Electronic Supplementary Information

Robust and color-tunable afterglows from guanidine derivatives

Zihao Zhao,^a Yuxuan Li,^a Xiaohong Chen,^{*ab} Yongming Zhang^{ab} and Wang Zhang Yuan^{*a}

^{*a*}School of Chemistry and Chemical Engineering, Frontiers Science Center for Transformative Molecules, Shanghai Key Lab of Electrical Insulation and Thermal Aging, Shanghai Jiao Tong University, No. 800 Dongchuan Rd., Minhang District, Shanghai 200240, China. E-mail: wzhyuan@sjtu.edu.cn ^{*b*}Institute of Advanced Materials, North China Electric Power University, Beijing 102206, China. E-mail: xhchen200905@ncepu.edu.cn

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Table S4 Transition configurations of DCDA. The ones labelled in red are allowed transitions.

Fig. S11 Electron densities of the HOMO and LUMO levels for monomer, dimer, trimer and tetramer of GCA.

Fig. S12 The ISC channels of (a) monomer, (b) dimer, (c) trimer and (d) tetramer of GCA. (e) The excitation energy diagram of monomer, dimer, trimer and tetramer of GCA.

Table S5 Transition configurations of GCA. The ones labelled in red are allowed transitions.

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Fig. S14 NTO (isovalue = 0.04) analyses and spin-orbit coupling coefficients for corresponding singlet and triplet states of monomer, dimer, trimer and tetramer of GCA.

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Fig. S16 (a) Photographs of GCA ground powders taken under or after ceasing 254, 312 and 365 nm UV lights. (b) Prompt and (c) delayed ($t_d = 1 \text{ ms}$) emission spectra of the ground powders for GCA with different $\lambda_{ex}s$. (d) Normalized prompt and (e) delayed ($t_d = 1 \text{ ms}$) emission spectra of the ground powders for GCA with different $\lambda_{ex}s$.

Fig. S17 (a) p-RTP lifetimes at varying λ_{em} s for GCA ground powders. (b) Fluorescence lifetimes at varying λ_{em} s for GCA ground powders ($\lambda_{ex} = 363$ nm).

Reference

Experimental Section

Materials. Dicyandiamide (DCDA, > 98.0%) and glycocyamine (GCA, > 97.0%) were purchased from TCI Development Co., Ltd. Methanol (MeOH), tetrahydrofuran (THF) and *N*,*N*dimethylformamide (DMF) were obtained from Sinopharm Chemical Reagent Co., Ltd. (China). Pure water (H₂O) was bought from Hangzhou Wahaha Group Co., Ltd (Zhejiang, China).

Instruments. Prompt and delayed emission spectra, time-resolved measurements in nanosecond (ns) and millisecond (ms) scales as well as the quantum efficiencies ($\lambda_{ex} = 312$ nm) for single crystals were conducted on an Edinburgh FLS1000 photoluminescence spectrometer. Emission spectra for solutions were collected on a PerkinElmer LS 55 fluorescence spectrometer. Absorption of single crystals were measured on a PerkinElmer Lambda 750s UV/Vis spectrometer. Single crystal diffractogram data were collected by a Bruker D8 VENTURE X-ray Diffractometer. All photographs and videos were taken by a digital camera (Sony α 7sII, Japan).

Purification of DCDA and GCA. DCDA was firstly dissolved into DMF to form nearly saturated solution and filtered. The preliminarily purified DCDA was precipitated by adding the filtered solution into pure water dropwise under stirring vigorously. The precipitates were collected through filtration and then dissolved in methanol for further single crystal cultivation. Single crystals of DCDA were obtained by evaporating its dilute solution slowly at room temperature and separated from the solvent prior to total evaporation to guarantee purity.

The purification process for GCA is as same as that for DCDA, except that methanol was used in the first step of dissolution rather than DMF.

Anti-counterfeiting pattern preparation. The pattern with corals and clownfishes was carved on a commercially available nonluminescent rubber stamper. Then, the carved pattern on the stamper was filled with ground powders of DCDA and directly pictured under 312 and 365 nm UV irradiation.

Computational method. The single molecule (monomer), dimer and trimer of DCDA were extracted from their single crystal data. Time dependent density functional theory (TD-DFT) was

used to calculate the HOMO and LUMO electron densities, energy levels and molecular electrostatic potential using B3LYP hybrid functional and 6-31G (d,p) basis set. All TD-DFT calculations were performed within Gaussian 09 (version 9.5) program. Nature transition orbital (NTO) analysis was carried out with Multiwfn at 6-31G (d,p) level of theory with B3LYP hybrid functional.¹ Spin-orbit coupling calculations were performed with ORCA 4.2.1 using B3LYP hybrid functional and 6-31G (d,p) basis set.



Fig. S1 Absorption spectra of different (a) DCDA/MeOH and (b) GCA/H₂O solutions.



Fig. S2 Emission spectra of 1 M DCDA/DMF solution with different λ_{exs} .



Fig. S3 (a, b) Photographs and (c, d) emission spectra under 312 nm UV irradiation of (a, c) DCDA in MeOH and MeOH/THF mixtures and (b, d) GCA in H₂O and H₂O/THF mixtures with different THF volume fractions. Concentration = 1×10^{-3} M.



Fig. S4 (a) Excitation spectra of DCDA single crystals. (b) Prompt and (c) delayed ($t_d = 1 \text{ ms}$) emission spectra of DCDA single crystals with different $\lambda_{ex}s$. (d) Normalized prompt and (e) delayed ($t_d = 1 \text{ ms}$) emission spectra of DCDA single crystals with different $\lambda_{ex}s$. (f) CIE coordinate diagram of the delayed emission for DCDA single crystals with different $\lambda_{ex}s$.



Fig. S5 Fluorescence lifetimes at varying emission wavelengths (λ_{ems}) for DCDA single crystals

 $(\lambda_{\rm ex} = 363 \text{ nm}).$

Table S1 Fluorescence lifetimes for single crystals of DCDA and GCA and ground powders of DCDA.^{a)}

Compound	λex [nm]	λem [nm]	<\tau>1 [ns]	< <i>t</i> >2 [ns]	<t>3 [ns]</t>	<\tau>4 [ns]	A ₁ [%]	A ₂ [%]	A3 [%]	A4 [%]	χ^2	<τ≻ _{f-ave} [ns]
DCDA		412	0.8	2.8	10.1	63.7	13.1	67.3	16.6	3.0	1.08	25.8
single	363	435	0.6	2.6	7.9	37.4	13.4	61.9	21.7	3.0	1.05	13.2
crystals		495	0.6	3.5	10.0	49.3	28.2	48.9	20.9	2.0	1.13	15.3
GCA	272	440	3.3				100				1.03	3.3
single crystals	363	590	1.4				100				1.14	1.4
DCDA		412	0.6	3.1	9.9	50.5	4.3	34.8	27.8	33.1	1.18	42.5
ground	363	435	0.6	2.8	8.9	48.4	12.3	62.7	16.3	8.7	1.22	29.6
powders		495	1.6	5.6	24.0	155.5	8.6	51.2	22.5	17.7	1.18	123.2
GCA	262	440	0.7	2.9			5.1	94.9			1.01	2.9
single crystals	363	590	0.5				100				1.12	0.5

^{a)} $<\tau >_{f-ave}$ is the average fluorescence lifetime of every components adopted from the lifetime measurements. $<\tau >_{f-ave} = (A_1\tau_1^2 + A_2\tau_2^2 + A_3\tau_3^2 + A_4\tau_4^2) / (A_1\tau_1 + A_2\tau_2 + A_3\tau_3 + A_4\tau_4)$

Table S2 p-RTP lifetimes for single crystals of DCDA and GCA and ground powders of DCDA.^{a)}

Compound	λex [nm]	λem [nm]	< <i>t</i> >1 [ms]	<τ>2 [ms]	<\tau>3 [ms]	< <i>t</i> >4 [ms]	A ₁ [%]	A ₂ [%]	A3 [%]	A4 [%]	χ^2	<τ>p-ave [ms]
DCDA	280	485	8.3	40.8	186.9	598.6	2.8	15.4	36.3	45.5	1.09	507.5
single	312	495	16.2	99.4	485.4		17.9	31.5	50.6		1.05	437.3
crystals	365	525	14.9	42.1	134.5	426.6	8.3	30.2	20.2	41.3	1.19	365.2
	254	520	1.9	13.1	70.6	260.3	11.7	35.1	32.0	21.2	1.11	194.0
GCA	285	510	13.5	54.9	274.2		20.8	33.1	46.1		1.08	242.2
single	285	690	0.5	6.1	41.0		9.9	37.9	52.2		1.10	37.5
crystals	312	475	19.2	74.7	383.5		18.4	22.7	58.9		1.21	357.0
	365	525	1.6	13.2	56.0	283.9	7.7	23.4	30.2	38.7	1.12	247.6
DCDA	280	485	3.4	36.3	364.6		17.0	31.8	51.2		1.21	344.5
ground	312	495	14.6	60.9	416.5		14.4	30.9	54.7		1.13	386.3
powders	365	525	20.1	60.2	331.0		24.3	34.7	41.0		1.21	286.6
	254	520	1.7	9.1	27.5	136.6	4.3	18.8	32.6	44.3	1.18	119.7
GCA	285	510	2.8	9.8	29.0	163.5	4.2	21.9	30.8	43.1	1.11	144.5
ground	285	690	0.2	6.5	7.7		4.6	37.5	57.6		1.15	7.3
powders	312	475	1.2	6.3	21.0	133.9	2.5	25.8	47.6	24.1	1.13	103.3
	365	525	1.4	6.7	26.7	134.2	3.3	12.8	31.4	52.5	1.17	121.5

^{a)} $<\tau >_{p-ave}$ is the average p-RTP lifetime of every components adopted from the lifetime measurements. $<\tau >_{p-ave} = (A_1\tau_1^2 + A_2\tau_2^2 + A_3\tau_3^2 + A_4\tau_4^2) / (A_1\tau_1 + A_2\tau_2 + A_3\tau_3 + A_4\tau_4)$



Fig. S6 (a) Excitation spectra of GCA single crystals. (b) Prompt and (c) delayed ($t_d = 1 \text{ ms}$) emission spectra of GCA single crystals with different λ_{exs} . (d) Normalized prompt and (e) delayed ($t_d = 1 \text{ ms}$) emission spectra of GCA single crystals with different λ_{exs} . (f) CIE coordinate diagram of the delayed emission for GCA single crystals with different λ_{exs} .



Fig. S7 (a) p-RTP lifetimes at varying λ_{ems} for GCA single crystals. (b) Fluorescence lifetimes at varying λ_{ems} for GCA single crystals ($\lambda_{ex} = 363$ nm).

	DCDA (296 K)	GCA (295 K)
Formula	$C_2H_4N_4$	$C_3H_7N_3O_2$
Formula Weight	84.09	117.12
Wavelength (Å)	1.54178	1.54178
Space Group	C2/c	P21/n
Cell Length (Å)	a=14.9827 (5) b=4.4969 (1) c=13.1144 (4)	a=4.9399 (2) b=6.0035 (3) c=17.1666 (8)
Cell Angle (°)	$\alpha = 90$ $\beta = 115.361(1)$ $\gamma = 90$	
Cell Volume (Å ³)	798.44 (4)	507.69 (4)
Z	8	4
Density (g/cm ³)	1.399	1.532
F (000)	352.0	248.0
$h_{\max}, k_{\max}, l_{\max}$	18, 5, 15	5, 7, 20

Table S3 Single crystal data of DCDA and GCA.



Fig. S8 Crystal structure with (a) hydrogen bonds and (b) N···N short contacts around one DCDA molecule. (c) Fragmental molecular packing with through-space conjugation (TSC) for DCDA.



Fig. S9 Crystal structure with (a) hydrogen bonds and (b) O…N short contacts around one GCA molecule. (c) Fragmental molecular packing with TSC for GCA.



Fig. S10 The ISC channels of (a) monomer, (b) dimer, (c) trimer and (d) tetramer of DCDA. (e) The excitation energy diagram of monomer, dimer, trimer and tetramer of DCDA.

Aggregation	Excited	Excitation	Transition
state	state	energy [eV]	configuration
	S_1	5.7706	H-1→L
monomer	T_1	4.3198	$H \rightarrow L$
	T_2	5.4077	H-1→L
	S_1	5.8664	$H \rightarrow L+1$
	T_1	4.4797	$H-1 \rightarrow L+1, H \rightarrow L$
dimer	T_2	4.4981	$H-1 \rightarrow L, H \rightarrow L+1$
	T ₃	5.5926	H-2→L
	T_4	5.5930	H-3→L
	S_1	5.6678	$H-2\rightarrow L, H\rightarrow L$
	T_1	4.4201	$\text{H-1} \rightarrow \text{L+1}, \text{H-1} \rightarrow \text{L+2}, \text{H} \rightarrow \text{L+1}, \text{H} \rightarrow \text{L+2}$
	T_2	4.4725	H-2→L, H-1→L
trimer	T ₃	4.5988	$H-1 \rightarrow L+2, H \rightarrow L+1$
	T_4	5.5297	$H-3 \rightarrow L+1, H-3 \rightarrow L+2$
	T 5	5.5770	H-5→L
	T_6	5.6594	$H-2\rightarrow L, H\rightarrow L$
	S_1	5.8101	$H-3\rightarrow L, H-2\rightarrow L+1, H\rightarrow L+1$
	T_1	4.4219	$\text{H-3}{\rightarrow}\text{L+1}, \text{H-2}{\rightarrow}\text{L}, \text{H-1}{\rightarrow}\text{L+1}, \text{H}{\rightarrow}\text{L+1}$
	T_2	4.4219	H-3→ L , H-3→ L +1, H-2→ L , H -2→ L +1, H-1→ L , H-1→ L +1
	T ₃	4.5747	H-3 \rightarrow L+3, H-2 \rightarrow L+2, H \rightarrow L+2
tetramer	T_4	4.5937	H-3 \rightarrow L+2, H-2 \rightarrow L+3, H \rightarrow L+3
	T 5	5.5328	$H-6\rightarrow L+1, H-5\rightarrow L+1, H-4\rightarrow L$
	T_6	5.5328	$\text{H-6}{\rightarrow}\text{L}, \text{H-5}{\rightarrow}\text{L}, \text{H-5}{\rightarrow}\text{L+1}, \text{H-4}{\rightarrow}\text{L}, \text{H-4}{\rightarrow}\text{L+1}$
	T 7	5.7074	H-7 \rightarrow L+1, H-6 \rightarrow L+2, H-6 \rightarrow L+3
	T ₈	5.7078	$\text{H-7} \rightarrow \text{L+2, H-6} \rightarrow \text{L+1, H-6} \rightarrow \text{L+2, H-5} \rightarrow \text{L+3, H-4} \rightarrow \text{L+2}$

Table S4 Transition configurations of DCDA. The ones labelled in red are allowed transitions.



Fig. S11 Electron densities of the HOMO and LUMO levels for monomer, dimer, trimer and tetramer of GCA.



Fig. S12 The ISC channels of (a) monomer, (b) dimer, (c) trimer and (d) tetramer of GCA. (e) The excitation energy diagram of monomer, dimer, trimer and tetramer of GCA.

For dimer and tetramer of GCA, since the S₁ and S₂ states both own the ¹(n, π^*) character and the energies of them are very close ($\Delta E_{S1S2, dimer} = 0.017 \text{ eV}$, $\Delta E_{S1S2, tetramer} = 0.002 \text{ eV}$), it is probable that the ISC process could happen from S₂ to corresponding triplet states.² Considering the allowed transitions between HOMOs and LUMOs (Fig. S12 and Table S5), only ISC from S₂ to T₂ can occur for dimer and tetramer of GCA, which is taken into account in the subsequent discussions.

Aggregation	Excited	Excitation	Transition
state	state	energy [eV]	configuration
monomor	S_1	4.2901	H→L
monomer	T_1	4.2383	H→L
	S_1	3.6427	H→L
dimor	\mathbf{S}_2	3.6603	H-1→L
unner	T_1	3.6149	H→L
	T_2	3.6332	H-1→L
trimor	S_1	3.5248	$H \rightarrow L, H \rightarrow L+1$
umer	T_1	3.5068	$H \rightarrow L, H \rightarrow L+1$
	S_1	3.3192	$H-1 \rightarrow L+3, H \rightarrow L+1, H \rightarrow L+2$
tatramar	S_2	3.3212	$H-1\rightarrow L+1, H-1\rightarrow L+2, H\rightarrow L+3$
tetramer	T_1	3.2979	$H-1 \rightarrow L+3, H \rightarrow L+1, H \rightarrow L+2$
	T_2	3.2983	$H-1\rightarrow L+1, H-1\rightarrow L+2, H\rightarrow L+3$

Table S5 Transition configurations of GCA. The ones labelled in red are allowed transitions.



Fig. S13 NTO (isovalue = 0.04) analyses and spin-orbit coupling coefficients for corresponding singlet and triplet states of monomer, dimer, trimer and tetramer of DCDA.



Fig. S14 NTO (isovalue = 0.04) analyses and spin-orbit coupling coefficients for corresponding singlet and triplet states of monomer, dimer, trimer and tetramer of GCA.



Fig. S15 (a) Normalized prompt and (b) delayed ($t_d = 1 \text{ ms}$) emission spectra of the ground powders for DCDA with different $\lambda_{ex}s$. (c) Fluorescence lifetimes at varying $\lambda_{em}s$ of the ground powders for DCDA ($\lambda_{ex} = 363 \text{ nm}$).



Fig. S16 (a) Photographs of GCA ground powders taken under or after ceasing 254, 312 and 365 nm UV lights. (b) Prompt and (c) delayed ($t_d = 1 \text{ ms}$) emission spectra of the ground powders for GCA with different $\lambda_{ex}s$. (d) Normalized prompt and (e) delayed ($t_d = 1 \text{ ms}$) emission spectra of the ground powders for GCA with different $\lambda_{ex}s$.



Fig. S17 (a) p-RTP lifetimes at varying λ_{em} s for GCA ground powders. (b) Fluorescence lifetimes at varying λ_{em} s for GCA ground powders ($\lambda_{ex} = 363$ nm).

Reference

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