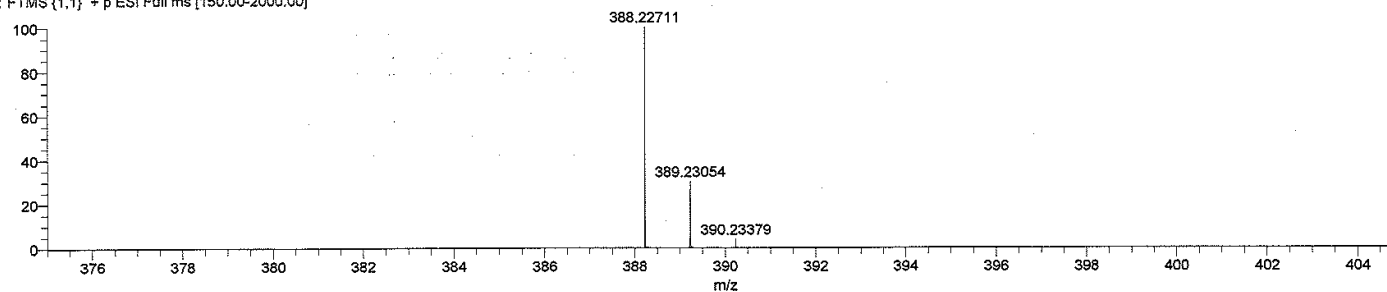
201251_RT13110_a_pn #21-26 RT: 0.30-0.34 AV: 3 NL: 1.63E7

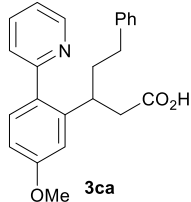
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201251_RT13110_a_pn #26-30 RT: 0.37-0.39 AV: 2 NL: 1.07E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





201247

3ca

Sample No. : C:\Xcalibur\...201247...13107_a_pn2

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

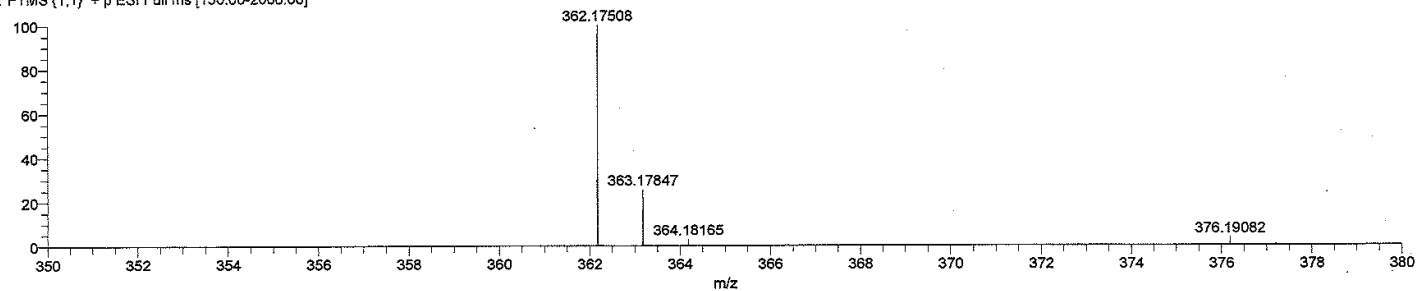
Date : 11/19/2020 2:24:55 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

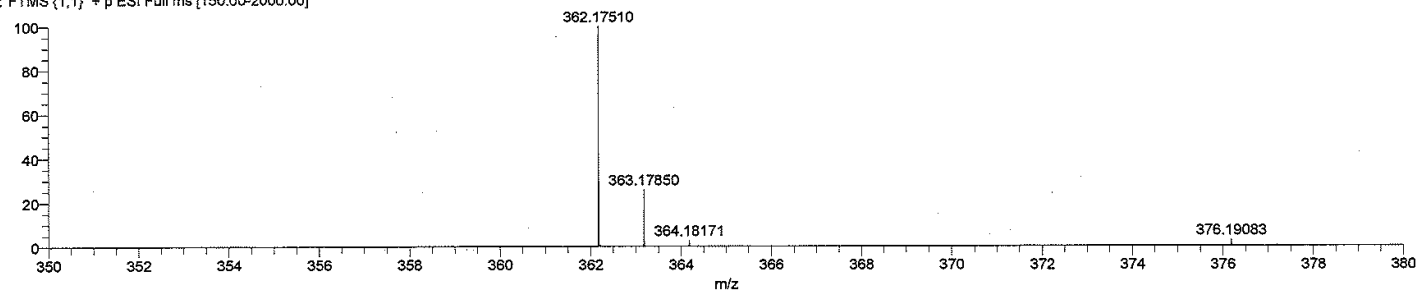
201247_RT13107_a_pn2 #17-21 RT: 0.25-0.30 AV: 3 NL: 1.24E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



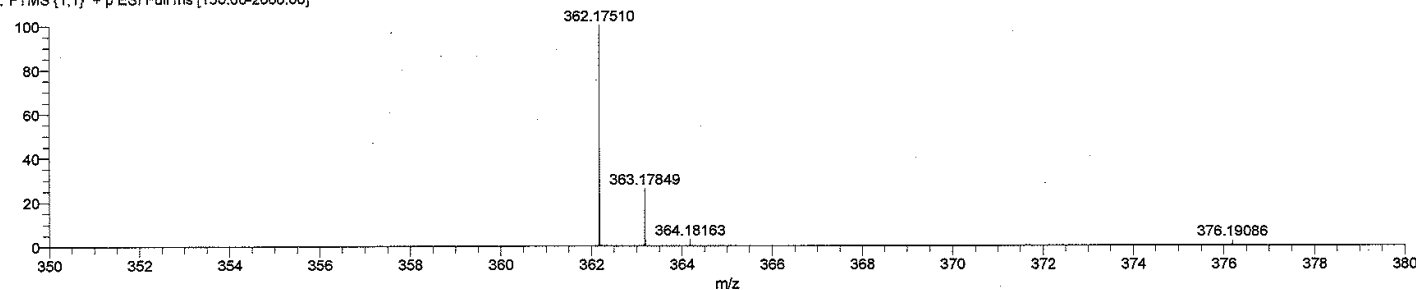
201247_RT13107_a_pn2 #21-25 RT: 0.30-0.35 AV: 3 NL: 9.83E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



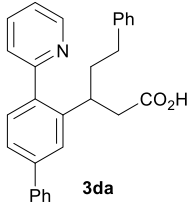
201247_RT13107_a_pn2 #25-30 RT: 0.35-0.39 AV: 3 NL: 7.94E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



3/a

201252



Sample No. : C:\Xcalibur\...1119\201202_RT13110_b_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Sample solvent : submitting solution

Operator name : hayashi harumi

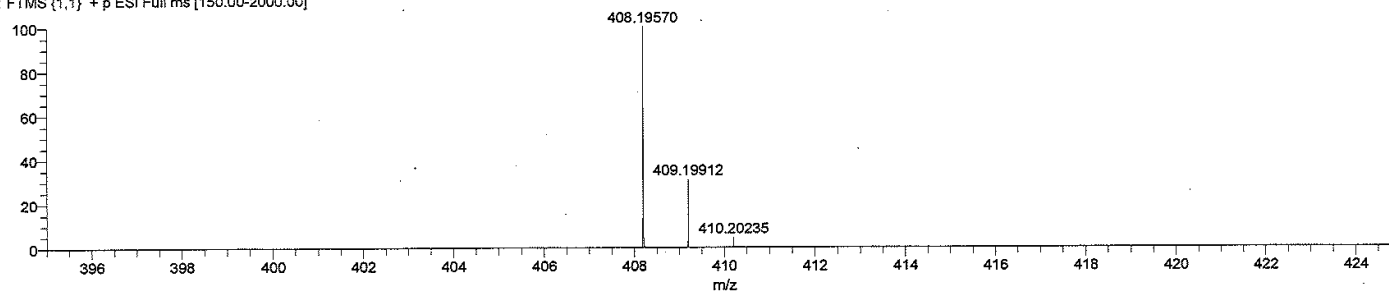
Date : 11/19/2020 2:40:03 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

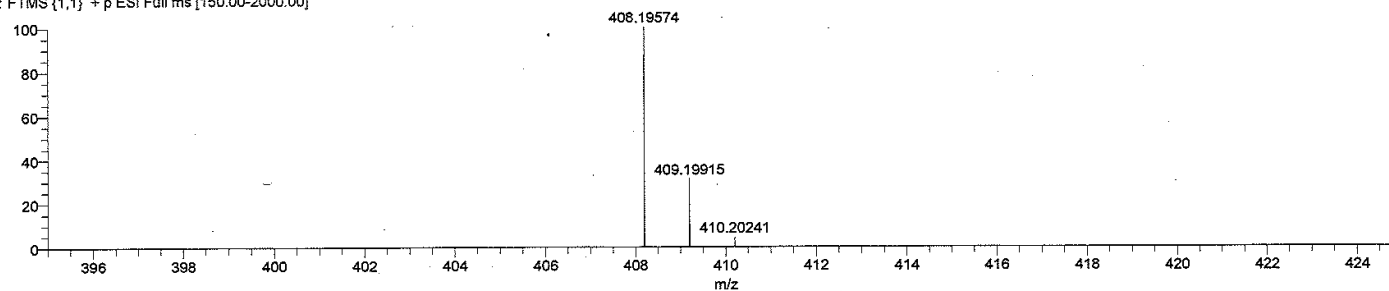
201252_RT13110_b_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 1.62E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



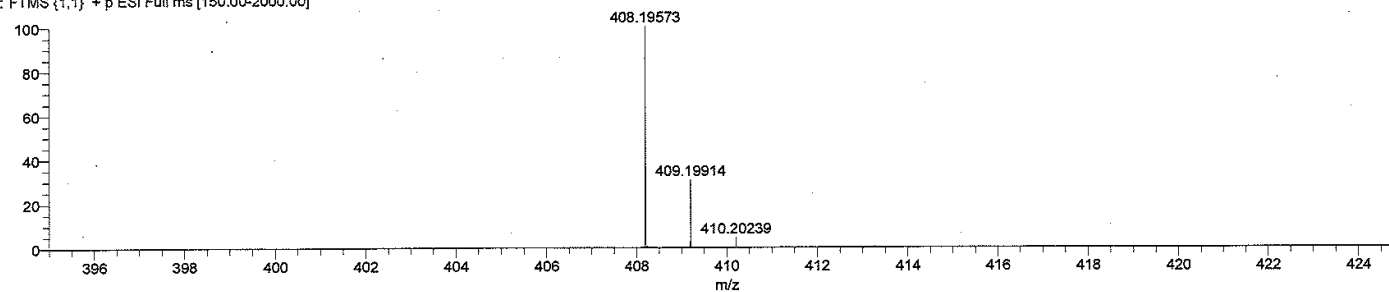
201252_RT13110_b_pn #21-25 RT: 0.30-0.35 AV: 3 NL: 1.20E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



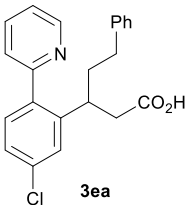
201252_RT13110_b_pn #25-30 RT: 0.35-0.39 AV: 3 NL: 9.66E6

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



3ea

201246



Sample No. : C:\Xcalibur\...1119\201246_RT13102_a_pn

Instrument : Exacte

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

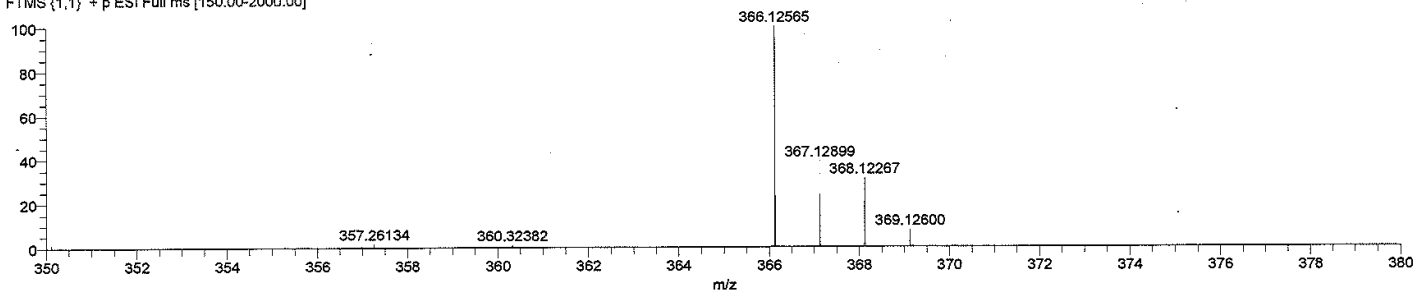
Date : 11/19/2020 1:30:55 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

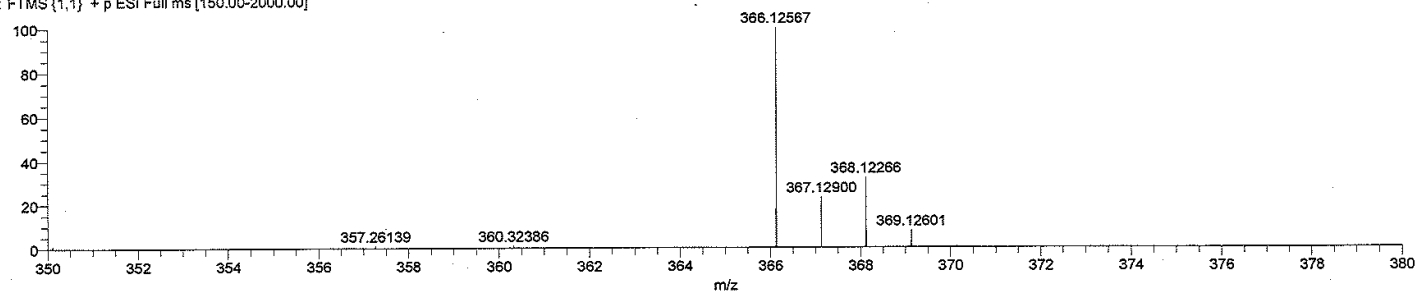
201246_RT13102_a_pn #16-19 RT: 0.27-0.30 AV: 2 NL: 1.63E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



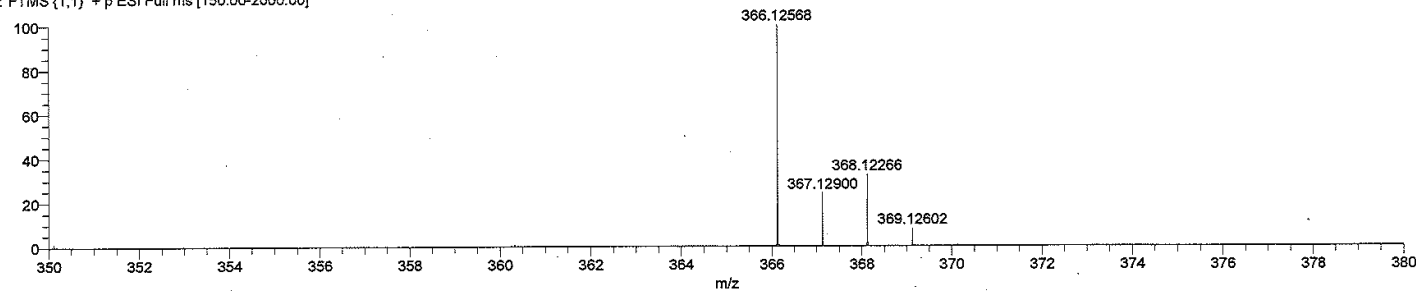
201246_RT13102_a_pn #19-23 RT: 0.30-0.35 AV: 3 NL: 1.22E6

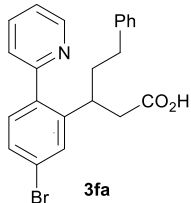
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201246_RT13102_a_pn #23-26 RT: 0.35-0.38 AV: 2 NL: 7.85E5

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201250

Sample No. : C:\Xcalibur\...1119\201250_RT13104_a_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

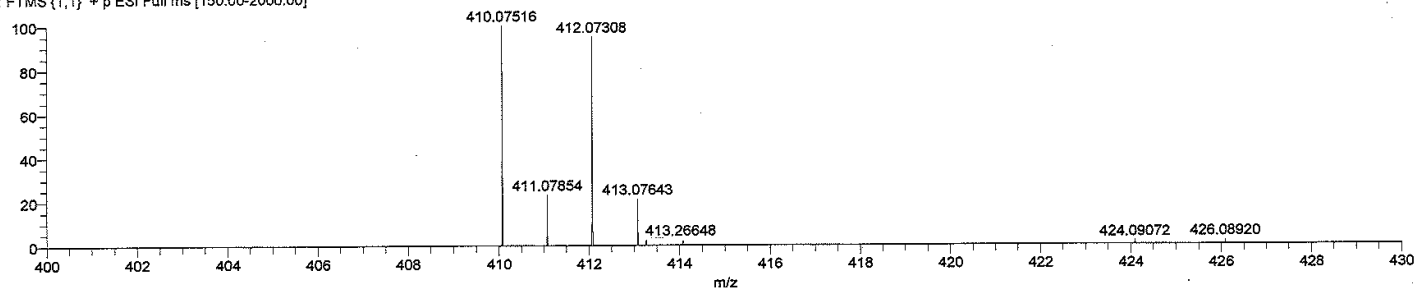
Date : 11/19/2020 2:29:59 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

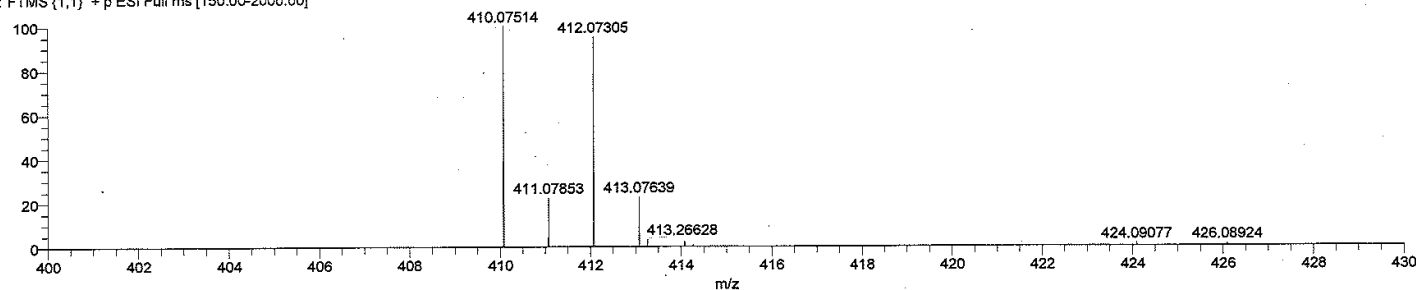
201250_RT13104_a_pn#17-22 RT: 0.25-0.29 AV: 3 NL: 4.69E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



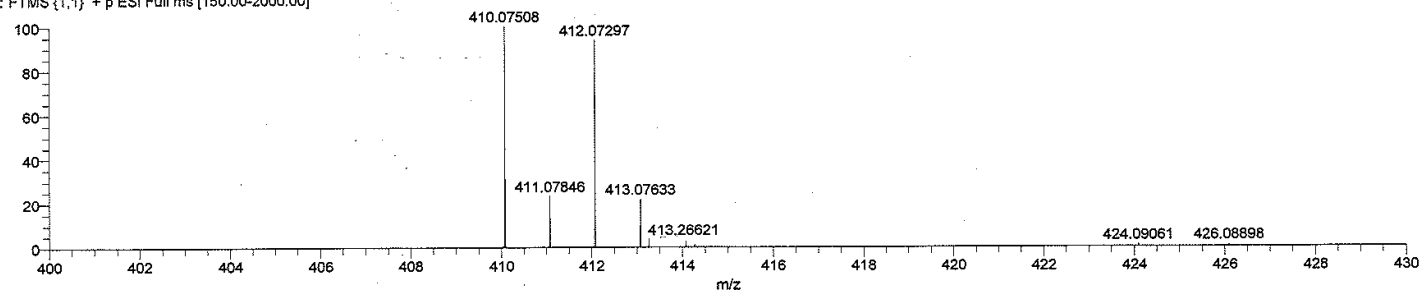
201250_RT13104_a_pn#22-26 RT: 0.32-0.34 AV: 2 NL: 3.53E6

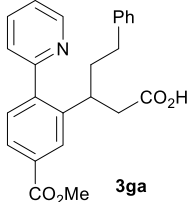
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201250_RT13104_a_pn#26-30 RT: 0.36-0.39 AV: 2 NL: 3.32E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201248

Sample No. : C:\Xcalibur\...1119\201248_RT13104_b_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Sample solvent : submitting solution

Operator name : hayashi harumi

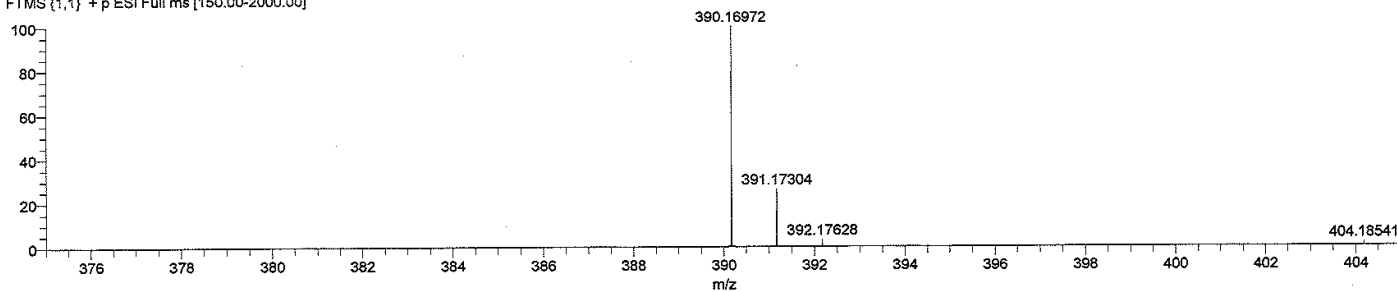
Date : 11/19/2020 1:56:47 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

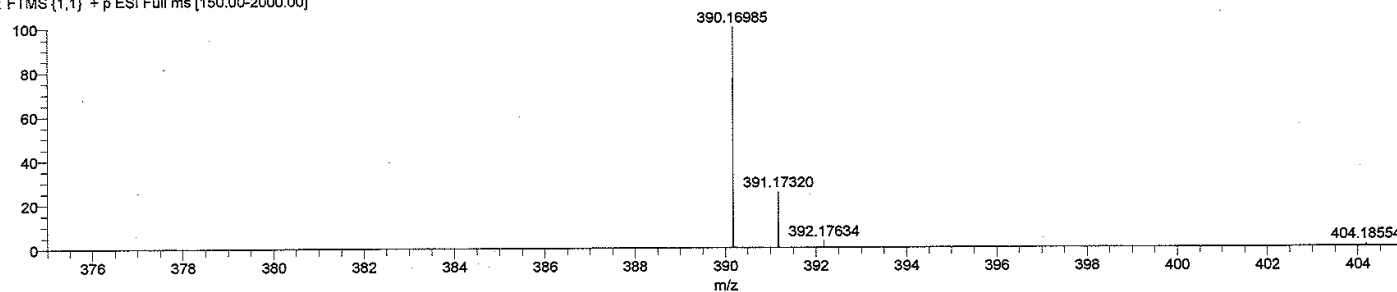
201248_RT13104_b_pn #17-22 RT: 0.25-0.29 AV: 3 NL: 9.95E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



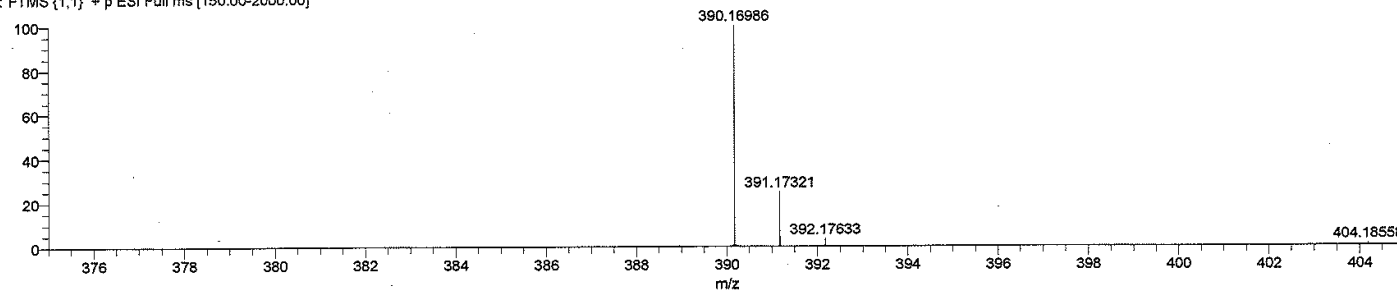
201248_RT13104_b_pn #22-26 RT: 0.32-0.34 AV: 2 NL: 9.82E6

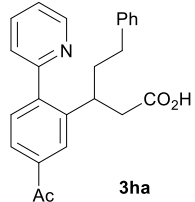
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201248_RT13104_b_pn #26-30 RT: 0.36-0.38 AV: 2 NL: 7.98E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201249

3ha

Sample No. : C:\Xcalibur\...1119\2012-2-9_RT13107_b_pn

Instrument : Exacte

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

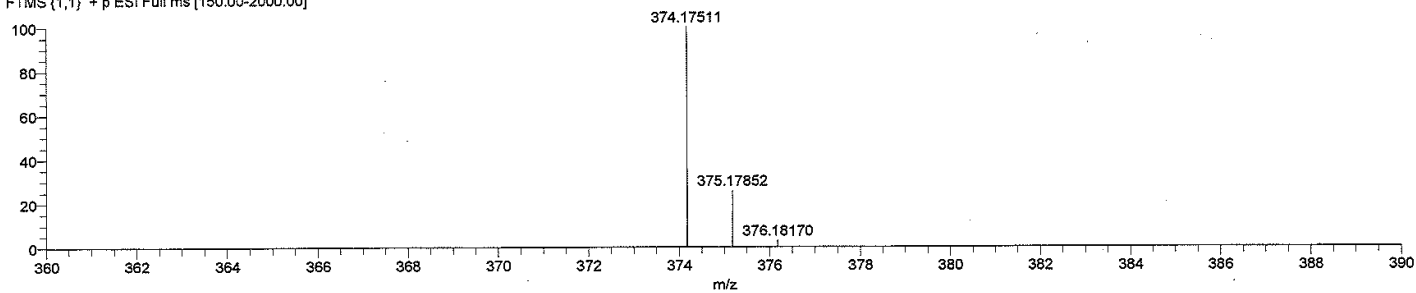
Date : 11/19/2020 2:01:49 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

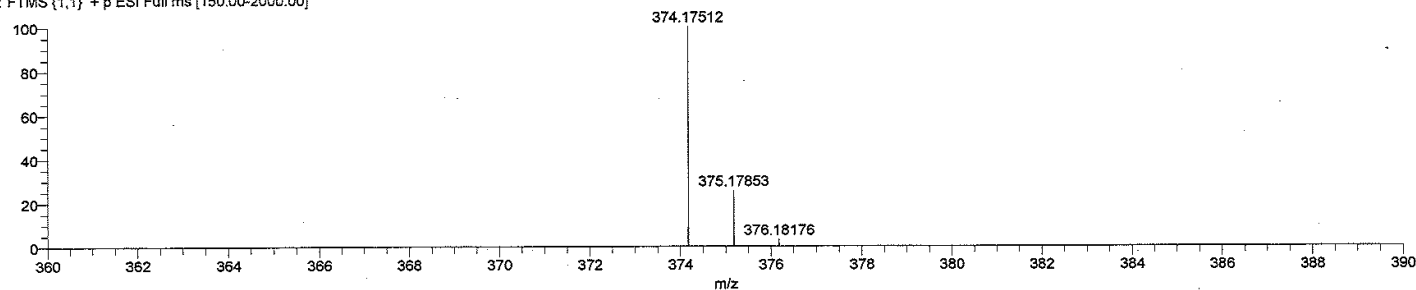
201249_RT13107_b_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 8.40E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



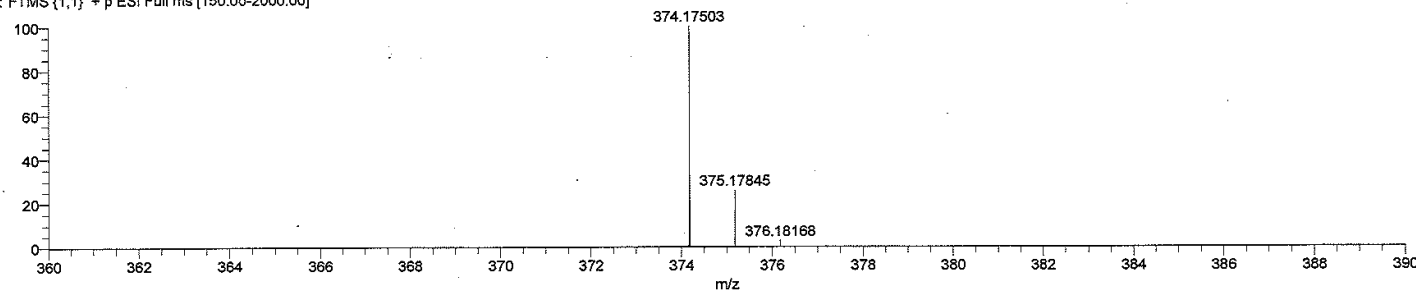
201249_RT13107_b_pn #21-26 RT: 0.30-0.34 AV: 3 NL: 7.23E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



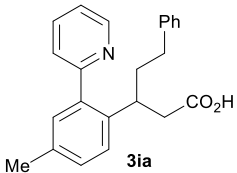
201249_RT13107_b_pn #26-30 RT: 0.36-0.39 AV: 2 NL: 5.94E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



31a

201253



Sample No. : C:\Xcalibur\...1119\201253_RT13137_pn

Instrument : Exacte

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

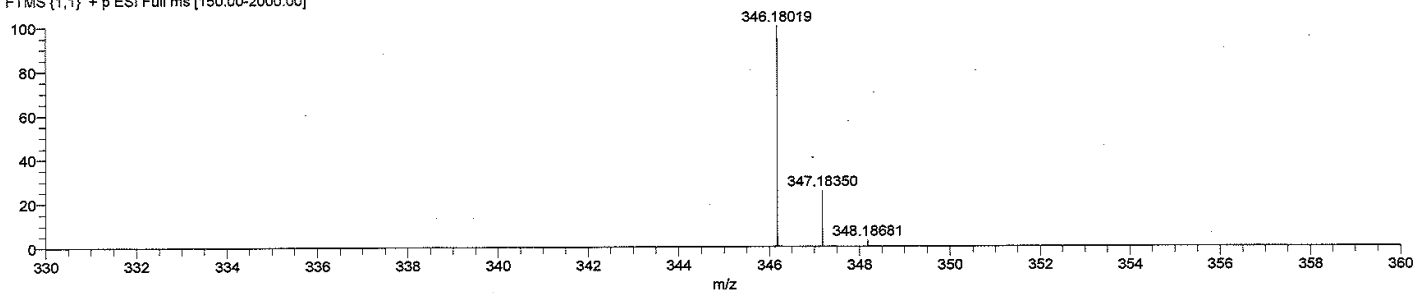
Date : 11/19/2020 2:45:05 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

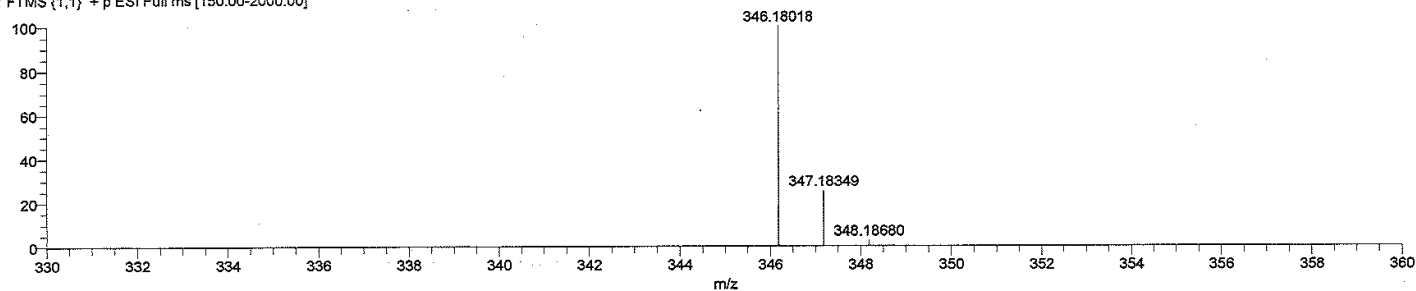
201253_RT13137_pn#17-21 RT: 0.25-0.29 AV: 3 NL: 2.54E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



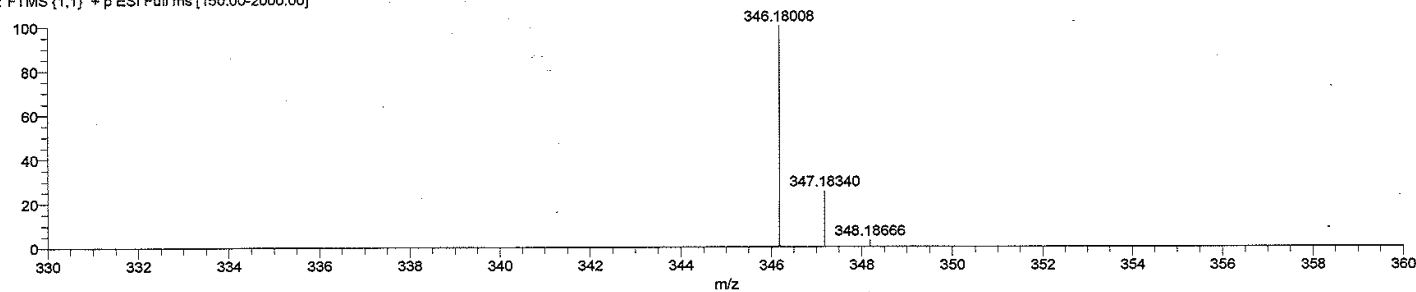
201253_RT13137_pn#21-26 RT: 0.29-0.34 AV: 3 NL: 1.94E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]

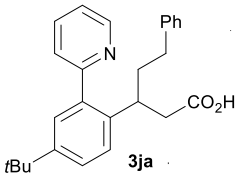


201253_RT13137_pn#26-30 RT: 0.36-0.39 AV: 2 NL: 1.50E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201256



Sample No. : C:\Xcalibur\11119\201256_RT13130_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

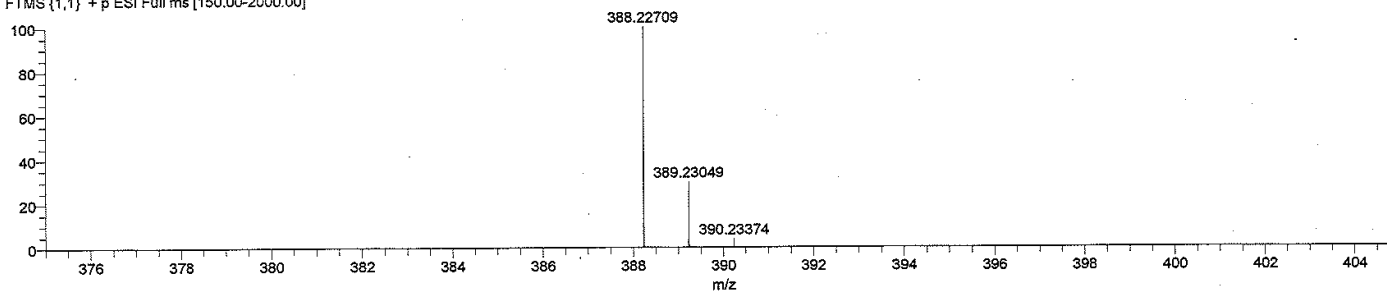
Date : 11/19/2020 3:00:12 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

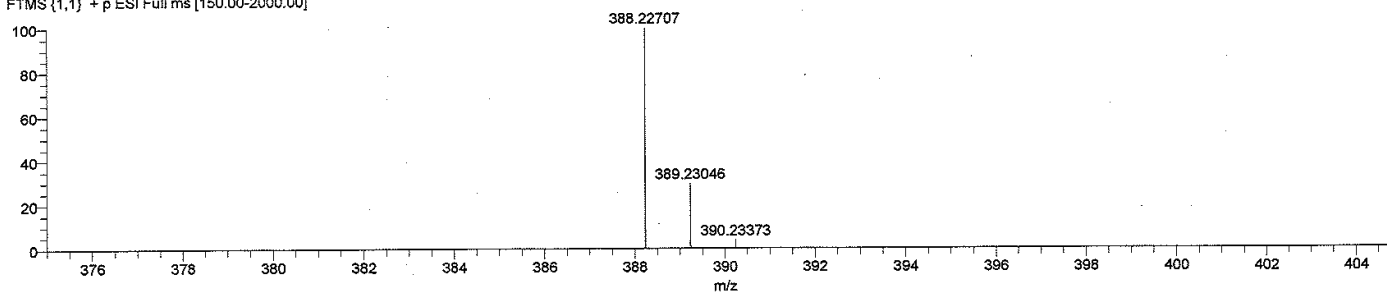
201256_RT13130_pn#17-21 RT: 0.25-0.30 AV: 3 NL: 2.46E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



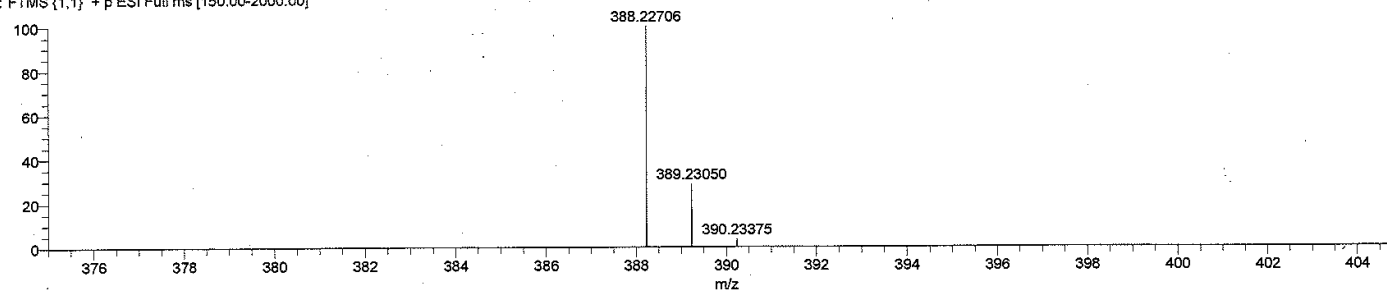
201256_RT13130_pn#21-26 RT: 0.30-0.34 AV: 3 NL: 1.88E7

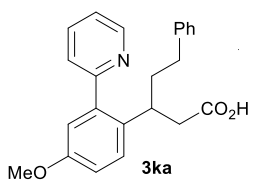
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201256_RT13130_pn#26-30 RT: 0.36-0.39 AV: 2 NL: 1.45E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201255

3ka

Sample No. : C:\Xcalibur\...1119\201255_RT13114_b_pn

Instrument : Exacte

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

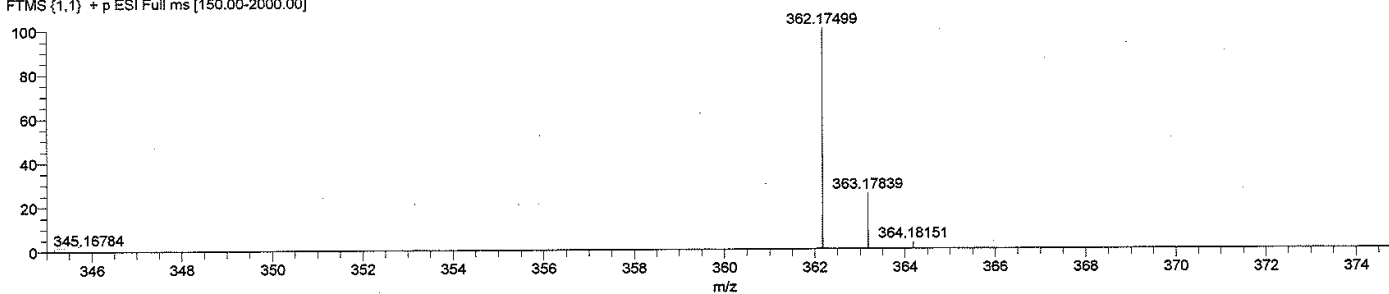
Date : 11/19/2020 2:55:10 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

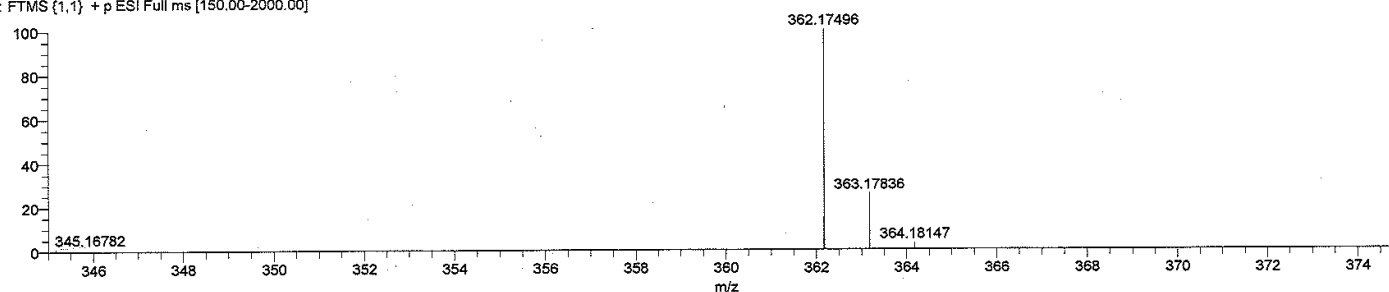
201255_RT13114_b_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 2.25E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



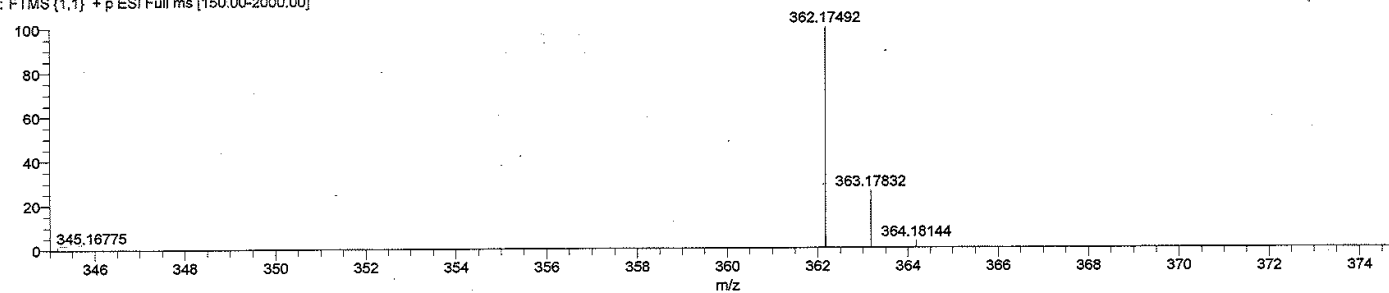
201255_RT13114_b_pn #21-26 RT: 0.30-0.34 AV: 3 NL: 1.77E7

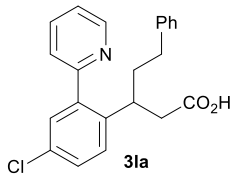
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201255_RT13114_b_pn #26-30 RT: 0.37-0.39 AV: 2 NL: 1.31E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201254

Sample No. : C:\Xcalibur\...1119\201204_RT13114_a_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

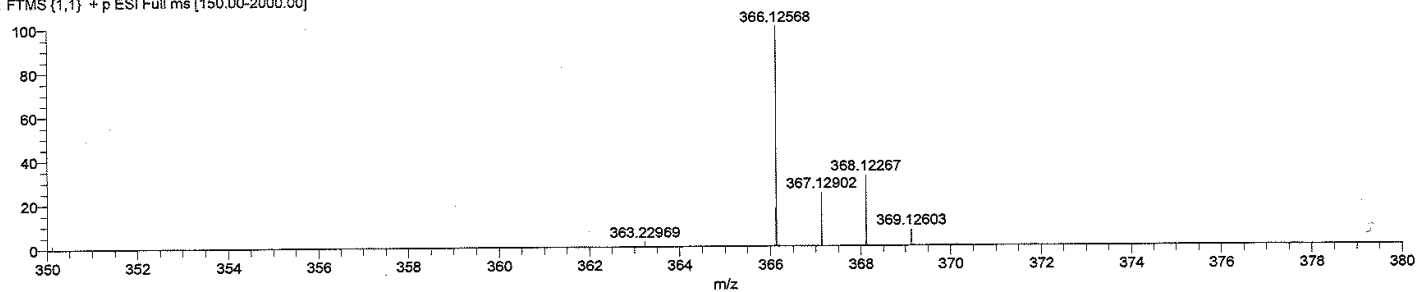
Date : 11/19/2020 2:50:07 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

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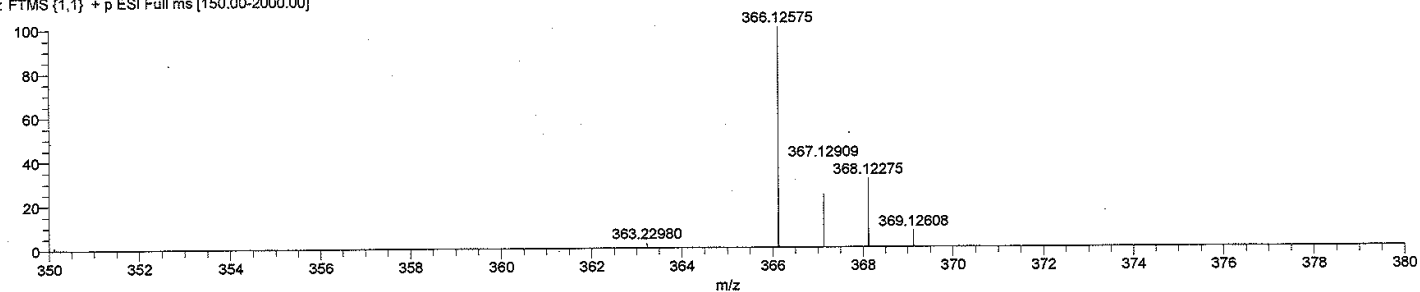
201254_RT13114_a_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 6.06E6

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



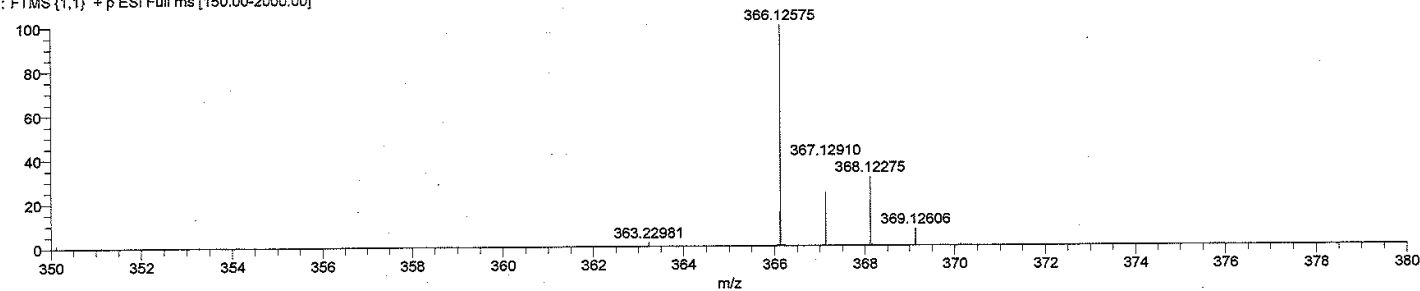
201254_RT13114_a_pn #21-25 RT: 0.30-0.34 AV: 3 NL: 5.22E6

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]

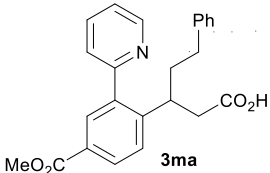


201254_RT13114_a_pn #25-30 RT: 0.34-0.39 AV: 3 NL: 3.83E6

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



5872



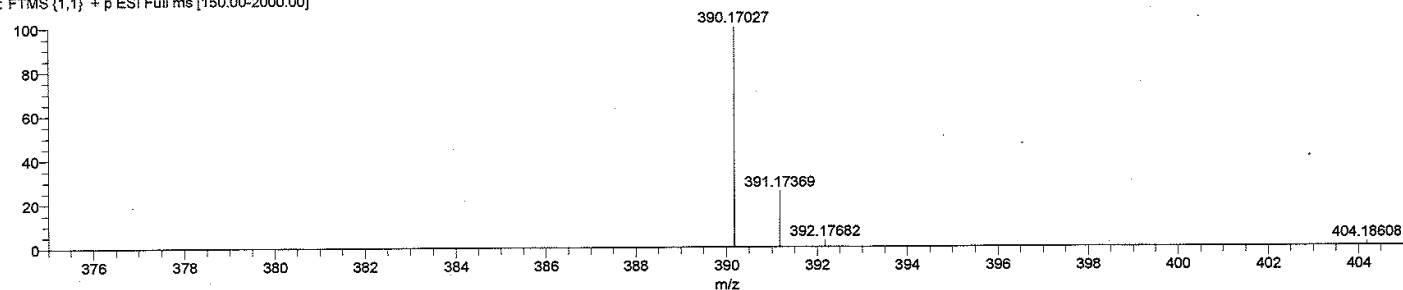
201257

Sample No. : C:\Xcalibur\...1119\201207_RT13131_a_pn
 Operator name : hayashi harumi
 Date : 11/19/2020 3:05:14 PM
 Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth
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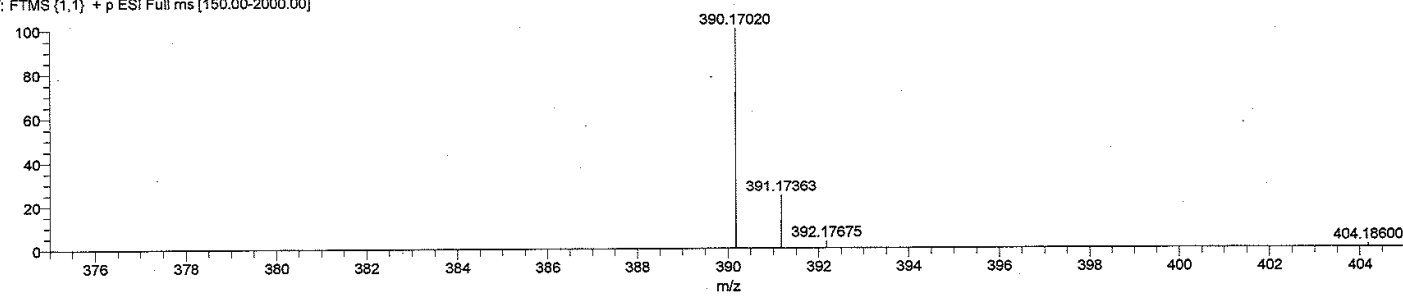
Instrument : Exactive

Mobile phase solvent : MeOH
 Sample solvent : submitting solution

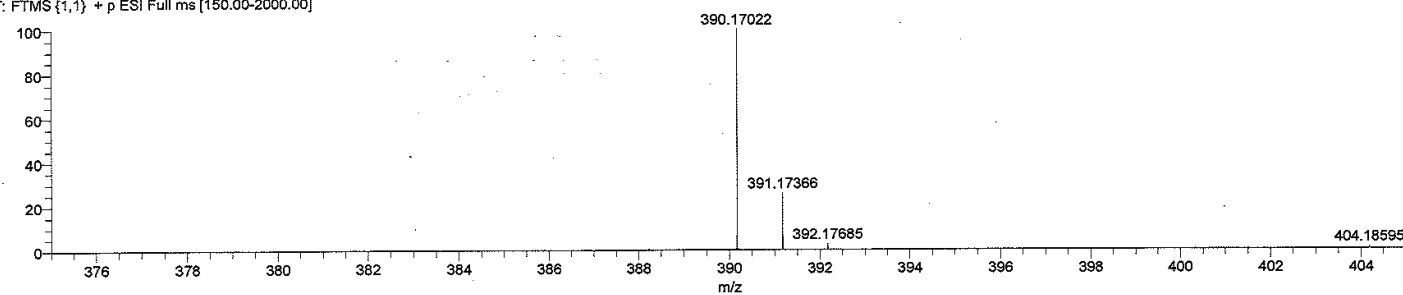
201257_RT13131_a_pn #17-21 RT: 0.25-0.30 AV: 3 NL: 5.90E6
 T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]

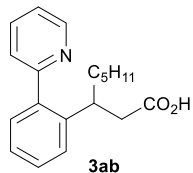


201257_RT13131_a_pn #21-26 RT: 0.30-0.34 AV: 3 NL: 4.96E6
 T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201257_RT13131_a_pn #26-30 RT: 0.37-0.39 AV: 2 NL: 4.23E6
 T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201526

3ab

Sample No.: C:\Xcalibur\1214\201526_RT13080_pn

Instrument: Exactive

Mobile phase solvent: MeOH

Operator name: hayashi harumi

Sample solvent: submitting solution

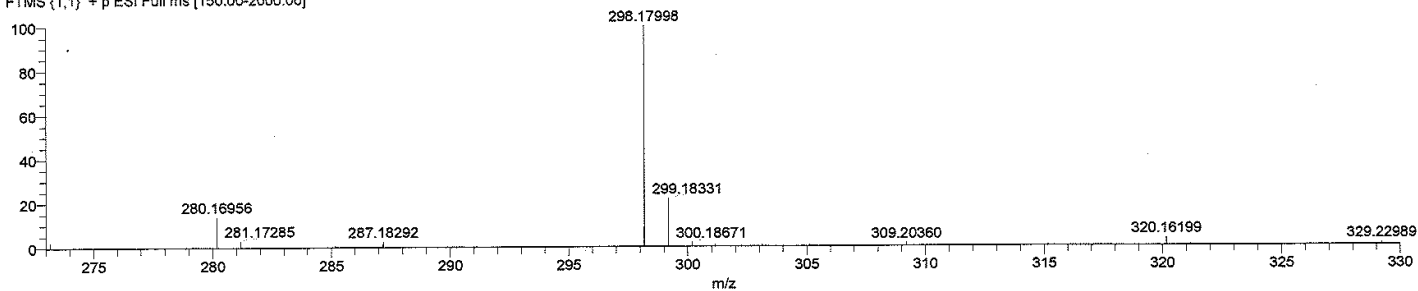
Date: 12/14/2020 11:35:22 AM

Instrumental method: C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

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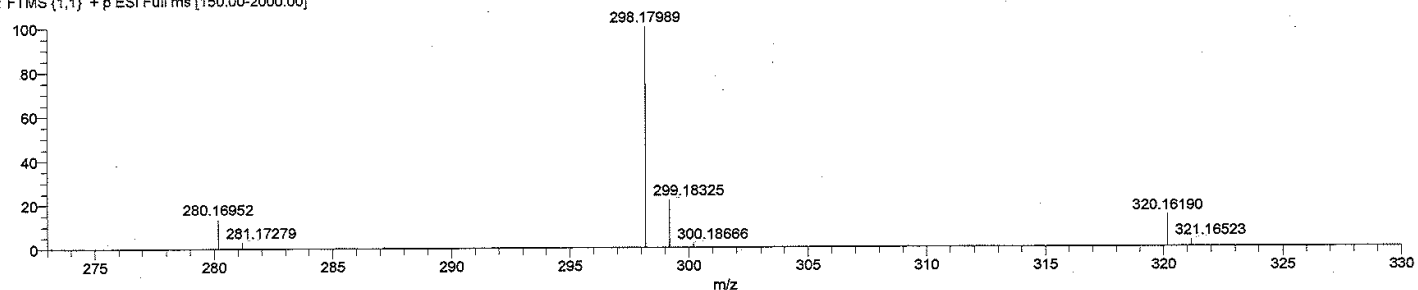
201526_RT13080_pn#16-19 RT: 0.27-0.31 AV: 2 NL: 1.84E6

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



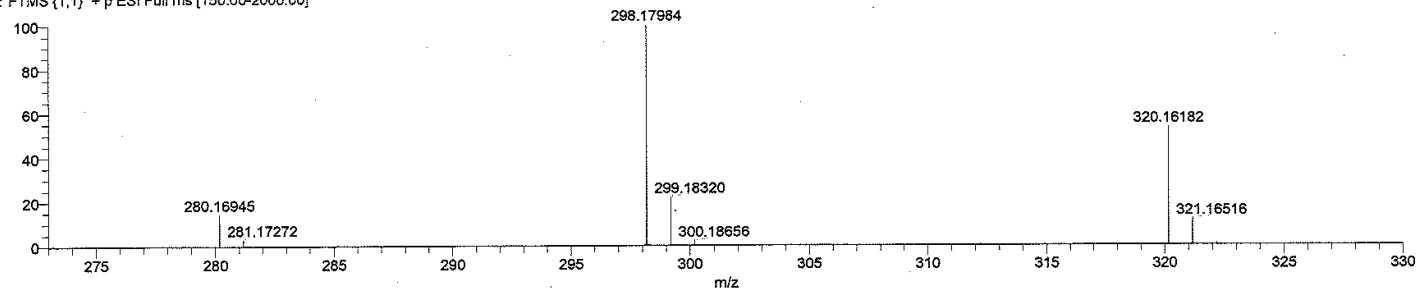
201526_RT13080_pn#19-22 RT: 0.31-0.33 AV: 2 NL: 5.76E6

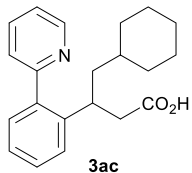
T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]



201526_RT13080_pn#22-26 RT: 0.36-0.38 AV: 2 NL: 1.43E7

T: FTMS {1,1} + p ESI Full ms [150.00-2000.00]





201527

Sample No. : C:\Xcalibur\...1214\201527_RT13142_b_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

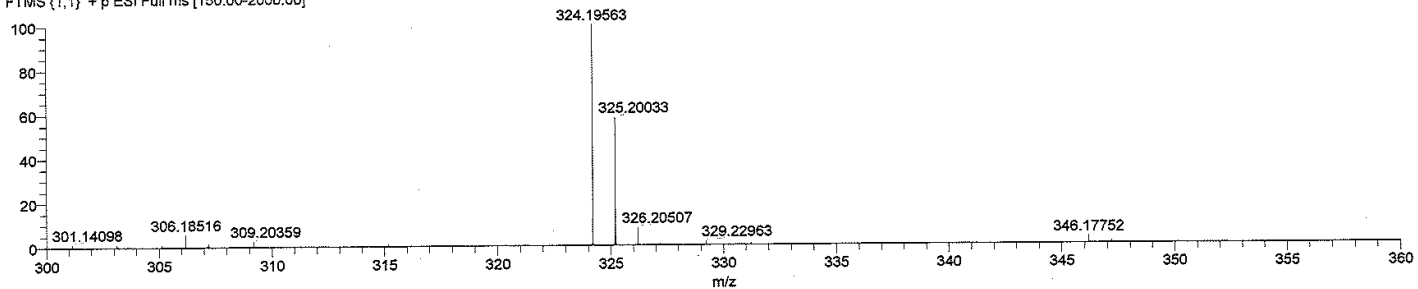
Date : 12/14/2020 11:40:24 AM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

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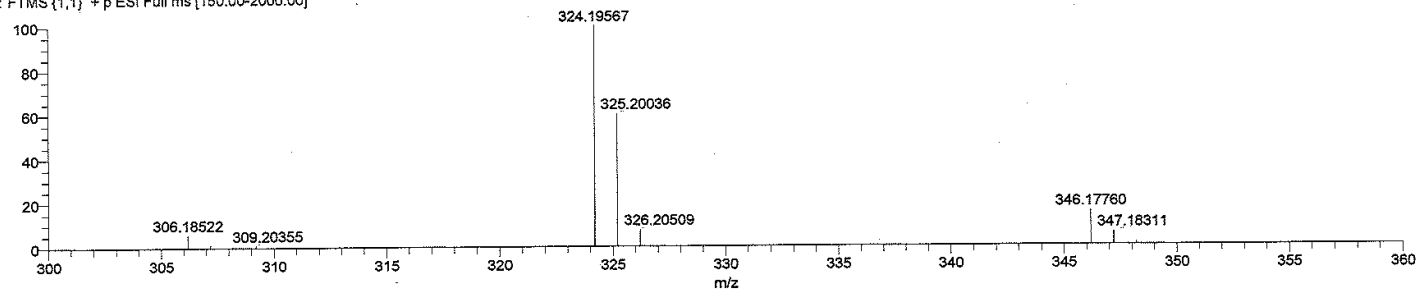
201527_RT13142_b_pn#16-19 RT: 0.28-0.31 AV: 2 NL: 1.09E6

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



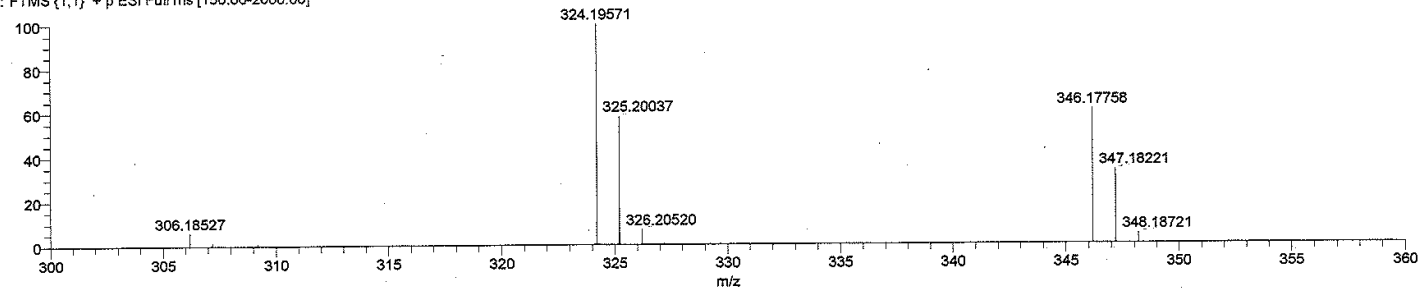
201527_RT13142_b_pn#19-22 RT: 0.31-0.34 AV: 2 NL: 3.91E6

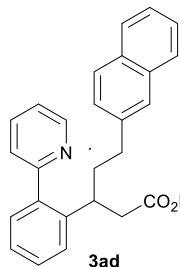
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201527_RT13142_b_pn#22-26 RT: 0.36-0.38 AV: 2 NL: 1.26E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





201528

3ad

Sample No. : C:\Xcalibur\...1214\201528_RT13145_a_pn

Instrument : Exacte

Mobile phase solvent : MeOH

Operator name : hayashi harumi

Sample solvent : submitting solution

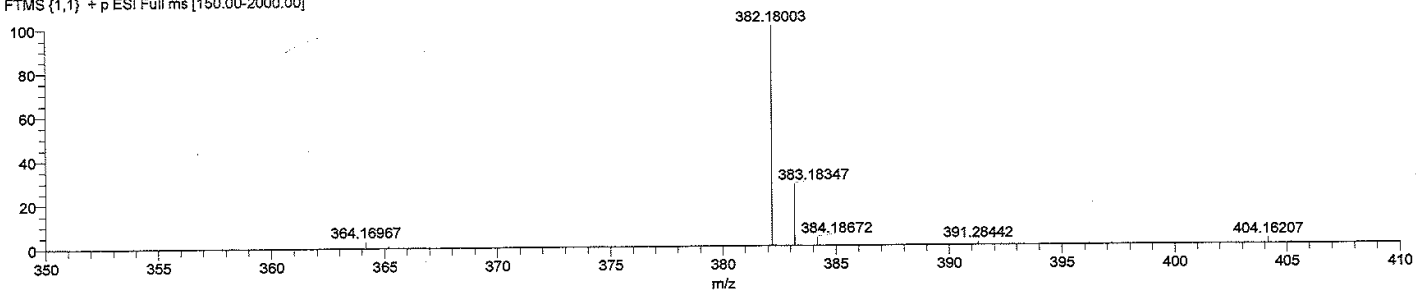
Date : 12/14/2020 11:45:26 AM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

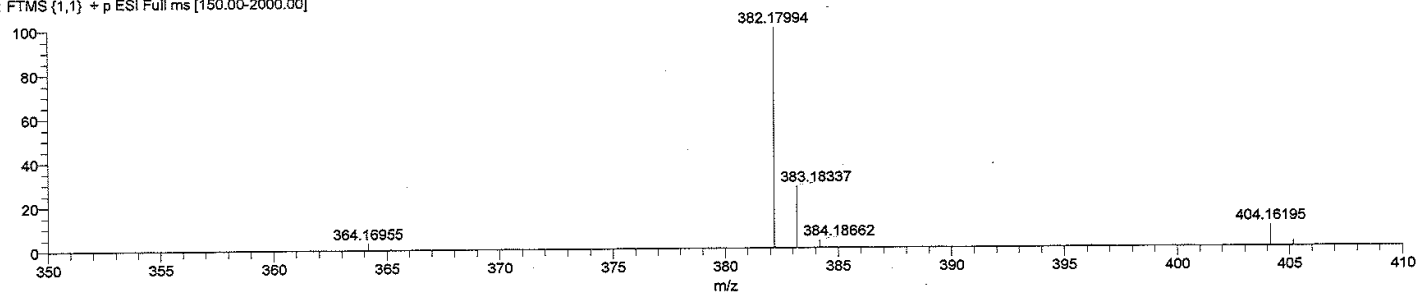
201528_RT13145_a_pn #16-19 RT: 0.27-0.30 AV: 2 NL: 8.85E5

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



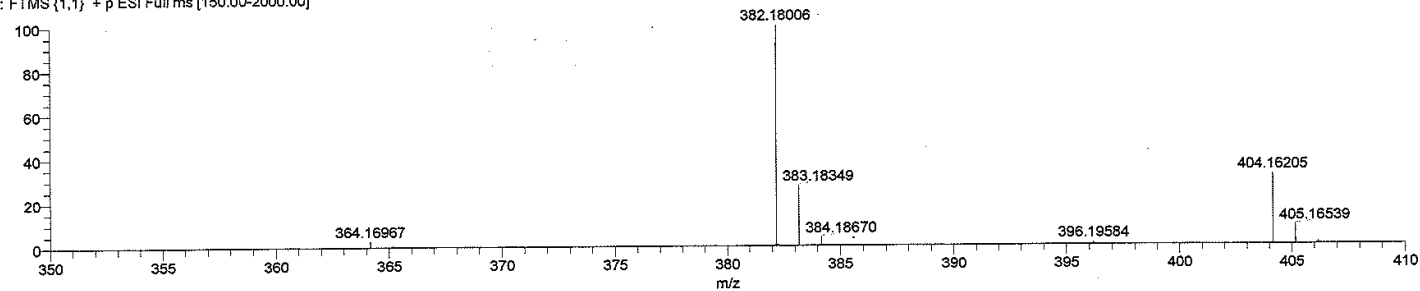
201528_RT13145_a_pn #19-22 RT: 0.30-0.33 AV: 2 NL: 3.43E6

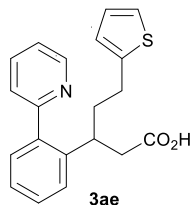
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201528_RT13145_a_pn #22-26 RT: 0.36-0.39 AV: 2 NL: 1.21E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





201813

Sample No. : C:\Xcalibur\...0121\201813_RT13146_a_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hirose tomohiro

Sample solvent : submitting solution

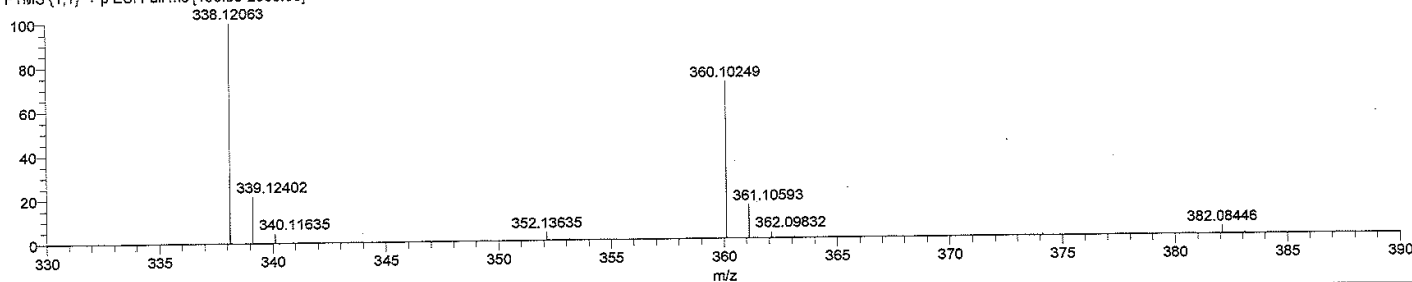
Date : 1/21/2021 1:39:00 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

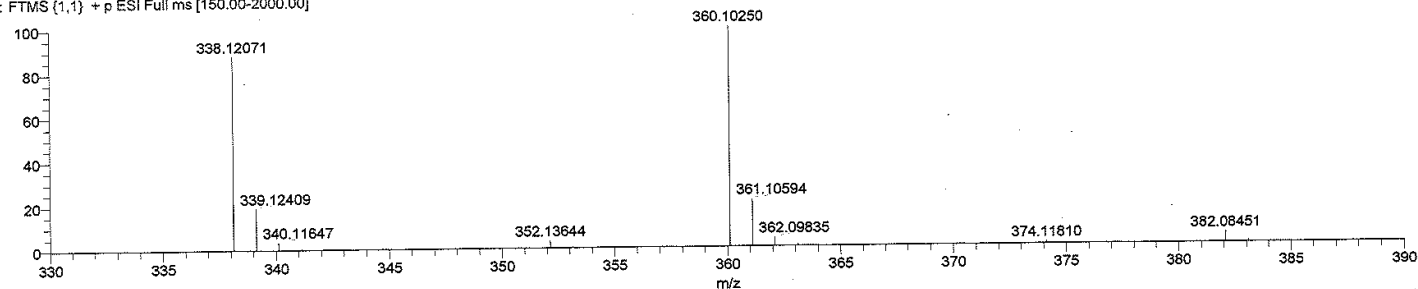
201813_RT13146_a_pn #22-25 RT: 0.34-0.37 AV: 2 NL: 1.97E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



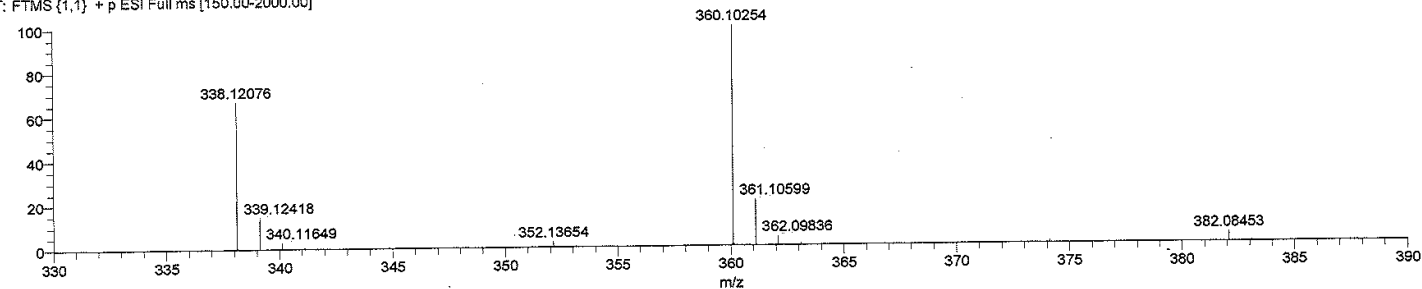
201813_RT13146_a_pn #25-30 RT: 0.37-0.41 AV: 3 NL: 2.43E7

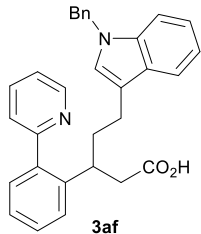
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201813_RT13146_a_pn #30 RT: 0.44 AV: 1 NL: 2.90E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





201815

Sample No. : C:\Xcalibur\...0121\201815_RT13151_b_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hirose tomohiro

Sample solvent : submitting solution

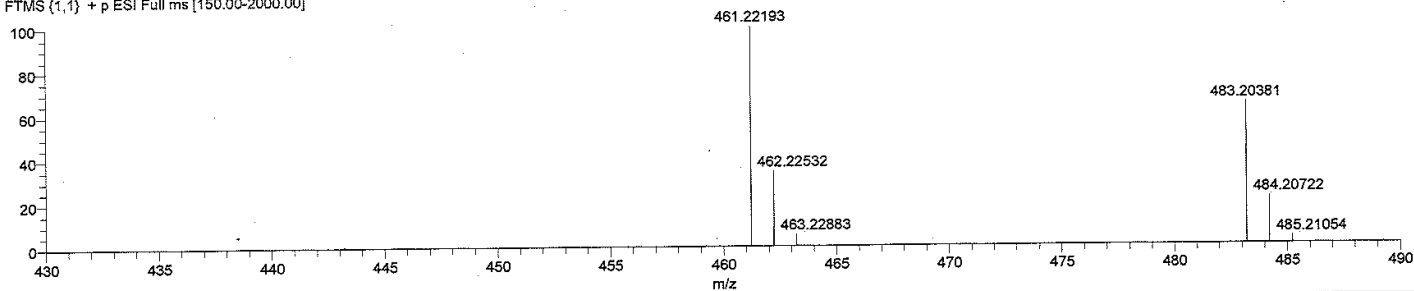
Date : 1/21/2021 1:49:07 PM

Instrumental method : C:\Xcalibur\methods\HESI_100ul\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

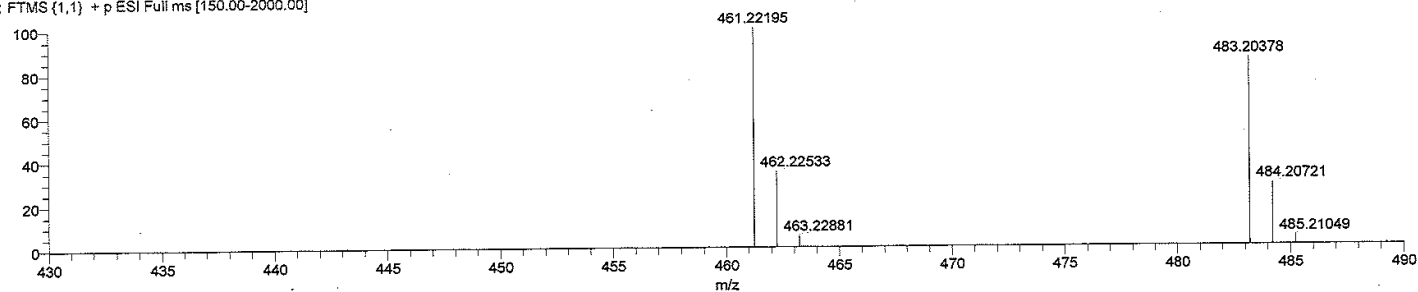
201815_RT13151_b_pn #22-25 RT: 0.35-0.37 AV: 2 NL: 2.27E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



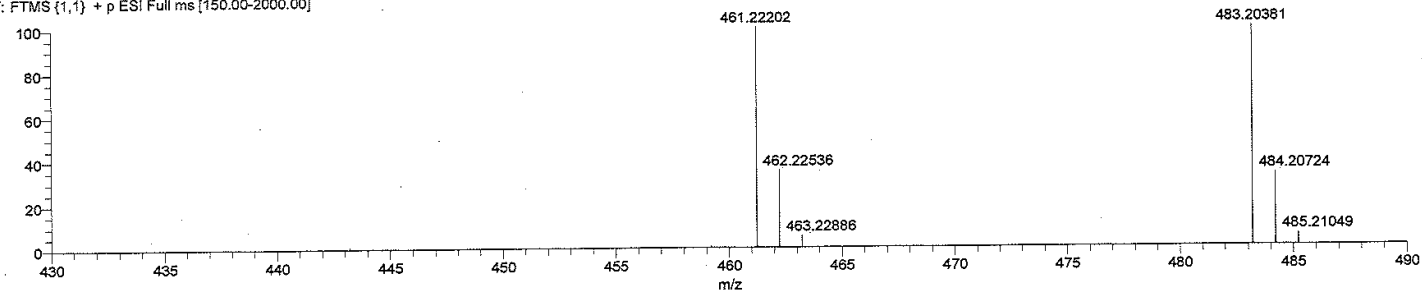
201815_RT13151_b_pn #25-29 RT: 0.37-0.42 AV: 3 NL: 2.42E7

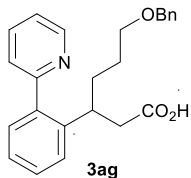
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201815_RT13151_b_pn #29-32 RT: 0.42-0.44 AV: 2 NL: 2.05E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





201814

Sample No. : C:\Xcalibur\...0121\201814_RT13151_a_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name : hirose tomohiro

Sample solvent : submitting solution

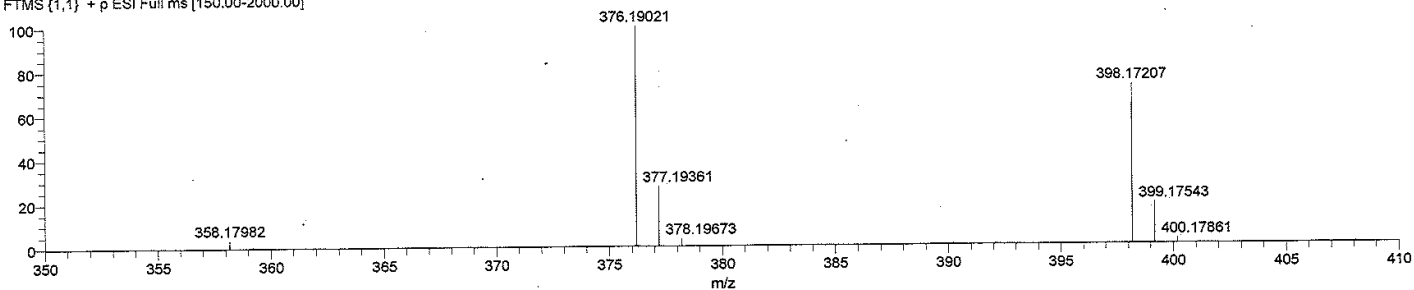
Date : 1/21/2021 1:44:03 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

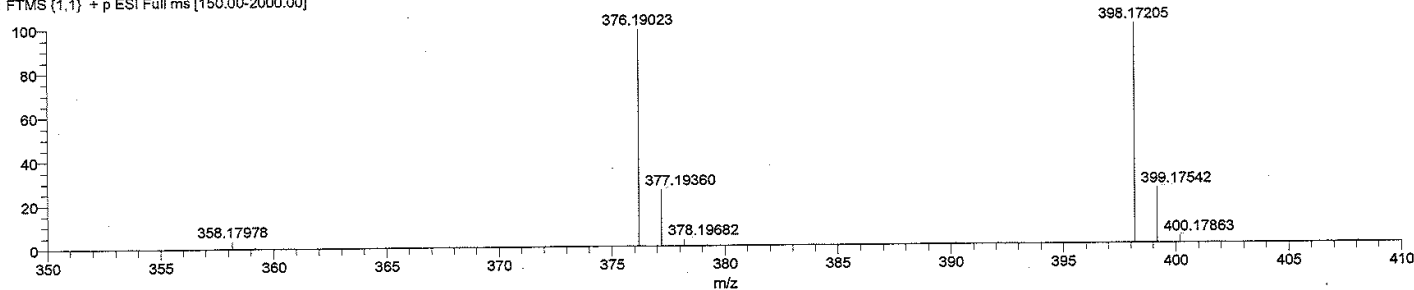
201814_RT13151_a_pn #22-26 RT: 0.34-0.36 AV: 2 NL: 4.06E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



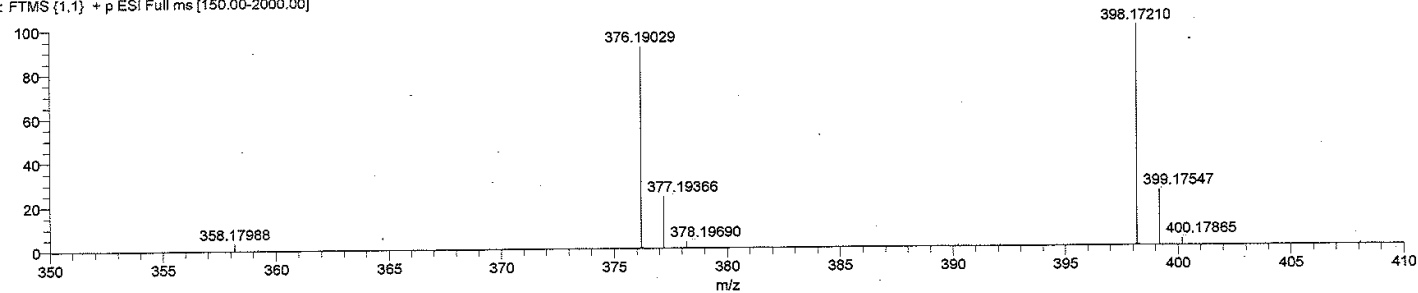
201814_RT13151_a_pn #26-30 RT: 0.38-0.41 AV: 2 NL: 4.69E7

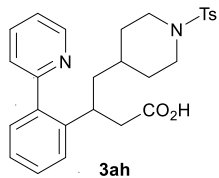
T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



201814_RT13151_a_pn #30-33 RT: 0.43-0.45 AV: 2 NL: 4.60E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]





202062

Sample No. : C:\Xcalibur\...10222\202006_13176_b_pn

Instrument : Exactive

Mobile phase solvent : MeOH

Operator name :

Sample solvent : submitting solution

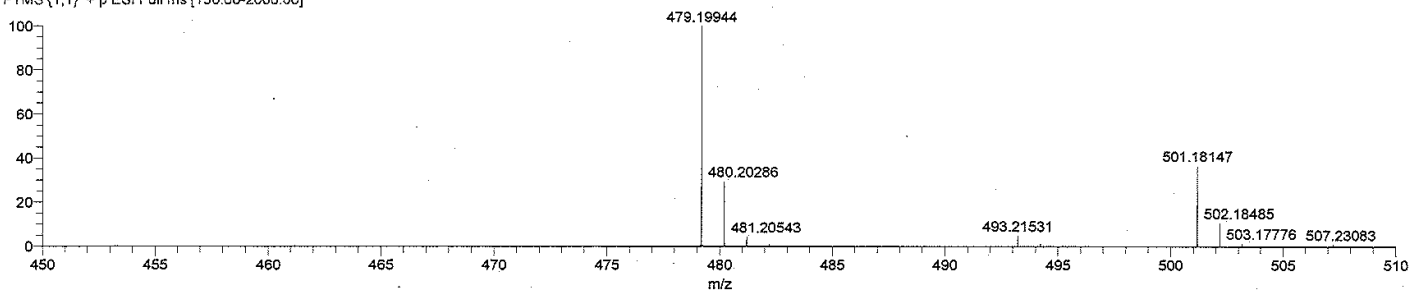
Date : 2/22/2021 6:06:10 PM

Instrumental method : C:\Xcalibur\methods\HESI_100u\pn_H80_S30.meth

Instrumental Analysis Division, Global Facility Center, Creative Research Institution, Hokkaido University

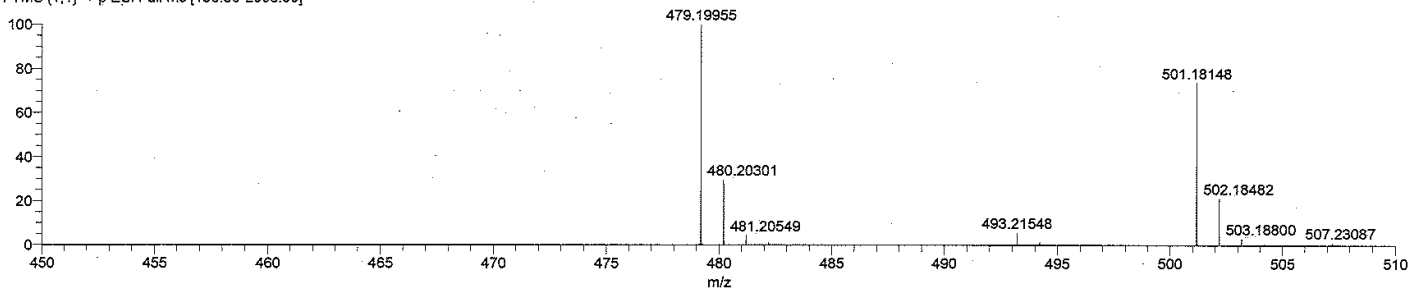
202062_13176_b_pn #22-25 RT: 0.34-0.37 AV: 2 NL: 3.72E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



202062_13176_b_pn #25-30 RT: 0.37-0.41 AV: 3 NL: 3.56E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]



202062_13176_b_pn #30 RT: 0.43 AV: 1 NL: 3.12E7

T: FTMS (1,1) + p ESI Full ms [150.00-2000.00]

