

Electronic Supplementary Information (ESI) for *CrystEngComm*

Regulating the Distortion Degree of the Square Antiprism

Coordination Geometry in the Dy-Na Single Ion Magnets

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Table S1. Crystal data and structure refinement parameters for complexes **1-2**.

Compounds	1	2
Formula ^a	C ₄₆ H ₄₄ DyN ₄ NaO ₁₀	C ₄₆ H ₄₄ DyN ₅ Na ₂ O ₁₃
Formula weight	998.34	1083.34
Crystal color	yellow	yellow
Crystal size (mm)	0.51×0.32×0.06	0.18×0.17×0.14
Crystal system	monoclinic	monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> (Å)	22.3221(6)	10.9971(4)
<i>b</i> (Å)	11.7740(3)	29.7015(10)
<i>c</i> (Å)	16.2459(4)	14.6760(4)
α (°)	90	90
β (°)	90.1500(10)	111.4410(10)
γ (°)	90	90
Unit cell volume (Å ³)	4269.74(19)	4461.9(3)
<i>T</i> (K)	173(2)	173(2)
<i>Z</i>	4	4
Radiation type	Mo/ <i>K</i> α	Mo/ <i>K</i> α
μ (mm ⁻¹)	1.825	1.767
<i>D</i> _c /g (cm ⁻³)	1.553	1.613
θ range	2.74-25.34	2.74- 25.33
Index ranges	-26 ≤ <i>h</i> ≤ 26 -14 ≤ <i>k</i> ≤ 14 -15 ≤ <i>l</i> ≤ 15	-13 ≤ <i>h</i> ≤ 13 -35 ≤ <i>k</i> ≤ 35 -17 ≤ <i>l</i> ≤ 17
<i>F</i> (000)	2020	2188
Reflections collected	35053	25671
Unique reflections [<i>R</i> _{int}]	7474	7813
Reflections with <i>I</i> > 2 σ (<i>I</i>)	4608	6085
Final <i>R</i> indices (<i>I</i> > 2 σ (<i>I</i>) ^{b,c}	<i>R</i> ₁ = 0.0471, <i>wR</i> ₂ = 0.0881	<i>R</i> ₁ = 0.0349, <i>wR</i> ₂ = 0.0712
Final <i>R</i> indices (all data)	<i>R</i> ₁ = 0.0964, <i>wR</i> ₂ = 0.1019	<i>R</i> ₁ = 0.0553, <i>wR</i> ₂ = 0.0765
<i>S</i> (all data)	1.085	1.025

^a The formula and the formula weights include the MeOH solvent molecules which were free.

^b $R_1 = \Sigma (||F_o| - |F_c||) / \Sigma |F_o|$.

^c $wR_2 = [\Sigma [w (F_o^2 - F_c^2)^2] / \Sigma [w (F_o^2)^2]]^{0.5}$, $w = 1/[\sigma^2(F_o^2) + [(ap)^2 + bp]$, where $p = [\max(F_o^2, 0) + 2F_c^2]/3$.

Table S2. Selected bond lengths (Å) and angles (°) for **1**.

Selected bond lengths for 1					
Dy1-O5	2.269(4)	Dy1-N4	2.531(5)	Na1-O6	2.414(5)
Dy1-O6	2.276(4)	Dy1-N2	2.531(5)	Na1-O4	2.514(6)
Dy1-O1	2.275(4)	Dy1-N1	2.550(5)	Na1-O2	2.528(5)
Dy1-O2	2.285(4)	Na1-O5	2.343(5)	Na1-O7	2.679(5)
Dy1-N3	2.496(5)	Na1-O10	2.373(6)	Na1-O8	2.720(6)
Selected bond angles for 1					
O5-Dy1-O6	74.34(15)	O5-Dy1-N3	72.07(16)	O5-Na1-O4	117.47(19)
O6-Dy1-O1	156.31(14)	O1-Dy1-N3	79.19(17)	O5-Na1-O2	72.77(16)
O5-Dy1-O2	78.87(16)	O6-Dy1-N4	69.79(16)	O6-Na1-O2	69.89(16)
O1-Dy1-O2	88.56(17)	N3-Dy1-N1	82.77(17)	O5-Na1-O7	62.04(15)
O5-Dy1-N3	72.07(16)	N4-Dy1-N1	71.86(17)	O10-Na1-O7	81.40(18)
O1-Dy1-N3	79.19(17)	N2-Dy1-N1	63.08(17)	O6-Na1-O7	127.07(19)
O1-Dy1-N4	129.91(17)	O5-Na1-O10	132.3(2)	O4-Na1-O7	98.30(19)
O5-Dy1-O2	78.87(16)	O5-Na1-O6	70.50(16)	O10-Na1-O6	118.5(2)
O1-Dy1-O2	88.56(17)				

Table S3. Selected bond lengths (Å) and angles (°) for **2**.

Selected bond lengths for 2					
Dy1-O1	2.260(3)	N5-O10	1.246(5)	Na1-O4	2.844(4)
Dy1-O2	2.272(3)	N5-Na1	2.850(4)	Na2-O6	2.436(3)
Dy1-O6	2.277(3)	N5-Na2	2.976(4)	Na2-O5	2.456(3)
Dy1-O5	2.286(3)	Na1-O2	2.393(3)	Na2-O9	2.483(4)
Dy1-N1	2.495(4)	Na1-O1	2.447(3)	Na2-O2	2.635(4)
Dy1-N3	2.497(3)	Na1-O9	2.484(4)	Na2-O11	2.681(5)
Dy1-N2	2.505(4)	Na1-O10	2.497(4)	Na2-O8	2.712(3)
Dy1-N4	2.519(3)	Na1-O7	2.607(4)	Na2-O4	2.721(4)
N5-O11	1.229(5)	Na1-O5	2.757(3)	Na2-N5	2.976(4)
N5-O9	1.241(5)	Na1-O3	2.772(4)		
Selected bond angles for 1					
O1-Dy1-O2	78.08(10)	O2-Na1-O1	72.27(10)	O5-Na2-O2	67.65(10)
O2-Dy1-O6	78.62(11)	O9-Na1-O10	51.02(11)	O11-Na2-O8	114.77(12)
O6-Dy1-O5	75.79(10)	O7-Na1-O5	57.63(9)	O9 Na2 O4	96.93(14)
O1-Dy1-N1	71.90(11)	N3-Dy1-N4	64.27(11)	O2 Na1 O4	57.69(10)

Table S4. The possible geometries of octa-coordination metal centers and Deviation parameters from each ideal polyhedron for Dy in complex **1**.

Point group	Geometry	Polyhedron	Dy1
D_{8h}	OP-8	Octagon	31.857
C_{7v}	HPY-8	Heptagonal pyramid	20.930
D_{6h}	HBPY-8	Hexagonal bipyramid	13.836
O_h	CU-8	Cube	9.365
D_{4d}	SAPR-8	Square antiprism	1.123
D_{2d}	TDD-8	Triangular dodecahedron	2.698
D_{2d}	JGBF-8	Johnson - Gyrobifastigium (J26)	13.407
D_{3h}	JETBPY-8	Johnson - Elongated triangular bipyramid (J14)	27.132
C_{2v}	JBTP-8	Johnson - Biaugmented trigonal prism (J50)	3.193
C_{2v}	BTPR-8	Biaugmented trigonal prism	2.423
D_{2d}	JSD-8	Snub disphenoid (J84)	5.109
T_d	TT-8	Triakis tetrahedron	10.117
D_{3h}	ETBPY-8	Elongated trigonal bipyramid	22.532

Table S5. The possible geometries of octa-coordination metal centers and Deviation parameters from each ideal polyhedron for Dy in complex **2**.

Point group	Geometry	Polyhedron	Dy1
D_{8h}	OP-8	Octagon	32.546
C_{7v}	HPY-8	Heptagonal pyramid	22.438
D_{6h}	HBPY-8	Hexagonal bipyramid	15.627
O_h	CU-8	Cube	10.293
D_{4d}	SAPR-8	Square antiprism	0.955
D_{2d}	TDD-8	Triangular dodecahedron	2.181
D_{2d}	JGBF-8	Johnson - Gyrobifastigium (J26)	14.520
D_{3h}	JETBPY-8	Johnson - Elongated triangular bipyramid (J14)	27.771
C_{2v}	JBTP-8	Johnson - Biaugmented trigonal prism (J50)	3.074
C_{2v}	BTPR-8	Biaugmented trigonal prism	2.526
D_{2d}	JSD-8	Snub disphenoid (J84)	5.006
T_d	TT-8	Triakis tetrahedron	11.101
D_{3h}	ETBPY-8	Elongated trigonal bipyramid	23.244

Table S6. The details for parameters involved in SAP geometry for **1** and **2**.

Skew angle ($\varnothing$) of complex 1 (°)		Skew angle ($\varnothing$) of complex 2 (°)	
$\varnothing_1 = 52.82(1)$	$\varnothing_2 = 43.05(1)$	$\varnothing_1 = 45.97(9)$	$\varnothing_2 = 44.32(1)$
$\varnothing_3 = 49.22(2)$	$\varnothing_4 = 43.91(2)$	$\varnothing_3 = 45.23(1)$	$\varnothing_4 = 47.18(9)$
Mean value of $\varnothing = 47.25(2)$		Mean value of $\varnothing = 45.68(5)$	
Parameter γ and magic angles α for complex 1 (°)		Parameter γ and magic angles α for complex 2 (°)	
$\gamma(\text{O1-Dy1-N2}) = 109.62(2)$	$\alpha(\text{O1-N2}) = 54.81(1)$	$\gamma(\text{O1-Dy1-N2}) = 107.74(1)$	$\alpha(\text{O1-N2}) = 53.87(1)$
$\gamma(\text{O2-Dy1-N1}) = 119.94(2)$	$\alpha(\text{O2-N1}) = 59.97(1)$	$\gamma(\text{O2-Dy1-N1}) = 114.95(1)$	$\alpha(\text{O2-N1}) = 57.48(1)$
$\gamma(\text{O5-Dy1-N4}) = 111.69(2)$	$\alpha(\text{O5-N4}) = 55.85(1)$	$\gamma(\text{O5-Dy1-N4}) = 114.71(1)$	$\alpha(\text{O5-N4}) = 57.36(1)$
$\gamma(\text{O6-Dy1-N3}) = 104.09(2)$	$\alpha(\text{O6-N3}) = 52.05(1)$	$\gamma(\text{O6-Dy1-N3}) = 105.80(1)$	$\alpha(\text{O6-N3}) = 52.91(1)$
Mean value of $\gamma = 111.34$	Mean value of $\alpha = 55.67$	Mean value of $\gamma = 110.80$	Mean value of $\alpha = 55.41$
Values of d_{in} for 1 (Å)		Values of d_{in} for 2 (Å)	
$d_{\text{in}}(\text{N1-N2}) = 2.66(8)$	$d_{\text{in}}(\text{N2-O2}) = 2.87(6)$	$d_{\text{in}}(\text{N1-N2}) = 2.65(5)$	$d_{\text{in}}(\text{N2-O2}) = 2.85(4)$
$d_{\text{in}}(\text{O1-O2}) = 3.18(6)$	$d_{\text{in}}(\text{N1-O1}) = 2.82(6)$	$d_{\text{in}}(\text{O1-O2}) = 2.85(4)$	$d_{\text{in}}(\text{N1-O1}) = 2.80(4)$
$d_{\text{in}}(\text{N3-N4}) = 2.65(6)$	$d_{\text{in}}(\text{N4-O6}) = 2.76(7)$	$d_{\text{in}}(\text{N3-N4}) = 2.67(4)$	$d_{\text{in}}(\text{N4-O6}) = 2.79(5)$
$d_{\text{in}}(\text{O5-O6}) = 2.74(6)$	$d_{\text{in}}(\text{N3-O5}) = 2.81(7)$	$d_{\text{in}}(\text{O5-O6}) = 2.80(3)$	$d_{\text{in}}(\text{N3-O5}) = 2.87(5)$
Mean value of $d_{\text{in}} = 2.81(6)$		Mean value of $d_{\text{in}} = 2.79(4)$	
$d_{\text{pp}}(\text{A-B})$ in 1 (Å)		$d_{\text{pp}}(\text{A-B})$ in 2 (Å)	
$d_{\text{pp}}(\text{N1-B}) = 2.82(5)$	$d_{\text{pp}}(\text{N2-B}) = 2.88(5)$	$d_{\text{pp}}(\text{N3-B}) = 2.87(3)$	$d_{\text{pp}}(\text{N4-B}) = 2.96(3)$
$d_{\text{pp}}(\text{O1-B}) = 2.65(4)$	$d_{\text{pp}}(\text{O2-B}) = 2.33(5)$	$d_{\text{pp}}(\text{O5-B}) = 2.28(3)$	$d_{\text{pp}}(\text{O6-B}) = 2.67(3)$
Mean value of $d_{\text{pp}}(\text{A-B}) = 2.67(5)$		Mean value of $d_{\text{pp}}(\text{A-B}) = 2.70(3)$	
$d_{\text{pp}}^* = 2.68$ for 1		$d_{\text{pp}}^* = 2.71$ for 2	
Dihedral angle (θ) for 1 (°)		Dihedral angle (θ) for 2 (°)	
$\theta = 7.73(1)$		$\theta = 10.26(7)$	
$\sigma_{\varnothing}^2$ and σ_{α}^2 for 1		$\sigma_{\varnothing}^2$ and σ_{α}^2 for 1	
$\sigma_{\varnothing}^2 = 20.99, \sigma_{\alpha}^2 = 8.96$		$\sigma_{\varnothing}^2 = 1.55, \sigma_{\alpha}^2 = 4.62$	

$\varnothing_1 = \varnothing(\text{O1N2}[\text{Dy1}]\text{O5N4})$, $\varnothing_2 = \varnothing(\text{O5N4}[\text{Dy1}]\text{O2N1})$, $\varnothing_3 = \varnothing(\text{O2N1}[\text{Dy1}]\text{O6N3})$ and $\varnothing_4 = \varnothing(\text{O6N3}[\text{Dy1}]\text{O1N2})$, in which [Dy] represents Dy used doubly with its left two atoms and its right two atoms to form two planes, for example, $\varnothing(\text{O1N2}[\text{Dy1}]\text{O5N4})$ means the dihedral angle of O1N2Dy1 plane and Dy1O5N4 plane. The $d_{\text{pp}}(\text{A-B})$ means the distance between coordination A in the upper plane and the lower plane B. d_{pp}^* values were calculated using the program Mercury by defining the centroids of the upper and lower planes and then measuring the distances of between centroids. The θ means the dihedral angle between the upper and lower planes.

Table S7. Best fitted parameters obtained for the extended Debye model with ac susceptibility data from SQUID magnetometer of compound **1** under a 1000 Oe dc applied field.

χ_s	χ_T	τ	α	Residual
0.01947	0.56932	0.02917	0.05455	0.00547
0.01156	0.53641	0.01483	0.08109	0.00186
0.00918	0.49304	0.00745	0.08532	0.00307
0.01313	0.45408	0.0038	0.06633	0.00502
0.00749	0.41116	0.00217	0.09428	0.00128
0.00869	0.37611	0.00131	0.0797	0.00117
0.01149	0.35369	8.58386E-4	0.07978	7.79053E-4
0.00834	0.33272	5.79112E-4	0.10039	4.19364E-4
0.01379	0.30912	4.11647E-4	0.0889	2.54813E-4
0.02035	0.28907	3.02129E-4	0.07483	3.8817E-4
1.25508E-15	0.2617	1.72854E-4	0.08916	0.00101
1.74329E-15	0.25968	1.52646E-4	0.10813	2.55041E-4
2.00091E-15	0.24865	1.18321E-4	0.11166	2.57206E-4
3.08763E-15	0.23627	9.17539E-5	0.11636	5.79399E-5
5.61773E-15	0.22642	7.33018E-5	0.10906	5.22785E-5
8.57844E-15	0.21572	5.86831E-5	0.09752	3.97937E-5
2.85902E-14	0.20418	5.00007E-5	0.04799	4.91185E-5
1.44781E-14	0.19746	3.97147E-5	0.05882	3.27553E-5
2.23747E-14	0.18955	3.344E-5	0.0658	1.25681E-5
2.52677E-14	0.18276	2.76605E-5	0.06323	8.66633E-6
3.39179E-14	0.17734	2.26577E-5	0.08821	8.42287E-6
0.084 (mean value)				

Table S8. Best fitted parameters obtained for the extended Debye model with ac susceptibility data from SQUID magnetometer of compound **2** under a 1000 Oe dc applied field.

χ_s	χ_T	τ	α	Residual
0.01104	0.59637	0.06728	0.10027	0.00144
0.01166	0.50847	0.03175	0.0484	0.00178
0.01224	0.45649	0.01629	0.01381	8.22788E-4
0.00843	0.42761	0.01015	0.04201	9.4092E-4
0.01006	0.39058	0.00596	0.02378	3.84582E-4
0.00844	0.36031	0.00396	0.02168	7.73638E-4
0.01027	0.33908	0.00264	0.01372	7.53728E-4
0.00758	0.3195	0.00188	0.02582	3.39378E-4
0.00587	0.30338	0.00137	0.03909	4.96191E-5
0.00698	0.28864	0.00102	0.03441	8.40152E-5
0.00905	0.26894	7.59953E-4	0.01536	8.48753E-5
0.00737	0.25707	5.89625E-4	0.02513	4.84957E-5
0.00667	0.24517	4.63357E-4	0.02726	5.07325E-5
0.00577	0.23541	3.67732E-4	0.03195	3.97054E-5
0.00863	0.22482	2.99905E-4	0.02499	2.27902E-5
0.00438	0.21537	2.36308E-4	0.03763	1.92678E-5
0.00283	0.20586	1.90007E-4	0.03537	6.23492E-5
0.00363	0.1974	1.5551E-4	0.02663	2.72784E-5
0.00273	0.19022	1.25633E-4	0.03304	4.71783E-5
6.65748E-17	0.18396	1.00214E-4	0.03628	3.68821E-5
1.95225E-18	0.17752	8.17214E-5	0.03081	3.82762E-5
6.71521E-18	0.17088	6.53799E-5	0.02086	4.21858E-5
0.031 (mean value)				

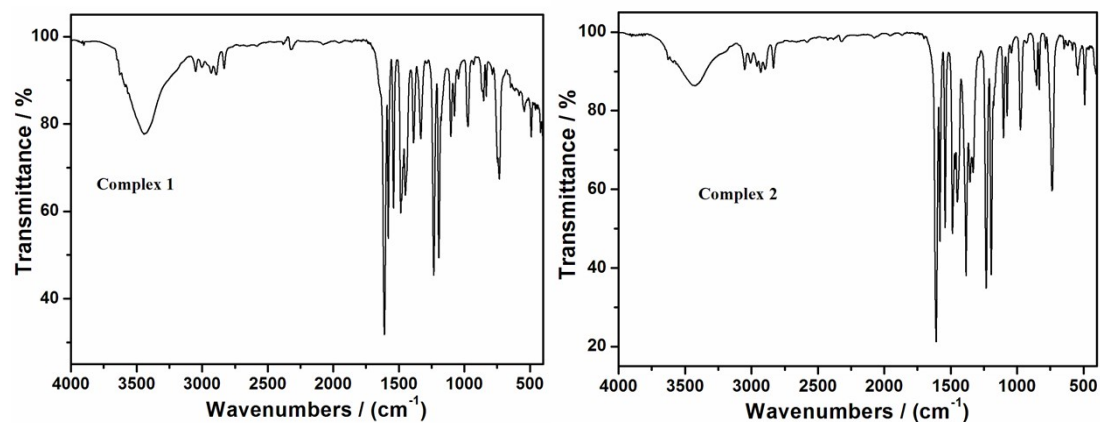


Figure S1. IR spectra of complexes **1** (left) and **2** (right).

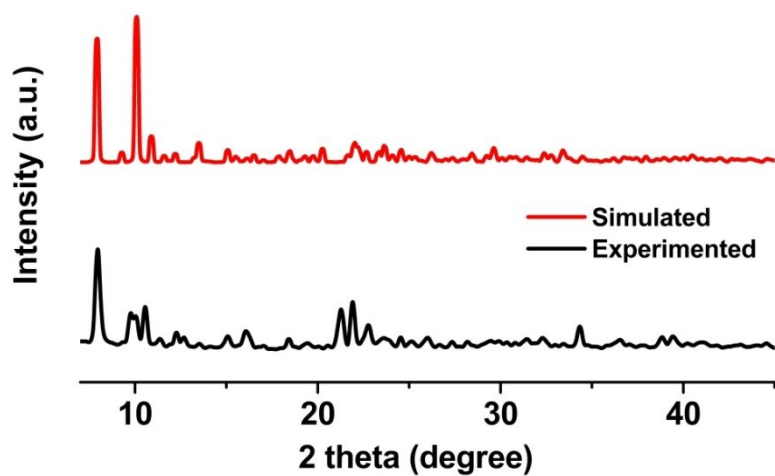


Figure S2. Experimental and simulated X-ray powder diffractograms for **1**.

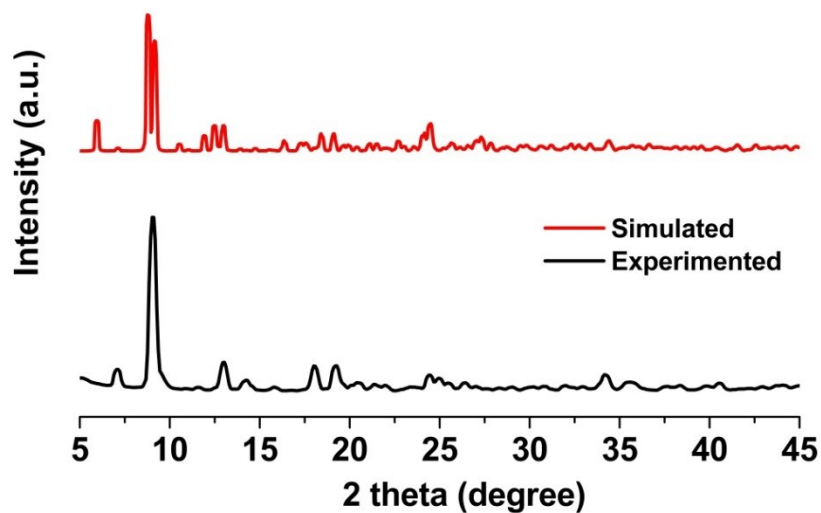


Figure S3. Experimental and simulated X-ray powder diffractograms for **2**.

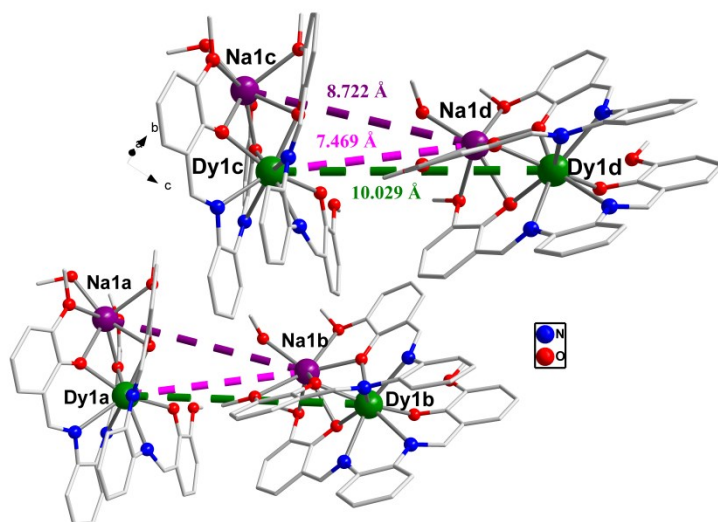


Figure S4. The plot showing the distances of $\text{Na}^{\text{I}} \cdots \text{Na}^{\text{I}}$ (violet), $\text{Dy}^{\text{III}} \cdots \text{Na}^{\text{I}}$ (pink) and $\text{Dy}^{\text{III}} \cdots \text{Dy}^{\text{III}}$ (green) in the crystal lattices of **1**. Symmetry codes: a (2+x, -1+y, z); b (2+x, 1.5-y, 0.5+z); c (2+x, y, z); d (2+x, 2.5-y, 0.5+z).

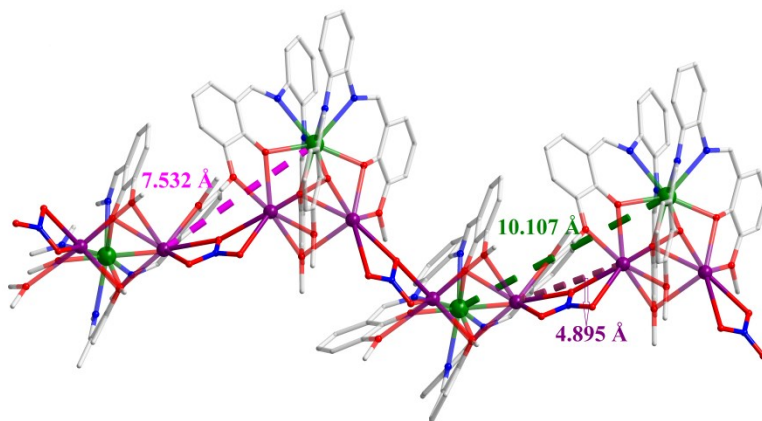


Figure S5. The plot showing the distances of $\text{Na}^{\text{I}} \cdots \text{Na}^{\text{I}}$ (violet), $\text{Dy}^{\text{III}} \cdots \text{Na}^{\text{I}}$ (pink) and $\text{Dy}^{\text{III}} \cdots \text{Dy}^{\text{III}}$ (green) between the neighboring DyNa_2 units within 1D chain of **2**.

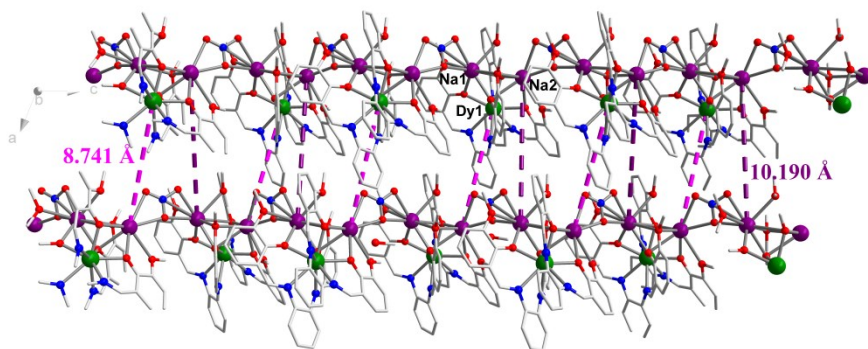


Figure S6. The plot showing the distances of $\text{Dy}^{\text{III}} \cdots \text{Na}^{\text{I}}$ (pink) and $\text{Na}^{\text{I}} \cdots \text{Na}^{\text{I}}$ (violet) between the 1D chains of **2**.

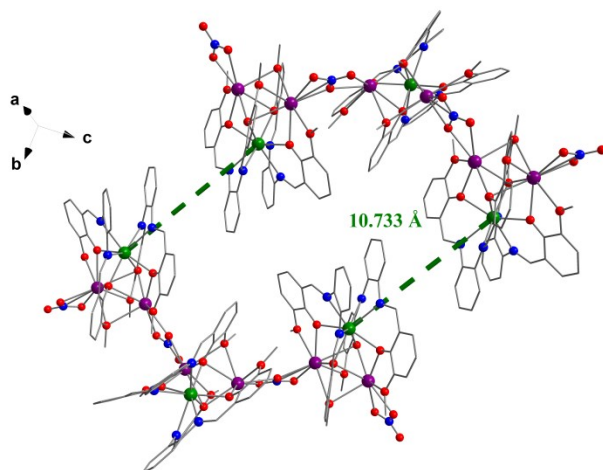


Figure S7. The plot showing the interchain distances of Dy^{III}...Dy^{III} (green) of **2**.

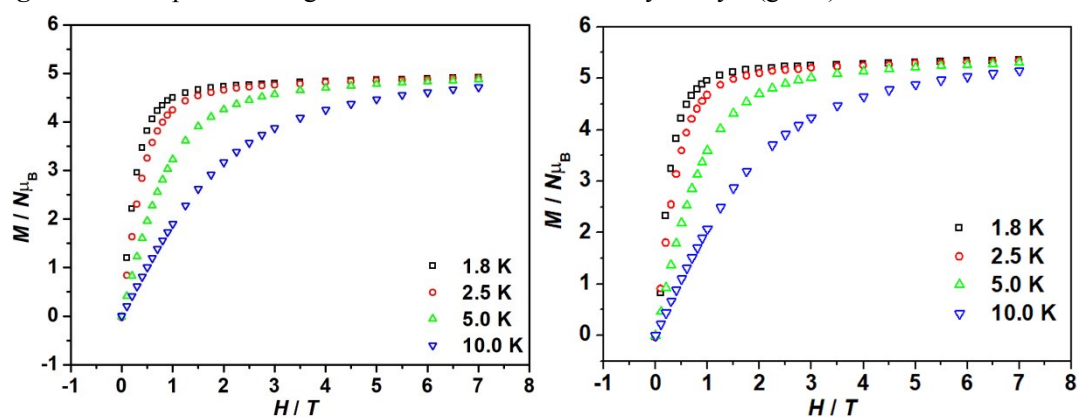


Figure S8. Field dependence of magnetization, M , at different temperature for complex **1** (left) and **2** (right).

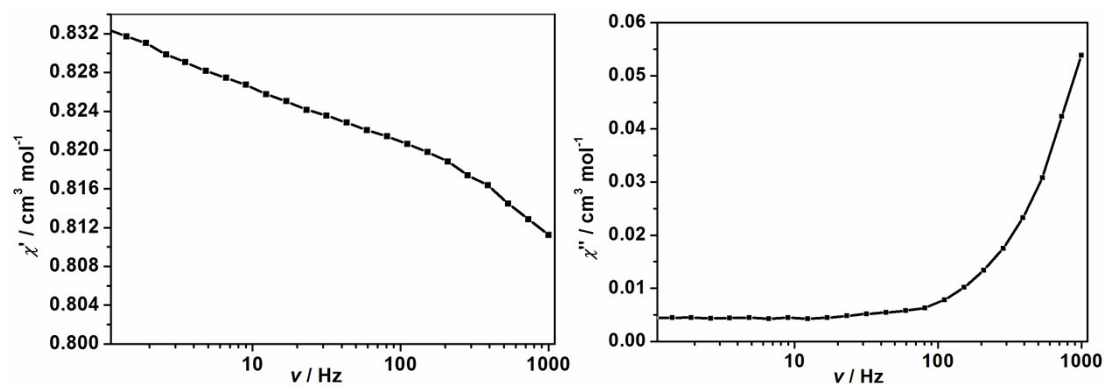


Figure S9. Frequency dependence of the in-phase (χ' , left) and the out-of-phase (χ'' , right) ac signals of **1** under a zero dc field at 3.0 K.

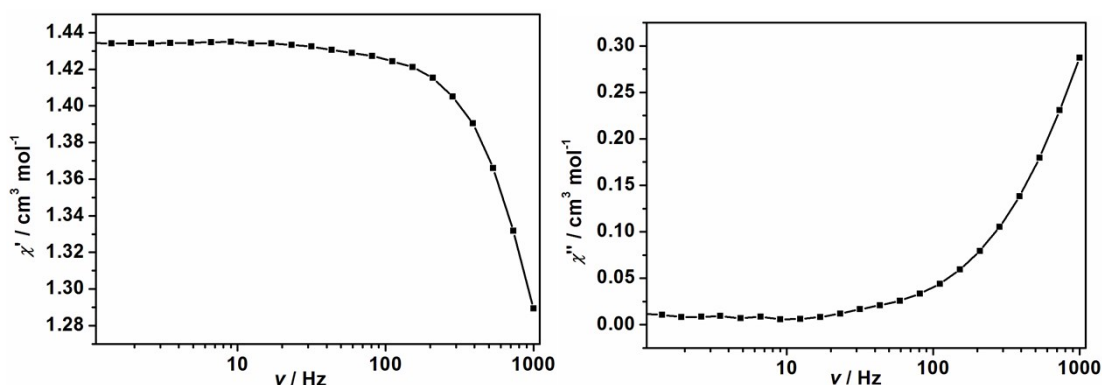


Figure S10. Frequency dependence of the in-phase (χ' , left) and the out-of-phase (χ'' , right) ac signals of **2** under a zero dc field at 1.8 K.

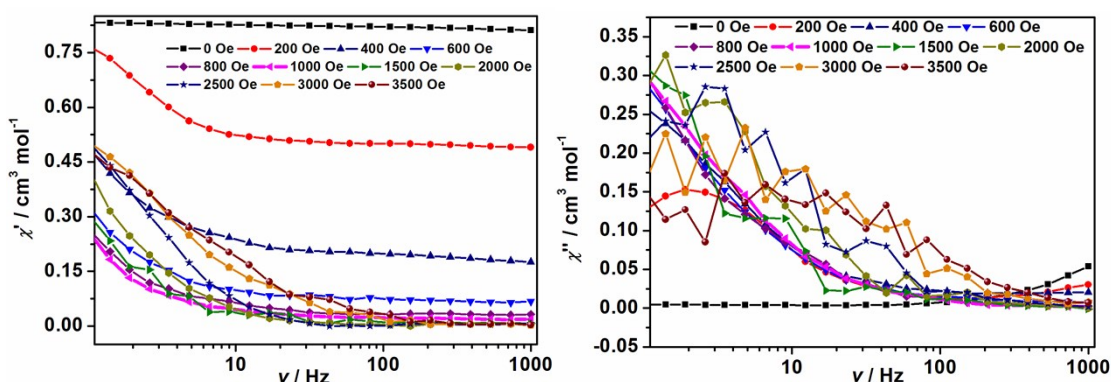


Figure S11. Frequency dependence of the in-phase (χ' , left) and out-of-phase (χ'' , right) ac signals for **1** under different dc fields at 3.0 K. To see easily, the χ' and χ'' ac susceptibility signals at the optimal dc field were represented by thick pink lines.

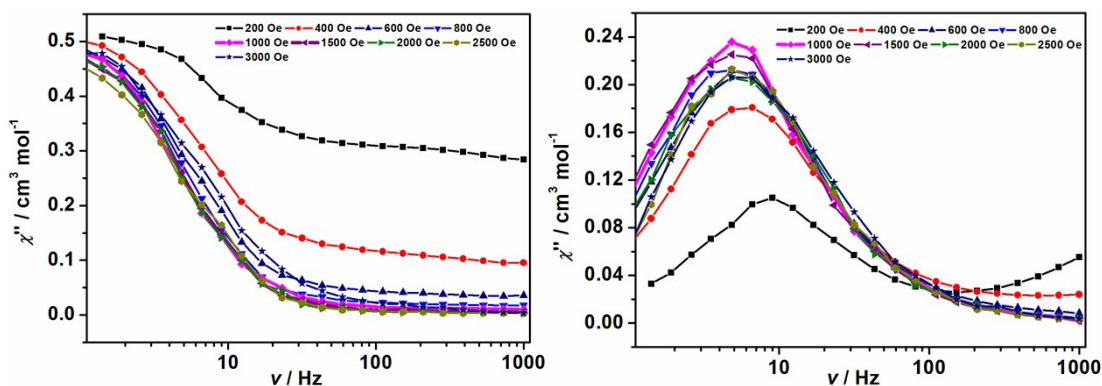


Figure S12. Frequency dependence of the in-phase (χ' , left) and out-of-phase (χ'' , right) ac signals for **2** under different dc fields at 5.0 K. To see easily, the χ' and χ'' ac susceptibility signals at the optimal dc field were represented by thick pink lines.

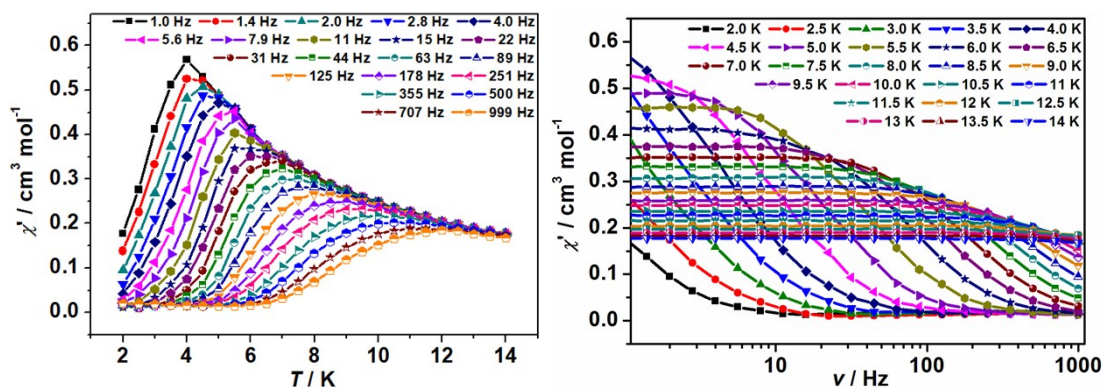


Figure S13. Temperature dependence (left) and frequency dependence (right) of the in-phase (χ') ac signals for **1** under a 1000 Oe dc field.

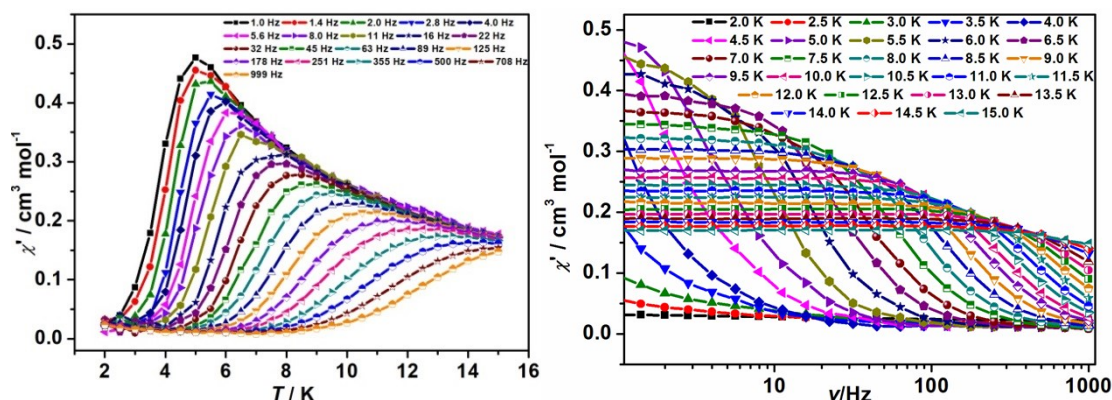


Figure S14. Temperature dependence (left) and frequency dependence (right) of the in-phase (χ') ac signals for **2** under a 1000 Oe dc field.

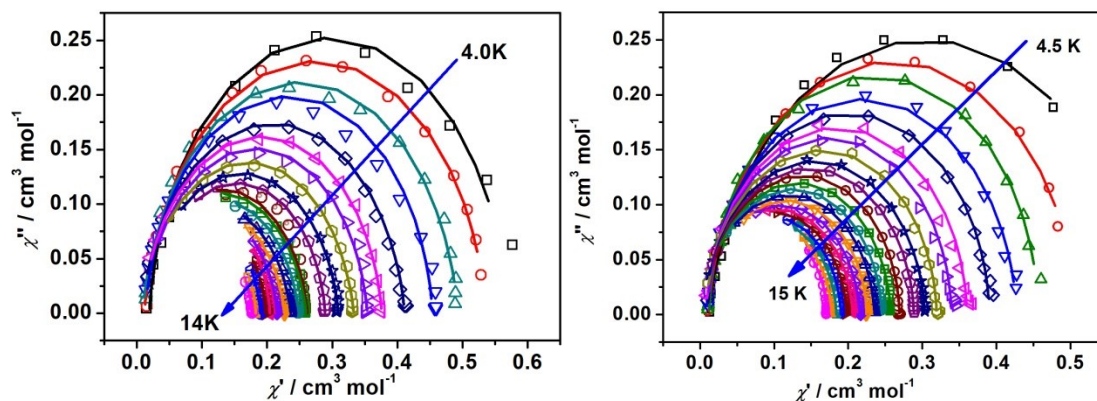


Figure S15. Cole-Cole plots for complex **1** (left) and **2** (right) under a 1000 Oe dc field. The solid lines represent the fitted results by using the generalized single relaxation process Debye model.

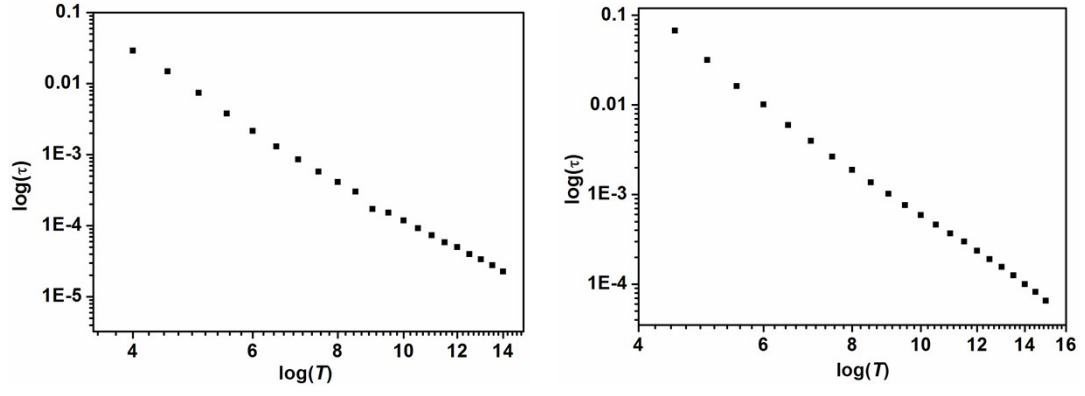


Figure S16. Plots of $\ln(\tau)$ vs. T^{-1} and $\log(\tau)$ vs. $\log(T)$ for **1** (left) and **2** (right).