

Supporting Information

Molecular motion of halogenated ethylammonium / [18]crown-6 supramolecular ions in nickel dithiolate magnetic crystals

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§1. Crystal data, crystal data, data collection, and reduction parameter.

Table S1. Crystal data, data collection, and reduction parameter for crystals **1**, **2**, **3** and **4** at the lowest temperatures.

	1@113K	2@93K	3@93K	4@106K
<i>Temperature / K</i>	113	93	93	106
<i>Crystal Dimensions / mm³</i>	0.724×0.378×0.207	0.507×0.211×0.09	0.69×0.60×0.18	0.40×0.11×0.07
<i>Chemical formula</i>	C ₂₀ H ₃₂ NNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNiO ₆ S ₁₀	C ₂₀ H ₃₁ ClNiO ₆ S ₁₀	C ₂₀ H ₃₁ BrNNiO ₆ S ₁₀
<i>Formula weight</i>	761.77	779.77	796.22	840.73
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.1765(4)	10.3645(3)	11.2107(3)	10.9869(3)
<i>b / Å</i>	12.2812(3)	12.6254(5)	12.3150(2)	12.0485(4)
<i>c / Å</i>	12.6527(4)	12.9123(4)	12.7451(3)	13.7559(5)
<i>α / deg</i>	87.364(2)	69.835(3)	92.2792(18)	115.833(4)
<i>β / deg</i>	68.465(3)	79.972(3)	112.126(2)	91.587(3)
<i>γ / deg</i>	80.094(2)	79.948(3)	96.0601(18)	100.906(3)
<i>V / Å³</i>	1591.12(9)	1549.82(10)	1614.83(7)	1596.83(10)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.590	1.671	1.638	1.748
<i>μ(Mo Kα) / cm⁻¹</i>	1.302	1.343	1.366	2.549
<i>2θ_{max} / deg</i>	61.784	61.986	61.838	61.140
<i>Reflections measured</i>	22773	18605	45358	19669
<i>Independent reflections</i>	7934	7679	8651	7891
<i>Reflections used</i>	7934	7679	8651	7891
<i>R₁^a</i>	0.0357	0.0324	0.0495	0.0356
<i>R_w(F²)^b</i>	0.0980	0.0792	0.1395	0.0703
<i>GOF</i>	1.051	1.070	1.098	0.995
<i>CCDC number</i>	2053279	2053282	2053390	2053298

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

Table S2. Crystal data, data collection, and reduction parameter for crystal **1**.

<i>Temperature / K</i>	293
<i>Crystal Dimensions / mm³</i>	0.724×0.378×0.207
<i>Chemical formula</i>	C ₂₀ H ₃₂ NNiO ₆ S ₁₀
<i>Formula weight</i>	761.77
<i>Crystal System</i>	Triclinic
<i>Space group</i>	<i>P</i> -1
<i>a / Å</i>	11.2486(6)
<i>b / Å</i>	12.3540(8)
<i>c / Å</i>	12.7375(8)
<i>α / deg</i>	87.498(5)
<i>β / deg</i>	69.343(6)
<i>γ / deg</i>	80.758(5)
<i>V / Å³</i>	1634.62(18)
<i>Z</i>	2
<i>D_{calc} / g·cm⁻³</i>	1.548
<i>μ(Mo Kα) / cm⁻¹</i>	1.267
<i>2θ_{max} / deg</i>	62.114
<i>Reflections measured</i>	20189
<i>Independent reflections</i>	8034
<i>Reflections used</i>	8034
<i>R₁^a</i>	0.0365
<i>R_w (F²)^b</i>	0.0945
<i>GOF</i>	1.051
<i>CCDC number</i>	2053280

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

Table S3. Crystal data, data collection, and reduction parameter for crystal **2** at different temperatures.

	2			
<i>Temperature / K</i>	113	133	153	173
<i>Crystal Dimensions / mm³</i>	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09
<i>Chemical formula</i>	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀
<i>Formula weight</i>	779.77	779.77	779.77	779.77
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	10.3864(2)	10.3884(3)	10.3994(3)	10.4067(3)
<i>b / Å</i>	12.6343(5)	12.6271(4)	12.6286(5)	12.6314(5)
<i>c / Å</i>	12.9338(5)	12.9344(4)	12.9431(4)	12.9479(4)
<i>α / deg</i>	69.944(3)	70.113(3)	70.317(3)	70.517(3)
<i>β / deg</i>	79.973(3)	79.966(2)	79.956(2)	79.892(2)
<i>γ / deg</i>	79.970(3)	80.068(2)	80.078(3)	80.138(3)
<i>V / Å³</i>	1557.90(10)	1559.30(9)	1564.10(9)	1567.96(9)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.662	1.661	1.656	1.650
<i>μ(Mo Kα) / cm⁻¹</i>	1.336	1.335	1.331	1.327
<i>2θ_{max} / deg</i>	62.114	61.916	61.652	61.248
<i>Reflections measured</i>	19702	19502	19288	19026
<i>Independent reflections</i>	7645	7689	7694	7718
<i>Reflections used</i>	7645	7689	7694	7718
<i>R₁^a</i>	0.0329	0.0386	0.0340	0.0366
<i>R_w (F²)^b</i>	0.0806	0.0941	0.0865	0.0925
<i>GOF</i>	1.054	0.990	1.075	1.075
<i>CCDC number</i>	2053283	2053284	2053285	2053286

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	193	213	233	253
<i>Crystal Dimensions / mm³</i>	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09
<i>Chemical formula</i>	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀
<i>Formula weight</i>	779.77	779.77	779.77	779.77
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	10.4204(3)	10.4363(3)	10.4520(3)	10.4754(3)
<i>b / Å</i>	12.6319(4)	12.6370(4)	12.6374(4)	12.6503(4)
<i>c / Å</i>	12.9511(4)	12.9511(4)	12.9527(4)	12.9637(4)
<i>α / deg</i>	70.735(3)	70.974(3)	71.178(3)	71.385(3)
<i>β / deg</i>	79.898(2)	79.852(2)	79.806(2)	79.802(2)
<i>γ / deg</i>	80.250(2)	80.316(2)	80.408(2)	80.550(2)
<i>V / Å³</i>	1572.90(9)	1578.17(8)	1582.77(9)	1591.61(9)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.646	1.641	1.636	1.627
<i>μ(Mo Kα) / cm⁻¹</i>	1.323	1.319	1.315	1.308
<i>2θ_{max} / deg</i>	61.70	61.50	62.07	61.34
<i>Reflections measured</i>	19734	19305	19876	19688
<i>Independent reflections</i>	7797	7814	7847	7844
<i>Reflections used</i>	7797	7814	7847	7844
<i>R₁^a</i>	0.0414	0.0336	0.0335	0.0343
<i>R_w (F²)^b</i>	0.1011	0.0838	0.0864	0.0810
<i>GOF</i>	1.016	1.050	1.044	1.047
<i>CCDC number</i>	2053287	2053288	2053289	2053290

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	273	293	313	333
<i>Crystal Dimensions / mm³</i>	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09	0.507×0.211×0.09
<i>Chemical formula</i>	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀
<i>Formula weight</i>	779.77	779.77	779.77	779.77
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	10.4883(2)	10.5115(3)	10.5377(3)	10.5475(3)
<i>b / Å</i>	12.6467(3)	12.6531(4)	12.6697(3)	12.6632(3)
<i>c / Å</i>	12.9675(3)	12.9821(3)	13.0079(4)	13.0124(4)
<i>α / deg</i>	71.543(2)	71.701(2)	71.903(2)	72.091(3)
<i>β / deg</i>	79.792(2)	79.787(2)	79.822(2)	79.842(3)
<i>γ / deg</i>	80.656(2)	80.782(2)	80.956(2)	81.154(2)
<i>V / Å³</i>	1595.30(6)	1603.24(8)	1615.05(8)	1618.65(8)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.623	1.615	1.603	1.600
<i>μ(Mo Kα) / cm⁻¹</i>	1.305	1.298	1.289	1.286
<i>2θ_{max} / deg</i>	61.89	61.438	58.006	57.812
<i>Reflections measured</i>	39570	20112	17913	19117
<i>Independent reflections</i>	8451	7909	7336	7320
<i>Reflections used</i>	8451	7909	7336	7320
<i>R₁^a</i>	0.0357	0.0336	0.0354	0.0383
<i>R_w (F²)^b</i>	0.1327	0.0801	0.0855	0.0986
<i>GOF</i>	0.870	1.055	1.042	1.077
<i>CCDC number</i>	2053291	2053292	2053293	2053294

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	353	373
<i>Crystal Dimensions / mm³</i>	0.507×0.211×0.09	0.507×0.211×0.09
<i>Chemical formula</i>	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀	C ₂₀ H ₃₁ FNNiO ₆ S ₁₀
<i>Formula weight</i>	779.77	779.77
<i>Crystal System</i>	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	10.5673(4)	10.5742(7)
<i>b / Å</i>	12.6724(4)	12.6426(7)
<i>c / Å</i>	13.0358(6)	13.0305(11)
<i>α / deg</i>	72.219(4)	72.446(6)
<i>β / deg</i>	79.801(4)	79.808(6)
<i>γ / deg</i>	81.293(3)	81.420(5)
<i>V / Å³</i>	1627.17(12)	1626.1(2)
<i>Z</i>	2	2
<i>D_{calc} / g·cm⁻³</i>	1.592	1.593
<i>μ(Mo Kα) / cm⁻¹</i>	1.279	1.280
<i>2θ_{max} / deg</i>	58.162	58.054
<i>Reflections measured</i>	18534	18911
<i>Independent reflections</i>	7331	7413
<i>Reflections used</i>	7331	7413
<i>R₁^a</i>	0.0459	0.0514
<i>R_w (F²)^b</i>	0.1373	0.1505
<i>GOF</i>	0.911	0.911
<i>CCDC number</i>	2053295	2053296

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

Table S4. Crystal data, data collection, and reduction parameter for crystal **3** at different temperatures.

3				
<i>Temperature / K</i>	113	133	153	173
<i>Crystal Dimensions / mm³</i>	0.69×0.60×0.18	0.69×0.60×0.18	0.69×0.60×0.18	0.290×0.249×0.157
<i>Chemical formula</i>	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀
<i>Formula weight</i>	796.22	796.22	796.22	796.22
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.2193(3)	11.2216(3)	11.2185(3)	11.2452(2)
<i>b / Å</i>	12.3191(2)	12.3162(3)	12.3084(3)	12.3442(2)
<i>c / Å</i>	12.7473(3)	12.7541(3)	12.7536(3)	12.8113(2)
<i>α / deg</i>	92.239(2)	92.155(2)	92.083(2)	91.8320(10)
<i>β / deg</i>	111.956(2)	111.864(2)	111.733(2)	111.4310(10)
<i>γ / deg</i>	96.066(2)	96.100(2)	96.128(2)	96.2430(10)
<i>V / Å³</i>	1618.98(7)	1620.96(7)	1621.02(7)	1640.65(5)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.633	1.631	1.631	1.612
<i>μ(Mo Kα) / cm⁻¹</i>	1.363	1.361	1.361	7.865
<i>2θ_{max} / deg</i>	52.742	52.744	61.84	144.97
<i>Reflections measured</i>	37298	35045	41879	39675
<i>Independent reflections</i>	6613	6623	8598	6266
<i>Reflections used</i>	6613	6623	8598	6266
<i>R₁^a</i>	0.0446	0.0419	0.0438	0.0385
<i>R_w (F²)^b</i>	0.1268	0.1168	0.1217	0.1156
<i>GOF</i>	1.077	1.056	1.068	1.020
<i>CCDC number</i>	2053392	2053393	2053394	2053395

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	193	213	233	253
<i>Crystal Dimensions / mm³</i>	0.371×0.624×0.96	0.290×0.249×0.15	0.290×0.249×0.15	0.290×0.249×0.15
<i>Chemical formula</i>	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀
<i>Formula weight</i>	796.22	796.22	796.22	796.22
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.25474(18)	11.2637(2)	11.27140(10)	11.27840(10)
<i>b / Å</i>	12.34881(18)	12.34200(10)	12.34060(10)	12.34600(10)
<i>c / Å</i>	12.82304(17)	12.8337(2)	12.84520(10)	12.86190(10)
<i>α / deg</i>	91.7415(11)	91.3720(10)	91.2410(10)	91.1450(10)
<i>β / deg</i>	111.2915(14)	110.9960(10)	110.8230(10)	110.6780(10)
<i>γ / deg</i>	96.3016(12)	96.4500(10)	96.4520(10)	96.4220(10)
<i>V / Å³</i>	1645.75(4)	1651.13(4)	1655.78(3)	1661.63(3)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.607	1.602	1.597	1.591
<i>μ(Mo Kα) / cm⁻¹</i>	1.341	7.815	7.793	7.765
<i>2θ_{max} / deg</i>	58.252	145.476	146.17	144.498
<i>Reflections measured</i>	33190	42270	41408	41385
<i>Independent reflections</i>	7803	6349	6351	6340
<i>Reflections used</i>	7803	6349	6351	6340
<i>R₁^a</i>	0.0347	0.0310	0.0321	0.0271
<i>R_w (F²)^b</i>	0.0910	0.0815	0.0845	0.0763
<i>GOF</i>	1.084	1.034	1.052	1.053
<i>CCDC number</i>	2053396	2053397	2053398	2053399

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	273	293	313	333
<i>Crystal Dimensions / mm³</i>	0.69×0.60×0.18	0.371×0.624×0.961	0.371×0.624×0.961	0.69×0.60×0.18
<i>Chemical formula</i>	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀
<i>Formula weight</i>	796.22	796.22	796.22	796.22
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.2803(3)	11.2892(3)	11.3149(3)	11.3046(5)
<i>b / Å</i>	12.3375(2)	12.3438(3)	12.3589(3)	12.3319(4)
<i>c / Å</i>	12.8611(3)	12.8805(3)	12.9090(3)	12.9008(6)
<i>α / deg</i>	91.103(2)	91.047(2)	90.9686(19)	90.991(3)
<i>β / deg</i>	110.634(2)	110.467(2)	110.340(2)	110.208(4)
<i>γ / deg</i>	96.375(2)	96.326(2)	96.269(2)	96.173(3)
<i>V / Å³</i>	1661.43(7)	1668.26(7)	1679.59(8)	1675.11(13)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.592	1.583	1.574	1.579
<i>μ(Mo Kα) / cm⁻¹</i>	1.328	1.322	1.314	1.317
<i>2θ_{max} / deg</i>	62.142	58.016	58.032	52.744
<i>Reflections measured</i>	49247	32304	32926	27105
<i>Independent reflections</i>	8848	7906	7989	6845
<i>Reflections used</i>	8848	7906	7989	6845
<i>R₁^a</i>	0.0389	0.0366	0.0389	0.0516
<i>R_w (F²)^b</i>	0.1292	0.1064	0.1150	0.1420
<i>GOF</i>	1.015	1.015	0.996	0.986
<i>CCDC number</i>	2053391	2053400	2053401	2053402

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	353	373
<i>Crystal Dimensions / mm³</i>	0.69×0.60×0.18	0.371×0.624×0.961
<i>Chemical formula</i>	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀	C ₂₀ H ₃₁ CINNiO ₆ S ₁₀
<i>Formula weight</i>	796.22	796.22
<i>Crystal System</i>	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.3244(5)	11.3571(2)
<i>b / Å</i>	12.3495(4)	12.3603(2)
<i>c / Å</i>	12.9385(6)	12.9511(3)
<i>α / deg</i>	90.918(3)	90.864(2)
<i>β / deg</i>	110.128(4)	109.988(2)
<i>γ / deg</i>	96.018(3)	95.958(2)
<i>V / Å³</i>	1686.95(13)	1696.81(6)
<i>Z</i>	2	2
<i>D_{calc} / g·cm⁻³</i>	1.568	1.558
<i>μ(Mo Kα) / cm⁻¹</i>	1.308	1.300
<i>2θ_{max} / deg</i>	58.424	58.26
<i>Reflections measured</i>	30129	33505
<i>Independent reflections</i>	8054	8059
<i>Reflections used</i>	8054	8059
<i>R₁^a</i>	0.0559	0.0396
<i>R_w (F²)^b</i>	0.1553	0.1222
<i>GOF</i>	0.916	0.968
<i>CCDC number</i>	2053403	2053404

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

Table S5. Crystal data, data collection, and reduction parameter for crystal **4** at different temperatures.

4				
<i>Temperature / K</i>	113	133	153	173
<i>Crystal Dimensions / mm³</i>	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07
<i>Chemical formula</i>	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁
<i>Formula weight</i>	840.68	840.68	840.68	840.68
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	10.9810(3)	11.0117(3)	11.0165(3)	11.0374(3)
<i>b / Å</i>	12.0448(3)	12.0652(3)	12.0569(4)	12.0627(3)
<i>c / Å</i>	13.7409(4)	13.7791(4)	13.7786(4)	13.7656(4)
<i>α / deg</i>	115.825(3)	115.895(3)	64.116(3)	64.349(3)
<i>β / deg</i>	91.535(2)	91.351(2)	79.239(3)	79.574(2)
<i>γ / deg</i>	100.927(2)	101.004(2)	78.965(2)	78.920(2)
<i>V / Å³</i>	1593.91(8)	1604.68(8)	1605.03(9)	1611.42(9)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.747	1.740	1.739	1.733
<i>μ(Mo Kα) / cm⁻¹</i>	2.554	2.537	2.536	2.526
<i>2θ_{max} / deg</i>	62.086	61.834	62.064	62.062
<i>Reflections measured</i>	38281	38326	39974	40700
<i>Independent reflections</i>	8378	8423	8484	8560
<i>Reflections used</i>	8378	8423	8484	8560
<i>R₁^a</i>	0.0315	0.0325	0.0337	0.0333
<i>R_w (F²)^b</i>	0.0677	0.0677	0.0706	0.0690
<i>GOF</i>	1.016	1.026	1.032	1.039
<i>CCDC number</i>	2053299	2053300	2053301	2053302

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	193	213	233	253
<i>Crystal Dimensions / mm³</i>	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07
<i>Chemical formula</i>	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁
<i>Formula weight</i>	840.68	840.68	840.68	840.68
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.0467(3)	11.0537(3)	11.0760(3)	11.0864(3)
<i>b / Å</i>	12.0603(3)	12.0557(3)	12.0662(3)	12.0734(3)
<i>c / Å</i>	13.7505(4)	13.7262(4)	13.7164(5)	13.7103(4)
<i>α / deg</i>	64.635(3)	64.890(3)	65.106(3)	65.253(3)
<i>β / deg</i>	79.967(2)	80.331(2)	80.573(3)	80.773(3)
<i>γ / deg</i>	78.865(2)	78.815(2)	78.719(2)	78.660(2)
<i>V / Å³</i>	1615.49(8)	1617.26(9)	1623.84(9)	1627.77(9)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.728	1.726	1.719	1.715
<i>μ(Mo Kα) / cm⁻¹</i>	2.520	2.517	2.507	2.501
<i>2θ_{max} / deg</i>	62.048	62.122	62.116	62.16
<i>Reflections measured</i>	40000	40716	40270	42066
<i>Independent reflections</i>	8603	8593	8614	8660
<i>Reflections used</i>	8603	8593	8614	8660
<i>R₁^a</i>	0.0375	0.0391	0.0405	0.0377
<i>R_w (F²)^b</i>	0.0851	0.0818	0.0923	0.0875
<i>GOF</i>	1.073	1.039	1.019	1.039
<i>CCDC number</i>	2053303	2053304	2053305	2053306

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	273	293	313	333
<i>Crystal Dimensions / mm³</i>	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07	0.40×0.11×0.07
<i>Chemical formula</i>	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁	C ₂₀ H ₃₁ BrNNiO ₆ S ₁
<i>Formula weight</i>	840.68	840.68	840.68	840.68
<i>Crystal System</i>	Triclinic	Triclinic	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.1022(4)	11.1201(5)	11.1694(4)	11.1865(5)
<i>b / Å</i>	12.0838(4)	12.1021(5)	12.1413(4)	12.1459(5)
<i>c / Å</i>	13.7055(5)	13.7058(7)	13.7408(6)	13.7395(6)
<i>α / deg</i>	65.348(3)	65.393(5)	65.543(4)	65.630(4)
<i>β / deg</i>	80.897(3)	80.958(4)	81.123(3)	81.204(4)
<i>γ / deg</i>	78.557(3)	78.454(4)	78.362(3)	78.269(4)
<i>V / Å³</i>	1632.00(11)	1637.40(15)	1656.09(12)	1659.95(14)
<i>Z</i>	2	2	2	2
<i>D_{calc} / g·cm⁻³</i>	1.709	1.705	1.686	1.682
<i>μ(Mo Kα) / cm⁻¹</i>	2.494	2.486	2.458	2.452
<i>2θ_{max} / deg</i>	62.156	62.106	58.064	58.072
<i>Reflections measured</i>	41574	42141	33146	34340
<i>Independent reflections</i>	8724	8744	7890	7868
<i>Reflections used</i>	8724	8744	7890	7868
<i>R₁^a</i>	0.0413	0.0431	0.0416	0.0879
<i>R_w (F²)^b</i>	0.0922	0.0958	0.0958	0.1086
<i>GOF</i>	1.020	1.031	1.049	1.027
<i>CCDC number</i>	2053307	2053308	2053309	2053310

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

<i>Temperature / K</i>	353	373
<i>Crystal Dimensions / mm³</i>	0.40×0.11×0.07	0.40×0.11×0.07
<i>Chemical formula</i>	C ₂₀ H ₃₁ BrNNiO ₆ S ₁₀	C ₂₀ H ₃₁ BrNNiO ₆ S ₁₀
<i>Formula weight</i>	840.68	840.68
<i>Crystal System</i>	Triclinic	Triclinic
<i>Space group</i>	<i>P</i> -1	<i>P</i> -1
<i>a / Å</i>	11.2110(4)	11.2167(4)
<i>b / Å</i>	12.1703(4)	12.1808(4)
<i>c / Å</i>	13.7456(5)	13.7373(5)
<i>α / deg</i>	65.635(3)	65.696(3)
<i>β / deg</i>	81.252(3)	81.308(3)
<i>γ / deg</i>	78.145(3)	78.012(3)
<i>V / Å³</i>	1667.21(10)	1668.72(11)
<i>Z</i>	2	2
<i>D_{calc} / g·cm⁻³</i>	1.675	1.673
<i>μ(Mo Kα) / cm⁻¹</i>	2.442	2.439
<i>2θ_{max} / deg</i>	58.086	58.11
<i>Reflections measured</i>	34600	35265
<i>Independent reflections</i>	7944	7932
<i>Reflections used</i>	7944	7932
<i>R₁^a</i>	0.0522	0.0470
<i>R_w (F²)^b</i>	0.1284	0.1118
<i>GOF</i>	1.016	1.008
<i>CCDC number</i>	2053311	2053312

^a $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ and ^b $R_w = (\sum \omega(|F_o| - |F_c|)^2 / \sum \omega F_o^2)^{1/2}$.

§2. Crystallographically independent molecules with atom numbering

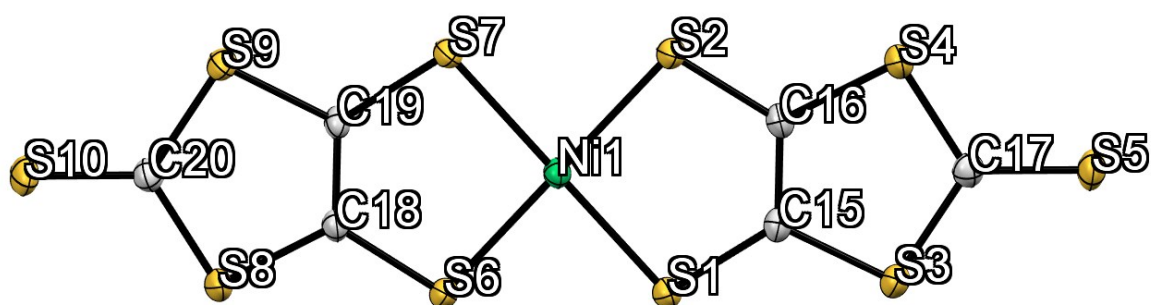
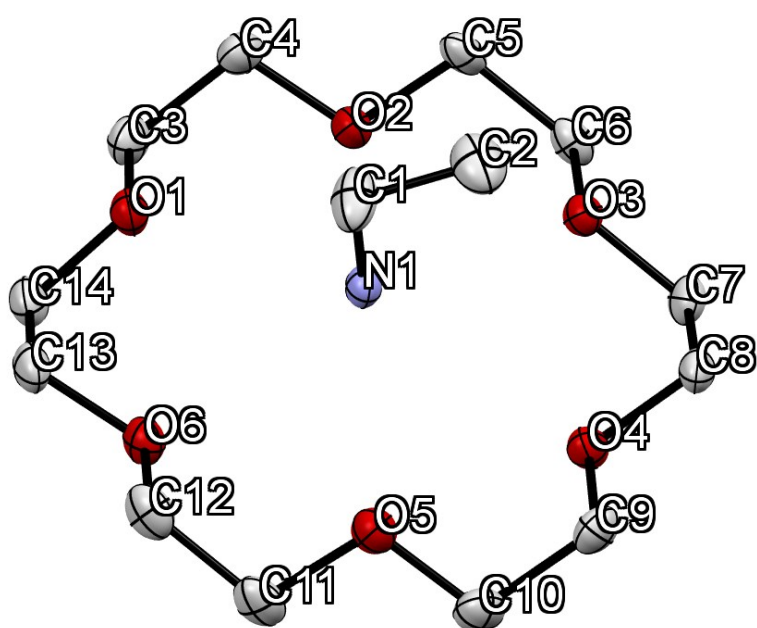


Figure S1. Crystallographically independent molecules in the crystal **1** at 113 K with atomic numbers.

Crystallographically independent molecular structure is depicted as ORTEP with 50% thermal ellipsoids probability level. Hydrogen atoms are omitted for clarity.

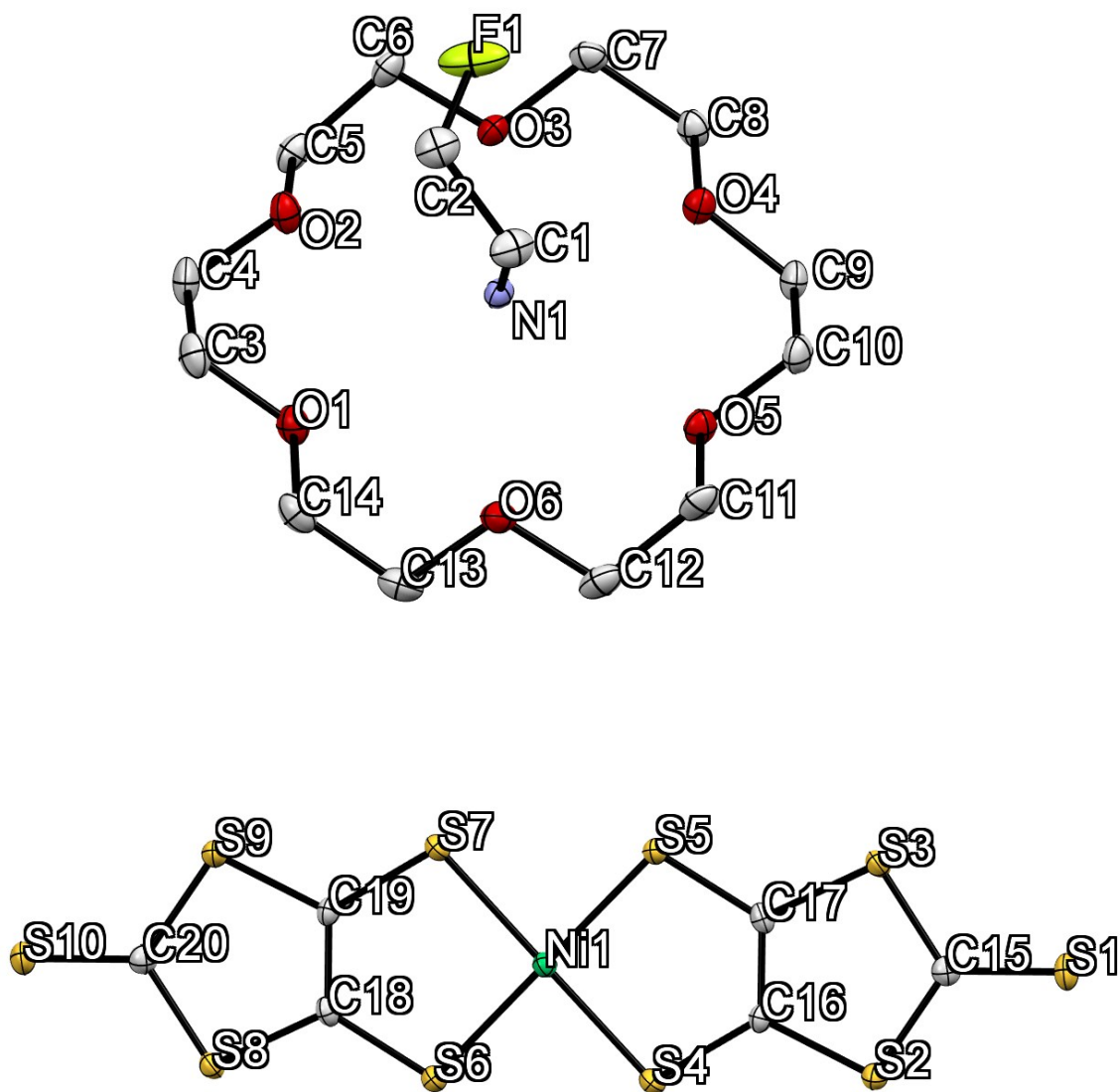


Figure S2. Crystallographically independent molecules in the crystal **2** at 93 K with atomic numbers. The structure is depicted as ORTEP with 50% probability. Hydrogen atoms are omitted for clarity.

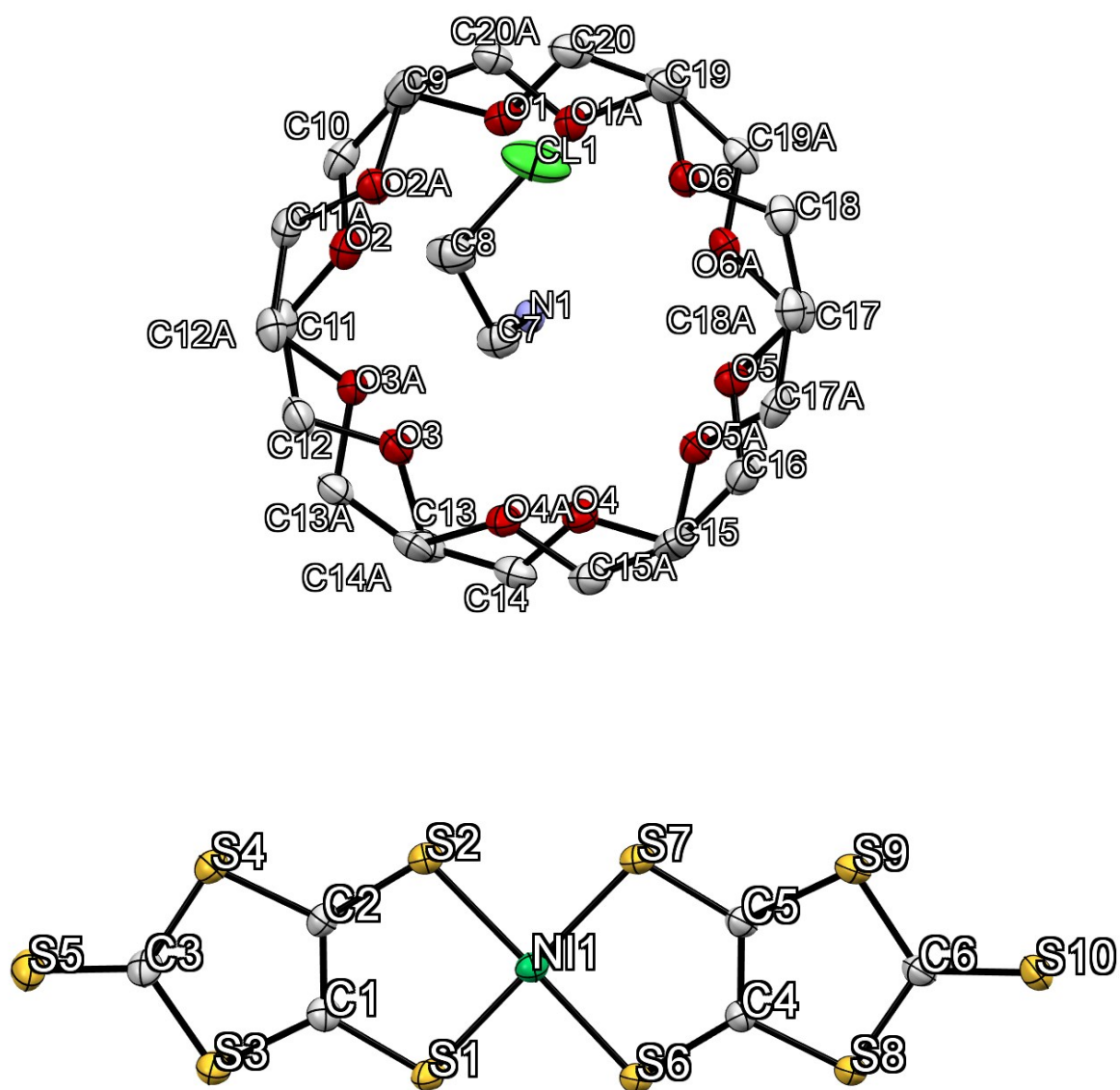


Figure S3. Crystallographically independent molecules in the crystal **3** at 93 K with atomic numbers. The structure is depicted as ORTEP with 50% probability. Hydrogen atoms are omitted for clarity.

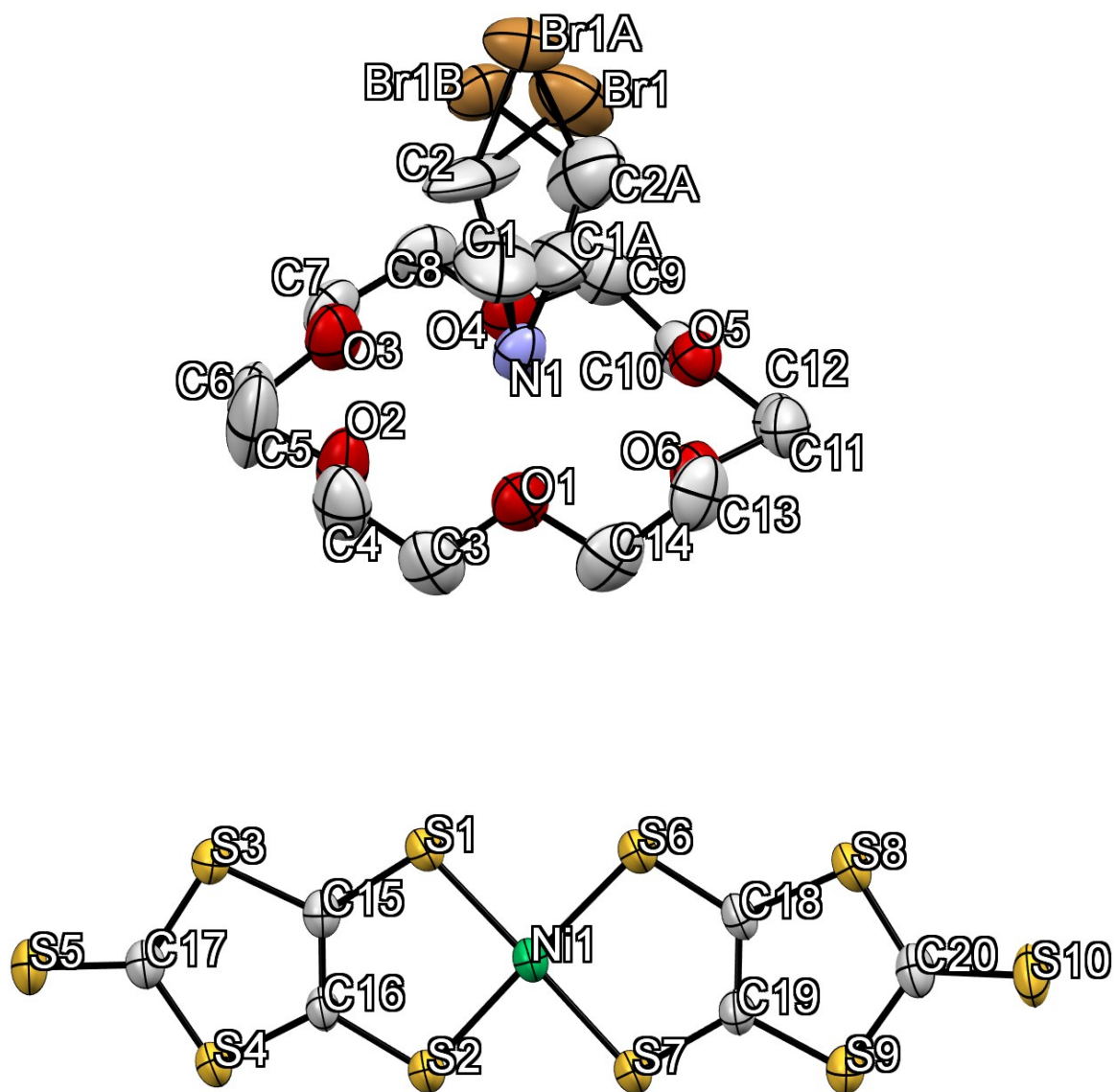


Figure S4. Crystallographically independent molecules in the crystal 4 at 293 K with atomic numbers. The structure is depicted as ORTEP with 50% probability. Hydrogen atoms are omitted for clarity.

§3. Packing structures of crystals 2-4.

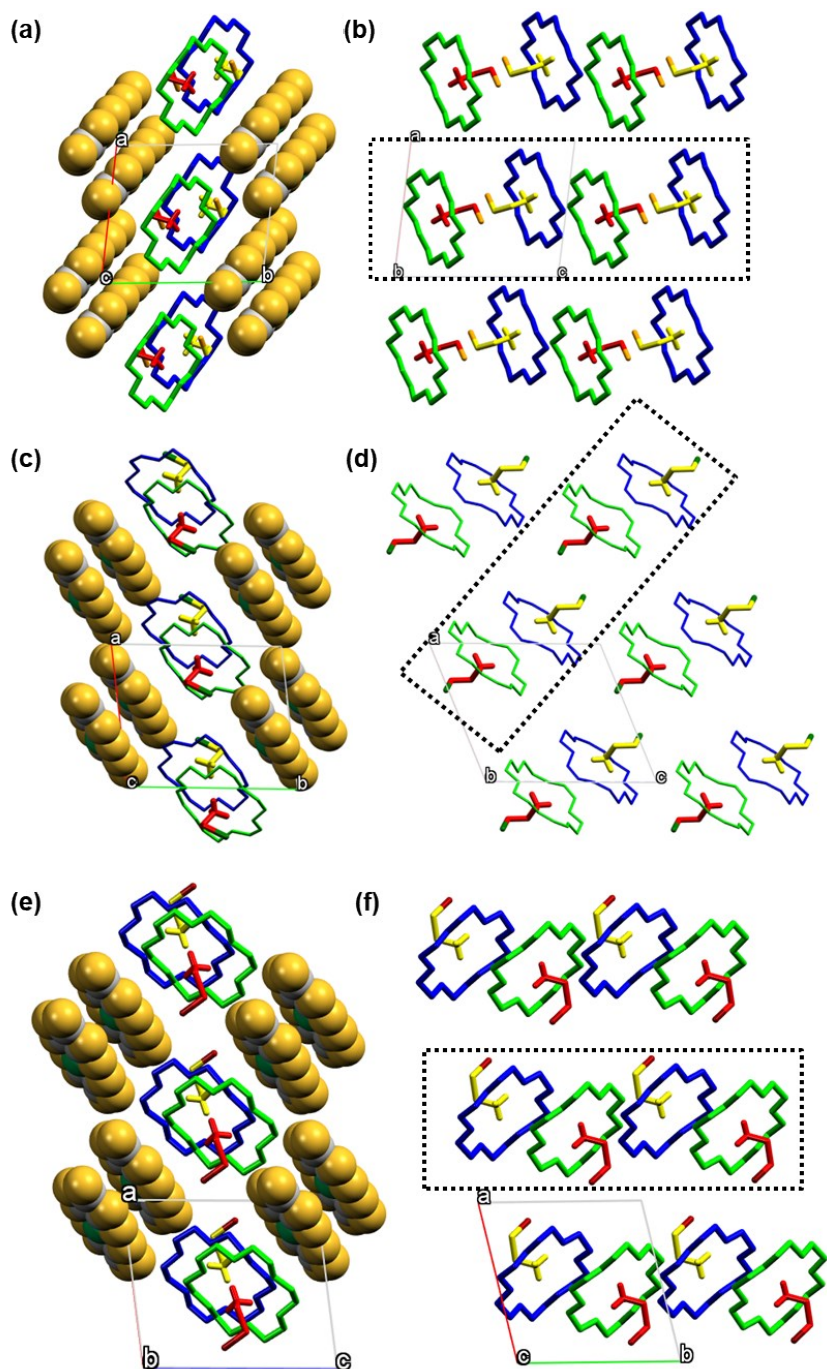


Figure S5. Packing structures of crystals 2, 3 and 4 viewed along the (a) *c*, (c) *c* and (e) *b* axis, respectively. 2D arrangement of supramolecular cations for crystals 2, 3 and 4 viewed along the (b) *b*, (d) *b* and (f) *c* axis, respectively. Dotted square encircles the 1D arrangement of supramolecular cations. Hydrogen atoms are omitted for clarity except the atoms connected to nitrogen atom. Sulfur, carbon, and nickel atoms in [Ni(dmit)₂] colored as yellow, grey, and green, respectively. Yellow X-C₂-NH₃ and blue [18]crown-6, and red X-C₂-NH₃ and green [18]crown-6 is supramolecular cation unit and the two unit form supramolecular cation pair.

§4. Detailed analysis of structures of crystal 1-4.

Table S6. Intermolecular interaction between supramolecular cations and density of the crystals at the lowest temperatures.

Crystals	^a Interaction	Å	^b $d_{c...c}$ / Å	D_{calc} / g cm ³
1	C-H...O	3.595(2)	3.726	1.590
2	C-H...O	3.633(2)	3.745	1.671
3			3.782	1.638
4	C-H...O	3.468(3), 3.477(4)	3.635	1.748

^a Intermolecular interaction between the neighboring supramolecular cations. Instead of H...O distance, distance of C...O was used for C-H...O interaction. Distance of weak hydrogen bonding is in the range of 3.2-4.0 Å between hydrogen-bonded donor...acceptor atoms.

^b Distance between two plane of neighboring [18]crown-6. The plane of [18]crown-6 used by six oxygen atom in [18]crown-6.

§5. Temperature- and frequency-dependent dielectric constants for crystals 1-4

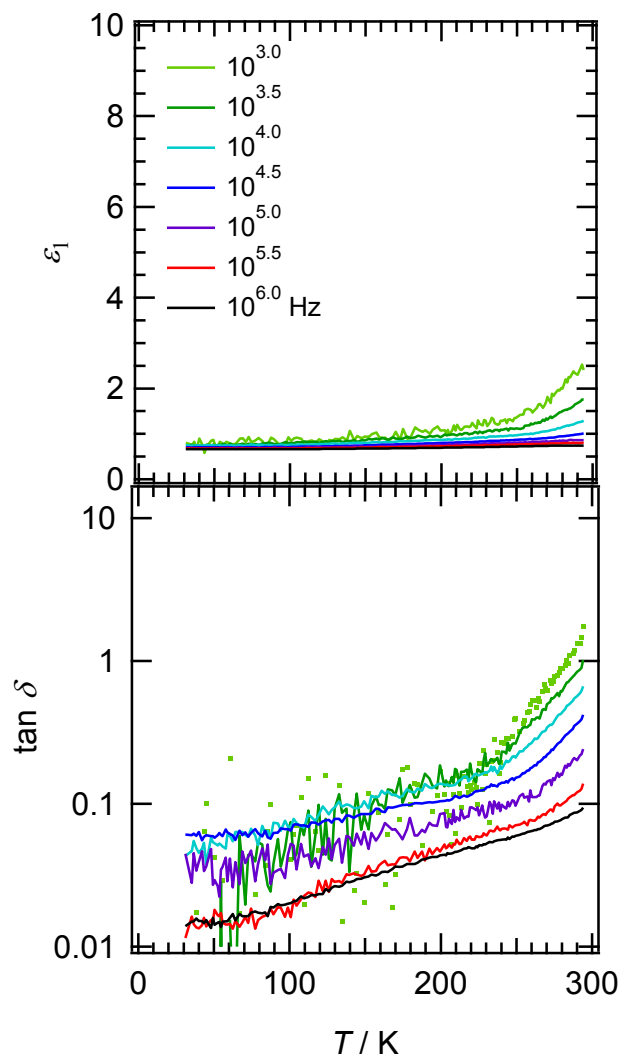


Figure S6. Temperature- and frequency-dependent real part of dielectric constant (ϵ_1) and tangent delta for crystal 1.

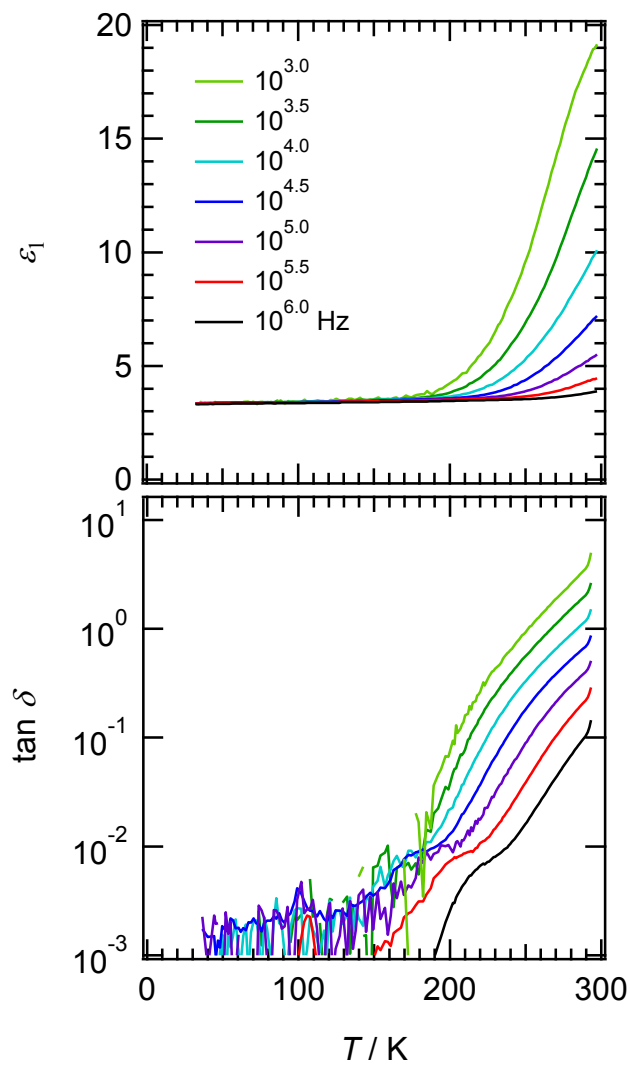


Figure S7. Temperature- and frequency-dependent real part of dielectric constant (ϵ_1) and tangent delta for crystal 2.

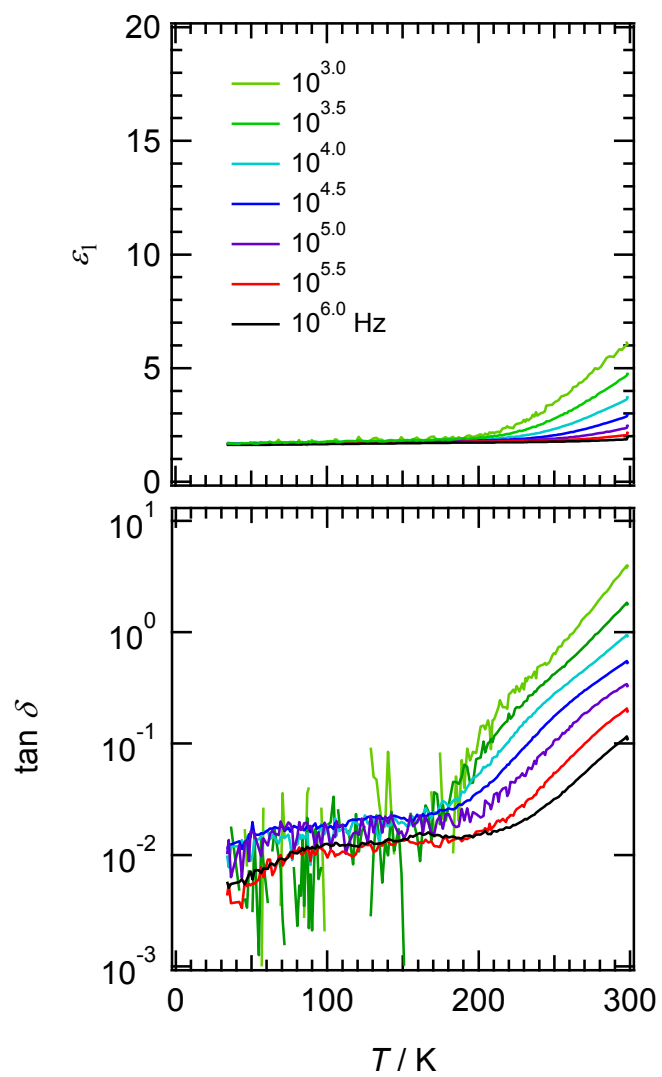


Figure S8. Temperature- and frequency-dependent real part of dielectric constant (ϵ_1) and tangent delta for crystal 3.

§6. Transfer integrals between $[\text{Ni}(\text{dmit})_2]^-$ molecules in crystals 1-4.

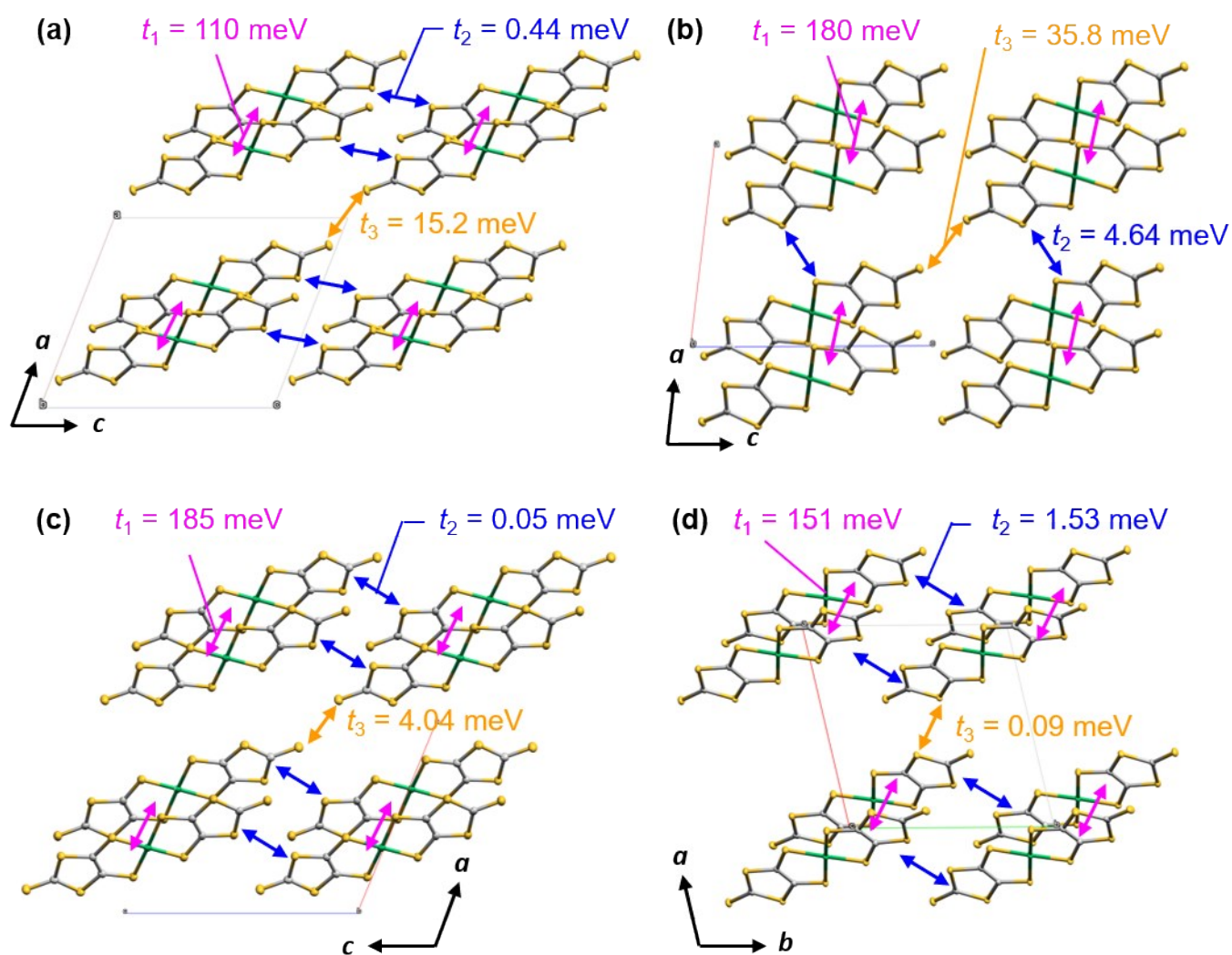


Figure S9. Transfer integral between $[\text{Ni}(\text{dmit})_2]^-$ anions in crystals (a) 1, (b) 2, (c) 3, and (d) 4. t_1 corresponds to the transfer integrals between the intradimer $[\text{Ni}(\text{dmit})_2]^-$ anions, whereas t_2 and t_3 are transfer integrals between interdimer $[\text{Ni}(\text{dmit})_2]^-$ anions.