

Assembly, structures and properties of polyoxometalate-based supramolecular complexes involving in situ transformation of single-branch N-donor cyano ligands

Xiang Wang,* Yunhui Li, Tong Zhang, Sijia Ma, Xiuli Wang*

Liaoning Professional Technology Innovation Center of Liaoning Province for Conversion Materials of Solar Cell, College of Chemistry and Materials Engineering, Bohai University, Jinzhou 121000, P. R. China

Table S1. Selected bond lengths (Å) and angles (°) of complexes **1-6**.

Complex 1			
Cu(1)-O(2)	1.963(6)	Cu(1)-N(1)	1.988(7)
Cu(1)-O(1W)	1.970(6)	Cu(1)- N(3)	1.978(7)
Cu(1)-O(2W)	2.263(6)	Cu(2)- N(4)	1.865(6)
Cu(2)-O(1)	2.367(5)	Cu(2)- N(5)	1.867(6)
O(2)-Cu(1)-O(2W)	90.4(2)	O(2)-Cu(1)-O(1W)	87.7(2)
O(2)-Cu(1)-N(1)	89.8(3)	O(2)-Cu(1)-N(3)	170.4(3)
O(1W)-Cu(1)-O(2W)	92.8(3)	O(1W)-Cu(1)-N(1)	172.8(3)
O(1W)-Cu(1)-N(3)	86.7(3)	N(1)-Cu(1)-O(2W)	94.0(3)
N(3)-Cu(1)-O(2W)	97.7(3)	N(3)-Cu(1)-N(1)	94.8(3)
N(4)-Cu(2)-O(1)	92.4(2)	N(4)-Cu(2)-N(5)	172.5(3)
N(5)-Cu(2)-O(1)	94.0(2)		
Complex 2			
Cu(1)-O(3)	2.476(6)	Cu(1)-N(1)	1.869(8)
Cu(1)-N(2)	1.863(8)	Cu(2)-N(6)	1.869(8)
Cu(2)-N(7)	1.877(8)	Cu(2)-O(15)	2.554(7)
Cu(3)-O(5)	2.613(8)	Cu(3)-N(10)	1.861(10)
Cu(3)-N(12)	1.888(11)	N(6)-Cu(2)-N(7)	176.6(4)
N(6)-Cu(2)-O(15)	91.4(3)	N(7)-Cu(2)-O(15)	90.6(3)
N(10)-Cu(3)-O(5)	98.8(4)	N(10)-Cu(3)-N(12)	166.3(5)
N(12)-Cu(3)-O(5)	90.8(4)		

Complex 3

Cu(1)-O(1W)	1.971(6)	Cu(1)-O(4)	1.966(5)
Cu(1)-O(2W)	2.261(6)	Cu(1)-N(12)	1.990(6)
Cu(1)-N(14)	1.972(7)	Cu(2)-N(2)	1.880(6)
Cu(2)-O(8)	2.375(5)	Cu(2)-N(8)	1.875(6)
O(1W)-Cu(1)-O(2W)	94.0(3)	O(1W)-Cu(1)-N(12)	87.3(3)
O(1W)-Cu(1)-N(14)	172.1(3)	O(4)-Cu(1)-O(1W)	87.3(2)
O(4)-Cu(1)-O(2W)	90.2(2)	O(4)-Cu(1)-N(12)	170.5(3)
O(4)-Cu(1)-N(14)	90.1(3)	N(12)-Cu(1)-O(2W)	97.9(3)
N(14)-Cu(1)-O(2W)	93.4(3)	N(14)-Cu(1)-N(12)	94.2(3)
N(2)-Cu(2)-O(8)	92.6(2)	N(8)-Cu(2)-N(2)	172.5(3)
N(8)-Cu(2)-O(8)	94.0(2)		

Complex 4

Cu(1)-N(1)	1.966(6)	Cu(1)-N(2)	2.003(6)
Cu(1)-O(5)	2.429(4)	Cu(1)-O(10)	2.392(4)
Cu(1)-O(14)	2.023(4)	Cu(1)-O(2W)	2.007(5)
Cu(2)-O(1W)	2.026(4)	Cu(2)-N(4)	1.999(5)
Cu(2)-N(5)	1.987(5)	Cu(2)-N(34)	2.090(5)
Cu(2)-N(39)	2.122(4)	N(1)-Cu(1)-N(2)	82.5(3)
N(1)-Cu(1)-O(5)	88.9(2)	N(1)-Cu(1)-O(10)	107.2(2)
N(1)-Cu(1)-O(14)	103.0(2)	N(1)-Cu(1)-O(2W)	168.3(2)
N(2)-Cu(1)-O(5)	96.8(2)	N(2)-Cu(1)-O(10)	114.7(2)
N(2)-Cu(1)-O(14)	171.7(2)	N(2)-Cu(1)-O(2W)	89.2(3)
O(10)-Cu(1)-O(5)	145.87(15)	O(14)-Cu(1)-O(5)	89.56(16)
O(14)-Cu(1)-O(10)	57.97(16)	O(2W)-Cu(1)-O(5)	83.8(2)
O(2W)-Cu(1)-O(10)	83.79(19)	O(2W)-Cu(1)-O(14)	86.2(2)
O(1W)-Cu(2)-N(34)	137.28(17)	O(1W)-Cu(2)-N(39)	119.37(18)
N(4)-Cu(2)-O(1W)	87.9(2)	N(4)-Cu(2)-N(34)	100.7(2)
N(4)-Cu(2)-N(39)	80.0(2)	N(5)-Cu(2)-O(1W)	89.3(2)
N(5)-Cu(2)-N(4)	176.8(2)	N(5)-Cu(2)-N(34)	80.4(2)
N(5)-Cu(2)-N(39)	102.67(19)	N(34)-Cu(2)-N(39)	103.35(17)

Complex 5

Cu(1)-N(5)	1.90(2)	Cu(1)-N(2)	1.99(2)
Cu(1)-N(4)	2.03(3)	Cu(1)-N(3)	2.07(3)
Cu(1)-N(1)	2.20(3)	N(5)-Cu(1)-N(2)	177.4(12)
N(5)-Cu(1)-N(4)	94.3(10)	N(2)-Cu(1)-N(4)	85.9(10)
N(5)-Cu(1)-N(3)	80.4(11)	N(2)-Cu(1)-N(3)	97.8(11)
N(4)-Cu(1)-N(3)	139.1(13)	N(5)-Cu(1)-N(1)	102.5(11)
N(2)-Cu(1)-N(1)	79.8(11)	N(4)-Cu(1)-N(1)	111.1(12)
N(3)-Cu(1)-N(1)	109.7(11)		

Complex 6

Co(1)-O(13)	2.128(15)	Co(1)-O(1W)	2.09(2)
Co(1)-N(1)	2.09(2)	Co(1)-N(4)	2.123(19)
Co(1)-N(3)	2.143(17)	Co(1)-O(2W)	2.053(18)
O(13)-Co(1)-N(3)	94.4(7)	O(1W)-Co(1)-O(13)	85.1(8)
O(1W)-Co(1)-N(4)	95.3(8)	O(1W)-Co(1)-N(3)	171.1(7)
N(1)-Co(1)-O(13)	169.7(7)	N(1)-Co(1)-O(1W)	88.5(8)
N(1)-Co(1)-N(4)	102.5(7)	N(1)-Co(1)-N(3)	93.2(7)
N(4) -Co(1)-O(13)	86.1(6)	N(4) -Co(1)-N(3)	75.9(7)
O(2W)-Co(1)-O(13)	85.0(6)	O(2W)-Co(1)-O(1W)	89.4(9)
O(2W)-Co(1)-N(1)	86.9(7)	O(2W)-Co(1)-N(4)	169.6(7)
O(2W)-Co(1)-N(3)	99.4(8)		

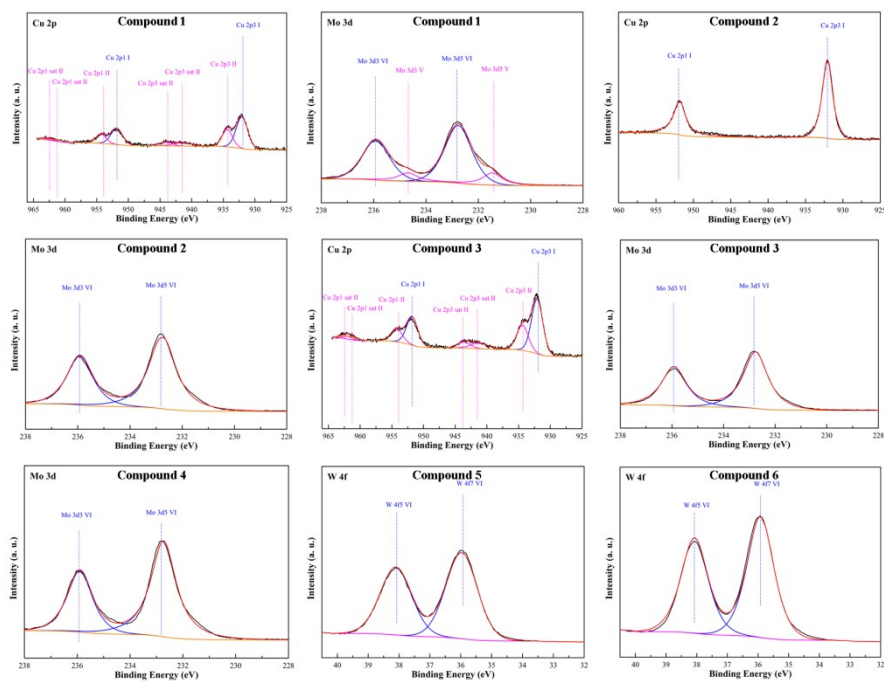


Figure S1 The XPS patterns of complexes 1-6.

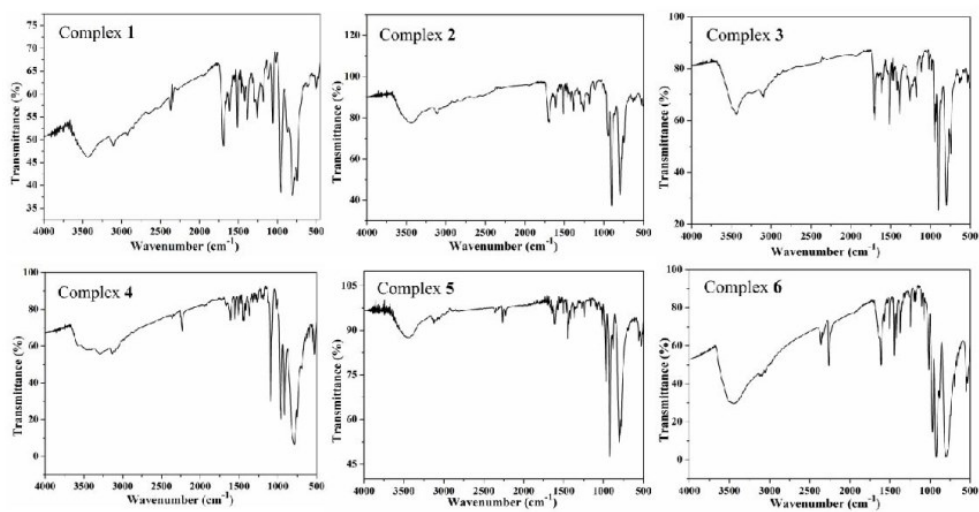


Figure S2 IR spectra of complexes 1-6.

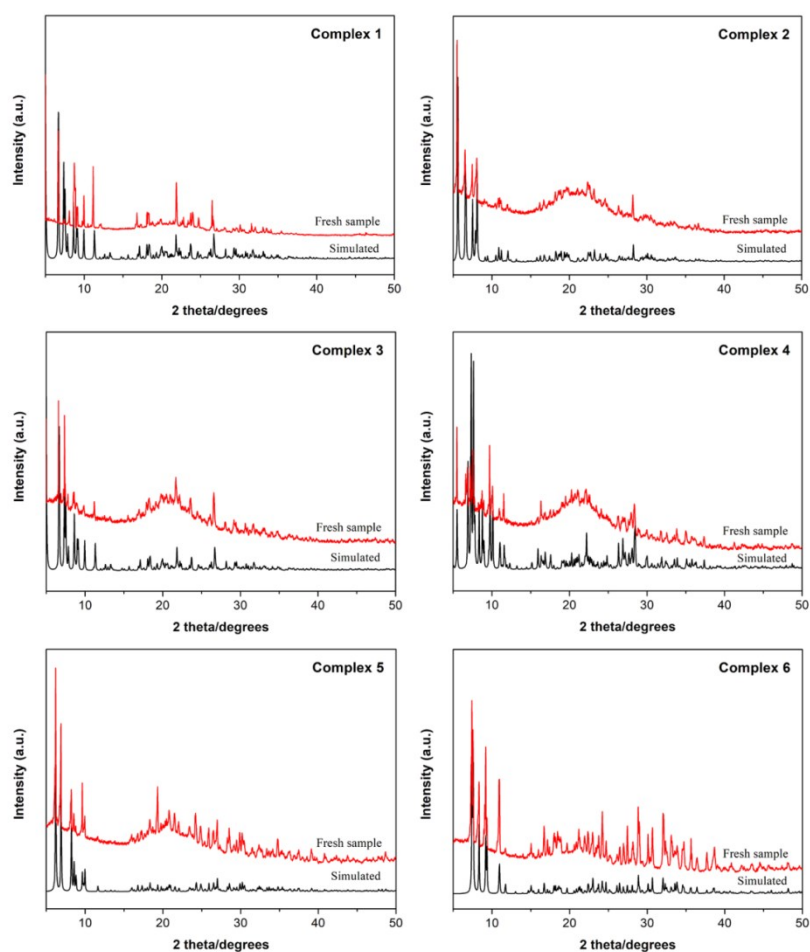


Figure S3 The PXRD patterns for complexes 1-6.

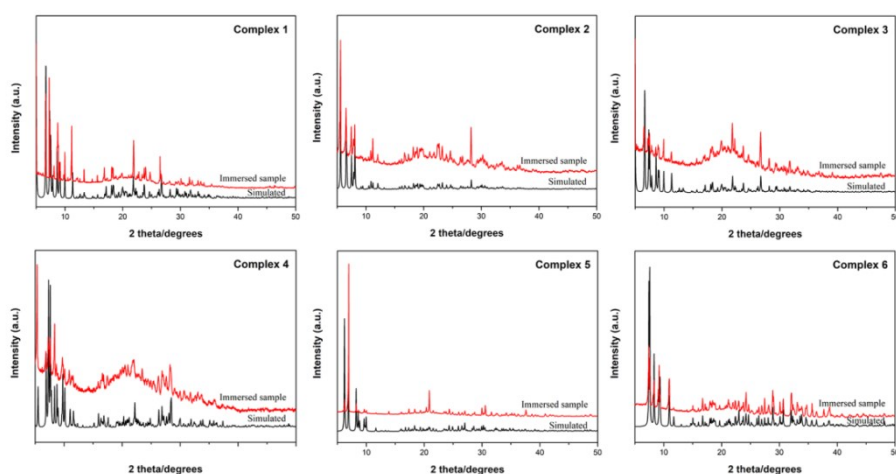


Figure S4 The PXRD patterns of 1-6 after immersion in 0.1 M H_2SO_4 + 0.5 M Na_2SO_4 aqueous solution.

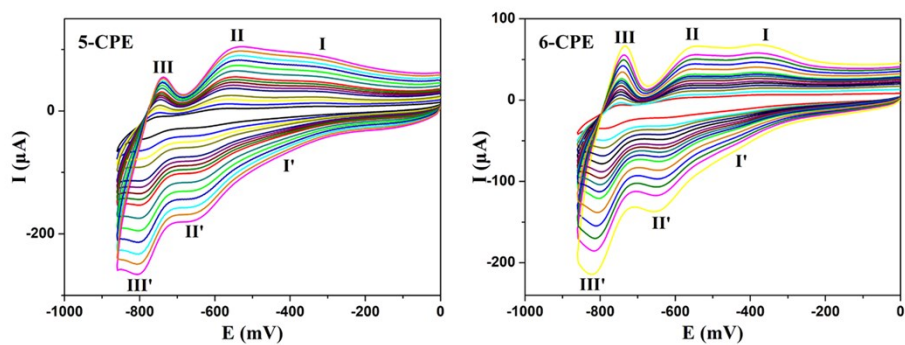


Figure S5 Cyclic voltammograms of **5**– and **6**–CPEs in 0.1 M H₂SO₄ + 0.5 M Na₂SO₄ aqueous solution at different scan rates.

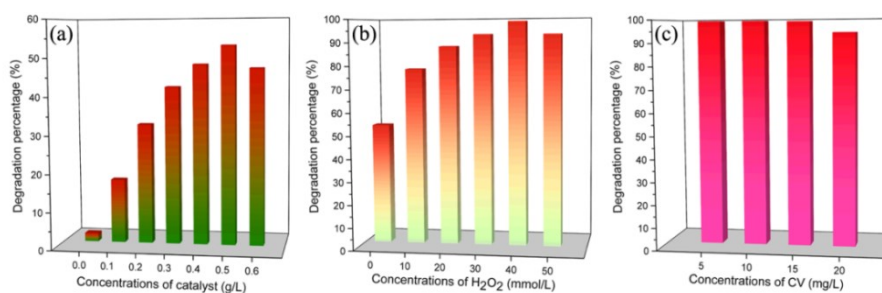


Figure S6 The influences on the degradation efficiencies of the concentrations of catalyst (a), H₂O₂ (b) and dye (c).

Table S2. The optimized concentrations of catalyst, H₂O₂ and dyes.

Complex	CV (10 mg/L)		MB (10 mg/L)	
	Catalyst (g/L)	H ₂ O ₂ (mmol/L)	Catalyst (g/L)	H ₂ O ₂ (mmol/L)
1	0.5	40	0.5	40
2	0.7	50	0.7	50
3	0.5	40	0.5	40
4	0.5	40	0.5	40
5	0.5	40	0.5	50
6	0.7	60	0.7	60

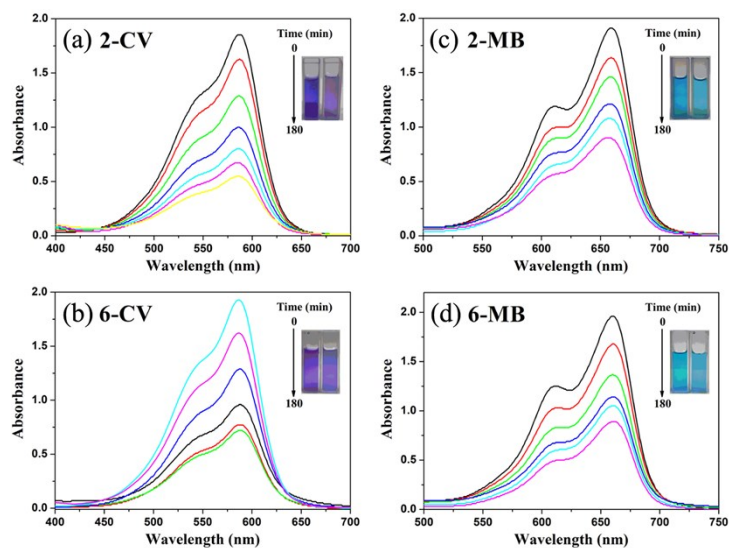


Figure S7 Absorption spectra of the CV (a-b) and MB (c-d) solutions under UV irradiation in the presence of complexes **2** and **6**.

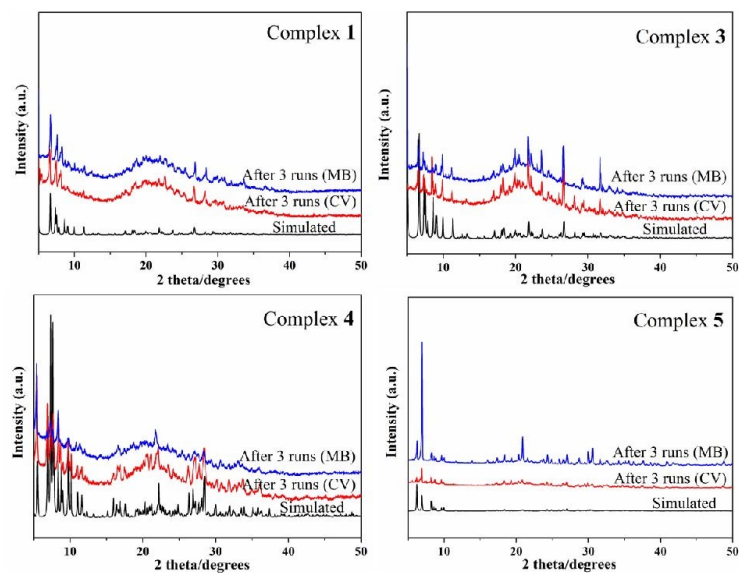


Figure S8 The PXRD patterns of simulated and recycled after acting as catalysts of **1**, **3**, **4** and **5**.